



Heriot-Watt University
Research Gateway

Repeated enrichment of trace metals and organic carbon on an Eocene high-energy shelf caused by anoxia and reworking

Citation for published version:

März, C, Wagner, T, Aqleh, S, Al-Alaween, M, van den Boorn, S, Podlaha, OG, Kolonic, S, Poulton, SW, Schnetger, B & Brumsack, H-J 2017, 'Repeated enrichment of trace metals and organic carbon on an Eocene high-energy shelf caused by anoxia and reworking', *Geology*, vol. 44, no. 12, pp. 1011-1014. <https://doi.org/10.1130/G38412.1>

Digital Object Identifier (DOI):

[10.1130/G38412.1](https://doi.org/10.1130/G38412.1)

Link:

[Link to publication record in Heriot-Watt Research Portal](#)

Document Version:

Peer reviewed version

Published In:

Geology

General rights

Copyright for the publications made accessible via Heriot-Watt Research Portal is retained by the author(s) and / or other copyright owners and it is a condition of accessing these publications that users recognise and abide by the legal requirements associated with these rights.

Take down policy

Heriot-Watt University has made every reasonable effort to ensure that the content in Heriot-Watt Research Portal complies with UK legislation. If you believe that the public display of this file breaches copyright please contact open.access@hw.ac.uk providing details, and we will remove access to the work immediately and investigate your claim.

Publisher: GSA
Journal: GEOL: Geology
DOI:10.1130/G38412.1

1 Repeated enrichment of trace metals and organic carbon on
2 an Eocene high energy shelf caused by anoxia and
3 reworking

4 Christian März^{1,2*}, T. Wagner^{1,3} S. Aqleh¹, M. Al-Alaween¹, S. van den Boorn⁴, O.G.
5 Podlaha⁴, S. Kolonic⁵, S.W. Poulton², B. Schnetger⁶, and H.-J. Brumsack⁶

6 ¹*School of Civil Engineering and Geosciences, Newcastle University, UK*

7 ²*School of Earth and Environment, University of Leeds, UK*

8 ³*Lyell Centre, Heriot-Watt University, Edinburgh, UK*

9 ⁴*Shell Global Solutions International B.V., Projects and Technology, Rijswijk, The*
10 *Netherlands*

11 ⁵*Shell Petroleum Development Company of Nigeria Ltd, Nigeria*

12 ⁶*ICBM, University of Oldenburg, Germany*

13 *E-mail: c.maerz@leeds.ac.uk

14 **ABSTRACT**

15 Petroleum source rocks are strongly enriched in organic carbon (OC), and their
16 trace metal (TM) contents often reach low grade ore levels. The mechanisms leading to
17 these co-enrichments are important for understanding how extreme environmental
18 conditions support the formation of natural resources. We therefore studied organic-rich
19 Eocene marls and limestones (oil shale) from the central Jordan Amzaq-Hazra sub-basin,
20 part of a Cretaceous-Paleogene shelf system along the southern Neo-Tethys margin.
21 Geochemical analyses on two cores show highly dynamic depositional conditions,
22 consistent with sedimentological and micropaleontological observations. Maximum and

23 average contents in OC (~26 and ~10 wt%), S (~7 and ~2.4 wt%), P (~10 and ~2 wt%),
24 Mo (>400 and ~130 ppm), Cr (>500 and ~350 ppm), V (>1,600 and ~550 ppm) and Zn
25 (>3,800 and ~900 ppm) are exceptional, in particular without any indication of
26 hydrothermal or epigenetic processes. We propose a combination of two processes:
27 Physical reworking of OC- and metal-rich material from locally exposed Cretaceous-
28 Paleogene sediments (as supported by re-worked nannofossils); and high marine
29 productivity fueled by chemical remobilization of nutrients and metals on land that
30 sustained anoxic-sulfidic conditions. Burial of high-quality organic matter (Hydrogen
31 Index 600–700 mgHC/gOC) was related to strongly reducing conditions, punctuated by
32 only short-lived oxygenation events, and to excess H₂S promoting organic matter
33 sulphurisation. These processes likely caused the OC/TM co-enrichments in a high-
34 energy shallow marine setting that contradicts common models for black shale formation,
35 but may explain similar geochemical patterns in other black shales.

36 INTRODUCTION

37 Marine shales rich in organic carbon (OC) and trace metals (TMs) represent
38 extreme environmental conditions under perturbed biogeochemical cycles in the
39 geological past. High primary productivity and anoxic/sulfidic redox conditions are
40 particularly conducive to their formation (Brumsack, 1980; Demaison and Moore, 1980).
41 Metal enrichments reaching low ore grade in black shales (Anjum et al., 2012) are
42 typically explained by secondary enrichment via metal-rich hydrothermal fluids or brines
43 (Eugster, 1985; Coveney and Glascock, 1989; Lehmann et al., 2007). Understanding
44 these processes is important to gain insight into extreme states of Earth's climate system,
45 and the formation and potential location of hydrocarbon and metal resources. To address

46 the topic of paired OC/TM enrichments, we applied geochemical analyses coupled with
47 sedimentological observations to Eocene, thermally immature OC-rich sediments ('oil
48 shale') from two wells in central Jordan. This region was part of an extended paleo-shelf
49 system throughout the Cretaceous-Paleogene, with numerous tectonically induced sub-
50 basins (Alqudah et al., 2015) covering large parts of the Arabian-African margins of the
51 Neo-Tethys (Fig. 1).

52 **METHODS**

53 The oil shale interval in cores OS22 and OS23 from central Jordan is ~225 m
54 thick (Fig. 1). Lithological units A and C are heterogeneous marls, phosphates, cherts and
55 limestones; unit B is more homogenous with interbedded laminated marls (Ali Hussein et
56 al., 2014a). The oil shale succession is part of the Umm Rijam Chalk Limestone
57 Formation, deposited across the Early-Middle Eocene (Ypresian-Lutetian) Amzaq-Hazra
58 sub-basin (Alqudah et al., 2014a, 2015). 186 samples were selected for geochemical
59 analysis. Total sulfur (S), total organic carbon (OC) and carbonate (calculated as CaCO_3)
60 were determined by Leco combustion analysis. The Hydrogen Index (HI) was analyzed
61 with a Rock Eval 6 analyzer. X-Ray Fluorescence (XRF) analysis was used to quantify
62 major (Al, Fe, P) and trace (Cr, Mo, V, Zn, Zr) element contents. TM excess contents
63 ($\text{element}_{\text{xs}}$) (Brumsack, 2006) were calculated relative to element/Al ratios of average
64 shale (Wedepohl, 1991). Sequential Fe (Poulton and Canfield, 2005) and chromium-
65 reducible S (CRS) extractions (Canfield et al., 1986) were used to calculate the ratios of
66 highly reactive to total Fe (FeHR/FeT), and of sulfide-bound to highly reactive Fe
67 (FeS/FeHR). Data are displayed against "adjusted depth" with zero depth (0 m) defined

68 as top of the oil shale interval. As both cores exhibit the same geochemical patterns, only
69 OS23 data are displayed in Figure 2 (for OS22, see Electronic Supplement).

70 **RESULTS**

71 The cores are subdivided into geochemical Units I to V (Fig. 2). Units I+II, III
72 and IV+V correlate with lithological units A, B and C (Ali Hussein et al., 2014a). Units II
73 and IV are geochemically more variable than the rather uniform Units I, III and V. The
74 sediment is dominantly composed of CaCO₃ (av. ~60 wt%) and TOC (av. ~10 wt%) with
75 high amplitude and high frequency variability (Fig. 2). Both S and P are enriched in parts
76 of the records (up to ~7 and ~10 wt%, respectively), with S showing a very strong
77 relationship with OC ($R^2 > 0.95$), while Al contents of ~0.1 to ~3.6 wt% (av. ~1.0 wt%)
78 indicate very little fine-grained siliciclastic material (Fig. 2A). Throughout the record,
79 OC is of excellent quality (kerogen type I/II), with Hydrogen Indices (HI) of 600–700
80 mgHC/gOC that do not match the OC fluctuations (Fig. 2A). Extreme enrichments of
81 Mo, Cr, V and Zn occur in Units II and IV (Fig. 2B), and to a lesser extent in Units I, III
82 and V, as evident from average excess contents (Table 1, Fig. DR1). Iron and S
83 speciation data indicate that > 80% of the S is bound to OC (Fig. 2B). High FeHR/FeT
84 ratios > 0.38 suggest anoxic bottom waters throughout the section, while fluctuations in
85 FeS/FeHR ratios around 0.7 indicate oscillations between sulfidic and non-sulfidic water
86 column conditions (Poulton and Canfield, 2011).

87 **DISCUSSION**

88 Our data provide intriguing insights into the depositional environment of the
89 Jordan oil shales. Units I, III and V tend to have FeS/FeHR ratios below 0.7 (Fig. 2B).
90 However, the observed ratios of ~0.6 are nevertheless quite high (Poulton and Canfield,

91 2011) and, when considered alongside significant TM enrichments (Figs. 2A, DR1, Table
92 1), indicate an anoxic water column during deposition of Units III and V, likely
93 fluctuating between sulfidic and non-sulfidic conditions. With no evidence for oxic
94 conditions, these laminated and rather homogenous Units likely document quiet
95 sedimentation under stable anoxia. This is supported by negative correlations between
96 OC and biogenic CaCO₃ in Unit III in both cores ($R^2 = 0.48$ and 0.76), implying dilution
97 of organic material by biogenic carbonate under steady hemi-pelagic sedimentation
98 (Alqudah et al., 2014a, b).

99 In contrast, Units II and especially IV show very different characteristics (Fig. 2).
100 High TM enrichments, in particular Mo, combined with Fe and S speciation data strongly
101 argue for at least episodically sulfidic bottom waters (Brumsack, 1980, 2006; Poulton and
102 Canfield, 2011). In support of this, high amounts of non-sulphide S and the occurrence of
103 Mo/Al peaks at maximum OC (Fig. 2) suggest that kerogen type I/II preservation was
104 favored by OM sulfurization due to excess H₂S availability relative to highly reactive Fe
105 (Tribovillard et al., 2004, 2015). However, the TM records (Fig. 2B) also suggest
106 significant redox variability during deposition of Units II and IV. Peaks in Zr and P (Fig.
107 2B) indicate a dynamic setting with episodically high depositional energies, probably
108 induced by currents and/or wave action enriching heavy minerals like zircon and apatite
109 (März et al., 2011). In agreement with convoluted and tilted bedding, tempestites, and
110 intensely bioturbated horizons in Units II and IV, this suggests deposition above the
111 storm or even fair weather wave base (Ali Hussein et al., 2014a, b). These observations
112 suggest extreme redox variability, from sulfidic water column conditions through to high-
113 energy oxic periods with mixing and ventilation. Such conditions are commonly regarded

114 as unfavorable for the preservation of high-quality kerogen, as ventilation should lead to
115 partial OC degradation (Demaison and Moore, 1980; Hedges and Keil, 1995).

116 Within this context it is notable that excess Mo, U, V, Zn, Ni and Cr for Units II,
117 III and IV exceed those of modern organic-rich deposits (the permanently sulfidic Black
118 Sea, and the episodically sulfidic Peru upwelling area. They are, however, comparable to
119 Cretaceous Oceanic Anoxic Event 2 sediments documenting widespread, long-lasting
120 anoxia/euxinia (Brumsack, 2006), and to the Cretaceous-Paleogene Belqa Group deposits
121 (Table 1, Fig. DR1) exposed near the coring sites in Jordan (Fleurance et al., 2013).
122 Finally, the Eocene TM patterns show similarities to metalliferous Cambrian (Lehmann
123 et al., 2007) and Carboniferous shales (Coveney and Glascock, 1989; Slack et al., 2015)
124 (Fig. DR1). For the latter, syn- or epigenetic hydrothermal input has been suggested as
125 the dominant TM source (Coveney and Glascock, 1989). Lithological and geochemical
126 properties of the Jordan shale (e.g., well-preserved sedimentary features, low thermal
127 maturity), however, support neither a sedimentary exhalative (SEDEX-type) mechanism
128 nor any post-depositional, epigenetic intrusion of metal-rich solutions (see Electronic
129 Supplement).

130 We instead propose that there were two key factors for TM enrichments and for
131 high quality OC production and preservation in the Eocene Jordan shales (Fig. 3): (1) The
132 water column was anoxic to euxinic during deposition of Units II, III and IV due to high
133 primary productivity and semi-restriction of the depositional basins (Demaison and
134 Moore, 1980; Alqudah et al., 2014a, b). (2) Different from interpretations of most black
135 shale deposits, the Eocene Jordan shale received material physically eroded and/or
136 chemically weathered from OC- and TM-rich Cretaceous-Paleocene limestones, marls

137 and shales exposed in proximal/transitional settings close to the study sites (Fleurance et
138 al., 2013). During the Cretaceous to Eocene, active tectonic processes (Alqudah et al.,
139 2014 a, b) favored re-distribution of erosional products from older strata across the shelf,
140 especially during periods of low relative sea level and high depositional energy, where
141 this material was mixed with autochthonous deposits. Alternative to, or in combination
142 with, physical re-distribution, chemical weathering of local OM- and TM-rich rocks
143 could have provided excess nutrients to the Eocene basin, enhancing primary productivity
144 particularly under the warm and humid Eocene climate (Pearson et al., 2007).

145 Evidence for physical erosion comes from the high abundance (sometimes
146 dominance) of re-worked Cretaceous to Eocene nannofossils in the cores studied
147 (Alqudah et al., 2014a, b) (Fig. 1). Localized tectonic processes causing syn-sedimentary
148 uplift and exposure of older sediments along the horst structures bordering the Amzaq-
149 Hazra sub-basin supported intense current- or wave-driven erosion events (Alqudah et al.,
150 2014a, b). This process would have resulted in re-deposition of older OC- and TM-rich
151 material, accompanied by drawdown of water column TMs and preservation of OC under
152 high primary productivity and sulfidic water column conditions (Fig. 3). Chemical
153 weathering and/or physical erosion of OC-, TM- and nutrient-rich lithologies exposed on
154 land might have contributed to increased nutrient supply, higher productivity, and
155 ultimately the OC/TM co-enrichment in the Eocene shale basins. However, weathering
156 and transport under oxic conditions would have partly degraded the organic matter before
157 it entered the anoxic/sulfidic shelf waters, which is inconsistent with the high kerogen
158 quality throughout the studied sediments. We therefore assume that input from land was

159 secondary to wave- and current-induced re-distribution processes on an anoxic/sulfidic
160 shelf (Baird and Brett, 1991).

161 Irrespective of the transport pathways of eroded material, reworking of
162 Cretaceous-Paleocene strata was variable but generally persistent throughout large parts
163 of the Jordan oil shale deposition (Alqudah et al., 2014a) (Fig. 1). The low-energy
164 conditions prevailing during deposition of Unit III should have favored lateral advection
165 of fine and fragile calcareous coccoliths, clay minerals, and associated OM. The lower
166 percentage of reworked nannofossils in parts of Units II and IV does not exclude that
167 erosion and reworking took place, but suggests that the fine material was winnowed
168 away, leaving only coarser/denser particles (rich in Zr, P and Cr) to settle and accumulate
169 (Fig. 3). It appears reasonable that a significant fraction of OC and TM in the sediment
170 may derive from this hydrodynamically sorted material. Similar processes have been
171 described for Paleozoic black shales (Baird and Brett, 1991; Formolo and Lyons, 2007),
172 highlighting the potential role of submarine erosion and re-working on anoxic high-
173 energy shelves in generating TM/OC enrichments. The variability in current and wave
174 intensity and continental export was probably linked to Eocene subtropical trade wind
175 dynamics, with alternations between periods of aridity (stronger winds, shelf reworking)
176 and humidity (weaker winds, continental export) (Huber and Goldner, 2012).

177 While this study describes environmental conditions and reworking processes for
178 one Eocene sub-basin of the Neo-Tethys shelf, it also confirms that OC- and TM-
179 enriched intervals are tens of meters thick in two cores drilled ~10 km apart, with
180 synchronous oil shale deposits spreading over ~100 km across the Amzaq-Hazra sub-
181 basin (Alqudah et al., 2015). We suggest that similar processes were active across large

182 parts of the Eocene Neo-Tethys shelf, and may also have operated during other periods of
183 widespread black shale deposition that occurred under similar environmental and tectonic
184 conditions. A notable example of OC/TM co-enrichment comparable to the Jordan shale
185 is a ~7 m thick Eocene black shale succession from the Central Arctic Lomonosov Ridge.
186 This interval is characterized by variable but partly high enrichments in OC, S and metals
187 (including Fe, Cr, Mo, Zn), indicative of at least episodically anoxic/sulfidic bottom
188 waters (Sangiorgi et al., 2008; März et al., 2011). But geochemical and mineralogical
189 proxies also support a dynamic setting with high depositional energies and very low
190 sedimentation rates in shallow waters (März et al., 2011). By analogy to the Jordan
191 setting, we suggest that chemical re-mobilization and/or physical erosion and reworking
192 of Cretaceous-Paleocene black shales on the Lomonosov Ridge contributed to the
193 observed element records. Black shale deposits from Mesozoic shallow water systems
194 also show features consistent with those reported from the Eocene Jordan shelf.
195 Petrographic and geochemical analyses of lower Toarcian black shales of the Posidonia
196 Shale Formation in the Dutch Central Graben (Trabucho-Alexandre et al., 2012), for
197 example, support dynamic energetic conditions driven by intense and variable bottom
198 currents as key controls on shale deposition, rather than vertical settling alone.

199 **CONCLUSIONS**

200 Based on geochemical data from two Eocene oil shale intervals in Jordan, we
201 present a model explaining high OC and TM enrichments in a shelf basin along the
202 southern margin of the Western Tethys. Critical factors are (1) overall high marine
203 productivity promoting anoxic/sulfidic water column conditions and OM sulphurisation,
204 and (2) the re-distribution of OM- and TM-rich deposits of Cretaceous-Paleogene age

205 from the inner shelf and hinterland. These factors, with a dominance of current- and
206 wave-induced reworking on the shelf, were strongly favored by the prevailing tectonic
207 regime, including small-scale extensional tectonics that generated semi-restricted shelf
208 basins in graben structures. Similar processes were likely active across large parts of the
209 Cretaceous-Paleogene Levantine shelf basins, and they should also be considered for
210 widespread shale deposits that formed under comparable paleo-environmental conditions
211 in tectonically active settings. This study challenges the common perception of stagnant
212 ('non-dynamic') depositional conditions during marine black shale formation. Instead, it
213 shifts the focus to dynamic, high-energy and tectonically active paleo-settings where
214 multiple climatic, depositional and tectonic processes overlap to produce extreme
215 geochemical shale properties. With this scope, our study also provides links to low-grade
216 ore resources generated without any syn- or post-depositional hydrothermal activity.

217 **ACKNOWLEDGMENTS**

218 Substantial financial and analytical support for this project from Shell Global
219 Solutions and JOSCo is gratefully acknowledged. We are indebted to C. Lehnert, E.
220 Gruendken and for P. Green for assistance with sample preparation and analysis, and to
221 M. Ali Hussein and M. Alqudah for fruitful discussions. N. Tribovillard, P. Meyers and
222 an anonymous reviewer provided insightful comments on an earlier version of the
223 manuscript.

224 **REFERENCES CITED**

225 Ali Hussein, M., Alqudah, M., Van den Boorn, S., Kolonic, S., Podlaha, O.G., and
226 Mutterlose, J., 2014a, Eocene oil shales from Jordan – their petrography, carbon and
227 oxygen stable isotopes: *GeoArabia*, v. 19, p. 139–162.

- 228 Ali Hussein, M., Alqudah, M., Podlaha, O.G., Van den Boorn, S., Kolonic, S., and
229 Mutterlose, J., 2014b, Ichnofabrics of Eocene oil shales from central Jordan and their
230 use for paleoenvironmental reconstructions: *GeoArabia*, v. 19, p. 145–160.
- 231 Alqudah, M., Ali Hussein, M., Podlaha, O.G., Van den Boorn, S., Kolonic, S., and
232 Mutterlose, J., 2014a, Calcareous nannofossil biostratigraphy of Eocene oil shales
233 from central Jordan: *GeoArabia*, v. 19, p. 117–140.
- 234 Alqudah, M., Ali Hussein, M., Van den Boorn, S., Giraldo, V.M., Kolonic, S., Podlaha,
235 O.G., and Mutterlose, J., 2014b, Eocene oil shales from Jordan - paleoenvironmental
236 implications from reworked microfossils: *Marine and Petroleum Geology*, v. 52,
237 p. 93–106, doi:10.1016/j.marpetgeo.2014.02.001.
- 238 Alqudah, M., Ali Hussein, M., Van den Boorn, S., Podlaha, O.G., and Mutterlose, J.,
239 2015, Biostratigraphy and depositional setting of Maastrichtian e Eocene oil shales
240 from Jordan: *Marine and Petroleum Geology*, v. 60, p. 87–104.
- 241 Anjum, F., Shahid, M., and Akcil, A., 2012, Biohydrometallurgy techniques of low grade
242 ores: a review on black shales: *Hydrometallurgy*, v. 117–118, p. 1–12,
243 doi:10.1016/j.hydromet.2012.01.007.
- 244 Baird, G.C., and Brett, C.E., 1991, Submarine erosion on the anoxic sea floor:
245 Stratigraphic, paleoenvironmental, and temporal significance of reworked pyrite-bone
246 deposits, *in* Tyson, R.V., and Pearson, T.H., eds., *Modern and Ancient Continental*
247 *Shelf Anoxia: Geological Society Special Publication 58*, p. 233–257,
248 doi:10.1144/GSL.SP.1991.058.01.16.

- 249 Brumsack, H.-J., 1980, Geochemistry of Cretaceous black shales from the Atlantic Ocean
250 (DSDP Legs 11, 14, 36 and 41): *Chemical Geology*, v. 31, p. 1–25,
251 doi:10.1016/0009-2541(80)90064-9.
- 252 Brumsack, H.-J., 2006, The trace metal content of recent organic carbon-rich sediments:
253 implication for Cretaceous black shale formation: *Palaeogeography*,
254 *Palaeoclimatology*, *Palaeoecology*, v. 232, p. 344–361,
255 doi:10.1016/j.palaeo.2005.05.011.
- 256 Canfield, D.E., Raiswell, R., Westrich, J.T., Reaves, C.M., and Berner, R.A., 1986, The
257 use of chromium reduction in the analysis of reduced inorganic sulphur in sediments
258 and shales: *Chemical Geology*, v. 54, p. 149–155, doi:10.1016/0009-2541(86)90078-
259 1.
- 260 Coveney, R.M., and Glascock, M.D., 1989, A review of the origins of metal-rich
261 Pennsylvanian black shales, central U.S.A., with an inferred role for basinal brines:
262 *Applied Geochemistry*, v. 4, p. 347–367, doi:10.1016/0883-2927(89)90012-7.
- 263 Demaison, G.J., and Moore, G.T., 1980, Anoxic environments and oil source bed
264 genesis: *Organic Geochemistry*, v. 2, p. 9–31, doi:10.1016/0146-6380(80)90017-0.
- 265 Eugster, H.P., 1985, Oil shales, evaporates and ore deposits: *Geochimica et*
266 *Cosmochimica Acta*, v. 49, p. 619–635, doi:10.1016/0016-7037(85)90158-9.
- 267 Fleurance, S., Cuney, M., Malartre, F., and Reyx, J., 2013, Origin of the extreme
268 polymetallic enrichment (Cd, Cr, Mo, Ni, U, V, Zn) of the Late Cretaceous-Early
269 Tertiary Belqa Group, central Jordan: *Palaeogeography*, *Palaeoclimatology*,
270 *Palaeoecology*, v. 369, p. 201–219, doi:10.1016/j.palaeo.2012.10.020.

- 271 Formolo, M.J., and Lyons, T.W., 2007, Accumulation and preservation of reworked
272 marine pyrite beneath an oxygen-rich Devonian atmosphere: constraints from sulfur
273 isotopes and framboid textures: *Journal of Sedimentary Research*, v. 77, p. 623–633,
274 doi:10.2110/jsr.2007.062.
- 275 Hedges, J.I., and Keil, R.G., 1995, Sedimentary organic matter preservation: An
276 assessment and speculative synthesis: *Marine Chemistry*, v. 49, p. 81–115,
277 doi:10.1016/0304-4203(95)00008-F.
- 278 Huber, M., and Goldner, A., 2012, Eocene monsoons: *Journal of Asian Earth Sciences*,
279 v. 44, p. 3–23, doi:10.1016/j.jseaes.2011.09.014.
- 280 Lehmann, B., Nägler, T.F., Holland, H.D., Wille, M., Mao, J.W., Pan, J.Y., Ma, D.S., and
281 Dulski, P., 2007, Highly metalliferous carbonaceous shale and Early Cambrian
282 seawater: *Geology*, v. 35, p. 403–406, doi:10.1130/G23543A.1.
- 283 März, C., Vogt, C., Schnetger, B., and Brumsack, H.-J., 2011, Variable Eocene-Miocene
284 sedimentation processes and bottom water redox conditions in the Central Arctic
285 Ocean (IODP Expedition 302): *Earth and Planetary Science Letters*, v. 310, p. 526–
286 537, doi:10.1016/j.epsl.2011.08.025.
- 287 Pearson, P.N., Van Dongen, B.E., Nicholas, C.J., Pancost, R.D., Schouten, S., Singano,
288 J.M., and Wade, B.S., 2007, Stable warm tropical climate through the Eocene epoch:
289 *Geology*, v. 35, p. 211–214, doi:10.1130/G23175A.1.
- 290 Poulton, S.W., and Canfield, D.E., 2005, Development of a sequential extraction
291 procedure for iron: implications for iron partitioning in continentally derived
292 particulates: *Chemical Geology*, v. 214, p. 209–221,
293 doi:10.1016/j.chemgeo.2004.09.003.

- 294 Poulton, S.W., and Canfield, D.E., 2011, Ferruginous conditions; a dominant feature of
295 the ocean through Earth's history: *Elements*, v. 7, p. 107–112,
296 doi:10.2113/gselements.7.2.107.
- 297 Sangiorgi, F., Brumsack, H.-J., Willard, D.A., Schouten, S., Stickley, C.E., O'Regan, M.,
298 Reichart, G.-J., Sinninghe Damsté, J.S., and Brinkhuis, H., 2008, A 26 million year
299 gap in the Central Arctic record at the greenhouse-icehouse transition: looking for
300 clues: *Paleoceanography*, v. 23, PA1S04, doi:10.1029/2007PA001477.
- 301 Slack, J.F., Selby, D., and Dumoulin, J.A., 2015, Hydrothermal. Biogenic, and seawater
302 components in metalliferous black shales of the Brooks Range, Alaska:
303 Synsedimentary metal enrichment in a carbonate ramp setting: *Economic Geology*
304 and the *Bulletin of the Society of Economic Geologists*, v. 110, p. 653–675,
305 doi:10.2113/econgeo.110.3.653.
- 306 Trabucho-Alexandre, J., Dirx, R., Veld, H., Klaver, G., Poppe, L., and de Boer, P.L.,
307 2012, Toarcian black shales in the Dutch central graben: record of energetic, variable
308 depositional conditions during an oceanic anoxic event: *Journal of Sedimentary*
309 *Research*, v. 82, p. 104–120, doi:10.2110/jsr.2012.5.
- 310 Tribovillard, N., Riboulleau, A., Lyons, T., and Baudin, F., 2004, Enhanced trapping of
311 molybdenum by sulfurized marine organic matter of marine origin in Mesozoic
312 limestones and shales: *Chemical Geology*, v. 213, p. 385–401,
313 doi:10.1016/j.chemgeo.2004.08.011.
- 314 Tribovillard, N., Hatén, E., Averbuch, O., Barbecot, F., Bout-Roumazeilles, V., and
315 Trentesaux, A., 2015, Iron availability as a dominant control on the primary

316 composition and diagenetic overprint of organic-matter-rich rocks: Chemical
317 Geology, v. 401, p. 67–82, doi:10.1016/j.chemgeo.2015.02.026.
318 Wedepohl, K.H., 1991, The composition of the upper earth's crust and the natural cycles
319 of selected metals. Metals in natural raw materials, *in* Merian, E., ed., Natural
320 Resources, Metals and Their Compounds in the Environment: Weinheim, VCH, p.
321 3–17.

322

323 FIGURE CAPTIONS

324

325 Figure 1. A: Paleogeographic map of Paleogene Neo-Tethys, with core location on the
326 shallow continental shelf. B: Close-up of southern Neo-Tethys margin, with location of
327 core locations and border of present-day Jordan (after Alqudah et al., 2014b) C:
328 Lithostratigraphic columns of cores OS23 and OS22 with sediment age, coring depth,
329 percentages of reworked Eocene-Cretaceous calcareous nannofossils, and bioturbated
330 intervals (after Ali Hussein et al., 2014b).

331

332 Figure 2. Core OS23 records of (A) CaCO_3 (wt%), Al (wt%), TOC (wt%), S (wt%), P/Al
333 (wt%/wt%), and Zr/Al (wt%/ppm); and (B) Mo/Al (ppm/wt%), Zn/Al (ppm/wt%), V/Al
334 (ppm/wt%), Cr/Al (ppm/wt%), FeHR/FeT, FeS/FeHR and non-sulphide S (% of total S),
335 against adjusted drilling depth (meters, 0 m = top of black shale succession). Columns on
336 right are geochemical Units (I-V) and lithological units (A-C).

337

338 Figure 3. Schematic illustration of depositional conditions during deposition of Unit III,
 339 and Units II + IV (anoxic/euxinic and oxic states). Black arrows = input of OC- and TM-
 340 rich Cretaceous-Paleogene detritus. Grey arrows = input of OC- and TM-rich matter from
 341 overlying water column.

342

343 ¹GSA Data Repository item 2016xxx, xxxxxxxx, is available online at

344 <http://www.geosociety.org/pubs/ft2016.htm> or on request from editing@geosociety.org.

345

346

347

348

349

350

351

352

353

TABLE 1. AVERAGE Mo, Cr, V, AND Zn EXCESS CONTENTS (ppm) RELATIVE TO AVERAGE SHALE COMPOSITION (WEDEPOHL, 1991) IN UNITS I TO V OF CORES OS22 AND OS23, CRETACEOUS-PALEOGENE BELQA GROUP SHALES AND LIMESTONES (FLEURANCE ET AL., 2013), BLACK SEA UNIT 2, MEDITERRANEAN SAPROPELS, AND OCEAN ANOXIC EVENT 2 DEPOSITS (BRUMSACK, 2006)

	Unit I (OS23 / OS22)	Unit II (OS23 / OS22)	Unit III (OS23 / OS22)	Unit IV (OS23 / OS22)	Unit V (OS23 / OS22)	Belqa Group ¹ (shale)	Belqa Group ¹ (limestone)	Black Sea ² (Unit 2)	Med. Sapropels ²	C/T Boundary (OAE 2) ²
Mo _{xs} (ppm, mean)	31 / 31	97 / 104	110 / 76	401 / 272	38 / 7	496	5	116	104	315
Cr _{xs} (ppm, mean)	248 / 210	265 / 296	360 / 373	514 / 490	186 / 222	392	125	12	69	124
V _{xs} (ppm, mean)	202 / 186	608 / 501	356 / 274	1638 / 1044	139 / 129	688	113	126	457	935
Zn _{xs} (ppm, mean)	521 / 566	703 / 958	825 / 671	1906 / 469	165 / 254	4666	220	31	51	1997

354

Figure 1

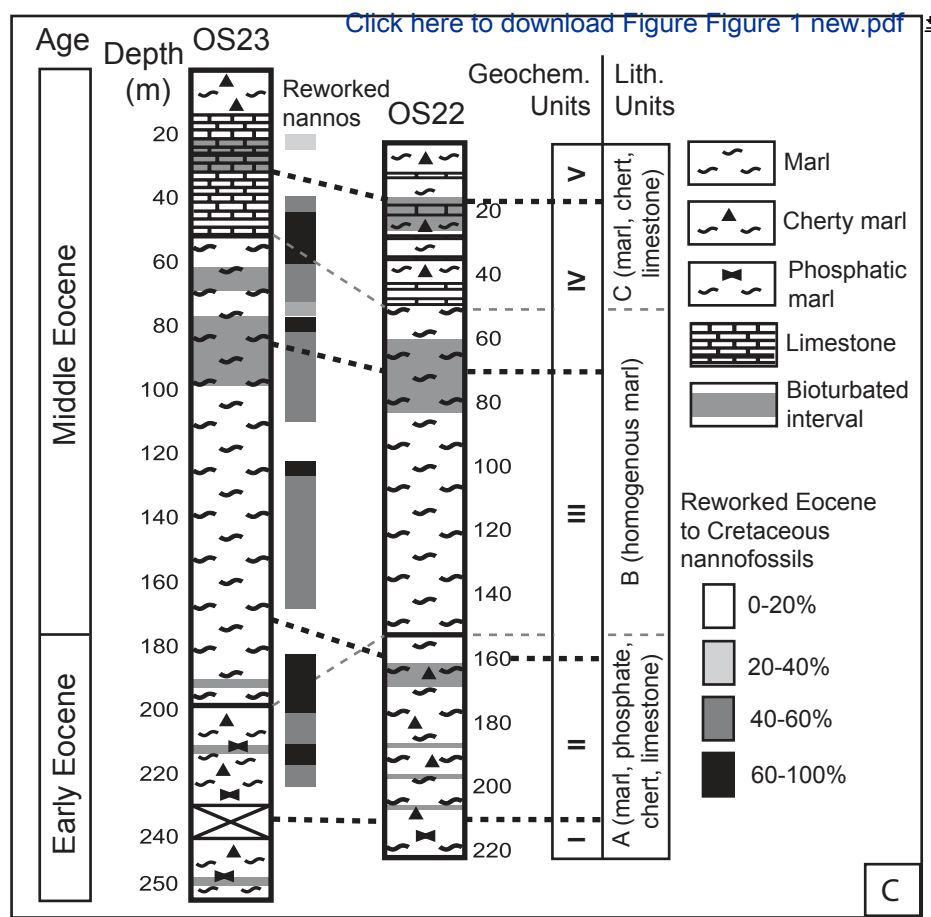
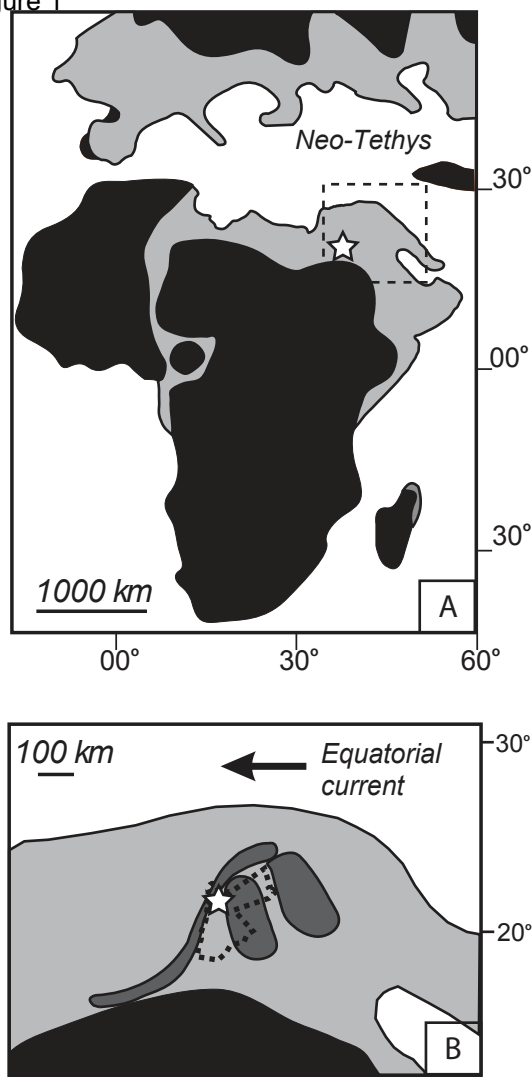


Figure 1

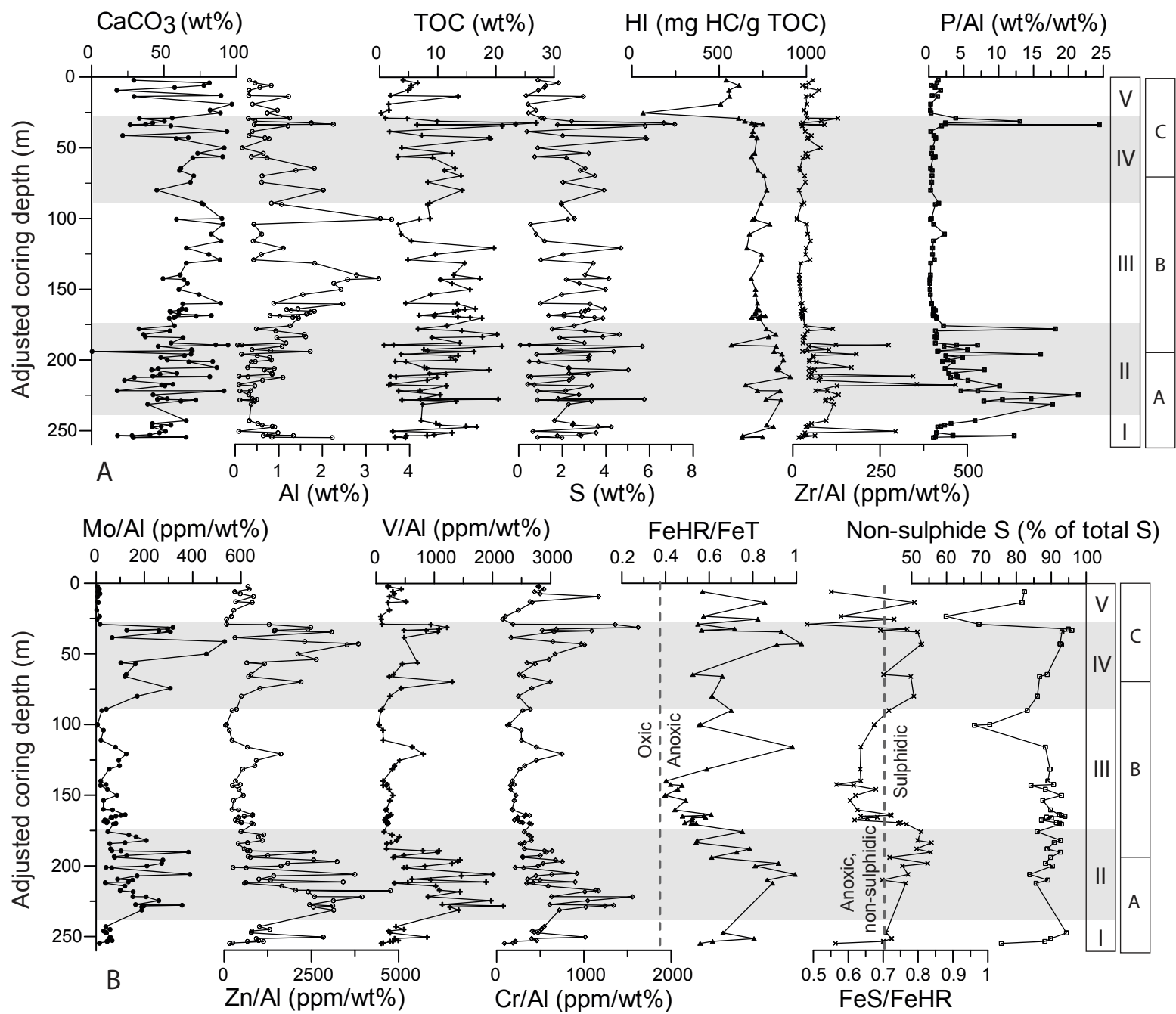
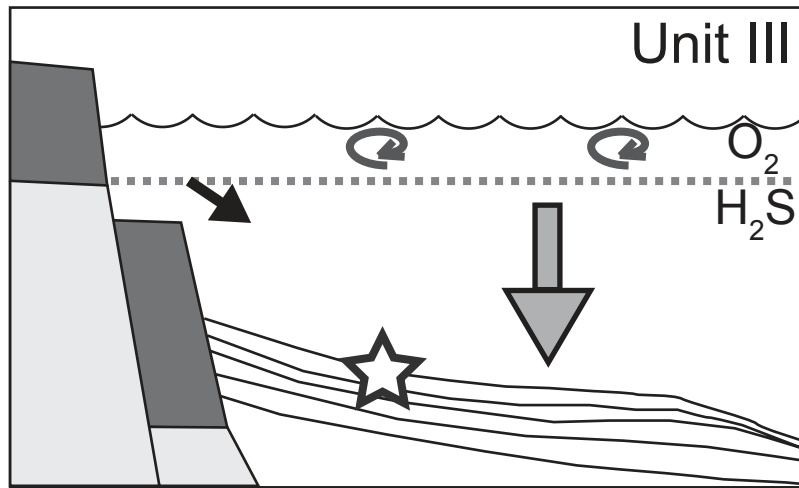
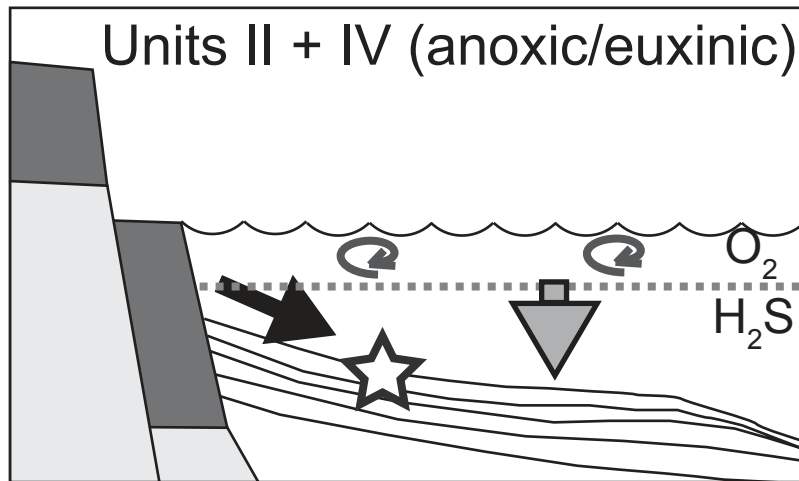


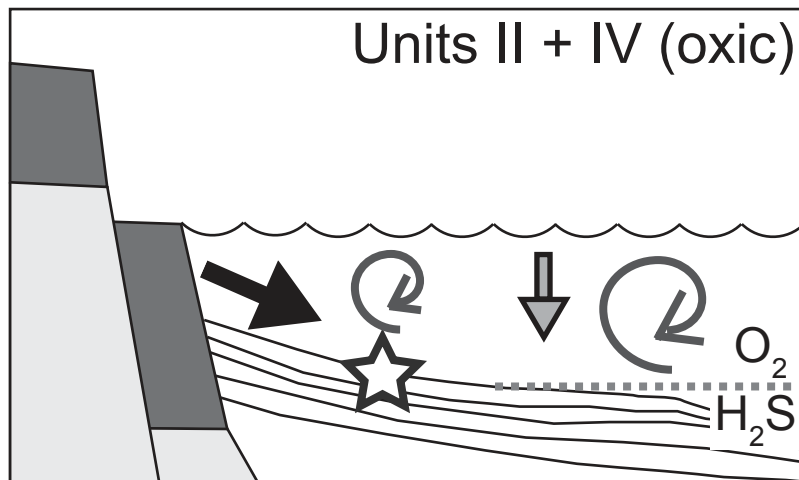
Figure 2



- * Deposition below wave base
- * Shallow redoxcline
- * Anoxic/euxinic seafloor
- * High primary productivity
- * Moderate reworking of older sediments
- * Reworked nannofossils



- * Deposition below wave base
- * Shallow redoxcline
- * Anoxic/euxinic seafloor
- * High primary productivity
- * Intense OC + TM reworking
- * Reworked nannofossils



- * Wave and/or current activity
- * Deep redoxcline
- * Oxic seafloor + bioturbation
- * High primary productivity
- * Intense OC + TM reworking
- * Winnowing of nannofossils
- * Heavy mineral enrichments

Figure 3

Table 1

	<i>Unit I</i> (OS23 / OS22)	<i>Unit II</i> (OS23 / OS22)	<i>Unit III</i> (OS23 / OS22)
<i>Mo_{xs}</i> (ppm, mean)	31 / 31	97 / 104	110 / 76
<i>Cr_{xs}</i> (ppm, mean)	248 / 210	265 / 296	360 / 373
<i>V_{xs}</i> (ppm, mean)	202 / 186	608 / 501	356 / 274
<i>Zn_{xs}</i> (ppm, mean)	521 / 566	703 / 958	825 / 671

<i>Unit IV</i> <i>(OS23 / OS22)</i>	<i>Unit V</i> <i>(OS23 / OS22)</i>	<i>Belqa Group</i> ¹ <i>(shale)</i>	<i>Belqa Group</i> ¹ <i>(limestone)</i>	<i>Black Sea</i> ² <i>(Unit 2)</i>
401 / 272	38 / 7	496	5	116
514 / 490	186 / 222	392	125	12
1638 / 1044	139 / 129	688	113	126
1906 / 469	165 / 254	4666	220	31

<i>Med.</i> <i>Sapropels</i> ²	<i>C/T Boundary</i> <i>(OAE 2)</i> ²
104	315
69	124
457	935
51	1997

1 **ARGUMENTS AGAINST HYDROTHERMAL OR EPIGENETIC METAL**

2 **ENRICHMENTS IN THE JORDAN OIL SHALES**

3 In sedimentary rocks, hyper-enrichments of a wide range of trace metals are often explained
4 by the syn-genetic input of metal-rich hydrothermal solutions or brines into the
5 contemporaneous sea water, or by the post-depositional (epigenetic) stratiform intrusion of
6 metal-rich fluids into sedimentary formations. Here we will argue against a contribution of
7 either of these processes to the trace metal hyper-enrichments in the studied Jordan oil shales.

8

9 *Ruling out epigenetic enrichments:*

- 10 - In the cores, no stratiform or strata-cutting metal-rich veins or layers are found, but
11 the metal enrichments appear to be finely dispersed within the sediment.
- 12 - The organic matter is of low thermal maturity, implying that the sediments were not
13 affected by post-depositional high-temperature overprint.
- 14 - Highest metal enrichments do not stand in any relationship to lowest carbonate
15 contents, which implies that metal enrichments did not form in carbonate dissolution
16 features.
- 17 - The classical epigenetic stratiform ore deposits are most strongly enriched in Zn, but
18 also in Pb, and not substantially enriched in Cr and V. However, the Jordan oil shales
19 lack significant Pb enrichments and are instead strongly enriched in Cr and V.
- 20 - Fossils (e.g., planktonic and benthic foraminifera, shell fragments) are generally well
21 to moderately preserved with sharp boundaries, and show no signs of dissolution by
22 acidic hydrothermal fluids.
- 23 - The sedimentary texture (e.g., bioturbation features, laminations) are pristine
24 throughout the cores and do not show any signs of post-depositional overprint.

25 *Ruling out syn-genetic hydrothermal enrichments:*

- 26 - There are no signs of volcanic activity on the Eocene paleo-shelf of Jordan, and no
27 signs of hydrothermal mounds, veins or pathways in older (Cretaceous-Paleocene)
28 strata in the area that could have delivered metal-rich hydrothermal solutions.
- 29 - Highest trace metal enrichments occur in Units II and IV, where enrichments of Zr
30 and P as well as sedimentary structures and bioturbation features indicate episodic
31 water column mixing. These conditions are not conducive to the spreading of a
32 hydrothermal plume or metal-rich brine, which would require stable salinity
33 stratification.
- 34 - Spreading of metal-rich brines is not supported by the continuous presence, and good
35 preservation, of planktonic and benthic fossils as well as bioturbation features that
36 exclude the presence of hyper-saline brines loaded with dissolved or particulate
37 metals at toxic levels.

38

39 **FIGURE CAPTIONS SUPPLEMENTARY FIGURES**

40 Figure DR1: Element excess contents relative to average shale composition (Wedepohl,
41 1971, 1991) for Jordan oil shales (average for Units I-V in cores OS22 and OS23) and
42 various modern and ancient organic-rich lithologies deposited under anoxic/sulfidic
43 conditions (data from Brumsack, 2006; Fleurance et al., 2013; Coveney and Glascock, 1989;
44 Slack et al., 2015; Lehmann et al., 2007).

45 Figure DR2: Core OS22 records of (A) CaCO₃ (wt%), Al (wt%), TOC (wt%), S (wt%), P/Al
46 (wt%/wt%), and Zr/Al (wt%/ppm); and (B) Mo/Al (ppm/wt%), Zn/Al (ppm/wt%), V/Al
47 (ppm/wt%), Cr/Al (ppm/wt%), FeHR/FeT, FeS/FeHR and non-sulphide S (% of total S),

48 against adjusted drilling depth (meters, 0 m = top of black shale succession). Columns on
49 right are geochemical Units (I-V) and lithological units (A-C).

50

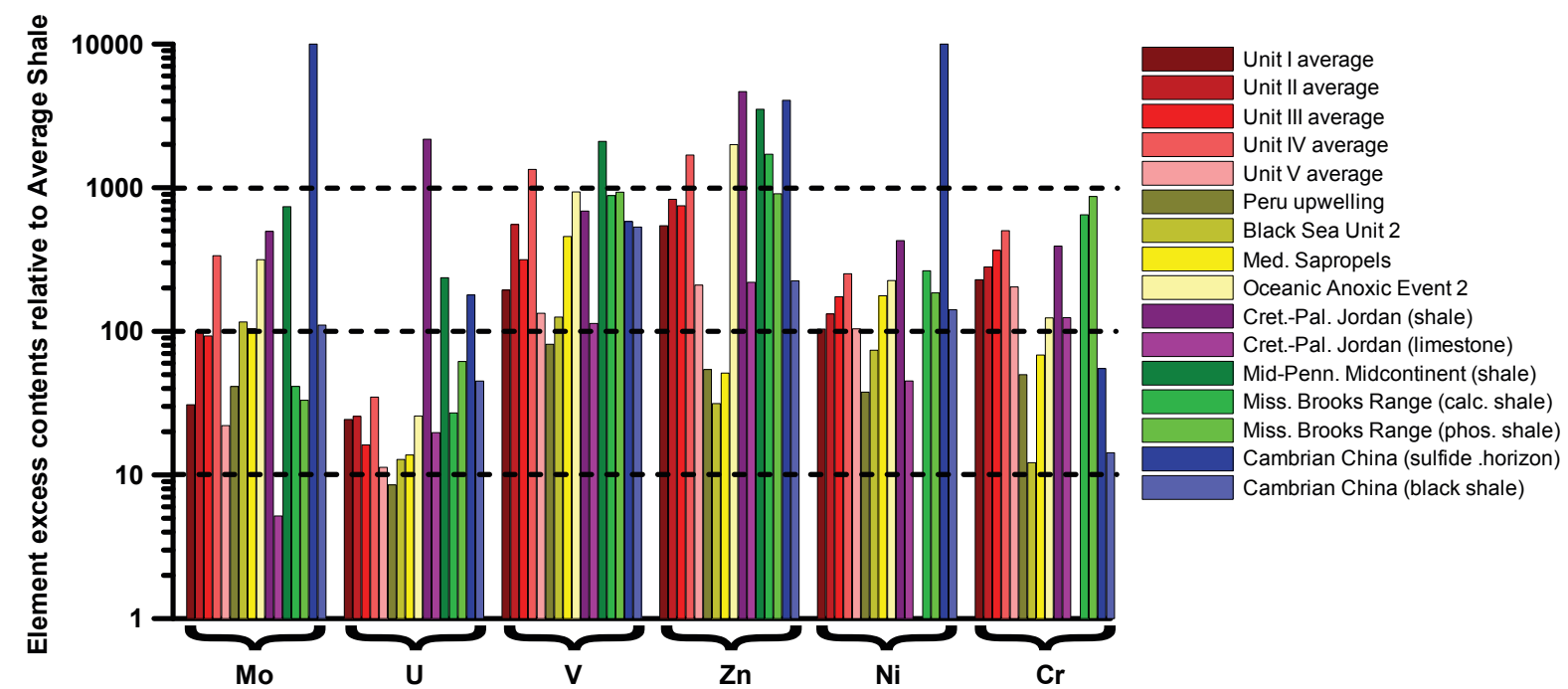


Figure DR1

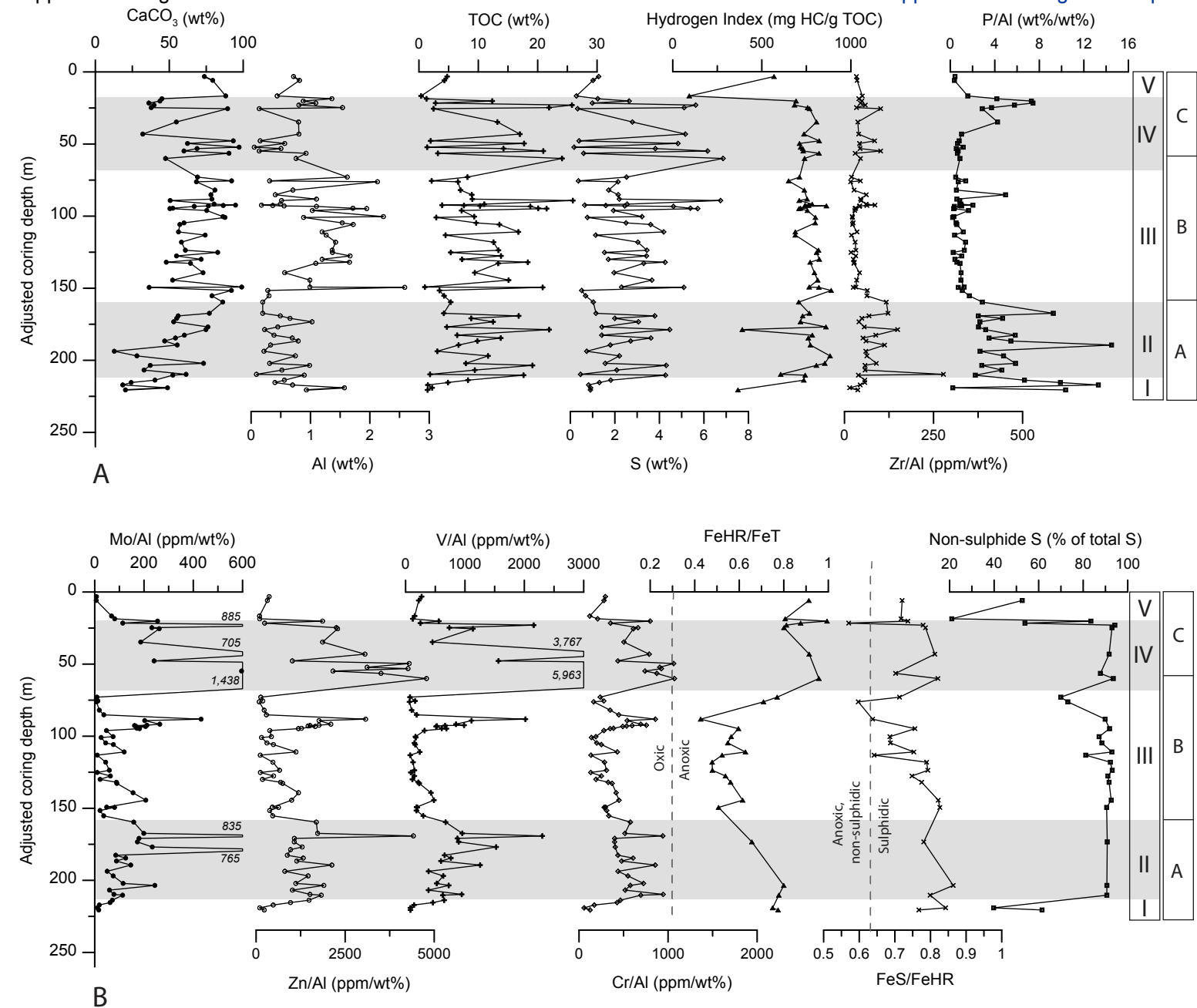


Figure DR2