

ARTICLE

Received 23 Sep 2014 | Accepted 26 Nov 2014 | Published 19 Jan 2015

DOI: 10.1038/ncomms6972

Revealing a double-inversion mechanism for the $F^- + CH_3Cl$ S_N2 reaction

István Szabó¹ & Gábor Czako¹

Stereo-specific reaction mechanisms play a fundamental role in chemistry. The back-side attack inversion and front-side attack retention pathways of the bimolecular nucleophilic substitution (S_N2) reactions are the textbook examples for stereo-specific chemical processes. Here, we report an accurate global analytic potential energy surface (PES) for the $F^- + CH_3Cl$ S_N2 reaction, which describes both the back-side and front-side attack substitution pathways as well as the proton-abstraction channel. Moreover, reaction dynamics simulations on this surface reveal a novel double-inversion mechanism, in which an abstraction-induced inversion via a $FH \cdots CH_2Cl^-$ transition state is followed by a second inversion via the usual $[F \cdots CH_3 \cdots Cl]^-$ saddle point, thereby opening a lower energy reaction path for retention than the front-side attack. Quasi-classical trajectory computations for the $F^- + CH_3Cl$ ($v_1 = 0, 1$) reactions show that the front-side attack is a fast direct, whereas the double inversion is a slow indirect process.

¹Laboratory of Molecular Structure and Dynamics, Institute of Chemistry, Eötvös University, P.O. Box 32, H-1518, Budapest, Hungary. Correspondence and requests for materials should be addressed to G.C. (email: czako@chem.elte.hu).

Bimolecular nucleophilic substitution (S_N2) reactions play a fundamental role in chemistry; therefore, many experimental and theoretical studies have focused on the atomic-level dynamics and mechanism of this class of reactions^{1–9}. In a typical S_N2 reaction, $X^- + H_3CY \rightarrow XCH_3 + Y^-$, the reactive events usually begin with a back-side attack forming a pre-reaction complex, which can be either a hydrogen-bonded $X^- \cdots HCH_2Y$ or a traditional ion-dipole $X^- \cdots H_3CY$ complex, then the system goes through a central transition state (TS), where a new $X-C$ bond forms and the $C-Y$ bond breaks while the umbrella motion of the CH_3 unit inverts the configuration around the tetrahedral carbon centre. This is the famous Walden inversion mechanism of S_N2 reactions, which is described in every organic chemistry textbook. An important feature of this mechanism is its stereo-specificity, which means that an inversion always occurs resulting in a specific configuration of the product molecule, which is the opposite of the reactant's configuration. Stereo-specificity has exceptional importance in nature, for example, all the natural amino acids exist in a specific configuration and their stereo-isomers (enantiomers) cannot be found in natural proteins. So far, we have thought that the stereo-chemistry of the S_N2 reactions is well-understood. Besides the well-known Walden inversion mechanism, there is a front-side attack pathway, which goes over a high energy barrier and results in retention of configuration. This front-side attack mechanism is much less studied than the Walden inversion^{10–13}. Due to the high barrier of the former, the S_N2 reactions are known to proceed via Walden inversion at low collision energies (E_{coll}) and the front-side attack pathway may open at higher E_{coll} . Note that one may distinguish between direct rebound and stripping, as well as indirect (ion-dipole and/or hydrogen-bonded complex formation, roundabout and barrier recrossing) mechanisms¹⁴, but in the present study we consider these as variants of the back-side attack inversion mechanism.

Here we perform high-level reaction dynamics simulations for the $F^- + CH_3Cl$ prototypical S_N2 reaction using a new *ab initio* global potential energy surface (PES). Since in a simulation one

can label the three H atoms, we can examine the configurations of the CH_3F products. As expected, we find that most of the CH_3F molecules have inverted configurations. However, some of the trajectories result in retention of configuration, which is very surprising, because this is found at low E_{coll} well below the barrier height of the front-side attack pathway. How can this happen? In what follows, we describe the details of the reaction dynamics computations and reveal a novel mechanism for S_N2 reactions.

Results

Potential energy surface. Reaction dynamics simulations require the knowledge of the PES, which governs the motion of the atoms in a chemical reaction^{15–17}. Full-dimensional analytical PESs that describe both the back- and front-side attack mechanisms have not been developed for S_N2 reactions. Following our previous work⁹, here we report a global PES for the $F^- + CH_3Cl$ reaction by fitting about 52,000 high-level *ab initio* energy points (see Supplementary Methods). As Fig. 1 shows, the PES describes the back and front-side attack substitution pathways, as well as the abstraction channel leading to $HF + CH_2Cl^-$. Furthermore, we have uncovered another pathway, which begins with an abstraction-induced inversion via a TS of C_s symmetry followed by a second inversion via the usual C_{3v} TS. We call this as a double-inversion mechanism, which results in the retention of configuration. It is also possible that the first induced inversion is not followed by a reactive substitution event; thus, the collision results in an inverted reactant.

Characterization of the stationary points. We have characterized the stationary points of the PES by a sophisticated composite focal-point analysis (FPA)¹⁸ approach considering extrapolation to the complete basis set limits, electron correlation beyond the 'gold-standard' CCSD(T) method, correlation of all the electrons (core and valence) and scalar relativistic effects. More details about the FPA are given in Supplementary Methods. As Fig. 1 shows, the relative energies corresponding to the fitted analytical PES agree well, usually within 1 kcal mol^{-1} , which is considered

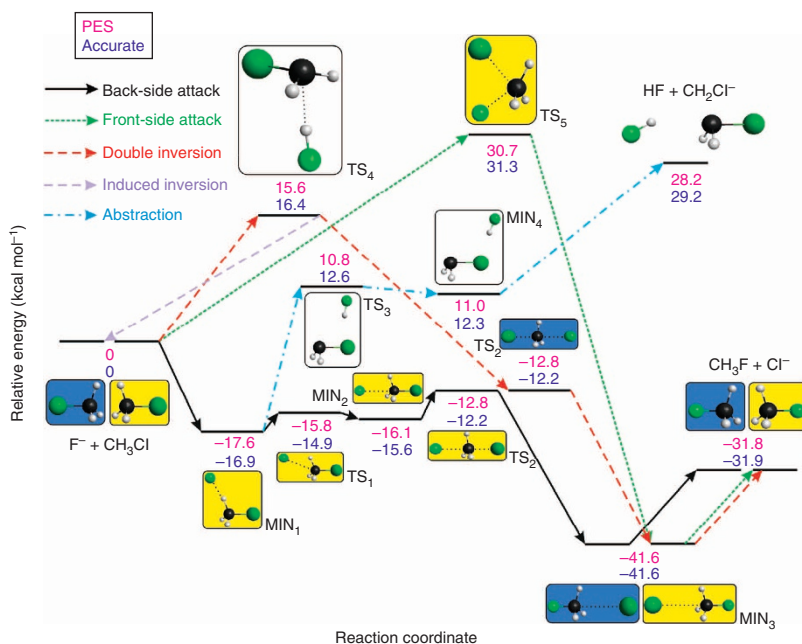


Figure 1 | Schematic of the global potential energy surface (PES) of the $F^- + CH_3Cl$ reaction. Arrows show the different stereo-specific reaction pathways leading to retention (yellow) and inversion (blue) of the initial configuration (yellow). The accurate benchmark energies and the PES values are relative to $F^- + CH_3Cl(eq)$.

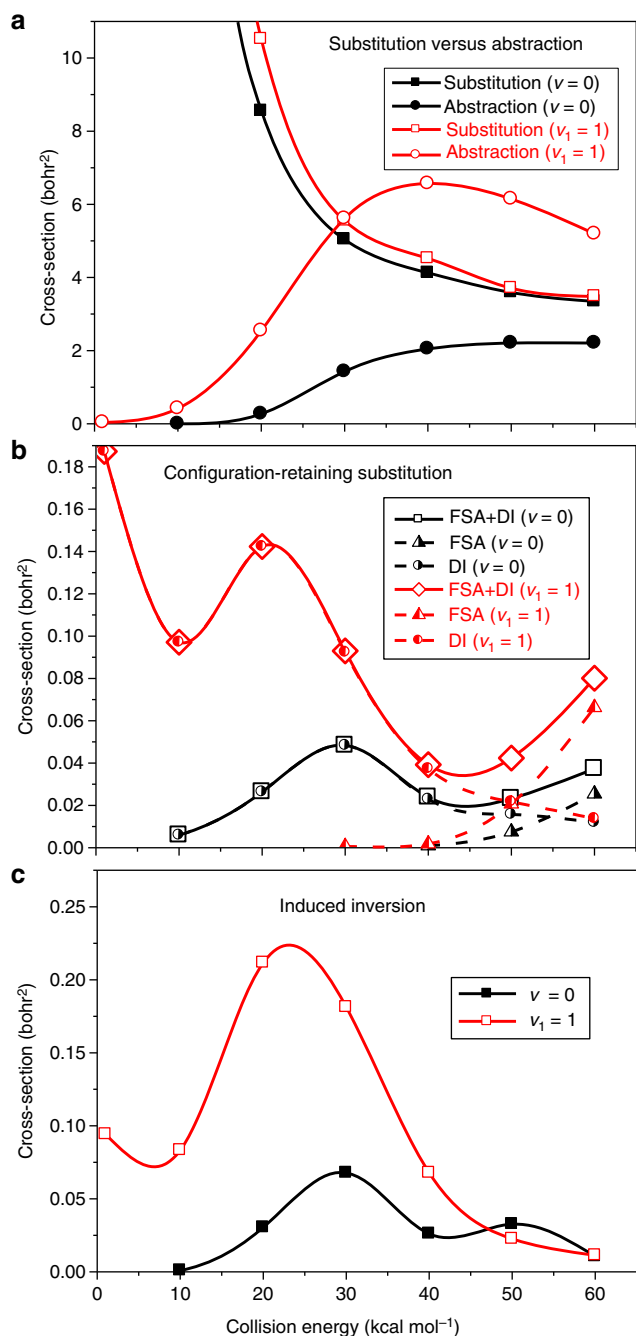


Figure 2 | Cross-sections as a function of collision energy for the ground-state and CH-stretching-excited $F^- + CH_3Cl(v_1 = 0, 1)$ reactions.

(a) back-side attack substitution ($Cl^- + CH_3F$) and abstraction ($HF + CH_2Cl^-$) channels, (b) retention of configuration via front-side attack (FSA) and double-inversion (DI) substitution mechanisms and (c) induced inversion of the reactant CH_3Cl .

as ‘chemical accuracy’, with the benchmark FPA data. The $F^- + CH_3Cl$ S_N2 reaction is highly exothermic, $\Delta E_e = -31.9 \text{ kcal mol}^{-1}$, whereas the abstraction channel is endothermic, $\Delta E_e = 29.2 \text{ kcal mol}^{-1}$. The analytical PES accurately describes many minima (complexes) and saddle points (transition states) that separate the reactants from the products. The back-side attack substitution pathway goes through H-bonded C_s and ion-dipole C_{3v} $F^- \cdots CH_3Cl$ complexes, a C_{3v} $[F \cdots CH_3 \cdots Cl]^-$ TS and a C_{3v} $FCH_3 \cdots Cl^-$ complex, which are all below the $F^- + CH_3Cl(eq)$ asymptote by $16.9 \text{ kcal mol}^{-1}$, $15.6 \text{ kcal mol}^{-1}$, $12.2 \text{ kcal mol}^{-1}$ and

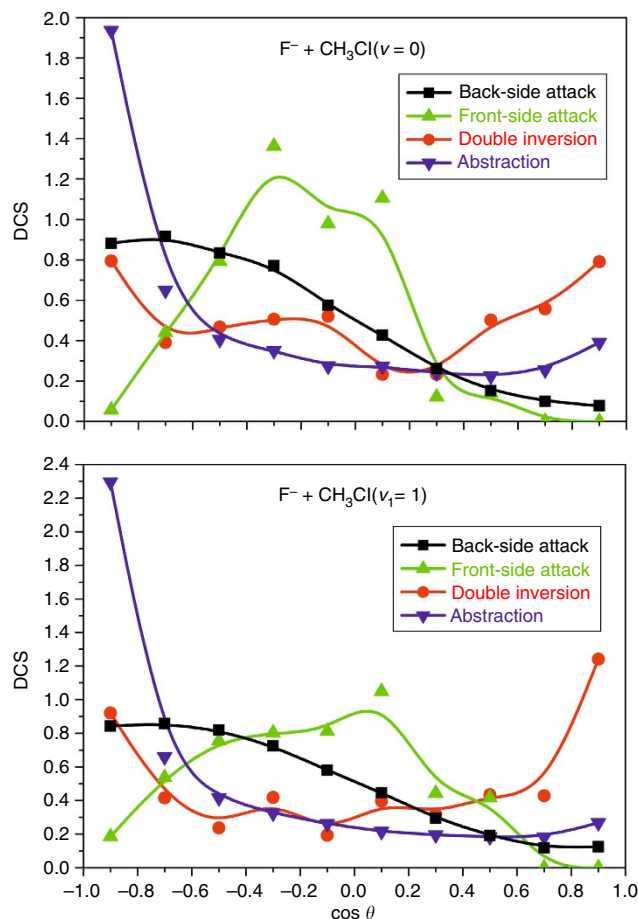


Figure 3 | Scattering angle distributions. Angular distributions of the ground-state and CH-stretching-excited $F^- + CH_3Cl(v_1 = 0, 1)$ reactions for the back-side attack, double-inversion and front-side attack substitution pathways ($Cl^- + CH_3F$) and for the abstraction channel ($HF + CH_2Cl^-$) at a collision energy of 50 kcal mol^{-1} . Each distribution has the same unit area.

$41.6 \text{ kcal mol}^{-1}$, respectively, whereas the front-side attack substitution has a high classical barrier of $31.3 \text{ kcal mol}^{-1}$. Furthermore, we have found a TS for the double-inversion mechanism as shown in Fig. 1, which opens a substantially lower energy configuration-retaining pathway with a classical barrier height of $16.4 \text{ kcal mol}^{-1}$ than the front-side attack mechanism. Structures and harmonic vibrational frequencies of all the stationary points can be found in the Supplementary Tables 1–4.

Reaction dynamics simulations. With an accurate full-dimensional analytical PES at hand, we can go much further than predicting reaction mechanisms based on stationary points. We have performed quasi-classical trajectory (QCT) computations on the PES, thereby following the motion of the atoms along the reaction path. We have run about four million trajectories for the ground-state and symmetric-CH-stretching-excited $F^- + CH_3Cl(v_1 = 0, 1)$ reactions (the initial conditions and analysis techniques are described in Supplementary Methods).

Integral cross-sections. Integral cross-sections (ICSs) as a function of E_{coll} are shown in Fig. 2. Since the exothermic back-side attack substitution reaction does not have a barrier, the S_N2 ICSs are large and decrease steeply with E_{coll} . The H-abstraction reaction is highly endothermic, its zero-point energy (ZPE)

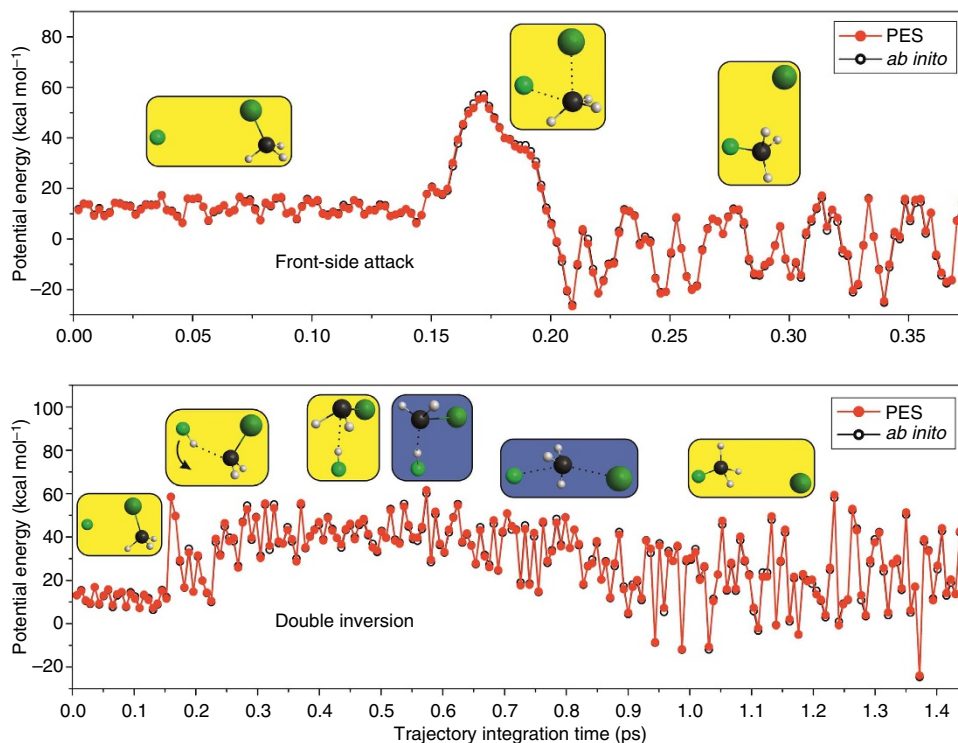


Figure 4 | S_N2 retention mechanisms. Snapshots of representative trajectories showing the front-side attack and double-inversion mechanisms of the $F^- + CH_3Cl(v=0)$ reaction at a collision energy of 50 kcal mol^{-1} . The potential energies, relative to $F^- + CH_3Cl(eq)$, obtained from the fitted PES and direct *ab initio* computations are shown as a function of time. Blue background denotes an inverted configuration relative to that of the reactant (yellow).

corrected reaction enthalpy is $24.5 \text{ kcal mol}^{-1}$ on the PES, and the abstraction saddle point is well below the product asymptote. The ICSs show that the $HF + CH_2Cl^-$ channel opens as soon as it becomes energetically available and the ICSs increase with E_{coll} . (The small reactivity at $E_{coll} < 24.5 \text{ kcal mol}^{-1}$ is due to ZPE violation of the products.) The $v_1=1$ excitation has only slight effect on the S_N2 reactivity, showing that the CH symmetric stretching vibrational mode behaves as a spectator in the back-side attack S_N2 reaction. In the case of the H-abstraction, as expected, CH stretching excitation substantially enhances the reactivity and lowers the reaction threshold. It is interesting to find that the $v_1=1$ abstraction ICSs increase up to a E_{coll} of $\sim 40 \text{ kcal mol}^{-1}$ and then a slight decay is seen, because the faster reactants have less time to interact with each other. This depression of reactivity with the increase of E_{coll} was also found in the $H + CD_4 \rightarrow HD + CD_3$ reaction¹⁹.

Figure 2 also shows the ICSs for the configuration-retaining substitution. These ICSs are much smaller than those of the inversion mechanism (1.1% and 2.3% at $E_{coll} = 60 \text{ kcal mol}^{-1}$ for $v=0$ and $v_1=1$, respectively), but the absolute ICSs are not negligible since, for example, the $H + CD_4$ reaction has similar small ICSs¹⁹. For $F^- + CH_3Cl(v=0)$, the configuration-retaining pathway opens at a E_{coll} of $\sim 10 \text{ kcal mol}^{-1}$, the ICSs rise up to $E_{coll} = \sim 30 \text{ kcal mol}^{-1}$, then decay and increase again around $E_{coll} = 40 - 50 \text{ kcal mol}^{-1}$. The adiabatic barrier height for the front-side attack path is $29.3 \text{ kcal mol}^{-1}$ on the PES; thus, this mechanism cannot produce CH_3F with retention of configuration in the $10 - 30 \text{ kcal mol}^{-1}$ E_{coll} range. Examination of many configuration-retaining trajectories has revealed a double-inversion mechanism, which has an adiabatic barrier height of only $12.6 \text{ kcal mol}^{-1}$ on the PES, thereby allowing retention of configuration at E_{coll} where the front-side attack pathway is closed. Based on the analysis of the integration time of

the reactive trajectories, we have found that none of the double-inversion trajectories finished within 0.65 ps, whereas all the front-side attack reactions occurred faster than 0.65 ps. Thus, on the basis of the integration time we could distinguish between double-inversion and front-side attack trajectories and we could get the mechanism-specific retention ICSs as shown in Fig. 2. As expected, at low E_{coll} every configuration-retaining substitution event goes via double inversion. The front-side attack pathway opens at a E_{coll} of $\sim 40 \text{ kcal mol}^{-1}$, well above the adiabatic barrier. Now we can explain the E_{coll} dependence of the ICSs: the ICSs start to increase when the double-inversion pathway opens, above E_{coll} of 30 kcal mol^{-1} the reactivity via double inversion decreases, because the large E_{coll} does not favour the indirect mechanism and the abstraction channel opens, then at $E_{coll} = \sim 40 \text{ kcal mol}^{-1}$ the front-side attack reactions raise the ICSs again.

After the discovery of the double-inversion mechanism, we can also expect to get inverted CH_3Cl molecules if the first inversion is not followed by substitution. Indeed, the analysis of the ‘non-reactive’ trajectories revealed that some of the reactants became inverted via an abstraction-induced inversion. As Fig. 2 shows, the E_{coll} dependence of the induced-inversion ICSs is consistent with that of the double-inversion ICSs: the induced inversion opens above $E_{coll} = \sim 10 \text{ kcal mol}^{-1}$ and the ICSs have a maximum at $E_{coll} = \sim 30 \text{ kcal mol}^{-1}$. Since the first step of the double- and induced-inversion mechanisms is pulling out a proton, CH-stretching excitation substantially enhances the double and induced inversions and diminishes their energy thresholds.

Differential cross-sections. Angular distributions for the $F^- + CH_3Cl(v_1=0, 1)$ reactions at $E_{coll} = 50 \text{ kcal mol}^{-1}$ are shown

in Fig. 3. As seen, the scattering angle distributions are very different for the various mechanisms. The back-side attack pathway results in mainly backward scattered products, as expected, since the $[F \cdots CH_3 \cdots Cl]^-$ TS is collinear and the direct rebound mechanism is dominant at a high E_{coll} (refs 9,20). The front-side attack substitutions are also direct; thus, dominance of sideways scattering is seen as expected based on the TS structure, because the F–C–Cl angle is about 80° at the TS. The double-inversion substitutions result in more-or-less isotropic angular distributions indicating an indirect mechanism, where long-lived complexes are formed. For the H-abstraction channel, a clear preference for backward scattering is found, suggesting that stripping is not significant at $E_{\text{coll}} = 50 \text{ kcal mol}^{-1}$. CH-stretching excitation virtually does not affect the angular distributions, perhaps the most significant effect is seen for the front-side attack, where $v_1 = 1$ excitation broadens the angular distributions. In Supplementary Fig. 1, angular distributions are also shown at a smaller E_{coll} of 30 kcal mol^{-1} . The shapes of the distributions are very similar at 30 and 50 kcal mol^{-1} E_{coll} . Note that the front-side attack pathway is not open at the lower E_{coll} .

Retention pathways step by step. Representative trajectories illustrating the key steps of the front-side attack and the double-inversion mechanisms, and showing the potential energy as a function of time are shown in Fig. 4. As seen, in the first 0.15 ps while the reactants approach each other, the potential energy oscillates around 12 kcal mol^{-1} , which corresponds to half of the ZPE of CH_3Cl , in agreement with the virial theorem. The front-side attack substitution is very direct, the system goes through a high barrier and vibrationally excited CH_3F product is formed. The double inversion occurs on a much longer time scale. First, the F^- abstracts a proton (H^+), but the system does not have enough energy to fall apart, thus HF starts to move around CH_2Cl^- and eventually a C–H bond forms again while the configuration around the carbon centre gets inverted. Second, a C–F bond forms and the C–Cl bond breaks resulting in a second inversion via the usual C_{3v} TS. Therefore, the double inversion results in retention of configuration via a long indirect mechanism. These unusual reaction pathways were verified by direct *ab initio* computations along selected trajectories. As shown in Fig. 4, the *ab initio* data reproduce remarkably well the energies obtained from the analytically fitted PES, thereby confirming the new findings.

Discussion

The double-inversion mechanism revealed here for the $F^- + CH_3Cl$ reaction may be a general pathway for substitution reactions, where hydrogen/proton abstraction is a competing channel. We plan to develop analytical global PESs for other systems to see if this is the case. Furthermore, it is important to note that the first step of the double inversion is reminiscent of the famous roaming mechanism discovered for the photodissociation of formaldehyde (H_2CO)²¹. In roaming, the radical fragments, $H \cdots HCO$, do not have enough energy in the dissociation coordinate to break apart; therefore, they follow a non-traditional path to form the molecular products, $H_2 + CO$. In double inversion, the $FH \cdots CH_2Cl^-$ complex cannot dissociate, but the system has enough energy to invert the configuration around the carbon atom, thereby opening a new way for retention in S_N2 reactions. The roundabout S_N2 mechanism also has features similar to roaming¹⁴, therefore, the present study may inspire future research to focus on the possibility of retention via the roundabout pathway and the abstraction-induced mechanisms in S_N2 reactions. It is possible that the new

double-inversion mechanism could be related with the already known roundabout mechanism. Since the first step of the double inversion is an abstraction-induced inversion, the reaction may result in an inverted reactant molecule as shown in the present study. Finding inverted reactants could be a signature for the double-inversion mechanism. Finally, one should note that the above atomistic mechanisms are defined based on classical dynamics. This approach is usually a good approximation for chemical reactions, but, of course, the nuclei do not exactly follow a defined trajectory. A time-dependent quantum mechanical treatment of the nuclear dynamics could show the regions of the configuration space in which the wave functions have non-negligible values. The comparison of the present findings with future quantum dynamical results may add to our understanding of chemical reaction mechanisms.

Methods

Potential energy surface. The global analytical full-dimensional PES for the $F^- + CH_3Cl$ reaction is obtained by fitting 52,393 *ab initio* energy points computed by an efficient composite method. We have selected structures that cover the configuration space and energy range of chemical importance. The composite *ab initio* energies are computed as

$$\begin{aligned} & \text{CCSD(T)/aug-cc-pVDZ} + \text{MP2/aug-cc-pVQZ} - \text{MP2/aug-cc-pVDZ} \\ & + (\text{AE-MP2/aug-cc-pCVTZ} - \text{FC-MP2/aug-cc-pCVTZ}), \end{aligned} \quad (1)$$

where in parenthesis the core correlation energy increment is given as a difference between all-electron (AE) and frozen-core (FC) energies. This composite method provides AE-CCSD(T)/aug-cc-pCVQZ quality results within a root-mean-square error of only $0.35 \text{ kcal mol}^{-1}$.

The analytical representation of the PES is obtained by a fifth-order fit using the permutationally invariant polynomial approach^{15,16} based on Morse-like variables, $\exp(-r_{ij}/a)$, where r_{ij} are the inter-atomic distances and $a = 3 \text{ bohr}$. The linear least-squares fit, with weight of $E_0/(E + E_0)$, where $E_0 = 31 \text{ kcal mol}^{-1}$ and E is relative to the global minimum, provides 3,313 coefficients. These coefficients are provided in Supplementary Data 1. The root-mean-square fitting errors are $0.31 \text{ kcal mol}^{-1}$, $0.51 \text{ kcal mol}^{-1}$ and $1.38 \text{ kcal mol}^{-1}$ for the energy ranges $0\text{--}31 \text{ kcal mol}^{-1}$, $31\text{--}63 \text{ kcal mol}^{-1}$ and $63\text{--}157 \text{ kcal mol}^{-1}$, respectively.

Benchmark *ab initio* thermochemistry. The best technically feasible relative energies for the stationary points of the $F^- + CH_3Cl$ reaction are obtained by the FPA¹⁸ approach. First, we compute the structures and harmonic frequencies at the AE-CCSD(T)/aug-cc-pCVQZ and FC-CCSD(T)/aug-cc-pVTZ levels of theory, respectively, for all the minima and saddle points shown in Fig. 1, except for TS₃, TS₄, TS₅ and MIN₄, where the structures are obtained at the AE-CCSD(T)/aug-cc-pCVTZ level. Second, the benchmark FPA relative energies are obtained by considering (a) extrapolation to the complete basis set limit using AE-CCSD(T)/aug-cc-pCVnZ ($n = Q(4)$ and 5) energies, (b) post-CCSD(T) correlation effects up to CCSDT(Q) based on AE-CCSDT/aug-cc-pCVDZ and FC-CCSDT(Q)/aug-cc-pVDZ energy computations and (c) scalar relativistic effects at the second-order Douglas–Kroll AE-CCSD(T)/aug-cc-pCVQZ level of theory.

QCT calculations. QCT computations are performed for the $F^- + CH_3Cl(v_1 = 0, 1)$ reactions using the new analytical PES. The initial conditions for the trajectories are as follows: (a) standard normal mode sampling is used, (b) trajectories are run at E_{coll} of 1, 10, 20, 30, 40, 50 and 60 kcal mol^{-1} , (c) the total number of $v = 0(v_1 = 1)$ trajectories are 145,000 (145,000), 125,000(125,000), 85,000(95,000), 645,000(725,000), 85,000(95,000), 645,000(725,000) and 85,000(95,000) at the above E_{coll} , respectively, and (d) the integration time step is 0.0726 fs and each trajectory is propagated until the maximum of the actual inter-atomic distances is 1 bohr larger than the initial one. We have found that basically no S_N2 trajectory violates the product ZPE; thus, the QCT analysis considers all the trajectories. The stereo-specific configurations of CH_3Cl and CH_3F are analyzed based on the procedure described in ref. 22.

References

- Chabiny, C. L., Craig, S. L., Regan, C. K. & Brauman, J. I. Gas-phase ionic reactions: dynamics and mechanism of nucleophilic displacements. *Science* **279**, 1882–1886 (1998).
- Sun, L., Song, K. & Hase, W. L. A S_N2 reaction that avoids its deep potential energy minimum. *Science* **296**, 875–878 (2002).
- Gonzales, J. M. *et al.* Definitive *ab initio* studies of model S_N2 reactions $CH_3X + F^-$ ($X = F, Cl, CN, OH, SH, NH_2, PH_2$). *Chem. Eur. J.* **9**, 2173–2192 (2003).

- Mikosch, J. *et al.* Imaging nucleophilic substitution dynamics. *Science* **319**, 183–186 (2008).
- Brauman, J. I. Not so simple. *Science* **319**, 168–168 (2008).
- Mikosch, J. *et al.* Indirect dynamics in a highly exoergic substitution reaction. *J. Am. Chem. Soc.* **135**, 4250–4259 (2013).
- Manikandan, P., Zhang, J. & Hase, W. L. Chemical dynamics simulations of $X^- + CH_3Y \rightarrow XCH_3 + Y^-$ gas-phase S_N2 nucleophilic substitution reactions. Nonstatistical dynamics and nontraditional reaction mechanisms. *J. Phys. Chem. A* **116**, 3061–3080 (2012).
- Zhang, J., Xu, Y., Chen, J. & Wang, D. Y. A multilayered-representation, quantum mechanical/molecular mechanics study of the $CH_3Cl + F^-$ reaction in aqueous solution: the reaction mechanism, solvent effects and potential of mean force. *Phys. Chem. Chem. Phys.* **16**, 7611–7617 (2014).
- Szabó, I., Császár, A. G. & Czakó, G. Dynamics of the $F^- + CH_3Cl \rightarrow Cl^- + CH_3F$ S_N2 reaction on a chemically accurate potential energy surface. *Chem. Sci.* **4**, 4362–4370 (2013).
- Glukhovtsev, M. N., Pross, A., Schlegel, H. B., Bach, R. D. & Radom, L. Gas-phase identity S_N2 reactions of halide anions and methyl halides with retention of configuration. *J. Am. Chem. Soc.* **118**, 11258–11264 (1996).
- Li, G. & Hase, W. L. *Ab initio* direct dynamics trajectory study of the $Cl^- + CH_3Cl$ S_N2 reaction at high reagent translational energy. *J. Am. Chem. Soc.* **121**, 7124–7129 (1999).
- Angel, L. A. & Ervin, K. M. Dynamics of the gas-phase reactions of fluoride ions with chloromethane. *J. Phys. Chem. A* **105**, 4042–4051 (2001).
- Yang, Z.-Z., Ding, Y.-L. & Zhao, D.-X. Theoretical analysis of gas-phase front-side attack identity $S_N2(C)$ and $S_N2(Si)$ reactions with retention of configuration. *J. Phys. Chem. A* **113**, 5432–5445 (2009).
- Xie, J. *et al.* Identification of atomic-level mechanisms for gas-phase $X^- + CH_3Y$ S_N2 reactions by combined experiments and simulations. *Acc. Chem. Res.* **47**, 2960–2969 (2014).
- Braams, B. J. & Bowman, J. M. Permutationally invariant potential energy surfaces in high dimensionality. *Int. Rev. Phys. Chem.* **28**, 577–606 (2009).
- Bowman, J. M., Czakó, G. & Fu, B. High-dimensional *ab initio* potential energy surfaces for reaction dynamics calculations. *Phys. Chem. Chem. Phys.* **13**, 8094–8111 (2011).
- Czakó, G. & Bowman, J. M. Dynamics of the reaction of methane with chlorine atom on an accurate potential energy surface. *Science* **334**, 343–346 (2011).
- Császár, A. G., Allen, W. D. & Schaefer, H. F. In pursuit of the *ab initio* limit for conformational energy prototypes. *J. Chem. Phys.* **108**, 9751–9764 (1998).
- Zhang, W. *et al.* Depression of reactivity by the collision energy in the single barrier $H + CD_4 \rightarrow HD + CD_3$ reaction. *Proc. Natl Acad. Sci. USA* **107**, 12782–12785 (2010).
- Su, T., Wang, H. & Hase, W. L. Trajectory studies of S_N2 nucleophilic substitution. 7. $F^- + CH_3Cl \rightarrow FCH_3 + Cl^-$. *J. Phys. Chem. A* **102**, 9819–9828 (1998).
- Townsend, D. *et al.* The roaming atom: straying from the reaction path in formaldehyde decomposition. *Science* **306**, 1158–1161 (2004).
- Czakó, G. Gaussian binning of the vibrational distributions for the $Cl + CH_4(v_{4/2} = 0, 1) \rightarrow H + CH_3Cl(n_1n_2n_3n_4n_5n_6)$ reactions. *J. Phys. Chem. A* **116**, 7467–7473 (2012).

Acknowledgements

G.C. was supported by the Scientific Research Fund of Hungary (OTKA, NK-83583) and the János Bolyai Research Scholarship of the Hungarian Academy of Sciences.

Author contributions

G.C. designed the research; I.S. performed the computations, analyzed the data and made the figures and tables under the guidance of G.C. and G.C. wrote the manuscript.

Additional information

Supplementary Information accompanies this paper at <http://www.nature.com/naturecommunications>

Competing financial interests: The authors declare no competing financial interests.

Reprints and permission information is available online at <http://npg.nature.com/reprintsandpermissions/>

How to cite this article: Szabó, I. *et al.* Revealing a double-inversion mechanism for the $F^- + CH_3Cl$ S_N2 reaction. *Nat. Commun.* **6**:5972 doi: 10.1038/ncomms6972 (2015).