REVIEW

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Recent advances in lignin valorization with bacterial cultures: microorganisms, metabolic pathways, and bio-products



Zhaoxian Xu¹, Peng Lei², Rui Zhai¹, Zhiqiang Wen¹ and Mingjie Jin^{1*}

Abstract

Lignin is the most abundant aromatic substrate on Earth and its valorization technologies are still under developed. Depolymerization and fragmentation are the predominant preparatory strategies for valorization of lignin to chemicals and fuels. However, due to the structural heterogeneity of lignin, depolymerization and fragmentation typically result in diverse product species, which require extensive separation and purification procedures to obtain target products. For lignin valorization, bacterial-based systems have attracted increasing attention because of their diverse metabolisms, which can be used to funnel multiple lignin-based compounds into specific target products. Here, recent advances in lignin valorization using bacteria are critically reviewed, including lignin-degrading bacteria that are able to degrade lignin and use lignin-associated aromatics, various associated metabolic pathways, and application of bacterial cultures for lignin valorization. This review will provide insight into the recent breakthroughs and future trends of lignin valorization based on bacterial systems.

Keywords: Biorefinery, Lignin valorization, Aromatic compounds, Metabolism of lignin-based aromatics, Biodegradation, Lignin-degrading bacteria, Lipid production, PHA production, *cis*, *cis*-muconic acid production

Background

Lignocellulose as the largest sustainable reservoir of organic material could be used to substitute for petroleum-based fuels and chemicals. Among the major components of lignocellulose, cellulose and hemicellulose have been converted to various chemicals and biofuels efficiently through biochemical route [1]. However, intense efforts are still needed to develop technologies to valorize lignin. Lignin, as a three-dimensional amorphous polymer, is composed of three different phenylpropane units: guaiacyl alcohol (G-type unit), *p*-coumaryl alcohol (H-type unit) and syringyl alcohol (S-type unit), which are linked mainly by aryl ether (β -O-4), phenylcoumaran (β -5), resinol (β - β), biphenyl ether (5-O-4), and dibenzodioxocin (5-5/ β -O-4) [2, 3] (Fig. 1). Although lignin

*Correspondence: jinmingjie@njust.edu.cn

¹ School of Environmental and Biological Engineering, Nanjing University of Science and Technology, Nanjing 210094, China

Full list of author information is available at the end of the article



accounts for approximate 15%–40% of lignocellulose, it is the most underutilized fraction of lignocellulose [4, 5].

As mentioned by Nguyen et al. the second-generation biofuels began to be commercialized in 2015 and 67 facilities were operated throughout the world in 2017, with more than 1/3 of these operating at commercial scale [6]. In September 2017, fifteen ministries in China jointly issued an announcement to promote ethanol production for gasoline blending. The announcement stated that the large-scale production of cellulosic ethanol should be achieved in 2025. This suggests the continued growth in global cellulosic ethanol production. In cellulosic ethanol production process, hexose and pentose are fermented to ethanol, leaving most lignin in the solid residue. When 1.0 L cellulosic ethanol is produced, about 1.0 kg lignin will be generated as byproducts. Moreover, the global pulp/paper industry would also generate about 50 million tons of lignin annually. Although lignin is generated in large amount, valorization of lignin is still a great challenge due to its inherent structural heterogeneity [7, 8]. Currently, most lignin is mainly burned for energy supply

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or discarded into the environment. The life cycle assessment model indicated that valorizing partial lignin to target chemicals may be more environmentally beneficial than burning for energy supply only [9]. To make use of lignin more effectively, lignin valorization technologies are urgently desired.

With the aforementioned information, it is of great interest and challenge to realize the valorization of lignin. Although the structure of lignin is more complex compared with that of cellulose and hemicellulose, the high carbon/oxygen ratio and the abundant reserves of aromatic skeletal in lignin render it a promising feedstock for chemicals and fuels [10]. The predominant strategies for valorization of lignin to chemicals and fuels are depolymerization and fragmentation (ring scission), including reduction to modify lignin, supercritical fluids to modify lignin, ionic liquids to modify lignin, and fractionation by ultrafiltration and selective precipitation [10–13]. However, the structural heterogeneity of lignin generally results in diverse product species, which requires extensive separation and purification procedures [13–15].

Biological treatment is another choice for lignin valorization. In previous studies, the main biological treatment studies focused on wood-rotting basidiomycetes. Even though the investigation of lignin biodegradation by fungi has been carried out for decades and several progresses have been made, there are few commercial processes using fungi for lignin valorization.

Recently, bacterial systems have attracted an increasing attention on lignin valorization because of their inherent "biological funneling" processes, which can funnel multiple aromatic streams into a uniform compound [16]. Thus, the challenges associated with lignin heterogeneity can potentially be overcome. To summarize the current breakthroughs and discuss future trends of lignin valorization with bacterial systems, here the recent advances in lignin valorization with bacterial systems were reviewed and discussed, including the lignindegrading bacteria and their screening methods, the lignin degradation pathways in bacteria, and the related bio-products produced from lignin components.

The lignin-degrading bacteria and their screening methods

Discovery of bacteria with strong lignin degradation capability and characterization of related enzymes have significant benefits for lignin valorization. Sample source is a key factor for discovering lignin-degrading bacteria with excellent performance. The lignin-degrading bacteria are generally abundant in natural or manmade ligninrich environments, such as leaf litter, sludge of pulp paper mill, compost soils, decomposing woods, and activated sludge. Some unusual samples also contain various lignin-degrading bacteria. For example, several lignindegrading bacteria were separated from the steeping fluid of eroded bamboo slips, which were unearthed from the ancient tomb of more than 1700 years ago [17-19]. Wood-eating termites play an important role in natural carbon cycle and most lignocellulosic materials are digested in termite hindgut [20]. Thus, the termite gut is a rich source for the isolation of lignin-degrading bacteria and some bacteria responsible for lignin degradation were isolated successfully [21, 22]. In addition, some endophytes can decompose plant residues rapidly when the plants die due to its lignocellulose degradation ability, and endophytic bacteria were also isolated for lignin degradation [23].

In addition to sample sources, the screening method is another key factor for obtaining lignin-degrading bacteria with excellent performance. The commonly used method is enriching isolates by inoculating sample sources into mineral salts medium supplemented with lignin, synthetic lignin, or lignin-based aromatics as the sole carbon source [23-28]. In this case, the bacteria that cannot break down lignin and utilize lignin fragments for cell growth would be weeded out and lignin-degrading bacteria can be isolated. For instance, guaiacylglycerol- β -guaiacyl ether, a popular lignin model compound containing β -O-4 bonds, was applied to screen lignin degraders as β -O-4 linkages are the most abundant bonds in lignin [22, 29]. To inhibit the false positives caused by lignin degradation fungi, cycloheximide was commonly added into the selection medium [24, 30]. Even though the application of lignin and lignin-mimicked compounds can generally discover amounts of lignin-degrading bacteria, it cannot distinguish the lignin degradation abilities among the screened strains. Thus, it often entails a secondary screening. Due to the structural similarity between lignin fragments and some dyes, bacteria that can decolorize/degrade dyes generally have the capability to degrade lignin. Thus, the decoloration could serve as an index to determine the capabilities of lignin degradation. The commonly used dyes for lignin-degrading bacteria screening include Azure B, Toluidene Blue O, Methylene Blue, Malachite Green, Remazol Brilliant Blue R, indulin AT, etc. [21, 26, 28, 31-33].

For efficient screening of bacteria with exceptional lignin degradation capability, two high-throughput strategies were designed recently. In one case, target strains were screened based on their 2,2'-azino-bis(3-ethylbenzothiazoline-6-sulfonic acid) (ABTS) oxidizing activity. As well known, laccase belongs to ligninolytic enzyme system and most lignin-degrading bacteria can secrete laccase. ABTS can produce a green radical cation on oxidation by laccase, which has a strong absorption at 420 nm. Thus, when the culture broth of candidate bacteria was mixed with ABTS solution, the laccase-producing strains can be easily distinguished [29]. In another case, Chong and co-workers screened lignin degradation microbes based on the sensitive prussian blue spectrophotometric method [34]. When $K_3Fe(CN)_6$ and $FeCl_3$ are mixed with phenolic hydroxyl groups, prussian blue is formed because of the redox reaction. By determining the absorbance of formed prussian blue at 710 nm, the lignin content in culture broth can be calculated [35, 36]. The culture broth with low lignin suggests the presence of bacteria with exceptional lignin degradation capability. The ABTS and prussian blue-based methods can be supervised with a spectrophotometric instrument; thus, it can be convenient to realize the high-throughput screening for lignin-degrading bacteria.

Based on the methods mentioned above, a considerable amount of lignin-degrading bacteria have been isolated.

These bacteria mainly belong to phyla Proteobacteria, Actinobacteria, and Firmicutes. Recent study demonstrated that archaeal phylum Bathyarchaeota members also play an important role in lignin degradation [37]. The detailed information of the recently discovered lignin-degrading bacteria and typical lignin-degrading bacteria is shown in Table 1. More genotype and phenotype information associated with lignin-degrading bacteria can be found in the literature [38, 39]. With further researches, inherent enzymes and metabolic pathways involved in lignin degradation by these bacteria have been characterized, including enzymes/ pathways catalyzing oxidative and hydroxylation reactions, depolymerizing phenolic and non-phenolic lignin polymers, demethylation reactions and opening the aromatic rings of lignin-based compounds [33, 40-45]. Now, some of these bacteria have been used in treatment of sewages from pulp and paper industry, degradation heterogeneous compounds, and pretreatment of lignocellulosic biomass [46–49]. In particular, among these isolates, some strains (e.g., Pseudomonas putida KT2440, Sphingobium sp. SYK-6, and Rhodococcus opacus PD630) have already been applied as typical cells for producing value-added products from lignin and mining new lignin-degrading enzymes. The detailed information will be elaborated in the following texts.

Degradation pathways of lignin-based aromatics in bacteria

As mentioned above, bacteria have evolved multiple metabolic pathways to decompose lignin and assimilate its aromatic building blocks, and these sophisticated metabolic pathways are essential for lignin degradation. With the development of multi-omics technology, lignin-related metabolic pathways are becoming more accessible [40, 44, 45, 50]. Similar to fungi, some bacteria depolymerize lignin using multifarious enzymes, such as laccases, manganese peroxidases, dye-decolorizing peroxidases, cytochrome P450s, non-heme iron enzymes, dioxygenase, superoxide dismutases, and β -etherase enzymes. These lignin-degrading enzymes have been summarized in some excellent reviews [51–55]. Similarities and differences between bacteria and fungi in lignin depolymerization mechanism were described comprehensively in the above reviews. Even though the lignin depolymerization capability of bacteria is less than that of some famous lignin-degradation fungi [56], some of these bacteria can utilize the depolymerized lignin (lignin monomers, dimers and other low-molecular weight aromatic compounds) efficiently. It was hypothesized that lignin decomposition in nature is mainly initiated with fungi, which excrete powerful extracellular enzymes for lignin depolymerization. When lignin is depolymerized to monomers and/or

Taxonomic groups	Strains	Sample resources	Screen methods	Special characteristics	References
a-Proteobacteria	Rhizobium sp. YS-1 r	Decaying biomass	Enrich bacteria in mineral salts medium supplemented with alkali lignin, and 0.02% cycloheximide	Possess nitrogen-fixing ability and aro- matic hydrocarbons ability; possess a variety of lignin peroxidase and phenol oxidase	[24]
	Novosphingobium sp. B-7	Steeping fluid of eroded bamboo slips in ancient tomb	1	Can utilize kraft lignin as a sole carbon source; possess high manganese per- oxidase activity and laccase activity	[19]
eta-Proteobacteria	Comamonas sp. B-9	Steeping fluid of eroded bamboo slips in ancient tomb	1	Can reduce COD and color of lignin solutions; possess high manganese peroxidase activity and laccase activity	[17, 18]
	Pandoraea norimbergensis LD001 (DSM 24563)	Soil from beneath decomposing wood	Enrich bacteria in phosphate buffered mineral salts medium supplemented with kraft lignin	Can utilize at least seven lignin-based aromatic monomers; Can decolour- ize several lignin-mimicking dyes	[140]
y-Proteobacteria	P. deceptionensis (DSM 105530)	Sediments from the Baltic Sea	Enrich bacteria in M9 medium sup- plemented with lignin-related compounds and cycloheximide	Can catabolize various aromatic compounds	[30]
	Pseudomonas sp. (DSM 104486)	Mature vegetable compost	Enrich bacteria in mineral salts medium with kraft pulping stream or technical kraft lignin as the sole carbon source	Can catabolize vanillin, vanillic acid, 4-hydroxybenzic acid, p-coumaric acid, benzoic acid, and ferulic acid	[141]
	P. putida NX-1	Leaf mold samples	Enrich bacteria in M9 medium sup- plemented with kraft lignin	Possess high lignin peroxidase activity, manganese peroxidase activity and laccase activity	[28]
	Burkholderia sp. strain CCA53	Soil samples from Higashi-Hiroshima	Enrich bacteria in mineral salts medium with alkali lignin as the sole carbon source	Can catabolize benzaldehyde, benzoic acid, catechol, 4-hydroxy benzal- dehyde, 4-hydroxy benzoic acid, 4-hydroxybenzyl alcohol, syringol, and vanillin	[142]
	Serratia sp. JHT01 (DSM 29580), Serratia liquefaciens sp. PT01 (DSM 29581)	Forest surface soil	Primary screening with kraft lignin as the sole carbon source; secondary screening by determining the decol- orization of Remazol Brilliant Blue R as indicator	Possess dye decolorization ability on Azure B, Methylene blue, Remazol Brilliant Blue R; Can catabolize syringaldehyde, syringic acid, vanillin, vanillic acid, vanillyl alcohol, guaiacol, veratyl alcohol, and biphenyl	[32]
	Enterobacter sp. PY12	Termites guts	Primary screening with lignin as the sole carbon source; secondary screening by determining the decol- orization of Azure-B	Possess high lignin peroxides activity; possess dye decolorization ability on Congo red, Neutral red, Azure-B, Malachite green, and Methylene blue	[21]
	Trabulsiella sp. IIPTG13	Termite guts	Use guaiacylglycerol- β -guaiacyl ether as the sole carbon source	Can degrade lignin to organic acids and some low-molecular weight aromatics	[22]
	Pantoea sp. Sd-1	Rice seeds	Use lignin, rice straw powder, or lignin- related monomers as the sole carbon source	Can degrade both of cellulose, hemi- cellulose and lignin; can reduce color of lignin solutions	[23]

Taxonomic groups	Strains	Sample resources	Screen methods	Special characteristics	References
	Enterobacter soli sp. nov. LF7	Soil from the natural reserve	Enrich bacteria in mineral salts medium with alkali lignin as the sole carbon source	Possess high hydrogen producing abil- ity; can produce the oxidizer of ABTS	[143]
	Pseudomonas sp. Q18	Rotten wood in forests	Enrich bacteria in mineral salts medium supplemented with lignin	Possess a novel bacterial DyP-type peroxidase;	[144]
Firmicutes	Bacillus ligniniphilus L1 (DSM 26145, JCM 18543)	Sediments from 3415 m depth of the South China Sea	Enrich bacteria in mineral salts medium with lignin as the sole carbon source	Can produce at least 15 kinds of aro- matic compounds when using alkali lignin as the substrates	[40]
	Acetoanaerobium sp. WJDL-Y2	Anaerobic sludge from paper mill	Determine COD reduction of the broth containing lignin	It is an anaerobic bacterial strain; can reduce COD of lignin solutions	[145]
	Aneurinibacillus aneurinilyticus	Activated sludge from an effluent treat- ment plant of a pulp paper mill	 Primary screening by determining color reduction of the broth contain- ing lignin; secondary screening by using lignin-related low-molecular weight aromatics as substrates 	Can reduce color and lignin content of lignin solutions; can degrade lignin to organic acids and some low- molecular weight aromatics	[27]
	Paenibacillus glucanolyticus	Black liquor	I	It is an anaerobic microorganism that can catabolize black liquor, cellulose, hemicellulose and lignin	[146]
	Bacillus sp. CS-1, Bacillus sp. CS-2	Forest soils of 0–15 cm depths	Determine the decolorization of Rema- zol Brilliant Blue R	Possess high laccase activity; can be used for biological pretreatment	[147]
	Bacillus atrophaeus LSSC3 and Bacillus pumilus strain CL29	Decayed plant material and soil in tropical rainforest	Determine isolates possessing laccase activity with ABTS as substrate	Can degrade guaiacylglycerol- eta - guaiacyl ether	[29]
	Bacillus sp. SHC1	Soils of 0–10 cm depths from palm oil plantation	Primary screening by using kraft lignin as the sole carbon source; secondary screening by determining the decol- orization of methylene blue	Possess high manganese peroxidase activity and lignin peroxidase activity	[33]
Actinobacteria	Streptomyces sp. Y-8	Soil samples	Enrich bacteria in mineral salts medium supplemented with lignin as the sole carbon source	Possess high laccase activity and lignin peroxidase activity	[148]
	Rhodococcus opacus PD630 (DSM 44193)	Soils from a gas-works plant	Enrich bacteria in mineral salts medium supplemented with phenyldecane as the sole carbon source	Can accumulate up to 87% (w/w) lipid in cell	[149]
	Rhodococcus jostii RHA1	Soils contaminated by y-hexachlorocyclohexane	Enrich bacteria in mineral salts medium supplemented with biphenyl as the sole carbon source	Can break down some polychlorinated biphenyl congeners into smaller molecules	[150]
	Rhodococcus pyridinivorans CCZU-B16	Soil sample	Use a high-throughput screening strat- egy using Prussian blue spectropho- tometric method	Can convert alkali lignin into microbial lipids	[34]

low-molecular weight aromatics, bacteria assimilate them for carbon and energy through their well-adapted metabolic pathways [57]. This section focuses on the bacterial metabolic versatility for the assimilation of lignin-related low-molecular weight aromatics (Fig. 2).

The degradation of G-lignin-based compounds (e.g., ferulic acid)

The G-lignin unit accounts for 37.8%, 98.3%, 38.3%, and 77.1% of the lignin in poplar (a typical hardwood) wood,

pine (a typical softwood) wood, corn (a typical monocotyledon) stover, and *Arabidopsis* (a typical dicotyledon) inflorescence stem, respectively [58]. Ferulic acid is a standard model compound for G-lignin. Structurally, it is covalently linked to C-5 of the L-arabinofuranosyl residue which is attached to the xylan backbone and acts as an anchor of lignification in herbaceous biomass [59, 60]. As mentioned in "The lignin-degrading bacteria and their screening methods", there are some bacteria that can degrade ferulic acid and utilize it as the sole



carbon source for cell growth. The degradation pathway of ferulic acid can be divided into four categories: nonoxidative decarboxylation pathway [61, 62], coenzyme A (CoA)-dependent non- β -oxidation pathway [63–66], CoA-dependent β -oxidation pathway [66, 67], and side chain reduction pathway [68]. Even though involved with different intermediates and enzymes, these four pathways are all funneled into vanillic acid for further degradation. Moreover, some other lignin-based aromatics were also degraded through these pathways, such as vanillin, vanillic acid, and dihydroferulic acid (Fig. 2). In other words, these lignin intermediate products, if present in hydrolysates, can also be assimilated by these pathways.

The degradation of p-coumaric acid

The H-lignin unit accounts for 0.3%, 1.7%, 2.8%, and 2.8% of the lignin in poplar wood, pine wood, corn stover, and Arabidopsis inflorescence stem, respectively [58]. It differs from G- and S-lignin units; there are no methoxy groups at the 3' or 5' positions of phenylpropane units. p-Courmaric acid and some other hydroxycinnamic acid have been commonly utilized to represent H-lignin units. Structurally, some p-coumaric acid is linked to the hemicellulose with ester linkages in herbaceous biomass, which can be released together with ferulic acid under alkaline hydrolysis conditions [59, 60, 69]. Generally, the degradation pathways of p-coumaric acid in bacteria can be divided into three categories: CoAdependent β -oxidation pathway [70, 71], CoA-dependent non- β -oxidation pathway [64, 69, 72], and CoA-independent pathway [4, 73]. Ultimately, all these three pathways converge at the intermediate of *p*-hydroxybenzoic acid, which is then converted to protocatechuic acid by p-hydroxybenzoate-3-hydroxylase for further metabolism (Fig. 2).

The degradation of S-lignin-based compounds (e.g., syringic acid)

The S-lignin unit accounts for 61.9%, 0, 58.9%, and 20.1% of the lignin in poplar wood, pine wood, corn stover, and *Arabidopsis* inflorescence stem, respectively [58]. There are two methoxy groups on the aromatic ring of S-lignin, in contrast to one and zero methoxy group on that of G-, and H-lignin, respectively, which make the degradation of S-lignin more difficult than that of G-, and H-lignin. Syringic acid is considered to be a model compound of S-lignin. Compared with ferulic acid and *p*-coumaric acid, there are fewer studies on microbes that can efficiently degrade syringic acid, which indicates the adverse effects of aromatic methoxy on S-lignin catabolism. The knowledge of bacterial S-lignin degradation pathway

is mainly derived from *Sphingomonas* sp. SYK-6, which was isolated as a 5,5'-dehydrodivanillate degrader in 1987 [74]. Now, it is one of the most widely used bacteria for lignin degradation study.

In *Sphingomonas* sp. SYK-6 strain, syringic acid is first *O*-demethylated to 3-*O*-methylgallate (3MGA) by a tetrahydrofolate-dependent *O*-demethylase (DesA). Subsequently, there are three pathways for 3MGA degradation: gallic acid (GA) as an intermediate [75, 76], 4-carboxy-2-hydroxy-6-methoxy-6-oxohexa-2,4-dienoate as an intermediate [77, 78], 3MGA is directly integrated into protocatechuic acid 4,5-cleavage pathway [78, 79]. With the aforementioned three pathways, syringic acid is assimilated into the protocatechuic acid 4,5-cleavage pathway and ultimately integrated to the tricarboxylic acid (TCA) cycle (Fig. 2).

The degradation of protocatechuic acid and catechol

Protocatechuic acid and catechol are widely distributed in various lignin hydrolysates and they are also two key branch points in bacterial lignin degradation pathways (Fig. 2). From the above discussion, both G- and H-lignin components are metabolized using protocatechuic acid as intermediate and S-lignin components are degraded through the protocatechuic acid 4,5-cleavage pathway. The degradation of protocatechuic acid in microbes has been classified into three categories: 3,4-cleavage pathway [80, 81], 4,5-cleavage pathway [82–84], and 2,3-cleavage pathway [85, 86]. Some other lignin-based aromatics (e.g., phenol, benzene, benzoate, toluene, naphthalene, etc.) could be funneled into catechol for further degradation. The degradation of catechol is mainly catalyzed by dioxygenases through ortho- or meta-cleavage pathway [87–92]. In nature, the *ortho*-cleavage pathway and *meta*cleavage pathway of catechol are not exclusive to each other; they can co-exist in a bacterium. However, they may be induced differently by different carbon sources. For example, when grown on salicylic acid, only the catechol ortho-pathway is induced in *P. cepacia*. In contrast, when grown on benzoate, the ortho- and meta-pathways could be induced simultaneously for catechol degradation [93].

To summarize, bacteria have evolved a wide variety of distinct pathways to metabolize lignin-based compounds. The biochemical richness in bacteria provides great opportunities for pathway engineering for overproduction of valuable intermediates. For example, the metabolisms of some G- and H-lignin compositions share protocatechuic acid as a common intermediate. Thus, these compounds can be funneled to various target products through protocatechuic acid.

Application of multiple metabolic pathways for lignin valorization

Traditionally research on processing lignocellulose to biofuels and chemicals has focused on the carbohydrate fractions. However, as hexose and pentose fermentation technologies near maturation, it is becoming increasingly apparent that it is desirable to develop useful strategies to make use of considerable amount of residual lignin. However, the lignin components in fermentation streams are heterogeneous, containing organic acids, fermentation intermediates, and residual enzymes (e.g., cellulases) in addition to the targeted lignin-based compounds, which hinder its application in many fields (e.g., nanomaterials, fine chemicals). Biological conversion is a commendable supplement for the thermo-chemical lignin valorization route because some bacteria can funnel lignin fragments and other fermentation residuals into target chemicals [5, 11, 51, 94, 95]. Actually, several wild lignin-degrading bacteria possess lipid or polyhydroxyalkanoates (PHA) synthesis capability. Moreover, engineered bacteria with pyruvate, lactate, pyrogallol, and vanillin synthesis capability have also been constructed to utilize lignin and its fragments as substrates. All these examples demonstrate that it is technically feasible to convert lignin-enriched streams to value-added products with bacteria. In this section, recent advances in lignin conversion to bioproducts are presented.

Application of lignin degradation pathways for lipid production

Lipids are attractive feedstocks for production of biofuels. Generally, oleaginous organisms can accumulate > 20% of their dry cell weight (DCW) as lipids. In recent years, researchers found that some oleaginous microbes can synthesize lipid from lignin-based aromatics and Rhodococcus is such a promising species among these microbes, because of its robust growth, tolerance to various aromatics, broad substrate specificity, as well as robust lipid production capability [96, 97]. As mentioned in the "Degradation pathways of lignin-based aromatics in bacteria", Rhodococcus bacteria can metabolize various ligninbased aromatics. These lignin-based aromatics undergo ring cleavage and are converted to acetyl-CoA, which is an essential precursor for lipid biosynthesis. For example, when *p*-hydroxybenzoic acid or vanillic acid was applied as the sole carbon source, both R. opacus PD630 and R. opacus DSM 1069 can survive well on these two lignin model monomers and accumulate approximately 20% of DCW as lipids under nitrogen limiting conditions [98]. Further studies demonstrate that blending lignin-based aromatics with glucose increased lipid contents considerably [99].

Compared with model aromatics, the actual residual lignin from biorefinery processes is of more interesting. Kosa and his co-workers validated the feasibility of lignin-to-lipid conversion with R. opacus, even though with limited lipid content and low lipid titer [100, 101] (Table 2). Monitoring lignin fractions during the fermentation process showed that the low-molecular weight lignin fractions were digested successfully, leaving the more recalcitrant lignin-based polymers undigested. It was suggested that these bacteria were not good at depolymerizing lignin, but can assimilate low-molecular weight lignin fractions efficiently. Thus, it was hypothesized that modifying lignin properties, such as decreasing the molecular weight and destructing obstinate chemical bonds, would be helpful for lipid accumulation with better efficiency. Thus, chemical lignin depolymerization methods and biological depolymerization methods were integrated with bacterial cultures for lipid production from lignin [35, 102]. In these ways, lignin was depolymerized first and then the depolymerized lignin can be utilized more efficiently by related bacteria. Moreover, bacteria co-culture systems were also applied for lipid production from lignin under the consideration that different bacteria can be complementary to each other on lignin depolymerization and assimilation [103].

In addition to lignin residuals, aqueous wastes containing lignin-based aromatics can be also applied for lipid production with bacteria. A large amount of waste effluent could be generated during the pretreatment process of lignocellulose, which contains different fractions based on the pretreatment method. For example, alkali pretreatment processing would generate waste effluent that contains lignin, aromatic monomers, oligosaccharides, acetate, and other carbohydrates, which can be utilized by some lipid producing microorganisms [16, 103]. Thus, the co-production of bioethanol and lipid in an integrated biorefinery route was realized aimed at making use of all lignocellulose composition [104]. Thermochemical process is another choice for the conversion of lignocellulosic biomass to liquid fuels in addition to the biochemical process. In hydrothermal liquefaction processes, biomass is pyrolyzed to decompose the solid polymeric structure into a mixture of gas, biocrude, solid residue, and aqueous waste. Generally, this aqueous waste contains high concentrations of organic acids, alcohols, ketones, ammonia and lignin-based aromatics, which can be converted into lipids by some bacteria, along with the reduction of chemical oxygen demand (COD) [105]. Biomass gasification is another thermochemical process where biomass is converted to synthetic gas at a higher temperature. During the biomass gasification process, a large amount of wastewater is generated when the synthetic gas is scrubbed and it contains

Products	Substrates	Strains	Main Strategies	Titers	References
Lipid	4-Hydroxybenzoic acid, vanil- lic acid and glucose as the co-substrates	Rhodococcus rhodochrous ATCC 21198	Use lignin model monomer and glucose as co-sub- strates for lipid production	>40% of DCW	[99]
	Ultrasonicated ethanol orga- nosolv lignin	R. opacus DSM 1069	Integrate <i>R. opacus</i> -based assimilation process and ultrasonication-based lignin pretreatment process	4.08% of DCW	[101]
	$\rm O_2$ pretreated kraft lignin	R. opacus DSM 1069	Integrate <i>R. opacus</i> -based assimilation process and O ₂ -based lignin pretreat- ment process	14.21% of DCW, 0.067 mg/mL	[102]
	Kraft lignin	R. opacus PD630	Integrate <i>R. opacus</i> -based assimilation process and laccase-based lignin depo- lymerization process	About 150 mg/L	[35]
	Alkali-extracted lignin	<i>R. opacus</i> PD630 and <i>R. jostii</i> RHA1 VanA ⁻	Co-culture of <i>R. opacus</i> and <i>R. jostii</i> RHA1	0.39 g lipid/g DCW	[103]
	Algal hydrothermal liquefac- tion aqueous wastes	<i>R. opacus</i> PD630, <i>R. jostii</i> RHA1, and VanA	Co-culture of <i>R. opacus</i> and <i>R. jostii</i> RHA1	0.46 g lipid/g DCW	[105]
	Biomass gasification waste- water	<i>R. opacus</i> DSM 43205	Supply wastewater with some mineral salt for bet- ter cell growth and lipid production	62.8% of DCW	[106]
	Effluent generated from a two-stage pretreatment of NaOH pre-extraction and alkaline H ₂ O ₂ post-treatment.	<i>R. opacus</i> PD630	Integrate <i>R. opacus</i> -based assimilation process and alkali/alkali-peroxide-based pretreatment	1.3 g/L and 42.1% of DCW	[104]
	Ammonia fiber expansion corn stover lignin	<i>R. opacus</i> NRRL B-3311	Apply ammonia fiber expan- sion corn stover lignin without pretreatment for lipid production	32 mg/L	[151]
	Lignin from combinatorial pretreatment	R. opacus PD630	Consolidate combinato- rial pretreatment, laccase addition and fed-batch fermentation processes	1.83 g/L	[152]

Table 2 Lipid production from lignin degradation pathways by bacteria

-, not mentioned in the paper

abundant lignin fractions. Using the biomass gasification wastewater with mineral salts as substrate, *R. opacus* DSM 43,205 can accumulate 62.8% g lipid/g DCW with a wastewater COD removal efficiency of 74% [106]. These two examples both indicated the prospect of bioconversion of underutilized aromatics in waste water from biorefinery process into useful products by selected bacteria.

Application of lignin degradation pathways for PHA production

PHA is a group of biopolyesters synthesized as energy reserve inside cells and can be produced by variety of microbes under nutrient imbalance conditions. In the last three decades, PHA as biodegradable plastics has attracted wide attention not only because they have compatible material performance but also because they could be produced from renewable carbon sources, even from inferior biomass constituents [107]. As mentioned in "Application of lignin degradation pathways for lipid production", lignin derivatives can be metabolized to acetyl-CoA, a precursor for lipid, as well as for PHA synthesis. PHA and lipid are all intracellular compounds, which can be separated from lignin streams just by centrifugation. Thus, it is a simpler process to convert lignin to PHA and lipid compared with pyrolyzing lignin to various chemicals which requires extensive separation and purification procedures.

There are various bacteria capable of producing PHA from lignin-based aromatics [16, 50, 108–111]. In addition to aromatic monomers, some actual lignin streams can also be utilized for PHA biosynthesis [112, 113] (Table 3). Just like lipid production from lignin-based streams, the PHA production from lignin streams by bacterial cultures is also generally concurrent with the reduction of the liquor color and COD [108, 114]. The

Products	Substrates	Strains	Main strategies	Titers	Yields	References
PHA	Thermo-chemical wastewater streams	Engineered <i>P. putida</i> KT2440	Construct a strain with high tolerance to highly toxic substrates	_	-	[153]
	Lignin from a combined pretreatment strategy	Engineered <i>P. putida</i> KT2440	Perform a fed-batch fermen- tation and use lignin from a combined pretreatment strategy as substrate	1.0 g/L	17.6% mol/mol	[115]
	Kraft lignin	Pandoraea sp. ISTKB	Apply a nitrogen-limited culture condition	18 mg/L	-	[108]
	Kraft lignin	Cupriavidus basilensis B-8	Perform a fed-batch fermen- tation	319.4 mg/L	-	[114]
	Insoluble kraft lignin	Engineered <i>P. putida</i> A514	Strengthen three functional modules of lignin depolym- erization system, aromatic compound catabolism system, and PHA synthesis system	75 mg/L		[116]
	Alkaline pretreated liquor	P. putida KT2440	Apply alkaline pretreated liq- uor directly without dilution	0.252 g/L	-	[16]
	Lignin	Oceanimonas doudoroffii	Perform a two-phase culture: the pre-culture with marine broth medium and the PHA production stage with mineral salt medium added lignin and lignin derivatives	0.2% of DCW	-	[154]
	benzoic acid	P. putida KT2440	Apply a nitrogen-limited culture condition	37.3% of DCW		[155]

Table 3 PHA production from lignin degradation pathways by bacteria

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PHA yield is relatively low when lignin residuals generated from traditional biorefinery process was used as substrates. As present in Table 3, only a milligram level of PHA was obtained when related bacteria were cultured in traditional lignin steams, such as kraft lignin and alkaliextracted lignin. Additional depolymerization steps can contribute to a better PHA production. Liu and co-workers applied lignin from an H₂SO₄ and NaOH combined pretreatment process as substrate for PHA production, and 1.0 g/L PHA was achieved. Further study indicated that this combined pretreatment process facilitated more lignin components accessible to PHA biosynthesis by increasing the contents of G- and H-lignin, reducing the β - β and β -O-4 bonds, and fractionating more aromatic monomers [115]. Besides integration of exogenous lignin depolymerization processes, enhancing the autologous lignin utilization capability of related bacteria is another choice for improving the lignin valorization efficiency. In this consideration, Lin and co-workers attempted to integrate three functional modules of lignin utilization in a wild P. putida strain, including the dye-decolorizing peroxidases-based lignin depolymerization system, the β -ketoadipate pathway-based aromatic compound catabolism system, and the PHA polymerase-based PHA synthesis system. As a result, this consolidated P. putida led to a sixfold increase of PHA titer [116]. To enable a broader slate of the produced PHA, the lignin valorization route was lengthened by consolidating the PHA production with a chemical catalysis. First, lignin stream or pretreated lignocellulosic liquor was converted to PHA by related microorganisms. Subsequently, the produced PHA was catalytically converted to alkenoic acids and hydrocarbons, which are precursors of diverse chemicals [16]. In this way, lignin can be transformed into biomaterials, chemical precursors and fuel-range hydrocarbons.

Application of lignin metabolic pathways for *cis, cis*-muconic acid production

The compound *cis*, *cis*-muconic acid (*cis*, *cis*-MA) attracts large amount of attention recently because it can be applied as an intermediate for adipic acid production, which is a bulk feedstock of fibers and plastics. There is a reported market greater than \$22 billion for *cis*, *cis*-MA globally [117]. Currently, the industrial production of *cis*, *cis*-MA depends mainly on chemical synthesis using petroleum-based feedstocks. Compared with the wide-spread challenges of petrochemical processes, *cis*, *cis*-MA from renewable biomass provides a feasible alternative to alleviate the concerns of environmental issues and finite fossil resources. As presented in Fig. 2, *cis*, *cis*-MA is an

intermediate of the β -ketoadipate pathway. Compared with lipid and PHA, *cis, cis*-MA is situated well before aromatics entering central carbon metabolism, which indicates that additional metabolic steps are not needed to divert aromatic carbons for cell growth and energy supply, and thus allows for a better atom efficiency.

As illustrated in Fig. 2, several aromatic-utilizing bacteria can employ catechol 1,2-dioxygenase to convert catechol to cis, cis-MA; however, cis, cis-MA is a metabolic intermediate and not accumulated by native strains. When cis, cis-MA degradation pathway was blocked, the recombinant strains can accumulate cis, cis-MA and secrete it into the culture broth. In native strains, phenol, benzene, benzoate, toluene, cinnamic acid and some other compounds, which wildly exist in lignin hydrolysate, can be funneled to cis, cis-MA through catechol. However, some other important components of lignin, such as *p*-coumaric acid, ferulic acid, and vanillin, were degraded using protocatechuic acid as a key intermediate, instead of catechol. Thus, these components of lignin cannot be converted to *cis, cis*-MA with the native metabolic pathways. To capture aromatics metabolized through protocatechuic acid degradation branch, Vardon and co-workers bridged the protocatechuic acid and catechol branches, along with blocking further metabolism of protocatechuic acid. Then, the modified strain was sequentially engineered to broaden its substrate spectrum (e.g., benzoate and phenol). Ultimately, the engineered strain could funnel multiple lignin-based aromatics and actual lignin steams to cis, cis-MA with a high efficiency [118] (Table 4). It is well known that almost all metabolic pathways are involved with multiple regulators as well as versatile key enzymes. The cis, cis-MA production pathway is no exception. The cis, cis-MA production capability was further improved by the co-expression of two genetically associated proteins of protocatechuic acid decarboxylase, and the deletion of the carbon catabolite repression control proteins [119, 120]. Recently, the titer, yield and productivity of MA from recombinant P. putida were constantly improved with a combination of gene overexpression, removal of global catabolic regulator, constant fed-batch and highpH feeding strategy. As a result, as much as 50 g/L MA was produced from p-coumaric acid and 3.7 g/L cis, cis-MA was produced from base-catalyzed depolymerized lignin [121]. Although *cis, cis*-MA is secreted into the lignin streams during fermentation process compared with lipid and PHA, its structure of dicarboxylic acid is different to lignin-based aromatics. Thus, it can be also separated efficiently from lignin-based solutions. Based on the high cis, cis-MA titer, a separation and purification scheme consisting of protein removal process, active carbon cleanup process, crystallization process, and ethanol purification process was applied to achieve high purity *cis*, *cis*-MA (99.8%) [122]. The practical feasibility of the entire route from lignin to nylon-6,6 was demonstrated with the procedures of depolymerization of lignin, fedbatch fermentation process, recovery and purification of *cis*, *cis*-MA, catalytic hydrogenation, and polymerization of adipic acid to nylon 6,6 [123].

In addition to *P. putida*, other bacteria were also found to produce cis, cis-MA from lignin-based aromatics. Amycolatopsis sp. ATCC 39,116 with a deletion of two cis, cis-muconate cycloisomerases could accumulate 3.1 g/L cis, cis-MA from guaiacol with a yield of 96%, and 255.8 mg/L cis, cis-MA from softwood lignin hydrolysate [124]. In a very recently reported achievement, Corynebacterium glutamicum was also applied in cis, cis-MA production due to its robust tolerance to lignin-based aromatics. The recombinant C. glutamicum can produce 85 g/L and 1.8 g/L cis, cis-MA from catechol and lignin hydrolysate, respectively [125]. In the above studies, additional glucose or organic acid was needed for cell growth. Sonoki et al. constructed an engineered strain to utilize S-lignin derivatives for cell growth and G-lignin derivatives for *cis*, *cis*-MA production. This way, hardwood lignin which contained abundant G-lignin and S-lignin components could be utilized comprehensively for cis, cis-MA production without additional glucose [126]. As one of the most famous historic commercial microorganisms, E. coli was also applied for utilizing lignin-based aromatics because of its fast growth, unambiguous genetic background, and readily available genetic tools [127-129]. With different gene expression strategies, 100-314 mg/L cis, cis-MA were produced from vanillin [117].

Application of lignin degradation pathways for aromatics

Since the European and US food legislations permitted the word "natural" to be used for products derived from biological sources, "natural" compounds have been considered superior and are more expensive than synthetic ones. Therefore, there is a growing interest in producing natural aromatics due to their wide applications. The unique aromatic structure and the reproducible characteristics of lignin make it an ideal feedstock for natural aromatics. However, vanillin, *p*-hydroxybenzoic acid, and pyrogallol are the only aromatics that have been produced from lignin through biological methods.

Vanillin is the main organoleptic ingredient of the vanilla pod and is widely applied by food, cosmetics, pharmaceutical, and other industries. Because of the limited supply and high price of natural vanillin, the current global demand for vanillin is mainly provided by chemical conversion of petrochemicals and thermo-chemical-mediated lignin degradation process [130]. Nowadays,

Products	Substrates	Strains	Main strategies	Titers	Yields	References
<i>cis, cis</i> -MA	Catechol	C. glutamicum MA-2	Delete <i>catB</i> gene; express <i>catA</i> gene; apply a fed-batch fermentation process	85 g/L	100% mol/mol	[125]
	<i>p</i> -Coumaric acid	<i>P. putida</i> KT2440-CJ103	Express <i>aroY</i> gene; delete <i>pcaHG</i> and <i>catB</i> genes; apply dissolved oxygen static fed-batch fermenta- tion	13.5 g/L	-	[118]
	<i>p</i> -Coumaric acid	P. putida KT2440-CJ184	Co-express ecdB and ecdD with aroY gene	15.59 g/L	1.01 mol/mol	[120]
	<i>p</i> -Coumaric acid	P. putida KT2440-CJ238	Delete genes encoding car- bon catabolite repression control protein	-	0.946 mol/mol	[119]
	<i>p</i> -Coumaric acid	P. putida KT2440-CJ242	Perform a fed-batch fermenta- tion with by using high pH solution of <i>p</i> -coumaric acid as feeding solution	50 g/L	-	[121]
	Guaiacol	Amycolatopsis sp. ATCC 39116	Delete two putative <i>catB</i> genes; Perform a fed-batch fermentation	3.1 g/L	0.96 mol/mol	[124]
	Vanillin	Engineered <i>E. coli</i>	Co-express four genes of vdh, desA, catA and aroY	341 mg/L	0.69 g/g	[117]
	Sodium benzoate and glucose	P. putida KT2440-CJ102	Perform a dissolved oxygen- state fed-batch fermenta- tion process	35.4 g/L	-	[122]
	Vanillin	Engineered E. coli	Co-express genes of vdh, vanA, vanB, catA, aroY and kpdB	-	-	[156]
	Lignin hydrolysate	C. glutamicum MA-2	Delete <i>catB</i> gene; express <i>catA</i> gene; apply a fed-batch fermentation process	1.8 g/L	-	[125]
	Alkaline pretreated liquor	<i>P. putida</i> KT2440-CJ103	Introduce <i>aroY</i> gene; delete <i>pcaHG</i> and <i>catB</i> genes	0.7 g/L	-	[118]
	Softwood lignin hydrolysate	P. putida IDPC/pTS110	Co-express <i>pcaHG</i> and <i>aroY</i> genes; apply a dissolved oxygen static batch fermen- tation process	-	0.3–0.331 mol/mol lignin- based aromatics	[126]
	Hardwood lignin hydrolysate	Sphingobium sp. SME257/ pTS084	Use G-lignin components for <i>cis, cis</i> -MA production and S-lignin components for cell growth	26.8 mg/L	0.41 mol/mol birch lignin derivatives	[126]
	Softwood lignin hydrolysate	P. putida MA-9	Construct a strain with high tolerance to catechol; enhance catechol 1,2-dioxy- genase expression levels; depolymerize softwood lignin in supercritical water	13 g/L	Nearly 100% from lignin- based aromatics	[123]
	Softwood lignin hydrolysate	Amycolatopsis sp. ATCC 39116	Delete two putative <i>catB</i> genes; pretreat lignin with low temperature hydrother- mal conversion method	255.8 mg/L	0.72 mol/mol	[124]
	Alkaline pretreated lignin liquor	KT2440-CJ475	Perform a constant fed-batch fermentation process	0.65 g/L	>100% ^a	[121]
	Base-catalyzed depolymerized lignin	KT2440-CJ242	Perform a constant fed-batch fermentation process	3.7 g/L	>100% ^a	[121]

Table 4 cis, cis-MA production from lignin degradation pathways by bacteria

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 $^{\rm a}~$ The MA yield was calculated as MA mol/(p-coumaric acid + ferulic acid) mol

consumers' demands for natural products have motivated extensive research into biological methods for vanillin production from glucose, phenolic stilbenes, isoeugenol, eugenol, or ferulic acid through fungi, bacteria and plant cells [68]. There are a lot of microbes reported to have the capability of decomposing lignin to vanillin, including fungi and bacteria. However, vanillin was detected only in trace amounts in most cases. Recently, the vanillin metabolism in *R. jostii* RHA1 was blocked and the mutant strain was found to accumulate 96 mg/L vanillin when grown on 2.5% wheat straw lignocellulose [131] (Table 5). Just recently, the microbial fuel cell system has been also designed for lignin depolymerization to aromatics. In this case, lignin was dissolved in the aerobic cathode chamber and microbial electrochemical cells were cultivated in the anode chamber. When these two chambers were connected with a salt bridge and an external wire, electrons generated from microbial electrochemical cells will reduce oxygen molecules to produce H_2O_2 at the cathode. Then, lignin was depolymerized through H_2O_2 -mediated oxidative reaction, with some vanillin produced [132]. It means that microbial fuel cell system may be a choice for the production of aromatics from lignin.

p-Hydroxybenzoic acid is an important mono-hydroxybenzoic acid due to its excellent antimicrobial and antioxidant properties and low toxicity. Its esters are wildly used as preservatives in food, flavors, cosmetics and pharmaceutical products [133]. Although the synthetic technology for *p*-hydroxybenzoic acid has been developed for

Table 5 Aromatic and other chemicals from lignin degradation pathways by bacteria

Products	Substrates	Strains	Main strategies	Titers	Yields	References
<i>p</i> -Hydroxybenzoic acid	<i>p</i> -Coumaric acid	Engineered Burkholde- ria glumae BGR1	Delete genes encod- ing <i>p</i> -hydroxyben- zoate-3-hydroxylase, benzoyl-CoA ligase; overexpress <i>p</i> -hydroxcinnmaoyl- CoA synthetase II	2.73 g/L	99.0% mol/mol	[69]
Pyrogallol, GA	Syringic acid	Engineered <i>E. coli</i>	Co-express <i>desA</i> , <i>ligM</i> and a GA decar- boxylase gene	7.3 mg/L pyrogallol, 18 mg/L GA	7.3 mg pyrogallol/g syringate, 18 mg GA/g syringate	[117]
Vanillin	2.5% Wheat straw lignocellulose	R. jostii	Delete <i>vdh</i> gene	96 mg/L	-	[131]
Vanillin	Lignin extracted from wheat straw	Shewanella putrefa- ciens	Construct a microbial fuel cell system for lignin depolymeri- zation	275 mg/L	-	[132]
Pyruvate	<i>p</i> -Coumaric acid	P. putida KT2440- CJ122	Choose protocat- echuic acid <i>meta-</i> cleavage pathway for <i>p</i> -coumaric acid conversation	-	0.414 g/g	[135]
Lactic acid	<i>p</i> -Coumaric acid	P. putida KT2440- CJ122	Choose protocat- echuic acid <i>meta-</i> cleavage pathway for <i>p</i> -coumaric acid conversation	-	0.411 g/g	[135]
Pyruvate	<i>p</i> -Coumaric acid	P. putida KT2440- CJ124	Choose protocat- echuic acid <i>ortho</i> - cleavage pathway for <i>p</i> -coumaric acid conversation	-	0.019 g/g	[135]
Lactic acid	<i>p</i> -Coumaric acid	P. putida KT2440- CJ124	Choose protocat- echuic acid <i>ortho</i> - cleavage pathway for <i>p</i> -coumaric acid conversation	-	0.145 g/g	[135]
Methane	Hydrolysis lignin	Anaerobic microor- ganisms	Degrade lignin by anaerobic digestion in a nylon bag		125 mL methane/g volatile solid	[136]

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many years, there are still several limitations in the production of *p*-hydroxybenzoic acid by chemical synthesis, such as low regional selectivity, harsh conditions, and by-product formation. Jung and co-workers engineered a natural pathway to produce *p*-hydroxybenzoic acid by deleting genes involved in p-hydroxybenzoic acid degradation. With *p*-coumaric acid as substrate, the mutant strain accumulated 2.73 g/L p-hydroxybenzoic acid with a 99% conversion [69]. Pyrogallol is a common raw material used in chemical synthesis to produce bioactive molecules. Recent research work indicated that pyrogallol has benign anti-proliferative effects on some cancer cells [134]. Thus, there is a huge demand for pyrogallol, especially the biologically produced pyrogallol for medicine preparation. Wu and co-workers attempted to convert lignin-based aromatics into pyrogallol [117]. In their studies, two demethylase genes with a GA decarboxylase gene were co-expressed. The results showed that the recombinant strain yielded about 7.3 mg/L pyrogallol and 18 mg/L GA from syringic acid, which was obtained from hydrogen peroxide-catalyzed lignin.

In addition to the products mentioned above, lignin can also be converted to succinate, acetyl-CoA, pyruvate, and lactic acid through various metabolic pathways. Moreover, different aromatic metabolic pathways differ in intermediates, reducing equivalents, and carbon emitted, which will ultimately lead to different products, and/or different yields of the targeted product [135]. For example, the *ortho*-cleavage pathways of both protocatechuic acid and catechol yield one succinate and one acetyl-CoA; the *meta*-cleavage of catechol and the 2,3 *meta*-cleavage pathway of protocatechuic acid yield one pyruvate and one acetyl-CoA; the 4,5 *meta*-cleavage pathway of protocatechuic acid ultimately yields two pyruvates (Fig. 2).

The above chemicals are all produced by aerobic fermentation process. As mentioned in "The lignindegrading bacteria and their screening methods", some anaerobic bacteria can also metabolize lignin. These bacteria were also applied for lignin valorization. For example, hydrolysis lignin was converted into biogas by anaerobic digestion. In this way, the cellulose and hemicellulose parts of lignocellulose can be converted to ethanol and the lignin part can be converted to methane, which will improve the energy yield significantly [136]. With the aforementioned information, lignin represents a potential renewable feedstock for aromatics and other platform bio-products if suitable bio-catalysis routes are developed.

Conclusion and perspectives

Even though various lignin-degrading bacteria were found widely and some lignin-based aromatics metabolic pathways have been elucidated and applied to produce bio-products. Based on the current knowledge, it has been suggested that the conversion of high-molecular weight lignin into bacteria available compounds (such as aromatic monomers or dimers) is the major bottleneck in the synthesis of bio-products from lignin. There is still a long way to go before lignin valorization at an industrial scale with bacteria can be realized technically and economically. The following perspectives should be considered for future studies:

- 1. Most lignin-degrading bacteria can only assimilate a fraction of lignin-based compounds. More efficient bacteria and metabolic pathways are in need for the comprehensive utilization of lignin or lignocellulosic biomass. As we further our understanding of lignin degradation process in bacteria, it is expected that pathway engineering can be applied in suitable bacterial hosts to assimilate more lignin components, as well as achieving high yields of the targeted products.
- 2. One drawback of lignin valorization by bacterial system is the low product titers. The product titers from lignin-based solutions are much lower than from glucose or other common substrates. Except the limited lignin utilization capability of target bacteria, the inhibition from lignin-based compounds is another key factor that causes the low microbial productivity. Fed-batch fermentation is a good solution for releasing the inhibition from high content lignin. Moreover, some microbes with high tolerance to ligninbased compounds can be obtained by screening, genetic engineering, or adapted evolution.
- 3. One major disadvantage of using microbes in lignin valorization processes is their low capability in utilizing water-insoluble and/or high-molecular weight lignin. Thus, appropriate depolymerization processes are required to disrupt lignin-enriched substrates into low-molecular weight and water-soluble species that can be assimilated by bacteria efficiently (Fig. 3). In particular, gasify the solid lignin and run a syngas bacterial culture may be a promising strategy for lignin valorization by bacteria cultures. In this case, the gasified components will be utilized more efficiently by related bacteria compared with the solid lignin or other lignin streams.
- 4. In previous lignocellulosic biorefinery designs, biomass pretreatment was generally designed for high content fermentable sugars. As different pretreatments can contribute to different lignin characteristics [137], lignin valorization is expected to be considered in addition to fermentable sugars when pretreatment and process are designed.
- 5. In addition to biological methods, other alternative methods have also applied wildly in lignin valoriza-



tion, e.g., reduction, supercritical fluids, ionic liquids treatment, and fractionation by ultrafiltration and selective precipitation [11, 13]. These methods are not standalone and different alternatives pathways must be integrated for better lignin valorization. For example, the present results suggest that the productivities of lignin valorization by biological methods are relatively low and many bacteria can only utilize small lignin fragments. Thus, it is promising to integrate the chemical or physical lignin depolymerization processes and biological assimilation processes.

Abbreviations

AroY: protocatechuic acid decarboxylase; CatA: catechol 1, 2-dioxygenase; CatB: *cis, cis*-muconate cycloisomerase; CatC: muconolactone isomerase; *cis, cis*-MA: *cis, cis*-muconic acid; CoA: coenzyme A; COD: chemical oxygen demand; DCW: dry cell weight; DesA: O-demethylase; GA: gallic acid; PcaHG: protocatechuic acid 3,4-dioxygenase; PHA: polyhydroxyalkanoates; PraA: protocatechuic acid 2,3-dioxygenase; TCA: tricarboxylic acid; VDH: vanillin dehydrogenase; 3MGA: 3-O-methylgallate.

Authors' contributions

ZXX, PL, and MJJ wrote the manuscript. All authors read and approved the final manuscript.

Author details

¹ School of Environmental and Biological Engineering, Nanjing University of Science and Technology, Nanjing 210094, China. ² Nanjing Institute for Comprehensive Utilization of Wild Plants, Nanjing 211111, China.

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Competing interests

The authors declare that they have no competing interests.

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All data generated or analyzed during this study are included.

Consent for publication

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Not applicable.

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