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**Specialty section:**

This article was submitted to  
*Environmental Toxicology*, a section of  
the journal *Frontiers in Environmental  
Science*

**Received:** 30 December 2014

**Accepted:** 28 February 2015

**Published:** 23 March 2015

**Citation:**

Srivastava AK, Srivastava S, Lokhande  
VH, D'Souza SF and Suprasanna P  
(2015) Salt stress reveals differential  
antioxidant and energetics responses  
in glycophyte (*Brassica juncea* L.) and  
halophyte (*Sesuvium portulacastrum*  
L.). *Front. Environ. Sci.* 3:19.  
doi: 10.3389/fenvs.2015.00019

# Salt stress reveals differential antioxidant and energetics responses in glycophyte (*Brassica juncea* L.) and halophyte (*Sesuvium portulacastrum* L.)

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Salt stress, considered as one of the major environmental factors, decreases crop productivity world-wide and hence, investigations are being made to understand the cellular basis of salt tolerance in plants. In our earlier studies, maintenance of redox homeostasis and energetics were found as key determinants of salt tolerance in a halophyte *Sesuvium portulacastrum* (high salt accumulator). The redox homeostasis is defined as integrated ratio of different redox couples present inside the cell. In recent years, it has also been proposed as general stress response regulator in plants, bacteria as well as animals. In view of this, present study was performed to compare responses of redox state and energetics of *S. portulacastrum* with a glycophyte *Brassica juncea* (low salt accumulator). The data revealed activation of antioxidant defense in *S. portulacastrum* which either avoided or delayed the accumulation of different reactive oxygen species (ROS). In contrast, due to the lack of co-ordination, although the non-enzymatic antioxidants were increased, significant oxidative damage was seen in *B. juncea*. Further, the decreased NADPH oxidase activity suggested that basal redox signaling was also affected in *B. juncea*. In order to correlate these changes with chloroplastic and mitochondrial electron transport chain, NADP/NADPH and NAD/NADH ratios were measured. The NADP/NADPH ratio suggested that the process of photosynthesis was minimally affected in *S. portulacastrum* which might have contributed to its lower level of ROS under salt stress. The comparatively lower NAD/NADH and ATP/ADP ratios in *S. portulacastrum* as compared to *B. juncea* indicated the active and better utilization of energy generated to support different processes associated with salt tolerance. Thus, the findings suggest that co-ordinated regulation of antioxidant defense to avoid oxidative damage and proper utilization of energy are the key determinants of salt-tolerance in plants.

**Keywords:** antioxidants, ascorbate, energetics, glutathione, redox state, reactive oxygen species

## Introduction

A major challenge toward world agriculture involves production of 70% more food crop for an additional 2.3 billion people by 2050 (FAO, 2009). Salinity is a major stress limiting the increase in the demand for food crops. More than 20% of cultivated land worldwide (~ about 45 hectares) is affected by salt stress and the amount is increasing day by day. Plants on the basis of adaptive evolution can be classified roughly into two major types: the halophytes (that can withstand salinity) and glycophytes (that cannot withstand salinity and eventually die). Majority of crop species belong to this second category. Thus salinity is considered as one of the most important stresses that hamper crop productivity worldwide (Gupta and Huang, 2014). During salinity stress, availability of atmospheric CO<sub>2</sub> is reduced because of an increased stomatal closure in order to avoid water loss via transpiration and hence, consumption of NADPH by the Calvin cycle is decreased. Due to the over-reduction of electron transport chain, electrons get misled from oxygen to form reactive oxygen species (ROs) like superoxide radicals (O<sub>2</sub><sup>-</sup>) and hydrogen peroxide (H<sub>2</sub>O<sub>2</sub>). Additionally, to meet the increased energy demand of the cell which can support different defense processes such as enhanced antioxidant capacity, osmolytes biosynthesis, ion transport and vacuolar sequestration, the activity of mitochondrial electron transport is also increased which further contributes to the generation of ROs (Munns and Tester, 2008). Plants have complex antioxidant defense mechanisms including superoxide dismutase (SOD) and the ascorbate (ASC)-glutathione (GSH) cycle (Mittler, 2002). SOD constitutes the first line of defense converting O<sub>2</sub><sup>•-</sup> to hydrogen peroxide (H<sub>2</sub>O<sub>2</sub>), which is further reduced to water and oxygen by ascorbate peroxidase (APX) and catalase (CAT). APX uses two molecules of ASC to reduce H<sub>2</sub>O<sub>2</sub> to water, with the concomitant generation of two molecules of monodehydroascorbate (MDHA), whereas CAT does not need any reductant for action against H<sub>2</sub>O<sub>2</sub>. Regeneration of ASC from MDHA occurs in sequential steps and utilizes GSH. This results in generation of oxidized glutathione (GSSG), which is in turn, re-reduced to GSH by NADPH, a reaction catalyzed by glutathione reductase (GR). Ascorbate and GSH both can accumulate in millimolar concentrations in cells and function as molecular antioxidants, in addition to serving various other roles, reacting directly with various ROS (Noctor and Foyer, 1998).

Among different experimental approaches adopted to understand plant responses to salt stress, one of the strategies is to compare halophytes and glycophytes. In this context, *Thellungiella salsuginea* is widely used as halophytic model and its responses are compared with its close relative *Arabidopsis thaliana*. The lipidomic analysis has revealed that remodeling of plastidic lipids is important for maintaining the integrity and fluidity of plastidic membranes which contributes to PEG-induced osmotic tolerance of *T. salsuginea* (Yu and Li, 2014). A proteomics study has been done to identify novel proteins associated with high salt tolerance of *T. salsuginea* (Vera-Estrella et al., 2014). The physiological responses of *T. salsuginea* and *A. thaliana* have also been studied to understand the mechanism of boron exclusion and tolerance in plants (Lamdan et al., 2012). The species of *Eutrema* such as

*Eutrema parvulum* and *E. salsugineum* have been contrasted with *A. thaliana* to understand the regulatory role of genes associated with aldehyde dehydrogenase gene superfamily (Hou and Bartels, 2015). Using *Mesembryanthemum crystallinum* (halophyte) and *Brassica juncea*, the mechanism of nickel accumulation and tolerance has been studied (Amari et al., 2014). The halophyte *Cakile maritima* has been compared with *Brassica juncea* to understand cadmium accumulation and tolerance (Taamalli et al., 2014). In the same line, *Sesuvium portulacastrum* is a mangrove associate which is known for its high capacity to accumulate salt. Its physiological, redox, and energetics behavior toward salt stress has been reported (Lokhande et al., 2011). In the present study, these responses were compared with that of *B. juncea* which is a glycophyte and low salt accumulator. The findings confirmed the significance of co-ordinated antioxidant responses and effective utilization of energy as important determinants of salt tolerance in plants.

## Materials and Methods

### Plant Material and Treatment Conditions

The seeds of Indian mustard (*B. juncea* cv. TM- 2) were surface sterilized with 30% ethanol for 3 min and then washed thoroughly to remove any traces of ethanol. They were then allowed to germinate in plastic pots containing sand: soil (1:1) mixture at 25°C under light providing 115 μmol photons m<sup>-2</sup>s<sup>-1</sup> illumination, with 12 h photoperiod. After 6-days of germination, water was changed with ½strength Murashige and Skoog's (MS) medium (Murashige and Skoog, 1962). After 15-days of germination, plants having secondary leaves were used for salt treatment. The *Sesuvium*—MH clone (Lokhande et al., 2010) was used for the present study. Four nodal sectors (~4.0 cm) containing single pre-existing axillary bud with two opposite leaves were planted in the plastic pots containing sand: soil (1:1) mixture. The 45-day-old plants were used for the present study. For salt treatment, *B. juncea* was subjected to 250 mM NaCl, while *S. portulacastrum* was given two different NaCl concentrations (250 or 1000 mM). All the salt solutions were prepared in ½strength MS medium. An independent set with no NaCl was maintained as control. At 2, 4, and 8-days after treatment, secondary leaves were harvested and stored at -80°C until further analyses. The NaCl concentrations were selected on the basis of preliminary experiments to assess plant tolerance in terms of growth to various NaCl concentrations (100–300 mM for *B. juncea* and 200–1000 mM for *S. portulacastrum*).

### Assay of Reactive Oxygen Species (Superoxide radicals, Hydrogen Peroxide) and the Level of Malondialdehyde

For the estimation of hydrogen peroxide (H<sub>2</sub>O<sub>2</sub>) levels, plant samples were homogenized in 0.5% (w/v) trichloroacetic acid (TCA) in an ice bath and centrifuged at 14,000 × *g* for 15 min at 4°C. For H<sub>2</sub>O<sub>2</sub> determination, 0.5 ml of supernatant was mixed with 0.5 ml 100 mM potassium phosphate buffer (pH 7.0) and 1 ml of freshly prepared 1 M potassium iodide. Reaction was allowed to develop for 1 h in dark and absorbance was measured at 390 nm (Alexieva et al., 2001). The amount of H<sub>2</sub>O<sub>2</sub>

was calculated from a standard curve prepared using the known concentrations of H<sub>2</sub>O<sub>2</sub>. The rate of superoxide radicals (O<sub>2</sub><sup>•-</sup>) production was measured following the method of Chaitanya and Naithani (1994). About 500 mg of fresh plant samples were homogenized under N<sub>2</sub> atmosphere in cold (0-4°C) in 100 mM sodium phosphate buffer (pH 7.2) containing 10 mM sodium azide to inhibit SOD activity. After centrifugation at 22,000 × g at 4°C for 20 min, the level of O<sub>2</sub><sup>•-</sup> was measured in the supernatant by its capacity to reduce NBT (extinction coefficient; ε = 12.8 mM<sup>-1</sup> cm<sup>-1</sup>). The reaction mixture (1 ml) contained 100 mM sodium phosphate buffer (pH 7.8), 0.05% (w/v) NBT (Nitro blue tetrazolium chloride), 10 mM sodium azide and 0.2 ml of extract. Absorbance was measured at 580 nm at 0 and 60 min. The level of O<sub>2</sub><sup>•-</sup> is expressed as an increase in absorbance per min per gram fresh weight. Lipid peroxidation was determined by estimation of the malondialdehyde (MDA) content following Heath and Packer (1968) as described previously (Srivastava et al., 2006).

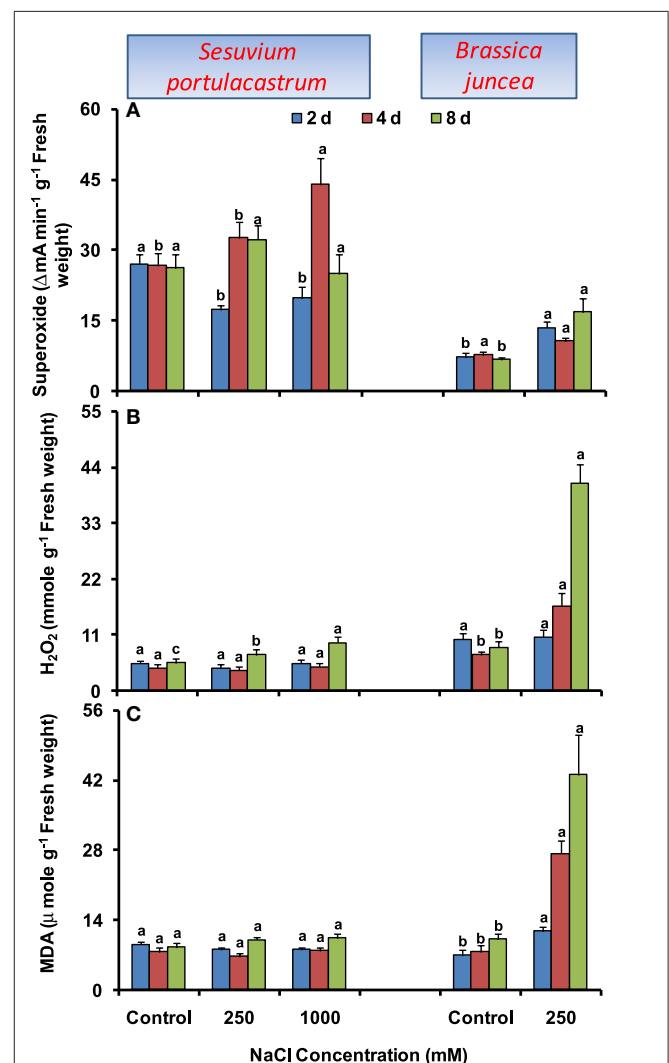
### Assay of NADPH Oxidase Activity

Plants were homogenized in 20 mM HEPES [4-(2-hydroxyethyl)-1-piperazineethanesulfonic acid; pH 7.0] containing 1 mM EDTA, 1 mM EGTA, 1 mM phenylmethyl sulfonyl fluoride (PMSF) and 2.5% polyvinylpyrrolidone (PVP) under chilled conditions. Homogenate was squeezed through four layers of cold cheese cloth and centrifuged at 12,000 × g for 15 min at 4°C. Protein content of supernatant was measured following the protocol of Lowry et al. (1951). NADPH-dependent O<sub>2</sub><sup>•-</sup> generation was measured using NBT as an electron acceptor, whose reduction was monitored at 530 nm. Monoformazan concentrations (and therefore O<sub>2</sub><sup>•-</sup> concentrations) were calculated using ε of 12.8 mM<sup>-1</sup> cm<sup>-1</sup>. The reaction mixture consisted of Tris buffer (50 mM Tris-HCl, pH 7.4), 5 mM NBT, 1 mM MgCl<sub>2</sub>, 1 mM CaCl<sub>2</sub>, 5 mM NADPH and a suitable aliquot of enzyme extract. The selective reduction of NBT by O<sub>2</sub><sup>•-</sup> was calculated from the difference in the NBT reduction rate in the presence and absence of SOD (50–100 units ml<sup>-1</sup>; Sigma, USA). No NBT reduction with NADPH was observed in the absence of protein fractions.

### Assays of Enzymes of Antioxidant System

For the assay of antioxidant enzymes, plant samples (500 mg) were homogenized in 100 mM chilled potassium phosphate buffer (pH 7.0) containing 1 mM EDTA, 1 mM PMSF and 1% PVP (soluble, MW 3,60,000) at 4°C. Homogenate was squeezed through four layers of cold cheese cloth and the extract thus obtained was centrifuged at 15,000 g for 15 min at 4°C. For all enzyme assays, respective sample and reagent blanks were run in duplicate. The activity of SOD (EC 1.15.1.1) was assayed following the method of Beauchamp and Fridovich (1971). The reaction mixture for SOD activity assay contained 40 mM phosphate buffer (pH 7.8), 13 mM methionine, 75 μM NBT, 2 μM riboflavin, 0.1 mM EDTA and a suitable aliquot of enzyme extract. After the reaction under light for 15 min, the absorbance was taken at 560 nm. One unit of activity is the amount of protein required to inhibit 50% initial reduction of NBT under light.

For measurement of the CAT (EC 1.11.1.6) activity, extraction was done in the buffer containing 50 mM Tris-HCl (pH 7.0), 0.1 mM EDTA, 1 mM PMSF and 0.3 g g<sup>-1</sup> fw PVP. Activity was measured by the method of Aebi (1974). The reaction mixture comprised of 50 mM sodium phosphate buffer (pH 7.0), 20 mM H<sub>2</sub>O<sub>2</sub> and suitable aliquot of enzyme. Decrease in the absorbance was taken at 240 nm (molar extinction coefficient of H<sub>2</sub>O<sub>2</sub> was 0.04 cm<sup>2</sup> μmol<sup>-1</sup>). Enzyme activity was expressed as units mg<sup>-1</sup> protein. The activity of APX (EC 1.11.1.11) was measured by estimating the rate of ascorbate oxidation (extinction coefficient 2.8 mM<sup>-1</sup> cm<sup>-1</sup>). The reaction mixture contained 50 mM phosphate buffer (pH 7.0), 0.1 mM H<sub>2</sub>O<sub>2</sub>, 0.5 mM sodium ascorbate, 0.1 mM EDTA and suitable aliquot of enzyme. The change in absorbance was monitored at 290 nm (Nakano and



**FIGURE 1 |** Time dependent modulation of superoxide radicals (A), hydrogen peroxide (B) and malondialdehyde (C) in NaCl-exposed *Sesuvium portulacastrum* and *Brassica juncea*. All values represent the mean of three replicates. Different letters indicate significantly different values at a particular duration (DMRT, P = 0.01).

Asada, 1981) and enzyme activity was expressed as units  $\text{mg}^{-1}$  protein. For the estimation of the GR (EC 1.6.4.2) activity plant material was extracted in 0.1 M potassium phosphate buffer (pH 7.5) containing 0.5 mM EDTA. Activity was assayed by following the method of Smith et al. (1988). The reaction was started by adding following in order; 1.0 ml of 0.2 M potassium phosphate buffer (pH 7.5) containing 1 mM EDTA, 500  $\mu\text{l}$  3 mM 5,5'-dithiobis (2-nitrobenzoic acid) in 0.01 M phosphate buffer (pH 7.5), 250  $\mu\text{l}$   $\text{H}_2\text{O}$ , 100  $\mu\text{l}$  2 mM NADPH, 50  $\mu\text{l}$  enzyme extract, and 100  $\mu\text{l}$  20 mM GSSG. The increase in absorbance was monitored for 5 min at 412 nm. The rate of enzyme activity was calculated using standard curve prepared by known amounts of GR (Sigma, USA). Activity of enzyme was expressed as units  $\text{mg}^{-1}$  protein.

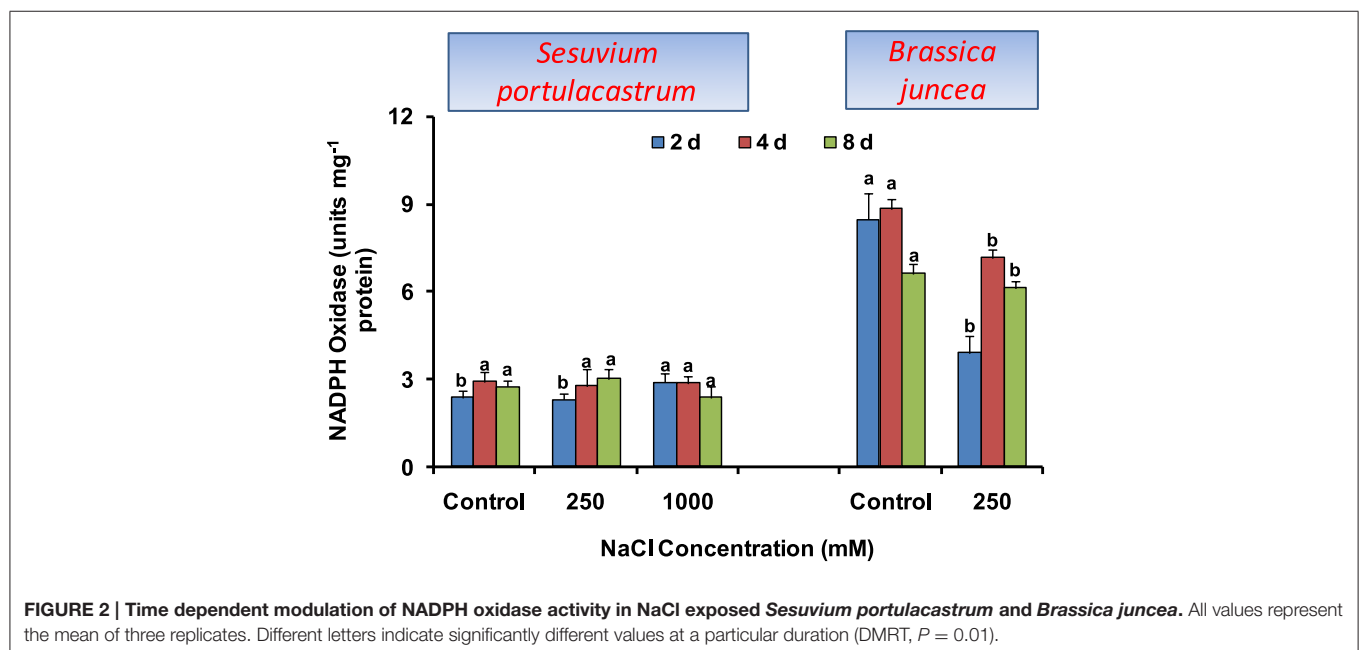
### Estimation of Redox Couples (GSH/GSSG and ASC/DHA Ratio)

The level of reduced (GSH) and oxidized (GSSG) glutathione was determined fluorometrically using o-phthalaldehyde (OPT) as fluorophore (Hissin and Hilf, 1976). The level of total, reduced and oxidized ascorbate (ASC) contents in plants was measured following the protocol of Gillespie and Ainsworth (2007). Plant samples (50 mg) were homogenized in 1 ml 6% trichloroacetic acid (TCA) under chilled conditions and centrifuged at  $13,000 \times g$  for 5 min at  $4^\circ\text{C}$ . To 200  $\mu\text{l}$  of sample 100  $\mu\text{l}$  75 mM phosphate buffer (pH 7.0) was added. In total ASC, 100  $\mu\text{l}$  DTT (dithiothreitol; 10 mM) was added and incubated for 10 min at room temperature to reduce the pool of oxidized ASC. Then, 100  $\mu\text{l}$  NEM (N-ethylmaleimide; 0.5%) was added to remove excess DTT. For reaction, 500  $\mu\text{l}$  10% TCA, 400  $\mu\text{l}$  43% orthophosphoric acid, 400  $\mu\text{l}$  4% 2,2'-bipyridyl, and 200  $\mu\text{l}$  3%  $\text{FeCl}_3$  were added to all tubes. After incubation at  $37^\circ\text{C}$  for 1 h, absorbance was measured at 525 nm. The level of dehydroascorbate

(DHA) was calculated by subtracting ASC values from total ASC.

### Determination of Adenine and Pyridine Nucleotides

The analysis of adenine and pyridine nucleotides was performed by High Performance Liquid Chromatography (HPLC) as described previously by Srivastava et al. (2011). In brief, samples (100 mg) were subjected to either acid extraction using 0.6 M perchloric acid (for the measurement of ATP, ADP, NADP, and NAD) or alkaline extraction using 0.5 M potassium hydroxide (for the measurement of NADPH and NADH). The extract was centrifuged at  $14,000 \times g$  at  $4^\circ\text{C}$  for 10 min followed by neutralization with either 0.5 M KOH or 1 M  $\text{KH}_2\text{PO}_4$ , respectively and re-centrifuged at  $14,000 \times g$  at  $4^\circ\text{C}$  for 10 min to remove the precipitate. Supernatant was filtered using 0.22  $\mu\text{m}$  syringe filters and used for the HPLC injection. The mobile phase consisted of 0.1 M  $\text{KH}_2\text{PO}_4$  solution at pH 6.0 (Buffer A) and a 0.1 M  $\text{KH}_2\text{PO}_4$  solution at pH 6.0, containing 10% (v/v) of  $\text{CH}_3\text{OH}$  (Buffer B). The chromatographic conditions were as follows: 8 min at 100% of buffer A, 7 min at up to 25% of buffer B, 2.5 min at up to 90% of buffer B, 2.5 min at up to 100% of buffer B, held for 7 min at 100% B, 5 min at up to 100% buffer A and held for 8 min at 100% buffer B to restore the initial condition. The flow rate was 1 mL/min and detection was performed at 254 nm (Waters 996, PDA detector). Separation was performed on a 10  $\mu\text{m}$  C18 analytical column (250  $\times$  4.6 mm) equipped with a guard column. The peaks were identified using the standard samples. The analytical recovery was tested by adding a known amount of standard compound prior to extraction and recovery was found to be 94 to 100% for different compounds. The data were analyzed using Empower software.



**FIGURE 2 |** Time dependent modulation of NADPH oxidase activity in NaCl exposed *Sesuvium portulacastrum* and *Brassica juncea*. All values represent the mean of three replicates. Different letters indicate significantly different values at a particular duration (DMRT,  $P = 0.01$ ).

### Statistical Analysis

The experiments were carried out in a randomized block design. One-Way analysis of variance (ANOVA) was done on all the data to confirm the variability of data and validity of results and Duncan's multiple range test (DMRT) was performed to determine the significant difference between treatments using the statistical software SPSS 10.0.

### Results

#### Effect of Salinity Stress on the Level of Reactive Oxygen Species and Lipid Peroxidation

In *S. portulacastrum*, no significant change in MDA level was observed at any salt concentration and duration. The increase in ROS levels was also limited to longer duration. At 8 d, H<sub>2</sub>O<sub>2</sub> level was increased under both 250 and 1000 mM NaCl treatment; while O<sub>2</sub><sup>•-</sup> level was increased only under 1000 mM NaCl treatment as compared to that of control (Figures 1A–C). By contrast, in *B. juncea*, time-dependent increase in MDA level was observed (Figure 1C) and ROS levels also increased significantly. The maximum increase of 2.5- and 4.8-fold in O<sub>2</sub><sup>•-</sup> and H<sub>2</sub>O<sub>2</sub> levels were observed at 8 d after treatment, as compared to that of control (Figures 1A–B).

#### Effect of Salinity Stress on the Activity of NADPH Oxidase

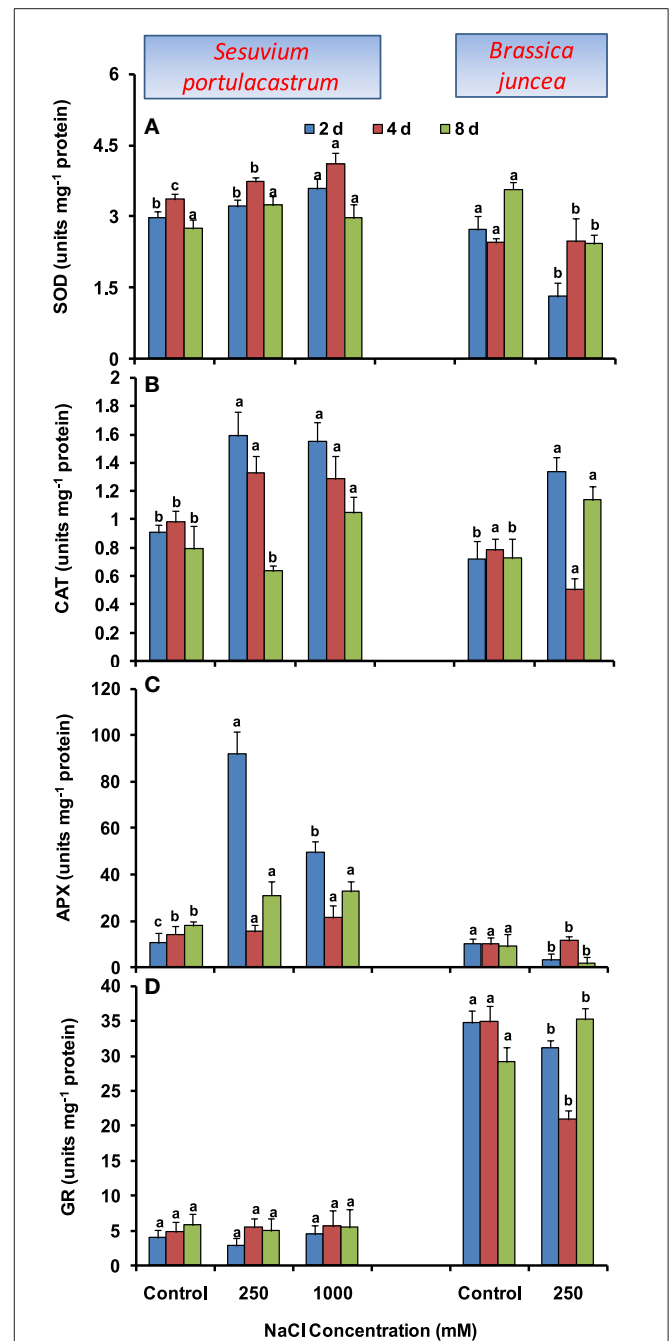
The activity of NADPH oxidase was not affected significantly in *S. portulacastrum* except for the 22% increase at 1000 mM NaCl after 2 d treatment. In *B. juncea*, NADPH oxidase activity declined significantly at all the durations as compared to that of control (Figure 2).

#### Effect of Salinity Stress on the Activities of Antioxidant Enzymes

In *S. portulacastrum*, dose-dependent increase in SOD activity was observed until 4 d, with the maximum of 30% under 1000 mM NaCl as compared to that of control. At 2 and 8 d after treatment, SOD activity was almost unchanged, except at 1000 mM at 2 d where it was increased by 20% as compared to that of control (Figure 3A). Unlike SOD, CAT activity was increased at all the durations, except at 8 d where the increase was observed only at 1000 mM NaCl concentration (Figure 3B). The APX activity increased at both NaCl concentrations; however, maximum of 2.8-fold was observed under 250 mM at 2 d after treatment as compared to that of control. At 4 and 8 d, comparable increase in APX activity was observed under both 250 and 1000 mM NaCl as compared to that of control (Figure 3C). The GR activity was almost unchanged at both NaCl concentrations and all durations (Figure 3D). In *B. juncea*, except for CAT, activities of SOD, APX, and GR decreased at all the durations (Figures 3A–D).

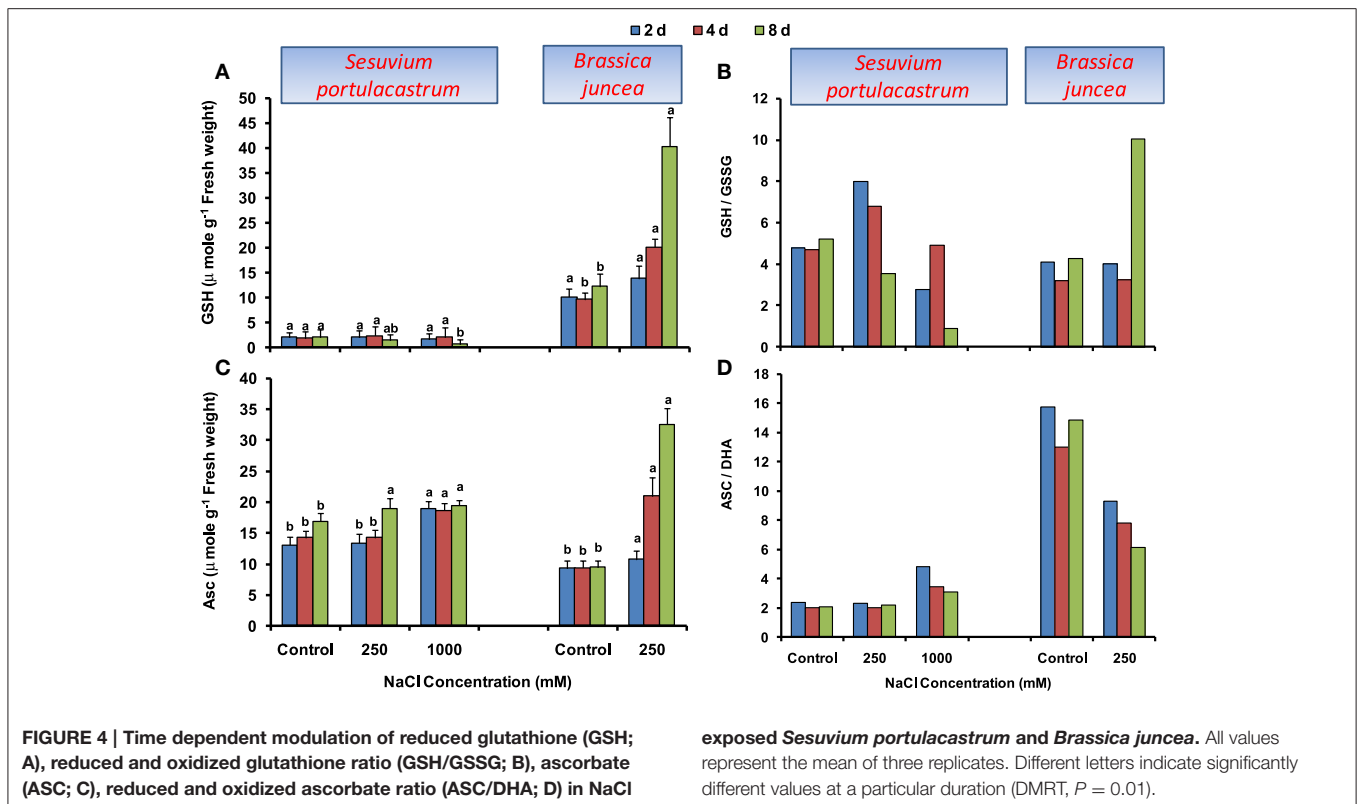
#### Effect of Salinity Stress on GSH and ASA Pools and their Redox Couples (GSH/GSSG and ASC/DHA Ratio)

In *S. portulacastrum*, GSH level was unaltered until 4 d; however; at 8 d it showed significant reduction with the maximum



**FIGURE 3 |** Time dependent modulation of activities of superoxide dismutase (SOD; A), catalase (CAT; B), ascorbate peroxide (APX; C), and glutathione reductase (GR; D) in NaCl exposed *Sesuvium portulacastrum* and *Brassica juncea*. All values represent the mean of three replicates. Different letters indicate significantly different values at a particular duration (DMRT, P = 0.01).

of 40% under 1000 mM NaCl as compared to that of control (Figure 4A). The GSH/GSSG ratio showed an increase in comparison to control until 4 d at 250 mM NaCl but was at par to control at 1000 mM NaCl. At 8 d, GSH/GSSG ratio was decreased



with the maximum of 66% under 1000 mM NaCl as compared to that of control (**Figure 4B**). In *B. juncea*, GSH level increased beyond 2 d and at 8 d, it was 2.5-fold higher than that of control. The GSH/GSSG ratio was at par with control till 4 d and at 8 d, it was 1.8-fold higher than that of control.

Under NaCl stress, a distinct increase in ASC level was observed in both *S. portulacastrum* and *B. juncea*, however, the level of increase was comparatively higher in *B. juncea* (**Figure 4C**). The ASC/DHA ratio was unchanged in *S. portulacastrum* at 250 mM NaCl; however, at 1000 mM, it significantly increased. In contrast, *B. juncea* showed a significant decrease of ASC/DHA ratio at all the durations (**Figure 4D**).

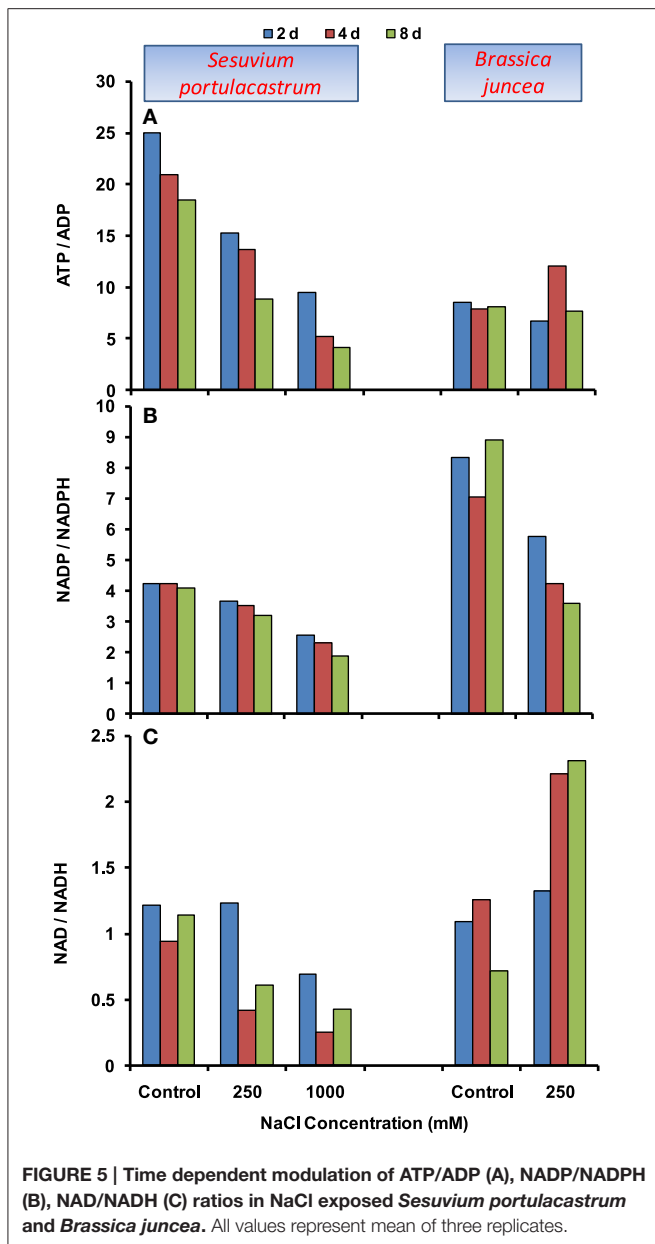
### Effect of Salinity Stress on the Level of Adenylate and Pyridine Nucleotides

In *S. portulacastrum*, ATP level was declined at both 250 and 1000 mM NaCl (Supplementary Figure 1A), hence, ATP/ADP ratio declined significantly (**Figure 5A**). Similar profile was also observed for NADP/NADPH and NAD/NADH ratios. The maximum decrease in these ratios was observed under 1000 mM NaCl at 8 d after treatment (**Figures 5B,C**). In *B. juncea*, ATP/ADP ratio remained at par with control, except at 4 d at which it was increased by 1.5-fold as compared to that of control (**Figure 5A**). The NADP and NADPH levels although increased (Supplementary Figures 2, 3), NADP/NADPH ratio was reduced with the maximum of 60% at 8 d after treatment as compared to that of control (**Figure 5B**). In contrast, the NAD/NADH ratio was increased beyond 2 d of stress treatment (**Figure 5C**).

## Discussion

The salinity stress in plants is known to drastically reduce its growth and productivity (Gupta and Huang, 2014). Thus, in order to maintain the crop productivity in saline affected regions, it is essential to understand the mechanism of salt toxicity in plants. Generally, most crop plants are glycophytes and can withstand salt concentration in the range of 50–250 mM; however, there are specific type of plants referred as halophytes which can tolerate upto 1 M salt concentration. In an earlier study, physiological responses of a halophyte, *S. portulacastrum* have been reported and redox homeostasis and energetics were found to be the key determinants of salt tolerance (Lokhande et al., 2011). To further strengthen this hypothesis, in the present work, similar responses were studied in *B. juncea* (Indian mustard – a glycophyte) and compared with that of *S. portulacastrum*.

In the presence of salt stress, series of events leading to perturbation of cellular metabolism are: less water availability, stomata closure, altered gaseous exchange, inhibition of photosynthesis, effect on electron flow in ETC in chloroplast and mitochondria, increase in the production of ROS and disturbed status of adenine (ATP) and pyridine nucleotides (NADH, NADPH). Thus, the oxidative damage and altered energetics are mainly responsible for affecting the general metabolism and plant growth under salt stress (Srivastava et al., 2011). In *S. portulacastrum*, the increase in ROS levels was seen only at longer duration; however, in *B. juncea*, the increase in ROS and MDA levels were seen as early



as 2 d after treatment (Figures 1A–C). This clearly suggested that *S. portulacastrum* has better ability to avoid or delay the oxidative damage under salt stress conditions. In recent years, significant progress has been made with respect to ROS function in plants and they are now not seen as negative by-product of oxidative metabolism but they are known to serve as important signaling mediators under stress (Gilroy et al., 2014). The ROS based signaling is termed as “redox signaling” and is important for growth and survival of the plants under normal as well as salt stress conditions (Srivastava and Suprasanna, in press). There are many pro-oxidant enzymes in plants which are responsible for the ROS formation. NADPH oxidase is one such pro-oxidant which is an important regulator of calcium signaling and downstream signal transduction associated with salt tolerance (Marino et al.,

2012; Drerup et al., 2013). The NADPH oxidase activity in *S. portulacastrum* was unaffected while in *B. juncea*, it decreased significantly (Figure 2). This indicates that apart from oxidative damage, the basal ROS signaling is also disturbed in glycophytes under salt stress. Thus, differential ability of halophytes and glycophytes to perceive stress seems to be responsible for differential stress response. The NADPH oxidase might be one of the possible candidates responsible for this; however, this needs to be investigated further.

Significant research has been conducted to establish that antioxidant defense is important for maintaining proper plant growth under different abiotic stress conditions, especially halophytes (Boestfleisch et al., 2014; Canalejo et al., 2014; Yildiztugay et al., 2014). In the same line, to see how the stress imposition and oxidative stress in two plant species (glycophytes and halophyte) affected their antioxidant metabolism and *vice versa*, activities of enzymatic antioxidant and levels of non-enzymatic antioxidants were measured. The enzymatic antioxidants showed a significant decline except CAT in *B. juncea*, whereas in *S. portulacastrum*, there was either a increase or no change (Figures 3A–D). Apart from suggesting the better ability of *S. portulacastrum* to activate its antioxidant defense under stress, this also indicated that halophytic enzymes are comparably more robust and stable than glycophytic enzymes. In recent years, various efforts have been made where antioxidant gene from a halophyte has been used to make transgenic glycophytes with higher salt tolerance. For instance, cytosolic copper/zinc SOD from a mangrove plant *Avicennia marina* has been used to increase the abiotic stress tolerance in rice (Prashanth et al., 2008). The peroxisomal APX from halophyte *Salicornia brachiata* was used to confer salt and drought tolerance in tobacco (Singh et al., 2014). Among molecular antioxidants, ASC and GSH are involved in the functioning of ASC-GSH cycle and can also independently function as antioxidants. Hence, maintenance of ASC/DHA and GSH/GSSG ratio is necessary to allow antioxidant functions to operate properly. In *B. juncea*, the GSH/GSSG and ASC/DHA ratio was either maintained or decreased which indicated plant’s inability to stimulate their ASC and GSH regenerating systems as observed in the case of GR. Thus, despite an increase in absolute ASC and GSH, ROS levels continued to increase significantly. On the contrary in *S. portulacastrum*, both ASC/DHA and GSH/GSSG showed increasing trend; except for the decrease in GSH/GSSG ratio at 1000 mM NaCl which signifies the active utilization of GSH to avoid the oxidative damage at such a higher salt concentration.

Apart from the lack of co-ordinated activation of antioxidant defense, the over-reduction of electron transport chain (ETC) is also an important contributor of oxidative damage under stress. The over-reduced ETC in chloroplasts and mitochondria apparently result in significant decline in ratios of NADP/NADPH and NAD/NADH, respectively (Munns and Tester, 2008). In the present study, both the plants demonstrated the decreasing trend in NADP/NADPH ratio; however the extent of decrease was higher in *B. juncea* than *S. portulacastrum*. This suggests two possibilities, first: either the photosynthesis was not affected to a significant level in *S. portulacastrum*; second: there might be the activation of alternative salvage pathway to utilize

reducing equivalents, such as photorespiration. Although, comparison of glycophyte *Arabidopsis* and its halophytic relative *Thellungiella* showed that photosynthesis and gas exchange in halophyte is minimally affected under salt conditions (Stepien and Johnson, 2009), the contribution of salvage pathway cannot be completely ignored, however this needs further investigation. To see the effects on plant energetic, the levels of adenylate (ATP, ADP) and pyridine nucleotides (NAD, NADH) were measured. In *B. juncea*, NaCl stress increases the NAD/NADH ratio which signifies the effect on energy metabolism. However, the ATP/ADP ratio remained unchanged which suggests that ATP formed is not actively utilized. By contrast, in *S. portulacastrum*, significant consumption of ATP and NAD was observed which led to the decline in ATP/ADP and NAD/NADH ratios. Thus, apart from the ATP and NAD generation, their proper channelization or consumption is also important to activate

different mechanisms responsible for imparting salt tolerance in plants.

In summary, two plant systems (*S. portulacastrum* – a halophyte and *B. juncea* – a glycophyte) differing in their salt-tolerance ability have been compared. The findings reveal better antioxidant defense to avoid oxidative damage and proper energy consumption as key determinants of salinity tolerance in these two contrasting plants. The differential response of NADPH oxidase is also seen which needs to be investigated further in the context of stress perception and downstream signaling.

## Supplementary Material

The Supplementary Material for this article can be found online at: <http://www.frontiersin.org/journal/10.3389/fenvs.2015.00019/abstract>

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**Conflict of Interest Statement:** The authors declare that the research was conducted in the absence of any commercial or financial relationships that could be construed as a potential conflict of interest.

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