

6-1-1987

## Search for Isotope Effect in Superconducting Y-Ba-Cu-O

L C. Bourne

*University of California - Berkeley*

M F. Crommie

*University of California - Berkeley*

A Zettl

*University of California - Berkeley*

Hans Conrad zur Loye


*University of South Carolina - Columbia, zurloye@sc.edu*

S W. Keller

*University of California - Berkeley*

*See next page for additional authors*

Follow this and additional works at: [https://scholarcommons.sc.edu/chem\\_facpub](https://scholarcommons.sc.edu/chem_facpub)

 Part of the [Chemistry Commons](#)

---

### Publication Info

*Physical Review Letters*, Volume 58, Issue 22, 1987, pages 2337-2339.

© 1987 The American Physical Society

URL: <http://link.aps.org/doi/10.1103/PhysRevLett.58.2337>

DOI: 10.1103/PhysRevLett.58.2337

PACS: 74.30.-e, 74.70.Ya

This Article is brought to you by the Chemistry and Biochemistry, Department of at Scholar Commons. It has been accepted for inclusion in Faculty Publications by an authorized administrator of Scholar Commons. For more information, please contact [digres@mailbox.sc.edu](mailto:digres@mailbox.sc.edu).

---

**Author(s)**

L C. Bourne, M F. Crommie, A Zettl, Hans Conrad zur Loye, S W. Keller, K L. Leary, Angelica M. Stacy, K J. Chang, Marvin L. Cohen, and Donald E. Morris

## Search for Isotope Effect in Superconducting Y-Ba-Cu-O

L. C. Bourne, M. F. Crommie, and A. Zettl

*Department of Physics, University of California, Berkeley, California 94720*

Hans-Conrad zur Loye, S. W. Keller, K. L. Leary, and Angelica M. Stacy

*Department of Chemistry, University of California, and Materials and Chemical Sciences Division, Lawrence Berkeley Laboratory, Berkeley, California 94720*

K. J. Chang and Marvin L. Cohen

*Department of Physics, University of California, and Materials and Chemical Sciences Division, Lawrence Berkeley Laboratory, Berkeley, California 94720*

and

Donald E. Morris

*Department of Physics, University of California, and Physics Division, Lawrence Berkeley Laboratory, Berkeley, California 94720*

(Received 28 April 1987)

An isotope effect has been searched for in the high- $T_c$  superconductor  $\text{YBa}_2\text{Cu}_3\text{O}_{7-\delta}$  through substitution of  $^{18}\text{O}$  for  $^{16}\text{O}$ . No shift in the superconducting transition temperature  $T_c$  is observed by electrical resistivity or magnetic susceptibility measurements. We discuss the implications of this result for mechanisms of superconductivity in the high- $T_c$  oxides.

PACS numbers: 74.30.-e, 74.70.Ya

The recent discovery of superconductivity above 30 K in  $\text{La}_{2-x}\text{Ba}_x\text{CuO}_4$  by Bednorz and Müller<sup>1</sup> has led to enormous success in the development of materials with high superconducting transition temperature  $T_c$ . Wu *et al.*<sup>2</sup> were the first to report superconductivity above 90 K in a mixed-phase Y-Ba-Cu-O compound; similar independent findings have since been published by Bourne *et al.*<sup>3</sup> and Hikami, Hirai, and Kagoshima.<sup>4</sup> Structural studies<sup>5,6</sup> indicate that the superconducting phase in these Y-Ba-Cu-O compounds is  $\text{YBa}_2\text{Cu}_3\text{O}_{7-\delta}$ .

An important feature common to both the layered perovskite  $\text{K}_2\text{NiF}_4$  structure of  $\text{La}_{2-x}\text{M}_x\text{CuO}_4$  ( $M = \text{Ba}$ ,  $\text{Sr}$ , or  $\text{Ca}$ ) and the distorted oxygen-defect perovskite structure of  $\text{R}\text{Ba}_2\text{Cu}_3\text{O}_{7-\delta}$  ( $R = \text{Y}$  or some lanthanide rare-earth elements) is the existence of two-dimensional copper-oxygen planes which are the most responsible for the metallic conduction. Band-structure calculations<sup>7,8</sup> demonstrate that the density of states at the Fermi level  $E_F$  is due largely to hybridized Cu  $3d$  and O  $2p$  states. It is therefore expected that vibration of the O ions would be involved if electron-phonon coupling is responsible for the superconductivity. For the La-based materials, it has been suggested<sup>9</sup> that the conventional electron-phonon Bardeen-Cooper-Schrieffer<sup>10</sup> (BCS) superconductivity mechanism is dominant. Strong electron-phonon couplings to O breathing-mode vibrations lead to electron-phonon coupling constants  $\lambda \geq 2$  and  $T_c$ 's on the order of 30–40 K.<sup>9</sup> On the other hand, the high transition temperatures and unusual two-dimensional electronic structure of the superconducting

oxides have also provoked theoretical descriptions which lie outside the conventional BCS theory.<sup>11,12</sup>

One of the most suggestive tests for electron-phonon coupling in conventional superconductors is the isotope effect,<sup>13</sup> where  $T_c$  is sensitive to ionic mass. In the original BCS treatment,<sup>10</sup>  $\alpha = \frac{1}{2}$  where  $\alpha$  is defined by

$$\alpha = -\partial(\ln T_c)/\partial(\ln M) = \frac{1}{2}(1 - \zeta). \quad (1)$$

Here,  $\zeta$  is a deviation from the ideal isotope effect where  $\alpha = \frac{1}{2}$ . The presence of an isotope effect demonstrates an important role played by the phonons in the superconductivity mechanisms.

We have searched for an isotope effect in  $\text{YBa}_2\text{Cu}_3\text{O}_{7-\delta}$ , by substituting the isotope  $^{18}\text{O}$  for  $^{16}\text{O}$ . With an estimated 90% of the oxygen sites replaced by  $^{18}\text{O}$  (see below), no shift in  $T_c$  is observed in either dc electrical resistivity measurements or dc magnetic susceptibility measurements. Our results show  $\alpha = 0.0 \pm 0.027$  indicating that, within our experimental resolution, there is no isotope shift for oxygen. We examine this value in the context of conventional BCS theory and other mechanisms for superconductivity.

Samples of  $\text{YBa}_2\text{Cu}_3\text{O}_{7-\delta}$  were prepared by solid-state reaction. Fine powders of  $\text{Y}_2\text{O}_3$ ,  $\text{BaCO}_3$ , and  $\text{CuO}$  were mixed in stoichiometric proportions, dissolved in nitric acid, evaporated to dryness, and decomposed at 800°C. The powder was then reground, pressed into a pellet, sintered in an  $^{16}\text{O}$  atmosphere at 950°C, and finally slow cooled. The uniform parent pellet was then broken into several pieces for subsequent treatment and

measurement. Resistivity measurements on the parent specimen indicated a sharp transition to the superconducting state at  $T_c = 90$  K (transition midpoint). Powder x-ray diffraction showed that the sample consisted of a single phase.

In order to determine the appropriate conditions for oxygen isotope substitution, a series of careful thermogravimetric analysis (TGA) experiments were performed on several  $\text{YBa}_2\text{Cu}_3\text{O}_{7-\delta}$  samples from different preparation batches. The time constant for oxygen diffusion was found to be very short, and dependent on temperature and also sample grain size.<sup>14</sup> It was found that oxygen in the sample freely exchanged with oxygen in the surrounding atmosphere at  $950^\circ\text{C}$ . A portion of the original  $\text{YBa}_2\text{Cu}_3\text{O}_{7-\delta}$  parent pellet was placed in an  $^{18}\text{O}$  atmosphere at  $950^\circ\text{C}$  for 12 h and slow cooled to  $200^\circ\text{C}$  over 6 h; the temperature and reaction time were chosen to ensure maximal  $^{18}\text{O}$  exchange with both weakly and tightly bound oxygen sites in the specimen. The isotope exchange takes place without change of the oxidation state of the material, and the superconducting transition remains sharp. A nearly identical treatment was performed on another portion of the parent pellet, with use of an  $^{16}\text{O}$  atmosphere. This treated  $^{16}\text{O}$  sample was used as the control sample.

Figure 1(a) shows  $R$  vs  $T$  for the  $^{16}\text{O}$  control sample

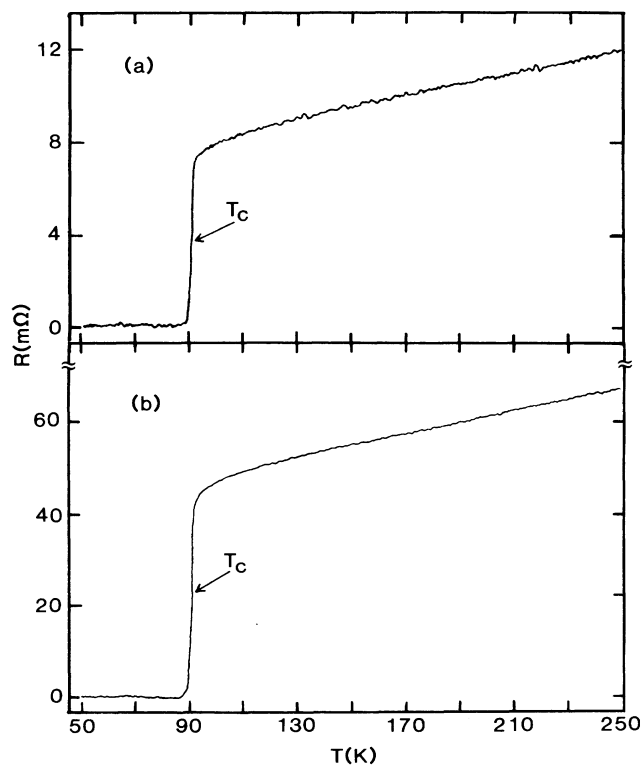


FIG. 1. (a)  $R$  vs  $T$  for  $\text{YBa}_2\text{Cu}_3\text{O}_{7-\delta}$  with  $^{16}\text{O}$ . (b) Same as (a) but for  $^{18}\text{O}$ . No isotope effect is observed.

of  $\text{YBa}_2\text{Cu}_3\text{O}_{7-\delta}$  near the superconducting transition region. The data indicate a midpoint transition  $T_c = 90.0$  K, nearly identical to that determined for the untreated parent pellet. Figure 1(b) shows similar  $R$  vs  $T$  data for the  $^{18}\text{O}$  sample. The midpoint of the transition is at  $90.3$  K. Within the experimental uncertainty of  $\sim 0.3$  K (taking into account possible temperature gradients across the measurement probe, and variations of  $T_c$  in different portions of the parent pellet), we find<sup>14,15</sup> *no isotope effect* in  $\text{YBa}_2\text{Cu}_3\text{O}_{7-\delta}$ , i.e.,  $\alpha = 0.0 \pm 0.027$ . A similar conclusion is drawn from a careful comparison of magnetic susceptibility  $\chi$  measurements of the parent pellet and the treated  $^{18}\text{O}$  specimen. In Fig. 2, the  $\chi$  for an  $^{18}\text{O}$  sample is displayed.

Temperature-programmed-desorption measurements (employing a quadrupole mass spectrometer) on the specimens treated in  $^{18}\text{O}$  determine how much of the  $^{16}\text{O}$  was actually replaced by  $^{18}\text{O}$ . A total of  $900 \mu\text{mol/g}$  of O were initially removed from the sample by heating to  $800^\circ\text{C}$  in flowing He. The oxygen desorbed as  $\text{O}_2$ , with an  $^{18}\text{O}$  to  $^{16}\text{O}$  ratio of 90%. In addition, small amounts of  $\text{H}_2\text{O}$ ,  $\text{CO}_2$  and CO desorbed (the samples had been exposed to air). The sample was then heated in flowing  $\text{H}_2$ , destructively reducing the sample to  $\text{Y}_2\text{O}_3$ ,  $\text{BaO}$ , and  $\text{CuO}$ , and an additional  $4100 \mu\text{mol/g}$  of O desorbed in the form of  $\text{H}_2\text{O}$  with an isotope ratio of 90%. Of the total amount of oxygen removed from the specimen (50% of the total specimen content), 90% was  $^{18}\text{O}$ .

For the La-based oxides, the frequency of the O breathing mode is found to be very high and this high frequency as a prefactor in the  $T_c$  equation is suggested to predominate in the high- $T_c$  superconductivity.<sup>7,9</sup> If we make a similar assumption for the Y-based samples studied here, then it is possible to examine the magnitudes of the parameters entering the BCS equation. In this analysis, it is assumed that the oxygen vibrations dominate and comparison is made with transition metals where a reduced and sometimes zero ( $\alpha \approx 0$ ,  $\zeta \approx 1$ ) iso-

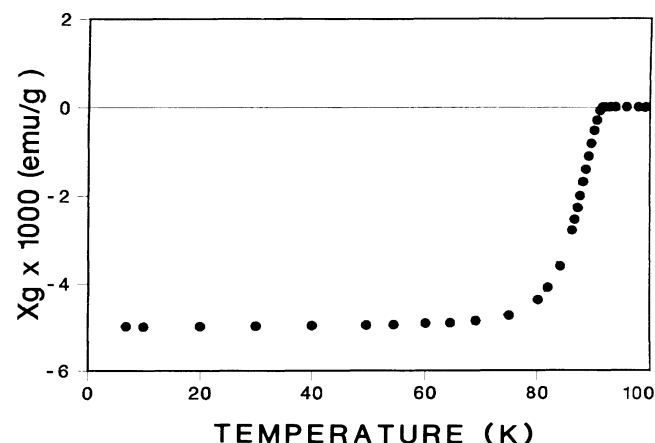


FIG. 2.  $\chi$  vs  $T$  for  $\text{YBa}_2\text{Cu}_3\text{O}_{7-\delta}$  with  $^{18}\text{O}$ .

tope effect is found<sup>16</sup> as in Ru and Zr. In contrast, for simple (*sp*) BCS superconductors  $\zeta \sim 0.1$ .

Garland<sup>17</sup> explored the reduced isotope effect for the transition metals. He found that the large and narrow *d*-band density of states changes the effective Coulomb interaction; thus the isotope effect can be absent despite the phonon mechanism being the dominant attractive interaction. If a simple two-square-well model<sup>18</sup> is used within the BCS theory, the parameter  $\zeta$  is algebraically obtained:

$$\zeta = [\mu^*/(\lambda^* - \mu^*)]^2, \quad (2)$$

where  $\lambda^* [= \lambda/(1+\lambda)]$  and  $\mu^*$  are the effective phonon and Coulomb interaction parameters, and  $\lambda$  is the electron-phonon coupling constant. The isotope effect is completely reduced when  $\mu^* \gg \lambda^*$  or  $\mu^* = \lambda^*/2$ . For the Y-Ba-Cu-O system, if we use the measured  $T_c$  and assume a Debye temperature  $\theta_D \approx 400$  K, then  $\mu^* = 0.63$  and  $\lambda^* = 1.25$  which is an unphysical result. When  $\theta_D \approx 800$  K is used,  $\lambda$  is in the range of 7 while  $\mu^*$  is about 0.45. We also tested an expression for  $T_c$  recently developed by Kresin<sup>19</sup> from the Eliashberg equation in the strong-coupling limit of  $\lambda$ . With the use of the same  $T_c$  and  $\theta_D$ , we found that  $\mu^*$  is approximately  $\lambda/2$  for  $\lambda$  in the range of 10. Since the  $T_c$  equation is obtained in the strong-coupling limit of  $\lambda$ , the value for  $\mu^*$  which yields the zero isotope effect is significantly larger than those found in most superconducting materials, suggesting that the Y-Ba-Cu-O compound has a strong effective Coulomb interaction.

Finally, the present experiments do not rule out non-phonon mechanisms such as the resonating-valence-bond model,<sup>11</sup> the two-dimensional plasmon model,<sup>12</sup> the negative-*U* mechanism,<sup>20</sup> or other Coulomb-induced attractions. Since these mechanisms are not necessarily related to the ionic mass, the isotope effect may not be expected.

In conclusion, we have found that the oxygen isotope effect is absent in superconducting Y-Ba-Cu-O. This result suggests that although the BCS phonon mechanism may still be appropriate for explaining the high- $T_c$  superconducting behavior of this material, the parameters,  $\theta_D$ ,  $\lambda^*$ , and  $\mu^*$  would be severely constrained. Hence the absence of an oxygen isotope shift suggests that a nonphonon mechanism may be important in this material.

This research was supported by National Science Foundation Grants No. DMR-83-19024 (K.J.C. and M.L.C.), No. CHE-83-51881 (A.M.S.), and No. DMR-

84-00041 (A.Z.), and by the Director, Office of Energy Research, Office of Basic Energy Sciences, Materials Sciences Division of the U.S. Department of Energy under Contract No. DE-AC03-76SF00098. One of us (A.Z.) acknowledges support from the Alfred P. Sloan Foundation, and another (D.E.M.) thanks Professor D. W. Fuerstenau for providing the TGA apparatus, U. M. Scheven for carrying out the TGA measurements, and Professor R. A. Muller for encouragement and support during the course of this work.

<sup>1</sup>J. G. Bednorz and K. A. Müller, *Z. Phys. B* **64**, 189 (1986).

<sup>2</sup>M. K. Wu, J. R. Ashburn, C. J. Torng, P. H. Hor, R. L. Meng, L. Gao, Z. J. Huang, Y. Q. Wang, and C. W. Chu, *Phys. Rev. Lett.* **58**, 908 (1987).

<sup>3</sup>L. C. Bourne, M. L. Cohen, W. N. Creager, M. F. Crommie, A. M. Stacy, and A. Zettl, *Phys. Lett.* **120**, 494 (1987).

<sup>4</sup>S. Hikami, T. Hirai, and S. Kagoshima, *Jpn. J. Appl. Phys.* **26**, L314 (1987).

<sup>5</sup>R. J. Cava, B. Batlogg, R. B. van Dover, D. W. Murphy, S. Sunshine, T. Siegrist, J. P. Remeika, E. A. Rietman, S. Zahvra, and G. P. Espinosa, *Phys. Rev. Lett.* **58**, 1676 (1987).

<sup>6</sup>P. M. Grant, R. B. Beyers, E. M. Engler, G. Lim, S. S. P. Parkin, M. L. Tamirez, V. Y. Lee, A. Nazzal, J. E. Vazyvez, and R. J. Savoy, *Phys. Rev. B* **35**, 7242 (1987).

<sup>7</sup>L. F. Mattheiss, *Phys. Rev. Lett.* **58**, 1028 (1987).

<sup>8</sup>J. Yu, A. J. Freeman, and J.-H. Xu, *Phys. Rev. Lett.* **58**, 1035 (1987).

<sup>9</sup>W. Weber, *Phys. Rev. Lett.* **58**, 1371 (1987).

<sup>10</sup>J. Bardeen, L. N. Cooper, and J. R. Schrieffer, *Phys. Rev.* **106**, 162 (1957), and **108**, 1175 (1957).

<sup>11</sup>P. W. Anderson, *Science* **235**, 1196 (1987).

<sup>12</sup>V. Z. Kresin, *Phys. Rev. B* (to be published).

<sup>13</sup>E. Maxwell, *Phys. Rev.* **78**, 477 (1950); C. A. Reynolds, B. Serin, W. H. Wright, and L. B. Nesbitt, *Phys. Rev.* **78**, 487 (1950).

<sup>14</sup>D. E. Morris *et al.*, in Proceedings of the Symposium S of the 1987 Spring Meeting of the Materials Research Society, 21-25 April 1987, Anaheim, California (Materials Research Society, Pittsburgh, PA, to be published), Vol. EA-11.

<sup>15</sup>M. L. Cohen, A. M. Stacy, and A. Zettl, in Ref. 14.

<sup>16</sup>C. Kittel, *Introduction to Solid State Physics* (Wiley, New York, 1976), p. 368.

<sup>17</sup>J. W. Garland, *Phys. Rev. Lett.* **11**, 111, 114 (1963).

<sup>18</sup>N. N. Bogoliubov, V. V. Tolmachev, and D. V. Shirkov, *A New Method in the Theory of Superconductivity* (Academy of Science, Moscow, 1958; translated by Consultants Bureau, New York, 1959).

<sup>19</sup>V. Z. Kresin, unpublished.

<sup>20</sup>H.-B. Schüttler, M. Jarrell, and D. J. Scalapino, *Phys. Rev. Lett.* **58**, 1147 (1987).