

Pro-Oxidant Natural Products as Anticancer Agents

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Abstract: Cancer cells produce high levels of reactive oxygen species (ROS) that lead to a state of increased basal oxidative stress. Since this state of oxidative stress makes cancer cells vulnerable to agents that further augment ROS levels, the use of pro-oxidant agents is emerging as an exciting strategy to selectively target tumor cells. Natural products have provided a significant contribution to the development of several drugs currently used in cancer chemotherapy. Although many natural products are known to affect the redox state of the cell, most studies on these compounds have focused on their antioxidant activity instead of on their pro-oxidant properties. This article provides an overview of natural products with pro-oxidant and anticancer activities, with special focus on plant secondary metabolites, and discusses their possible use as cancer chemotherapeutic agents.

Keywords: Alkaloids, cancer, hydrogen peroxide, phenolic compounds, plant secondary metabolites, prooxidant, reactive oxygen species, terpenoids.

INTRODUCTION

Cancer kills over seven million people worldwide every year [1]. The mortality rate of this disease has not changed much in the past few decades even in developed countries as the United States [2]. Although cancer therapy in the form of surgery or radiotherapy is effective when the disease is early detected, many cancers are still diagnosed when cells from a primary tumor have already metastasized to other parts of the body. The main form of treatment at this point is chemotherapy, which consists of delivering drugs systemically so that they can reach and kill the tumor cells. But most of these drugs cause severe side effects in patients and, therefore, need to be used at suboptimal levels. The low efficacy of chemotherapy in patients with advanced cancers is reflected in the low 5-year survival rates observed in these patients [2]. For example, cancer statistics show that the most commonly diagnosed cancer in the world is lung cancer [1], that approximately 50% of patients diagnosed with this type of cancer have distant metastasis [2] and that only 3% of these patients manage to survive more than 5 years [2]. The low efficacy of cancer therapy for the treatment of patients with metastasis makes the development of novel chemotherapeutic agents necessary.

Despite the recent interest by pharmaceutical companies in molecular modeling, combinatorial chemistry and other synthetic chemistry techniques, natural products and medicinal plants continue to be an important source of new drugs. Natural products are not only used as therapeutic agents, but are also a source of lead compounds that have provided the basis for the semisynthesis or total synthesis of new drugs. An analysis of the sources of drugs approved from January

1981 to the middle of October 2008 revealed that 6% of the 1024 new chemical entities were unmodified natural products and that only 37% were drugs not related to natural products [3, 4]. The role these latter in drug discovery is particularly relevant in oncology. It is estimated that over the time frame from around the 1940s to 2006, of the 155 small molecules approved for cancer therapy, only 27% were not related to natural products [3]. The first plant-derived anticancer agents to advance into clinical use were the *Vinca* alkaloids vinblastine and vincristine. Other important plant-derived anticancer compounds include paclitaxel (taxol), the epipodophyllotoxin derivative etoposide, and the camptothecin derivatives topotecan and irinotecan [5]. The mechanism of action of these drugs is considered to consist in the inhibition of microtubule assembly (*Vinca* alkaloids and paclitaxel), inhibition of DNA topoisomerase II (etoposide) and inhibition of DNA topoisomerase I (camptothecin derivatives). Recent evidence suggests that the formation of reactive oxygen species (ROS) may also contribute to the anticancer effects of these drugs [6-8]. The induction of oxidative stress by pro-oxidant agents is indeed emerging as an attractive anticancer strategy that may be used to target cancer cells selectively [9-13]. After discussing the role of oxidative stress in cancer and the possible use of pro-oxidant agents in cancer therapy, this article provides an overview of pro-oxidant natural products with anticancer activity and examines their potential as cancer chemotherapeutic agents.

OXIDATIVE STRESS IN CANCER CELLS

Oxidative stress is an imbalance between the generation and elimination of reactive oxygen species in favor of the former, causing excessive oxidative damage to macromolecules, cells and tissues. Reactive oxygen species (ROS) is the collective term used to name oxygen radicals (including hydroxyl radical and superoxide radical) and some other non-radical derivatives of oxygen, such as hydrogen peroxide

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(H₂O₂). ROS can easily generate free radicals (any species containing one or more unpaired electrons) and/or cause oxidative damage [14]. ROS are generated by all aerobic organisms and their production seems to be needed for signal-transduction pathways that regulate several different physiological processes. Excessive amounts of ROS, however, can start toxic and lethal chain reactions, which oxidize and disable structures that are required for cellular integrity and survival. ROS are generated in multiple compartments and by multiple enzymes within the cell. Important contributions include proteins within the plasma membrane, such as the growing family of NADPH oxidases; lipid metabolism within the peroxisomes; as well as the activity of various cytosolic enzymes such as cyclooxygenases. Although all these sources contribute to the overall ROS production, the vast majority of cellular ROS can be traced back to the mitochondria [15, 16].

Accumulating evidence indicates that cancer cells generate excessive levels of ROS and have a state of oxidative stress. Many malignant cells produce high levels of ROS in culture. For instance, Szatrowski and Nathan reported that several tumor cell lines, representing a variety of tissue types, constitutively produced large amounts of H₂O₂. They observed that the cumulative amount of H₂O₂ produced after 4 h by these tumor cells was comparable to the amount of H₂O₂ produced by similar numbers of phorbol ester-triggered neutrophils [17]. The increased production of ROS by cancer cells observed *in vitro* has also been found *in vivo*. For example, chronic lymphocytic leukemia cells freshly taken from patients showed increased ROS production compared with normal lymphocytes. This was also observed with B-cell lines from patients with Burkitt's lymphoma associated with Epstein-Barr virus infection and malignant B-cells from patients with hairy cell leukemia (see [18] and references therein). For solid tumors, however, demonstrating increased ROS production *in vivo* is difficult to achieve owing to methodological inadequacies, so most researchers have studied oxidative damage levels rather than ROS production. Such studies have shown increased levels of oxidative damage (e.g. 8OHdG) in human cancers and in animal cancers induced by a wide range of carcinogens (reviewed in [18]). Interestingly, the most important carcinogenic agents and behaviors induce oxidative stress, including most chemical carcinogens (e.g. *N*-nitrosamines, asbestos, arsenic), ultraviolet radiation, cancer-associated viruses or bacteria, inflammation, alcohol, tobacco smoke and obesity. It is also recognized that age is the principal risk factor for most cancers and that oxidative stress may be the most important causal factor in aging (see [19, 20] and references therein).

The increased levels of ROS of cancer cells seem to play a key role in cancer development [12, 18, 21]. ROS such as H₂O₂ can induce cell malignant transformation, and the malignant phenotype of tumor cells can be reversed by decreasing the levels of ROS [12, 22-24]. For instance, expression of the ROS generation system Nox1 in normal NIH3T3 fibroblasts resulted in cells with malignant characteristics that produced tumors in athymic mice. These transformed cells showed a 10-fold increase in H₂O₂ levels. When human catalase was expressed in these transformed cells, H₂O₂ concentration decreased, and the cells reverted to a normal appearance, the growth rate normalized, and cells no longer pro-

duced tumors in athymic mice [24]. In addition, ROS have been shown to participate in the most relevant aspects of carcinogenesis. Most researchers consider that cancer is a genetic disease caused by the acquisition of multiple mutations in genes that control cell proliferation, cell death and genomic instability [25]. It is also accepted that cells must develop several acquired capabilities in order to become a malignant cancer: increased cell proliferation (caused, in part, by resistance to growth inhibition and independence from mitogenic stimulation), apoptosis resistance, cellular immortalization, increased angiogenesis, invasion and metastasis. In addition, genetic instability is considered to be a key event that enables the acquisition of these capabilities [26, 27]. Accumulating experimental data indicate that an increase in the cellular concentrations of ROS such as H₂O₂ can explain all these hallmarks of cancer. It is known that an increase in the levels of H₂O₂ can lead to DNA damage, mutations, and genetic instability [26-31]; H₂O₂-induced DNA damage seems to be mediated by hydroxyl radical generated from H₂O₂ by the Fenton reaction [28-30]. Several studies have also demonstrated that ROS can induce cell proliferation [31], apoptosis resistance [32, 33], increased angiogenesis [34, 35], and invasion and metastasis [36-38]. Indeed, these studies showed that an increase in the levels of H₂O₂-detoxifying enzymes could reduce cell proliferation, promote apoptosis, and inhibit invasion, metastasis and angiogenesis. In short, cancer cells produce high levels of ROS that lead to a state of increased basal oxidative stress. Such state of oxidative stress is induced by the most important human carcinogens and plays an important role in cancer development.

SELECTIVE ANTICANCER ACTIVITY OF PRO-OXIDANT AGENTS

Since cancer cells have increased levels of ROS that play an important role in carcinogenesis, agents with antioxidant activity may induce cancer preventive effects by reducing and/or preventing such increase in the cellular levels of ROS. Because pro-oxidant agents increase the cellular levels of ROS, it is recognized that these agents can induce carcinogenic effects. But when pro-oxidant agents increase the cellular levels of ROS to cytotoxic levels, these agents may induce selective killing of cancer cells and be therapeutically useful. It is important to mention that all these effects can be achieved by agents with both antioxidant and pro-oxidant properties (e.g. curcumin), which can act as cancer chemopreventive, carcinogenic, and chemotherapeutic agents mainly depending on the concentration by which they are used [12, 39].

The role of ROS in cancer therapy is increasingly being acknowledged and the induction of oxidative stress by pro-oxidant agents is emerging as an attractive anticancer strategy [9-12, 40-43]. Recent data suggests that ROS participate in the anticancer activity of many chemotherapeutic agents commonly used in the clinic, including paclitaxel, docetaxel, cisplatin, doxorubicin, arsenic trioxide, bortezomib, procarbazine and etoposide [6-8, 10, 40, 44-54]. For instance, although it has been known for many years that the microtubule protein tubulin is the therapeutic target for paclitaxel (taxol), recent experiments have shown that the accumulation of H₂O₂ is a crucial step for paclitaxel-induced cancer cell death both *in vitro* and *in vivo* [6, 8]. H₂O₂ seems to be a

key player in oxidative stress-induced cancer cell death. Many anticancer agents increase the levels of H_2O_2 [6, 45, 47], and H_2O_2 is known to be an efficient inducer of cell death in cancer cells [12, 48, 55]. Interestingly, cancer cells are more susceptible to H_2O_2 -induced cell death than non-malignant cells [56-58]. Using several cancer and normal cell lines, Chen *et al.* [56] observed that high concentrations of ascorbic acid selectively killed a variety of cancer cells and that this effect was mediated by H_2O_2 . They showed, for instance, that a concentration of 50 μM H_2O_2 induced more percentage of cell death in Burkitt's lymphoma cells than 250 μM in normal lymphocytes and normal monocytes [56]. This selective effect of H_2O_2 has also been observed in cells derived from solid tumors. Using lung cancer cells and non-malignant lung fibroblasts under the same experimental conditions, we recently found that specific concentrations of H_2O_2 and of the H_2O_2 -generating agent pyrogallol induced selective killing of the cancer cells [59].

It is not clear why specific concentrations of H_2O_2 (and of pro-oxidant agents) can kill cancer cells selectively. *In vitro* and *in vivo* data indicate that tumor cells produce higher concentrations of H_2O_2 than their normal counterparts [17, 18, 31, 60-62]. This, and the fact that there is a threshold of H_2O_2 above which cells cannot survive, may explain why specific concentrations of H_2O_2 induce selective killing of cancer cells [12]. Excessive cellular accumulation of H_2O_2 may cause cell death through the induction of DNA damage, which seems to be mediated by hydroxyl radical generated from H_2O_2 in the presence of iron or copper (Fenton reaction) [28-30]. Unlike non-malignant cells, cancer cells have mutations in DNA repair genes and cannot properly repair specific types of DNA damage [25, 63]. It is possible that some cancers may have a reduced capacity to repair ROS-induced DNA damage and be more vulnerable than normal cells to the cytotoxic activity of ROS. It has also been proposed that the increased levels of copper found in various malignancies may explain why some pro-oxidant agents (e.g. plant polyphenols) can induce selective killing of cancer cells [64]. The increased levels of copper of cancer cells would favor the formation of higher levels of hydroxyl radical through the Fenton reaction.

PRO-OXIDANT NATURAL PRODUCTS WITH ANTICANCER ACTIVITY

An overview of natural products with both pro-oxidant and anticancer activities is presented in Table 1. The name of the natural product, the type of compound, the natural source (representative species) and the references are provided. The first part of the table comprises plant compounds of primary metabolism and their derivatives. Then, plant secondary metabolites, including phenolic compounds, terpenoids and alkaloids, are compiled. The last section includes other natural products from different natural sources (compounds of animal, microorganism, or marine origin, vitamins, etc). The mechanism involved in the generation of ROS is not available for most compounds and is not included. The general mechanisms involved in ROS generation by a variety of pro-oxidant agents (from natural and synthetic origin) have been discussed extensively elsewhere [42].

Some natural products reported in Table 1 are drugs currently used in cancer chemotherapy (e.g. paclitaxel, vincristine, vinblastine, bleomycin, mitomycin, doxorubicin, idarubicin, aclarubicin and actinomycin D). Others have entered clinical trials for the treatment of specific types of cancer (e.g. curcumin, epigallocatechin-3-gallate, genistein, resveratrol, camptothecin, perillal alcohol, lycopene, phenylethyl isothiocyanate, sulforaphane, aplidin, eicosapentaenoic acid, linoleic acid, ursodeoxycholic acid, vitamin C, vitamin D₂ and vitamin D₃; see <http://clinicaltrials.gov/>). The chemical structures of these compounds are represented in Fig. (1) and Fig. (2). The anticancer activity of most compounds compiled in Table 1 has only been evaluated in pre-clinical models.

It is important to mention that, although the pro-oxidant effect of a specific natural product may not be the most important cytotoxic mechanism of action, this pro-oxidant effect may be responsible for the selective anticancer activity of the compound. For instance, it is known that the main mechanism of action of paclitaxel consists in the inhibition of microtubule assembly. A drug that only inhibits microtubule assembly would be equally cytotoxic in cells with the same proliferating rate, as microtubules are necessary for cell proliferation. Because it is known that cancer cells are more vulnerable to paclitaxel than highly proliferating non-malignant cells, it has been enigmatic for many years why this drug has certain selectivity for cancer cells [25]. Recently, it was observed that the accumulation of H_2O_2 is crucial for paclitaxel-induced cancer cell death both *in vitro* and *in vivo* [6, 8]. Being well known that H_2O_2 can induce selective killing of cancer cells, it seems possible that paclitaxel-induced H_2O_2 production plays a role in the selective anticancer effects of this natural product.

Since the redox state of the cell is important for many cellular processes, it has been discussed that pro-oxidant agents may act as "dirty" drugs (agents that modulate multiple molecular targets through pleiotropic interactions). However, recent research suggests that this pleiotropic mode of action may be an advantage to overcome cancer cell drug resistance typical of drugs acting on just one target [42]. Although pro-oxidant agents could be used as stand-alone drugs, evidence suggests that they could also be used in combination [42]. Indeed, although ROS induce cancer cell death, tumor cells are known to develop mechanisms that prevent ROS from reaching cytotoxic levels. The glutathione and thioredoxin antioxidant systems are crucial for detoxifying ROS. These antioxidant systems are activated in cancer cells and play an important role in the development of resistance to many anticancer agents [65-71]. The possible drug resistance induced by pro-oxidant agents could be reduced with glycolysis inhibitors [13]. Evidence indicates that pro-oxidant agents can increase the cellular levels of H_2O_2 and that glycolysis inhibitors can reduce the capacity of cells to detoxify H_2O_2 . Experimental data have shown that malignant cells are more susceptible to glucose deprivation than non-transformed cells, and that an increase in the levels of H_2O_2 may mediate the cytotoxic effect induced by glucose deprivation [62, 72-74]. Two possible mechanisms may explain why the activation of glycolysis plays an important function in protecting tumor cells from H_2O_2 -induced cell death. First, the activation of glycolysis increases the formation of pyru-

Table 1. Natural Products with Pro-Oxidant and Anticancer Activities

Compound	Type of Compound	Natural Source	References
Compounds of Primary Metabolism and derivatives			
Abrin	Aminoacid	<i>Abrus precatorius</i> (Fabaceae)	[81, 82]
Ajoene	Organosulfur	<i>Allium sativum</i> (Alliaceae)	[83, 84]
Allicin	Organosulfur	<i>Allium sativum</i> (Alliaceae)	[85-87]
Benzyl isothiocyanate	Organosulfur	<i>Brassica</i> spp. (Brassicaceae)	[88, 89]
Diallyl disulfide	Organosulfur	<i>Allium</i> spp. (Alliaceae)	[90, 91]
Dimethyl disulfide	Organosulfur	<i>Allium</i> spp. (Alliaceae)	[92]
Jasmonic acid	Fatty acid	<i>Jasminum</i> spp. (Oleaceae), widespread	[93]
Linoleic acid	Fatty acid	<i>Carthamus tinctorius</i> (Asteraceae)	[94, 95]
Linolenic acid	Fatty acid	<i>Perilla frutescens</i> (Lamiaceae)	[95-97]
L-Mimosine	Aminoacid	<i>Mimosa</i> spp., <i>Leucaena</i> spp. (Fabaceae)	[98, 99]
Melatonin	Aminoacid	<i>Prunus cerasus</i> (Rosaceae)	[100, 101]
Methyl jasmonate	Fatty acid	<i>Jasminum</i> spp. (Oleaceae), widespread	[93, 102]
Phenylethylisothiocyanate	Organosulfur	<i>Brassica</i> spp. (Brassicaceae)	[89, 103, 104]
Sorbitol	Sugar alcohol	<i>Malus</i> spp. (Rosaceae)	[105]
Sulforaphane	Organosulfur	<i>Brassica</i> spp. (Brassicaceae)	[106, 107]
Phenolic Compounds			
2'-Hydroxycinnamaldehyde	Phenolic acid	<i>Cinnamomum</i> spp. (Lauraceae)	[108, 109]
3,7,4'-trihydroxyflavone	Flavone	<i>Rhus chinensis</i> (Anacardiaceae)	[110, 111]
4'-Hydroxycinnamaldehyde	4'-Hydroxycinnamaldehyde	<i>Alpinia galanga</i> (Zingiberaceae)	[112]
4-Hydroxycinnamic acid	Hydroxycinnamic acid	<i>Erythrina fusca</i> (Fabaceae), widespread	[113, 114]
6- Dehydrogingerdione	Aryl alkanones	<i>Zingiber officinale</i> (Zingiberaceae)	[115, 116]
6-Gingerol	Aryl alkanones	<i>Zingiber officinale</i> (Zingiberaceae)	[117, 118]
6-Shogaol	Aryl alkanones	<i>Zingiber officinale</i> (Zingiberaceae)	[119]
8-Shogaol	Aryl alkanones	<i>Zingiber officinale</i> (Zingiberaceae)	[120]
Acacetin	Flavone	<i>Robinia pseudoacacia</i> (Fabaceae)	[121, 122]
Aesculetin	Coumarin	<i>Aesculus hippocastanum</i> (Hippocastanaceae)	[123]
Aloe-emodin	Anthraquinone	<i>Rheum</i> spp. (Polygonaceae), <i>Cassia</i> spp. (Fabaceae)	[124, 125]
Apigenin	Flavone	<i>Petroselinum crispum</i> (Apiaceae), widespread	[126, 127]
Baicalein	Flavone	<i>Scutellaria baicalensis</i> (Lamiaceae), <i>Oroxylum indicum</i> (Bignoniaceae)	[128, 129]
Baicalin	Flavone	<i>Scutellaria</i> spp. (Lamiaceae)	[130, 131]
Benzaldehyde	Aromatic aldehyde	<i>Prunus</i> spp. (Rosaceae), widespread	[132, 133]
Betuletol 3-methyl ether	Flavonol	<i>Allagopappus viscosissimus</i> (Asteraceae)	[134]
Butein	Chalcone	<i>Rhus verniciflua</i> (Anacardiaceae)	[135]
Caffeic acid	Phenolic acid	<i>Coffea</i> spp. (Rubiaceae), widespread	[136, 137]

(Table 1) contd....

Compound	Type of Compound	Natural Source	References
Cajanol	Isoflavanone	<i>Cajanus cajan</i> (Fabaceae)	[138]
Catechin	Flavan-3-ol	<i>Acacia catechu</i> (Fabaceae), widespread	[139, 140]
Catechol	Simple phenol	<i>Cola</i> spp. (Malvaceae), <i>Gaultheria</i> spp. (Ericaceae)	[141-143]
Chebulinic acid	Hydrolyzable tannin	<i>Terminalia chebula</i> (Combretaceae)	[144, 145]
Chlorogenic acid	Phenolic acid	<i>Coffea</i> spp. (Rubiaceae), widespread	[146-148]
Chrysin	Flavone	<i>Prunus</i> spp. (Rosaceae),	[149-151]
Chrysoeriol	Flavone	<i>Medicago sativa</i> (Fabaceae), widespread	[152]
Chrysophanol	Anthraquinone	<i>Rhamnus</i> spp. (Rhamnaceae), <i>Rheum</i> spp. (Polygonaceae)	[153]
Curcumin	Diarylheptanoid	<i>Curcuma longa</i> (Zingiberaceae)	[39, 154, 155]
Cyanidin	Anthocyanidin	<i>Vaccinium</i> spp. (Ericaceae), <i>Prunus</i> spp. (Rosaceae)	[156, 157]
Cyanidin 3-glucoside	Anthocyanin	<i>Vaccinium</i> spp. (Ericaceae), <i>Prunus</i> spp. (Rosaceae)	[157, 158]
Cyanidin-3-rutinoside	Anthocyanin	<i>Vaccinium</i> spp. (Ericaceae), <i>Prunus</i> spp. (Rosaceae)	[157, 159]
Daidzein	Isoflavone	<i>Glycine max</i> (Fabaceae), widespread	[160]
Dantron	Anthraquinone	<i>Xyris semifuscata</i> (Xyridaceae)	[161, 162]
Daphnetin	Coumarin	<i>Daphne</i> spp. (Thymelaeaceae)	[163]
Delphinidin	Anthocyanidin	<i>Delphinium</i> spp. (Ranunculaceae)	[157, 164]
Delphinidin 3-sambubioside	Anthocyanin	<i>Hibiscus</i> spp. (Malvaceae)	[165]
Diospyrin	Naphthoquinone	<i>Diospyros montana</i> (Ebenaceae)	[166]
Ellagic acid	Phenolic acid	<i>Vaccinium</i> spp. (Ericaceae), widespread	[167]
Emodin	Anthraquinone	<i>Rheum</i> spp. (Polygonaceae)	[168]
Epicatechin	Flavan-3-ol	<i>Acacia catechu</i> (Fabaceae), widespread	[139, 169]
Epicatechin-gallate	Flavan-3-ol	<i>Camellia sinensis</i> (Theaceae)	[170]
Epigallocatechin	Flavan-3-ol	<i>Camellia sinensis</i> (Theaceae)	[170]
Epigallocatechin-3-gallate	Flavan-3-ol	<i>Camellia sinensis</i> (Theaceae)	[170, 171]
Eriodictyol	Flavanone	<i>Eriodictyon californicum</i> (Boraginaceae)	[172]
Esculetin (Aesculetin)	Coumarin	<i>Aesculus hippocastanum</i> (Hippocastanaceae)	[163, 173]
Eugenol	Phenylpropanoid	<i>Eugenia caryophyllata</i> (Myrtaceae)	[174]
Eupafolin	Flavone	<i>Eupatorium perfoliatum</i> (Asteraceae), <i>Artemisia princeps</i> (Asteraceae)	[175-177]
Ferulic acid	Phenolic acid	<i>Ferula communis</i> (Apiaceae), widespread	[136]
Fisetin	Flavonol	<i>Fragaria</i> spp. (Rosaceae), widespread	[178, 179]
Flavokawain B	Chalcone	<i>Piper methysticum</i> (Piperaceae)	[180]
Fraxetin	Coumarin	<i>Fraxinus</i> spp. (Oleaceae)	[173, 181]
Gallic acid	Phenolic acid	<i>Kalanchoe</i> spp. (Crassulaceae), widespread	[182, 183]
Gambogic acid	Xanthone	<i>Garcinia hanburyi</i> (Clusiaceae)	[184, 185]

(Table 1) contd....

Compound	Type of Compound	Natural Source	References
Genistein	Isoflavone	<i>Genista</i> spp. (Fabaceae), widespread	[186]
Gentianaclausin	Xanthone	<i>Gentiana kochiana</i> (Gentianaceae)	[187]
Gentianaclausin	Xanthone	<i>Gentiana kochiana</i> (Gentianaceae)	[187]
Guttiferone-A	Benzophenone	<i>Garcinia livingstonei</i> (Clusiaceae)	[188]
Hesperetin	Flavanone	<i>Citrus</i> spp. (Rutaceae)	[189, 190]
Hydroxytyrosol	Simple Phenol	<i>Olea europaea</i> (Oleaceae)	[191]
Icariin	Flavonol glycoside	<i>Epimedium</i> spp. (Berberidaceae)	[192]
Isoeugenol	Phenylpropanoid	<i>Eugenia caryophyllata</i> (Myrtaceae)	[193, 194]
Isoliquiritigenin	Chalcone	<i>Glycyrrhiza glabra</i> (Fabaceae)	[178, 195]
Juglone	Naphthoquinone	<i>Juglans regia</i> (Juglandaceae)	[196, 197]
Kaempferol	Flavonol	<i>Kaempferia galanga</i> (Zingiberaceae), widespread	[198, 199]
Liquiritigenin	Flavanone	<i>Glycyrrhiza glabra</i> (Fabaceae)	[200]
Luteolin	Flavone	<i>Reseda luteola</i> (Resedaceae), widespread	[201, 202]
Malvidin	Anthocyanidin	<i>Althaea rosea</i> (Malvaceae)	[157, 203]
Malvidin 3-glucoside	Anthocyanin	<i>Vitis</i> spp. (Vitaceae)	[157, 204]
Methyl gallate	Phenolic acid	<i>Camellia sinensis</i> (Theaceae)	[205]
Morin	Flavone	<i>Maclura</i> spp. (Moraceae)	[206]
Myricetin	Flavonol	<i>Myrica rubra</i> (Myricaceae), widespread	[59, 207]
Naringenin	Flavanone	<i>Citrus</i> spp. (Rutaceae), widespread	[208, 209]
Nordihydroguaiaretic acid	Phenolic acid	<i>Larrea</i> spp. (Zygophyllaceae)	[210-212]
Norwogonin	Flavone	<i>Scutellaria</i> spp. (Lamiaceae)	[213]
Pelargonidin	Anthocyanidin	<i>Pelargonium</i> spp. (Geraniaceae)	[157, 203]
Pelargonidin 3-glucoside	Anthocyanin	<i>Vaccinium</i> spp. (Ericaceae)	[157, 214]
Pentagalloyl glucose	Hydrolyzable tannin	<i>Quercus infectoria</i> (Fagaceae)	[215, 216]
Peonidin	Anthocyanidin	<i>Paeonia</i> spp. (Ranunculaceae)	[157, 203]
Peonidin 3-glucoside	Anthocyanin	<i>Vaccinium</i> spp. (Ericaceae)	[157, 217]
Phloretin	Chalcone	<i>Malus</i> spp. (Rosaceae)	[178, 218]
Plumbagin	Naphthoquinone	<i>Drosera</i> spp. (Droseraceae)	[219, 220]
Procyanidin B2	Proanthocyanidin	<i>Cinnamomum cassia</i> (Lauraceae), <i>Vaccinium</i> spp. (Ericaceae)	[221, 222]
Protoapigenone	Flavone	<i>Thelypteris torresiana</i> (Thelypteridaceae)	[223]
Psoralen	Furanocoumarin	<i>Psoralea corylifolia</i> (Fabaceae)	[224, 225]
Pterostilbene	Stilbenoid	<i>Vitis</i> spp. (Vitaceae)	[226]
Quercetin	Flavonol	<i>Citrus</i> spp. (Rutaceae), widespread	[207]
Resveratrol	Stilbenoid	<i>Vitis</i> spp. (Vitaceae)	[227, 228]
Rhein	Anthraquinone	<i>Rheum</i> spp. (Polygonaceae)	[229, 230]
Rosmarinic acid	Hydroxycinnamic acid	<i>Rosmarinus officinalis</i> (Lamiaceae), widespread	[231, 232]

(Table 1) contd....

Compound	Type of Compound	Natural Source	References
Rottlerin	Phloroglucinol	<i>Mallotus philippinensis</i> (Euphorbiaceae)	[233, 234]
Rutin	Flavonol	<i>Ruta</i> spp. (Rutaceae), widespread	[235-238]
Salicylic acid	Phenolic acid	<i>Salix</i> spp. (Salicaceae)	[239-241]
Shikonin	Naphthoquinone	<i>Lithospermum erythrorhizon</i> (Boraginaceae)	[242, 243]
Sinapic acid	Phenolic acid	<i>Brassica</i> spp. (Brassicaceae), widespread	[113, 114]
Sophoranone	Flavanone	<i>Sophora subprostrata</i> (Fabaceae)	[244]
Tannic acid	Tannin	<i>Quercus</i> spp. (Fagaceae), widespread	[245, 246]
Taxifolin	Flavanonol	<i>Silybum marianum</i> (Asteraceae)	[111, 149, 247]
Tricetin	Flavone	<i>Oryza sativa</i> (Poaceae)	[248]
Usnic acid	Dibenzofuran	<i>Usnea</i> spp. (Parmeliaceae)	[249, 250]
Vanillin	Phenolic acid	<i>Vanilla</i> spp. (Orchidaceae)	[251-253]
Wogonin	Flavone	<i>Scutellaria baicalensis</i> (Lamiaceae)	[254, 255]
Xanthohumol	Chalcone	<i>Humulus lupulus</i> (Cannabaceae)	[256, 257]
Xanthotoxin	Furanocoumarin	<i>Ammi majus</i> (Apiaceae)	[258, 259]
Terpenoids			
18 β -Glycyrrhetic acid	Triterpenoid	<i>Glycyrrhiza glabra</i> (Fabaceae)	[260, 261]
Andrographolide	Diterpenoid	<i>Andrographis paniculata</i> (Acanthaceae)	[262]
Artemisinin	Lactone sesquiterpenoid	<i>Artemisia annua</i> (Asteraceae)	[263-265]
Asiatic acid	Triterpenoid	<i>Centella asiatica</i> (Mackinlayaceae)	[266]
Astilbotriterpenic acid	Triterpenoid	<i>Astilbe chinensis</i> (Saxifragaceae)	[267]
Betulinic acid	Triterpenoid	<i>Betula</i> spp. (Betulaceae)	[268]
Bixin	Apocarotenoid	<i>Bixa orellana</i> (Bixaceae)	[269]
Bufalin	Cardiac glycoside	<i>Bufo bufo</i> (Bufonidae)	[270]
Cannabidiol	Cannabinoid	<i>Cannabis sativa</i> (Cannabaceae)	[271-273]
Costunolide	Sesquiterpenoid	<i>Laurus nobilis</i> (Lauraceae)	[274, 275]
Cucurbitacin B	Triterpenoid	<i>Iberis amara</i> (Brassicaceae)	[276]
Dioscin	Steroidal saponin	<i>Dioscorea</i> spp. (Dioscoreaceae)	[277]
Diosgenin	Steroidal sapogenin	<i>Dioscorea</i> spp. (Dioscoreaceae)	[278]
Erythrodil	Triterpenoid	<i>Olea europaea</i> (Oleaceae)	[279, 280]
Farnesol	Sesquiterpenoid	<i>Vachellia farnesiana</i> (Fabaceae), widespread	[281-283]
Ginkgolide B	Diterpenoid	<i>Ginkgo biloba</i> (Ginkgoaceae)	[284, 285]
Ginsenoside RH-2	Triterpenoid saponin	<i>Ginkgo biloba</i> (Ginkgoaceae)	[286, 287]
Glaucocalyxin A	Diterpenoid	<i>Rabdosia japonica</i> var. <i>glaucocalyx</i> (Lamiaceae)	[288]
Guggulsterone	Triterpenoid	<i>Commiphora mukul</i> (Burseraceae)	[289, 290]
Gypenosides	Triterpenoid	<i>Gynostemma pentaphyllum</i> (Cucurbitaceae)	[291, 292]
Helenalin	Sesquiterpenoid	<i>Arnica</i> spp. (Asteraceae)	[293]
Linalool	Monoterpenoid	<i>Coriandrum sativum</i> (Apiaceae), widespread	[294]

(Table 1) contd....

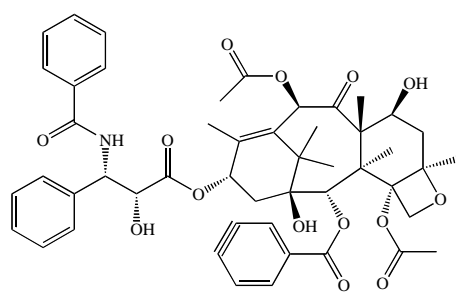
Compound	Type of Compound	Natural Source	References
Lupeol	Triterpenoid	<i>Mangifera</i> spp. (Anacardiaceae), widespread	[295]
Lycopene	Carotenoid	<i>Solanum lycopersicum</i> (Solanaceae)	[296-298]
Oleandrin	Cardiac glycoside	<i>Nerium oleander</i> (Apocynaceae)	[299]
Oleanolic acid	Triterpenoid	<i>Olea europaea</i> (Oleaceae), widespread	[300, 301]
Oleuropein	Iridoid	<i>Olea europaea</i> (Oleaceae)	[302-304]
Oridonin	Diterpenoid	<i>Rabdosia rubescens</i> (Lamiaceae)	[305]
Ouabain	Cardiac glycoside	<i>Strophanthus gratus</i> , <i>S. kombe</i> (Apocynaceae)	[306]
Ovatodiolide	Diterpenoid	<i>Anisomeles indica</i> (Lamiaceae)	[307]
Taxol	Diterpenoid	<i>Taxus brevifolia</i> (Taxaceae)	[6, 308, 309]
Parthenolide	Sesquiterpenoid	<i>Chrysanthemum parthenium</i> (Asteraceae)	[310, 311]
Perillyl alcohol	Monoterpenoid	<i>Perilla frutescens</i> (Lamiaceae)	[312, 313]
Polygodial	Sesquiterpenoid	<i>Tasmannia</i> spp. (Winteraceae)	[314, 315]
Pristimerin	Triterpenoid	<i>Maytenus heterophylla</i> (Celastraceae)	[316, 317]
Protopanaxadiol	Triterpenoid saponin	<i>Panax ginseng</i> (Araliaceae)	[318, 319]
Sarsasapogenin	Steroidal sapogenin	<i>Smilax</i> spp. (Smilacaceae)	[320]
Tetrahydrocannabinol	Cannabinoid	<i>Cannabis sativa</i> (Cannabaceae)	[321]
Thymol	Monoterpenoid	<i>Thymus</i> spp. (Lamiaceae), widespread	[322]
Triptolide	Diterpenoid	<i>Tripterygium wilfordii</i> (Celastraceae)	[323, 324]
Ursolic acid	Triterpenoid	<i>Arctostaphylos uva-ursi</i> (Ericaceae), widespread	[325]
Uvaol	Triterpenoid	<i>Olea europaea</i> (Oleaceae)	[280]
Withaferin	Withasteroid	<i>Withania somnifera</i> (Solanaceae)	[326, 327]
α -Hederin	Triterpenoid saponin	<i>Hedera helix</i> (Araliaceae)	[328]
α -humulene	Sesquiterpenoid	<i>Humulus lupulus</i> (Cannabaceae)	[329]
β -Amyrin	Triterpenoid	<i>Medicago sativa</i> (Fabaceae), widespread	[330]
β -carotene	Carotenoid	<i>Daucus carota</i> (Apiaceae), widespread	[331-333]
β -Escin (aescin)	Triterpenoid saponin	<i>Aesculus hippocastanum</i> (Hippocastanaceae)	[334]
Atractylolide	Diterpenoid	<i>Atractylis</i> spp. (Asteraceae)	[335, 336]
β -Sitosterol	Phytosterol	<i>Serenoa repens</i> (Arecaceae), widespread	[337]
Vernolepin	Lactone sesquiterpenoid	<i>Vernonia hymenolepis</i> (Compositae)	[338]
Alkaloids			
6-Methoxydihydrosanguinarine	Benzophenanthridine	<i>Hylomecon hylomeconoides</i> (Papaveraceae)	[339, 340]
Berberine	Isoquinoline	<i>Berberis</i> spp. (Berberidaceae)	[341, 342]
Boldine	Aporphine	<i>Peumus boldus</i> (Monimiaceae)	[343, 344]
Caffeine	Xanthine	<i>Coffea</i> spp. (Rubiaceae), <i>Camellia sinensis</i> (Theaceae)	[345-347]
Camptothecin	Quinoline	<i>Camptotheca acuminata</i> (Nyssaceae)	[7, 348]
Cepharanthine	Isoquinoline	<i>Stephania cepharantha</i> (Menispermaceae)	[349, 350]

(Table 1) contd....

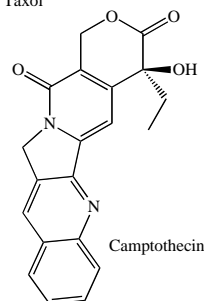
Compound	Type of Compound	Natural Source	References
Chelerythrine	Phenanthridine	<i>Chelidonium majus</i> (Papaveraceae)	[351-353]
Ellipticine	Pyridocarbazole	<i>Ochrosia elliptica</i> (Apocynaceae)	[354-356]
Homoharringtonine	Cephalotaxine	<i>Cephalotaxus harringtonia</i> (Cephalotaxaceae)	[357]
Indole acetic acid	Indole	<i>Arabidopsis thaliana</i> (Brassicaceae)	[358, 359]
Indole-3-carbinol	Indole	<i>Brassica</i> spp. (Brassicaceae)	[360]
Lycopodine	Quinolizidine	<i>Lycopodium clavatum</i> (Lycopodiaceae)	[361]
Morphine	Phenanthrene	<i>Papaver somniferum</i> (Papaveraceae)	[362, 363]
Oxymatrine	Quinolizidine	<i>Sophora flavescens</i> (Fabaceae)	[364]
Pancratistatin	Phenanthridine	<i>Hymenocallis</i> spp. (Amaryllidaceae)	[365, 366]
Piperine	Piperidine	<i>Piper</i> spp. (Piperaceae)	[367, 368]
Sampangine	Aporphine	<i>Cananga odorata</i> (Annonaceae)	[369]
Sanguinarine	Benzylisoquinoline	<i>Sanguinaria canadensis</i> (Papaveraceae)	[353, 370]
Tetrandrine	Bis-benzylisoquinoline	<i>Stephania tetrandra</i> (Menispermaceae)	[371, 372]
Tomatine	Steroidal	<i>Solanum lycopersicum</i> (Solanaceae)	[373, 374]
Vinblastine	Bis-indole	<i>Catharanthus roseus</i> (Apocynaceae)	[375, 376]
Vincristine	Bis-indole	<i>Catharanthus roseus</i> (Apocynaceae)	[377, 378]
Other Natural Products			
4-Acetyl-12,13-epoxyl-9-trichothecene-3,15-diol	Macrocyclic Trichocenes	<i>Isaria japonica</i> (Onygenaceae)	[379]
Aclarubicin	anthracycline	<i>Streptomyces galilaeus</i> (Streptomycetaceae)	[380]
Actinomycin-D	Polypeptide	<i>Streptomyces</i> spp. (Streptomycetaceae)	[381, 382]
Aplidin	Depsipeptide	<i>Aplidium albicans</i> (Clavelinidae)	[383-385]
Arachidonic acid	Fatty acid	Widespread in vertebrates	[96, 386, 387]
Ascididemin	Pyridoacridine	<i>Cystodytes dellechiaiei</i> (Polycitoridae)	[388, 389]
Bleomycin	Glucopeptide	<i>Streptomyces verticillus</i> (Streptomycetaceae)	[390, 391]
Boningmycin	Glucopeptide	<i>Streptomyces verticillus</i> var. <i>pingyangensis</i> (Streptomycetaceae)	[392]
Butenolide	Lactone	<i>Angelica</i> spp. (Apiaceae)	[393]
Capsaicin	Capsaicinoid	<i>Capsicum</i> spp. (Solanaceae)	[394, 395]
Chenodeoxycholic acid	Bile acid	Liver of animals	[396-398]
Cholic acid	Bile acid	Liver of animals	[399, 400]
C-phycocyanin	Phycobiliprotein	<i>Aphanizomenon flos-aquae</i> (Nostocaceae)	[401]
Cribrostatin 6	Quinone	<i>Cribrochalina</i> spp. (Haliclonidae)	[402]
Daunomycin	anthracycline	<i>Streptomyces peucetius</i> (Streptomycetaceae)	[403]
Deoxycholic acid	Bile acid	Liver of animals	[397, 398, 404]
Deoxynivalenol (Vomitoxin)	Epoxy-sesquiterpenoid	<i>Fusarium</i> spp. (Nectriaceae)	[405, 406]
Docosahexaenoic acid (DHA)	Fatty acid	<i>Cryptocodinium cohnii</i> , <i>Schizochytrium</i> spp.	[407]

(Table 1) contd....

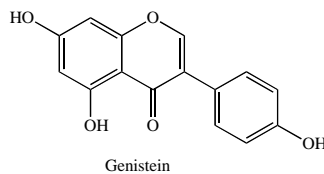
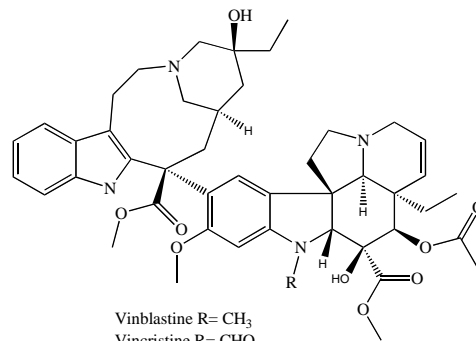
Compound	Type of Compound	Natural Source	References
Doxorubicin	Anthracycline	<i>Streptomyces</i> spp. (Streptomycetaceae)	[377, 408, 409]
Eicosapentaenoic acid (EPA)	Fatty acid	<i>Cryptocodinium cohnii</i> , <i>Parietochloris incise</i>	[96, 410, 411]
F-2 Mycotoxin (Zearalenone)	Trichothecene	<i>Gibberella</i> spp. (Nectriaceae)	[412, 413]
Fucoanthin	Carotenoid	<i>Undaria pinnatifida</i> (Alariaceae)	[414]
Isoobtusilactone A	Butanolide	<i>Cinnamomum kotoense</i> (Lauraceae)	[415, 416]
Kotomolide A	Butyrolactone	<i>Cinnamomum kotoense</i> (Lauraceae)	[417]
Mitomycin C	Aziridine	<i>Streptomyces caespitosus</i> (Streptomycetaceae)	[418-420]
Neocarzinostatin	chromoprotein enediyne	<i>Streptomyces carzinostaticus</i> (Streptomycetaceae)	[421]
Norharman	β -carboline alkaloid	<i>Passiflora incarnata</i> (Passifloraceae)	[422, 423]
Ochratoxin A	Pentaketide	<i>Aspergillus ochraceus</i> (Trichocomaceae)	[424-426]
Patulin	Fuopyranone	<i>Penicillium</i> spp. (Trichocomaceae)	[427-429]
Putrescine-1,4-dicinnamide	Phenylpropanoid	<i>Pholiota spumosa</i> (Strophariaceae)	[430]
Secotenuifolide	Butanolide	<i>Cinnamomum tenuifolium</i> (Lauraceae)	[431]
T-2 mycotoxin	Trichothecene	<i>Fusarium</i> spp. (Nectriaceae)	[432, 433]
Ursodeoxycholic acid	Bile acid	Liver of animals	[434]
Vitamin A (retinol)	Carotenoid	<i>Daucus carota</i> (Apiaceae)	[435-437]
Vitamin C (Ascorbic acid)	Butenolide	<i>Citrus</i> spp. (Rutaceae), widespread	[56, 438, 439]
Vitamin D2 (Ergocalciferol)	Steroid	<i>Lentinus edodes</i> (Marasmiaceae)	[440]
Vitamin D3 (Cholecalciferol)	Steroid	Animal origin	[441, 442]
Vitamin K2	Naftoquinone	<i>Brassica</i> spp. (Brassicaceae), widespread	[443, 444]
Vitamin K3	Naftoquinone	<i>Brassica</i> spp. (Brassicaceae) , widespread	[445, 446]



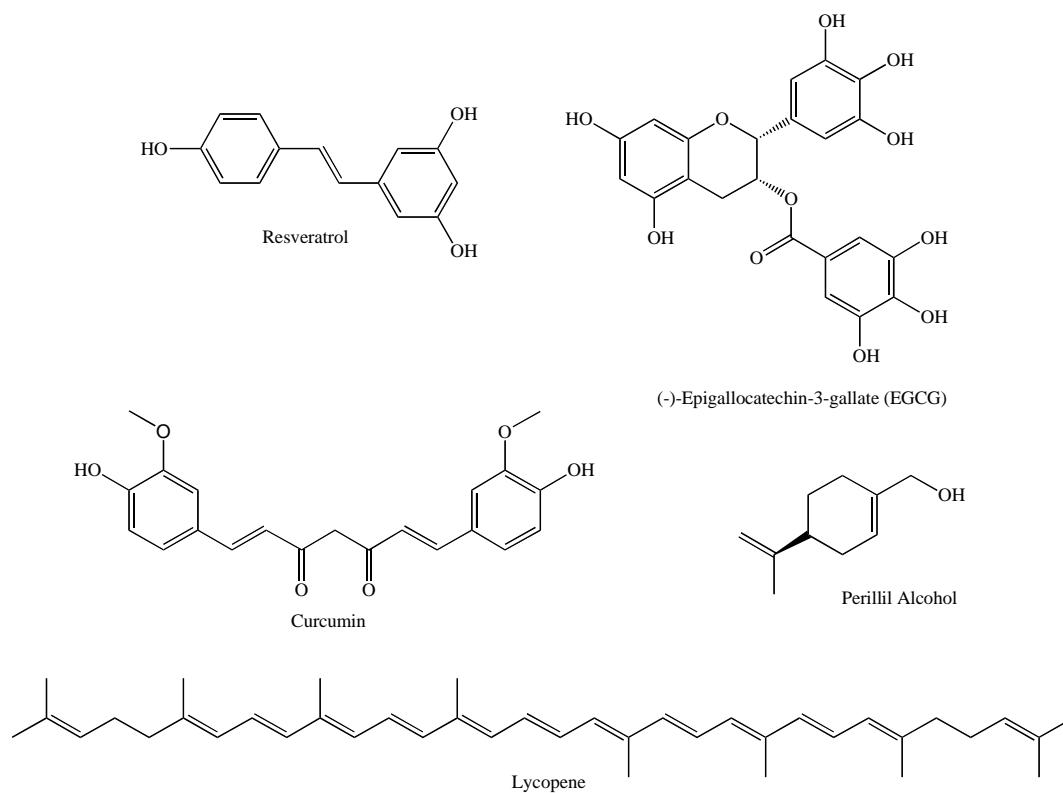
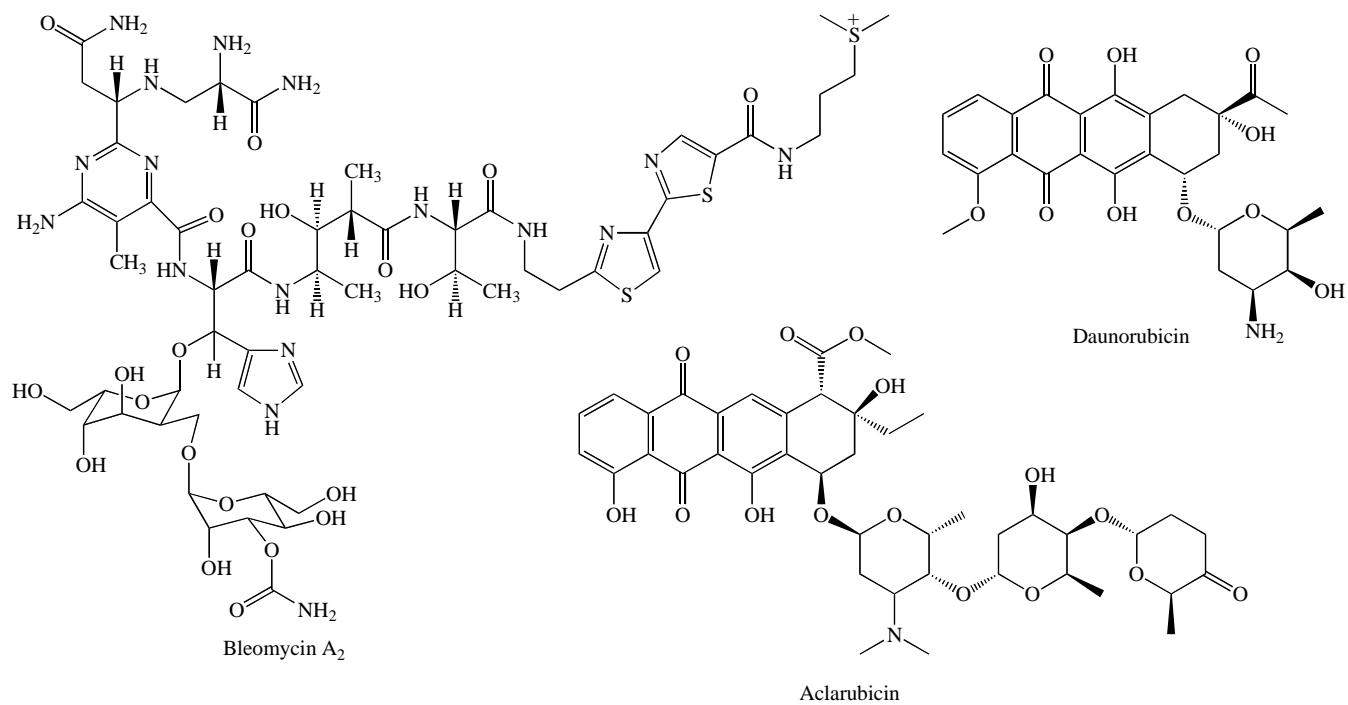
Taxol



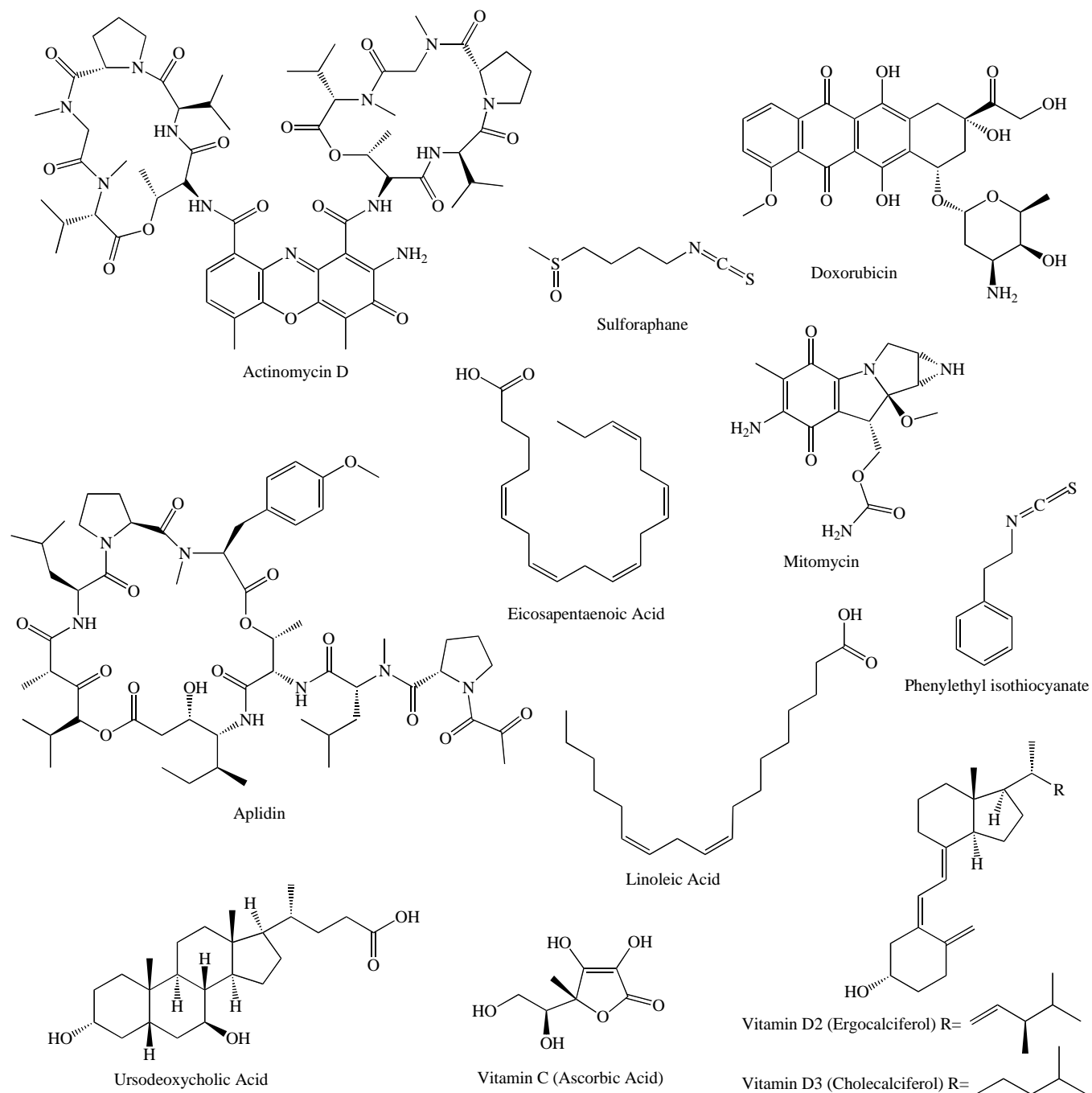
Camptothecin



(Fig. 1) contd...

**Fig. (1).** Selected plant secondary metabolites with pro-oxidant and anticancer activities.

(Fig. 2) contd...

**Fig. (2).** Selected natural products, excluding plant secondary metabolites, with pro-oxidant and anticancer activities.

vate, which is an efficient scavenger of H_2O_2 [75-78]. Second, glucose metabolism through the pentose phosphate pathway regenerates NADPH from NADP^+ in a reaction in which glucose-6-phosphate is converted into 6-phosphogluconolactone by the enzyme glucose-6-phosphate dehydrogenase. The regeneration of NADPH is required for H_2O_2 detoxification through the glutathione peroxidase/glutathione reductase system and through the thioredoxin peroxidase/thioredoxin reductase system [73, 79, 80] (reviewed in [13]). Therefore, the anticancer potential of pro-

oxidant natural products could be maximized in combination with glycolysis inhibitors.

In conclusion, natural products have made a significant contribution to the development of many anticancer drugs currently used in chemotherapy. Recent observations suggest that pro-oxidant agents may represent a new class of anticancer drugs with capacity to target tumor cells selectively. In this article, we have provided an overview of pro-oxidant natural products with anticancer activity and discussed their anticancer potential.

CONFLICT OF INTEREST

Declared none.

ACKNOWLEDGEMENT

Declared none.

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