

## RADIOCARBON DATING OF LIME FRACTIONS AND ORGANIC MATERIAL FROM BUILDINGS

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**ABSTRACT.** We have dated carbonate fractions and organic material from different types of mortar from two sites in Belgium. We demonstrate the difficulties in obtaining good dates from carbonate samples. We also discuss the need for new types of dating material when the mortar comes from contaminated and disturbed sites, where even charcoal can yield aberrant results.

### INTRODUCTION

In previous papers we examined the absorption mechanism of CO<sub>2</sub> on lime mortar (Pachiaudi *et al.* 1986; Van Strydonck, Dupas & Keppens 1989) and its effect on the radiocarbon age and isotope fractionation of different mortar fractions. We have also studied the influence of contamination by fossil carbonate on different fractions (Van Strydonck *et al.* 1986). In this paper, we investigate different types of samples and compare their (apparent) radiocarbon ages. Where possible, we compared carbonate dates with dates on organic material. We chose locations that represent the worst possible conditions for mortar samples: 1) underground ruins, subject to severe perturbation (construction of new walls over older ones, digging of graves); 2) a building facade subject to temperature changes, air pollution and restoration.

### ANALYTICAL METHODS

Charcoal samples were picked from the carbonate matrix and pretreated by the acid-base-acid (ABA) method before combustion. Mortar samples were ground, passed through a 250- $\mu$ m sieve and dried. A small amount of sample was taken to measure its CO<sub>2</sub> content. The rest was divided into two parts. From each part, the CO<sub>2</sub> was released by a standard 1.15 N HCl solution. The reagent was added slowly to the mortar powder, held in suspension in CO<sub>2</sub>-free water while constantly stirred. From one part, the CO<sub>2</sub> was released completely (fraction = 100%). The amount of acid added to the second part was just enough to release 10% of the total CO<sub>2</sub> (fraction = 10%) (Van Strydonck *et al.* 1986). The one exception was AK-11(ML), where we took the first 22 and 57%. The amount of CO<sub>2</sub> was measured by routine chemical analyses (Dupas 1981). If we had enough material, we measured the samples by gas counting. Small samples were AMS dated at Utrecht or Oxford.

### SAMPLE DESCRIPTION

#### Onze-Lieve-Vrouw Cathedral, Antwerpen, Belgium

Mortar and charcoal samples were collected during excavation of the foundations of the Romanesque church (50°13'16"N, 4°23'60"E), demolished when a Gothic cathedral was built over it. The context of this excavation presents a complex situation. At the same site are the remains of the foundations of several churches built in different phases. These construction activities, in addition to many grave trenches, contribute to the disturbance of this subterranean site. Figure 1 shows the locations of the samples relative to the floor plan of the cathedral.

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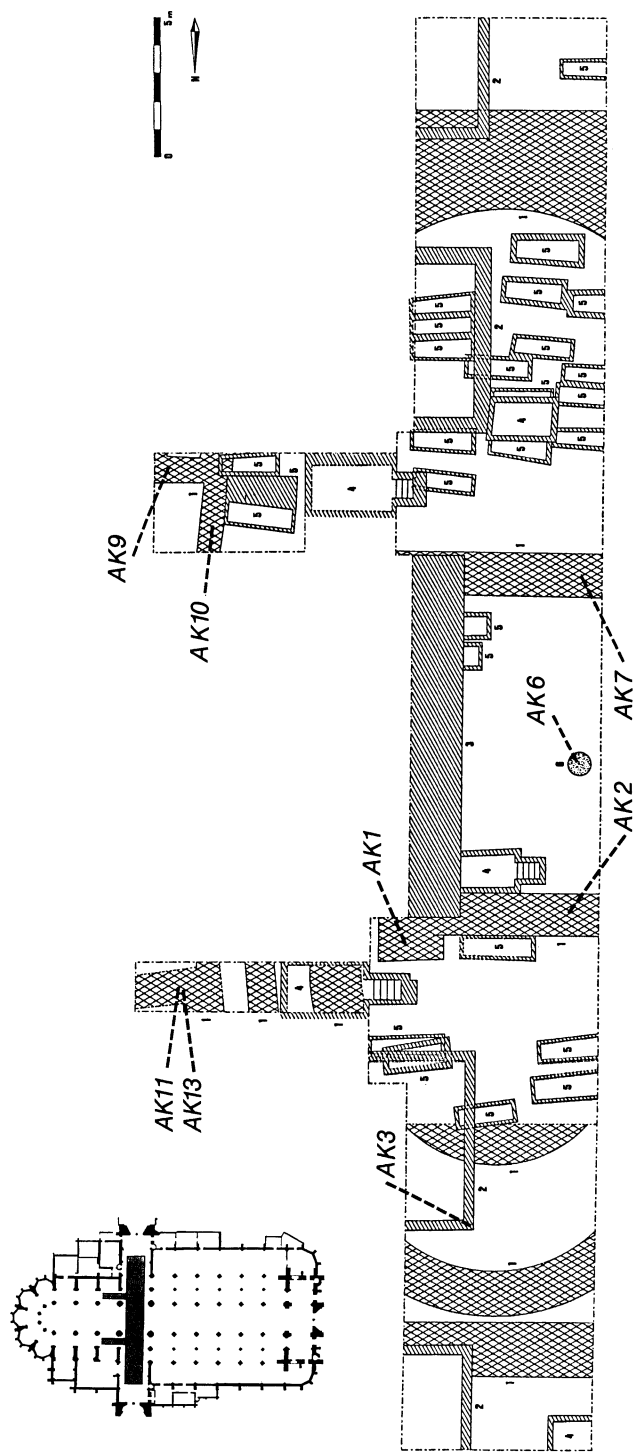


Fig. 1. Plan of the excavation of Onze-Lieve-Vrouw-Cathedral. Samples AK-2, -6 and -7 belong to the same construction. A written document describing the church in AD 1119 serves as a *terminus ante quem* date for this construction phase. Samples AK-9, -10, -11 and -13 date between the beginning of the 12th century and 1352, based on archaeological estimates.

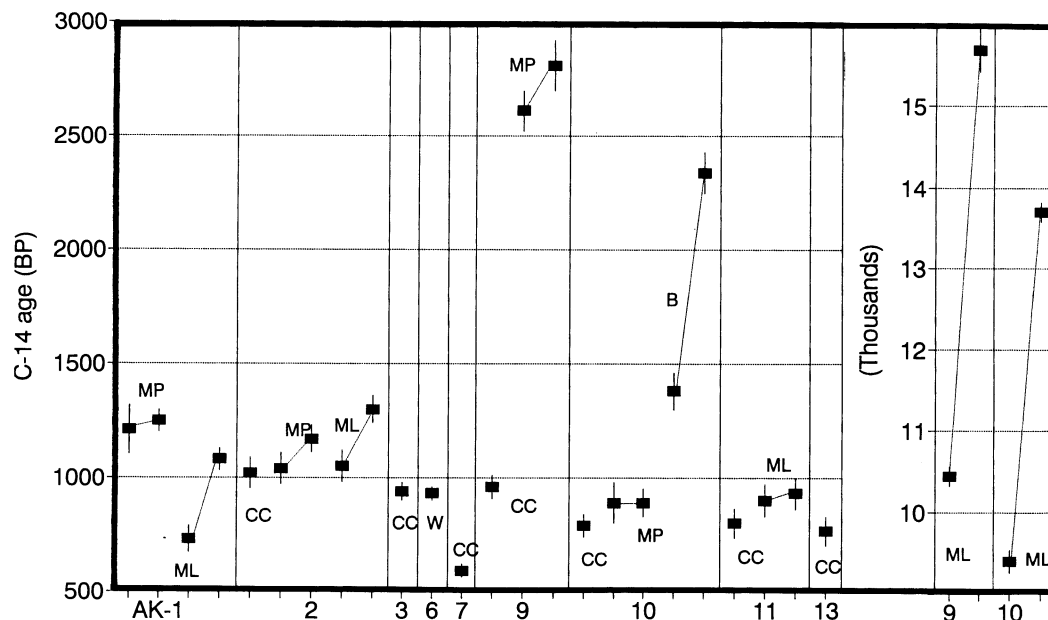


Fig. 2. <sup>14</sup>C results of mortar samples from Onze-Lieve-Vrouw Cathedral (uncalibrated ± 1 σ). CC-charcoal; MP-mortar powder; ML-mortar lumps; W-wood; B-brown sand.

Figure 2 plots the dates of the samples.

*Sample AK-1.* A very sandy beige mortar powder (MP) containing 2.33% CO<sub>2</sub>. In some areas, the mortar forms a conglomerate. Small white nodules (ML) are present in the powder with a maximum diameter of 4 mm containing 27.8% CO<sub>2</sub>.

*Sample AK-2.* From a rather hard brown-gray mortar layer (MP) containing 9.99% CO<sub>2</sub>, with very hard white nodules (ML), with varying diameters from about 10 mm to 5 cm, containing 17.90% CO<sub>2</sub>. Shells and some charcoal fragments (CC) (<0.5 g) were also visible in this layer.

*Sample AK-3.* From a charcoal (CC) layer mixed with sand under tiled pavement (>5 g).

*Sample AK-6.* Twigs (W) that held a wooden well together.

*Sample AK-7.* 1.8 g of small charcoal fragments (CC) from a wall.

*Sample AK-9.* Very powdery mortar (MP) containing 12.75% CO<sub>2</sub> with small white nodules (ML) containing 41.90% CO<sub>2</sub>. This high content of CO<sub>2</sub> means that the nodules (ML) are almost pure CaCO<sub>3</sub>. We collected a total amount of 0.27 g of charcoal (CC) from the lumps.

*Sample AK-10.* Mortar containing several fragments, 1–2.5 cm long, of a rather large burned wooden branch (diameter 2 cm) (CC). The sample is very hard gray-white mortar (MP) containing 20.25% CO<sub>2</sub> with white inclusions (ML) of 100% CaCO<sub>3</sub>. Brown sand containing 11.4% CO<sub>2</sub> was collected with the mortar but dated separately (B) and contained large and small mortar nodules.

*Sample AK-11.* White mortar and charcoal from a layer between the top of the foundation wall and the pavement. CaCO<sub>3</sub> content of the sample (98%) indicated complete carbonation of pure lime. We collected about 3 g of charcoal (CC).

*Sample AK-13.* Rather large pieces of charcoal, as in AK-10, collected from the same wall as AK-11.

*Sample W.* 5.97 mg of candle wax.

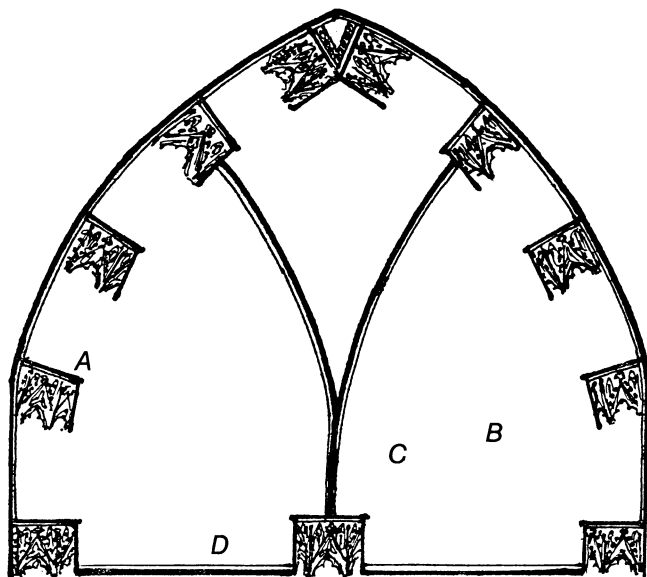


Fig. 3. Plan of the pediment of the portico of the Portail du Bethléem erected in the beginning of the 14th century. A–D indicate sample locations.

#### Portail du Bethléem, Huy-Liège, Belgium

We collected mortar samples that served to connect the statues to the pediment of the portico of the Portail du Bethléem at the Cathedral of Huy (50°31'N, 05°14'E) (Fig. 3). No surface mortar was taken because of the risks of collecting samples contaminated by air pollution.

*Samples A, B, C.* Original mortar from three locations on the pediment. Two samples (a & b, Fig. 4) were collected from location B. The CO<sub>2</sub> content of the carbonate samples varied from 16.51 to 20.89%.

*Sample D.* Mortar from an ancient restoration with a CO<sub>2</sub> content of 18.86%.

*Sample H.* Horse hair used to temper mortar sample D.

#### RESULTS

Table 1 and Figures 2 and 4 show the radiocarbon results and isotopic fractionation of each sample and each fraction.

#### DISCUSSION

##### Onze-Lieve-Vrouw Cathedral, Antwerpen

In general, the  $\delta^{13}\text{C}$  values confirm the results of our previous study (Van Strydonck *et al.* 1989). The first fraction is more depleted than the total sample and the heavily contaminated samples have values close to those expected for fossil rock carbonate ( $\sim 0\text{‰}$ ). Mortar powder tends to yield better  $^{14}\text{C}$  results than mortar nodules (Folk & Valastro 1976, 1979) but the correlation is too vague to be conclusive. In any case, it is no criterion for sample selection.

TABLE 1. Radiocarbon Measurements of Lime Fractions and Organic Materials

| Sample no.    | Lab. no. | Fraction (%) | <sup>14</sup> C age (yr BP) | δ <sup>13</sup> C (‰) |
|---------------|----------|--------------|-----------------------------|-----------------------|
| AK-1(MP-10)   | UtC-946  | 10           | 1210 ± 110                  | -20.3                 |
| AK-1(MP-100)  | UtC-945  | 100          | 1250 ± 50                   | -18.5                 |
| AK-1(ML-10)   | UtC-944  | 10           | 730 ± 60                    | -18.9                 |
| AK-1(ML-100)  | UtC-943  | 100          | 1080 ± 50                   | -17.2                 |
| AK-2(CC)      | UtC-998  |              | 1020 ± 70                   | -25.3                 |
| AK-2(MP-10)   | UtC-983  | 10           | 1040 ± 70                   | -16.5                 |
| AK-2(MP-100)  | UtC-984  | 100          | 1170 ± 60                   | -15.7                 |
| AK-2(ML-10)   | UtC-985  | 10           | 1050 ± 70                   | -13.6                 |
| AK-2(ML-100)  | UtC-986  | 100          | 1300 ± 60                   | -12.1                 |
| AK-3(CC)      | IRPA-948 |              | 940 ± 40                    | -27.0                 |
| AK-6(W)       | IRPA-967 |              | 930 ± 30                    | -25.3                 |
| AK-7(CC)      | UtC-1046 |              | 590 ± 30                    | -26.3                 |
| AK-9(CC)      | UtC-979  |              | 960 ± 50                    | -25.5                 |
| AK-9(MP-10)   | UtC-983  | 10           | 2610 ± 90                   | -14.1                 |
| AK-9(MP-100)  | UtC-984  | 100          | 2810 ± 110                  | -13.9                 |
| AK-9(ML-10)   | UtC-985  | 10           | 10,440 ± 120                | - 0.3                 |
| AK-9(ML-100)  | UtC-986  | 100          | 15,680 ± 260                | - 0.5                 |
| AK-10(CC)     | IRPA-972 |              | 790 ± 50                    | -26.5                 |
| AK-10(MP-10)  | UtC-948  | 10           | 890 ± 90                    | -15.5                 |
| AK-10(MP-100) | UtC-946  | 100          | 890 ± 60                    | -14.1                 |
| AK-10(ML-10)  | UtC-950  | 10           | 9390 ± 140                  | - 8.5                 |
| AK-10(ML-100) | UtC-949  | 100          | 13,700 ± 120                | - 2.7                 |
| AK-10(B-10)   | UtC-952  | 10           | 1380 ± 80                   | -16.8                 |
| AK-10(B-100)  | UtC-951  | 100          | 2340 ± 90                   | -13.8                 |
| AK-11(CC)     | IRPA-855 |              | 800 ± 65                    | -26.3                 |
| AK-11(ML-22)  | OxA-1707 | 22           | 900 ± 70                    | -16.6                 |
| AK-11(ML-57)  | OxA-1708 | 57           | 930 ± 70                    | -11.6                 |
| AK-13(CC)     | IRPA-856 |              | 765 ± 65                    | -27.4                 |
| W             | UtC-1167 |              | 480 ± 60                    | -28.1                 |
| A-10          | UtC-992  | 10           | 680 ± 80                    | -15.1                 |
| A-100         | UtC-987  | 100          | 840 ± 90                    | -11.9                 |
| B-10a         | UtC-993  | 10           | 790 ± 90                    | -12.4                 |
| B-100a        | UtC-988  | 100          | 890 ± 60                    | -11.6                 |
| B-10b         | OxA-1709 | 10           | 690 ± 65                    | -11.6                 |
| B-100b        | OxA-1710 | 100          | 620 ± 65                    | - 9.9                 |
| C-10          | UtC-994  | 10           | 800 ± 80                    | -12.5                 |
| C-100         | UtC-989  | 100          | 850 ± 80                    | -11.8                 |
| D-10          | UtC-995  | 10           | 550 ± 70                    | -12.5                 |
| D-100         | UtC-990  | 100          | 680 ± 60                    | -13.0                 |
| H             | UtC-991  |              | 410 ± 90                    | -23.1                 |

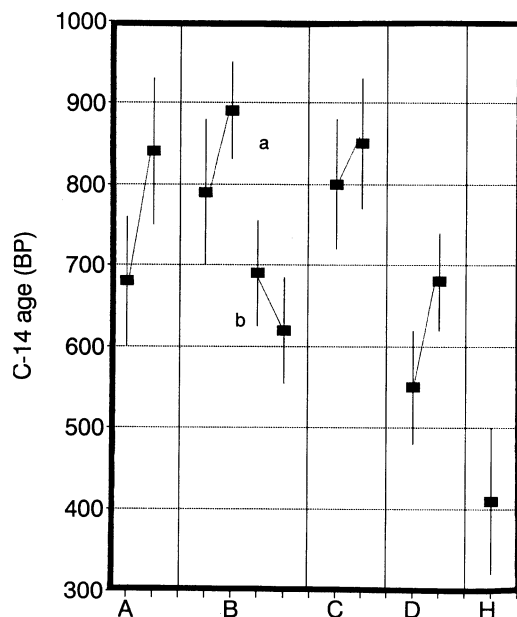


Fig. 4.  $^{14}\text{C}$  results of mortar samples from Portail du Bethléem (uncalibrated  $\pm 1 \sigma$ )

The sample ages of charcoal, mortar nodules and powder agree very well for AK-2. It was the only sample that had the same characteristics (solidity, homogeneity, color, *etc.*) as the good-quality mortars found in the standing walls. The results show very clearly that fossil carbonate is the major cause of the deviant results.

The charcoal in the samples showed other dating problems. Based on archaeological estimates, samples AK-2, -6 and -7 should come from the same construction phase, and samples AK-9, -10, -11 and -13 from another. Thus, AK-7 appears to be too young and AK-9 too old. Presumably, these are all sample errors, which means that the samples of charcoal and mortar lumps are reworked or from a younger inclusion. To overcome this general problem of dating a building, we asked the archaeologists and restorers to find samples with a closer relationship between the material and the historical event to be dated. A first and very important measurement was executed on a drop of candle wax (W), which was stuck between a wall painting and its painted-over surface. The result, cal AD 1396-1472 ( $1 \sigma$ ), agrees with stylistic and stratigraphic estimates (Van Strydonck 1990).

#### Portail du Bethléem, Huy-Liège, Belgium

The two fractions of the original mortar samples do not show the same important age differences as in the previous case. The mean age of samples A-10, B-10a, B-10b and C-10,  $740 \pm 63$  BP, cal AD 1215-1297 ( $1 \sigma$ ) is comparable, but older than the historic date (beginning of the 14th century). The results show very clearly the lack of accuracy in mortar dating, even when only a small amount of fossil carbonate is present. We also found a different type of sample, horse hair, which was used to temper the mortar.

#### CONCLUSION

The organic material in this study yielded better results than those of Tubbs and Kinder (1990), who obtained very old ages on organic material in lime mortar, because we tried to collect only

samples that were expected to be contemporaneous with an archaeological event. At first, we searched only for charcoal, but later tried to find material with a more specific origin.

The results of the carbonate samples show a large scatter not only among samples, but also within one sample. Good results are possible only if the mortar is of high quality, which is not often the case in building foundations.

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