## Quantification of dissolved CO<sub>2</sub> in silicate glasses using micro-Raman spectroscopy

## YANN MORIZET<sup>1,2,\*</sup>, RICHARD A. BROOKER<sup>3</sup>, GIADA IACONO-MARZIANO<sup>2</sup> AND BRUCE A. KJARSGAARD<sup>4</sup>

<sup>1</sup>Laboratoire de Planétologie et Géodynamique de Nantes (LPGN) UMR CNRS 6112, Université de Nantes, Nantes Atlantique Universités, 2 rue de la Houssinière, 44322 Nantes, France

<sup>2</sup>Institut des Sciences de la Terre D'Orléans (ISTO) UMR CNRS 7327, Université d'Orléans, Campus Géosciences, 1A rue de la Férolerie, 45071 Orléans Cedex 2, France

<sup>3</sup>School of Earth Sciences, University of Bristol, Wills Memorial Building, Queens Road, Bristol BS8 1RJ, U.K. <sup>4</sup>Geological Survey of Canada, 601 Booth Street, Ottawa, Ontario K1A 0E8, Canada

## ABSTRACT

This study investigates the potential use of confocal micro-Raman spectroscopy for the quantification of CO<sub>2</sub> in geologically relevant glass compositions. A calibration is developed using a wide range of both natural and synthetic glasses that have  $CO_2$  dissolved as carbonate ( $CO_3^{-}$ ) in the concentration range from 0.2 to 16 wt%. Spectra were acquired in the 200 and 1350 cm<sup>-1</sup> frequency region that includes the  $v_1$  Raman active vibration for carbonate at 1062–1092 cm<sup>-1</sup> and the intensity of this peak is compared to various other peaks representing the aluminosilicate glass structure. The most precise and accurate calibration is found when carbonate peaks are compared to aluminosilicate spectral features in the high-frequency region (HF: 700-1200 cm<sup>-1</sup>), which can be simulated with several Gaussian peaks, directly related to different structural species in the glass. In some samples the "dissolved"  $CO_{2}^{-}$  appears to have two different Raman bands, one sharper than the other. This may be consistent with previous suggestions that  $CO_{3}^{2-}$  has several structural environments in the glass, and is not related to any precipitation of crystalline carbonate from the melt during quenching. The calibration derived using the HF peaks appears linear for both the full range of glass composition considered and the range of CO<sub>2</sub> concentrations, even when multiple carbonate peaks are involved. We propose the following, compositionally independent linear equation to quantify the  $CO_2$  content in glass with micro-Raman spectroscopy

wt%  $CO_2 = 15.17 \times CO_3/HF$ 

where  $CO_3/HF$  is the area ratio of the fitted  $v_1$  carbonate peak(s) at 1062–1092 cm<sup>-1</sup> to the remaining area of the fitted aluminosilicate envelope from 700–1200 cm<sup>-1</sup>. This is similar to the Raman calibration developed for water, but is complicated by the overlapping of these two fitted components. Using error propagation, we propose the calibration accuracy is better than ±0.4 wt% CO<sub>2</sub> for our data set.

The  $v_1$  Raman peak position for carbonate is not constant and appears to be correlated with the density of the melt (or glass) in some way rather than the chemical composition.

Keywords: Raman spectroscopy, CO<sub>2</sub> content, silicate glass, melt density, CO<sub>3</sub> molecules Raman shift