

# Potassium contributes to zinc stress tolerance in peach (*Prunus persica*) seedlings by enhancing photosynthesis and the antioxidant defense system

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**ABSTRACT.** Zinc (Zn) is considered to be a major industrial pollutant because excessive amounts can impair plant growth. In this paper, we found that peach 'Yoshihime' seedlings are promising Zn tolerant plants. However, heavy Zn toxicity (2 mM) damaged plant performance by disrupting biochemical processes, including photosynthesis, proline production, and K<sup>+</sup> nutrition. Notably, elevated external K<sup>+</sup> supply (10 mM) alleviated peach seedlings from Zn toxicity, evidenced by enhanced photosynthesis, antioxidant defense systems, and plant K<sup>+</sup> nutritional status. Moreover, the transcript levels of *KUP* (K<sup>+</sup> uptake) genes involved in K<sup>+</sup> acquisition, transport, and homeostasis were

significantly upregulated following supply of sufficient  $K^+$  upon Zn toxicity. In general,  $K^+$  favorably contributes to improvements in internal  $K^+$  homeostasis, via the help of  $K^+$  transporters, further protecting plant photosynthesis and the antioxidative defense system. Our findings further benefit the study of the mechanisms underpinning heavy metal tolerance in woody plants.

**Key words:** Potassium; Photosynthesis; Antioxidant defense system; Peach; Zinc stress

#### **INTRODUCTION**

High concentrations of heavy metals can adversely affect the growth, fruit formation, and development of fruit trees (Worthington, 2001; Lombardi and Sebastiani, 2005; Foyer and Noctor, 2005). Yet, zinc (Zn) sprays at critical periods can improve the fruit quality of apple trees (Zhang et al., 2013), and foliar application of Zn contributes to soluble carbohydrate and oil contents in olive cultivars during fruit ripening (Saadati et al., 2013). However, due to its high mobility and bioavailability, Zn is considered to be a major industrial pollutant of both the terrestrial and aquatic environments (Barak and Helmke, 1993; Morillo et al., 2004). Excessive amounts of Zn cause leaf chlorosis, nutrient imbalance, and inhibition of photosynthesis at various stages, resulting in damage to plant growth (Kabata-Pendias and Pendias, 2001; Morillo et al., 2004; Street et al., 2007). Common plant seedlings, such as rice, *Arabidopsis*, and maize, experienced mortality under 0.5 mM Zn stress (Song ZZ, Guo SL, Ma RJ and Yu ML, unpublished results). Despite this, certain plant species, such as *Silene vulgaris* and *Halimione portulacoides*, have evolved a tolerance to Zn that enables them to grow in Zn-contaminated soils, remaining unaffected by the elevated Zn levels (Ernst et al., 2000; Cambrollé et al., 2012).

Plants suffering from adverse abiotic stresses, including heavy-metal stress, produce excess internal reactive oxygen species (ROS), such as superoxide  $(O_2^{-})$ , hydrogen peroxide  $(H_2O_2)$ , and hydroxyl radicals (·OH<sup>-</sup>), which negatively elicit oxidative stress on cellular structures and metabolism (Foyer and Noctor, 2005; Turchi et al., 2012; Singh et al., 2012; Song et al., 2014a). In particular, plants can cope with oxidative stress via the antioxidant defense system. The activity of antioxidant enzymes, i.e., superoxide dismutase (SOD), peroxidase (POD), catalase (CAT), and ascorbate peroxidase (APX), are correlated with plant tolerance to abiotic stresses (Wang et al., 2009; Huang et al., 2013). Previous reports illustrate that heavy metal toxicity results in oxidative damage to plants and that heavy metal tolerant plants possess high antioxidant enzyme activity (Turchi et al., 2012; Singh et al., 2012). In particular, proline accumulates heavily under stresses, providing the plants protection against damage by ROS, which helps to protect enzyme activities, stabilize protein synthesis machinery, or regulate cytosolic acidity (Choudhary et al., 2007).

As one of the most abundant cations in cells, potassium ( $K^+$ ) not only contributes to plant growth, development, and fruit quality and yield (Maathuis and Sanders, 1996; Véry and Sentenac, 2003; Demiral and Köseoglu, 2005; Hartz et al., 2005), but it also favorably improves plant tolerance to various abiotic stresses, such as drought (Li et al., 2011; Song and Su, 2013; Song et al., 2014a), salt (Li et al., 2011; Mian et al., 2011; Song et al., 2014b), and cold (Rai et al., 2008). However, studies of K<sup>+</sup> in relation to plant tolerance to heavy metals have not been undertaken to date.

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In particular, the KT/HAK/KUP family of transporters, as part of the high-affinity K<sup>+</sup> transport system, are effective in transporting external K<sup>+</sup>, via involvement in root hair elongation or cell expansion growth (Véry and Sentenac, 2003; Grabov, 2007; Alemán et al., 2011; Coskun et al., 2013). Notably, K<sup>+</sup> transporters in the grape berry (*Vitis vinifera* L.) were involved in potassium accumulation in fruit (Davies et al., 2006). Since the first plant member of the KT/HAK/KUP family was identified in barley (*Hordeum*) (Santa-Maria et al., 1997), subsequent studies have been conducted on evolutionarily diverse organisms (mentioned in Grabov, 2007).

In previous studies, we identified 16 putative KUP (K<sup>+</sup> uptake) transporters in the peach genome (Song et al., 2015). In this present study, we evaluated the physiological response of peach seedlings exposed to various external Zn stress levels (ranging from 0.015-2 mM) and subsequently estimated the capability of improved Zn tolerance in relation to elevated K<sup>+</sup> supply. This present study directly indicates that K<sup>+</sup> contributes to improving plant tolerance to heavy-metal stress, which provides evidence for evaluating K<sup>+</sup> for its potential role in phytoremediation of increasing heavy metal polluted waterbodies.

#### **MATERIAL AND METHODS**

## Plant material and culture conditions

Peach (*Prunus persica* cv. 'Yoshihime') seedlings were used throughout this study. Seed kernels were washed with distilled water and germinated in soil in a greenhouse. Germinated seedlings of a similar growth status were transferred from soil to 1/2 MS solution (containing 0.015 mM Zn<sup>2+</sup>, Murashige and Skoog, 1962), and exposed to different treatments in aerated plastic containers. Greenhouse conditions were maintained at 25° ± 2°C and 12/12h light/dark, with 60% relative humidity. The nutrient solution was refreshed every other day.

#### **Stress treatments**

For the  $Zn^{2+}$  toxicity treatments, we carried out three parallel sets of experiments using 8 peach seedlings in each. The final  $Zn^{2+}$  concentration was adjusted to the desired amounts of 0.015 (the control, present in 1/2 MS basic media), 0.1, 0.5, or 2 mM.

To determine the role of external K<sup>+</sup> supply in plant performance under Zn toxicity, three parallel sets of experiments using 8 peach seedlings each were exposed to 2 mM ZnCl<sub>2</sub>, supplied with 1 (1K+Zn treatment) or 10 mM (10K+Zn treatment) KCl, respectively. All treatments were maintained for 10 days (for phenotype assay) or 21 days (for gene expression and antioxidant enzyme activity analysis) prior to analysis.

#### **Physiological measurements**

Seedlings were rinsed in distilled water, soaked, and then weighed to obtain fresh weight. The roots were scanned with an Epson Rhizo scanner (RHIZO 2004b, Long Beach, CA, USA), and total root length and surface area were acquired using the Epson WinRHIZO software installed in the Rhizo scanner. The seedlings were separated into roots, stems, and leaves, dried in an oven at 105°C for 30 min, and then baked at 70°C for 48 h to obtain the dry weight of the biomass. The dried samples were ground into fine powder and were fully di-

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gested with HNO<sub>3</sub>-HClO<sub>4</sub>, and K<sup>+</sup> and Zn<sup>2+</sup> concentrations were measured using flame atomic absorption spectrometry. Photosynthetic analysis was carried out on a portable photosynthetic system LI-6400 (Li-COR, Lincoln, NE, USA) to determine net photosynthetic rate ( $P_N$ ) and stomatal conductance ( $g_3$ ) at the terminal leaflet of fully grown second leaves, as described by Kumar et al. (2006). Chlorophyll and proline determination were carried out according to the description of Song et al. (2014a). The total protein concentration of leaf samples was determined according to the method specified by Bradford (1976), and was then used immediately for antioxidant enzyme assays. SOD, POD, CAT, and APX activities, and H<sub>2</sub>O<sub>2</sub> level were determined using the relevant detection kits (Jiancheng Bioengineering Institute, Nanjing, China), according to the manufacturer protocol.

#### Isolation of candidate genes involved in the antioxidant defense system

Sequences of genes encoding putative antioxidant enzymes used in this study were obtained by querying the protein sequences of *Arabidopsis* SOD (AT4G25100 and AT1G08830), POD (AT4G30170), CAT (AT1G20630), or APX (AT1G07890) against the BLAST peach genome in the Genome Database for Rosaceae. Successfully, we obtained 4 putative SOD (*PpeSOD1-4*), 3 POD (*PpePOD1-3*), 1 CAT (*PpeCAT1*), and 4 APX (*PpeAPX1-4*) encoding genes, respectively.

## **RNA extraction and quantitative real-time PCR (qRT-PCR)**

Total RNA was extracted from leaves, stems, and roots using a Plant RNA Kit (Omega, Norcross, GA, USA) and was reverse transcribed into cDNA using a reverse transcription kit (TaKaRa, Tokyo, Japan). qRT-PCR was carried out in a 7500 Real Time PCR System (Applied Biosystems, USA). Specific primers for genes and the control gene *ubiquitin* in peach were designed using the NCBI/Primer-BLAST online server (listed in Table 1). PCR conditions for thermal cycling were as follows: 95°C for 30 s, 40 cycles of 95°C for 5 s and 60°C for 30 s. The relative expression levels of the target genes were presented after normalization to the internal control from four independent biological repeats.

#### Statistical and graphical analysis

Data were statistically analyzed using the SPSS 13.0 software (SPSS Chicago, IL, USA). Details are provided in both the text and figure legends. Graphs were produced using the Origin 8.0 software.

#### RESULTS

#### Physiological responses of peach seedlings to external Zn stress

Peach, as a perennial woody plant, exhibits good Zn tolerance and can survive in 0.5 mM  $Zn^{2+}$  stress (Figure 1). Compared to control conditions (0.015 mM  $Zn^{2+}$ ), shoot growth and root elongation of peach seedlings were inhibited after being subjected to 2 mM Zn toxicity (Figure 1a), evidenced by dramatically reduced total fresh weight, total root length, and surface area (Figure 1b-d). Notably, Zn toxicity also significantly reduced leaf chlorophyll

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Table 1. Primer pairs used for quantitative RT-PCR experiments in peach seedlings exposed to zinc toxicity.

Gene	Pfam annotation	Primer (5' to 3')	Amplicon size (bp)
PpeSOD1	Putative iron superoxide dismutase	F: GAAGGGCGTTGCTGTTCTTG	163
(ppa008765m)	I diative non superoxide distributase	R: GCAACCGTTGGTTGTGTCTC	105
	Putative copper/zinc superoxide dismutase		163
PpeSOD2	Putative copper/zinc superoxide distilutase	F: GAAGGGCGTTGCTGTTCTTG	105
(ppa012845m)		R: GCAACCGTTGGTTGTGTCTC	1(0
PpeSOD3	Putative copper/zinc superoxide dismutase	F: CCCAGATGGAGTTGCTGAGG	160
(ppa012725m)		R: TCTTGCTCCTGCATTCCCAG	1.77
PpeSOD4	Putative copper/zinc superoxide dismutase	F: GGCCTCTCTGTTAAGCTCCC	177
(ppa009729m)		R: CACAGTTGTAGGCCCATCGT	
PpePOD1	Putative peroxidase precursor	F: TACGCATTGCAGCTCAGACA	193
(ppa008569m)		R: CCAACTTGGTCATGGCTTGC	
PpePOD2	Putative peroxidase precursor	F: AGGGTTGTGATGCTTCGGTT	249
(ppa008489m)		R: CCGTCCAATCTCCCCAACTC	
PpePOD3	Putative peroxidase precursor	F: AAATGGGGACGCAGAGAAGG	232
(ppa015245m)		R: AACGACCTGAGAGGCTTGTG	
PpeCAT1	Putative catalase protein	F: CACATGGAAGGCTCTGGTGT	160
(ppa004776m)	*	R: CCTTGGTAGCATGGCTGTGA	
PpeAPXI	Putative ascorbate peroxidase	F: GGTCACACTCTGGGAAGGTG	182
(ppa010431m)		R: TCAACAAGAGGGCGGAAGAC	
PpeAPX2	Putative ascorbate peroxidase	F: TGGATATTGCGGTCAGGCTC	159
(ppa010426m)	i unitte ascolonic peroxicase	R: GTGGCGGTTCTTGTTTGTCC	109
PpeAPX3	Putative ascorbate peroxidase	F: TACGACGCCAAGACGAAGAC	181
(ppa009538m)	I utative ascorbate peroxidase	R: GCTCAACTGCAACAACACCC	101
	Dutativa agaanhata manavidaga	F: CCAGATGCCAAACAAGGTGC	154
PpeAPX4	Putative ascorbate peroxidase		134
(ppa009582m)	Destations water a investment and an	R: AAGGCTCATTCGTCCAAGGG	224
PpeKUP1	Putative potassium transporter	F: AGGCGTTGAAGGATGGGTTT	234
(ppa025447m)		R: GGCCAAAACACAGCTTCTGG	
PpeKUP2	Putative potassium transporter	F: AGCGGAGGAAGGAGGTTAGT	165
(ppa002879m)		R: GTGATACCCCACTTGCCACA	
PpeKUP3	Putative potassium transporter	F: TAGAGGAGACTCAACCGGCA	213
(ppa001638m)		R: CGAGTATAAACGCCGTCCCA	
PpeKUP4	Putative potassium transporter	F: CCGACCCAACGAAGGAAGAA	240
(ppa001724m)		R: TACGCTGAGTGCATAGCTGG	
PpeKUP5	Putative potassium transporter	F:TGGGTCCCCATTGCTCTTTC	152
(ppa003114m)		R: CGGACCCTAACAATCCCCAG	
PpeKUP6	Putative potassium transporter	F: AGTCTGTGAGCTTGGTGAGC	196
(ppa022818m)		R: AAGTACGCTGTGCCACTCTC	
PpeKUP7	Putative potassium transporter	F: GTGCATCGTCAGGTATGGGT	235
(ppa001652m)	* *	R: GATGGAGTCGCTCACACCAA	
PpeKUP8	Putative potassium transporter	F: AACTCCTGAAGGTGCTGTGG	182
(ppa022284m)		R: TAGGTGGCCTGCAAGTTCTG	
PpeKUP9	Putative potassium transporter	F: GACAATCAGAGCACCGGGAA	181
(ppa001335m)	i uutive poussium nunsporter	R: GACACGCCGGAAAAGAAACC	101
PpeKUP10	Putative potassium transporter	F: TTGTCCAAGGCATCCCATCC	178
(ppa001373m)	i dative potassium transporter	R: CGGGCTATACAGCGGAACAT	170
	Dutative notacium transportar		201
PpeKUP11	Putative potassium transporter	F: CGGAAAGATTGGCAACGCAT	201
(ppa016701m)		R: AGAGGAAGGAAACCGCCTTG	1.50
PpeKUP12	Putative potassium transporter	F: CATACCTTCCGTCCACTCCG	159
(ppa015062m)		R: TCAACCCTGTCGTTGTAGCC	22 (
PpeKUP13	Putative potassium transporter	F: GGCATACACTGGACAAGCCT	226
(ppa001648m)		R: CCTGGCCGATGTGTGCACGA	
PpeKUP14	Putative potassium transporter	F: TGCCTGCGATCCATTCTGTT	249
(ppa001603m)		R: CATCCGAGTCTGAACACCCC	
PpeKUP15	Putative potassium transporter	F: AATTGGCAATGCCTCTGGGA	204
(ppa001626m)		R: GCGATCACAAGAGGAACCCA	
PpeKUP16	Putative potassium transporter	F: GTCAATCAAGGTGGGTGGGT	246
(ppa016662m)		R: GGCGGGAAGGTTGGTGATAA	
Ubiquitin	Ubiquitin	F: GGCTAAGATCCAGGACAAGGAG	145
(ppa005503m)	•	R: CCACGCAGACGGAGCACCAAG	

Primer sequences of putative *KUP* antioxidant enzyme encoding genes were obtained using a query protein sequence of *Arabidopsis* KUP1 (AT2G30070), SOD (AT4G25100 and AT1G08830), POD (AT4G30170), CAT (AT1G20630) or APX (AT1G07890), respectively, against the BLAST Peach genome at Genome Database for Rosaceae.

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concentration (Figure 2a),  $P_N$  (Figure 2b), and  $g_s$  (Figure 2c), but enhanced leaf proline concentration (Figure 2d). Simultaneously,  $Zn^{2+}$  accumulation was reinforced with increasing external  $Zn^{2+}$  supplies, especially in the aboveground parts of the seedlings (Figure 2e). In contrast, K<sup>+</sup> accumulation was simultaneously decreased, especially in the aboveground parts of the seedlings (Figure 2f).

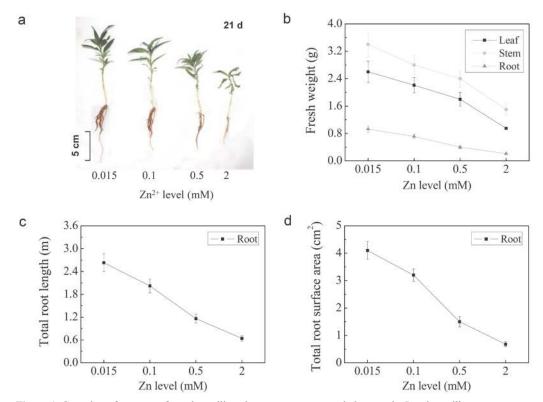


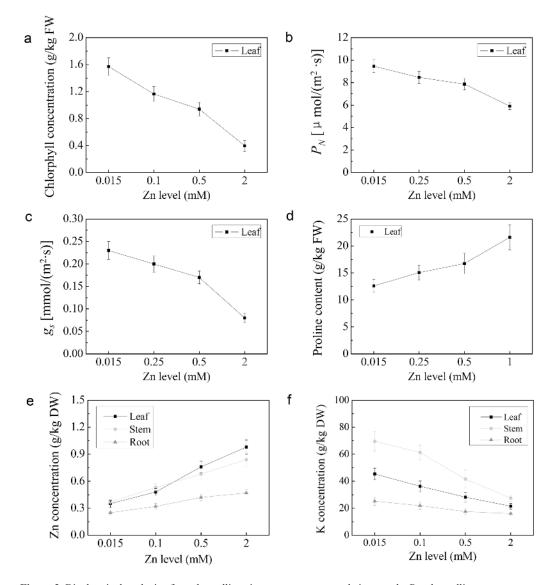
Figure 1. Growth performance of peach seedlings in response to external zinc supply. Peach seedlings were grown in a 1/2 MS solution supplemented with 0.015, 0.1, 0.5, or 2 mM ZnCl<sub>2</sub> for 21 days before analysis. **a.** Phenotype; **b.** total fresh weight; **c.** total root length and surface area; **d.** leaf chlorophyll concentration. The values are reported as means  $\pm$  SE (N = 24), calculated from 3 sets of experiments.

#### Elevated K<sup>+</sup> supply contributes to tolerance to zinc toxicity

Compared with control conditions, the addition of 2 mM Zn<sup>2+</sup> (1K+Zn treatment) caused severe damage (Figure 3), evidenced by reduced total fresh biomass (approximately 68% loss, P < 0.01, *t*-test), whereas the application of 10 mM K<sup>+</sup> (10K+Zn treatment) protected the seedlings from such damage, resulting in a significant increase in fresh biomass (P < 0.01, *t*-test) and root elongation (P < 0.01, *t*-test), accompanied by increased chlorophyll concentration in the leaves (0.01 < P < 0.05, *t*-test), increased  $g_s$  (P < 0.01, *t*-test), and increased  $P_N$  (P < 0.01, *t*-test) (Table 2). Compared with the 1K+Zn treatment, elevated K<sup>+</sup> supply (10K+Zn) also significantly reduced tissue Zn<sup>2+</sup> concentration (33% at the whole-plant

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level, P < 0.01, *t*-test) but enhanced K<sup>+</sup> accumulation (25%, 0.01 < P < 0.05, *t*-test) (Table 2). Moreover, elevated K<sup>+</sup> supply (10 mM K<sup>+</sup>) significantly reduced proline concentration (approximately 30%, 0.01 < P < 0.05, *t*-test) under 2 mM Zn toxicity, compared to that of the 1 mM K<sup>+</sup> supply (Table 2).



**Figure 2.** Biochemical analysis of peach seedlings in response to external zinc supply. Peach seedlings were grown in a 1/2 MS solution supplemented with 0.015, 0.1, 0.5, or 2 mM ZnCl<sub>2</sub> for 21 days before analysis. **a.** Leaf chlorophyll concentration; **b.** photosynthetic rate  $(P_N)$ ; **c.** stomatal conductance  $(g_s)$ ; **d.** proline content; **e.** tissue zinc concentration; **f.** tissue potassium concentration. The values are reported as means  $\pm$  SE (N = 24), calculated from 3 sets of experiments.

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Zinc stress tolerance in peach seedlings

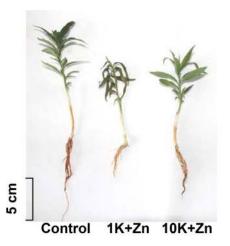


Figure 3. Morphology of peach seedlings under zinc toxicity and K<sup>+</sup> supply. Seedlings were treated for 10 days prior to analysis. +Zn, means 2 mM ZnCl, stress; 1 or 10 K, indicates 1 or 10 mM K<sup>+</sup>.

Seedling property	Treatments			
	Control (1K)	1K+Zn	10K+Zn	
Total fresh weight (g)	$5.9 \pm 0.4 **$	$1.9 \pm 0.1$	3.6 ± 0.3**	
Total root length (m)	$2.6 \pm 0.2 **$	$0.6 \pm 0.1$	$1.3 \pm 0.1 **$	
Total root surface area (cm <sup>2</sup> )	$4.1 \pm 0.3 **$	$0.7 \pm 0.1$	$1.9 \pm 0.1*$	
Chlorophyll content (g/kg FW)	$1.6 \pm 0.1$	$1.1 \pm 0.1$	$1.4 \pm 0.1*$	
$P_{\rm v}$ [µmol·m <sup>-2</sup> ·s]	$9.3 \pm 0.3$	$5.9 \pm 0.2 **$	$8.2 \pm 0.4$	
g [mmol·m <sup>-2</sup> ·s]	$0.2 \pm 0.00$	$0.1 \pm 0.02 **$	$0.2 \pm 0.01$	
Total K <sup>+</sup> content (g/kg DW)	$52.7 \pm 4.6 **$	$29.4 \pm 6.2$	$36.7 \pm 6.3*$	
Total Zn <sup>2+</sup> content (g/kg DW)	$0.3 \pm 0.05$	$0.7 \pm 0.08 **$	$0.4 \pm 0.06$	
Proline content (g/kg FW)	$14.6 \pm 1.2^*$	$11.2 \pm 0.9*$	$8.4 \pm 0.7$	
H <sub>2</sub> O <sub>2</sub> content (mmol/kg FW)	$0.2 \pm 0.04$	$0.8 \pm 0.09 **$	$0.5 \pm 0.06^{*2}$	

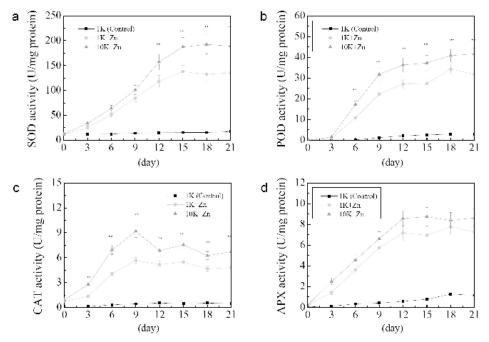
Seedlings were subjected to treatments for 21 days prior to examination. 1K+Zn, 2 mM K<sup>+</sup> plus 2 mM Zn<sup>2+</sup>; 10K+Zn, 10 mM K<sup>+</sup> plus 2 mM Zn<sup>2+</sup>;  $P_N$ , photosynthetic rate;  $g_s$ , stomatal conductance. The values are reported as means  $\pm$  SE, N = 24. \*Significant difference among treatments (0.01 < \*P < 0.05, \*\*P < 0.01, independent samples *t*-test).

#### Escalated K<sup>+</sup> supply enhances activity of antioxidant enzymes under Zn toxicity

Considering that the "Control+10K" treatment caused little effect to either growth status or physiological processes, including photosynthesis and antioxidant enzyme activities (SOD, POD, CAT, and APX), compared to the control (1 mM K<sup>+</sup>) treatment (Song ZZ, Guo SL, Ma RJ and Yu ML, unpublished results), we carried out further studies focused on the application of normal K<sup>+</sup> (1 mM K<sup>+</sup>) and escalated K<sup>+</sup> (10 mM K<sup>+</sup>) supplies under high Zn toxicity, to illuminate whether external K<sup>+</sup> has any practical means in improving plant tolerance to Zn stress. Compared with the 1K+Zn (2 mM Zn toxicity supplied with 1 mM K<sup>+</sup>) treatment, elevated K<sup>+</sup> supply (10 mM) significantly reduced endogenous H<sub>2</sub>O<sub>2</sub> concentration, with approximately 38% loss (Table 2, P < 0.01, *t*-test). Meanwhile, significantly higher activities of endogenous SOD, POD, and CAT were observed in the leaves of plants exposed to the 10K+Zn treatment, especially from day 15 to day 21, compared with that of the 1K+Zn treat-

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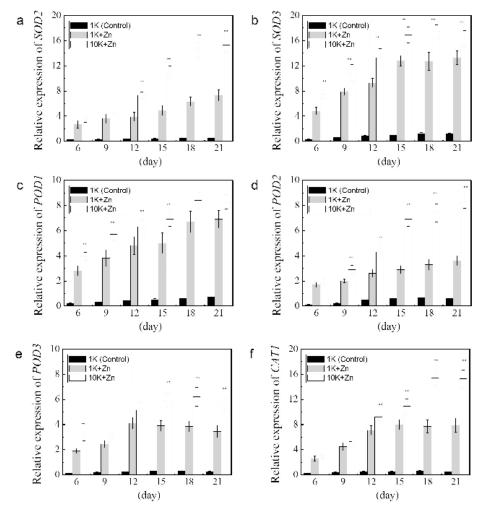
ment (Figure 4a-c, 0.01 < P < 0.05, *t*-test). However, no significant difference in APX activity was detected between the additions of distinct K<sup>+</sup> supplies (Figure 4d).



**Figure 4.** Activity of antioxidant enzymes in peach seedlings under zinc toxicity and K<sup>+</sup> supply. Seedlings were subjected to treatments for 21 days and leaves were collected for analysis at desired times. **a.** SOD activity; **b.** POD activity; **c.** CAT activity; **d.** APX activity. +Zn, means 2 mM ZnCl<sub>2</sub> stress; 1 or 10 K, indicates 1 or 10 mM K<sup>+</sup>. The values are reported as means  $\pm$  SE (N = 24), calculated from 3 sets of experiments. \*Statistical differences between seedlings grown under 1 mM ZnCl, supplied with 1 and 10 mM KCl (\*\*P < 0.01, independent samples *t*-test).

To elucidate the molecular mechanisms underlying the relationship between antioxidant enzyme activities and Zn toxicity, transcriptional profiles of putative antioxidation-related genes in peach leaves were analyzed in response to Zn toxicity. The genes encoding SOD, POD, and CAT had higher expression levels in peach leaves, with different dynamics, under the 10K+Zn treatment compared to the 1K+Zn treatment. In particular, expression levels of the SOD2, SOD3, POD2, POD3, and CAT1 genes were significantly higher in peach leaves exposed to the 10K+Zn treatment than in leaves exposed to the 1K+Zn treatment (Figure 5, 0.01 < P < 0.05, P < 0.01, t-test), especially between days 15 to 21. However, no obvious difference was observed in the transcript levels of the SOD1, SOD4, and PpeAPX genes (data not shown). In detail, expression of SOD2 and POD2 was slightly increased under the 1K+Zn treatment within the time points tested, but was dramatically enhanced in leaves under the 10K+Zn treatment (Figure 5a and d, 0.01 < P < 0.05, t-test). Notably, expression of SOD3, POD3, and CAT1 was upregulated in plant leaves under the 1K+Zn treatment at the early stage but changed little after 15 days, whereas expression of these genes steadily increased in leaves after being exposed to the 10K+Zn treatment within the first 18 days (Figure 5b, e, and f,  $P \le 0.01$ , *t*-test). Although *POD1* expression steadily increased under both treatments, it was slightly higher in leaves under the 10K+Zn treatment than the 1K+Zn treatment at the several tested time points (Figure 5c).

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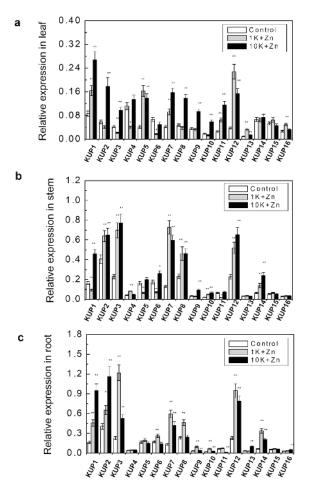
**Figure 5.** Relative expression of genes involved in the antioxidant defense system of peach seedlings under zinc toxicity. **a.** *SOD2*, **b.** *SOD3*, **c.** *POD1*, **d.** *POD2*, **e.** *POD3*, and **f.** *CAT1*. All treatments were maintained for 21 days prior to analysis. Expression values are given as a ratio relative to the values of actin. +Zn, means 2 mM ZnCl<sub>2</sub> stress; 1 or 10 K, indicates 1 or 10 mM K<sup>+</sup>. Data are reported as means of values obtained from four independent replicates  $\pm$  SE. Statistical differences between seedlings grown under 1 mM ZnCl<sub>2</sub> supplied with 1 and 10 mM KCl (\*\*P < 0.01, independent samples *t*-test).

## Expression profiles of the KT/HAK/KUP family genes in response to Zn toxicity

Notably, Zn toxicity dramatically decreased tissue  $K^+$  accumulation, while escalated  $K^+$  supply (10K+Zn) practically restored tissue  $K^+$  concentration under Zn toxicity (Table 2). To further elucidate the molecular mechanisms underlying the relationship between  $K^+$  acquisition and the KT/HAK/KUP family  $K^+$  transporters, expression profiles of 16 putative *KUP* genes in peach were evaluated in response to different  $K^+$  supplies under Zn toxicity (for 21 days). Compared to control conditions, expression of seven, eight, and eight *KUP* genes was

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significantly induced in leaf, stem and root tissues (0.01 < P < 0.05, P < 0.01, t-test), respectively, under Zn toxicity (1K+Zn), whereas transcript levels of three genes (i.e., *KUP3*, *KUP4*, and *KUP6*) in leaf and four genes (including *KUP1*, *KUP5*, *KUP6*, and *KUP11*) in stem were reduced (0.01 < P < 0.05, P < 0.01, t-test). Expression of *KUP15* changed little, either under 1K+Zn or 10K+Zn treatment (Figure 6). Together, the differential response of these genes may intricately contribute to the reduced tissue K<sup>+</sup> concentration under Zn toxicity. Compared to control conditions, elevated K<sup>+</sup> supply (10K+Zn) significantly enhanced expression of eleven, ten, and ten *KUP* genes in leaf, stem and root tissues respectively, under Zn toxicity (Figure 6, 0.01 < P < 0.05, P < 0.01, t-test), which may partially explain the increased K<sup>+</sup> accumulation in tested organs under Zn toxicity (Table 2).



**Figure 6.** Relative expression of *PpeKUP* genes under zinc toxicity. All treatments were maintained for 21 days prior to analysis. Expression values are given as a ratio relative to the values of actin. +Zn, means 2 mM ZnCl<sub>2</sub> stress; 1 or 10 K, indicates 1 or 10 mM K<sup>+</sup>. The values are reported as means  $\pm$  SE (N = 24), calculated from 3 sets of experiments. \*Statistical differences between seedlings grown under control and ZnCl<sub>2</sub> stresses (\*\*P < 0.01, independent samples *t*-test).

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## DISCUSSION

As an essential microelement involved in numerous physiological processes, trace amounts of Zn application contribute to fruit quality in particular (Zhang et al., 2013; Saadati et al., 2013). However, excessive amounts of Zn can lead to toxicity symptoms in plants. In this present study, peach seedlings exhibited better tolerance to heavy Zn toxicity than several annual plants, including maize, rice, and *Arabidopsis*. Notably, peach as a perennial woody plant contains more lignin than these annual plants, which may partially explain the improved tolerance to heavy metals. It is also worth mentioning that the Zn<sup>2+</sup> concentration in shoots was simultaneously reinforced with increasing Zn<sup>2+</sup> supplies, implying that peach 'Yoshihime' seedlings may have an excellent capacity for Zn<sup>2+</sup> accumulation, and excess Zn<sup>2+</sup> may be further transported or deposited to the aboveground parts of the seedlings. Nonetheless, our findings imply that peach 'Yoshihime' seedlings may have higher Zn tolerance and Zn accumulation capacities than normal crops

Heavy Zn toxicity has been reported to reduce plant growth by causing internal physiological changes (Kozlov, 2004; Street et al., 2007). Indeed, Zn toxicity (2 mM) resulted in severe growth damage and affected biochemical processes, especially photosynthetic performance, in peach seedlings in the current study. Interestingly, elevated K<sup>+</sup> supply (10 mM K<sup>+</sup>) beneficially abolished the damage caused by Zn toxicity, and endowed peach seedlings with dramatically strengthened photosynthetic performance, evidenced by enhanced total chlorophyll concentration,  $g_s$  and  $P_{NP}$  which may directly explain their enhanced growth performance and increased number of green leaves.

Notably, elevated  $K^+$  supply reduced the internal  $H_2O_2$  concentration, suggesting an enhanced capacity for protection from oxidative damage by Zn toxicity. Simultaneously, significantly enhanced activities of endogenous SOD, POD, and CAT in peach leaves were observed under the 10K+Zn treatment. Correspondingly, the transcripts of relative genes involved in plant antioxidant systems were significantly increased in the leaves of peaches under test. In particular, proline is an important stress-related metabolite that contributes to osmotic adjustment and enzyme stabilization, enabling plants to better tolerate adverse abiotic stresses (Choudhary et al., 2007; Huang et al., 2013). Under Zn toxicity, elevated K<sup>+</sup> supply (10 mM K<sup>+</sup>) significantly reduced leaf proline concentrations, which may partially support the theory that proline contributes to plant performance and tolerance under Zn toxicity. We speculate that escalated K<sup>+</sup> supply favorably contributes to plant antioxidant systems by enhancing ROSscavenging capacity for oxidative damage protection, which further provides peach seedlings with better performance under Zn toxicity.

 $K^+$  beneficially contributes to plant tolerance to various abiotic stresses, such as drought, salt, and cold (Rai et al., 2008; Mian et al., 2011; Li et al., 2011; Song and Su, 2013; Song et al., 2014a, 2015). Zn toxicity dramatically reduces many metabolic processes in plants, including  $K^+$  nutrition and homeostasis. To cope with such adverse toxicity stress, plants may have to improve their internal nutritional status or strengthen metabolic systems to maintain basic growth. Notably, expression of 15 out of 16 *KUP* genes was greatly increased in individual tissues under Zn toxicity stress, directly indicating that  $K^+$  acquisition and transport systems were enhanced, which possibly mobilized the 'luxury uptake' of external  $K^+$  via  $K^+$ transporters to protect  $K^+$ -dependent cellular processes and maintain basic growth. Moreover, elevated  $K^+$  supply beneficially strengthened peach root formation under Zn stress, which possibly ensured powerful  $K^+$  uptake via plant roots when exposed to sufficient  $K^+$  supply. Si-

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multaneously, all of the *KUP* genes, except *KUP4* and *KUP15*, were significantly upregulated in individual tissues under Zn toxicity after supply of plentiful amounts of K<sup>+</sup>, indicating that enhanced expression of K<sup>+</sup> transporter genes surely guarantees K<sup>+</sup> uptake, which is then sufficiently transported or dispatched into the aboveground components, further protecting K<sup>+</sup>-dependent metabolic processes, including photosynthesis. Indeed, elevated K<sup>+</sup> supply favorably increased tissue K<sup>+</sup> accumulation under Zn toxicity. Escalated K<sup>+</sup> supply definitely favorably enhanced the expression of genes involved in plant antioxidant systems, which better explains the corresponding strengthened activity of antioxidant enzymes (Figure 4). Nonetheless, these findings directly suggest that escalated K<sup>+</sup> supply contributes to enhancing plant antioxidant systems, which may partially explain their improved performance under Zn toxicity.

Although the molecular basis of KT/HAK/KUP family transporters is largely unclear in peach, these transporters were closely involved in  $K^+$  nutrition and homeostasis, especially under Zn toxicity, further protecting plant photosynthetic performance and the antioxidative defense system. Nonetheless, our findings convincingly support the key role of  $K^+$  in improving plant tolerance to abiotic stresses.

#### **Conflicts of interest**

The authors declare no conflict of interest.

#### **ACKNOWLEDGMENTS**

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