

# Lawrence Berkeley National Laboratory

## Recent Work

### Title

REDETERMINATION AND ABSOLUTE CONFIGURATION OF SODIUM URANYL ACETATE

### Permalink

<https://escholarship.org/uc/item/70m6z3px>

### Author

Templeton, D.H.

### Publication Date

1985-04-01



# Lawrence Berkeley Laboratory

UNIVERSITY OF CALIFORNIA

## Materials & Molecular Research Division

RECEIVED  
LAWRENCE  
BERKELEY LABORATORY

MAY 16 1985

LIBRARY AND  
DOCUMENTS SECTION

Submitted to Acta Crystallographica, Section C

REDETERMINATION AND ABSOLUTE CONFIGURATION OF  
SODIUM URANYL ACETATE

D.H. Templeton, A. Zalkin, H. Ruben, and  
L.K. Templeton

April 1985

**TWO-WEEK LOAN COPY**

*This is a Library Circulating Copy  
which may be borrowed for two weeks*



LBL-19471  
c.2

## **DISCLAIMER**

This document was prepared as an account of work sponsored by the United States Government. While this document is believed to contain correct information, neither the United States Government nor any agency thereof, nor the Regents of the University of California, nor any of their employees, makes any warranty, express or implied, or assumes any legal responsibility for the accuracy, completeness, or usefulness of any information, apparatus, product, or process disclosed, or represents that its use would not infringe privately owned rights. Reference herein to any specific commercial product, process, or service by its trade name, trademark, manufacturer, or otherwise, does not necessarily constitute or imply its endorsement, recommendation, or favoring by the United States Government or any agency thereof, or the Regents of the University of California. The views and opinions of authors expressed herein do not necessarily state or reflect those of the United States Government or any agency thereof or the Regents of the University of California.

Redetermination and Absolute Configuration of Sodium Uranyl Acetate

By David H. Templeton, Allan Zalkin, Helena Ruben  
and Lieselotte K. Templeton

Materials and Molecular Research Division,  
Lawrence Berkeley Laboratory and  
Department of Chemistry, University of California,  
Berkeley, California 94720, USA

**Abstract.** Sodium tris(acetato)dioxouranate(1-),  $\text{NaUO}_2(\text{C}_2\text{H}_3\text{O}_2)_3$ ,  $M_r = 470.15$ , cubic,  $P2_13$ ,  $a = 10.689(2) \text{ \AA}$ ,  $V = 1221.3 \text{ \AA}^3$ ,  $Z = 4$ ,  $D_m = 2.562$ ,  $D_x = 2.557(2) \text{ g cm}^{-3}$ ,  $\text{Mo K}\alpha$ ,  $\lambda(\alpha_1) = 0.70930 \text{ \AA}$ ,  $\mu = 126.5 \text{ cm}^{-1}$ ,  $F(000) = 848$ ,  $T = 294 \text{ K}$ ,  $R = 0.021$  for 1195 unique reflections. A determination of the anomalous scattering term  $f''$  for uranium and Mo K $\alpha$  radiation gives  $9.7(2) \text{ e/atom}$ . Each uranyl ion [ $\text{U-O(av.)} = 1.758(3) \text{ \AA}$ ] lies on a 3-fold axis and is surrounded equatorially by six oxygen atoms of three acetate groups [ $\text{U-O(av.)} = 2.464(2) \text{ \AA}$ ]. The absolute configuration determined by the anomalous X-ray scattering is correlated with the sign of the optical activity.

**Introduction.** The structure of sodium uranyl acetate was studied by Fankuchen (1935) and was determined more completely by Zachariasen & Plettinger (1959). We studied it again to obtain more accurate parameters for use in the analysis of experiments with synchrotron radiation (Templeton & Templeton, 1982). The purpose of these experiments was to measure anomalous scattering terms and the effect of polarization on them.

Incidental to this work we determined that the absolute configuration of the structure, if described in a right-handed coordinate system with the atomic parameters reported by Zachariasen & Plettinger, is that of a crystal which rotates the plane of polarization of visible light to the left.

**Experimental.** Crystals were made by dissolving reagent-grade uranyl acetate and sodium acetate in water, followed by slow evaporation. Crystal with 18 faces of forms  $\{111\}$ ,  $\{11\bar{1}\}$  and  $\{110\}$ , 0.11 x 0.11 x 0.23 mm.  $D_m$  is taken from Fankuchen (1935). Picker FACS-I diffractometer, graphite monochromator,  $\theta$ - $2\theta$  scan; cell dimension from 12 reflections  $42^\circ < 2\theta < 48^\circ$ ; analytical absorption correction, range 2.70 to 3.64; max.  $(\sin\theta)/\lambda = 0.705 \text{ \AA}^{-1}$ ,  $h$  0 to 15,  $k$  -15 to 15,  $l$  -15 to 0; three standard reflections,  $\sigma = 1.3, 1.0, 1.4\%$ , no correction for decay; 3922 data, 1207 unique (including 12 observed less than background),  $R_{\text{int}}(I) = 0.030$ ; structure from Zachariasen & Plettinger (1959) refined on  $F$ , 62 parameters including  $f''$  of U, anisotropic thermal parameters for all atoms except H, H atoms found in  $\Delta F$  map and refined with isotropic thermal parameters and subject to restraints on H-H and C-H distances,  $R = 0.021$  for 1195 reflections,  $wR = 0.015$ ,  $S = 1.01$ ,  $w = [\sigma(F)]^{-2}$ , derived from  $\sigma^2(F^2) = [(\sigma^2(F^2))$ , counting statistics only,  $+ (0.014 F^2)^2]$ ; max  $\Delta/\sigma = 0.10$ ; max empirical isotropic correction for extinction 10% of  $F$ ; max. and min. of  $\Delta F$  synthesis 0.6 and  $-0.6 e \text{ \AA}^{-3}$ ; atomic  $f$  including dispersion for neutral U, Na, O, C and spherical bonded H from International Tables for X-ray Crystallography (1974); local unpublished programs and ORTEP (Johnson, 1965). Optical activity (positive rotation of 2 or 3°) was easy to observe in a well-formed crystal of thickness 1.5 mm with a polarizing microscope illuminated with white light. This crystal was too large for reliable observation of the signs of Bijvoet differences. A small fragment was broken from a corner.

Its diffraction intensities for several Bijvoet pairs, selected to be sensitive to configuration, were consistently reversed from those observed and calculated for the crystal used in the structure determination. Atomic parameters are listed in Table 1.\*

-----

\*Lists of structure factors, anisotropic thermal parameters and H-atom parameters have been deposited with the British Library Lending Division as Supplementary Publication No. (7 pp.). Copies may be obtained through The Executive Secretary, International Union of Crystallography, 5 Abbey Square, Chester CH1 2HU, England.

-----

**Discussion.** This work confirms, with about a 10-fold improvement of precision, the structure which was determined in projection by Zachariasen & Plettinger (1959). Each uranyl is complexed with six oxygen atoms from three acetate ions (Fig. 1 and Table 2), giving the uranium atom the hexagonal-bipyramidal coordination which is commonplace for uranyl salts. The thermal motion of the methyl carbon atom, C(2), is quite anisotropic, with principal rms amplitudes of 0.18, 0.21 and 0.38 Å. The largest amplitude is nearly perpendicular to the plane of the acetate ion. The other acetate atoms exhibit excess motion in the same direction, but to a lesser extent. This motion explains the short C(1)-C(2) bond length; after correction according to the riding model, it is quite normal (Table 2).

Zachariasen & Plettinger (1959) used the uranyl bond distance in this compound (1.71(4) Å) to revise upward to 1.70 Å the end point of a curve of uranium-oxygen bond lengths vs bond strength. The present results (ave. = 1.758(3) Å) are more in line with other recent values for "bond strength =

2.00" structures, and some further revision of the curve is suggested. A few examples are 1.75(2) Å in bis(tetrahydrofuran)dioxouranium(IV) nitrate (Reynolds, Zalkin & Templeton, 1977), 1.76(1) Å in pentakis(urea)dioxouranium(VI) nitrate (Zalkin, Ruben & Templeton, 1979), and 1.746(4) Å in rubidium uranyl nitrate (to be reported elsewhere). All these values are listed without any correction for thermal motion.

The chiral nature of the molecular environment in this crystal is evident in Fig. 1, which depicts the absolute configuration which produces negative optical rotation. To associate the present atomic coordinates with another specimen, two steps are required: correct designation of axes a, b, and c, and determination of the enantiomer. There are two distinct ways in which right-handed axes can be assigned to a specimen. They can be distinguished by the relative intensities of reflection pairs  $hk0$ ,  $h0k$  (e.g.,  $I(720)/I(702) = 52$  for our setting) which are independent of the enantiomer. This ratio is inverted for the other setting. Then the handedness can be determined by Bijvoet pairs such as  $I(712)/I(7,1,-2) = 1.58$ ,  $I(721)/I(7,2,-1) = 1.26$  (for coordinates of Table 1), or by observation of optical rotation.

This data set permits a good determination of  $f''$  for uranium at Mo  $K\alpha$ ; the result 9.7(2) e/atom is in agreement with 9.654 calculated by Cromer & Liberman (1970) for Mo  $K\alpha_1$ .

This work was supported by the National Science Foundation under grant No. CHE-8217443 and by the Director, Office of Energy Research, Office of Basic Energy Sciences, Chemical Sciences Division of the U.S. Department of Energy under Contract No. DE-AC03-76SF00098.

### References

- Cromer, D. T. & Liberman, D. (1970). *J. Chem. Phys.* **53**, 1891-1898.
- Fankuchen, I. (1935). *Z. Kristallogr.* **91**, 473-479.
- International Tables for X-ray Crystallography (1974). Vol. IV.  
Birmingham: Kynoch Press. (Present distributor D. Reidel, Dordrecht.)
- Johnson, C. K. (1965). ORTEP. Report ORNL-3794. Oak Ridge National  
Laboratory, Tennessee.
- Reynolds, J. G., Zalkin, A. & Templeton, D. H. (1977). *Inorg. Chem.* **16**,  
3357-3359.
- Templeton, D. H. & Templeton, L. K. (1982). *Acta Cryst.* **A38**, 62-67.
- Zachariasen, W. H. & Plettinger, H. A. (1959). *Acta Cryst.* **12**, 526-530.
- Zalkin, A., Ruben, H. W. & Templeton, D. H. (1979). *Inorg. Chem.* **18**,  
519-521.



Table 1. Fractional coordinates and equivalent isotropic thermal parameters (with e.s.d.'s in parentheses)

$$B_{eq} = \frac{1}{3} \sum_i \sum_j B_{ij} a_i^* a_j^* a_i \cdot a_j$$

	x	y	z	$B_{eq} (\text{\AA}^2)$
U	.42935(1)	.42935	.42935	1.89
Na	.82865(12)	.82865	.82865	2.35
O(1)	.3343(2)	.3343	.3343	3.33
O(2)	.5242(2)	.5242	.5242	3.07
O(3)	.3834(2)	.2945(2)	.6110(2)	3.05
O(4)	.5464(2)	.2443(2)	.5007(2)	2.97
C(1)	.4786(4)	.2260(3)	.5950(3)	2.95
C(2)	.5088(6)	.1240(6)	.6862(6)	5.68

Table 2. Bond distances (Å) and angles (°)

U-O(1)	1.759(4)	1.774*	O(1)-U-O(2)	180
U-O(2)	1.756(4)	1.769*	O(1)-U-O(3)	90.12(6)
U-3 O(3)	2.467(2)	2.475*	O(1)-U-O(4)	90.48(7)
U-3 O(4)	2.462(2)	2.469*	O(3)-U-O(4)	52.24(8)
C(1)-O(3)	1.264(4)	1.268*	O(3)-C(1)-O(4)	118.9(3)
C(1)-O(4)	1.256(4)	1.258*	O(3)-C(1)-C(2)	120.4(3)
C(1)-C(2)	1.498(5)	1.530*	O(4)-C(1)-C(2)	120.7(4)
Na-3 O(3)	2.385(3)		U-C(1)-C(2)	176.8(4)
Na-3 O(4)	2.403(3)			

\*Corrected for thermal motion according to the riding model.

Figure Caption:

Fig. 1. View of the structure down [111]. O(1) is below the central U atom, while O(2) and Na are above it. The central molecule has three more neighbors above it, generated by unit translations of the three depicted below it; e.g., the upper-left molecule shifted by  $c$ . Symmetry code: i,  $1 - x, -1/2 + x, 1/2 - x$ ; ii,  $-1/2 + x, 1/2 - x, 1 - x$ .

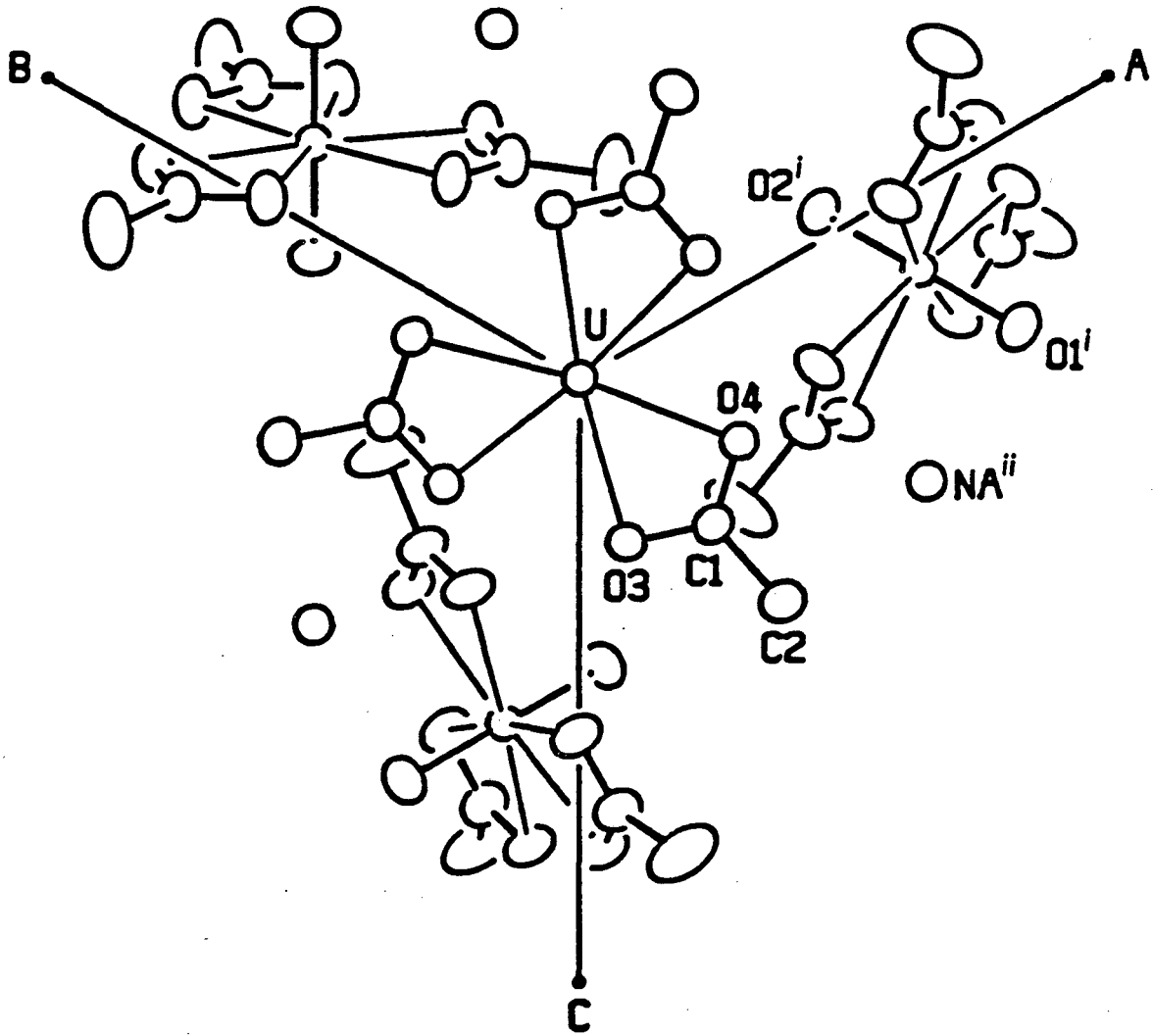


Fig. 1

Supplementary Material for:

Redetermination and Absolute Configuration of Sodium Uranyl Acetate

By David H. Templeton, Allan Zalkin, Helena Ruben  
and Lieselotte K. Templeton

Materials and Molecular Research Division,  
Lawrence Berkeley Laboratory and  
Department of Chemistry, University of California,  
Berkeley, California 94720, USA

**Abstract.** Sodium tris(acetato)dioxouranate(1-),  $\text{NaUO}_2(\text{C}_2\text{H}_3\text{O}_2)_3$ ,  $M_r = 470.15$ , cubic,  $P2_13$ ,  $a = 10.689(2)$  Å,  $V = 1221.3$  Å<sup>3</sup>,  $Z = 4$ ,  $D_m = 2.562$ ,  $D_x = 2.557(2)$  g cm<sup>-3</sup>, Mo K $\alpha$ ,  $\lambda(\alpha_1) = 0.70930$  Å,  $\mu = 126.5$  cm<sup>-1</sup>,  $F(000) = 848$ ,  $T = 294$  K,  $R = 0.021$  for 1195 unique reflections. A determination of the anomalous scattering term  $f''$  for uranium and Mo K $\alpha$  radiation gives  $9.7(2)$  e/atom. Each uranyl ion [ $\text{U-O}(\text{av.}) = 1.758(3)$  Å] lies on a 3-fold axis and is surrounded equatorially by six oxygen atoms of three acetate groups [ $\text{U-O}(\text{av.}) = 2.464(2)$  Å]. The absolute configuration determined by the anomalous X-ray scattering is correlated with the sign of the optical activity.

Supplementary Table 1. Fractional coordinates and isotropic thermal parameters for hydrogen atoms in sodium uranyl acetate

	x	y	z	B(A <sup>2</sup> )
H(1)	.508(6)	.042(3)	.651(5)	11.7(27)
H(2)	.448(4)	.119(5)	.746(4)	8.1(16)
H(3)	.592(3)	.143(5)	.707(5)	8.8(19)

Note: These atoms were refined subject to restraints on distances,  
 C(1)-H = 2.06(7) Å,  
 C(2)-H = 0.95(3) Å,  
 H-H = 1.55(5) Å.

Supplementary Table 2. Anisotropic thermal parameters for sodium uranyl acetate

	B <sub>11</sub>	B <sub>22</sub>	B <sub>33</sub>	B <sub>12</sub>	B <sub>13</sub>	B <sub>23</sub>	B <sub>eq</sub> (A <sup>2</sup> )
U	1.892(5)	1.892	1.892	.071(4)	.071	.071	1.892(2)
Na	2.35(4)	2.35	2.35	.05(4)	.05	.05	2.35(1)
O(1)	3.33(9)	3.33	3.33	-.59(9)	-.59	-.59	3.33(3)
O(2)	3.07(9)	3.07	3.07	-.59(8)	-.59	-.59	3.07(3)
O(3)	2.9(1)	3.1(1)	3.1(1)	.71(9)	.95(9)	.88(9)	3.05(6)
O(4)	3.1(1)	3.2(1)	2.52(9)	1.02(9)	.67(9)	.78(8)	2.97(6)
C(1)	3.4(2)	2.7(1)	2.7(2)	.6(1)	.5(1)	.6(1)	2.95(9)
C(2)	6.0(3)	6.3(3)	4.7(3)	2.9(2)	2.2(2)	3.0(2)	5.7(2)

Supplementary Table 3. Observed structure factors, standard deviations, and deltas for sodium uranyl acetate (next pages)

OBSERVED STRUCTURE FACTORS, STANDARD DEVIATIONS, AND DIFFERENCES (ALL X 4.0)  
SODIUM URANYL ACETATE MOKA

F(0,0,0) = 3227 (on this scale, including effect of anomalous scattering)

FOB AND FCA ARE THE OBSERVED AND CALCULATED STRUCTURE FACTORS.

SG = ESTIMATED STANDARD DEVIATION OF FOB. DEL = |FOB| - |FCA|.

\* INDICATES ZERO WEIGHTED DATA (observed less than background).

L	FOB	SG	DEL	L	FOB	SG	DEL	L	FOB	SG	DEL	L	FOB	SG	DEL	L	FOB	SG	DEL
H,K= 1,	1			-3	816	6	-14	H,K= 5,	3			5	387	4	-4	2	308	3	-3
1	910	7	3	H,K= 4,	2			4	404	3	0	4	247	3	1	1	254	3	2
0	700	5	2	3	225	2	2	3	578	5	1	3	445	4	3	0	545	4	-3
-1	957	7	7	2	476	4	5	2	449	4	-1	2	346	3	-2	-1	266	3	2
H,K= 2,	0			1	541	4	2	1	278	3	0	1	373	3	-0	-2	285	3	-6
0	814	8	-8	0	152	2	7	0	336	3	2	0	518	4	8	-3	181	3	-2
-1	784	6	16	-1	514	4	-2	-1	294	3	4	-1	368	3	5	-4	200	3	1
H,K= 2,	1			-2	495	4	3	-2	438	3	-2	-2	350	3	-3	-5	222	3	-5
1	571	4	12	-3	254	2	0	-3	555	4	-5	-3	422	4	1	-6	250	5	1
0	428	3	7	H,K= 4,	3			-4	373	3	-0	-4	257	3	-3	H,K= 7,	0		
-1	525	4	8	3	263	2	4	H,K= 5,	4			-5	383	3	-1	-1	301	3	-1
H,K= 2,	2			2	633	5	10	4	130	3	-1	H,K= 6,	3			-2	79	3	-1
2	528	4	4	1	402	3	2	3	60	7	-2	5	365	3	2	-3	741	6	-10
1	653	7	-2	0	920	7	1	2	246	2	-2	4	185	3	3	-4	83	3	2
0	379	3	7	-1	388	3	3	1	608	5	0	3	138	3	1	-5	453	4	-4
-1	673	5	-4	-2	659	5	6	0	460	4	-10	2	281	3	1	-6	24	24	6
-2	456	4	-1	-3	276	2	2	-1	607	5	0	1	572	5	0	H,K= 7,	1		
H,K= 3,	0			H,K= 4,	4			-2	273	3	-3	0	452	4	-4	6	105	6	-6
-1	240	2	6	4	559	6	-3	-3	39	7	-1	-1	588	5	-4	5	364	3	-4
-2	652	5	-3	3	201	2	2	-4	112	4	1	-2	270	3	0	4	317	3	-2
H,K= 3,	1			2	735	6	-1	H,K= 5,	5			-3	147	2	-2	3	99	3	-3
2	574	4	5	1	276	3	4	5	291	4	-6	-4	151	3	-1	2	286	3	3
1	319	2	6	0	166	2	2	4	291	3	1	-5	377	3	-3	1	562	5	-5
0	685	5	1	-1	297	2	7	3	426	4	-2	H,K= 6,	4			0	22	24	3
-1	346	3	6	-2	735	5	-8	2	331	3	3	5	456	5	-4	-1	565	4	-6
-2	529	4	5	-3	171	2	-1	1	270	3	3	4	255	3	-3	-2	232	2	-2
H,K= 3,	2			-4	599	5	7	0	452	4	-6	3	546	5	1	-3	96	3	-2
2	594	4	8	H,K= 5,	0			-1	269	3	1	2	201	3	3	-4	309	3	-3
1	824	6	8	-1	133	2	0	-2	333	3	0	1	347	3	6	-5	359	3	-4
0	38	4	3	-2	586	4	-4	-3	447	4	-2	0	100	3	2	-6	115	4	-0
-1	796	6	2	-3	423	3	-1	-4	301	3	1	-1	319	3	2	H,K= 7,	2		
-2	587	4	3	-4	546	4	-4	-5	328	4	-8	-2	218	3	3	6	368	3	5
H,K= 3,	3			H,K= 5,	1			H,K= 6,	0			-3	535	4	-3	5	314	3	-2
3	941	7	-12	4	237	2	1	0	585	4	-4	-4	261	3	-5	4	58	6	6
2	163	2	-0	3	241	2	-0	-1	267	2	-5	-5	439	4	-6	3	418	4	-1
1	427	3	6	2	470	4	-2	-2	563	4	1	H,K= 6,	5			2	334	3	-2
0	84	2	-3	1	476	4	-7	-3	30	7	-5	5	200	3	0	1	211	2	-2
-1	464	3	5	0	656	5	1	-4	53	5	-2	4	360	4	-1	0	566	4	-3
-2	198	2	2	-1	490	4	-2	-5	257	3	-2	3	149	3	-1	-1	190	3	-4
-3	953	7	-1	-2	488	4	0	H,K= 6,	1			2	312	3	2	-2	320	3	1
H,K= 4,	0			-3	267	2	3	5	315	3	-2	1	478	4	-0	-3	440	4	1
0	110	2	-1	-4	247	2	2	4	540	5	-5	0	274	2	5	-4	79	4	1
-1	749	5	-10	H,K= 5,	2			3	354	3	1	-1	479	4	1	-5	274	3	2
-2	57	2	2	4	374	3	-3	2	487	4	5	-2	303	3	-2	-6	377	3	1
-3	165	2	3	3	395	3	2	1	327	3	2	-3	155	3	1	H,K= 7,	3		
H,K= 4,	1			2	334	3	-0	0	59	3	-1	-4	351	3	-4	6	286	3	1
3	866	6	-3	1	443	3	-5	-1	292	3	-1	-5	207	3	1	5	105	4	3
2	241	2	3	0	706	5	7	-2	508	4	2	H,K= 6,	6			4	562	5	-2
1	268	2	4	-1	457	3	-6	-3	384	3	1	6	272	4	5	3	53	6	5
0	434	3	-1	-2	311	3	-2	-4	538	4	-1	5	225	3	2	2	462	4	-0
-1	289	2	3	-3	374	3	1	-5	302	3	-5	4	202	3	0	1	107	3	6
-2	252	2	4	-4	366	3	-3	H,K= 6,	2			3	165	3	5	0	138	2	-5

STRUCTURE FACTORS CONTINUED FOR  
SODIUM URANYL ACETATE

MOKA

L	FOB	SG	DEL	L	FOB	SG	DEL	L	FOB	SG	DEL	L	FOB	SG	DEL	L	FOB	SG	DEL
-1	119	4	1	0	28	18	8	1	470	4	-3	-7	110	7	-2	-3	233	3	4
-2	463	4	-3	-1	459	4	-6	0	189	2	-1	H,K=	8,	7	-4	145	3	6	
-3	37	10	1	-2	62	8	-6	-1	487	4	-6	7	0	35	-14*	-5	199	4	-2
-4	565	5	-5	-3	126	4	3	-2	195	3	2	6	128	5	-7	-6	267	4	2
-5	91	5	-8	-4	52	12	6	-3	139	4	-8	5	116	5	2	-7	240	4	3
-6	256	3	-2	-5	261	3	5	-4	105	4	-3	4	346	4	3	-8	216	4	-3
	H,K=	7,	4	-6	40	14	19	-5	329	3	2	3	157	4	2	H,K=	9,	2	
6	442	4	-5	-7	357	5	1	-6	225	3	1	2	321	4	0	8	171	4	-1
5	110	4	2	H,K=	8,	0		-7	377	4	-3	1	93	6	-7	7	235	3	3
4	69	5	3	0	539	4	-1	H,K=	8,	4		0	0	28	-3*	6	244	3	-2
3	127	3	10	-1	274	2	-5	7	52	10	2	-1	106	7	-2	5	229	3	2
2	388	3	-2	-2	360	3	-1	6	147	4	7	-2	334	3	-3	4	204	3	-1
1	52	7	-5	-3	34	9	2	5	385	4	-4	-3	131	4	1	3	241	3	-0
0	677	5	-11	-4	177	3	1	4	175	3	3	-4	368	3	-2	2	273	3	3
-1	54	7	-2	-5	102	3	-7	3	444	4	4	-5	124	6	6	1	267	3	-0
-2	385	3	-1	-6	344	3	0	2	234	3	-3	-6	146	4	6	0	239	3	-4
-3	147	3	8	-7	152	3	-3	1	253	3	1	-7	33	28	16	-1	301	3	2
-4	95	4	-2	H,K=	8,	1		0	142	3	-2	H,K=	8,	8		-2	273	3	-1
-5	99	4	11	7	131	4	-2	-1	244	3	3	8	261	7	6	-3	239	3	-0
-6	443	4	-3	6	193	3	-0	-2	240	3	-1	7	126	7	3	-4	213	3	-1
	H,K=	7,	5	5	163	3	1	-3	425	4	2	6	235	4	0	-5	230	3	-4
6	158	3	14	4	522	5	-6	-4	188	3	3	5	112	6	-7	-6	243	4	-3
5	221	3	2	3	253	3	-2	-5	355	4	-8	4	76	8	-7	-7	239	3	2
4	418	4	3	2	407	4	-4	-6	148	3	2	3	153	4	-0	-8	180	4	5
3	30	19	-18	1	262	3	-2	-7	49	11	-4	2	210	4	1	H,K=	9,	3	
2	417	4	4	0	123	3	-2	H,K=	8,	5		1	149	5	5	8	99	8	7
1	339	3	-3	-1	240	3	0	7	286	4	1	0	355	3	1	7	82	12	6
0	71	5	2	-2	415	4	-3	6	132	5	4	-1	151	4	8	6	122	5	-1
-1	336	3	-3	-3	250	3	3	5	210	3	-1	-2	203	3	-5	5	251	4	-5
-2	392	3	-2	-4	526	5	-11	4	240	3	-4	-3	165	4	1	4	306	3	-1
-3	45	11	-3	-5	136	4	-3	3	194	3	2	-4	71	9	-2	3	360	4	3
-4	422	4	1	-6	211	3	-3	2	242	3	-1	-5	128	5	6	2	253	3	-3
-5	207	3	-4	-7	149	4	-5	1	391	4	0	-6	241	4	4	1	172	5	-1
-6	166	3	2	H,K=	8,	2		0	144	3	-2	-7	114	6	-4	0	64	5	-0
	H,K=	7,	6	7	133	4	1	-1	366	4	-4	-8	269	6	2	-1	185	3	2
6	200	4	-3	6	288	4	2	-2	237	3	-1	H,K=	9,	0		-2	241	5	-4
5	283	4	-2	5	336	4	-6	-3	196	3	-2	-1	187	2	1	-3	334	4	-3
4	51	12	-1	4	278	4	-2	-4	244	3	-2	-2	376	4	-6	-4	293	3	1
3	458	5	-2	3	402	4	1	-5	201	3	-0	-3	412	4	-4	-5	267	3	4
2	136	5	3	2	234	3	4	-6	149	4	8	-4	413	3	0	-6	115	6	-3
1	227	3	-7	1	202	3	-4	-7	282	3	-0	-5	180	3	-1	-7	78	7	-8
0	289	3	0	0	295	3	-7	H,K=	8,	6		-6	134	4	-1	-8	102	6	9
-1	229	3	-3	-1	197	3	0	7	128	5	0	-7	69	9	2	H,K=	9,	4	
-2	131	4	4	-2	260	3	-1	6	310	3	1	-8	145	5	0	8	190	4	2
-3	473	4	-4	-3	402	4	-3	5	190	3	-0	H,K=	9,	1		7	297	3	2
-4	64	9	-5	-4	259	4	-7	4	94	5	8	8	229	4	3	6	229	4	-2
-5	284	3	-4	-5	339	4	-4	3	184	4	1	7	244	4	2	5	201	4	1
-6	204	4	0	-6	292	4	-2	2	314	4	-6	6	251	3	0	4	99	6	-2
	H,K=	7,	7	-7	151	4	5	1	213	3	-0	5	193	4	1	3	94	5	1
7	369	8	-10	H,K=	8,	3		0	387	3	-0	4	123	4	-1	2	184	3	-0
6	38	16	22	7	391	4	-3	-1	192	3	-4	3	219	3	0	1	323	3	5
5	253	3	-3	6	200	4	-3	-2	315	3	1	2	309	3	-1	0	393	3	-4
4	46	15	-4	5	312	3	3	-3	192	4	-0	1	303	3	-3	-1	327	3	4
3	126	4	-3	4	115	4	-1	-4	71	7	-1	0	379	3	-1	-2	192	4	2
2	72	6	-0	3	146	4	2	-5	184	5	-2	-1	305	3	-7	-3	79	8	-4
1	453	4	-3	2	193	3	-0	-6	300	3	-1	-2	330	3	-1	-4	95	5	-2

STRUCTURE FACTORS CONTINUED FOR  
SODIUM URANYL ACETATE

MOKA

L	FOB	SG	DEL	L	FOB	SG	DEL	L	FOB	SG	DEL	L	FOB	SG	DEL	L	FOB	SG	DEL
-5	199	3	1	-7	202	4	-1	3	380	4	-1	7	76	11	-7	-9	156	5	7
-6	218	5	-4	-8	201	4	-2	2	133	4	0	6	157	4	5	H,K=	10,	7	
-7	278	4	-1	H,K=	9,	8		1	217	3	-6	5	110	6	-0	8	0	40	-19*
-8	188	5	-5	8	100	8	5	0	154	3	-0	4	400	4	-1	7	42	24	17
	H,K=	9,	5	7	89	8	4	-1	216	3	-2	3	107	5	7	6	51	17	2
8	168	5	1	6	125	6	6	-2	134	4	-2	2	276	3	1	5	250	4	-2
7	129	5	6	5	159	5	5	-3	369	4	-5	1	98	6	2	4	107	7	13
6	200	5	6	4	216	4	-4	-4	100	5	4	0	23	27	9	3	337	4	1
5	217	3	4	3	221	4	7	-5	259	3	-7	-1	93	6	2	2	85	8	-5
4	261	4	2	2	182	4	6	-6	145	6	-7	-2	290	3	-3	1	145	6	-1
3	284	3	-2	1	167	4	-4	-7	53	19	27	-3	89	6	-7	0	0	31	-29*
2	199	4	-2	0	82	7	5	-8	125	5	-4	-4	399	4	-5	-1	147	5	5
1	237	3	2	-1	168	4	1	-9	212	4	-2	-5	106	5	-2	-2	80	9	-5
0	256	3	-5	-2	177	4	-1	H,K=	10,	2		-6	153	6	-6	-3	326	4	-2
-1	234	3	-3	-3	217	4	-2	9	112	7	-1	-7	68	10	-10	-4	86	9	-9
-2	210	3	-2	-4	230	4	0	8	116	6	5	-8	130	7	-5	-5	258	5	-3
-3	289	3	-0	-5	151	5	-2	7	217	4	-0	-9	55	18	2	-6	62	14	6
-4	260	3	1	-6	117	6	3	6	150	7	-1	H,K=	10,	5		-7	35	42	13
-5	226	3	8	-7	96	11	1	5	169	4	3	9	121	7	6	-8	0	46	-22*
-6	193	4	5	-8	103	8	10	4	279	4	-3	8	213	4	3	H,K=	10,	8	
-7	120	6	-0	H,K=	9,	9		3	121	5	-0	7	80	9	10	7	222	5	8
-8	147	5	-4	8	114	7	4	2	343	3	2	6	219	3	-4	6	92	7	22
	H,K=	9,	6	7	129	6	11	1	320	4	-5	5	203	4	-1	5	162	5	0
8	126	9	1	6	117	7	-2	0	49	11	7	4	116	5	6	4	130	5	-3
7	123	9	-1	5	163	5	2	-1	320	3	-4	3	227	4	-2	3	87	8	-2
6	151	5	1	4	147	5	2	-2	346	4	-3	2	201	3	2	2	88	8	-3
5	242	3	-0	3	172	5	0	-3	126	4	-5	1	130	5	-2	1	248	5	2
4	243	4	-0	2	183	4	8	-4	262	4	-1	0	312	3	-0	0	90	5	7
3	232	4	-2	1	167	4	6	-5	163	4	-4	-1	139	4	4	-1	250	4	4
2	257	3	2	0	176	4	-1	-6	148	6	1	-2	195	4	7	-2	81	10	1
1	156	4	0	-1	159	5	-1	-7	229	5	0	-3	238	3	-2	-3	92	10	3
0	146	4	-2	-2	180	5	4	-8	128	6	5	-4	102	5	-2	-4	118	6	-4
-1	146	4	9	-3	175	5	3	-9	96	7	-11	-5	211	4	-6	-5	164	5	-0
-2	247	3	-1	-4	132	5	-5	H,K=	10,	3		-6	216	4	2	-6	62	13	0
-3	245	3	2	-5	151	5	-5	9	64	13	-9	-7	77	10	2	-7	210	4	-6
-4	258	4	-4	-6	128	6	8	8	282	4	1	-8	202	5	1	H,K=	10,	9	
-5	247	4	6	-7	125	6	12	7	104	6	14	-9	106	11	-4	6	167	5	2
-6	151	4	7	-8	109	7	-1	6	327	4	-1	H,K=	10,	6		5	123	7	6
-7	137	6	6	H,K=	10,	0		5	136	4	2	9	154	7	9	4	73	14	2
-8	129	8	3	0	100	5	3	4	129	5	2	8	50	19	2	3	188	7	10
	H,K=	9,	7	-1	456	4	-3	3	89	7	0	7	251	4	3	2	150	6	0
8	202	4	6	-2	155	3	-4	2	259	4	3	6	87	7	-0	1	110	7	4
7	198	4	-1	-3	167	3	-4	1	129	5	7	5	161	5	2	0	199	4	-3
6	240	4	5	-4	0	29	-7*	0	460	4	-2	4	163	5	-5	-1	117	6	2
5	130	5	-1	-5	187	3	-4	-1	96	9	2	3	59	13	-7	-2	147	5	5
4	66	10	-7	-6	63	8	18	-2	247	3	-2	2	148	4	-1	-3	189	5	5
3	76	10	14	-7	347	3	-1	-3	98	6	5	1	305	3	0	-4	64	13	-8
2	182	4	-4	-8	113	5	-1	-4	119	4	4	0	75	6	-4	-5	120	7	-0
1	240	4	1	-9	170	4	1	-5	139	7	-0	-1	291	4	-2	-6	164	5	-5
0	326	3	2	H,K=	10,	1		-6	324	4	-5	-2	155	4	8	H,K=	10,	10	
-1	237	3	2	9	209	5	-4	-7	100	8	-6	-3	70	11	0	5	68	13	-1
-2	174	3	2	8	135	5	3	-8	276	4	3	-4	159	5	-1	4	224	4	10
-3	78	7	9	7	33	38	-4	-9	80	8	7	-5	168	4	4	3	52	18	5
-4	44	16	-14	6	149	7	-7	H,K=	10,	4		-6	103	7	17	2	210	5	7
-5	132	5	0	5	275	3	4	9	60	16	7	-7	243	4	-6	1	60	15	-1
-6	237	4	-5	4	87	7	-5	8	130	7	-3	-8	49	17	-3	0	41	19	13



STRUCTURE FACTORS CONTINUED FOR  
SODIUM URANYL ACETATE

MOKA

L	FOB	SG	DEL	L	FOB	SG	DEL	L	FOB	SG	DEL	L	FOB	SG	DEL	L	FOB	SG	DEL
-1	71	11	5	-7	242	4	-0	-2	126	7	4	H,K= 11,	9	2	211	4	4		
-2	205	5	3	-8	70	10	-9	-3	274	4	-1	5	127	7	6	1	186	4	-4
-3	69	14	16	-9	147	5	13	-4	73	8	10	4	75	11	13	0	149	4	-7
-4	210	4	4	-10	154	6	2	-5	211	4	2	3	191	5	8	-1	192	4	-3
-5	86	9	14	H,K= 11,	3	-6	164	4	-0	2	112	8	8	-2	198	4	-4		
H,K= 11,	0	9	188	5	1	-7	31	44	-13	1	80	16	-2	-3	180	4	-5		
-1	101	4	6	8	52	24	0	-8	136	6	4	0	184	4	8	-4	207	6	5
-2	316	3	-3	7	48	21	24	-9	129	8	1	-1	97	10	11	-5	146	5	-0
-3	55	9	8	6	100	6	10	H,K= 11,	6	-2	92	8	-8	-6	157	5	-0		
-4	364	3	-4	5	268	4	-0	8	74	15	-5	-3	191	6	5	-7	139	5	3
-5	50	11	-2	4	63	12	11	7	127	7	11	-4	35	43	-21	-8	104	9	9
-6	149	4	4	3	348	4	1	6	112	6	-22	-5	131	8	1	H,K= 12,	3		
-7	46	14	31	2	110	5	1	5	95	10	-3	H,K= 11,	10	8	188	5	13		
-8	154	4	7	1	175	4	0	4	270	4	1	2	35	40	-20	7	136	6	-3
-9	57	11	23	0	109	4	2	3	60	11	-10	1	215	5	5	6	200	5	-0
-10	226	4	2	-1	173	4	-2	2	247	4	-1	0	0	36	-26*	5	128	5	4
H,K= 11,	1	-2	115	5	-5	1	123	5	7	-1	217	5	1	4	49	15	2		
10	79	17	2	-3	354	3	2	0	42	17	29	-2	48	20	-8	3	74	12	-8
9	82	15	-7	-4	47	13	-12	-1	140	6	10	H,K= 12,	0	2	178	4	-2		
8	246	4	-6	-5	269	3	2	-2	243	4	0	0	240	3	-0	1	207	4	3
7	66	11	27	-6	87	7	4	-3	77	13	8	-1	255	3	-3	0	305	3	-1
6	252	4	-2	-7	32	40	11	-4	270	4	0	-2	78	6	-5	-1	207	4	5
5	135	5	1	-8	61	12	19	-5	111	7	3	-3	59	11	0	-2	177	4	-2
4	54	14	3	-9	177	5	4	-6	126	6	-2	-4	48	12	8	-3	92	7	7
3	212	3	3	H,K= 11,	4	-7	125	6	9	-5	130	4	3	-4	59	12	4		
2	240	3	-5	9	136	8	-1	-8	78	12	2	-6	159	4	5	-5	115	6	-6
1	115	5	1	8	39	30	-19	H,K= 11,	7	-7	241	3	3	-6	205	4	0		
0	424	3	-7	7	277	4	-1	7	57	15	23	-8	164	4	5	-7	137	6	2
-1	116	6	-3	6	86	8	10	6	234	4	-2	-9	143	6	1	-8	177	5	8
-2	228	4	-7	5	218	4	-2	5	42	21	15	H,K= 12,	1	H,K= 12,	4				
-3	201	3	5	4	65	9	-6	4	72	10	10	9	136	7	3	8	43	25	-18
-4	65	8	20	3	96	6	-1	3	0	40	-16*	8	113	7	1	7	51	24	-6
-5	136	6	2	2	57	15	-4	2	182	4	-0	7	59	18	-5	6	94	8	-12
-6	255	5	1	1	330	4	1	1	39	26	-15	6	101	8	5	5	119	7	1
-7	62	15	13	0	67	8	0	0	290	3	4	5	192	4	1	4	247	4	-4
-8	255	4	0	-1	332	3	-1	-1	50	16	-13	4	177	5	3	3	179	5	0
-9	74	10	-11	-2	67	9	8	-2	190	4	-1	3	253	5	7	2	215	5	3
-10	67	12	-8	-3	98	6	5	-3	0	39	-18*	2	131	5	-7	1	111	8	9
H,K= 11,	2	-4	81	7	1	-4	63	11	-3	1	161	4	-0	0	49	12	6		
10	148	7	-4	-5	222	4	4	-5	54	17	19	0	131	5	-3	-1	95	9	-5
9	135	6	-7	-6	64	12	2	-6	236	4	3	-1	156	5	-3	-2	215	4	0
8	83	9	1	-7	289	4	0	-7	58	16	22	-2	131	6	-10	-3	187	4	-1
7	233	4	2	-8	34	36	-14	H,K= 11,	8	-3	232	4	4	-4	256	4	4		
6	114	7	-1	-9	139	7	1	6	103	7	6	-4	171	4	2	-5	106	9	-11
5	166	4	-2	H,K= 11,	5	5	70	11	3	-5	195	4	-1	-6	121	6	0		
4	203	4	0	9	135	7	2	4	255	5	3	-6	109	8	11	-7	70	10	20
3	16	33	-21	8	127	6	-0	3	48	18	-5	-7	61	16	4	-8	81	10	18
2	175	6	5	7	32	38	-9	2	221	4	10	-8	114	7	9	H,K= 12,	5		
1	311	3	-2	6	159	5	-1	1	108	7	3	-9	147	6	4	7	124	7	4
0	35	14	10	5	208	4	2	0	0	34	-27*	H,K= 12,	2	6	151	7	-10		
-1	314	3	-4	4	73	12	2	-1	112	8	3	8	87	11	0	5	141	6	4
-2	178	4	3	3	280	4	-0	-2	211	5	6	7	146	6	4	4	108	6	5
-3	35	20	-5	2	130	5	2	-3	58	14	11	6	156	6	3	3	150	5	2
-4	198	3	0	1	140	5	5	-4	255	4	6	5	154	5	3	2	150	7	-4
-5	162	4	5	0	187	4	-5	-5	64	20	-6	4	212	4	-3	1	123	6	3
-6	98	11	-12	-1	137	4	-3	-6	112	7	17	3	176	4	-7	0	185	4	2

STRUCTURE FACTORS CONTINUED FOR  
SODIUM URANYL ACETATE

MOKA

L	FOB	SG	DEL	L	FOB	SG	DEL	L	FOB	SG	DEL	L	FOB	SG	DEL	L	FOB	SG	DEL
-1	129	5	3	7	184	5	-1	-2	176	4	3	4	34	46	-6				
-2	155	5	1	6	117	7	-5	-3	46	17	1	3	192	5	1				
-3	145	5	8	5	152	5	1	-4	66	12	3	2	109	7	9				
-4	112	6	10	4	112	7	1	-5	76	16	12	1	82	9	-0				
-5	142	5	-0	3	73	9	-4	-6	174	5	2	0	183	4	2				
-6	163	6	-5	2	146	5	4		H,K= 13,	5		-1	100	7	8				
-7	122	7	4	1	209	4	2	5	121	7	-3	-2	98	7	11				
	H,K= 12,	6		0	152	4	1	4	165	8	-4	-3	201	4	-0				
6	147	7	6	-1	213	4	-2	3	114	7	3	-4	46	24	8				
5	125	6	-6	-2	134	5	-5	2	151	6	5	-5	129	8	-1				
4	108	7	-5	-3	91	10	12	1	131	6	-3		H,K= 14,	3					
3	89	8	5	-4	127	5	10	0	91	6	-1	4	77	10	8				
2	125	6	-1	-5	160	5	5	-1	141	8	8	3	61	22	3				
1	183	5	0	-6	121	7	7	-2	147	6	0	2	51	19	9				
0	107	5	-1	-7	192	4	12	-3	122	7	10	1	225	5	-6				
-1	179	6	-1		H,K= 13,	2		-4	176	5	1	0	32	33	22				
-2	125	6	-4	7	98	9	-3	-5	121	8	-1	-1	230	5	7				
-3	100	7	10	6	158	7	6		H,K= 13,	6		-2	52	16	9				
-4	119	6	1	5	139	8	-13	4	115	7	5	-3	65	13	3				
-5	133	6	-1	4	108	7	2	3	185	6	1	-4	67	12	-1				
-6	143	6	10	3	159	5	0	2	90	9	-13		H,K= 14,	4					
	H,K= 12,	7		2	158	5	-0	1	95	10	0	3	227	4	1				
5	160	5	-1	1	108	9	2	0	89	7	-4	2	66	12	9				
4	153	7	3	0	202	3	-3	-1	101	12	5	1	79	11	-4				
3	213	5	3	-1	121	6	9	-2	105	7	2	0	69	9	9				
2	135	6	-3	-2	165	5	-2	-3	183	6	1	-1	90	8	11				
1	100	10	7	-3	144	6	-1	-4	106	9	-3	-2	59	19	2				
0	31	36	28	-4	109	6	2		H,K= 13,	7		-3	222	4	-3				
-1	83	9	-4	-5	151	6	-2	3	76	12	23		H,K= 14,	5					
-2	126	6	-4	-6	150	7	-2	2	72	13	-6	2	107	10	7				
-3	209	4	-2	-7	93	9	-6	1	203	5	8	1	178	5	10				
-4	159	5	13		H,K= 13,	3		0	136	5	4	0	34	36	26				
-5	149	6	3	7	43	26	-4	-1	187	5	1	-1	183	5	8				
	H,K= 12,	8		6	83	12	-6	-2	95	8	13	-2	99	9	4				
4	52	30	-12	5	116	8	4	-3	54	18	2		H,K= 15,	0					
3	88	9	7	4	243	4	3		H,K= 14,	0		-1	78	10	-3				
2	75	12	-11	3	108	7	-7	0	236	5	-4		H,K= 15,	1					
1	191	5	4	2	197	4	4	-1	32	35	0	1	182	6	-0				
0	144	6	-2	1	85	11	7	-2	165	6	1	0	65	11	-5				
-1	176	5	-7	0	74	8	7	-3	0	33	-7*	-1	171	5	-7				
-2	83	10	-8	-1	87	8	11	-4	40	21	-6								
-3	93	9	13	-2	180	4	-4	-5	0	36	-9*								
-4	80	10	14	-3	124	6	1		H,K= 14,	1									
	H,K= 12,	9		-4	242	4	2	5	74	13	5								
1	116	8	8	-5	118	9	0	4	220	4	-5								
0	144	5	1	-6	113	7	11	3	51	17	11								
-1	106	8	-9	-7	73	13	27	2	202	5	5								
	H,K= 13,	0			H,K= 13,	4		1	84	9	1								
-1	131	4	0	6	165	8	-7	0	30	34	28								
-2	141	4	-1	5	54	19	-13	-1	78	12	-7								
-3	257	3	2	4	54	16	-3	-2	205	5	2								
-4	136	5	1	3	50	18	-5	-3	53	16	8								
-5	198	4	1	2	183	4	13	-4	233	6	5								
-6	56	14	-4	1	119	6	-4	-5	78	14	6								
-7	41	20	37	0	246	3	-0		H,K= 14,	2									
	H,K= 13,	1		-1	132	6	2	5	122	8	-5								

This report was done with support from the Department of Energy. Any conclusions or opinions expressed in this report represent solely those of the author(s) and not necessarily those of The Regents of the University of California, the Lawrence Berkeley Laboratory or the Department of Energy.

Reference to a company or product name does not imply approval or recommendation of the product by the University of California or the U.S. Department of Energy to the exclusion of others that may be suitable.

TECHNICAL INFORMATION DEPARTMENT  
LAWRENCE BERKELEY LABORATORY  
UNIVERSITY OF CALIFORNIA  
BERKELEY, CALIFORNIA 94720