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Title

Role of Nonequilibrium Water Vapor Diffusion in Thermal Energy Storage Systems in the Vadose Zone

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1	ROLE OF NONEQUILIBRIUM WATER VAPOR DIFFUSION IN GEOTHERMAL
2	ENERGY STORAGE SYSTEMS IN THE VADOSE ZONE
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5	and J.S. McCartney, Ph.D., P.E., M.ASCE ⁸
6	Abstract: Although siting of geothermal energy storage systems in the vadose zone may be
7	beneficial due to the low heat losses associated with the low thermal conductivity of unsaturated
8	soils, water phase change and vapor diffusion in soils surrounding geothermal heat exchangers
9	may play important roles in both the heat injection and retention processes that are not considered
10	in established design models for these systems. This study incorporates recently-developed
11	coupled thermo-hydraulic constitutive relationships for unsaturated soils into a coupled heat
12	transfer and water flow model that considers time-dependent, nonequilibrium water phase change
13	and enhanced vapor diffusion to study the behavior of geothermal energy storage systems in the
14	vadose zone. After calibration of key parameters using a tank-scale heating test on compacted silt,
15	the ground response during 90 days of heat injection from a vertical geothermal heat exchanger

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16 followed by 90 days of ambient cooling was investigated. Significant decreases in degree of 17 saturation and thermal conductivity of the ground surrounding the vertical geothermal heat 18 exchanger were observed during the heat injection period that were not recovered during the 19 cooling period. This effect can lead to a greater amount of heat retained in the ground beyond that 20 estimated in conduction-based design models.

21 INTRODUCTION

An important challenge facing society is the storage of energy collected from renewable 22 sources. One such application is the storage of heat collected from solar thermal panels in the 23 24 subsurface so that it can be harvested later (Claesson and Hellström 1981; Nordell and Hellström 2000; Chapuis and Bernier 2009). A practical mode of heat injection into the subsurface involves 25 circulation of a heated carrier fluid through a closely-spaced array of closed-loop geothermal heat 26 exchangers in boreholes to reach ground temperatures ranging from 35 to 80 °C (Sibbitt et al. 27 2012; Başer et al. 2016a; McCartney et al. 2017). Unsaturated soils in the vadose zone are ideal 28 29 thermal energy storage media because low heat losses can be expected due to the low thermal conductivity of unsaturated soils (McCartney et al. 2013). The mode of heat transfer during 30 injection of heat into unsaturated soils is complex as it may be coupled with thermally-induced 31 32 water flow in either liquid or vapor forms along with latent heat transfer associated with phase change. However, most design models for geothermal heat storage systems focus on ground 33 temperature changes during heating and do not consider coupled heat transfer and water transport 34 35 (Claesson and Hellström 1981; Eskilson 1987). Although some recent studies on geothermal energy storage systems highlighted the importance of considering coupled heat transfer and water 36 37 flow in their performance evaluation (Catolico et al. 2016; Moradi et al. 2016), the impact of water 38 vapor diffusion and phase change in unsaturated soils during heat injection on the heat retention

during a subsequent ambient cooling phase is an important topic that has not been investigated. 39 This paper presents simulations of the response of a low-permeability, low activity, 40 incompressible, unsaturated silt layer surrounding a single geothermal heat exchanger to 41 understand the impacts of considering water vapor diffusion and water phase change on the 42 transient heat injection and retention processes. Comparison of the simulation results from the 43 44 coupled heat transfer and water flow model with a simpler heat transfer model without water vapor diffusion or phase change permits an evaluation of the importance of these heat transfer 45 mechanisms in simulating geothermal energy storage systems in the vadose zone. 46

47 BACKGROUND

Most models of heat transfer from geothermal heat exchangers employ analytical solutions to 48 the heat equation assuming conduction is the primary mechanism of heat transfer, using constant 49 thermal properties that do not consider the effects of changes in degree of saturation expected 50 during heat transfer in unsaturated soils (e.g., Kavanaugh 1985; Eskilson 1987). Analytical 51 solutions have been developed for geothermal heat exchanger geometries including the infinite 52 line source (Ingersoll and Plass 1948; Beier et al. 2014), finite line source (Acuña et al. 2012; 53 Lamarche and Beauchamp 2007), hollow cylinder source (Ingersoll et al. 1954; Gehlin 2002), 54 55 finite plate source (Ciriello et al. 2015), and one- and two-dimensional solid cylinder sources (Tarn and Wang 2004). Although numerical simulations of geothermal heat exchangers have also been 56 performed, most have also considered conduction as the primary mechanism of heat transfer 57 58 (Ozudogru et al. 2015; Welsch et al. 2015; Başer et al. 2016a). While these conduction-based analytical models and numerical simulations may be practical for the design of geothermal heat 59 60 exchangers in dry or saturated low permeability soils, they may not be practical for design of those 61 in unsaturated soils due to the potential for convective heat transfer associated with thermally62 induced liquid water or water vapor flow, which may result in irreversible changes in behavior 63 during cyclic heat injection and extraction (or ambient cooling). Further, the thermal properties of 64 unsaturated soils are highly dependent on the degree of saturation, even when conduction is 65 assumed to be the primary mode of heat transfer (e.g., Farouki 1981; Côté and Konrad 2005; Smits 66 et al. 2013; Lu and Dong 2015). Conduction-only models may also not be practical for use in 67 saturated soils with high permeability due to the potential for thermally-induced convection of 68 water from buoyancy effects (Catolico et al. 2016).

Because the properties of water in liquid and gas forms are dependent on temperature, heat 69 70 transfer in the unsaturated soils in the vadose zone leads to thermally induced water flow through soil. Specifically, temperature dependency of the density of liquid water ρ_w (Hillel 1980), dynamic 71 viscosity of liquid water μ_w (Lide 2001), surface tension of soil water σ (Saito et al. 2006), relative 72 humidity at equilibrium R_{h,eq} (Philip and de Vries 1957), saturated vapor concentration in the gas 73 phase $c_{y,sat}$ (Campbell 1985), vapor diffusion coefficient in air D_y (Campbell 1985), and the latent 74 75 heat of water vaporization L_w (Monteith and Unworth 1990) may lead to thermally-induced water 76 flow through unsaturated soils. The movement of water in soil caused by thermal and hydraulic gradients and the associated impacts on heat transfer have been studied experimentally for more 77 than 100 years (Bouyoucos 1915; Lewis 1937; Smith 1943; Gurr et al. 1952; Baladi et al. 1981; 78 Shah et al. 1983; Ewen 1988; Gens et al. 1998, 2007, 2009; Cleall et al. 2011; Smits et al. 2011; 79 Moradi et al. 2015, 2016; Başer et al. 2016b). Some general observations from these studies are: 80 81 (1) heat transfer occurs in unsaturated porous media by conduction, convection in both liquid and gas phases, and latent heat transfer associated with water phase change; (2) water movement due 82 to a temperature gradient is controlled by both vaporization/condensation processes as well as 83 84 development of suction gradients caused by changes in water properties with temperature (i.e.,

density, viscosity, solid-liquid contact angle) and drying effects; (3) vapor diffusion may occur at greater rates than predicted by Fick's law, (4) the magnitude of thermally induced water flow depends on the initial degree of saturation; and (5) the times required to reach steady-state distributions in degree of saturation and temperature may be different depending on the coupling between the thermal and hydraulic properties of a given soil.

The governing equations for coupled heat transfer and flow of water in liquid and vapor forms 90 91 have been investigated for unsaturated porous media in nondeformable conditions (Philip and de Vries 1957; Ewen and Thomas 1989; Thomas and King 1991; Thomas and Sansom 1995; Thomas 92 93 et al. 2001; Smits et al. 2011), deformable conditions (Thomas and He 1996; Thomas et al. 1996), and in the presence of pore fluids containing salts or chemicals (Cleall et al. 2007; Olivella et al. 94 1996; Guimaraes et al. 2007, 2013). Most models for coupled heat transfer and water flow in liquid 95 and vapor forms in nondeformable unsaturated soils are based on the model of Philip and de Vries 96 (1957), who proposed the "liquid island" theory as an explanation for observations from studies 97 like Gurr et al. (1952) that vapor diffusion occurred at a faster rate than predicted by Fick's law. 98 99 Their theory is a pore-scale explanation where local thermal gradients is assumed to be higher 100 across microscopic air-filled pores than the global thermal gradient across a soil element, and 101 where water vapor diffusion is enhanced by evaporation and condensation from water held between soil particles by capillarity (liquid islands), effectively increasing the area available for 102 vapor diffusion through a soil element. They implemented their pore-scale theory on a 103 104 macroscopic scale by extending the vapor diffusion theory of Penman (1940) through inclusion of a soil-specific enhancement factor to correct the vapor diffusion rate calculated from Fick's law. 105 Cass et al. (1984) found that the enhancement factor approaches 1.0 (no enhancement) for dry soils 106 107 and increases significantly with increasing degree of saturation. Although the model of Philip and

108 de Vries (1957) has been used in many coupled heat transfer and water flow problems in 109 nondeformable soils, their model does not account for convective transport in the gas or liquid water phases, nonequilibrium phase change, vapor dispersion, or sensible heat dispersion in the 110 liquid phase (Smits et al. 2011). Of these issues, consideration of nonequilibrium phase change in 111 the model is expected to lead to more accurate identification of the appropriate vapor enhancement 112 113 factor for a given soil (Smits et al. 2011). Lozano et al. (2008) observed that phase change may become the process limiting evaporation at low saturations rather than vapor diffusion as 114 classically believed. 115

116 In the model of Philip and de Vries (1957), it is assumed that the water in liquid and gas phases are in equilibrium, which means that phase change occurs instantaneously in response to a change 117 in vapor pressure. However, experimental studies have identified that time is required for liquid 118 119 water to volatilize in response to a change in vapor pressure in a pore resulting from vapor diffusion in response to gradients in vapor pressure and/or temperature (Bénet and Jouanna 1982; Armstrong 120 et al. 1994; Chammari et al. 2008, Bénet et al. 2009). To account for this in a model of coupled 121 122 heat transfer and water flow, a source term for the liquid/gas phase change rate is added to the mass balance equations of liquid and vapor that is based on irreversible thermodynamics, first 123 124 order reaction kinetics, or the kinetic theory of gases and contains a fitting coefficient that can calibrated for a given soil (Bénet and Jouanna 1982; Bixler 1985; Zhang and Datta 2004). Smits 125 et al. (2011) adopted the source term of Bixler (1985) because it was derived from the kinetic 126 127 theory of gases and is thus inherently temperature dependent. In the model of Bixler (1985), the vaporization rate is proportional to the difference between local equilibrium vapor pressure and 128 local partial vapor pressure and the difference between the local degree of saturation and residual 129 saturation. Smits et al. (2011) compared predictions of coupled heat transfer and water flow from 130

equilibrium and nonequilibrium models, and found major differences in the early stages of the flow process, with greater differences for soils with initially lower degrees of saturation. Smits et al. (2011) and Trautz et al. (2015) also found that nonequilibrium models provide a better match to experimental data from column tests involving evaporation from fine sand with a heated surface than the model of Philip and de Vries (1957), indicating that the nonequilibrium assumption for phase change may better capture the transient process of thermally-induced drying.

137 MODEL CALIBRATION

138 Model Description

139 A non-equilibrium, non-isothermal, and coupled heat transfer and water flow numerical model developed by Smits et al. (2011) and extended by Moradi et al. (2016) was used to consider the 140 behavior of an unsaturated soil layer during heating and cooling of a single vertical geothermal 141 heat exchanger. The governing equations and primary variables used in the formulation are given 142 in Table 1. Calibration of the model requires soil-specific quantification of the parameters for the 143 thermo-hydraulic constitutive relationships governing water retention, hydraulic conductivity, 144 thermal conductivity, and volumetric heat capacity, as well as estimates of parameters a and b in 145 Equations (4) and (5) that govern the rates of vapor diffusion and phase change, respectively. 146

The model used in the simulations incorporates recently-developed thermo-hydraulic constitutive relationships for unsaturated soils (Lu and Dong 2015; Baser et al. 2016c). The experimental approach used by Lu and Dong (2015) was used to obtain the data for calibration of these coupled thermo-hydraulic constitutive relationships. Lu and Dong (2015) used a modified form of the transient-release and imbibition method (TRIM) of Wayllace and Lu (2012) that included a dual-needle thermal probe to measure the thermal conductivity and volumetric heat capacity during monotonic drying of different unsaturated soils under isothermal conditions. 154 TRIM uses an inverse analysis to estimate the parameters of the soil-water retention curve (SWRC) and hydraulic conductivity function (HCF) given by van Genuchten (1980). These parameters 155 156 include α_{vG} , which represents the inverse of the air entry suction in the SWRC, N_{vG}, which represents the pore size distribution in the SWRC, and k_{sw}, which is the hydraulic conductivity of 157 158 saturated soil. The value of k_{sw} obtained from a test at room temperature can be used to calculate 159 the intrinsic permeability κ in Equation (1). Although the saturation-dependent relative 160 permeability to water (the HCF) was assumed not to vary with temperature, the hydraulic conductivity of the unsaturated soil will vary with temperature because the dynamic viscosity and 161 density of water vary with temperature according to the relationships presented in Lide (2001) and 162 Hillel (1980), respectively. The relative permeability to gas was not measured in this study, but 163 was assumed to equal $k_{rg}=1-k_{rw}$. The temperature-dependent surface tension σ relationship 164 presented by Saito et al. (2006) was used in the temperature correction for capillary pressure of 165 Grant and Salehzadeh (1996), given as follows: 166

$$P_{c}(T) = P_{c}(T_{ref})[\sigma(T)/\sigma(T_{ref})]$$
(7)

167 where σ is the surface tension (N/m), T is the temperature (K), and T_{ref} is the initial reference 168 temperature of 293.15 K.

Lu and Dong (2015) defined a thermal conductivity function (TCF) that can capture transitions in the thermal conductivity in the capillary, funicular, and pendular water retention regimes of the SWRC, given as follows:

$$\frac{\lambda - \lambda_{\rm dry}}{\lambda_{\rm sat} - \lambda_{\rm dry}} = 1 - \left[1 + \left(\frac{S_{\rm rw}}{S_{\rm f}}\right)^m\right]^{1/m-1}$$
(8)

where λ_{dry} and λ_{sat} are the thermal conductivities of dry and saturated soil specimens, respectively, S_f is a parameter representing the degree of saturation at the onset of the funicular regime, and *m* 174 is a parameter related to the pore fluid network connectivity. Lu and Dong (2015) correlated the parameters of the TCFs and SWRCs of several soils and found that the *m* parameter in the TCF is 175 related to the pore-size parameter $N_{\nu G}$ in the SWRC model of van Genuchten (1980), and can be 176 estimated to be $3.0-0.2N_{\rm vG}$. Evaluation of the form of Equation 8 indicates that the thermal 177 conductivity will not reduce to the value of λ_{sat} when S_e=1, so Lu and Dong (2015) treated λ_{sat} as 178 a fitting parameter. Although Smits et al. (2013) observed that the TCF may vary with temperature, 179 180 this temperature dependency is likely due to vapor diffusion and phase change that was not 181 accounted for in their simulations. Because the simulations in this study account for vapor diffusion and phase change explicitly, the TCF and VCHF measured at 20 °C were used in the 182 183 coupled heat transfer and water flow simulations.

Başer et al. (2016c) presented trends in the volumetric heat capacity of compacted silt during monotonic drying and found that it also depends on the degree of saturation in a similar manner to the thermal conductivity, and defined a volumetric heat capacity function (VCHF) that has the same form as the THF of Lu and Dong (2015), as follows:

$$\frac{C_{v} - C_{v \, dry}}{C_{v \, sat} - C_{v \, dry}} = 1 - \left[1 + \left(\frac{S_{rw}}{S_{f}}\right)^{m}\right]^{1/m-1}$$
(9)

where C_{vdry} and C_{vsat} are the volumetric heat capacities of dry and saturated soil, respectively, and are similarly treated as fitting parameters, and S_f and *m* are the same parameters as in Equation (8). Başer et al. (2016c) found that this model and the assumptions regarding the parameters provided a good match to the volumetric heat capacity data measured in the TRIM tests on different soils performed by Lu and Dong (2015) that were not reported in their paper due to its focus on the thermal conductivity.

195 Calibration of Thermo-Hydraulic Constitutive Relationships

196 The soil investigated in this study is Bonny silt, which is classified as ML (inorganic silt) according to the Unified Soil Classification System (USCS), and has a specific gravity of 2.65. 197 Silt was selected for this evaluation because it is not expected to deform significantly during 198 changes in temperature or degree of saturation, and its low activity of 0.33 (plasticity index of 4 199 divided by clay size fraction of 12%) indicates that it will not have significant diffuse double layer 200 effects which could complicate thermo-hydraulic analyses. The silt specimens used in the 201 calibration process were prepared using compaction at a gravimetric water content of 13.7% to a 202 dry unit weight of 14.0 kN/m³, which correspond to an initial degree of saturation of 0.42 and a 203 porosity of 0.46. For reference, the optimum water content and the maximum dry unit weight 204 corresponding to the standard Proctor compaction effort are 13.6% and 16.3 kN/m³ respectively. 205

206 The SWRC and HCF (in terms of the relative permeability to water) obtained from the modified TRIM test on compacted Bonny silt are shown in Figure 1(a) along with the model 207 parameters. The shape of the SWRC indicates that an appreciable amount of water will be retained 208 209 in the soil several meters above the water table under hydrostatic conditions. The compaction conditions for these curves are the same as those mentioned above, even though Lu and Dong 210 (2015) report a different porosity due to a lower value of G_s used in their calculations. An intrinsic 211 permeability of 1.27×10^{-14} m² was calculated from the hydraulic conductivity of saturated soil of 212 1.24×10⁻⁷ m/s and the values of water viscosity and density at 20 °C (293.15 K). The TCF and 213 214 VHCF for Bonny silt are shown in Figure 1(b), along with the TCF and VHCF parameters in Equations (8) and (9). The value of m = 2.62 measured in the experiment for Bonny silt was used 215 in the simulations, which is lower than the value of 2.68 obtained from the correlation between m 216 217 and N_{vG} of Lu and Dong (2015). The experimental value still reflects the coupling between the

thermo-hydraulic properties as they were defined in the same test. The thermal conductivity ranges from 1.25 to 0.37 W/mK for saturated to dry conditions, respectively, while the volumetric heat capacity ranges from 2.75 to 1.30 MJ/m³K for saturated to dry conditions, respectively. The variations in these parameters with degree of saturation indicate that changes in heat retention within a storage volume may occur if a soil experiences drying during heat injection.

223 Calibration of Vapor Diffusion and Phase Change Parameters

To define the parameters a and b, a tank-scale heat injection experiment was performed in an 224 instrumented layer of compacted Bonny silt, which was then simulated using the parameters from 225 the thermo-hydraulic constitutive relationships defined in Figure 1. A schematic of the 226 experimental setup is shown in Figure 2. Bonny silt was compacted in 9 lifts in a cylindrical 227 aluminum container having a diameter of 550 mm and a height of 477 mm. A 215 mm-long 228 229 cylindrical cartridge heater having a diameter of 10 mm was used as the heating source. During 230 heat injection, a temperature control unit was used to impose a constant temperature boundary condition of 60 °C on the heating rod. To monitor changes in temperature and degree of saturation 231 232 during heating of the soil, ten 5TM dielectric sensors manufactured by Decagon Devices of 233 Pullman, WA were placed at the locations shown in Figure 2. After all the lifts and sensors were placed, the top of the soil layer was covered with several layers of plastic wrap to minimize loss 234 235 of water vapor to the laboratory air. The top and sides of the tank were then wrapped in insulation, 236 and thermocouples were used to monitor the temperatures of the boundaries of the tank. The soil 237 had an initial temperature of 23.5 °C and an initial degree of saturation of 0.42.

In the simulations of the tank-scale tests, no mass flux boundary conditions were applied for both liquid water and vapor flow for all boundaries of the tank. The top boundary was thermally insulated, convective heat flux boundaries were defined for the side boundaries to consider heat loss from the tank, a constant temperature boundary condition was used for the heating rod, and a 242 constant temperature of 18 °C was applied at the bottom of tank. The system of partial differential equations in Table 1 was solved simultaneously using COMSOL Multiphysics software. Results 243 from the numerical analyses were then compared with the experimental results to calibrate the 244 parameters a and b. Comparisons of predicted and measured time series of temperature and degree 245 246 of saturation inferred from dielectric sensor #3 are shown in Figures 3(a) to 3(d). Sensor #3 was 247 selected as the primary location for calibration of the model as it is near the center of the heating rod and is relatively close to the heat exchanger. The predicted time series in these figures include 248 curves for different values of the fitting parameters a and b. The parameters are observed to have 249 250 a greater effect on the change in degree of saturation as they control the rates of vapor diffusion and water phase change. Simulations from a model where no vapor diffusion or phase change is 251 considered (hereafter referred to as the "no vapor" case) are also shown in Figures 3(a) and 3(b), 252 253 which indicate slower increases in temperature to lower magnitudes at this location as well as a negligible change in degree of saturation. Values of a = 30 and $b = 5 \times 10^{-7}$ s/m² were found to best 254 fit the Bonny silt data based on visual inspection, a similar approach used by Smits et al. (2011). 255 256 To evaluate the calibration, the spatial distributions of temperature and degree of saturation along Transects B and A at the end of the heating from the numerical simulations and the 257 258 experiments are shown in Figures 4(a) through 4(d). In most of the cases, the predicted profiles show good agreement with the measured data, except in the case of the degree of saturation 259

measured by the sensor nearest the edge of container in Figure 4(b). This sensor may have malfunctioned due to the compaction process. Overall, the comparisons between model predictions and experimental results in Figures 3 and 4 indicate that the calibrated values of a and b can be assumed to be representative of Bonny silt under these compaction conditions.

264 EVALUATION OF VAPOR DIFFUSION AND PHASE CHANGE AROUND A FIELD 265 SCALE GEOTHERMAL HEAT EXCHANGER

266 Scenario Considered

The main goal of this study is to use the calibrated parameters to understand the changes in the 267 behavior of a layer of unsaturated Bonny silt surrounding a geothermal heat exchanger during a 268 heating and cooling cycle representative of geothermal energy storage systems. Although 269 geothermal heat storage systems typically involve an array of geothermal heat exchangers with 270 spacings as close at 1.5 m (Baser et al. 2016a), this study focuses on the changes in soil behavior 271 272 around a single geothermal heat exchanger. This choice simplifies the boundary conditions and permits evaluation of the relative effects of the different heat transfer mechanisms. It is possible 273 that the close spacing between geothermal heat exchangers may lead to different distributions in 274 temperature and degree of saturation than those observed in this evaluation due to interactions 275 between heat exchangers, but the simpler scenario of a single vertical geothermal heat exchanger 276 is evaluated in this paper to help establish the impact of a heating-cooling cycle on the distributions 277 in temperature and degree of saturation in the surrounding unsaturated silt layer. 278

The vertical geothermal heat exchanger investigated in this study has a length of 25 m and a 279 280 radius of 0.04 m, embedded at a depth of 1 m from the surface. The embedment is consistent with the practice of installing geothermal heat exchangers below the frost depth. Even though this 281 scenario could be investigated using an axisymmetric analysis, a 3D simulation was performed for 282 283 a rectangular domain so that the domain could be modified to incorporate additional geothermal heat exchangers in future studies. The quarter domain having a height of 30 m and a width of 10 m 284 285 with the geothermal heat exchanger along one edge is shown in Figure 5. The entire domain was assumed to be uniform and isotropic, and the soil was discretized into 101,073 elements (394,394 286

degrees of freedom) with finer elements around the heat exchanger. The hydraulic and thermal 287 boundary conditions for the models are also shown in Figure 5. For liquid water and gas flow, 288 Neumann boundary conditions (no mass flux) were assumed for all boundaries except the bottom 289 290 boundary, which was set to be a constant head boundary condition corresponding to the water 291 table. For heat transfer, a constant temperature that represents an average mean subsurface soil temperature of 21 °C was applied at the bottom while at the outer boundaries the temperature 292 293 varied with depth. No flux boundary conditions were applied to the planes of symmetry. The size of the domain was selected to be large enough that a constant temperature and zero fluid flux could 294 295 be assumed on the outer vertical boundaries.

The initial conditions are shown in the color bars in Figure 5. The initial ambient temperature of the domain was assumed to be a function of depth until a certain depth of 9 m from surface, and this initial temperature profile is a representative of early summer months in San Diego (specifically May 2015). A hydrostatic initial condition was assumed, so the soil along the length of the heat exchanger is unsaturated with initial degrees of saturation ranging from 0.50 to 0.21 depending on the height from the water table. Two locations of interest that will be investigated further are noted in Figure 5(b) having different initial degrees of saturation.

During heat injection, a constant heat flux of 50 W/m² was applied to the outer boundary of the geothermal heat exchanger. This heat flux was converted to a volumetric heat source to obtain the value of Q in Equation (6). Although the magnitude of heat flux used in this study is representative of average value in geothermal energy storage systems (Acuña et al. 2012; Welsch et al. 2015; McCartney et al. 2017), a constant heat flux is not expected in a system where solar thermal panels are the heat source. In these cases, the input temperature from the solar thermal panels will remain relatively constant, which means that the heat flux will decrease with time as the subsurface warms (Welsch et al. 2015). Although use of a constant heat flux will lead to greater increases in ground temperature than those expected when using solar thermal panels as the heat source, it provides a simple boundary condition for evaluating the roles of different heat transfer mechanisms in unsaturated soils.

314 Heat Transfer and Water Flow Evaluation

The temperature time series at a distance of 0.05 m from the geothermal heat exchanger and a 315 depth of 8.5 m from the surface (i.e., $S_{r0} = 0.25$) is shown in Figure 6(a). After the 90-day heat 316 injection period, the ground temperature reached a maximum value of 45.6 °C at this location. For 317 318 comparison, the temperature time series from the model where no vapor diffusion or phase change is considered (the no vapor case) is also shown in Figure 6(a). In addition to showing a slower rate 319 of increase in temperature, a lower maximum temperature of 36.3 °C was observed for the no 320 vapor case. After the heating injection period, the heat flux was set to 0 W/m^2 and the soil was 321 allowed to cool ambiently. After 180 days from the start of the simulation (90 days after the end 322 of heat injection), the temperature decreased to 22.8 °C for the model with vapor diffusion and 323 324 phase change and to 21.8 °C for the no vapor case. Although the gradients for heat loss are higher in the model with vapor diffusion and phase change, more heat is retained at this location after the 325 326 180-day cooling period due to the decrease in degree of saturation of the soil during heat injection due to vapor diffusion and latent heat transfer observed in Figure 6(b). Specifically, at the end of 327 the heat injection period, a decrease in degree of saturation of 0.14 at this depth was observed for 328 329 the model with vapor diffusion and phase change, while a negligible decrease in degree of saturation of 0.01 was observed for the no vapor case. The greater decrease in the degree of 330 saturation for the model with vapor diffusion and phase change led to a decrease in thermal 331 332 conductivity according to the TCF (from 0.84 to 0.49 W/mK), which will be assessed in more

detail later. Another interesting observation from Figure 6(b) is that at the end of the 90-day
cooling period, only 16.5% of the decrease in degree of saturation observed during heat injection
was recovered, indicating that the drying near the heat exchanger was permanent from a practical
point of view. This may have an impact on subsequent heat injection and cooling cycles, and may
be one of the reasons that an increase in the ground temperature is observed after several cycles of
heat injection and extraction in practice (Sibbitt et al. 2012).

Radial distributions in temperature at a depth of 8.5 m from the surface at the ends of the heat 339 injection and cooling periods are shown in Figure 7(a). Heat injection led to a notable change in 340 341 temperature up to a distance of about 3 m from the heat exchanger. The temperature at the location of the heat exchanger was nearly 10 °C greater when vapor diffusion and latent heat transfer was 342 considered than the case when it was not, and the temperature at the end of the ambient cooling 343 period was greater throughout the zone of influence. A decrease in degree of saturation was only 344 observed within approximately 1 m from heat exchanger for the model with vapor diffusion and 345 phase change, as shown in Figure 7(b). A slight decrease in degree of saturation was observed near 346 347 the heat exchanger for the no vapor case due to thermally-induced liquid flow. The zone of influence for temperature changes is greater than the zone of influence for degree of saturation 348 349 changes for the conditions evaluated. For Bonny silt, this indicates that an overlap in the effects of different heat exchangers may be observed for the typical geothermal heat exchanger spacing of 350 1.5 m in geothermal energy storage systems. 351

Profiles of temperature with depth at horizontal distances of 0.05 m and 0.20 m from the heat exchanger at the end the 90-day heat injection period are shown in Figure 8(a). The temperature profiles varied nonlinearly with depth and had a maximum value at a depth of 4.5 m from the surface. For comparison, the temperature profiles for the no vapor case show more uniform

356 distributions in temperature with depth at the end of the heat injection period. The difference in temperature observed with depth in both models is due to the thermo-hydraulic properties with 357 depth associated with the variations in initial degree of saturation with depth shown in Figure 5. A 358 significant decrease in the in degree of saturation with depth is observed at both horizontal 359 distances for the model with vapor diffusion and phase change in Figure 8(b), while only a slight 360 361 decrease was observed for the no vapor case. Profiles of temperature after the ambient cooling are shown in Figure 8(c), with the profiles at horizontal distances of 0.05 m and 0.20 m overlapping. 362 Although most of the heat injected has dissipated away from the heat exchanger, a greater amount 363 364 of heat was retained in the soil close to the heat exchanger for the model with vapor diffusion and phase change. The profiles of degree of saturation at the end of the cooling period shown in Figure 365 8(d) only show slight increases from the profiles observed in Figure 8(b). 366

The impact of the initial degree of saturation on heat transfer and water flow can be investigated 367 by evaluating the transient response at different depths in the soil profile, which have different 368 initial degrees of saturation. Time series of temperature at depths of 8.5 m and 24.5 m from the 369 370 surface at a horizontal distance of 0.05 m from the heat exchanger corresponding to initial degrees of saturation of 0.25 and 0.50 are shown in Figure 9(a). Increases in temperature of 45.6 and 371 372 42.3 °C at the end of the heat injection period were observed when the initial degree of saturation was doubled from 0.25 to 0.50. However, decreases in degree of saturation of 0.14 and 0.35 were 373 observed at the end of the heat injection period for the same depths, as shown in Figure 9(b). The 374 375 greater decrease in degree of saturation for the initially wetter soil is likely due to the availability of water to evaporate from the region near the heat exchanger. The horizontal zone of influence of 376 377 the change in temperature is similar for the two depths as shown in Figure 9(c), but the horizontal 378 zone of influence of the change in degree of saturation was greater at the depth of 8.5 m as shown

in Figure 9(d). This is consistent with observations that dryer initial conditions lead to greater zones of influence of vapor diffusion (e.g., Smits et al. 2011). The soil with $S_{r0}=0.50$ also shows a slight wetting front due to the movement of water away from the heat exchanger.

382 Assessment of Heat Transfer Mechanisms and Effects of Coupled Flow

Profiles of thermal conductivity and volumetric heat capacity that correspond to the profiles of 383 384 degree of saturation in Figure 8(b) are shown in Figures 10(a) and 10(b), respectively. Despite the nonlinear decrease in degree of saturation along the length of the heat exchanger, a comparatively 385 uniform decrease of approximately 0.3 W/mK is observed at 0.05 m from the heat exchanger. A 386 387 more nonlinear decrease in thermal conductivity is observed further away from the heat exchanger at 0.20 m. The shapes of the profiles of the volumetric heat capacity are the same as those for the 388 thermal conductivity due to the same parameters used in Equations (8) and (9), but because of the 389 range of the two relationships for Bonny silt the volumetric heat capacity decreased by as much as 390 25% while the thermal conductivity decreased by as much as 70%. This is a positive finding for 391 geothermal energy storage in similar soil deposits, as it means that lower heat losses can be 392 expected without a significant reduction in the quantity of heat stored. 393

The vapor diffusion and latent heat transfer that results in the drying around the heat exchanger also leads to a suction gradient that may result in liquid water flow back toward the heat exchanger. Horizontal profiles of suction profiles at a depth of 8.5 m ($S_{r0}=0.25$) at different times are shown in figure 11(a). Large increases in suction are observed within 0.6 m from the heat exchanger, with decreases in suction beyond that point. Despite the large gradient associated with the suction distribution at the end of the heating period, the suction did not return to its original distribution during the ambient cooling period. This may have been due to the order of magnitude decrease in the hydraulic conductivity (adjusted for temperature effects) shown in Figure 11(b), indicating that
more time may be needed for liquid flow to occur than permitted in the 90-day cooling period.

Vapor concentrations (kg/m^3) near the heat exchanger normalized by the equilibrium vapor 403 concentration (kg/m³) are shown in Figure 12(a) for Bonny silt with initial degrees of saturation 404 of 0.25 and 0.50 (depths of 8.5 and 24.5 m). When the normalized vapor concentration is greater 405 406 or equal to 0.75, the phase change in the soil can be assumed to be near equilibrium (Lozano et al. 2008). For an initial degree of saturation of 0.25, the normalized vapor concentration soon after 407 the start of heating was smaller than this limit. The normalized vapor concentration decreased to 408 409 0.63 at the end of the heating period, indicating that use of the nonequilibrium model was justified. For an initial degree of saturation of 0.5, the normalized vapor concentration was 0.82 soon after 410 the start of heating and remained above 0.75 indicating that an equilibrium phase change 411 assumption may be valid for initially wetter soils. The time series in Figure 12(a) indicate that the 412 phase change process did not reach steady state conditions by the end of the heat injection period. 413 Horizontal profiles of the normalized vapor concentrations at the end of heating shown in Figure 414 12(b) indicate that lower vapor concentrations were present near the heat source and had a similar 415 zone of influence to the degree of saturation in Figure 7(b). Despite the higher magnitudes of 416 417 normalized vapor concentration, greater changes in normalized vapor concentration with horizontal distance are observed for the initially wetter soil ($S_{r0}=0.5$), which may be the reason for 418 the greater change in degree of saturation at this location. Vertical profiles of normalized vapor 419 420 concentration at the end in Figure 12(c) are similar to those for the degree of saturation in Figure 8(b). A higher vapor concentration was observed close to the surface because of the lower initial 421 422 degrees of saturation and higher temperatures and because of upward movement of water vapor 423 due to buoyancy effects.

424 Horizontal profiles of the latent heat transfer rate, calculated as the product of $L_w R_{gw}$ in Equation (6), at the end of heat injection for soil at depths corresponding to initial degrees of 425 saturation of 0.25 and 0.50 are shown in Figure 13(a). While a positive latent heat transfer rate was 426 427 higher near the heat source indicating evaporation, a very slight value less than zero was observed at a distance about 1 m away from the heat exchanger. This indicates that condensation is occurring 428 429 in the soil at lower temperatures further from the heat source. A comparison between the total thermal energy injected into the geothermal heat exchanger and the total latent heat (i.e., the total 430 energy associated with phase change calculated by integrating the product of $L_w R_{gw}$ over the 431 432 volume of the domain and over time) is shown in Figure 13(b). The total energy associated with phase change at the end of 90 days was 44 MJ, which is approximately 24% of the total heat 433 injected into the system of 180 MJ. This indicates that an appreciable amount of the thermal energy 434 injected into the system leads to phase change and further justifies the need to accurately account 435 for nonequilibrium phase change effects in unsaturated soils. The total latent heat appears to have 436 stabilized over the heat injection period, which may be because of the decrease in degree of 437 438 saturation and reduction in availability of water to change phase near the heat exchanger.

439 CONCLUSIONS

A model that includes a recently-developed set of thermo-hydraulic constitutive relationships was used to understand the roles of vapor diffusion and phase change on the coupled heat transfer and water flow in a fine-grained, non-deformable unsaturated silt layer initially under hydrostatic conditions surrounding a vertical geothermal heat exchanger during heat injection and ambient cooling. In general, the modeling results confirm the importance of considering vapor diffusion and water phase change in simulations of geothermal heat exchangers in unsaturated soils, as well as the relevance of considering nonequilibrium phase change in initially drier soil layers. Although quantitative conclusions from the simulations are specific to the given soil and geometry
investigated, several conclusions can be drawn regarding the use of geothermal heat exchangers
in geothermal energy storage systems in unsaturated soils, including:

A greater rate of increase in temperature and greater magnitude of temperature change were
 observed in the soil near the heat exchanger during heat injection in a model that includes
 enhanced vapor diffusion and phase change. This conclusion indicates that conduction-only
 design models may underestimate the heat injection response of geothermal energy storage
 systems in the vadose zone.

455 The heat retained in the soil near the heat exchanger during an ambient cooling period was • 456 greater when considering vapor diffusion and phase change, despite the greater thermal 457 gradient compared to a model with no vapor. This was found to be due to the decrease in 458 thermal conductivity associated with drying during heat injection. The drying during heat injection can be considered permanent for practical purposes within the time frame of ambient 459 460 cooling considered. This may partially be due to the decrease in hydraulic conductivity due to thermally-induced drying, leading to a negligible amount of liquid water flow back toward the 461 heat exchanger during ambient cooling. 462

Although reductions in both thermal conductivity and volumetric heat capacity are observed during thermally induced drying of the soil near the geothermal heat exchanger, the percentage reductions in thermal conductivity were greater. This indicates that the greater amount of heat retention can be expected in unsaturated soils during ambient cooling, but the maximum possible heat stored will not decrease by as large of an amount.

The zone of influence of changes in temperature was observed to be greater than the zone of
 influence of changes in degree of saturation for the silt under investigation, but both zones of

- 470 influence are appreciable enough that overlap is expected in geothermal energy storage471 systems with closely-spaced geothermal heat exchangers (i.e., 1.5 to 2.0 m).
- The normalized vapor concentrations in the initially drier soil near the ground surface were
 below the limit at which nonequilibrium phase change is expected to occur, justifying the use
 of this more advanced modeling approach. The vapor concentration gradient was greater in the
 initially wetter soil deeper in the profile
- The initial degree of saturation was observed to influence both heat transfer and water flow in
 the model with vapor diffusion and phase change, with the greatest change in the degree of
 saturation occurring for soil with initially higher degrees of saturation.

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Equation	Number	Referen
Nonisothermal liquid flow governing equation:	(1)	(Bear 1972;
$nS_{rw}\frac{\partial\rho_{w}}{\partial t} + n\rho_{w}\frac{dS_{rw}}{dP_{c}}\frac{\partial P_{c}}{\partial t} + \nabla \cdot \left[\rho_{w}\left(-\frac{k_{rw}}{\mu_{w}}\right)\nabla\left(P_{w} + \rho_{w}gz\right)\right] = -R_{gw}$		Moradi al. 2016
n=porosity (m^3/m^3), S _{rw} =degree of water saturation (m^3/m^3),		
ρ_w =temperature-dependent density of water (kg/m ³) (Hillel 1980),		
t=time(s), $P_c=P_w-P_g=$ capillary pressure (Pa), $P_w=$ pore water pressure (Pa),		
P_g =pore gas pressure (Pa), k_{rw} =relative permeability function for water		
(m/s); κ =intrinsic permeability (m ²); μ_w =temperature-dependent water		
dynamic viscosity (kg/(ms)) (Lide 2001), g=acceleration due to gravity $(m/c^2) R$. Phase sharpe rate $(leg/m^3 c)$		
(m/s ²) R _{gw} =Phase change rate (kg/m ³ s) Nonisothermal gas flow governing equation:	(2)	(Bear
$nS_{rg} \frac{\partial \rho_g}{\partial t} + n\rho_g \frac{dS_{rg}}{dP_c} \frac{\partial P_c}{\partial t} + \nabla \cdot \left[\rho_g \left(-\frac{k_r g}{\mu_g} \right) \nabla \left(P_g + \rho_g g z \right) \right] = R_{gw}$	(2)	1972; Moradi al. 2016
S_{rg} =degree of gas saturation (m ³ /m ³), ρ_g =temperature-dependent density of gas (kg/m ³) (Smits et al. 2011), k _{rg} =relative permeability function for gas (m/s); μ_g =temperature-dependent gas dynamic viscosity (kg/(ms))		
Water vapor mass balance equation:	(3)	(Smits e
		al. 2011
$n\frac{\partial\left(\rho_{g}S_{rg}w_{v}\right)}{\partial t}+\nabla\cdot\left(\rho_{g}u_{g}w_{v}-D_{e}\rho_{g}\nabla w_{v}\right)=R_{gw}$		
$D_e=D_v\tau$ =effective diffusion coefficient (m ² /s), D_v =diffusion coefficient of water vapor in air (m ² /s) (Campbell 1985), w_v =mass fraction of water vapor		
in the gas phase (kg/kg), $\tau = n^{1/3} S_{rg}^{7/3} \eta$ =tortuosity (Millington and Quirk 1961)		
Enhancement factor for vapor diffusion, η :	(4)	(Cass et
$\eta = a + 3S_{rw} - (a-1)\exp\left\{-\left[\left(1 + \frac{2.6}{\sqrt{f_c}}\right)S_{rw}\right]^3\right\}$		al. 1984
a=empirical fitting parameter, f _c = clay content		
Nonequilibrium gas phase change rate, R_{gw} (kg/m ³ s):	(5)	(Bixler 1985;
$R_{gw} = \left(\frac{bS_{rw}RT}{M_{vv}}\right) \left(\rho_{veq} - \rho_{v}\right)$		Zhang a
$\mathbf{n}_{gw} = \left(\frac{1}{M_{w}} \right) \left(\frac{\rho_{veq} - \rho_{v}}{\rho_{veq}} \right)$		Datta
w b=empirical fitting parameter (s/m ²), R=universal gas constant (J/molK),		2004;
$\rho_{\text{veq}}=c_{\text{v,sat}}R_{\text{h,eq}}=\text{equilibrium vapor density (kg/m3), T=Temperature (K),}$		Moradi al. 2010
$\rho_v = \rho_g w_v = vapor density (kg/m3), M_w = molecular weight of water (kg/mol)$		
Heat transfer energy balance:	(6)	(Whital
$\left(\rho C_{p}\right)\frac{\partial T}{\partial t}+\nabla \cdot\left(\left(\rho_{w}C_{pw}\right)u_{w}T+\left(\rho_{g}C_{pg}\right)u_{g}T-\left(\lambda\nabla T\right)\right)=-L_{w}R_{gw}+Q$		1977; Moradi e
ρ =total density of soil (kg/m ³), C _p =specific heat of soil (J/kgK),		al. 2010
C _{pw} =specific heat capacity of water (J/kgK), C _{pg} =specific heat capacity of		

TABLE 1. Equations used in the numerical analyses

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values of b

FIG. 1. Hydraulic and thermal constitutive relationships and relevant parameters for Bonny silt:
(a) SWRC and HCF; (b) TCF and VHCF

- FIG. 2. Cross-sectional elevation view of the experimental setup for model calibration with
 dielectric sensor locations
- FIG. 3. Predicted and measured time series from the tank-scale heating test: (a) Soil temperatures
 for different values of *a*; (b) Changes in degree of saturation for different values of *a*; (c)
 Soil temperatures for different values of *b*; (d) Changes in degree of saturation for different
- FIG. 4. Simulated and observed temperatures and changes in degree of saturation: (a) Horizontal
 temperature profile; (b) Horizontal change in degree of saturation profile; (c) Vertical
 temperature profile; (d) Vertical change in degree of saturation profile
- FIG. 5. Initial and boundary conditions on the quarter domain model for a field-scale geothermal
 heat exchanger: (a) Thermal; (b) Hydraulic
- FIG. 6. Comparisons of the effects of heat transfer mechanisms and heat flux for a 90-day heat
 injection period followed by cooling: (a) Temperature time series; (b) Degree of saturation
 time series
- FIG. 7. Effects of including vapor flow in the coupled analysis; (a) Radial profiles of temperature
 at a depth of 9.5 m (an initial degree of saturation of 0.25) at the end of heating and cooling;
- (b) Radial profiles of degree of saturation profiles before and after heating
- FIG. 8. Vertical profiles for comparison of different heat transfer mechanisms: (a) Temperature at
 the end of a 90-day heat injection period; (b) Degree of saturation at the end of a 90-day

- heat injection period; (c) Temperature at the end of a 90-day cooling period; (d) Degree of
 saturation at the end of a 90-day cooling period
- FIG. 9. Effect of initial degree of saturation: (a) Time series of temperature; (b) Time series of
 degree of saturation; (c) Radial temperature profiles; (d) Radial degree of saturation
 profiles
- FIG. 10. Thermal property evaluation at the end of heat injection: (a) Thermal conductivity
 profiles; (b) Volumetric heat capacity profiles
- FIG. 11. Liquid water flow evaluation: (a) Horizontal profiles of suction at different times;
 (b) Horizontal profiles of hydraulic conductivity at different times
- FIG. 12. Vapor concentrations normalized by the equilibrium vapor concentration: (a) Time
 series; (b) Horizontal profiles at the end of heating; (c) Vertical profiles at the end of
 heating
- FIG. 13. (a) Distributions of latent heat transfer rate at depths corresponding to initial degrees of
 saturation of 0.25 and 0.50; (b) Comparison of the total heat injected with the energy due
 to phase change





































