

Quantification of energy losses in organic solar cells from temperature-dependent device characteristics

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Owing to the excitonic nature of photoexcitations in organic semiconductors, the working mechanism of organic solar cells relies on the donor-acceptor (D/A) concept enabling photoinduced charge transfer at the interface between two organic materials with suitable energy-level alignment. However, the introduction of such a heterojunction is accompanied by additional energy losses compared to an inorganic homojunction cell due to the presence of a charge-transfer (CT) state at the D/A interface. By careful examination of planar heterojunctions of the molecular semiconductors diindenoperylene (DIP) and C₆₀ we demonstrate that three different analysis techniques of the temperature dependence of solar-cell characteristics yield reliable values for the effective photovoltaic energy gap at the D/A interface. The retrieved energies are shown to be consistent with direct spectroscopic measurements and the D/A energy-level offset determined by photoemission spectroscopy. Furthermore, we verify the widespread assumption that the activation energy of the dark saturation current ΔE and the CT energy E_{CT} may be regarded as identical. The temperature-dependent analysis of open-circuit voltage V_{OC} and dark saturation current is then applied to a variety of molecular planar heterojunctions. The congruency of ΔE and E_{CT} is again found for all material systems with the exception of copper phthalocyanine/C₆₀. The general rule of thumb for organic semiconductor heterojunctions, that V_{OC} at room temperature is roughly half a volt below the CT energy, is traced back to comparable intermolecular electronic coupling in all investigated systems.

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I. INTRODUCTION

Organic solar cells (OSCs) have recently surmounted the important barrier of 10% efficiency for the conversion of incoming solar to electrical power,¹ and the current record value for the power-conversion efficiency of molecular OSCs has even reached 12%.² The rapid progress over recent years poses the question of where the fundamental efficiency limits of these materials are.³

For inorganic photovoltaics, this question was answered more than 50 years ago by the well-known Shockley-Queisser (SQ) limit.⁴ Based on very few fundamental assumptions, the SQ theory is able to predict an upper limit for the power-conversion efficiency of single-junction cells with the energy gap E_g of the active semiconductor material as the only input parameter. Owing to their excitonic nature, however, an important modification of this concept is required for OSCs, where a donor-acceptor heterojunction is used to enable charge separation.⁵ This is related to the existence of a charge-transfer (CT) state [see Fig. 1(a)] as an intermediate in the dissociation process from the photoexcited (excitonic) state on one molecular species towards free charge-separated states on both molecular species.^{6–9} The inclusion of the energy E_{CT} and absorption strength α_{CT} of the CT state at the donor-acceptor interface in thermodynamic efficiency limits for OSCs has recently been demonstrated by different authors.^{10–13} They found that the photocurrent is not severely affected by the existence of a CT state, as long as its absorption strength is weak compared to that of the main absorber material. In that case the *number* of generated charge carriers remains governed by the fundamental absorption across the optical gap of the absorber. The open-circuit voltage V_{OC} , on the

other hand, is a measure of the *chemical potential* of the photogenerated charge-carrier pairs. As such, it is strongly affected by the introduction of the CT energy, which needs to be lower than the optical gap in order to enable charge transfer. Because of this fact E_{CT} takes the role of E_g in the derivation of the thermodynamic efficiency limits of OSCs and is sometimes also termed photovoltaic gap¹⁴ or effective gap.¹⁵ As a consequence, calculations predict both lower power-conversion efficiencies (20% to 25%) compared to the classical SQ model and larger values for the optical gap, where the highest efficiency is achieved.^{11,13} Note that practical efficiency limits are lower by a factor of 2 if more realistic values ($\sim 75\%$ each) are chosen instead of 100% quantum efficiency and ideal fill factors.^{3,6,12,16,17}

In the meantime the concept of an interfacial CT state is well established in polymer-fullerene bulk-heterojunction solar cells. Its existence has unambiguously been proven by spectral subgap absorption and photocurrent measurements as well as by electroluminescence spectra,^{18,19} and a consistent description of temperature-dependent device characteristics was obtained.²⁰ These investigations could directly show that E_{CT} sets the upper limit for V_{OC} as temperature approaches 0 K. However, for molecular semiconductors such a conclusive picture of the role of CT states at mostly planar donor-acceptor (D/A) heterojunctions is still missing. Rand *et al.* have presented the first systematic study of the interrelation between the measured open-circuit voltage and the energy-level offset E_{DA} on a range of D/A systems.²¹ They observed an increase of the open-circuit voltage V_{OC} with decreasing temperature until a saturation was reached. This maximum V_{OC} was found to correlate with the energy-level offset at the D/A interface, as estimated from the ionization potentials of the donor and

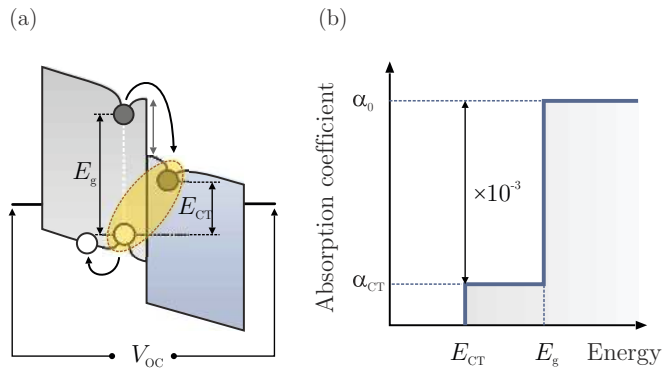


FIG. 1. (Color online) (a) Schematic illustration of the formation of a CT state with energy E_{CT} . (b) Steplike absorption in a D/A heterojunction solar cell. In addition to the absorption α_0 of photons with energy exceeding the optical energy gap E_g , absorption from the CT state α_{CT} already takes place for photon energies above E_{CT} but is typically several orders of magnitude weaker.

acceptor and the optical gap of the absorber. Potscavage *et al.* used the thermally activated behavior of the dark saturation current of solar cells to identify the crucial role of the energy-level offset at the D/A interface for V_{OC} .²² Perez *et al.* noted that in addition to the activation energy ΔE of the dark current under reverse-bias conditions, its prefactor, depending on the electronic coupling between both molecular species, is also important for the magnitude of the recombination current and thus determines V_{OC} .²³ Giebink *et al.* have put forward an extended diode model for organic heterojunctions based on polaron pair generation and recombination processes.²⁴ Remarkably, they predict two different ideality factors related to trapping in both materials forming the interface and apply their model to analyze current-voltage characteristics in terms of polaron pair recombination. However, in spite of the successful description of the temperature-dependent device behavior in the dark and under illumination, no direct spectroscopic evidence for the existence of a CT state at the respective energies was given. Note that the claimed evidence for an exciplex emission of CuPc/ C_{60} in Ref. 24 has to be considered as indirect evidence because no clear spectroscopic signature was found. Furthermore, the two different ideality factors postulated are expected to cause two different slopes in $V_{OC}(T)$, which have not actually been observed.²⁴

The aims of this paper are therefore, on the one hand, to establish temperature-dependent current-voltage characteristics (in the dark and under illumination) of *planar heterojunction* OSCs as a straightforward method for determining the effective photovoltaic gap and, on the other hand, to validate these results by comparison with photoelectron spectroscopy, optical absorption, and electroluminescence spectra. In particular, we will demonstrate that the different quantities E_{CT} , E_{DA} , and ΔE used in the literature are, within an experimental error of ± 0.1 eV, identical. For this purpose we will use the aromatic hydrocarbon diindenoperylene (DIP; $C_{32}H_{16}$) as the donor in combination with the fullerene C_{60} as the electron acceptor in planar heterojunction solar cells, where a broad range of experimental data has been gathered since this system was first introduced by Wagner *et al.*²⁵ Thereafter the developed analysis will be applied to a broader range of molecular

materials, including sexithiophene (6T), pentacene (Pen), and Cu phthalocyanine (CuPc) as donors together with C_{60} and DIP as acceptors, to obtain a more general picture of energy losses in molecular OSCs.

II. THEORETICAL CONCEPTS AND STATE OF THE ART

A. Dark-current activation energy

In a simplistic picture the electrical behavior of a solar cell can be described as a current source in parallel with a diode, representing the voltage response of the device in the dark. These two contributions are added and yield the current density vs voltage (j - V) characteristics:^{26,27}

$$j = j_{\text{dark}} - j_{\text{photo}} = j_0[\exp(qV/nk_B T) - 1] - j_{\text{SC}}, \quad (1)$$

where the expression for j_{dark} is also known as the Shockley equation. j_{SC} is the short-circuit current density of the solar cell under illumination and is assumed to be identical to the photocurrent j_{photo} . j_0 is the reverse saturation current density, q is the elementary charge, n is the diode ideality factor, k_B is Boltzmann's constant, and T is the absolute temperature.

The saturation current density j_0 , which sets the magnitude of the current in the absence of illumination, is also dependent on temperature according to²⁶

$$j_0 = j_{00} \exp(-\Delta E/nk_B T), \quad (2)$$

where j_{00} depends on the material system and is a measure for the coupling between the electronic states at the heterojunction. ΔE is the energy barrier at the interface, which has to be surmounted to generate a pair of oppositely charged carriers. In a classical p - n homojunction diode this would be the band gap of the semiconductor, while in a heterojunction device it denotes the band offset at the interface between the two different semiconductors.²⁸ Accordingly, in organic heterojunctions this quantity can be identified with the energy offset between the highest occupied molecular orbital (HOMO) of the donor, where holes flow, and the lowest unoccupied molecular orbital (LUMO) of the acceptor, which is the relevant transport level for electrons. In inorganic p - n junctions the ideality factor n , appearing in both equations, takes a value of 1 if direct recombination in the bulk dominates and 2 if trap-assisted recombination in the junction region is dominant.²⁶ Note that Potscavage *et al.* have introduced two different ideality factors in both expressions, however, at the same time noting that they are virtually identical in actual devices.²² According to the Shockley equation, an analysis of temperature-dependent j - V characteristics in the dark should thus allow for a determination of ΔE as the activation energy of j_0 . This analysis has been applied by various authors,^{21–24,29} and the outcome was compared to the energy-level difference E_{DA} at the D/A interface predicted from photoelectron spectroscopy or cyclic voltammetry. Note that some authors used a modification of Eq. (2) with an additional factor of 2 in the denominator^{21,23} to obtain a better agreement with expected values for E_{DA} ; however, as we will briefly address later on, this seems to be a peculiarity of the investigated systems and not generic.

If the cell is illuminated (typically under simulated sunlight conditions) without an external load, there will be no net current ($j_{\text{dark}} = j_{\text{photo}}$), and the voltage drop between both

contacts equals the open-circuit voltage V_{OC} , which is therefore given by

$$qV_{OC} = nk_B T \ln(j_{SC}/j_0 - 1). \quad (3)$$

Inserting Eq. (2) yields

$$qV_{OC} \approx \Delta E - nk_B T \ln(j_0/j_{SC}), \quad (4)$$

where the assumption that $\exp(qV_{OC}/nk_B T) \gg 1$ has been used. This identifies $\Delta E/q$ as the limit of V_{OC} for $T \rightarrow 0$.

B. The detailed balance approach

Following the approach taken by Shockley and Queisser, i.e., a detailed balance limit, the photocurrent of the cell j_{SC} and the dark saturation current j_0 can be expressed by integrals over the radiation absorbed and emitted by the cell, respectively. The absorbed light is given by the solar spectrum (at the surface of the earth) $\Phi_{sun}(E)$, and the emitted radiation is given by a blackbody spectrum at 300 K $\Phi_{cell}(E)$, both weighted by the absorption coefficient $\alpha(E)$ of the cell.⁴ This yields

$$j_{SC} = q \int_0^\infty \alpha(E) \Phi_{sun}(E) dE, \quad (5)$$

$$j_0 = q \int_0^\infty \alpha(E) \Phi_{cell}(E) dE. \quad (6)$$

Therein, the absorption spectrum $\alpha(E)$, originally resembling a step function with $\alpha(E) = 0$ for $E < E_g$ and $\alpha(E) = 1$ for $E > E_g$, needs to be modified to account for the presence of a CT state with energy E_{CT} at the D/A interface in organic solar cells. For example, a double-step function is commonly used [see Fig. 1(b)].^{11–13}

$$\alpha(E) = \begin{cases} 0, & E < E_{CT}, \\ \alpha_{CT}, & E_{CT} < E < E_g, \\ \alpha_0, & E > E_g, \end{cases} \quad (7)$$

where α_{CT} stands for the absorption strength of the CT state (relative to the fundamental absorption across the gap being set to unity). Following this approach, V_{OC} is readily calculated from Eq. (3) by the postulate that under open-circuit conditions all photogenerated carriers have to recombine so that no net current flows in the external circuit. Numerical calculations (for details see Ref. 13) predict an almost-linear increase in V_{OC} with decreasing temperature, leading to E_{CT}/q as the upper limit at $T = 0$.

Note that Eq. (4) also predicts a linear temperature dependence of V_{OC} with ΔE as the limiting value for $T \rightarrow 0$. However, in the following we will identify the extrapolated upper limit of V_{OC} with the energy E_{CT} of the interfacial CT state, where recombination of charge carriers occurs. We do not *a priori* set this value equal to ΔE , which is relevant for carrier generation across the heterojunction barrier in the reverse-bias direction.

In contrast to the idealized situation in the original SQ model, where all recombination was assumed to be radiative, the occurrence of predominantly nonradiative recombination in real cells can be taken into account by an enhanced recombination current $j_0^{(nonrad)} = k_n j_0^{(rad)}$.¹³ Rau *et al.* have shown that k_n is equal to the inverse of the electroluminescence quantum efficiency of a solar cell.³⁰ Performing this procedure

for varying temperature, it follows again that $qV_{OC}(T \rightarrow 0) = E_{CT}$; that is, the upper limit of the open-circuit voltage remains unchanged.¹³ Furthermore, knowing the relative strength of nonradiative recombination, it is now possible to express losses in V_{OC} at finite temperature as radiative and nonradiative ones:

$$V_{OC}(T) = E_{CT}/q - \Delta V_{OC}^{(rad)} - \Delta V_{OC}^{(nonrad)}. \quad (8)$$

Vandewal *et al.* have recently applied the detailed balance approach to polymer bulk-heterojunction solar cells.²⁰ They have been able to relate V_{OC} and its losses to spectroscopically accessible quantities at the donor-acceptor interface. These are, in particular, the energy of the CT state E_{CT} , the reorganization energy λ_0 in forming the CT state by light absorption, and a parameter f that is proportional to the number of CT states in the device as well as the electronic coupling between CT and the ground state:

$$V_{OC}(T) = E_{CT}/q + (k_B T/q) \ln \left(\frac{j_{SC} h^3 c^2}{2\pi f q (E_{CT} - \lambda_0)} \right) + (k_B T/q) \ln(\eta_{EQE,EL}), \quad (9)$$

where h and c are Planck's constant and the speed of light, respectively, and $\eta_{EQE,EL}$ denotes the electroluminescence quantum efficiency. Using experimental parameters from photocurrent and electroluminescence spectra, the authors have been able to achieve a consistent description of the temperature dependence of V_{OC} for a range of polymer-fullerene-blend solar cells. In particular, they could show that the open-circuit voltage measured at room temperature (under one sun illumination) is typically between 0.5 and 0.6 V less than the CT energy of the respective D/A interface.²⁰

In the following we will use DIP/C₆₀ as a prototypical molecular donor-acceptor system to establish a conclusive model for the origin of the open-circuit voltage from the offset energy at the D/A interface. By *a priori* independent consideration of the energy values obtained by photoemission spectroscopy (E_{DA}), dark current analysis (ΔE), and the detailed balance approach (E_{CT}) as well as optical spectroscopy (E_{CT}^{opt}), we show that these energies are related. We present various methods that consistently yield the effective photovoltaic energy gap at the heterojunction.

III. MATERIALS AND METHODS

Organic solar cells in planar heterojunction (PHJ) architecture [see Fig. 2(a)] were fabricated on commercially available indium tin oxide (ITO) coated glass substrates (purchased from Thin Film Devices, Inc., Anaheim, CA; sheet resistance $\approx 20 \Omega/\text{square}$). Prior to the evaporation of the active D/A heterojunction, an approximately 30-nm-thick layer of poly(3,4-ethylenedioxythiophene):poly(styrenesulfonate) (PEDOT:PSS; purchased from Heraeus Clevis GmbH, Leverkusen, Germany) was spin coated from an aqueous solution and annealed at 125 °C for 45 min on a hotplate under ambient conditions. Subsequently, the active donor and acceptor layers as well as a 5-nm layer of the exciton blocking material bathocuproine (BCP; purchased from Sigma Aldrich as sublimed grade and used without further purification) were deposited using the standard thermal evaporation procedure at base pressures of 10^{-6} – 10^{-7} mbar. Finally, 100 nm of

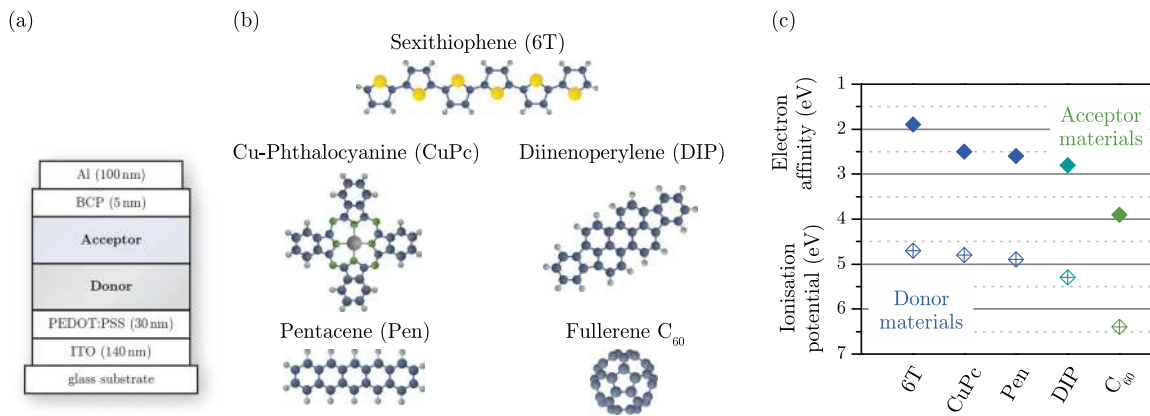


FIG. 2. (Color online) (a) Device stack of the investigated planar heterojunctions. (b) Chemical structures of the active organic semiconductors α -sexithiophene (6T), copper phthalocyanine (CuPc), diindenoperylene (DIP), pentacene (Pen), and Buckminster fullerene (C_{60}). (c) Energies of the HOMO (open symbols) and LUMO (solid symbols) levels for the investigated materials. The values are taken from the literature.^{14,31,32} The LUMO levels of CuPc and Pen are estimated by adding transport gaps of 2.3 eV (Ref. 33) and 2.2 eV (Ref. 34) to the respective HOMOs. The LUMO of 6T is estimated from the optical gap of 6T ($E_g = 2.4$ eV) and the relation between the optical gap and the transport gap, both found in Ref. 35.

aluminum were evaporated through a shadow mask as a top electrode, resulting in solar cells with an active area of $A = 4 \text{ mm}^2$. The active materials, diindenoperylene (DIP; purchased from S. Hirschmann, Universität Stuttgart, Stuttgart, Germany), α -sexithiophene (6T; purchased from Sigma Aldrich), and C_{60} (purchased from Creaphys, Dresden, Germany), were purified twice by gradient sublimation. Copper phthalocyanine (CuPc) was purchased from Sigma Aldrich as sublimed grade and was additionally purified once by gradient sublimation. Pentacene (Pen; purchased from Sigma Aldrich) was used without further purification. The chemical structures of the organic semiconductors as well as the literature values of their transport levels are depicted in Figs. 2(b) and 2(c). For the PHJ solar cells, the exact film thicknesses of the active layers are specified in the results section. Please note that growth conditions and film thicknesses have not been optimized for all devices. All evaporated organic layers have been grown while the substrate was kept at room temperature. Only in the case of DIP donor films in solar cells was an elevated substrate temperature of 100°C used to avoid injection barriers.³⁶

Current density vs voltage (j - V) characteristics of the solar cells were recorded using a source measure unit (Keithley 236 SMU) in the dark and under illumination. Unless otherwise stated, a white light-emitting diode (LED; Luxeon LXHL-NWE8) was used as the light source. Since the illumination conditions do not fulfill the AM 1.5 g standard, values for power-conversion efficiencies are not specified. However, representative data can be found in Ref. 25.

Temperature-dependent j - V characteristics were measured in an inert-gas atmosphere (300-mbar N_2) using a liquid-nitrogen-cooled cryostat (Cryovac) into which the samples were transferred without air exposure. Temperature was measured using a sensor mounted in close proximity to the sample. Temperature-dependent measurements of V_{OC} and j_{SC} have been recorded for different illumination intensities ranging from approximately 0.5 to 0.5×10^{-3} sun.

Photothermal deflection spectroscopy (PDS) was applied to neat and mixed films (ratio of approximately 1:1) of DIP and

C_{60} (co)evaporated on quartz glass substrates. Perfluorohexane was used as the deflection medium, which assures high transmittance in the energy region of 0.6–6 eV and large changes in index of refraction with small changes in temperature. Moreover, it dissolves neither DIP nor C_{60} .

Electroluminescence (EL) measurements were carried out on cells with a planar-mixed heterojunction (PM-HJ) with a coevaporated active layer of DIP: C_{60} and compared to single-layer devices with neat DIP or C_{60} . EL measurements were realized by using a liquid-nitrogen-cooled CCD camera (PyLoN:100BR, Princeton Instruments) with a spectral sensitivity in the wavelength range of approximately 300–1000 nm. All EL measurements were performed on encapsulated solar cells under a dc voltage drive from a Keithley source meter.

IV. EXPERIMENTAL RESULTS AND DISCUSSION

A. DIP/ C_{60} as a prototypical molecular donor-acceptor system

1. Temperature-dependent j - V characteristics in the dark: Analysis of saturation current and ideality factor

Earlier studies on the material combination DIP/ C_{60} have proven its suitability as a donor-acceptor pair in planar and mixed heterojunction solar cells.^{25,36} Concerning its application as a donor in organic solar cells, the main advantage of DIP can be found in its high ionization potential and the concomitant favorable energy-level alignment with the acceptor C_{60} , leading to high open-circuit voltages exceeding 0.9 V.²⁵ Moreover, favorable film morphology together with high crystalline order allows for improved transport properties of both excitons³⁷ and charges³⁸ in photovoltaic cells, resulting in high fill factors up to 75%.²⁵ To study its temperature-dependent behavior, dark j - V characteristics have been recorded in the temperature range from 350 to 130 K for a planar heterojunction cell consisting of ITO/PEDOT:PSS/DIP(50 nm)/ C_{60} (80 nm)/BCP(5 nm)/Al [see open symbols in Fig. 3(a)]. Based on the general diode equation (1), the exponential part of the forward-bias

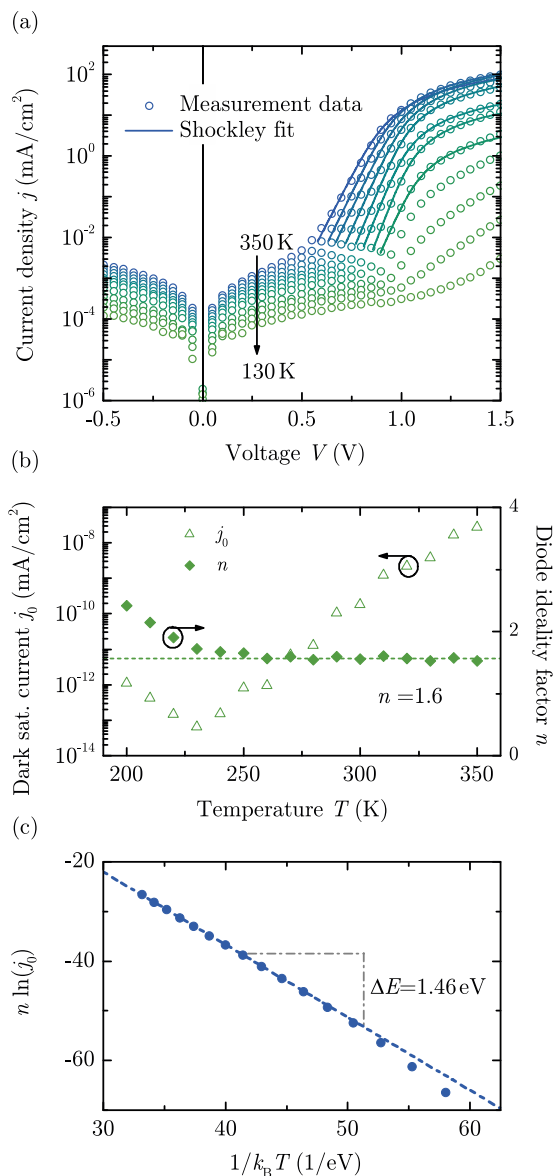


FIG. 3. (Color online) (a) Dark j - V characteristics of a planar heterojunction with structure ITO/PEDOT:PSS/DIP(50 nm)/C₆₀(80 nm)/BCP(5 nm)/Al recorded for several temperatures ranging from 130 to 350 K (open symbols). The solid lines are fitted curves in the forward-bias region using the Shockley equation. (b) Dark saturation current density j_0 and diode ideality factor n depending on temperature, analyzed with the Shockley equation. (c) Temperature dependence of the dark saturation current evaluated by studying $n \ln(j_0)$ plotted versus $1/k_B T$ in order to estimate the activation energy ΔE from Eq. (2). For clarity only a subset of temperatures is shown in (a).

characteristics can be described using a modified Shockley equation:

$$j(V) = j_0 \left\{ \exp \left[\frac{q[V - j(V)R_S A]}{nk_B T} \right] - 1 \right\}, \quad (10)$$

where an additional specific series resistance $R_S A$ accounts for the slower increase of the current within the drift-dominated regime at high voltages. While the Shockley equation was originally applied for analyzing and parameterizing the

j - V characteristics of inorganic p - n junction solar cells,²⁶ it was recently successfully adapted to organic solar cells,^{39,40} even in its modified variant.²³ By a fit with the modified Shockley equation (10) j_0 and n can be extracted from the data.

The obtained temperature dependencies of n and j_0 are depicted in Fig. 3(b). Within the wide range of $240 \text{ K} \leq T \leq 350 \text{ K}$, the ideality factor shows an almost-constant value of $n \approx 1.6$ (dashed line), while the dark saturation current obeys a temperature dependence of the form $\ln(j_0) \propto 1/k_B T$. The deviation from this behavior below 240 K may be attributed to nonideal shunt and series resistances that become relevant at low temperatures.⁴¹ According to Eq. (2), an activation energy of $\Delta E = (1.46 \pm 0.03) \text{ eV}$ can be determined from plotting $n \ln(j_0)$ versus $1/k_B T$ [see Fig. 3(c)].

2. Temperature-dependent j - V characteristics under illumination: Analysis and extrapolation of open-circuit voltage

The corresponding j - V curves of the same PHJ DIP/C₆₀ solar cell under white LED illumination corresponding to roughly 0.2 sun are shown in Fig. 4(a). The short-circuit current density stays almost constant in the high-temperature range; however, when cooling below $T = 180 \text{ K}$, it markedly decreases, and the curves are characterized by pronounced S shapes; that is, they show a decrease of the current close to V_{OC} and in the forward direction of the diode. Assuming a temperature-independent exciton generation rate, the reasons for the continuously decreasing j_{SC} can, on the one hand, be found in a reduction of charge-carrier extraction efficiency as a result of thermally activated charge-carrier mobility. On the other hand, a decline of exciton transport may contribute to the temperature dependence of j_{SC} .

Regarding the temperature dependence of the open-circuit voltage, an almost-linear increase in V_{OC} with decreasing T is observed in a wide temperature range [see Fig. 4(b)]. A similar relation has experimentally been found for a variety of material systems.^{21,42,43} The deviation from the linear behavior for temperatures $T < 150 \text{ K}$ can be attributed to the series resistance, which is found to strongly depend on temperature, as will be discussed below. Theoretical considerations predict this approximately linear temperature dependence of V_{OC} .^{13,44} This is based on the notion that the contribution of entropy to the electrochemical potential vanishes at 0 K (Ref. 45) and the energy gap that is responsible for charge recombination (E_{CT}) sets the upper limit for qV_{OC} . Following this approach, an extrapolation of the linear behavior towards $T = 0 \text{ K}$ yields a value of $E_{CT} = (1.42 \pm 0.02) \text{ eV}$.

Equation (4) shows that the dark saturation current has a decisive impact on the open-circuit voltage and presumably also on its temperature dependence. According to this relation, the open-circuit voltage can be calculated solely based on ΔE , n , and j_0 extracted from dark j - V characteristics, together with the short-circuit current density j_{SC} under given illumination conditions. As can be seen from Fig. 4(b) (blue diamonds), a very good agreement with the measured open-circuit voltage is obtained. The slight difference between the values determined for ΔE and E_{CT} becomes visible in the small deviation of reconstructed V_{OC} from measured V_{OC} data.

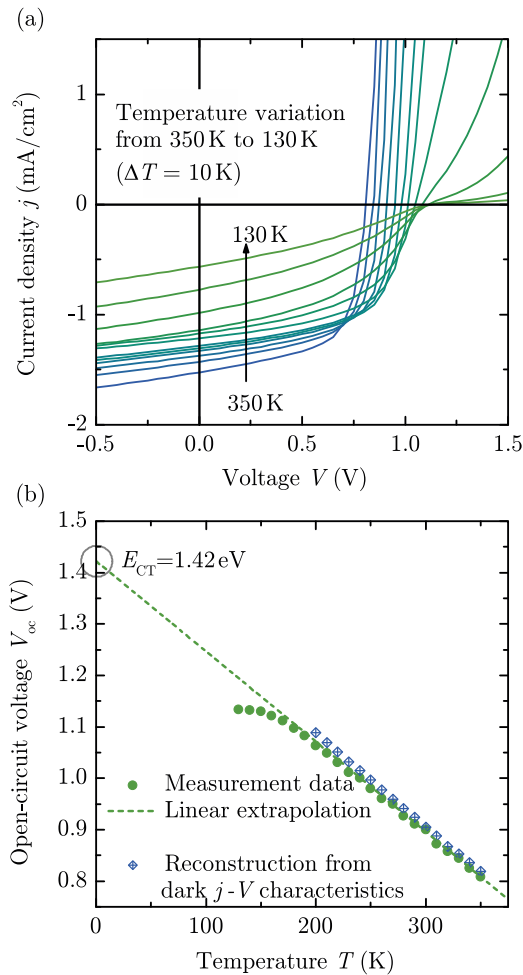


FIG. 4. (Color online) (a) j - V characteristics of a planar heterojunction solar cell under white LED illumination in the temperature range $130 \text{ K} \leq T \leq 350 \text{ K}$, $\Delta T = 10 \text{ K}$. The device has the following structure: ITO/PEDOT:PSS/DIP(50 nm)/C₆₀(80 nm)/BCP(5 nm)/Al. (b) Open-circuit voltage V_{OC} depending on temperature revealed from illuminated j - V data shown in (a) (green circles) with an extrapolation of the linear regime (dashed line) and reconstructed data according to Eq. (4) (diamond-shaped blue symbols). For clarity only a subset of temperatures is shown in (a).

Please note that the reconstruction has only been performed down to 200 K, where j_{SC} starts decreasing more substantially.

3. Intensity and temperature-dependent open-circuit voltage: Analysis of saturation current and ideality factor

Figure 5(a) shows the temperature dependence of the open-circuit voltage for different illumination intensities from 0.5 to 0.5×10^{-3} sun using a solar simulator and neutral density filters, measured on a DIP/C₆₀-based PHJ solar cell with a structure similar to that investigated before. The solid lines in Fig. 5(a) show the simulated temperature dependence for the different intensities. By solving Eqs. (6) and (3) and accounting for additional nonradiative recombination, the temperature dependence of V_{OC} can be calculated with a single value of $E_{CT} = 1.4 \text{ eV}$ for all intensities (for details refer to Ref. 13). However, the reduced photocurrent alone [as indicated by Eq. (3)] is insufficient to explain the steeper slope at low

intensities. Significantly larger values of the dark saturation current j_0 have to be used. To account for this circumstance, a biexponential function relating the nonradiative recombination current to light intensity can be used (not shown here).

Just like the simulated curves, the extrapolation of the linear regimes for the various illumination intensities all coincide in the same value of $E_{CT} = 1.4 \text{ eV}$ at $T = 0 \text{ K}$, as expected from Eq. (4). In fact the simulated lines in Fig. 5(a) can hardly be distinguished from a simple linear extrapolation. This demonstrates that the determination of the CT energy by linear extrapolation of V_{OC} to 0 K is actually valid. However, this does not *a priori* mean that all curves can be described with a single set of parameters ΔE , n , and j_{00} .

For a closer look into this issue, the measured $V_{OC}(T)$ data sets are plotted vs j_{SC} , which is approximately proportional to the illumination intensity if the slight decrease of j_{SC} with decreasing temperature [see Fig. 4(a)] is ignored. Figure 5(b) shows that, apart from the lowest two intensities, the proportionality of V_{OC} to $\log(j_{SC})$ according to Eq. (3) is well fulfilled. The fitted straight lines yield a temperature-dependent set of values for n and $\ln(j_0)$ that are plotted versus $1/k_B T$, as shown in Fig. 5(c). The perfect linear relation confirms the validity of this approach and yields an activation energy of $\Delta E = (1.43 \pm 0.03) \text{ eV}$, in excellent agreement with the analysis of the dark saturation current shown in Fig. 3. For the lowest two light intensities V_{OC} falls below the straight lines in Fig. 5(b). This shows that V_{OC} is disproportionately reduced at low intensity, implying severely increased nonradiative recombination losses. This is in accordance with what has been observed by simulation and can be seen as an indication that trap-assisted recombination becomes dominant at low light intensity, leading to enhanced nonradiative recombination.^{36,46}

4. Direct spectroscopic evidence for the charge-transfer state from absorption and electroluminescence measurements

While the electrical investigations discussed so far give more indirect information on the energetics at the D/A interface, direct spectroscopic evidence can be obtained from absorption and electroluminescence measurements. Note that for clear distinction the CT energy derived from spectroscopic measurements is denoted by E_{CT}^{opt} . As CT absorption features are typically characterized by low absorption coefficients, direct detection requires very sensitive spectroscopic techniques. Among the most common methods is PDS, which allows the determination of values for α_{CT} that are several orders of magnitude smaller compared to the bulk material absorption. PDS is based on local heating of the sample which is caused by even very weak absorption of light. Figure 6(a) shows PDS data of a coevaporated mixed layer of DIP and C₆₀ compared to the spectra of the single materials (0.5–1 μm on quartz). To obtain absolute values of the absorption coefficient, the PDS signal is calibrated using standard reflection/transmission measurements in the high-energy regime (open symbols), where both measurement ranges overlap.¹³

The DIP spectrum is characterized by a steep increase of absorption around $E = 2.0 \text{ eV}$, presumably originating from tail states of the DOS distribution below the optical gap at about 2.25 eV. The tiny peak at 1.1 eV is a multiphonon

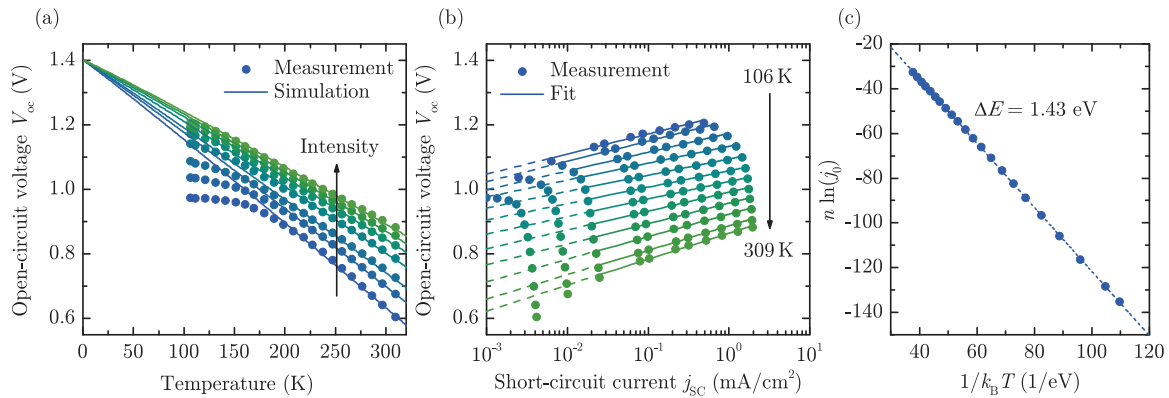


FIG. 5. (Color online) (a) Temperature dependence of V_{OC} measured for different illumination intensities. The solid lines show simulated behavior calculated with a modified SQ theory¹³ and $E_{CT} = 1.4$ eV. (b) V_{OC} as a function of j_{SC} measured for different temperatures and illumination intensities. The straight lines are fits according to Eq. (3). (c) Temperature dependence of the dark saturation current evaluated by studying $n \ln(j_0)$ plotted vs $1/k_B T$ as given by the fit results extracted from (b). For clarity only a subset of intensities/temperatures is shown in (a) and (b).

absorption due to vibrational stretching modes of C-H bonds, whereas the high-energy part (>2.25 eV) is due to $\pi-\pi^*$ transitions (for details see Ref. 47). In C_{60} the increase of absorption already starts around 1.75 eV, in accordance with an optical gap commonly taken to be 1.9 eV. The spectrum of the mixed film of DIP: C_{60} is characterized by signatures of both materials: While at low energies the multiphonon absorption of DIP can be detected, the high-energy region is governed by the characteristic absorption of the fullerene, which implies that the main contribution to light absorption in a heterojunction device with comparable layer thickness of both materials comes from the fullerene. However, what clearly distinguishes the simple superposition of the single spectra from the measured spectrum of the blend film is the extra, broad absorption in the energy range between 1.3 and 1.7 eV (hatched area), which cannot be found in any of the neat materials. This extra feature can be considered as a direct measure of the CT state as it presents the energetically lowest state to absorb light.

Figure 6(b) shows EL spectra of a PM-HJ DIP: C_{60} solar cell (blend layer of 50 nm sandwiched between neat layers of 5-nm DIP and 10-nm C_{60}) that was operated as an LED under varying forward bias, starting just above V_{OC} . As in PDS, a coevaporated blend was chosen to increase the D/A interface, allowing for a larger signal from weak CT emission. The spectra are compared to those of single-layer devices with DIP and C_{60} (active layer of 100 nm). The luminescence signals were corrected for the spectral sensitivity of the detector and normalized to their respective maximum values. The relative signal strengths are indicated by their magnification factors (given in brackets next to the respective curves). Due to the detection limit of the CCD camera the spectra are cut off at 1010 nm.

The EL signal of a pure C_{60} device is extremely low and only detectable when applying a forward bias of 6 V. One reason can be found in the symmetry-forbidden HOMO-LUMO transition of C_{60} :⁴⁸ Both the ground state S_0 and the first excited state S_1 exhibit even parity, which renders the transition between them dipole forbidden. However, vibronic coupling as well

as disorder in the film may lead to a certain degree of mixed components of even and odd states, which can partially allow the violation of parity conservation.⁴⁹ The results are comparable with literature values for the photoluminescence of C_{60} deposited on a quartz substrate, as found in Ref. 50: The authors attribute the main band at $E = 1.71$ eV to the Frenkel exciton within one C_{60} molecule followed by its vibronic progression on the low-energy side.

The DIP single-layer device shows a pronounced EL signal already at moderate forward bias. The main EL peak positions are similar to those found by photoluminescence measurements, characterized by the 0-0 transition at $E \approx 2.14$ eV, which is followed by pronounced vibronic progression with peak energies of 1.98 and 1.80 eV.^{47,51} The broad band around 1.7–1.6 eV is assigned to an aggregate emission due to the high crystallinity of DIP. Apart from the spectra of the single-layer devices, Fig. 6(b) comprises the EL spectra of a PM-HJ solar cell. As expected, luminescence bands of both materials are identifiable in the wavelength range between 600 and 900 nm as the two materials form a phase-separated blend when they are evaporated simultaneously.²⁵ However, EL offers the possibility to suppress the signals of the individual materials by reducing the applied voltage to a minimum, just above the open-circuit voltage. Thus, we observed a pronounced luminescence feature in the low-energy part of the spectrum (900–1000 nm), which becomes the dominant signature when the voltage is reduced to $V = 1.1$ V. As there is no indication for light emission from any of the neat materials in this near-infrared spectral range, we attribute it to the transition from the interfacial CT state to the ground state. We note that the apparent shift of the peak maximum for increasing bias voltage reminds us of a Stark effect. However, the complex morphology of the blend film⁵² makes a detailed interpretation very difficult and is beyond the scope of this paper.

In a recent publication, Tvingstedt *et al.* report electroluminescence from several types of polymer-fullerene BHJ solar cells biased in the forward direction, deducing E_{CT}^{opt} from the peak energy of the CT emission.⁵³ This approach was refined by Vandewal *et al.*:²⁰ Based on the framework of the

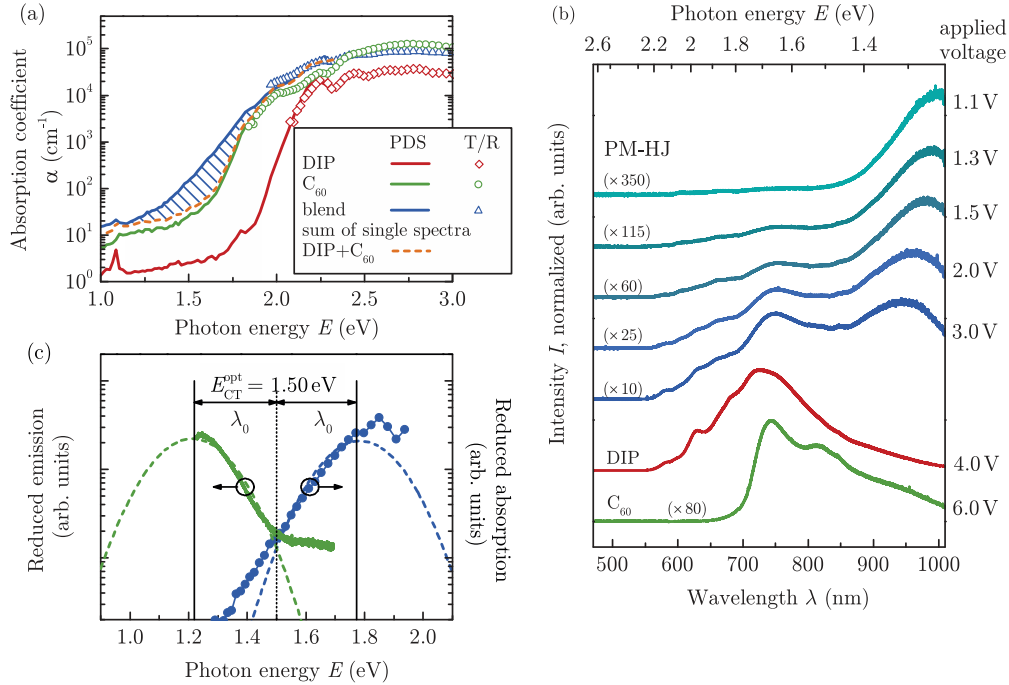


FIG. 6. (Color online) (a) Absorption spectrum for a blend of DIP:C₆₀ [lines: photothermal deflection spectroscopy (PDS);¹³ open symbols: transmission/reflection measurement (T/R)]. For comparison, the single spectra of both materials as well as their sum are plotted to demonstrate the additional feature caused by the charge-transfer-state absorption. (b) Electroluminescence spectra of PM-HJ solar cells based on DIP:C₆₀. The curves are obtained by applying a forward voltage of given magnitude. For better comparability all curves are normalized to their maximum values; however, the magnification factors are given in brackets to illustrate the strong variations in signal height. (c) Reduced absorption from PDS, blue circles, corresponding to the difference between the absorption spectrum of the blend and the sum of single absorption spectra) and emission (from EL, green straight line) for DIP:C₆₀ measured at room temperature. The dashed curves are fits according to Eq. (11), using the same values for E_{CT}^{opt} and λ_0 on both sides.

Marcus theory, the authors extracted the CT energy E_{CT}^{opt} by analyzing the absorption and emission spectra of a series of polymer-PCBM blends. They fitted the reduced absorption $[\sigma(E)E]$ and emission $[I_f(E)/E]$ spectra of the CT manifold, respectively, by

$$\left. \begin{array}{l} \sigma(E) E \\ I_f(E) E^{-1} \end{array} \right\} = \frac{f_{\sigma, I_f}}{\sqrt{4\pi\lambda_0 k_B T}} \exp\left(\frac{-(E_{CT}^{opt} \pm \lambda_0 - E)^2}{4\lambda_0 k_B T}\right). \quad (11)$$

As in Eq. (9), λ_0 denotes the reorganization energy (related to the Stokes shift). f_{σ} and f_{I_f} are proportional to the square of the electronic coupling matrix element. The plus sign is used for the reduced absorption, the minus sign is used for the reduced emission.

The reduced absorption and emission spectra for the DIP- and C₆₀-based devices are shown in Fig. 6(c). By fitting both spectra with Eq. (11), a reorganization energy of $\lambda_0 \approx 0.26$ eV as well as an energy of the CT state of $E_{CT}^{opt} \approx (1.50 \pm 0.06)$ eV can be deduced. The reorganization energy λ_0 corresponds to nonradiative energy losses as a result of relaxation, and its magnitude is found to be similar to values determined for a variety of polymer-fullerene blends.²⁰ Note, however, that a relatively large error of 0.06 eV has to be assumed for E_{CT}^{opt} since no clear absorption maximum is visible in the data, which reduces the reliability of the fit. Additional uncertainty is introduced by the fact that a mixed film has been used

instead of a bilayer in order to increase the signal strengths. Furthermore, the potential temperature dependence of E_{CT} has not been investigated. If present, these effects are expected to be within 0.1 eV.²⁰

5. Discussion

As a key outcome, the presented results convincingly show that different temperature-dependent electrical device characterizations and optical spectroscopies yield very similar values for the effective photovoltaic gap at the D/A interface in DIP/C₆₀ solar cells. The energy of the CT state $E_{CT}^{opt} = 1.5$ eV obtained from PDS and EL spectra can be identified with the value $E_{CT} = 1.42$ eV extracted from the extrapolation of V_{OC} towards 0 K. Both are in very good agreement with the activation energy $\Delta E = 1.46$ eV extracted from the temperature dependence of the dark saturation current. In addition, we want to note that an interfacial energy gap $E_{DA} = 1.40 \pm 0.15$ eV at the DIP/C₆₀ heterojunction has recently been determined by means of photoelectron spectroscopy by Wilke *et al.*¹⁴ Together with this work, the correspondence between E_{CT}^{opt} , E_{CT} , ΔE , and E_{DA} is thus clearly demonstrated for the material system DIP/C₆₀. The *a priori* different energies are basically the same within an error of 0.1 eV. This fact qualifies the extrapolation of V_{OC} to 0 K and the analysis of the thermally activated j_0 as suitable methods to identify the intermolecular energy gap at the D/A interface of an organic heterojunction. These methods are especially helpful since they allow the

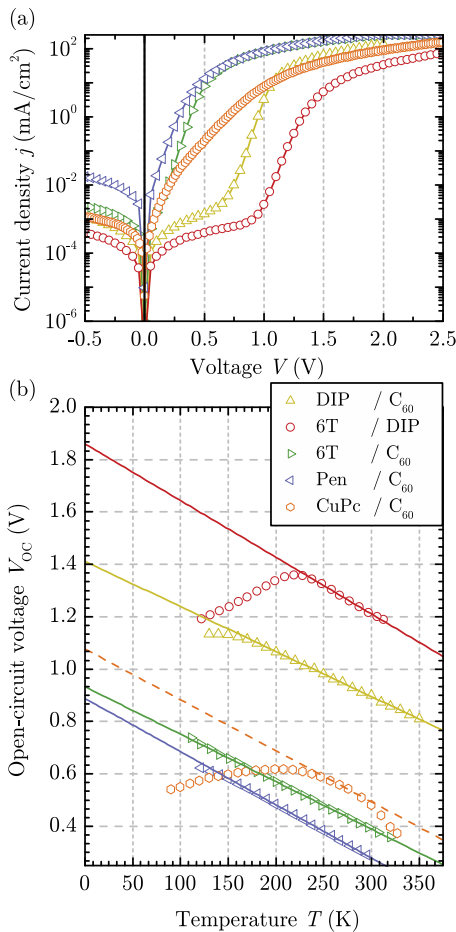


FIG. 7. (Color online) (a) Dark j - V characteristics of a series of PHJ solar cells employing different donor-acceptor combinations recorded at room temperature. The respective device structures and parameters are summarized in Table I. (b) Temperature dependence of V_{OC} of different PHJ solar cells. The extrapolation of the linear regime towards $T = 0$ K gives a measure of the energy of the charge-transfer state E_{CT} .

examination of the buried interface in device geometry. We would like to stress that even though the methods presented in Figs. 3 and 5 eventually rely on the analysis of j_0 , they have to be regarded as substantially different. In particular the cell is analyzed in the dark in the former case and under illumination in the latter one.

B. Application to other molecular D/A systems

1. Temperature-dependent j - V characteristics in the dark and under illumination

After the clear demonstration that temperature-dependent electrical device characteristics yield direct access to the effective photovoltaic energy gap at the D/A interface, which sets the upper limit for V_{OC} in organic heterojunction photovoltaic cells, we now apply this approach to a series of molecular D/A combinations. An overview of their current-voltage characteristics in the dark and at room temperature is given in Fig. 7(a). Characteristic device parameters extracted from the fits according to the modified Shockley equation, i.e., $R_S A$, j_0 , and n , are given in Table I. An analysis of

the temperature-dependent measurements was done for each material combination following the same procedure as exemplarily shown above for the DIP/ C_{60} PHJ cell. The activation energy ΔE extracted from the temperature dependence of j_0 according to Eq. (2) is also included in Table I.

The characteristic solar-cell parameters under illumination, such as V_{OC} , j_{SC} , and FF, were determined for each cell at room temperature at illumination intensities of about 0.2–0.4 sun (see Table I). Remarkably, there is a strong variation of V_{OC} between ~ 0.3 V for Pen/ C_{60} and 1.24 V for 6T/DIP, which cannot be caused by the slight variation of j_{SC} . As in the case of DIP/ C_{60} , temperature-dependent V_{OC} measurements were extrapolated linearly towards $T = 0$ K, yielding E_{CT} . Figure 7(b) depicts the measured data for the different material systems (symbols) recorded in a temperature range of 110 K $< T < 320$ K. The linear fits are included as straight lines, and the values for E_{CT} are summarized in Table I. In the case of CuPc/ C_{60} the selection of the fit data obviously has a strong impact on the value extracted for E_{CT} . However, the dashed line in Fig. 7(b) indicates a fit that is consistent with the literature from both direct spectroscopy⁵⁴ and extrapolation of $V_{OC}(T)$.¹⁵

The experimentally determined values for E_{CT} are compared to values of the intermolecular energy gap E_{DA} as found in the literature. The good correlation confirms the suitability of our approach. Furthermore, for most of the heterojunctions remarkably good agreement between ΔE (as extracted from the temperature behavior of j_0) and E_{CT} (as extracted from the linear extrapolation of V_{OC} versus T) is observed. The only exception is the material combination CuPc/ C_{60} , which shows a large discrepancy between both energies: The activation energy ΔE of the CuPc/ C_{60} system is almost twice as high as E_{CT} . We attribute this to a different recombination mechanism, as will be discussed below.

Note that while Pen/ C_{60} and 6T/ C_{60} show a linear behavior over the entire temperature range, the other material systems deviate from linearity below a certain temperature. Comparable effects are reported in the literature and have been explained by energetic disorder⁵⁵ and injection barriers.⁵⁶ We attribute the observed saturation or even decrease of V_{OC} to injection barriers and insufficient carrier mobility at low temperatures. This manifests in a strongly increased series resistance R_S visible in a reduced dark injection current [cf. Fig. 3(a)], which is accompanied by a distinct S kink of the light j - V curve around V_{OC} . Additionally, this leads to a strongly reduced fill factor. Finally, j_{SC} is decreased, which hints at severe recombination losses [cf. Fig. 4(a)]. As predicted by Eq. (3), V_{OC} depends on j_{SC} , representing the photocurrent. In the thermodynamic limit, which forms the basis for the linear extrapolation to $T = 0$ K, however, the photocurrent solely depends on the exciton generation rate and is thus a constant value. By replacing the ideal photocurrent with the measured short-circuit current, additional recombination losses caused by R_S are included. This directly yields the sublinear dependence of V_{OC} observed at low temperatures.

Overall, the extrapolation of the temperature-dependent V_{OC} seems to represent a reliable method to determine E_{CT} not only for DIP/ C_{60} but also for a whole range of material combinations. This indicates a more general suitability of this method.

TABLE I. Characteristic parameters for a series of PHJ photovoltaic devices with different donor-acceptor materials. All cells have a similar device structure of ITO/PEDOT(≈ 30 nm)/donor/acceptor/BCP(5 nm)/Al, with donor and acceptor layer thicknesses (in nm) as given below. Quantities extracted from illuminated j - V characteristics at room temperature are open-circuit voltage V_{OC} , short-circuit current density j_{sc} , and fill factor. Light illumination inside the cryostat was realized with a white LED with an estimated intensity I_L . E_{CT} denotes the energy of the charge-transfer state as extracted from a linear extrapolation of the temperature-dependent V_{OC} towards $T = 0$ K. The comparatively small differences in I_L [cf. Fig. 5(a)] are considered negligible for the values of E_{CT} . Quantities extracted from dark j - V characteristics at room temperature are specific series resistance $R_S A$, dark saturation current density j_0 , and ideality factor n . ΔE denotes the activation energy as deduced from the temperature dependence of the dark saturation current. With these parameters j_{00} was derived from Eq. (2). For comparison, the intermolecular energy gap $E_{DA} = |E_{HOMO}^{donor} - E_{LUMO}^{acceptor}|$ as taken from literature values is listed.

D/A	Illuminated j - V characteristics					Dark j - V characteristics					Literature
	V_{OC} (V)	j_{sc} (mA/cm ²)	FF (%)	I_L (mW/cm ²)	E_{CT} (eV)	$R_S A$ (Ω cm ²)	j_0 (mA/cm ²)	n	ΔE (eV)	j_{00} (mA/cm ²)	E_{DA} (eV)
DIP(50)/C ₆₀ (80)	0.91	-1.3	63.6	18	1.42	5	1.0×10^{-10}	1.6	1.46	2.13×10^5	1.4 (Ref. 14)
6T(60)/DIP(60)	1.24	-1.3	42.5	37	1.86	14	8.0×10^{-11}	2.3	2.03	5.37×10^4	1.8 (Ref. 14)
6T(40)/C ₆₀ (80)	0.41	-2.3	60.0	32	0.93	6	7.1×10^{-5}	1.7	0.94	1.38×10^5	0.95 (Ref. 14)
Pen(50)/C ₆₀ (80)	0.29	-3.1	48.3	36	0.89	5	6.0×10^{-3}	2.1	0.98	4.15×10^5	1.05 (Refs. 57 and 58)
CuPc(25)/C ₆₀ (50)	0.51	-1.9	34.0	28	1.08	8	2.0×10^{-3}	4.9	1.98	1.23×10^4	1.0 (Ref. 31)

2. Energy losses from the CT state to open-circuit voltage

Figure 8 summarizes the overall result of the present studies by showing the relation between V_{OC} at room temperature and E_{CT} of the different devices. Here, an excellent correlation is found for all of the material systems, clearly affirming that the open-circuit voltage is determined by the energy of the CT state, as predicted by Eq. (8). By applying a best fit to the data the linear dependence can be described by

$$V_{OC} = E_{CT}/q - 0.55 \text{ V.} \quad (12)$$

In this context, it has to be mentioned that the V_{OC} loss ΔV_{OC} derived from experimental dark j - V curves by $q \Delta V_{OC} = nk_B T \ln(j_{00}/j_{sc})$ [cf. Eq. (4)] is a result of both radiative and nonradiative recombination currents. Separation of the two components can be achieved by simulation if the absorption strength of the CT state α_{CT} is known. For a DIP/C₆₀ PHJ

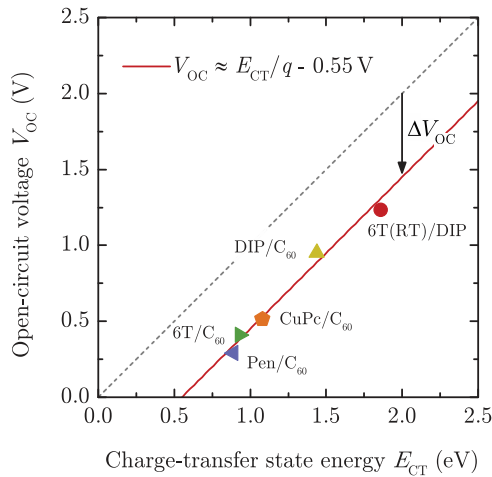


FIG. 8. (Color online) Relation between V_{OC} and E_{CT} (as extracted from the linear extrapolation of V_{OC} towards 0 K) for the different studied photovoltaic devices as listed in Table I. Their linear dependence for all material combinations is confirmed by a fit according to $V_{OC} = E_{CT}/q - 0.55$ V. The dashed line marks a one-to-one correspondence.

cell this has been done by Gruber *et al.*; at room temperature $\Delta V_{OC}^{(rad)} \approx 0.1$ V and $\Delta V_{OC}^{(nonrad)} \approx 0.4$ V were found.¹³

This indicates that in all of the investigated molecular D/A cells there is an energy loss of roughly half a volt between the energy of the CT state and V_{OC} at room temperature and that the dominant fraction is nonradiative losses.

3. Discussion

For most of the investigated heterojunctions we could observe remarkably good agreement between ΔE (as extracted from the temperature behavior of j_0) and E_{CT} (as extracted from the linear extrapolation of V_{OC} versus T). The direct correlation between ΔE and E_{CT} indicates with great confidence that the activation energy for the dark saturation current ΔE corresponds in a one-to-one manner to the CT energy. Both quantities represent the effective energy gap for an organic heterojunction solar cell. The only material system which shows deviations from this general rule is the CuPc/C₆₀ heterojunction. This deviation is already indicated by the shape of the dark j - V curves: As can be seen from Fig. 7(a), the current in the diffusion-dominated voltage regime is characterized by two differently steep slopes, a feature which becomes less pronounced at low temperatures. In literature, this double-exponential j - V characteristic observed for CuPc/C₆₀ has been addressed by the introduction of two independent diode ideality factors which consider the recombination of free acceptor electrons with trapped donor holes and vice versa.²⁴ Another peculiarity of CuPc/C₆₀ is that CuPc shows very fast intersystem crossing and that its T_1 triplet state lies energetically close to the donor-acceptor CT state.^{54,59,60} The exceptional importance of triplets at this interface might possibly lead to a more complex recombination mechanism compared to other systems. In any case, this observation suggests that CuPc/C₆₀ is not the material combination of choice to investigate a correlation between ΔE and E_{CT} . The data might lead to the assumption that ΔE corresponds to twice the effective band gap.^{21,23,61} This is, however, not true for a broad range of other material systems and needs to be carefully reviewed.

Concerning the band-gap-voltage offset, i.e., the difference between the effective gap at the heterojunction (e.g., E_{CT}/q) and the V_{OC} measured at room temperature, Fig. 8 clearly shows that about 0.5–0.6 V is lost for each investigated heterojunction. These losses are caused by recombination, both radiative and nonradiative. Both parts can be summarized in one experimentally accessible quantity $\Delta V_{OC} \propto \ln(j_{00}/j_{SC})$. j_{00} is generally considered to be a measure of the intermolecular coupling between the donor and acceptor molecules, where low j_{00} corresponds to weak coupling, a property which can be modified by varying the strength of the π interaction.^{23,29,61,62} Consequently, the experimentally apparent independence of the absolute V_{OC} loss on the material system becomes evident by considering these coupling constants (cf. Table I). Even though j_{00} varies by roughly one order of magnitude, the value enters logarithmically into ΔV_{OC} . This means that the electronic coupling is virtually identical for all investigated heterojunctions, independent of the effective photovoltaic energy gap. This results in very similar absolute losses. Even though a high short-circuit current will also favor a high V_{OC} , direct comparison between the values found for j_{00} and j_{SC} shows that the coupling constant dominates the impact on V_{OC} by far.

These findings emphasize that in order to achieve the highest possible V_{OC} two fundamental requirements have to be met by the material system: (1) The effective gap (be it ΔE or E_{CT}) has to be as large as possible in order to enable a large chemical potential of the charge-carrier pairs. (2) Recombination has to be suppressed to a minimum in order to maintain the chemical potential. The second requirement crucially depends on minimizing the coupling constant j_{00} while at the same time keeping j_{SC} high. Note that considering the mechanisms of charge generation in excitonic solar cells, weak electronic coupling at the D/A interface and large j_{SC} have to be regarded as potentially concurring aims. However, the predominant recombination losses are nonradiative and thus avoidable.^{13,43}

The fact that ΔV_{OC} is identical for all systems implies that the relative loss with respect to the particular effective photovoltaic gap is larger for material combinations with small intermolecular gaps. This is shown in Fig. 9. The measured relative V_{OC} loss with respect to E_{CT}/q derived from solar-cell characteristics under illumination (i.e., $1 - qV_{OC}/E_{CT}$) is plotted vs the predicted relative losses derived from dark j - V characteristics using Eq. (4) [i.e., $(nk_B T/\Delta E) \ln(j_{00}/j_{SC})$]. The almost-perfect one-to-one match of predicted and measured values shows that V_{OC} losses observed for the illuminated solar cell emerge already from the dark behavior of the diode, presumably from the strength of the donor-acceptor electronic coupling. This again underlines the fact that both techniques for the determination of the effective photovoltaic energy gap are fully equivalent.

In passing, we want to note that, interestingly, even inorganic single-junction solar cells exhibit band-gap-voltage offsets between 0.4 and 0.6 eV, more or less independent of the absolute magnitude of the semiconductor band gap.⁶³ This may be taken as a further confirmation that both organic and inorganic heterojunctions obey basically the same physics.

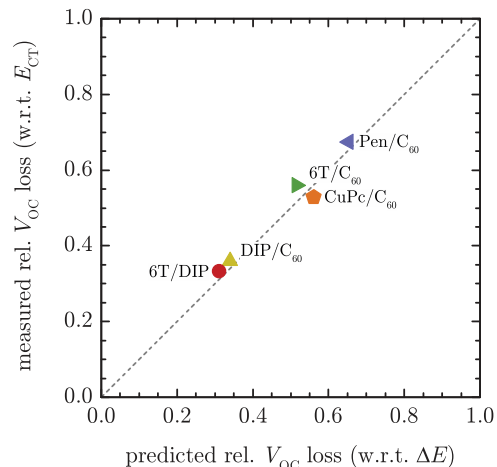


FIG. 9. (Color online) Relative V_{OC} loss of the different studied photovoltaic devices as listed in Table I, measured under illumination vs the predicted relative loss using parameters from dark characteristics and Eq. (4). Measured values are normalized by E_{CT} ; predicted values are normalized by ΔE (extracted from light and dark characteristics, respectively). The dashed line marks a one-to-one correspondence.

V. CONCLUSIONS

In this paper the effective photovoltaic gap acting as the upper limit of the open-circuit voltage of organic molecular heterojunction solar cells has been analyzed, and energy losses have been identified. For this purpose the quantities ΔE , E_{CT} , and E_{DA} typically used interchangeably in the literature have *a priori* been considered to be independent. In this scope, ΔE denotes the activation energy of the dark saturation current and can be regarded as the minimum energy required for charge generation. E_{CT} denotes the energy of the lowest radiative transition and is thus dominated by recombination. E_{DA} is the D/A intermolecular gap as determined by photoemission spectroscopy. These quantities were then shown to have equal values, indeed rendering them interchangeable.

In the first part of this work a thorough investigation of the CT energies of planar heterojunction DIP/ C_{60} cells was carried out. ΔE and E_{CT} were determined by two complementary methods each. While ΔE was extracted from the temperature dependence of the dark saturation current in both cases, j_0 itself was determined by distinctly different experiments. In the first case the dark injection current under forward bias was evaluated; in the second case the intensity dependence of V_{OC} was analyzed. Despite completely different solar-cell conditions, almost-identical results were extracted. E_{CT} was extracted from the temperature dependence of V_{OC} by extrapolation to 0 K. This value was compared to direct spectroscopic detection of the CT state by combined analysis of emission and absorption. Again, a close match of both values was obtained. By comparison of ΔE , E_{CT} , and E_{DA} from the literature (determined by photoemission and inverse photoemission spectroscopy)¹⁴ all three values could clearly be identified to be identical for the DIP/ C_{60} system. This shows that all four methods may reliably be applied to retrieve the effective energy gap at organic heterojunctions.

In the second part the temperature dependences of the dark injection current and of V_{OC} were analyzed for a series of material systems. Good agreement of ΔE and E_{CT} was found for all systems except CuPc/C₆₀, where recombination via trap states or triplets is assumed to play a major role.^{24,59} The general rule well established for bulk-heterojunction solar cells, that V_{OC} typically is 0.5–0.6 V below the effective photovoltaic gap, could clearly be verified for molecular planar heterojunction devices. Furthermore, we attribute this apparent generality to similar donor-acceptor electronic coupling strength, as indicated by relatively small variations of j_{00} . Additionally, we could show that the relative V_{OC} loss observed under illumination can already be predicted from the dark characteristics of the solar cell. The overall good agreement of values macroscopically observed and calculated from microscopic quantities shows that the temperature-dependent analysis of solar-cell characteristics

provides a simple and convenient tool to probe the effective energy gap at the donor-acceptor interface directly in device geometry.

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¹<http://www.nrel.gov/ncpv/>.

²<http://www.heliatek.com>.

³R. A. J. Jansen and J. Nelson, *Adv. Mater.* **25**, 1847 (2013).

⁴W. Shockley and H. J. Queisser, *J. Appl. Phys.* **32**, 510 (1961).

⁵B. A. Gregg and M. C. Hanna, *J. Appl. Phys.* **93**, 3605 (2003).

⁶D. Veldman, S. C. J. Meskers, and R. A. J. Janssen, *Adv. Funct. Mater.* **19**, 1939 (2009).

⁷T. M. Clarke and J. R. Durrant, *Chem. Rev.* **110**, 6736 (2010).

⁸C. Deibel, T. Strobel, and V. Dyakonov, *Adv. Mater.* **22**, 4097 (2010).

⁹K. Vandewal, K. Tvingstedt, and O. Inganäs, in *Quantum Efficiency in Complex Systems, Part II: From Molecular Aggregates to Organic Solar Cells*, edited by M. T. Uli Würfel and E. R. Weber, Semiconductors and Semimetals Vol. 85 (Elsevier, Amsterdam, 2011), pp. 261–295.

¹⁰T. Kirchartz, J. Mattheis, and U. Rau, *Phys. Rev. B* **78**, 235320 (2008).

¹¹N. C. Giebink, G. P. Wiederrecht, M. R. Wasielewski, and S. R. Forrest, *Phys. Rev. B* **83**, 195326 (2011).

¹²L. J. A. Koster, S. E. Shaheen, and J. C. Hummelen, *Adv. Energy Mater.* **2**, 1246 (2012).

¹³M. Gruber, J. Wagner, K. Klein, U. Hörmann, A. Opitz, M. Stutzmann, and W. Brütting, *Adv. Energy Mater.* **2**, 1100 (2012).

¹⁴A. Wilke, J. Endres, U. Hörmann, J. Niederhausen, R. Schlesinger, J. Frisch, P. Amsalem, J. Wagner, M. Gruber, A. Opitz, A. Vollmer, W. Brütting, A. Kahn, and N. Koch, *Appl. Phys. Lett.* **101**, 233301 (2012).

¹⁵J. Widmer, M. Tietze, K. Leo, and M. Riede, *Adv. Funct. Mater.* (to be published).

¹⁶M. C. Scharber, D. Mühlbacher, M. Koppe, P. Denk, C. Waldauf, A. J. Heeger, and C. J. Brabec, *Adv. Mater.* **18**, 789 (2006).

¹⁷R. R. Lunt, T. P. Osedach, P. R. Brown, J. A. Rowehl, and V. Bulović, *Adv. Mater.* **23**, 5712 (2011).

¹⁸K. Vandewal, A. Gadisa, W. D. Oosterbaan, S. Bertho, F. Banishoeib, I. Van Severen, L. Lutsen, T. J. Cleij, D. Vanderzande, and J. V. Manca, *Adv. Funct. Mater.* **18**, 2064 (2008).

¹⁹K. Vandewal, K. Tvingstedt, A. Gadisa, O. Inganäs, and J. V. Manca, *Nat. Mater.* **8**, 904 (2009).

²⁰K. Vandewal, K. Tvingstedt, A. Gadisa, O. Inganäs, and J. V. Manca, *Phys. Rev. B* **81**, 125204 (2010).

²¹B. P. Rand, D. P. Burk, and S. R. Forrest, *Phys. Rev. B* **75**, 115327 (2007).

²²W. J. Potscavage, A. Sharma, and B. Kippelen, *Acc. Chem. Res.* **42**, 1758 (2009).

²³M. D. Perez, C. Borek, S. R. Forrest, and M. E. Thompson, *J. Am. Chem. Soc.* **131**, 9281 (2009).

²⁴N. C. Giebink, G. P. Wiederrecht, M. R. Wasielewski, and S. R. Forrest, *Phys. Rev. B* **82**, 155305 (2010).

²⁵J. Wagner, M. Gruber, A. Hinderhofer, A. Wilke, B. Bröker, J. Frisch, P. Amsalem, A. Vollmer, A. Opitz, N. Koch, F. Schreiber, and W. Brütting, *Adv. Funct. Mater.* **20**, 4295 (2010).

²⁶S. Sze, *Physics of Semiconductor Devices* (Wiley, New York, 1982).

²⁷J. Nelson, *The Physics of Solar Cells* (Imperial College Press, London, 2003).

²⁸A. Fahrenbruch and J. Aranovich, in *Solar Energy Conversion*, edited by B. Seraphin, Topics in Applied Physics Vol. 31 (Springer, Berlin, 1979), pp. 257–326.

²⁹S. Yamamoto, A. Orimo, H. Ohkita, H. Benten, and S. Ito, *Adv. Energy Mater.* **2**, 229 (2012).

³⁰U. Rau, *Phys. Rev. B* **76**, 085303 (2007).

³¹A. Wilke, T. Mizokuro, R.-P. Blum, J. P. Rabe, and N. Koch, *IEEE J. Sel. Top. Quantum Electron.* **16**, 1732 (2010).

³²S. Duhm, I. Salzmann, G. Heimel, M. Oehzelt, A. Haase, R. L. Johnson, J. P. Rabe, and N. Koch, *Appl. Phys. Lett.* **94**, 033304 (2009).

³³I. Hill, A. Kahn, Z. Soos, and R. Pascal, *Chem. Phys. Lett.* **327**, 181 (2000).

³⁴A. Kahn, N. Koch, and W. Gao, *J. Polym. Sci., Part B* **41**, 2529 (2003).

³⁵P. I. Djurovich, E. I. Mayo, S. R. Forrest, and M. E. Thompson, *Org. Electron.* **10**, 515 (2009).

³⁶J. Wagner, M. Gruber, A. Wilke, Y. Tanaka, K. Topczak, A. Steindamm, U. Hörmann, A. Opitz, Y. Nakayama, H. Ishii,

- J. Pflaum, N. Koch, and W. Brütting, *J. Appl. Phys.* **111**, 054509 (2012).
- ³⁷D. Kurrle and J. Pflaum, *Appl. Phys. Lett.* **92**, 133306 (2008).
- ³⁸M. Horlet, M. Kraus, W. Brütting, and A. Opitz, *Appl. Phys. Lett.* **98**, 233304 (2011).
- ³⁹S. Yoo, B. Domercq, and B. Kippelen, *J. Appl. Phys.* **97**, 103706 (2005).
- ⁴⁰J. D. Servaites, S. Yeganeh, T. J. Marks, and M. A. Ratner, *Adv. Funct. Mater.* **20**, 97 (2010).
- ⁴¹G. A. H. Wetzelaer, L. J. A. Koster, and P. W. M. Blom, *Phys. Rev. Lett.* **107**, 066605 (2011).
- ⁴²V. D. Mihailetschi, P. W. M. Blom, J. C. Hummelen, and M. T. Rispens, *J. Appl. Phys.* **94**, 6849 (2003).
- ⁴³K. Vandewal, K. Tvingstedt, J. Manca, and O. Inganäs, *IEEE J. Sel. Top. Quantum Electron.* **16**, 1676 (2010).
- ⁴⁴M. A. Green, *Prog. Photovoltaics* **11**, 333 (2003).
- ⁴⁵P. Würfel, *Physics of Solar Cells: From Basic Principles to Advanced Concepts* (Wiley-VCH, Weinheim, 2009).
- ⁴⁶K. S. Nalwa, H. K. Kodali, B. Ganapathysubramanian, and S. Chaudhary, *Appl. Phys. Lett.* **99**, 263301 (2011).
- ⁴⁷D. Zhang, A. Horneber, J. Mihaljevic, U. Heinemeyer, K. Braun, F. Schreiber, R. Scholz, and A. Meixner, *J. Lumin.* **131**, 502 (2011).
- ⁴⁸N. Sariciftci, *Prog. Quantum Electron.* **19**, 131 (1995).
- ⁴⁹M. Fox, *Optische Eigenschaften von Festkörpern* (Oldenbourg Wissenschaftsverlag, Munich, Germany, 2012).
- ⁵⁰V. Capozzi, G. Casamassima, G. Lorusso, A. Minafra, R. Piccolo, T. Trovato, and A. Valentini, *Solid State Commun.* **98**, 853 (1996).
- ⁵¹M. Heilig, M. Domhan, and H. Port, *J. Lumin.* **110**, 290 (2004).
- ⁵²M. Gruber, M. Rawolle, J. Wagner, D. Magerl, U. Hörmann, J. Perlich, S. V. Roth, A. Opitz, F. Schreiber, P. Müller-Buschbaum, and W. Brütting, *Adv. Energy Mater.* **3**, 1075 (2013).
- ⁵³K. Tvingstedt, K. Vandewal, A. Gadisa, F. Zhang, J. Manca, and O. Inganäs, *J. Am. Chem. Soc.* **131**, 11819 (2009).
- ⁵⁴F. Piersimoni, D. Cheyngs, K. Vandewal, J. V. Manca, and B. P. Rand, *J. Phys. Chem. Lett.* **3**, 2064 (2012).
- ⁵⁵A. Manor and E. A. Katz, *Sol. Energy Mater. Sol. Cells* **97**, 132 (2012).
- ⁵⁶D. Rauh, A. Wagenpfahl, C. Deibel, and V. Dyakonov, *Appl. Phys. Lett.* **98**, 133301 (2011).
- ⁵⁷I. Salzmänn, S. Duhm, R. Opitz, R. L. Johnson, J. P. Rabe, and N. Koch, *J. Appl. Phys.* **104**, 114518 (2008).
- ⁵⁸J. C. Nolasco, A. Sánchez-Díaz, R. Cabré, J. Ferré-Borrull, L. F. Marsal, E. Palomares, and J. Pallarès, *Appl. Phys. Lett.* **97**, 013305 (2010).
- ⁵⁹G. J. Dutton and S. W. Robey, *J. Phys. Chem. C* **116**, 19173 (2012).
- ⁶⁰A. E. Jailaubekov, A. P. Willard, J. R. Tritsch, W.-L. Chan, N. Sai, R. Gearba, L. G. Kaake, K. J. Williams, K. Leung, P. J. Rossky, and X.-Y. Zhu, *Nat. Mater.* **12**, 66 (2013).
- ⁶¹P. Erwin and M. E. Thompson, *Appl. Phys. Lett.* **98**, 223305 (2011).
- ⁶²C. W. Schlenker and M. E. Thompson, *Chem. Commun.* **47**, 3702 (2011).
- ⁶³R. R. King, D. Bhusari, A. Boca, D. Larrabee, X.-Q. Liu, C. M. Fetzer, D. C. Law, and N. H. Karam, *Prog. Photovoltaics* **19**, 797 (2011).