

Purealidins J—R, New Bromotyrosine Alkaloids from the Okinawan Marine Sponge *Psammaphysilla porea*

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Nine new bromotyrosine alkaloids, purealidins J—R (1—9), have been isolated from the Okinawan marine sponge *Psammaphysilla porea* and the structures were elucidated on the basis of spectroscopic data. A hydroxy group at C-1 of purealidins M—O (4—6) may be biosynthetically derived from ring-opening of a spirocyclohexadienylisoxazole unit of purealidin J (1), aerophobin-1 (10), and purealidin L (3), respectively. Purealidins N (5), P (7), and Q (8) were cytotoxic to tumor cell lines, while purealidins J (1), K (2), P (7), and Q (8) showed moderate inhibitory activity against epidermal growth factor (EGF) receptor kinase.

Key words sponge; *Psammaphysilla porea*; bromotyrosine alkaloid; purealidin; cytotoxic; epidermal growth factor receptor kinase

Marine sponges of the order Verongidae have been found to contain a number of bromotyrosine alkaloids¹⁾ such as aerophysinin 1,²⁾ fistularin 3,³⁾ and bastadin 1.⁴⁾ In our search for bioactive substances from marine organisms,⁵⁾ a series of bromotyrosine alkaloids have been isolated from a Verongid marine sponge *Psammaphysilla porea*.⁶⁾ Further examination of the extract of *P. porea* resulted in isolation of nine new bromotyrosine-derived alkaloids, purealidins J—R (1—9), which might be closely related to one another on biogenesis of them. This paper describes the isolation and structure elucidation of 1—9.

EtOAc-soluble material of the methanolic extract of the sponge *P. porea* collected off Ishigaki island, Okinawa, was separated on silica gel and C₁₈ columns, and C₁₈ HPLC to yield purealidins J (1, 0.004% wet weight), K (2, 0.002%), L (3, 0.0002%), M (4, 0.002%), N (5, 0.0003%), O (6, 0.0001%), P (7, 0.0008%), Q (8, 0.0004%), and R (9, 0.0004%), which were obtained as the trifluoroacetic acid (TFA) salts, except for 5 and 9, together with known related compounds, aerophobin-1⁷⁾ (10, 0.005%), purealin^{6a)} (11), and purealidin B.^{6d)}

Purealidins J—L (1—3) revealed pseudomolecular ions in the ratio of 1:2:1 at *m/z* 490, 492, and 494 (1), at

m/z 506, 508, and 510 (2), and at *m/z* 494, 496, and 498 (3), respectively, in the FAB-MS spectra, indicating the presence of two bromine atoms in each molecule. HR-FAB-MS data of 1—3 revealed the molecular formulae, C₁₅H₁₇Br₂N₅O₄ (*m/z* 491.9682, M⁺ + 2 + H, Δ - 2.3 mmu), C₁₅H₁₇Br₂N₅O₅ (*m/z* 507.9672, M⁺ + 2 + H, Δ + 1.8 mmu), and C₁₅H₂₁Br₂N₅O₄ (*m/z* 496.0035, M⁺ + 2 + H, Δ + 1.7 mmu), respectively. IR absorptions at 3400 and 1680—1660 cm⁻¹ in 1—3 were attributed to NH/OH and amide carbonyl groups, respectively. Comparison of the ¹H- and ¹³C-NMR data (Table I) of 1—3 with those of aerophobin-1 (10) indicated that 1—3 possessed a common spirocyclohexadienylisoxazole unit with different structures for each C-10—C-14 segment.

Observation that purealidin J (1, [α]_D²¹ + 24° (c = 0.98, MeOH)) was positive to Sakaguchi test suggested the presence of guanidine group(s). The ¹H-NMR spectrum of 1 containing an NH₂ (δ_H 7.40, 2H) and an olefin proton (δ_H 6.62) signals was similar to that of the 2-aminohistamine unit in purealin (11) or purealidin A.^{6c)} The carbon chemical shifts (Table I) of C-12, C-13, and C-14 of 1 were coincident with those (δ_C 124.11, 109.00, and 146.74, respectively) of the corresponding 2-

TABLE I. ¹³C-NMR Data for Purealidins J—O and R (1—6 and 9) and Aerophobin-1 (10)

Positn.	1 ^{a)}	2 ^{a)}	3 ^{b)}	4 ^{c)}	5 ^{b)}	6 ^{b)}	9 ^{b)}	10 ^{a)}
1	73.6	73.6	76.3	152.8	155.0	155.2	75.5	73.5
2	113.0	113.1	114.9	107.6	108.8	108.2	114.2	113.1
3	147.1	147.2	150.1	152.7	154.3	153.6	149.3	147.1
4	120.8	120.9	123.5	105.5	107.4	108.2	122.7	120.9
5	131.2	131.3	133.0	132.2	134.6	135.4	132.3	131.2
6	90.2	90.4	93.2	122.0	122.7	123.7	92.6	90.3
7	39.2 ^{d)}	39.5 ^{d)}	40.1 ^{d)}	27.0	25.5	26.4	40.0	39.4
8	154.3	154.3	159.5	150.3	151.5	152.6	155.2	154.4
9	158.9	157.9	162.5	164.7	167.3	167.8	163.6	159.1
10	37.3	35.0	40.5	37.6	39.2	40.7		37.7
11	24.2	30.2	28.3	24.7	25.8	29.0		24.1
12	124.1	56.9	28.0	124.2	132.2	28.4		130.8
13	109.4	174.8	42.9	109.3	117.7	42.9		161.1
14	146.9	159.2	156.1	146.7	134.9	156.4		133.6
3-OCH ₃	59.6	59.7	61.2	60.0	60.8	61.7	60.4	59.6

a) In DMSO-*d*₆. b) In MeOH-*d*₄. c) In DMSO-*d*₆ with a drop of 1 N HCl. d) This carbon signal overlapped the DMSO signal.

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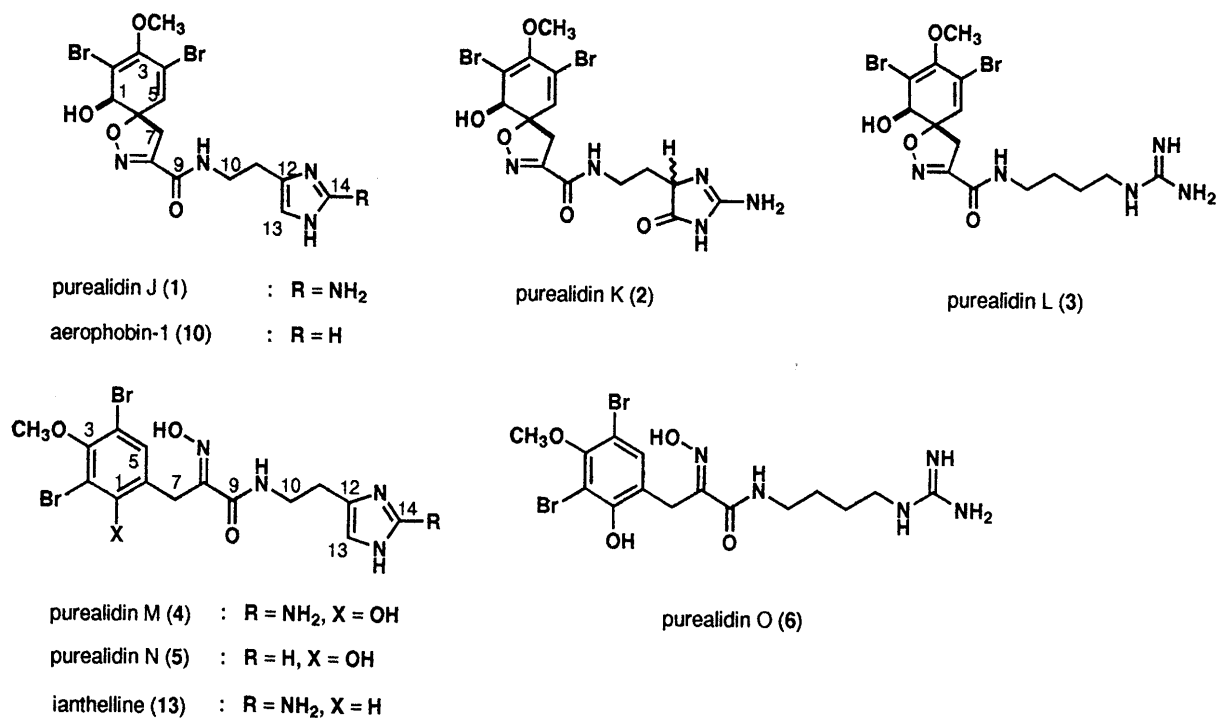


Fig. 1

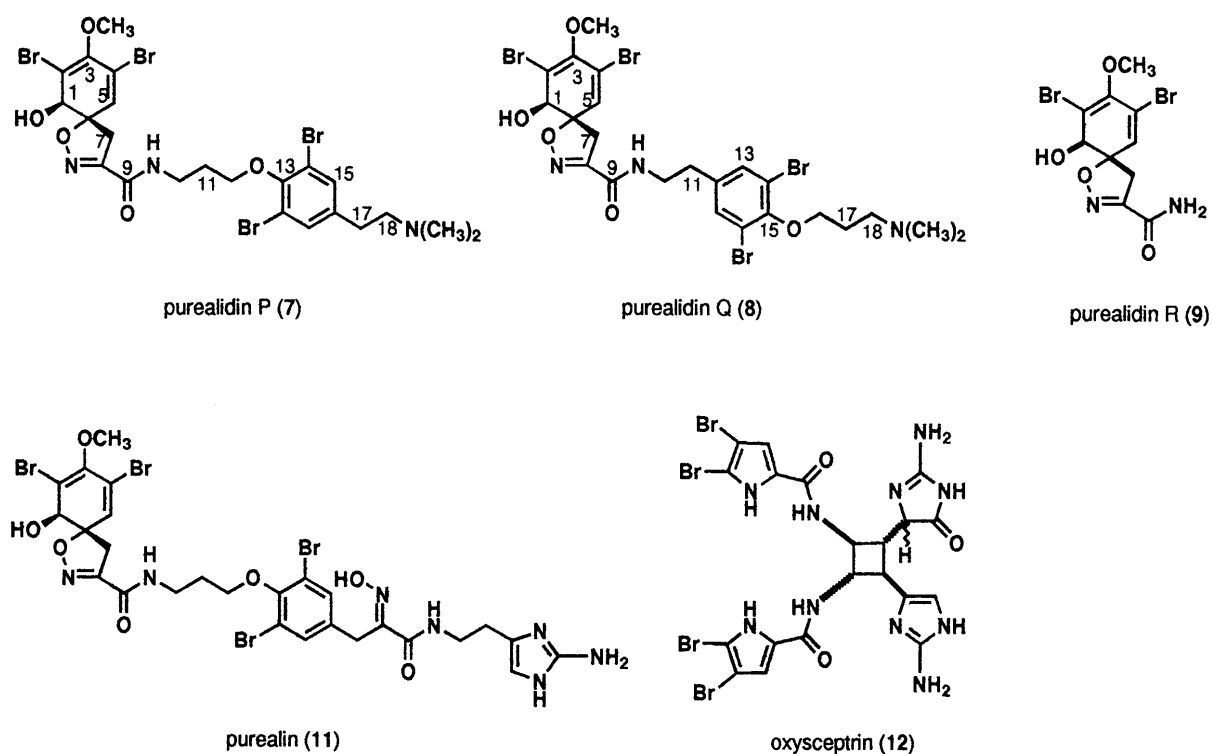


Fig. 2

aminoimidazole ring in **11**. Thus the structure of purealidin J was elucidated to be **1**.

The molecular weight of purealidin K (**2**, [α]_D²⁴ +26° (*c* = 0.38, MeOH)) was larger than that of **1** by 16 Da. The signals due to NH (δ_{H} 9.64) and a carbonyl carbon (δ_{C} 174.8) observed in the ¹H- and ¹³C-NMR spectra, respectively, were indicative of the presence of an additional amide carbonyl group. The position of the amide carbonyl (C-13) was assigned by HMBC correlations for H-11/C-12,

H-11/C-13 and 13-NH/C-13. The chemical shifts at C-12, C-13, and C-14 in **2** were close to those (δ_{C} 61.7, 174.1, and 160.5, respectively) of the aminoimidazolone ring in oxysceptrin⁸⁾ (**12**). Thus the structure of purealidin K was assigned to be **2**. Purealidin K (**2**) was subjected to ozonolysis followed by oxidation with H₂O₂ and subsequently acid hydrolysis. Standard amino acid analysis and chiral HPLC analyses of the hydrolysate revealed D- and L-2,4-diaminobutyric acid in the ratio of 1 : 1, so

that C-12 in **2** was racemic.

The ^{13}C -NMR data (Table I) of purealidin L (**3**, $[\alpha]_{\text{D}}^{24} + 27^\circ$ ($c=0.18$, MeOH)) showed characteristic resonances due to four sp^3 methylenes at δ_{C} 42.9, 40.5, 28.3, and 28.0 and an sp^2 quaternary carbon at δ_{C} 156.1. The sp^2 carbon chemical shift at C-14 as well as positive coloration in the Sakaguchi test implied the presence of a guanidino group. An agmatine (4-(aminobutyl)guanidine) moiety in **3** was assigned by the following ^1H - ^1H COSY cross-peaks: 9-NH/ H_2 -10, H_2 -10/ H_2 -11, H_2 -11/ H_2 -12, H_2 -12/ H_2 -13, and H_2 -13/ H_2 -13-NH. Thus the structure of purealidin L was elucidated to be **3**.

The molecular formula, $\text{C}_{15}\text{H}_{16}\text{Br}_2\text{N}_4\text{O}_4$, of purealidin M (**4**) was established by the HR-FAB-MS (m/z 476.9564, $\text{M}^+ + 2 + \text{H}$, $\Delta - 3.2$ mmu). Though the ^1H - and ^{13}C -NMR data (Table I) of **4** were similar to those of ianthelline⁹⁾ (**13**), differences were found for the aromatic ring (C-1—C-6): **13** possessed a symmetrical tetrasubstituted benzene ring, while the ^1H -NMR spectrum of **4** showed signals due to a pentasubstituted benzene ring (δ_{H} 7.26, 1H) and an additional phenol hydroxy group (δ_{H} 10.48). Assignment of C-1—C-6 was based on HMBC correlations for H-5/C-1, H-5/C-3, H-5/C-4, H-5/C-6, H_2 -7/C-1, H_2 -7/C-5, H_2 -7/C-6, and 3-OMe/C-3. Thus the structure of purealidin M was determined to be **4**. *E*-Geometry of the oxime at C-8 of **4**—**6** was inferred from the carbon chemical shift of C-7 (δ_{C} 25.5—27.0).¹⁰⁾

The molecular formulae, $\text{C}_{15}\text{H}_{17}\text{Br}_2\text{N}_5\text{O}_4$ and $\text{C}_{15}\text{H}_{21}\text{Br}_2\text{N}_5\text{O}_4$, of purealidins N (**5**) and O (**6**) were the same as those of aerophobin-1 (**10**) and purealidin L (**3**), respectively. Comparison of the ^1H - and ^{13}C -NMR data (Table I) of **5** and **6** with those of purealidin M (**4**) suggested that **5** and **6** had the same pentasubstituted bromotyrosine moiety (C-1—C-9) as **4**. The carbon resonances of C-10—C-14 in **5** were similar to those of the histamine unit in **10**, while those for **6** corresponded to the agmatine unit in **3**. The structures of purealidins N and O were, therefore, concluded to be **5** and **6**, respectively.

Purealidin P (**7**) showed the pseudomolecular ions at m/z 742, 744, 746, 748, and 750 (1:4:6:4:1) in the FAB-MS spectrum, and the molecular formula, $\text{C}_{23}\text{H}_{27}\text{Br}_4\text{N}_3\text{O}_5$, was established by the HR-FAB-MS (m/z 745.8776, $\text{M}^+ + 4 + \text{H}$, $\Delta + 5.5$ mmu). The ^1H - and ^{13}C -NMR data for **7** were almost same as those of purealidin B^{6d)} except for an *N*-methyl resonance (δ_{C} 59.2), which was shifted to higher field than that (δ_{C} 64.2) of purealidin B. A 6H singlet proton signal resonated at δ_{H} 2.78 was assigned to be a dimethylamino group. NOEs for H_2 -15/ H_2 -17 and H_2 -18/18-NMe₂ in addition to the above data established the structure of purealidin P to be **7**.

Purealidin Q (**8**) possessed the same molecular formula

as that of **7**, and the ^1H - and ^{13}C -NMR data suggested that **7** and **8** had a common partial structure. Analyses of the ^1H - ^1H COSY spectrum of **8** revealed proton connectivities of 9-NH— H_2 -11 and H_2 -16— H_2 -18, and the NOE observed for H_2 -18/18-NMe₂ suggested that an *N*-dimethyl group was attached to a C₃ unit. Thus the structure of purealidin Q was assigned to be **8**.

Purealidin R (**9**, $[\alpha]_{\text{D}}^{24} + 86^\circ$ ($c=0.19$, MeOH)) showed molecular ions at m/z 380, 382, and 384 (1:2:1) in the EI-MS spectrum. The molecular formula, $\text{C}_{10}\text{H}_{10}\text{Br}_2\text{N}_2\text{O}_4$, was determined by HR-EI-MS (m/z 381.8901, $\text{M}^+ + 2$, $\Delta + 0.3$ mmu). The ^1H - and ^{13}C -NMR data (Table I) revealed the presence of a spirocyclohexadienylisoxazole unit and a primary amide NH₂, thus suggesting that the structure of purealidin R is **9**.

Stereochemistry at C-1 and C-6 of the spiroisoxazole ring in purealidins J (**1**), K (**2**), L (**3**), P (**7**), Q (**8**), and R (**9**), and aerophobin-1 (**10**) was deduced to be *trans* from the proton chemical shift (*ca.* δ_{H} 4.05) of H-1 in CD₃OD.¹¹⁾ These compounds were found to be dextrorotatory from the signs of optical rotations and Cotton effects at 248 and 284 nm in the CD spectra, indicating that absolute configurations at C-1 and C-6 are 1*R* and 6*S*.^{6a,6d,12)} The $[\alpha]_{\text{D}}$ value (+68°) of compound **10** is smaller than that ($[\alpha]_{\text{D}}$ +139°) in the literature,⁷⁾ indicating that aerophobin-1 (**10**) isolated from this sponge may be an enantiomeric mixture containing an excess of the dextrorotamer (*ca.* 50% ee).

Purealidins M (**4**), N (**5**) and O (**6**) are the second examples¹³⁾ of bromotyrosines with a phenolic hydroxy group at C-1 from natural origin.¹⁴⁾ Biosynthetically compounds **1**—**9** may be closely related to one another, presumably being generated through condensation of bromotyrosine units (C-1—C-9) with various amines such as agmatine, brominated tyramine with C₃ units, ammonia, or histamine. The spirocyclohexadienylisoxazole ring may be derived from an arene oxide intermediate (**a**) as proposed by Andersen and Faulkner¹⁵⁾ (Chart 1). Ring-opening of the spirocyclohexadienylisoxazole units of purealidins J (**1**), L (**3**), and aerophobin-1 (**10**) may generate purealidins M (**4**), O (**6**), and N (**5**), respectively, with a hydroxy group at C-1. The aminohistamine units of **1** and **4** may be biogenetically derived from the agmatine units of **3** and **6**. Cyclization between C-12 and 14-N in **3** may afford **1**, which is probably oxygenated to yield **2**.

Purealidins N (**5**), P (**7**), and Q (**8**) exhibited cytotoxicity against murine lymphoma L1210 cells (IC₅₀ values: 0.07, 2.8, and 0.95 $\mu\text{g}/\text{ml}$, respectively) and human epidermoid carcinoma KB cells (IC₅₀: 0.074, 7.6, and 1.2 $\mu\text{g}/\text{ml}$, respectively) *in vitro*, while purealidins J—M (**1**—**4**), O (**6**) and R (**9**), and aerophobin-1 (**10**) had no cytotoxicity (IC₅₀

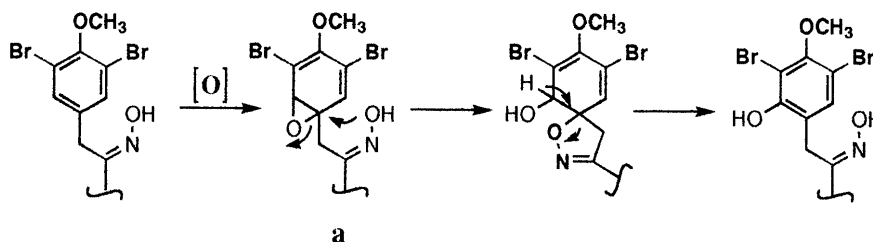


Chart 1. Plausible Biogenetic Path of Bromotyrosines with a Spirocyclohexadienylisoxazole Unit and/or a Hydroxy Group at C-1

> 10 $\mu\text{g/ml}$). Puralidins J (1), K (2), P (7), and Q (8) showed inhibitory activity against epidermal growth factor (EGF) receptor kinase¹⁶⁾ (IC_{50} : 23, 14, 18, and 11 $\mu\text{g/ml}$, respectively).

Experimental

General Methods UV and IR spectra were taken on a JASCO Ubest-35 and a JASCO IR Report-100 spectrometer, respectively. ¹H- and ¹³C-NMR spectra were conducted with a JEOL EX-400 and GSX-270 spectrometer in CD₃OD and DMSO-*d*₆. The resonances of MeOH at δ_{H} 3.30 and δ_{C} 49.0 were used as internal references for ¹H- and ¹³C-NMR spectra, respectively. The resonances of residual DMSO at δ_{H} 2.50 and δ_{C} 39.5 were used as internal references for ¹H- and ¹³C-NMR spectra, respectively. FAB-MS spectra were recorded employing a JEOL HX-110 spectrometer by using glycerol as a matrix. EI-MS spectra were obtained on a JEOL DX-303 spectrometer operating at 70 eV.

Collection, Extraction, and Isolation The dark brown sponge, *Psammaplysilla purea* Carter, was collected off Ishigaki Island, Okinawa and kept frozen until needed. The sponge (1.5 kg, wet weight) was extracted with MeOH (1.3 l and then 1 l). After evaporation under reduced pressure the residue (55.6 g) was partitioned between EtOAc (500 ml \times 3) and H₂O (500 ml). The EtOAc soluble material (3.30 g) was subjected to a silica gel column with CHCl₃-*n*-BuOH-AcOH-H₂O (1.5:6:1:1). The fraction (134 mg, 760–940 ml) was chromatographed on a C₁₈ column (Develosil LOP ODS 24S, Nomura Chemical, 30 \times 300 mm) with CH₃CN-H₂O-CF₃CO₂H (35:65:0.1) and then MeOH to give two fractions. The fraction (34 mg, 45–100 ml) was separated by C₁₈ HPLC (YMC Pack AM323 ODS, 10 \times 250 mm; eluent, CH₃CN-H₂O-CF₃CO₂H, 35:75:0.1; flow rate, 2.5 ml/min; UV detection at 254 nm) to afford puralidins J (1, 0.004%, *t*_R 10.8 min), K (2, 0.002%, *t*_R 10.2 min) and L (3, 0.0002%, *t*_R 11.5 min). The other fraction (36.4 mg) eluted with MeOH from the C₁₈ column was subjected to C₁₈ HPLC (YMC Pack AM323 ODS, 10 \times 250 mm; eluent, CH₃CN-H₂O-CF₃CO₂H, 42:58:0.1; flow rate, 2.5 ml/min; UV detection at 254 nm) to afford puralidins N (5, 0.0003%, *t*_R 12.0 min), O (6, 0.0001%, *t*_R 14.0 min), and R (9, 0.0004%, *t*_R 19.0 min). The other fraction (219.2 mg) eluted at 950–1080 ml from the first silica gel column was subjected to a C₁₈ column (Develosil LOP ODS 24S, 30 \times 300 mm; eluent, CH₃CN-H₂O-CF₃CO₂H, 45:55:0.1) followed by purification by C₁₈ HPLC (YMC Pack AM323 ODS, 10 \times 250 mm; eluent, CH₃CN-H₂O-CF₃CO₂H, 45:55:0.1; flow rate, 2.5 ml/min; UV detection at 254 nm) to afford puralidins M (4, 0.002%, *t*_R 8.4 min), P (7, 0.0008%, *t*_R 17.4 min), and Q (8, 0.0004%, *t*_R 16.2 min) together with aerophobin-1 (10, 0.005%).

Puralidin J (1) TFA Salt Colorless oil, $[\alpha]_{\text{D}}^{21} + 24^\circ$ ($c=0.98$, MeOH). IR (KBr): 3400, 2930, 2845, 1680, 1540, 1430, 1200, 1135 cm^{-1} . UV $\lambda_{\text{max}}^{\text{MeOH}}$ nm (ϵ): 277 (1700), 284 (1400). CD ($c=0.64 \times 10^{-4}$, MeOH) $\Delta\epsilon^{30}$ (nm): +2.2 (255), +2.1 (287). ¹H-NMR (DMSO-*d*₆) δ : 2.61 (2H, t, $J=6.4$ Hz, H₂-11), 3.19 (1H, d, $J=18.6$ Hz, H-7), 3.37 (2H, br t, $J=6.4$ Hz, H₂-10), 3.61 (1H, d, $J=18.6$ Hz, H-7), 3.64 (3H, s, 3-OMe), 3.91 (1H, d, $J=6.4$ Hz, H-1), 6.37 (1H, d, $J=6.4$ Hz, 1-OH), 6.58 (1H, s, H-5), 6.62 (1H, br s, H-13), 7.40 (2H, s, 14-NH₂), 8.63 (1H, t, $J=5.6$ Hz, 9-NH), 11.71 (1H, br s, 12-NH), 12.13 (1H, br s, 13-NH). ¹³C-NMR: see Table I. FAB-MS m/z : 490, 492, 494 ($\text{M}^+ + \text{H}$, 1:2:1). HR-FAB-MS Calcd for C₁₅H₁₉Br₂N₅O₄ ($\text{M}^+ + \text{H}$)⁺: 491.9705. Found: 491.9682.

Puralidin K (2) TFA Salt Colorless oil, $[\alpha]_{\text{D}}^{24} + 26^\circ$ ($c=0.38$, MeOH). IR (KBr): 3400, 2920, 2850, 1670, 1540, 1430, 1200, 1135, 1125 cm^{-1} . UV $\lambda_{\text{max}}^{\text{MeOH}}$ nm (ϵ): 231 (6300), 284 (2400). CD ($c=0.73 \times 10^{-4}$, MeOH) $\Delta\epsilon^{30}$ (nm): +2.3 (255), +2.1 (289). ¹H-NMR (DMSO-*d*₆) δ : 1.86 (1H, m, H-11), 1.96 (1H, m, H-11), 3.19 (1H, d, $J=18.0$ Hz, H-7), 3.35 (2H, m, H₂-10), 3.62 (1H, d, $J=18.0$ Hz, H-7), 3.65 (3H, s, 3-OMe), 3.92 (1H, d, $J=6.3$ Hz, H-1), 4.30 (1H, dd, $J=4.9$, 7.8 Hz, H-12), 6.36 (1H, d, $J=6.3$ Hz, 1-OH), 6.58 (1H, s, H-5), 8.63 (1H, t, $J=5.1$ Hz, 9-NH), 8.96 (2H, s, 14-NH₂), 9.64 (1H, br s, 13-NH), 12.5 (1H, br, 12-NH). ¹³C-NMR: see Table I. FAB-MS m/z : 506, 508, 510 ($\text{M}^+ + \text{H}$, 1:2:1). HR-FAB-MS Calcd for C₁₅H₁₈Br₂N₅O₅ ($\text{M}^+ + \text{H}$)⁺: 507.9654. Found: 507.9672.

Puralidin L (3) TFA Salt Colorless oil, $[\alpha]_{\text{D}}^{24} + 27^\circ$ ($c=0.18$, MeOH). IR (KBr): 3400, 2920, 1660, 1520, 1470, 1210 cm^{-1} . UV $\lambda_{\text{max}}^{\text{MeOH}}$ nm (ϵ): 228 (8600), 290 (3400). CD ($c=1.4 \times 10^{-4}$, MeOH) $\Delta\epsilon^{30}$ (nm): +2.3 (248), +1.9 (284). ¹H-NMR (DMSO-*d*₆) δ : 1.45 (2H, m, H₂-12), 1.47 (2H, m, H₂-11), 3.09 (2H, m, H₂-13), 3.16 (2H, m, H₂-10), 3.19 (1H,

d, $J=18.2$ Hz, H-7), 3.62 (1H, d, $J=18.2$ Hz, H-7), 3.64 (3H, s, 3-OMe), 3.91 (1H, d, $J=7.9$ Hz, H-1), 6.34 (1H, d, $J=7.9$ Hz, 1-OH), 6.57 (1H, s, H-5), 7.1–7.3 (4H, br, 14-C(-NH₂)NH₂), 7.44 (1H, br s, 13-NH), 8.54 (1H, t, $J=5.8$ Hz, 9-NH). ¹³C-NMR: see Table I. FAB-MS m/z : 494, 496, 498 ($\text{M}^+ + \text{H}$, 1:2:1). HR-FAB-MS Calcd for C₁₅H₂₂Br₂N₅O₄ ($\text{M}^+ + \text{H}$)⁺: 496.0018. Found: 496.0035.

Puralidin M (4) TFA Salt Colorless oil. IR (KBr): 3400, 2920, 1680, 1520, 1470, 1200, 1135, 1120 cm^{-1} . UV $\lambda_{\text{max}}^{\text{MeOH}}$ nm (ϵ): 277 (1700), 284 (1400). ¹H-NMR (DMSO-*d*₆) δ : 2.63 (2H, t, $J=6.9$ Hz, H₂-11), 3.40 (2H, dt, $J=5.9$, 6.9 Hz, H₂-10), 3.70 (2H, s, H₂-7), 3.74 (3H, s, 3-OMe), 6.60 (1H, s, H-13), 7.26 (1H, s, H-5), 7.33 (2H, s, 14-NH₂), 8.55 (1H, t, $J=5.9$ Hz, 9-NH), 10.48 (1H, br s, 1-OH), 11.55 (1H, br s, 12-NH), 11.99 (1H, br s, 13-NH), 12.19 (1H, s, 8-NOH). ¹³C-NMR: see Table I. FAB-MS m/z : 490, 492, 494 ($\text{M}^+ + \text{H}$, 1:2:1). HR-FAB-MS Calcd for C₁₅H₁₈Br₂N₅O₄ ($\text{M}^+ + \text{H}$)⁺: 491.9705. Found 491.9740.

Puralidin N (5) TFA Salt Colorless oil. IR (KBr): 3400, 2920, 2845, 1680, 1520, 1135, 1120 cm^{-1} . UV $\lambda_{\text{max}}^{\text{MeOH}}$ nm (ϵ): 235 (2600), 290 (2800). ¹H-NMR (DMSO-*d*₆) δ : 2.85 (2H, t, $J=6.7$ Hz, H₂-11), 3.48 (2H, dt, $J=5.9$, 6.7 Hz, H₂-10), 3.70 (2H, s, H₂-7), 3.74 (3H, s, 3-OMe), 7.26 (1H, s, H-5), 7.40 (1H, s, H-13), 8.59 (1H, t, $J=5.9$ Hz, 9-NH), 8.89 (1H, s, H-14), 10.45 (1H, br s, 1-OH), 12.20 (1H, s, 8-NOH), 14.1 (1H, br s, 13-NH). ¹³C-NMR: see Table I. FAB-MS m/z : 475, 477, 479 ($\text{M}^+ + \text{H}$, 1:2:1). HR-FAB-MS Calcd for C₁₅H₁₇Br₂N₄O₄ ($\text{M}^+ + \text{H}$)⁺: 476.9596. Found: 476.9564.

Puralidin O (6) TFA Salt Colorless oil. IR (KBr): 3400, 2920, 2850, 1680, 1635, 1135 cm^{-1} . UV $\lambda_{\text{max}}^{\text{MeOH}}$ nm (ϵ): 235 nm (2600), 287 (700). ¹H-NMR (DMSO-*d*₆) δ : 1.42 (2H, m, H₂-12), 1.44 (2H, m, H₂-11), 3.09 (2H, m, H₂-13), 3.18 (2H, m, H₂-10), 3.70 (2H, s, H₂-7), 3.74 (3H, s, 3-OMe), 7.1–7.3 (4H, br, 14-C(-NH₂)NH₂), 7.31 (1H, s, H-5), 7.43 (1H, br s, 13-NH), 8.56 (1H, t, $J=5.9$ Hz, 9-NH), 10.68 (1H, br s, 1-OH), 12.17 (1H, s, 8-NOH). ¹³C-NMR: see Table I. FAB-MS m/z : 494, 496, 498 ($\text{M}^+ + \text{H}$, 1:2:1). HR-FAB-MS Calcd for C₁₅H₂₂Br₂N₅O₄ ($\text{M}^+ + \text{H}$)⁺: 496.0018. Found: 496.0028.

Puralidin P (7) TFA Salt Colorless oil, $[\alpha]_{\text{D}}^{19} + 6.6^\circ$ ($c=0.75$, MeOH). IR (KBr): 3400, 2940, 2845, 1670, 1520, 1470, 1135, 1120 cm^{-1} . UV $\lambda_{\text{max}}^{\text{MeOH}}$ nm (ϵ): 277 (1700), 284 (1400). CD ($c=0.37 \times 10^{-4}$, MeOH) $\Delta\epsilon^{30}$ (nm): +0.9 (248), +1.1 (284). ¹H-NMR (DMSO-*d*₆) δ_{H} : 2.01 (2H, m, H₂-11), 2.78 (6H, s, 18-NMe₂), 2.93 (2H, t, $J=8.1$ Hz, H₂-17), 3.23 (1H, d, $J=18.0$ Hz, H-7), 3.35 (2H, m, H₂-18), 3.41 (2H, m, H₂-10), 3.62 (1H, d, $J=18.0$ Hz, H-7), 3.65 (3H, s, 3-OMe), 3.92 (1H, d, $J=8.1$ Hz, H-1), 3.97 (2H, t, $J=6.3$ Hz, H₂-12), 6.36 (1H, d, $J=8.1$ Hz, 1-OH), 6.59 (1H, s, H-5), 7.64 (2H, s, H-15, 15'), 8.57 (1H, t, $J=5.8$ Hz, 9-NH). ¹³C-NMR (CD₃OD) δ_{C} : 30.3 (t, C-17), 30.6 (t, C-11), 37.9 (t, C-10), 40.1 (t, C-7), 43.6 (2C, q, 18-NMe₂), 59.2 (t, C-18), 60.4 (q, 1-OMe), 72.3 (t, C-12), 75.5 (d, C-1), 92.5 (s, C-6), 114.2 (s, C-2), 119.5 (2C, s, C-14, 14'), 122.8 (s, C-4), 132.3 (d, C-5), 134.4 (d, C-15), 134.5 (d, C-15'), 136.5 (s, C-16), 149.4 (s, C-3), 153.7 (s, C-13), 155.3 (s, C-8), 161.6 (s, C-9). FAB-MS m/z : 742, 744, 746, 748, 750 ($\text{M}^+ + \text{H}$, 1:4:6:4:1). HR-FAB-MS Calcd for C₂₃H₂₈Br₄N₃O₅ ($\text{M}^+ + \text{H}$)⁺: 745.8721. Found: 745.8776.

Puralidin Q (8) TFA Salt Colorless oil, $[\alpha]_{\text{D}}^{19} + 9.1^\circ$ ($c=0.39$, MeOH). IR (KBr): 3400, 2940, 2850, 1675, 1470, 1200, 1135, 1120 cm^{-1} . UV $\lambda_{\text{max}}^{\text{MeOH}}$ nm (ϵ): 277 (1700), 284 (1400). CD ($c=0.52 \times 10^{-4}$, MeOH) $\Delta\epsilon^{30}$ (nm): +1.5 (250), +1.4 (284). ¹H-NMR (DMSO-*d*₆) δ_{H} : 2.15 (2H, m, H₂-17), 2.76 (2H, t, $J=6.9$ Hz, H₂-11), 2.83 (6H, s, 18-NMe₂), 3.18 (1H, d, $J=18.2$ Hz, H-7), 3.36 (2H, m, H₂-10), 3.40 (2H, m, H₂-18), 3.60 (1H, d, $J=18.2$ Hz, H-7), 3.64 (3H, s, 3-OMe), 3.92 (1H, d, $J=8.2$ Hz, H-1), 3.99 (2H, t, $J=6.3$ Hz, H₂-16), 6.34 (1H, d, $J=8.2$ Hz, 1-OH), 6.57 (1H, s, H-5), 7.54 (2H, s, H-15, 15'), 8.58 (1H, t, $J=5.7$ Hz, 9-NH). ¹³C-NMR (CD₃OD) δ_{C} : 26.4 (t, C-17), 35.1 (t, C-11), 40.1 (t, C-10), 41.4 (t, C-7), 43.7 (2C, q, 18-NMe₂), 57.2 (t, C-18), 60.4 (q, 1-OMe), 71.2 (t, C-16), 75.5 (d, C-1), 92.4 (s, C-6), 114.2 (s, C-2), 118.8 (2C, s, C-14, 14'), 122.8 (s, C-4), 132.2 (d, C-5), 134.5 (d, C-13), 134.6 (d, C-13'), 140.3 (s, C-12), 149.3 (s, C-3), 152.2 (s, C-15), 155.2 (s, C-8), 161.6 (s, C-9). FAB-MS m/z : 742, 744, 746, 748, 750 ($\text{M}^+ + \text{H}$, 1:4:6:4:1). HR-FAB-MS Calcd for C₂₃H₂₈Br₄N₃O₅ ($\text{M}^+ + \text{H}$)⁺: 745.8721. Found: 745.8729.

Puralidin R (9) TFA Salt Colorless oil, $[\alpha]_{\text{D}}^{24} + 86^\circ$ ($c=0.19$, MeOH). IR (KBr): 3400, 2940, 1675, 1135, 1120 cm^{-1} . UV $\lambda_{\text{max}}^{\text{MeOH}}$ nm (ϵ): 228 (8000), 290 (2000). CD ($c=0.52 \times 10^{-4}$, MeOH) $\Delta\epsilon^{30}$ (nm): +4.0 (248), +4.3 (284). ¹H-NMR (DMSO-*d*₆) δ : 3.18 (1H, d, $J=18.2$ Hz, H-7), 3.60 (1H, d, $J=18.2$ Hz, H-7), 3.65 (3H, s, 3-OMe), 3.92 (1H, d, $J=6.2$ Hz, H-1), 6.35 (1H, d, $J=6.2$ Hz, 1-OH), 6.58 (1H, s, H-5), 7.58 (1H, br s, NH-9), 7.82 (1H, br s, NH-9). ¹³C-NMR: see Table I. EI-MS m/z : 380, 382, 384

(M⁺ + H, 1:2:1). HR-EL-MS Calcd for C₁₀H₁₀Br₂N₂O₄ (M+2)⁺: 381.8898. Found: 381.8901.

Determination of the Stereochemistry of C-12 in Puralidin K (2) A solution of purealidin K (2, 1.0 mg) in MeOH (100 μl) was bubbled with O₃ at -78 °C for 1 min. After removal of excess O₃ with a stream of N₂, the solvent was evaporated under reduced pressure, and to the residue were added HCO₂H (250 ml) and 35% H₂O₂ (25 μl). The mixture was stirred for 1 h at 0 °C and then for 18 h at room temperature. The solvent was evaporated, and then the residue was dissolved in 6 N HCl (100 μl); this solution was heated at 110 °C for 24 h. Standard amino acid analysis was performed on a Hitachi amino acid autoanalyzer (Model 835, column #2617 4.0 × 250 mm) with Na buffer at a flow rate of 0.275 ml/min and detected at 570 nm. 2,4-Diaminobutyric acid (t_R 96.42 min) was found in the hydrolysate. The absolute stereochemistry of 2,4-diaminobutyric acid was examined by chiral HPLC analysis (Sumichiral OA-6000, Sumika Chemical Analysis Service, Ltd., 4.6 × 150 mm; flow rate, 0.5 ml/min; UV detection at 254 nm; eluent, 1 mM aqueous CuSO₄). The retention times of authentic L- and D-2,4-diaminobutyric acid were 30.5 and 29.6 min, respectively. The retention times of the 2,4-diaminobutyric acid in the hydrolysate was found to be 30.5 and 29.6 min (ratio of ca. 1:1).

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