



Rigby, M., Montzka, S. A., Prinn, R. G., White, J. W. C., Young, D., O'Doherty, S., Lunt, M. F., Ganesan, A. L., Manning, A. J., Simmonds, P. G., Salameh, P. K., Harth, C. M., Mühle, J., Weiss, R. F., Fraser, P. J., Steele, L. P., Krummel, P. B., McCulloch, A., & Park, S. (2017). Role of atmospheric oxidation in recent methane growth. *Proceedings of the National Academy of Sciences of the United States of America*, 114(21), 5373-5377. https://doi.org/10.1073/pnas.1616426114

Peer reviewed version

Link to published version (if available): 10.1073/pnas.1616426114

Link to publication record in Explore Bristol Research PDF-document

This is the author accepted manuscript (AAM). The final published version (version of record) is available online via PNAS at http://www.pnas.org/content/early/2017/04/11/1616426114. Please refer to any applicable terms of use of the publisher.

University of Bristol - Explore Bristol Research General rights

This document is made available in accordance with publisher policies. Please cite only the published version using the reference above. Full terms of use are available: http://www.bristol.ac.uk/red/research-policy/pure/user-quides/ebr-terms/

The role of atmospheric oxidation in recent methane growth

Matt Rigby^{a,1}, Stephen A. Montzka^b, Ronald G. Prinn^c, James W. C. White^d, Dickon Young^a, Simon O'Doherty^a, Mark Lunt^a, Anita L. Ganesan^e, Alistair Manning^f, Peter Simmonds^a, Peter K. Salameh^g, Chris M. Harth^g, Jens Mühle^g, Ray F. Weiss^g, Paul J. Fraser^h, L. Paul Steele^h, Paul B. Krummel^h, Archie McCulloch^a, and Sunyoung Parkⁱ

^aSchool of Chemistry, University of Bristol, Bristol, BS8 1TS, UK; ^bNational Oceanic and Atmospheric Administration, Earth System Research Laboratory, Boulder, Colorado, 80303, USA; ^cCenter for Global Change Science, Massachusetts Institute of Technology, Cambridge, Massachusetts, 02139, USA; ^dInstitute of Arctic and Alpine Research, University of Colorado, Bounder, Colorado, 80309, USA; ^eSchool of Geographical Sciences, University of Bristol, Bristol, BS8 1SS, UK; ^fHadley Centre, Met Office, Exeter, EX1 3PB, UK; ^gScripps Institution of Oceanography, University of California, San Diego, La Jolla, CA 92093, USA; ^hCSIRO Oceans and Atmosphere, Climate Science Centre, Aspendale, Victoria 3195, Australia; ¹Department of Oceanography, Kyungpook National University, Sangju 742-711, Republic of Korea

This manuscript was compiled on February 28, 2017

The growth in global methane (CH₄) concentration, which had been ongoing since the industrial revolution, stalled around the year 2000, before resuming globally in 2007. We evaluate the role of the hydroxyl radical (OH), the major CH₄ sink, in the recent CH₄ growth. We also examine the influence of systematic uncertainties in OH concentrations on CH₄ emissions inferred from atmospheric observations. We use observations of 1,1,1-trichloroethane (CH3CCI3), which is lost primarily through reaction with OH, to estimate OH levels as well as CH₃CCl₃ emissions, whose uncertainty has previously limited the accuracy of OH estimates. We find a 64% - 70% probability that a decline in OH has contributed to the post-2007 methane rise. Our median solution suggests that CH₄ emissions increased relatively steadily during the late 1990s and early 2000s, after which, growth was more modest. This solution obviates the need for a sudden, statistically significant change in total CH₄ emissions around the year 2007 to explain the atmospheric observations, and can explain some of the decline in the atmospheric ¹³CH₄/¹²CH₄ ratio and the recent growth in C₂H₆. Our approach indicates that significant OH-related uncertainties in the CH₄ budget remain, and we find that it is not possible to implicate, with a high degree of confidence, rapid global CH₄ emissions changes as the primary driver of recent trends, when our inferred OH trends and these uncertainties are considered.

Methane | Hydroxyl radical | Methyl chloroform | Isotope

ethane (CH₄), the second most important partially anthropogenic greenhouse gas, is observed to vary markedly in its year-to-year growth rate (Figure 1). The causes of these variations have been the subject of much controversy and uncertainty, primarily because there are a wide range of poorly quantified sources and because its sinks are ill-constrained[1]. Of particular recent interest is the cause of the "pause" in CH₄ growth between 1999 and 2007, and the renewed growth from 2007 onwards[2–7]. It is important that we understand these changes if we are to better project future CH₄ changes and effectively mitigate enhanced radiative forcing due to anthropogenic methane emissions.

The major sources of CH₄ include wetlands (natural and agricultural), fossil fuel extraction and distribution, enteric fermentation in ruminant animals and solid and liquid waste. Our understanding of the sources of CH₄ come from two approaches: "bottom-up", in which inventories or process models are used to predict fluxes, or "top-down", in which fluxes are inferred from observations assimilated into atmospheric chemical transport models. Bottom-up methods suffer from uncertainties and potential biases in the available activity data or emissions factors, or the extrapolation to large scales of a

relatively small number of observations. Furthermore, there is no constraint on the global total emissions from bottom-up techniques. The top-down approach is limited by incomplete or imperfect observations and our understanding of atmospheric transport and chemical sinks. For CH_4 , these difficulties result in a significant mismatch between the two methods [1].

The primary CH₄ sink is the hydroxyl radical (OH) in the troposphere, although smaller sinks also exist, such as methanotrophic bacteria in soils, oxidation by chlorine radicals in the marine boundary layer and photochemical destruction in the stratosphere. Predictions of the magnitude and variability of OH in the current generation of atmospheric models have been shown to be diverse[8]. Furthermore, due to its short lifetime, it is difficult to infer global OH concentrations using direct observations. Therefore, indirect observational methods are needed. The most commonly used approach has been to monitor the trends in 1,1,1-trichloroethane (CH₃CCl₃), whose major sink is reaction with OH, and, by making assumptions about its emissions into the atmosphere, infer global OH concentrations [9–13]. Recent work using this approach indicated that OH changes could have played a role in the pause in CH₄ that occurred after 1998 [3, 14].

Previous studies have shown that OH trends inferred using

Significance Statement

Methane, the second most important greenhouse gas, has varied markedly in its atmospheric growth rate. The cause of these fluctuations remains poorly understood. Recent efforts to determine the drivers of the pause in growth in 1999 and renewed growth from 2007 onwards have focused primarily on changes in sources alone. Here we show that changes in the major methane sink, the hydroxyl radical, have likely played a substantial role in the global methane growth rate. This work has significant implications for our understanding of the methane budget, which is important if we are to better predict future changes in this potent greenhouse gas, and effectively mitigate enhanced radiative forcing due to anthropogenic emissions.

MR devised and carried out the research and wrote the manuscript. SAM and RGP helped to devise the research, provided data and contributed to the writing of the manuscript. JWCW, DY, SOD, PS, PKS, CMH, JM, RFW, LPS, PJF, PBK and SP contributed observations and helped with the writing of the manuscript. MFL, AG and AM carried out a regional analysis of the CH₃ CCl₃ observations. AMC provided guidance on the CH₃ CCl₃ emissions model.

The authors declare no conflicts of interest.

¹ To whom correspondence should be addressed. E-mail: matt.rigby@bristol.ac.uk

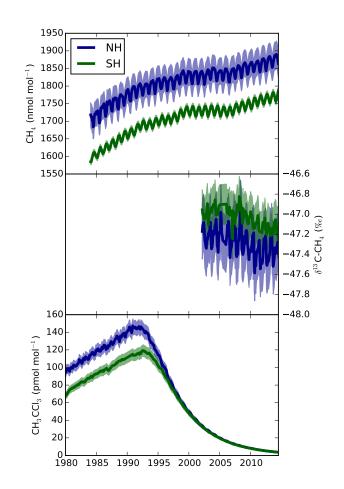


Fig. 1. First panel: NOAA observations of CH4; second panel: INSTAAR observations of δ^{13} C-CH₄: third panel: AGAGE observations of CH₂CCl₂. Each plot shows the northern hemisphere (NH) and southern hemisphere (SH) mean, and shading indicates the assumed 1-sigma model and measurement uncertainty, as defined in the SI Materials and Methods.

CH₃CCl₃ could be highly sensitive to systematic errors in the assumed emissions trends, particularly in the 1980s and early 1990s when emissions were changing rapidly [15]. Some authors have attempted to reduce this source of uncertainty by including CH₃CCl₃ emissions as part of the inversion[12]. However, these studies assumed that emissions uncertainties were Gaussian and uncorrelated between years, potentially reducing the impact of systematic errors in the a priori emissions model. Furthermore, with a few exceptions [16], most work has derived OH separately to CH_4 and its global $^{13}C/^{12}C$ source signature, limiting the propagation of uncertainty in OH through to the derived CH₄ fluxes. The inability to quantify CH₃CCl₃ systematic emissions uncertainties may be particularly problematic in recent years, when, as a result of its production and consumption ban under the Montreal Protocol, reported consumption has dropped to very low levels, but evidence of continued emissions can still be seen in atmospheric observations (Figure S1)[17, 18]. Therefore, the assumptions that were used in early estimates of CH₃CCl₃ emissions, which were based on industry surveys at a time when CH₃CCl₃ was widely used[19], are unlikely to hold in recent decades.

In contrast to previous approaches, the method used in this

paper explicitly includes a model of the CH₃CCl₃ emissions 187 processes in the estimation scheme. Information regarding the global emissions of long-lived trace gases such as CH₃CCl₃ can 189 be derived simultaneously with their atmospheric sinks, by 190 jointly considering factors such as the long-term trend in concentration and the inter-hemispheric gradient [20]. We extend 192 this approach here by including the uncertain emissions and at- 193 mospheric model parameters jointly in a hierarchical Bayesian 194 estimation framework that is informed by atmospheric data 195 from multiple species. This ensures that uncertainties in each 196 component are propagated throughout the system. A full list 197 of model parameters explored in the inversion is given in Table

198

199

200

201

204

205

206

213

217

218

228

229

230

231

232

233

234

235

236

237

238

239

240

241

242

243

244

245

246

In order to focus on the uncertainties in the CH₃CCl₃ emissions model, we chose to use a computationally efficient "box model" of atmospheric transport and chemistry that included two tropospheric boxes and one stratospheric box. 203 Previous authors have noted that the use of atmospheric box models with annually repeating transport can cause erroneous fluctuations in derived OH concentrations over periods of around three years or less, particularly during periods when 207 emissions of CH₃CCl₃ were relatively large[15]. However, 208 recent studies have shown that, at least in recent years when atmospheric CH₃CCl₃ gradients are small, OH inversions based on box models agree very closely (to within $\sim 1\%$) with three- 211 dimensional model inversions using analysed meteorology[13], or that OH variations derived using box models can be used to simulate realistic CH₃CCl₃ trends using three-dimensional models[14]. Therefore, in this paper, we primarily focus on longer-term OH trends, and we expect that our findings for recent decades would not be substantially different if a more complex model were used.

The atmospheric and emissions model parameters were 219 constrained in a multi-species inversion using monthly mean 220 observations of atmospheric CH₃CCl₃ from both the Advanced 221 Global Atmospheric Gases Experiment (AGAGE)[21] and Na- 222 tional Oceanic and Atmospheric Administration (NOAA)[4, 223 13] networks, along with NOAA CH₄ data and ¹³C-CH₄ observations from the University of Colorado's Institute of Arctic and Alpine Research (INSTAAR)[22, 23] (Figures 1). Colocated AGAGE and NOAA observations were found exhibit somewhat different long-term CH₃CCl₃ trends. Therefore, two sets of inversions were performed, based on the CH₃CCl₃ observations from each network (Figure S2). AGAGE CH₄ observations were not used in the main part of this study as they were found to agree very closely with NOAA data, but cover a shorter time period. Further details of about the observations are provided in the supplementary materials and methods, and the site locations are shown in Table S2.

Results

The first two panels of Figure 2 show the simultaneouslyderived OH concentrations and CH₃CCl₃ emissions inferred from independent application of our approach using AGAGE or NOAA observations. A comparison between the observations and the model is shown Figure S3 and numerical values for quantities in the figure are provided in the Supplement. The median solution shows a relatively small OH trend in the 1980s and 1990s (with smaller inter-annual variability than previous CH₃CCl₃ inversions[11, 12, 24]), followed by an upward trend in OH concentration on the order of 10% from the late 1990s

125

126

127

128

129

130

131

132

133

134

135

136

137

138

139

140

141

142

143

144

145

146

147

148

149

150

151

152

153

154

155

156

157

158

159

160

161

162

163

164

165

166

167

168

169

170

171

172

173

174

175

176

177

178

179

180

181

182

183

184

185

249 to 2004 (11 \pm 13% and 9 \pm 12% increase for AGAGE and NOAA, respectively, between 1998 and 2004). This trend is of a similar size to those highlighted in previous studies using 252CH₃CCl₃[14, 24]. Post-2004, our median estimate shows a 253decline in OH. This finding would suggest that at least some 254fraction of the post-2007 CH₄ growth could be attributable to 255 declining OH. By carrying out a set of linear regressions on 256 the post-2007 OH estimates from our a posteriori ensemble of model states, we find a 70% or 64% probability that OH 257 258 exhibited some level of negative trend during this period, when 259 AGAGE or NOAA data were used, respectively (the mean 260 difference between the 2004 and 2014 OH concentrations was -8261 \pm 11 % and -11 \pm 11 %, respectively). In addition to this trend 262are several features of our OH inversion that are important 263to note. Firstly, significant uncertainties remain in the global OH concentration, such that it is possible to draw a "constant 264 265OH" line that is consistent with the observation-derived OH, 266 within its uncertainties. Secondly, small differences in the 267 CH₃CCl₃ trend and inter-hemispheric gradient measured by 268the two independent networks lead to variations in the derived 269 OH concentration and CH₃CCl₃ emissions. However, these 270 differences are small compared to the other uncertainties in 271 the system.

Differences between our derived CH₃CCl₃ emissions and those assumed previously (Figure 2, second panel) explain part of the discrepancy between our OH trends and those derived in previous studies (Figure 2, first panel), although other factors such as the treatment of the ocean sink also contribute (see SI text). Our global CH₃CCl₃ emissions estimates differ to the previous estimates shown in the Figure in that they have been adjusted in the inversion to be consistent with atmospheric observations (and in particular, the inter-hemispheric CH₃CCl₃ mole fraction gradient), instead of being imposed based on bottom-up models or an assumed rate of decline [13, 24]. The CH₃CCl₃ emissions derived in our inversion indicate that there was ongoing release of CH₃CCl₃ to the atmosphere, at least through 2014, despite national reports indicating that use of this substance ceased in 2013[25]. Analysis of high-frequency AGAGE data confirms that emissions persisted throughout this period, upwind of some monitoring sites (Figure S1).

In addition to our multi-species inversion, we carried out an inversion for OH concentrations and CH₃CCl₃ emissions using only CH₃CCl₃ observations (Figure S4). We find that the OH concentrations and variability derived in this analysis leads to a similar result to the multi-species inversion, indicating that the constraint on OH is primarily from CH₃CCl₃, rather than CH₄ and its ¹³C/¹²C ratio. Therefore, the timing of the rise and fall in inferred OH has not been significantly influenced by "knowledge" of the pause and renewed growth in CH₄.

Our multi-species inversion allows us to propagate information on the derived OH concentration, and its uncertainty, through to estimates of CH₄ emissions. We find that, similarly to OH concentration, it is possible to draw a "constant CH₄ emissions" line within the derived uncertainties (Figure 2, third panel). However, the median solution suggests a relatively steady upward trend from the mid-1990s to the mid-2000s, followed by a period of smaller growth. We note that our result does not require a sudden, statistically significant, increase in CH₄ emissions in 2007, as suggested elsewhere, to explain the 309 observations [5-7, 26, 27]. Instead, it is implied that the rise 310 in atmospheric mole fractions in 2007 is consistent with the decline in OH concentrations post-2004, overlaid on a gradual rise in CH₄ emissions with some additional inter-annual variability on the order of 10 Tg yr^{-1} .

311

312

313 314

315

316

317

318

319

320 321

322

323

324

325

326

327

328 329

330

331

332

333

334

335

336

337

338

339

340

341

342

343

344

345

346

347

348

349

350

351

352

353

354

355

356

357

358

359

360

361

362

363

364

365

366

367

368

369

370

371

372

The third panel in Figure 2 also shows an inversion where OH is constrained to be inter-annually repeating. In this scenario, CH₄ emissions remain at a relatively low level throughout the 2000s, compared to the varying-OH inversions, until around 2007, when they sharply increase. Compared to the 5-year period before 2007, emissions from 2007 to 2011 (inclusive) were $22\pm9~{\rm Tg~yr}^{-1}$ higher in this scenario (similarly to other studies that had assumed constant OH[28]). In contrast, for the inversions with the OH changes derived from AGAGE or NOAA CH₃CCl₃, this difference was found to be 4±23 Tg yr^{-1} or 9 ± 22 Tg yr^{-1} , respectively.

In our inversion, we determine the global ¹³CH₄/¹²CH₄ source signature that would be required to match the observed atmospheric δ^{13} C-CH₄ (see SI materials and methods), considering changes in OH and global CH₄ emissions (Figure 2, fourth panel). The observations and modeling framework provide relatively weak constraints on this term, such that the uncertainties on annual ¹³CH₄/¹²CH₄ source ratios are around an order of magnitude larger, at around 1‰, than the changes that would be required to match the observed trends, which are of the order of 0.1%. Furthermore, we find that, due to the very long timescales over which methane isotopologues respond to source or sink perturbations[29], our derived source ratio values are significantly auto-correlated, meaning that, in our inversion, the derived annual values cannot be considered fully independent of one another (Figure S5).

272

273

274

275

276

277

278

279

280

281

282

283

284

285

286

287

288

289

290

291

292

293

294

295

296

297

298

299

300

301

304

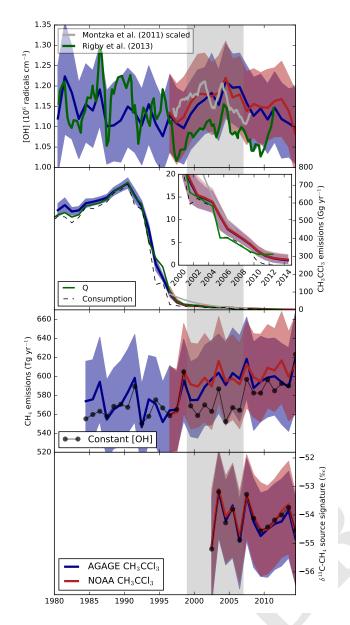


Fig. 2. First panel: inferred tropospheric annual mean OH concentration: second panel: global CH₃CCl₃ emissions; third panel: global CH₄ emissions; fourth panel: ¹³C/¹²C source isotope ratio of CH₄. The blue lines and shading show quantities inferred when AGAGE CH3CCl3 were used, and red lines and shading shows those inferred using NOAA CH3CCI3. Lines indicate the median and the shading shows the 16th to 84th percentile (approximately ± 1 sigma). The green and grey lines in the top two panels show estimates from previous studies that used the same observations, but different methodologies and emissions[13, 24]. The black line in the lower two panels shows the methane and isotopologue changes inferred when inter-annually repeating OH was used. The grey shading shows the approximate start and end of the methane "pause". Numerical values for the top two panels are available in the Supplementary Material.

Discussion

We have presented an inversion that derives global OH concentrations simultaneously with CH₃CCl₃ and CH₄ emissions and the ¹³CH₄/¹²CH₄ source ratio, using observations of CH₃CCl₃, $\mathrm{CH_4}$ and $\delta^{13}\mathrm{C\text{-}CH_4}.$ Our median solution shows that OH increased from the late 1990s until 2004, before declining until 2014, albeit with an uncertainty that is of similar magnitude to the change. The median solution suggests that OH changes

have contributed to the recent pause and growth in CH₄, as re- 435 flected in the median CH₄ emissions, which only change slowly 436 after the late 1990s. In contrast, our "constant OH" inversion 437 shows a relatively sudden emissions increase in 2007. It is in-438 teresting to note that these two sets of derived emissions agree 439 relatively well during the 1990s (at levels of approximately 440 560 Tg yr^{-1}), and after 2010 (approximately 600 Tg yr⁻¹), 441 but the trajectory of the transition is different, with most 442 of the increase occurring in the late 1990s if OH is allowed 443 to change, but primarily around 2007 if it isn't. However, 444 it is also important to note that the median solution of the 445 constant OH inversion falls within the 1-sigma range of the 446 "varying OH" inversions.

Notwithstanding the uncertainties, our findings are in contrast to recent work in which a three-dimensional model of atmospheric transport and chemistry predicted only a gradual decrease in methane lifetime over the last three decades, and therefore that emissions changes were primarily responsible for the CH₄ growth[7]. We also provide an alternative perspective to another study that attributed much of the recent growth in CH_4 and $\delta^{13}C$ - CH_4 to tropical wetland emissions, based partly on the finding that there was no clear signal of an OH change in other reduced chemical tracers (CH₃CCl₃ had not been considered)[6]. Other authors had investigated and ruled-out OH changes as being the sole driver of recent trends, in studies that used δ^{13} C-CH₄ and ethane (C₂H₆) to assign the growth in methane to livestock and oil and gas extraction, respectively [5, 26].

Forward model simulations with our derived OH and a constant $^{13}\text{C-CH}_4$ source show a decline in atmospheric $\delta^{13}\text{C-CH}_4$ post-2006, demonstrating that OH trends likely contributed to the recent δ^{13} C-CH₄ trends in our inversion (Figure S6). Whilst the precise contribution of OH to the observed trend is difficult to isolate from other influences, it is likely that our derived changes are not sufficient to explain the entire recent decline in δ^{13} C-CH₄, and that some change in the source signature has also occurred, as has been suggested previously [26]. However, as described above, the uncertainties on the source signature in our inversion are much larger than the required change in source signature, making the precise identification of a change in one or more source sectors difficult.

Some recent studies have pointed to an "upturn" in global 477 concentrations of ethane (C_2H_6) , coincident with the recent 478 rise in $CH_4[5, 30, 31]$, which may imply an increase in CH_4 479 emissions due to an increase in oil and gas extraction. Column averaged measurements in the background atmosphere reveal trends in C_2H_6 between 2007 and 2014 of 23 (18, 28) pmol $\text{mol}^{-1} \text{ yr}^{-1} \text{ and } -4 \text{ (-6, -1)} \text{ pmol } \text{mol}^{-1} \text{ yr}^{-1} \text{ (95 percent)}$ confidence intervals) in the northern and southern hemispheres, respectively[5]. Because C₂H₆ is primarily removed from the atmosphere via reaction with OH, we also expect changes in OH to have an impact on C₂H₆ concentrations, even if emissions have not changed. By running our model forward 488 with constant C_2H_6 emissions (which were tuned to match the mean northern and southern hemispheric observed mole fractions[5], Figure S7) and our derived OH concentrations, 491 we find that it is possible to explain a global background C_2H_6 growth rate of 9 (-11, 30) pmol mol⁻¹ yr⁻¹ and 3 (-4, 11) pmol mol⁻¹ yr⁻¹ (95 percent confidence interval) in the 494 northern and southern hemispheres respectively from 2007 to 495 2014. The timing of transition from declining to growing C₂H₆ 496

447

448

450

451

452

453

454

455

456

457

458

459

460

461

462

463

464

465

466

467

468

469

470

471

472

473

474

475

476

485

486

487

373

374

375

376

377

378

379 380

381

382

383

384

385

386

387

388

389

390

391

392

393

394

395

396

397

398

399

400

401

402

403

404

405

406

407

408

409

410

411

412

413

414

415

416

417

418

419

420

421

422

423

424

425

426

427

428

429

430

431

432

433

497 mole fractions in the northern hemisphere coincides to within one or two years with change from growing to declining OH in our inversion (Figure S7). Therefore, it is possible that some 500 of the recent upturn in northern hemispheric C₂H₆ is also 501 due to changes in OH concentration. Our constant emissions 502 simulation does not match the continued downward trend in 503 southern hemispheric C_2H_6 , although the uncertainties in our 504 estimates overlap with the observed trend.

As we stress above, it is important to note the magnitude of 506 the uncertainties in our inversions, which we believe are more comprehensive than previous work, as they incorporate several systematic factors, particularly relating to CH₃CCl₃ emissions. If OH changes and their uncertainty are not considered, a sudden and statistically significant increase in CH₄ emissions after 2006 is required to fit the observations. Whilst we cannot rule out this scenario, in our inversions in which the recent CH₃CCl₃ budget is objectively considered, a trajectory in which CH₄ emissions have changed more gradually during the late 2000s is also plausible. Our study highlights that, without careful consideration of the CH₄ sink and its uncertainty, it would be possible to draw misleading conclusions regarding the emissions trend when long-term records of background atmospheric observations are used. Our median estimate suggests an important role for OH in the recent CH₄ pause and growth, overlaid on a relatively gradual increase in CH₄ emissions over the last two decades.

Materials and Methods

507

509

512

513

514

515

516

517

521

522

523

524

525

526

527

528

529

530

531

532

533

534

535

536

537

538

539

540

541

542

543

544

545

546

547

548

549

550

551

552

553

554

555

556

Atmospheric mole fractions were simulated using a box model atmosphere, which accounted for mixing between the two tropospheric hemispheres, and exchange with the stratosphere. Loss of CH₃CCl₃ and CH₄ occurred primarily through reaction with OH in the model troposphere (with the potential for differences in the northern and southern OH concentration[32]). The model also included a firstorder loss of each compound in the stratosphere (all stratospheric losses were considered to contribute to a single stratospheric loss rate), first-order sinks for CH₄ in the troposphere due to reaction with chlorine and uptake by methanotrophs in soils[1], and an ocean uptake for CH₃CCl₃ according to previous ocean model estimates[33]. Isotopic fractionation of CH₄ was assumed to occur for each sink based on recent estimates [34-37]. Emissions of CH₃CCl₃ were estimated using a model that took as an input consumption or use of CH₃CCl₃. Uncertain parameters in the atmospheric and emissions model were estimated in the inversion, along with estimates of the annual, hemispheric CH₄ surface flux and ${}^{13}\mathrm{CH_4}/{}^{12}\mathrm{CH_4}$ source signature and global, annual OH concentration. By exploring some of the major unknown parameters in this multi-species framework, the influence of uncertainties in each parameter and the atmospheric data could be propagated through the system (see Table S1 for a list of model parameters). AGAGE, NOAA and INSTAAR data (Figure 1) were used to constrain the model parameters using a hierarchical Bayesian framework, which was solved using a Markov Chain Monte Carlo (MCMC) algorithm[38]. The MCMC approach iteratively explores model states, randomly accepting or rejecting proposed parameter values with a probability dependent on the ratio of posterior probability density of the "current" and proposed states. The outcome is a chain of parameter values that span the posterior probability density functions. Atmospheric data from a subset of the three networks were used, where predominantly "background" (unpolluted) air masses were sampled, and where time series of the order of a decade or more were available (SI Text). The delta notation for observations of $^{13}\text{C}/^{12}\text{C}$ ratio in CH₄ is defined as:

$$\delta^{13}C - CH_4 = 1000 \left(\frac{R}{R_{std}} - 1\right)$$
 [1]

where R is the $^{13}\text{C}/^{12}\text{C}$ ratio in CH₄, and R_{std} refers to a reference ratio[39] and values are quoted in per mille (‰). Further details are provided in the SI Text.

ACKNOWLEDGMENTS. Matt Rigby is supported by a NERC Advanced Research Fellowship (NE/I021365/1). The operations of the AGAGE instruments at Mace Head, Trinidad Head, Cape Matatula, Ragged Point, and Cape Grim are supported by the National Aeronautics and Space Administration (NASA) (grants NAG5-12669, NNX07AE89G, and NNX11AF17G to MIT and grants NNX07AE87G, NNX07AF09G, NNX11AF15G, and NNX11AF16G to SIO), the Department of Energy and Climate Change (DECC, UK) contract GA01081 to the University of Bristol, and the Commonwealth Scientific and Industrial Research Organisation (CSIRO Australia) and Bureau of Meteorology (Australia). We are indebted to E. Dlugokencky for his continuing efforts to produce the NOAA CH₄ dataset, and for his helpful comments on our manuscript. NOAA measurements are supported in part by the NOAA Climate Program Office's AC4 program and benefited from the technical assistance of C. Siso, B. Hall, G. Dutton, and J. Elkins.

559

560

561

562

563

564

565

566

567

568

569

570

571

572

573

574

575

576

577

578

579

580

581

582

583

584

585

586

587

588

589

590

591

592

593

594

595

596

597

598

599

600

601

602

603

604

605

606

607

608

609

610

611

612

613

614

615

616

617

618

619

- 1. Kirschke S et al. (2013) Three decades of global methane sources and sinks. Nature Geoscience 6(10):813-823.
- 2. Dlugokencky EJ et al. (2003) Atmospheric methane levels off: Temporary pause or a new steady-state? Geophysical Research Letters 30(19):3-6
- 3. Rigby M et al. (2008) Renewed growth of atmospheric methane. Geophysical Research Letters 35(22):L22805
- Dlugokencky EJ et al. (2009) Observational constraints on recent increases in the atmospheric CH4 burden. Geophysical Research Letters 36:L18803.
- Hausmann P, Sussmann R, Smale D (2016) Contribution of oil and natural gas production to renewed increase in atmospheric methane (2007-2014): top-down estimate from ethane and methane column observations. Atmospheric Chemistry and Physics 16(5):3227-3244.
- Nisbet EG et al. (2016) Rising atmospheric methane: 2007-2014 growth and isotopic shift. Global Biogeochemical Cycles 30(9):1356-1370.
- Dalsøren SB, Isaksen ISA (2006) CTM study of changes in tropospheric hydroxyl distribution 1990-2001 and its impact on methane. Geophysical Research Letters 33:L23811.
- Voulgarakis A et al. (2013) Analysis of present day and future OH and methane lifetime in the ACCMIP simulations. Atmospheric Chemistry and Physics 13(5):2563-2587
- Lovelock JE (1977) Methyl chloroform in the troposphere as an indicator of OH radical abundance. Nature 267(5606):32-32.
- Prinn RG et al. (2001) Evidence for substantial variations of atmospheric hydroxyl radicals in the past two decades. Science 292:1882-1888.
- 11. Prinn RG et al. (2005) Evidence for variability of atmospheric hydroxyl radicals over the past quarter century. Geophysical Research Letters 32:L07809.
- Bousquet P. Hauglustaine DA. Pevlin P. Carouge C. Ciais P (2005) Two decades of OH variability as inferred by an inversion of atmospheric transport and chemistry of methyl chloroform. Atmospheric Chemistry and Physics 5:2635-2656
- 13. Montzka SA et al. (2011) Small Interannual Variability of Global Atmospheric Hydroxyl. Science 331(6013):67-69.
- 14. McNorton J et al. (2016) Role of OH variability in the stalling of the global atmospheric CH4 growth rate from 1999 to 2006. Atmospheric Chemistry and Physics 16(12):7943-7956.
- Krol M. Lelieveld J (2003) Can the variability in tropospheric OH be deduced from mea surements of 1.1.1-trichloroethane (methyl chloroform)? Journal of Geophysical Research 108(D3):4125.
- Pison I. Bousquet P. Chevallier F. Szopa S. Hauglustaine D (2009) Multi-species inversion of CH4, CO and H2; emissions from surface measurements. Atmospheric Chemistry and Physics 9(14):5281-5297
- 17. Krol MC et al. (2003) Continuing emissions of methyl chloroform from Europe. Nature 421(6919):131-135.
- Reimann S et al. (2005) Low European methyl chloroform emissions inferred from long-term atmospheric measurements. Nature 433(7025):506-8.
- McCulloch A, Midgley PM (2001) The history of methyl chloroform emissions: 1951-2000. Atmospheric Environment 35(31):5311-5319.
- Liang Q et al. (2014) Constraining the carbon tetrachloride (CCl4) budget using its global trend and inter-hemispheric gradient. Geophysical Research Letters 41(14):5307-5315.
- Prinn RG et al. (2000) A history of chemically and radiatively important gases in air deduced from ALE/GAGE/AGAGE. Journal of Geophysical Research 105(D14):17751-17792.
- White J, Vaughn BH (2015) University of Colorado, Institute of Arctic and Alpine Research (INSTAAR), Stable Isotopic Composition of Atmospheric Methane (13c) from the NOAA ESRL Carbon Cycle Cooperative Global Air Sampling Network, 1998-2014, Version: 2015-08-03.
- Miller JB et al. (2002) Development of analytical methods and measurements of 13c/12c in atmospheric CH4 from the NOAA Climate Monitoring and Diagnostics Laboratory Global Air Sampling Network. Journal of Geophysical Research 107(D13).
- Rigby M et al. (2013) Re-evaluation of the lifetimes of the major CFCs and CH3ccl3 using atmospheric trends. Atmospheric Chemistry and Physics 13(5):2691-2702.
- UNEP (2016) UNEP Ozone Secretariat Data Centre.
- Schaefer H et al. (2016) A 21st-century shift from fossil-fuel to biogenic methane emissions indicated by 13ch4. Science 352(6281):80-84
- Turner AJ et al. (2016) A large increase in U.S. methane emissions over the past decade inferred from satellite data and surface observations. Geophysical Research Letters
- Bergamaschi P et al. (2013) Atmospheric CH4 in the first decade of the 21st century; Inverse modeling analysis using SCIAMACHY satellite retrievals and NOAA surface measurements. Journal of Geophysical Research: Atmospheres 118(13):7350-7369

621 29. Tans PP (1997) A note on isotopic ratios and the global atmospheric methane budget. *Global Biogeochemical Cycles* 11(1):77.

623

624

625

626

627

628

629

630

631

632

633

634

635

636

637

638

639

640

641

642

643

644

645

646

647

648

649

650

651

652

653

654

655

656

657

658

659

660

661 662

663 664

665 666

667

668 669

670 671

672

673

674

675

676

677

678

679

680

 $681 \\ 682$

- Helmig D et al. (2016) Reversal of global atmospheric ethane and propane trends largely due to US oil and natural gas production. Nature Geoscience 9(7):490–495.
- Franco B et al. (2015) Retrieval of ethane from ground-based FTIR solar spectra using improved spectroscopy: Recent burden increase above Jungfraujoch. *Journal of Quantitative Spectroscopy and Radiative Transfer* 160:36–49.
- Patra PK et al. (2014) Observational evidence for interhemispheric hydroxyl-radical parity. Nature 513(7517):219–223.
- Wennberg PO, Peacock S, Randerson JT, Bleck R (2004) Recent changes in the air-sea gas exchange of methyl choloroform. Geophysical Research Letters 31(16):3

 –6.
- Brenninkmeijer CAM, Lowe DC, Manning MR, Sparks RJ, van Velthoven PFJ (1995) The 13c, 14c, and 18o isotopic composition of CO, CH4, and CO2 in the higher southern latitudes lower stratosphere. *Journal of Geophysical Research* 100(D12):26163–26172.
- Allan W, Struthers H, Lowe DC (2007) Methane carbon isotope effects caused by atomic chlorine in the marine boundary layer: Global model results compared with Southern Hemisphere measurements. *Journal of Geophysical Research* 112(D4).
- Saueressig G, Crowley JN, Bergamaschi P, Brenninkmeijer CAM, Fischer H (2001) Carbon 13 and D kinetic isotope effects in the reactions of CH4 with O(1d) and OH: New Laboratory measurements and their implications for the isotopic composition of stratospheric methane. *Journal of Geophysical Research* 106(D19):23127–23138.
- Lassey KR, Etheridge DM, Lowe DC, Smith AM, Ferretti DF (2007) Centennial evolution of the atmospheric methane budget: what do the carbon isotopes tell us? Atmospheric Chemistry and Physics 7(8):2119–2139.
- Hastings WK (1970) Monte Carlo sampling methods using Markov chains and their applications. Biometrika 57(1):97–109.
- Craig H (1957) Isotopic standards for carbon and oxygen and correction factors for massspectrometric analysis of carbon dioxide. Geochimica et Cosmochimica Acta 12(1-2):133– 149
- Dlugokencky EJ, Steele LP, Lang PM, Masarie Ka (1994) The growth rate and distribution of atmospheric methane. *Journal of Geophysical Research* 99(D8):17021–17043.
- O'Doherty S et al. (2001) In situ chloroform measurements at Advanced Global Atmospheric Gases Experiment atmospheric research stations from 1994 to 1998. *Journal of Geophysical Research* 106(D17):20429–20444.
- Miller BR et al. (2008) Medusa: A Sample Preconcentration and GC / MS Detector System for in Situ Measurements of Atmospheric Trace Halocarbons, Hydrocarbons, and Sulfur Compounds. Analytical Chemistry 80(5):1536–1545.
- Cunnold DM et al. (2002) In situ measurements of atmospheric methane at GAGE/AGAGE sites during 1985–2000 and resulting source inferences. *Journal of Geophysical Research* 107(D14):4225.
- Stevens CM, Rust FE (1982) The carbon isotopic composition of atmospheric methane. Journal of Geophysical Research 87(C7):4879.
- Tyler SC (1986) Stable carbon isotope ratios in atmospheric methane and some of its sources. Journal of Geophysical Research 91 (D12):13232.
- 46. Lowe DC, Brenninkmeijer CAM, Tyler SC, Dlugkencky EJ (1991) Determination of the iso-

topic composition of atmospheric methane and its application in the Antarctic. *Journal of Geophysical Research* 96(D8):15455.

684

685

686

687

688

689

690

691

692

693

697

698

699

700

701

702

703

704

705

706

707

708

709

710

711

712

713

714

715

716

717

718

719

720

721

 $722 \\ 723$

724 725

 $726 \\ 727$

728

 $729 \\ 730$

731 732

733

734

735

736

737

738

739

740

741

 $742 \\ 743$

- Patra PK et al. (2011) TransCom model simulations of CH4 and related species: linking transport, surface flux and chemical loss with CH4 variability in the troposphere and lower stratosphere. Atmospheric Chemistry and Physics 11(24):12813–12837.
- Cunnold DM et al. (1983) The Atmospheric Lifetime Experiment 3. Lifetime Methodology and Application to Three Years of CFCl3 Data. *Journal of Geophysical Research* 88(C13):8379– 8400.
- Cunnold DM et al. (1994) Global trends and annual releases of CCI3f and CCI2f2 estimated from ALE/GAGE and other measurements from July 1978 to June 1991. *Journal of Geophysical Research* 99(D1):1107–1126.
- Sander SP et al. (2011) Chemical Kinetics and Photochemical Data for Use in Atmospheric Studies: Evaluation Number 17, (NASA Jet Propulsion Laboratory), Technical Report 17.
- Spivakovsky CM et al. (2000) Three-dimensional climatological distribution of tropospheric OH: Update and evaluation. *Journal of Geophysical Research* 105(D7):8931–8980.
- Morice CP, Kennedy JJ, Rayner NA, Jones PD (2012) Quantifying uncertainties in global and regional temperature change using an ensemble of observational estimates: The HadCRUT4 data set. Journal of Geophysical Research: Atmospheres 117:D08101.
 Chipperfield MP, et al. (2013) Model Estimates of Lifetimes in SPARC Report on the Lifetimes 696
- Chipperfield MP et al. (2013) Model Estimates of Lifetimes in SPARC Report on the Lifetimes of Stratospheric Ozone-Depleting Substances, Their Replacements, and Related Species, eds. Reimann S, Ko. MKW, Newman PA, Strahan SE. (WMO/ICSU/IOC World Climate Research Programme, Zurich, Switzerland) No. SPARC Report No. 6, WCRP-15/2013.
- Ganesan AL et al. (2014) Characterization of uncertainties in atmospheric trace gas inversions using hierarchical Bayesian methods. Atmospheric Chemistry and Physics 14(8):3855

 3864
- Roberts G, Gelman A, Gilks W (1997) Weak convergence and optimal scaling of random walk Metropolis algorithms. The Annals of Applied Probability 7(1):110–120.
- Engel A et al. (2013) Inferred Lifetimes from Observed Trace-Gas Distributions in SPARC Report on the Lifetimes of Stratospheric Ozone-Depleting Substances, Their Replacements, and Related Species, eds. Reimann S, Ko. MKW, Newman PA, Strahan SE. (WMO/ICSU/IOC World Climate Research Programme, Zurich, Switzerland) No. SPARC Report No. 6, WCRP-15/2013
- Whiticar M, Schaefer H (2007) Constraining past global tropospheric methane budgets with carbon and hydrogen isotope ratios in ice. *Philosophical Transactions of the Royal Society A* 365:1793–1828.
- Snover AK, Quay PD, Hao WM (2000) The D/H content of methane emitted from biomass burning. Global Biogeochemical Cycles 14(1):11–24.
- Rigby M, Manning AJ, Prinn RG (2012) The value of high-frequency, high-precision methane isotopologue measurements for source and sink estimation. *Journal of Geophysical Re*search 117(D12):1–14.
- Levin I et al. (2012) No inter-hemispheric \$\delta\$13ch4 trend observed. Natur 486(7404):E3–E4.
- Lassey KR, Lowe DC, Manning MR (2000) The trend in atmospheric methane 13c and implications for isotopic constraints on the global methane budget. Global Biogeochemical Cycles 14(1):41–49.