Electronic Supplementary Information

Removal of toxic Cr (VI) by UV-active functionalized graphene oxide for water purification

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Scheme S1. Schematic diagram of the mechanism for functionalization of graphene oxide.



Fig. S2 Low range XPS spectra of (a) GO and (b) DAP-RGO composite. Appearance of nitrogen peak and largely reduced oxygen peak indicates that GO is largely reduced to DAP-RGO composite.



Fig.S3 (a) UV-Vis absorption spectra for different Cr (VI) concentrations after 40 mins of adsorption at pH 1, (b) 500 mg/L Cr (VI) soln. with different adsorbent doses after 180 min of adsorption at pH 1. (c) 500 mg/L Cr (VI) soln. at different pH after 6h of adsorption with a fixed adsorbent dose 1 g/L.



Fig. S4 The pseudo-first-order kinetics for Cr (VI) adsorption on DAP-RGO composite



Fig. S5 TEM images of DAP-RGO composite of (a) before and (c) after Cr adsorption. EDX spectra and atomic% of DAP-RGO composite (b) before and (d) after Cr adsorption.

Table S1: Standard Deviation (±) data of different Cr (VI) concentrations

 50
 .0559
 .0334
 .0003

 mg/L
 .0581
 .0412
 .0443
 .0503
 0

 mg/L
 .0581
 .0412
 .0443
 .0503
 0

 200
 .1112
 .1428
 .0447
 .0723
 .0634
 .0448
 .0008

 mg/L
 .0381
 .0443
 .0533
 .0836
 .0559
 .1145
 .0919
 .0403
 .0558
 .0073
 .0005

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 1.5g/L .0379 .0251 .0042 0

 1g/L .0443 .0837 .0919 .0404 .0559 .0073 .0005

 .75g/L .5089 .1931 .3026 .0772 .0750 .1828 .0345 .0264 .0007

 .5g/L .3742 1.1728 .9015 .2038 .8962 .0658 .0992 .1156 .1133 .0680 .0411 .0009

Table 55: Standa	ard Deviation	(\pm) data of Cr (v I) at different	рн	
With UV at pH5	.20179	.14832	.07905	.00045	
Without UV at pH 5	.22829	.59582	.55946	.42778	
With UV at pH 4	.63319	.09711	.02886	.0010	
Without UV at pH 4	.49295	.5174	.12787	.14612	

Table S3: Standard Deviation (±) data of Cr (VI) at different pH