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# Seasonal Variation and Sources of Carbonaceous Species and Elements in PM2.5 and PM10 Over the Eastern Himalaya

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1	Seasonal variation and sources of carbonaceous species and elements in $PM_{2.5}$ and $PM_{10}$
2	over the eastern Himalaya
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#### 30 Abstract

31 The study represents the seasonal characteristics (carbonaceous aerosols and elements) and contribution of 32 prominent sources PM<sub>2.5</sub> and PM<sub>10</sub> in the high altitude of the eastern Himalaya (Darjeeling) during August 2018-33 July 2019. Carbonaceous aerosols [organic carbon (OC), elemental carbon (EC) and water soluble organic carbon 34 (WSOC)] and elements (Al, Fe, Ti, Cu, Zn, Mn, Cr, Ni, Mo, Cl, P, S, K, Zr, Pb, Na, Mg, Ca, and B) in PM<sub>2.5</sub> and PM<sub>10</sub> were analyzed to estimate their possible sources. The annual average concentration of PM<sub>2.5</sub> and PM<sub>10</sub> were 35 computed as 37±12 µg m<sup>-3</sup> and 58±18 µg m<sup>-3</sup>, respectively. In the present case, total carbonaceous species in PM<sub>2.5</sub> 36 37 and PM<sub>10</sub> were accounted for 20.6% of PM<sub>2.5</sub> and 18.6% of PM<sub>10</sub>, respectively. Whereas, trace elements in PM<sub>2.5</sub> 38 and  $PM_{10}$  were estimated as 15% of  $PM_{2.5}$  and 12% of  $PM_{10}$ , respectively. Monthly are seasonal variations in 39 concentrations of carbonaceous aerosols and elements in PM2.5 and PM10 were also observed during the 40 observational period. The positive relationship between OC & EC and OC & WSOC of PM2.5 and PM10 during all 41 the seasons (except monsoon in case of  $PM_{10}$ ) indicate rheir common sources. The enrichment factors (EFs) and 42 significant positive correlation of Al with othe crustal elements (Fe, Ca, Mg and Ti) of fine and coarse mode 43 aerosols indicates the influence of mineral dust at the Darjeeling. Principal component analysis (PCA) resolved the 44 four common sources (biomass burning + fossil fuel combustion (BB+FFC), crustal/soil dust, vehicular emissions 45 (VE) and industrial emissions (IE)) of PM<sub>2.5</sub> and PM<sub>10</sub> in Darjeeling.

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47 Keywords: PM<sub>2.5</sub>, PM<sub>10</sub>, OC, EC, WSOC, Elements, PCA

#### 49 1. Introduction

50 The urban agglomeration of hilly region of India is under serious threat various forms of anthropogenic activities 51 and pollutants (Gajananda et al. 2005; Sharma et al. 2020a;b). Due to rapid urbanization, industrialization, 52 increasing vehicular traffic due to increased tourism-related activities, increase in energy demands (for lighting, 53 cooking and heating etc.) have resulted in contamination/deterioration of ambient air quality, vegetation and climate 54 of the hilly region (Chatterjee et al. 2010; Rai et al. 2020a). From last few decades the fine and coarse mode of 55 aerosol/particulate matter (PM) has been altered the atmospheric particle chemistry. Earth's climate system 56 (radiative balance), and human health (Pant and Harrison 2012; Bond et al. 2013). Therefore, quantification and 57 identification of chemical species and sources of aerosol is essential to explore the appropriate mitigation option to 58 improve the ambient air quality, human health and climate (Ramana et al. 2010; Cao et al. 2006; Bond et al. 2013; 59 Pope et al. 2009; Ramgolam et al. 2009; Sharma et al. 2018 a; b; Rai et al., 2020b).

60 Carbonaceous aerosols (CAs) are the major fractions (OC, and EC) of PM<sub>2.5</sub> and PM<sub>10</sub> in the atmosphere and 61 have significant role in the Earth's radiative balance, visibility degradation, alteration of atmospheric chemistry 62 (Lim and Turpin 2002; Hansen et al. 2005; Bond et al. 2013; Pope et al. 2009). Various toxic gases (CO, CO<sub>2</sub>, SO<sub>3</sub>, 63 NH<sub>3</sub>, and NOx, etc.,) organics and hydrocarbons (like, VOCs: volatile organic compounds; PAHs: polycyclic 64 aromatic hydrocarbons) are produced during the combustion process/emissions of CAs and therefore affect the 65 respiratory/cardiovascular system of the human (Lighty et al. 2000; Pope et al. 2009). Both primary and secondary 66 organic aerosols significantly control of physico-chemical properties of particles/aerosols (Kanakidou et al. 2005) 67 and influenced the formation cloud condensation nuclei (CCN), whereas EC absorb solar radiation and contribute 68 for radiative change (Bond et al., 2013).

69 PM consists of organics (OC, EC, and other components), mineral/crustal/soil dust, metals, non-metals, 70 inorganic pollutants as well as sea salts and relative exuberance of these components are highly variable both 71 spatially and temporally (Ram et al. 2011; Jain et al. 2017). Generally, the elements linked with the PM are non-72 volatile in nature and remain unaffected even though they go for regional as well as long-range transportation 73 (Morawska and Zhang, 2002). Some of the transition/toxic metals (like Fe, Cu, Mn, Zn, Ni, Cr, As, Pb, Hg) which 74 are coming from the various sources into the ambient air have acute toxic and mutagenic effects on human health, 75 when inhaled at higher concentrations. Elements like Fe, Al, Si, Ca and Ti are also available in the fine and coarse 76 mode PM is originated from mineral dust/crustal dust (Sharma et al. 2014a). Soluble K in PM mostly originated from biomass burning, however, it is also attributed to dust in PM (Viana et al. 2008), whereas Cl originated from
sea salt is also considered from coal burning in aerosols (Pant and Harrison 2012).

79 Several studies conducted in past on carbonaceous aerosols, inorganic aerosols and elements in particulate 80 matter and their potential sources in urban (Ram and Sarin, 2011; Mandal et al. 2014; Sharma et al. 2016; Gupta et 81 al. 2018; Shivani et al. 2019; Jain et al. 2020a;b), rural, remote (Begam et al. 2017) as well as high altitude 82 atmosphere (Ram et al. 2008; Kumar and Attri, 2016; Sarkar et al. 2017; Kaushal et al. 2018) but very limited study 83 (Sharma et al. 2020b) has been conducted in the eastern Himalayan region of India. Considering the importance of 84 atmospheric carbonaceous species and elements in  $PM_{2.5}$  and  $PM_{10}$  and from their perspective on climate change, 85 the present study has been carried out at Darjeeling (the eastern Himalaya). In this paper, we report the 86 carbonaceous components (OC, EC, and WSOC) and elements (Al, Fe, Ti, Cu, Zn, Mn, Cr, Ni, Mo, Cl, P, S, K, Zr, 87 Pb, Na, Mg, Ca, and B) of PM<sub>2.5</sub> and PM<sub>10</sub> over the eastern Himalaya (Darjeeling) during August 2018 – July 2019. Principal component analysis (PCA) and HYSPLIT trajectory were applied to resolve the potential sources and 88 89 source region, respectively of PM<sub>2.5</sub> and PM<sub>10</sub> at Darjeeling.

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#### 91 2. Materials and methods

#### 92 **2.1 Description of study site**

93 The study was carried out in the campus of Bose Institute, Darjeeling (27.01 °N and 88.15°E; 2200 m asl) 94 situated in the Eastern Himalaya (Fig. 1). The hilly districts of Darjeeling are situated within the lower and sub-95 Himalayan ranges of the eastern Himalaya. Darjeeling is surrounded by Sikkim Himalayas, Bhutan Himalaya and 96 Nepal Himalaya from the north, east and west sides, respectively (Chatterjee et al. 2021). The study area is 97 considered as a semi-urban and is under he influence of biomass burning, vehicle emissions, agriculture and tourism 98 activities (Adak et al. 2014). The climate of Darjeeling is subtropical and temperate with wet summer/pre-monsoon 99 and monsoonal rains (June-September). The annual average maximum temperature was estimated to be 12.2 °C (in 100 summer: March-May) whereas the average minimum temperature was estimated as 3.4 °C (in winter: January-February) with monthly mean temperatures ranging from 5.8 to 17.2 °C. The monthly average meteorology 101 102 (ambient temperature, relative humidity (RH), wind speed and wind directions) during the study period are depicted 103 in Fig. S1 (see the supplementary information). The detail information on topography, weather, township and 104 influences of local pollution sources at the study site is described in Chatterjee et al. (2021).

#### 106 2.2 Sample collection and analysis

107  $PM_{2.5}$  (n = 94) and  $PM_{10}$  (n = 102) samples were collected simultaneously on pre-combusted (at 550 °C) 108 Pallflex tissue quartz filters (47mm for  $PM_{2.5}$  and  $20 \times 25$  cm<sup>2</sup> for  $PM_{10}$ ) using Fine Particle Sampler (flow rate:  $1m^3$ 109  $h^{-1} \pm 2\%$ ) and Respirable Particle Sampler (flow rate: 1.2 m<sup>3</sup> min<sup>-1</sup>  $\pm 2\%$ ), respectively for 24h from August 2018-July 2019 (PM2.5 sampling was not carried out in July 2019). The particle sampler was placed on the rooftop of the 110 Bose Institute at the height of 15m above the ground level. The gravimetric mass (in  $\mu$ g) of PM<sub>2.5</sub> and PM<sub>10</sub> was 111 112 computed by the difference between the initial and final weight using a calibrated weighing balance (M/s. Sartorius, 113 resolution:  $\pm 10 \ \mu g$ ) of the filter. The concentrations of PM<sub>2.5</sub> and PM<sub>10</sub> (in  $\mu g \ m^{-3}$ ) were further computed by 114 dividing the PM mass to the total volume of air passed during the sampling period (24h). Before chemical analysis 115 the samples were stored at  $-20^{\circ}$ C.

116 The concentrations of OC and EC in  $PM_{2.5}$  and  $PM_{10}$  samples were estimated by OC/EC Carbon analyzer 117 (Model: DRI 2001A; Make: Atmoslytic Inc., Calabasas, CA, USA) using IMPROVE-A Protocol (Chow et al. 118 2004). OC/EC carbon analyzer is working on the preferential oxidation of OC and EC at different temperatures 119 plateaus (140, 280, 480 and 580°C; for OC1, OC2, OC3 and OC4, respectively) in pure helium and three 120 temperature plateaus (580, 740 and 840°C for EC1, EC2 and EC3, respectively) in 98 % helium and 2 % oxygen 121 (Chow et al. 2004). A proper punch of  $\sim 0.536$  cm<sup>2</sup> area of the filter was cut and analyzed in triplicate along with 122 field blank filters. The standard calibration for peak area verification was performed daily using 5% CH<sub>4</sub>+balance 123 helium gas (before and after sample analysis). Calibration of the OC/EC analyzer was performed by 4.8% of CO<sub>2</sub> + 124 balance He gas along with known amounts of KHP (potassium hydrogen phthalate) and sucrose solution (Sharma et 125 al. 2020b). In the present case, repeatability error of OC and EC analysis were estimated as 3-7% (n=3). Total 126 carbonaceous aerosols (TCA) of PM<sub>2.5</sub> and PM<sub>10</sub> are computed as (1.6×OC)+EC (Rengarajan et al. 2007; Srinivas and Sarin 2014). 127

128 A known punch size of  $PM_{2.5}$  (~3.46 cm<sup>2</sup>) and  $PM_{10}$  (~7.07 cm<sup>2</sup>) filters were cut into four halves and was 129 soaked in 20 ml of de-ionized water (18.2 M $\Omega$ -cm) and ultrasonicated three times for 10 min each. The 10ml filtered 130 extract was transferred into a pre-cleaned glass vial and analyzed for WSOC and WIOC using a TOC analyzer 131 operating on catalytically-aided combustion oxidation (Model: Shimadzu TOC-L CPH/CPN, Japan). Field blank 132 filters were also analyzed using the same analytical procedure and subtracted from the respective  $PM_{2.5}$  and  $PM_{10}$ 

133	samples to obtain the analytical results (WSOC and WIOC = OC-WSOC)). Calibration of the instrument was						
134	performed before analysis of the samples using standard solution of five different concentrations. 3-10% of						
135	repeatability errors were estimated in WSOC analysis ( $n=3$ ). The detail, analytical procedures is described in Rai et						
136	al. (2020a).						
137	Wavelength Dispersive X-ray Fluorescence Spectrometer (WD-XRF) was used to analyzed the concentrations						
138	of Al, Fe, Ti, Cu, Zn, Mn, Cr, Ni, Mo, Cl, P, S, K, Zr, Pb, Na, Mg, Ca, and B in PM <sub>2.5</sub> and PM <sub>10</sub> (with repeatability						
139	errors 5-10%). The samples were analyzed in triplicate and for major and trace elements. The detail, analytical						
140	procedures is available in Jain et al. (2020a;b).						
141							
142	2.3 Primary organic carbon (POC) and secondary organic carbon (SOC)						
143	The concentrations of POC in PM <sub>2.5</sub> and PM <sub>10</sub> samples are estimated using minimum OC/EC ratio for the each						
144	season (winter, summer, monsoon and post-monsoon). Both OC and EC in ambient aerosols are considered to be						
145	originated from combustion sources and EC is a good marker for POC (Castro et al. 1999). POC are computed using						
146	the following equation,						
147	$POC = [OC/EC]_{min} \times [EC] + c $ (1)						
148	where, c is the contribution from non-combustion sources which is negligible in the present case. SOC is estimated						
148 149	where, c is the contribution from non-combustion sources which is negligible in the present case. SOC is estimated as the difference of OC and POC (SOC= OC-POC).						
149							
149 150	as the difference of OC and POC (SOC= OC–POC).						
149 150 151	as the difference of OC and POC (SOC= OC–POC). 2.4 Enrichment Factors (EFs)						
149 150 151 152	as the difference of OC and POC (SOC= OC-POC). <b>2.4 Enrichment Factors (EFs)</b> Crustal EFs referes the origin of the elements (either anthropogenic or natural) and their abundance in the						
149 150 151 152 153	as the difference of OC and POC (SOC= OC–POC). <b>2.4 Enrichment Factors (EFs)</b> Crustal EFs referes the origin of the elements (either anthropogenic or natural) and their abundance in the ambient particulates (Amato et al., 2016). The EFs of the elements present in PM <sub>2.5</sub> and PM <sub>10</sub> samples are computed						
149 150 151 152 153 154	as the difference of OC and POC (SOC= OC-POC). <b>2.4 Enrichment Factors (EFs)</b> Crustal EFs referes the origin of the elements (either anthropogenic or natural) and their abundance in the ambient particulates (Amato et al., 2016). The EFs of the elements present in PM <sub>2.5</sub> and PM <sub>10</sub> samples are computed (Taylor and McLennan, 1995) as:						
149 150 151 152 153 154 155	as the difference of OC and POC (SOC= OC–POC). <b>2.4 Enrichment Factors (EFs)</b> Crustal EFs referes the origin of the elements (either anthropogenic or natural) and their abundance in the ambient particulates (Amato et al., 2016). The EFs of the elements present in PM <sub>2.5</sub> and PM <sub>10</sub> samples are computed (Taylor and McLennan, 1995) as: $EF = \frac{El_{ample}/X_{sample}}{El_{cms}/X_{sample}} \qquad (2)$						
149 150 151 152 153 154 155 156	as the difference of OC and POC (SOC= OC–POC). 2.4 Enrichment Factors (EFs) Crustal EFs referes the origin of the elements (either anthropogenic or natural) and their abundance in the ambient particulates (Amato et al., 2016). The EFs of the elements present in PM <sub>2.5</sub> and PM <sub>10</sub> samples are computed (Taylor and McLennan, 1995) as: $EF = \frac{El_{sample}/X_{sample}}{El_{ons}/X_{sample}}$ (2) where,						
149 150 151 152 153 154 155 156 157	as the difference of OC and POC (SOC= OC-POC). 2.4 Enrichment Factors (EFs) Crustal EFs referes the origin of the elements (either anthropogenic or natural) and their abundance in the ambient particulates (Amato et al., 2016). The EFs of the elements present in PM <sub>2.5</sub> and PM <sub>10</sub> samples are computed (Taylor and McLennan, 1995) as: $EF = \frac{El_{umpk}/X_{sample}}{El_{rum}/X_{sample}}$ (2) where, El <sub>sample</sub> = element (EI) mass concentration						

Aluminium (Al) is used as the reference element in this study, which is also supported by the previous studies(Sharma et al. 2014a; 2020b).

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#### 164 2.5 Principal component analysis (PCA)

In this study, PCA was applied on chemical constituents of  $PM_{2.5}$  and  $PM_{10}$  to identify the potential sources contributing to fine and coarse fractions aerosols. It uses orthogonal decomposition to identify individual groups of components which are then tied to variables by loading factors (Viana et al. 2008). The total variability of the data sets is shared between these components, whereas the maximum is shared by the first component (Belize et al. 2013). In PCA, the chemical data are transformed into a dimensionless standardized form:

170 
$$Zij = \frac{Cij - \overline{Cj}}{\sigma j}$$
(2)

where i = 1, ..., n samples; j=1, ...m elements; *Cij* is the concentration of element *j* in sample *i*; and  $\overline{Cj}$  and  $\sigma j$  are the arithmetic mean concentration and the standard deviation for element *j*, respectively. The PCA model is expressed as:

174 
$$Z_{ij} = \sum_{k=1}^{p} g_{ik} h_{kj}$$
 (3)

175 Where k = 1, ..., p sources, and  $g_{ik}$  and  $h_{kj}$  are the factor loadings and the factor scores, respectively. This equation is 176 solved by Eigenvector decomposition (Song et al. 2006).

177

#### 178 **3. Results and discussion**

#### 179 3.1. Seasonal variations of chemical species of PM<sub>2.5</sub> and PM<sub>10</sub>

#### 180 3.1.1. Seasonal variability of PM<sub>2.5</sub> and PM<sub>10</sub>

181 The annual average PM<sub>2.5</sub> and PM<sub>10</sub> concentrations were estimated as  $37\pm12 \ \mu g \ m^{-3}$  (range: 16–77  $\mu g \ m^{-3}$ ) and  $56\pm18 \ \mu g \ m^{-3}$  (range: 21–116  $\mu g \ m^{-3}$ ), respectively (Table 1). The maximum monthly average mass concentration of 182  $PM_{2.5}$  was recorded in October (58 µg m<sup>-3</sup>) and the minimum monthly average concentration of  $PM_{2.5}$  was observed 183 in January (27 µg m<sup>-3</sup>). Whereas, monthly average maxima of PM<sub>10</sub> was observed in March (73 µg m<sup>-3</sup>) and monthly 184 average minima in the month of January (43 µg m<sup>-3</sup>). The monthly average and temporal variations in PM<sub>2.5</sub> and 185  $PM_{10}$  concentrations are shown in Figs.S2-S3 (see the supplementary information). The highest seasonal average 186 concentration of both PM<sub>2.5</sub> (41±14  $\mu$ g m<sup>-3</sup>) and PM<sub>10</sub> (64±20  $\mu$ g m<sup>-3</sup>) were found during summer and minimum 187 188 seasonal average mass concentration of both  $PM_{2.5}$  (31±9 µg m<sup>-3</sup>) and  $PM_{10}$  (51±17 µg m<sup>-3</sup>) were recorded during

189 winter. In post-monsoon, the mass concentrations of  $PM_{2.5}$  (40±11 µg m<sup>-3</sup>) and  $PM_{10}$  (55±17 µg m<sup>-3</sup>) were recorded 190 higher than the winter and monsoon (Table 1). In the present case, non-significant seasonal variation in mass 191 concentrations of both  $PM_{2.5}$  and  $PM_{10}$  was recorded at Darjeeling (except winter). Sharma et al. (2014b) reported 192 the similar concentration of PM<sub>2.5</sub> (at Kullu:  $34\pm2 \ \mu g \ m^{-3}$ ; at Shimla:  $32\pm3 \ \mu g \ m^{-3}$ ) at the north-western Himalayan 193 region during winter 2013, whereas Kaushal et al. (2018) recorded the 52±18  $\mu$ g m<sup>-3</sup> of PM<sub>10</sub> concentration at 194 Pohara of north-western Himalaya (Himachal Pradesh) during winter 2015. Sharma et al. (2020a) reported the  $PM_{10}$ concentration as  $54\pm12 \mu g$  m<sup>-3</sup> and  $39\pm10 \mu g$  m<sup>-3</sup> during post-monsoon and winter, respectively at Nainital a central 195 196 Himalaya. Sharma et al. (2020b) also reported the  $PM_{10}$  concentration as  $48\pm16 \ \mu g \ m^{-3}$  during winter in Darjeeling. 197 The monthly variation in mass concentration of PM<sub>2.5</sub> and PM<sub>10</sub> in relation ambient temperature, RH, wind direction 198 and wind speed is depicted in Figs.S1-S2 (see the supplementary information). It has been observed that the 199 prevailing meteorology of the sampling site influenced the seasonal variation in mass concentrations of PM<sub>2.5</sub> and 200  $PM_{10}$  at Darjeeling. However, the higher concentration of pollutants during the summer season at Darjeeling might 201 be due to more influence of tourism activities (Gajananda et al. 2005; Chatterjee et al. 2021) as well as long-range 202 transportation of pollutants at sampling site of the Darjeeling (Rai et al. 2020a).

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#### 204 3.1.2. Variation of OC, EC and WSOC in PM<sub>2.5</sub> and PM<sub>10</sub>

205 Temporal variations in OC, EC and WSOC concentrations of PM<sub>2.5</sub> and PM<sub>10</sub> during study period are shown in Fig.S3 (b-d) (see the supplementary information). The annual average concentrations of OC, EC, WSOC and WIOC 206 of PM<sub>2.5</sub> were  $3.46\pm1.59 \ \mu g \ m^{-3}$ ,  $1.93\pm0.93 \ \mu g \ m^{-3}$ ,  $1.88\pm1.05 \ \mu g \ m^{-3}$  and  $1.69\pm0.85 \ \mu g \ m^{-3}$ , respectively. Whereas 207 the annual average of OC, EC, WSOC and WIOC in  $PM_{10}$  were 5.06±2.03 µg m<sup>-3</sup>, 2.34±1.18 µg m<sup>-3</sup>, 3.55±1.75 µg 208 209  $m^{-3}$  and 1.51±0.92 µg  $m^{-3}$ , respectively. Figure 2 shows the monthly average of OC, EC and WSOC in PM<sub>2.5</sub> and 210  $PM_{10}$  during study period in Darjeeling. Highest monthly average OC in  $PM_{2.5}$  (4.76 µg m<sup>-3</sup>) and  $PM_{10}$  (6.30 µg m<sup>-3</sup>) 211 was found in November (post-monsoon), whereas monthly average minima of OC in  $PM_{2.5}$  (2.03 µg m<sup>-3</sup>) and  $PM_{10}$ 212 (3.31 µg m<sup>-3</sup>) was found in August (monsoon) and May (summer), respectively. Similar monthly average maxima 213 and minima of EC (in  $PM_{2.5}$  and  $PM_{10}$ ) was recorded in March and August, respectively (Fig. 2). Highest monthly 214 average WSOC concentration in both PM<sub>2.5</sub> (2.77 µg m<sup>-3</sup>) and PM<sub>10</sub> (5.51 µg m<sup>-3</sup>) was recorded in November (post-215 monsoon) might be due to the influence of stubble burning in northern states of India (Punjab, Haryana and western 216 Utter Pradesh of IGP). Figure 3 shows the seasonal average OC, EC, WSOC, WIOC, POC, SOC and TCA concentrations of  $PM_{2.5}$  and  $PM_{10}$  during winter, summer, monsoon and post-monsoon seasons in Darjeeling. Highest seasonal average OC concentration of  $PM_{2.5}$  (4.07±1.55 µg m<sup>-3</sup>) and  $PM_{10}$  (5.69±2.09 µg m<sup>-3</sup>) was recorded in post-monsoon season and minimum seasonal average concentration of OC in  $PM_{2.5}$  (2.21±0.86 µg m<sup>-3</sup>) and  $PM_{10}$ (3.62±0.86 µg m<sup>-3</sup>) was recorded during monsoon (Table 1). Similarly, seasonal average maxima and minima of EC in  $PM_{2.5}$  and  $PM_{10}$  were recorded in summer and monsoon seasons, respectively. Average WSOC in both  $PM_{2.5}$ (2.17±0.98 µg m<sup>-3</sup>) and  $PM_{10}$  (4.44±1.55 µg m<sup>-3</sup>) were found highest during post-monsoon season and minimum in monsoon season (Fig. 3).

224 The annual average concentration of TCA contributes 20.6% of  $PM_{2.5}$  (7.6±3.4 µg m<sup>-3</sup>) and 18.6% of  $PM_{10}$ 225 (10.4±4.3 µg m<sup>-3</sup>). The highest TCA in PM<sub>2.5</sub> and PM<sub>10</sub> were recorded in winter (22.7% of PM<sub>2.5</sub> and 22.4% of 226  $PM_{10}$ ) season followed by post-monsoon (21.7% of  $PM_{2.5}$  and 20.3% of  $PM_{10}$ ), summer (21.6% of  $PM_{2.5}$  and 17.6% 227 of PM<sub>10</sub>) and monsoon (12.5% of PM<sub>2.5</sub> and 13.4% of PM<sub>10</sub>) seasons. The annual average concentration of POC in 228  $PM_{2.5}$  was recorded to be 2.35 ± 1.06 µg m<sup>-3</sup> (66% of OC), and SOC was 1.19±0.57 µg m<sup>-3</sup> (34% of OC). Similarly, the annual average POC concentration in PM<sub>10</sub> was recorded as  $3.18 \pm 1.13 \ \mu g \ m^{-3}$  (63% of OC), whereas, SOC was 229 recorded as 2.05±0.98 µg m<sup>-3</sup> (37% of OC). In PM<sub>2.5</sub> the seasonal contribution of POC and SOC were ranging from 230 231 55-77% and 33-45% of OC, respectively, whereas in PM<sub>10</sub> the seasonal contribution of POC and SOC were ranging 232 from 51-73% and 37-49% of OC, respectively. Sen et al. (2018) reported the similar OC (7.7±0.7 µg m<sup>-3</sup>) and EC 233  $(3.7\pm0.6 \ \mu g \ m^{-3})$  concentrations in PM<sub>10</sub> at Darjeeling and high OC  $(10.3\pm5.6 \ \mu g \ m^{-3})$  and EC  $(5.4\pm2.9 \ \mu g \ m^{-3})$ 234 values in PM<sub>10</sub> at Kullu-Mohal (north-western Himalayas) during winter. Whereas, Rai et al. (2020a) observed the low OC (3.7±1.3  $\mu$ g m<sup>-3</sup>) and EC (1.3±0.6  $\mu$ g m<sup>-3</sup>) in PM<sub>10</sub> at Darjeeling. Sharma et al. (2020a) also found the 235 236 similar OC and EC concentrations in  $PM_{10}$  over Nainital (central Himalayas) during post-monsoon (OC: 4.7±1.1 µg 237  $m^{-3}$  and EC:  $1.1\pm0.5 \ \mu g \ m^{-3}$ ) and winter (OC:  $3.2\pm1.1 \ \mu g \ m^{-3}$  and EC:  $1.4\pm0.6 \ \mu g \ m^{-3}$ ) seasons.

Figure 2 shows the monthly average OC/EC and WSOC/OC ratios of  $PM_{2.5}$  and  $PM_{10}$  at Darjeeling, whereas seasonal relationship between OC and EC, WSOC and OC of  $PM_{2.5}$  and  $PM_{10}$  are depicted in Figs. S4-S5 (see the supplementary information). The seasonal average OC/EC of  $PM_{2.5}$  was 1.87 (range: 1.40 - 2.81), 1.92 (range: 1.22 -2.82), 1.85 (range: 1.30 - 2.57) and 1.94 (range: 1.05 -2.87) during winter, summer, monsoon and post-monsoon, respectively. The seasonal average WSOC/OC ratio of  $PM_{2.5}$  was computed as 0.50, 0.52, 0.53 and 0.54 during winter, summer, monsoon and post-monsoon, respectively. Similarly, the seasonal average OC/EC of  $PM_{10}$  was 2.04, 1.74, 3.37 and 3.01 during winter, summer, monsoon and post-monsoon, respectively, whereas, the seasonal 245 average WSOC/OC of PM10 was 0.71, 0.61, 0.65 and 0.79 during winter, summer, monsoon and post-monsoon, 246 respectively. Hegde et al. (2016) also observed a similar (0.51±0.06 for TSP) WSOC/OC value in Nainital during 247 winter. Due to poor solubility of organics emitted from the combustion of liquid fossil fuels the WSOC/OC values 248 for vehicular emissions are low as compared to biomass burning. In this study, significant positive correlation 249 between OC vs. EC (for PM<sub>2.5</sub>:  $R^2 = 0.88$ , 0.86, 0.63 and 0.80 at p < 0.05; for PM<sub>10</sub>:  $R^2 = 0.77$ , 0.85, 0.09 and 0.74 at 250 p <0.05 during winter, summer, monsoon and post-monsoon seasons, respectively) of PM<sub>2.5</sub> and PM<sub>10</sub> has been 251 observed during all the seasons (except monsoon season for  $PM_{10}$ ) (Figs. S4-S5; in supplementary information), 252 which is indicative of their common sources (Rengarajan et al. 2007; Ram and Sarin 2011). The scatter plots between WSOC and OC (for PM<sub>2.5</sub>:  $R^2 = 0.63$ , 0.73, 0.62 and 0.83; and for PM<sub>10</sub>:  $R^2 = 0.69$ , 0.91, 0.17 and 0.87 253 254 during winter, summer, monsoon and post-monsoon seasons, respectively) of PM<sub>2.5</sub> and PM<sub>10</sub> (except monsoon 255 season for PM<sub>10</sub>) shows significant positive correlation suggesting that both OC and EC are obtained from the same 256 primary emission source or by similar secondary processes (Figs. S4-S5; in supplementary information). The non-257 significant positive correlation of K (a tracer of biomass) with Ca, Mg, WSOC of PM<sub>2.5</sub> and significant positive 258 correlation of K with Ca, Mg, WSOC of  $PM_{10}$  during all the seasons (except post-monsoon season for  $PM_{2.5}$ ) 259 demonstrate the abundance of soil/road dust contributed by soluble organic sources to PM<sub>2.5</sub> and PM<sub>10</sub> at study site 260 (Tables: S1-S8; in supplementary information). I may also be considered that the soil suspension, fuel combustion 261 (Urban et al. 2012), and formation of secondary water soluble organic aerosols (Lim et al. 2010) might be the some 262 other sources of WSOC in the sampling site of Darjeeling.

- 263
- 264 3.1.3. Major and trace elements in PM<sub>2.5</sub> and PM<sub>10</sub>

265 During the sampling 17 common elements (Al, Fe, Ti, Cu, Zn, Mn, Cr, Ni, Mo, Cl, P, S, K, Zr, Pb, Na, Mg, Ca, 266 and B) were extracted in PM2.5 and PM10 (Cu and Pb in PM10 but traced in few PM2.5 samples also) using X-RF 267 technique of sampling site (Table 1). In PM<sub>2.5</sub> samples, Na has observed highest annual average concentration 268  $(1.014 \pm 0.392 \ \mu g \ m^{-3})$ , followed by Ca  $(0.918 \pm 0.813 \ \mu g \ m^{-3})$ , S  $(0.667 \pm 0.404 \ \mu g \ m^{-3})$ , Fe  $(0.502 \pm 0.195 \ \mu g \ m^{-3})$ , 269 K (0.388  $\pm$  0.224 µg m<sup>-3</sup>), and so on with higher loading in summer season (Table 1). Whereas in case of PM<sub>10</sub>, S 270 has recorded highest annual average concentration  $(1.260 \pm 0.744 \ \mu g \ m^{-3})$  followed by Ca  $(0.985 \pm 0.299 \ \mu g \ m^{-3})$ , 271 Al  $(0.922 \pm 0.446 \ \mu g \ m^{-3})$ , K  $(0.650 \pm 0.376 \ \mu g \ m^{-3})$ , Fe  $(0.635 \pm 0.266 \ \mu g \ m^{-3})$ , Na  $(0.607 \pm 0.396 \ \mu g \ m^{-3})$ , and so on. The annual average concentrations of elements in  $PM_{2.5}$  and  $PM_{10}$  are 5.71 ± 3.67 µg m<sup>-3</sup> (accounted for 15% of 272

273  $PM_{2.5}$ ) and 6.44 ± 3.58 µg m<sup>-3</sup> (accounted for 12% of  $PM_{10}$ ), respectively. The monthly average concentrations of 274 elements of  $PM_{2.5}$  and  $PM_{10}$  are shown in Fig. S6 (see the supplementary information). Higher average 275 concentrations of elements are recorded during summer (21% of PM<sub>2.5</sub> and 12% of PM<sub>10</sub>) followed by winter 276 (16.5% of PM<sub>2.5</sub> and 11.7 of PM<sub>10</sub>), post-monsoon (11.6% of PM<sub>2.5</sub> and 11% of PM<sub>10</sub>) and monsoon (9% of PM<sub>2.5</sub> 277 and 13% of PM<sub>10</sub>) seasons (Fig. 4). During winter 2013 land campaign, Sharma et al. (2014b) found the Na, Mg, Ca, 278 Al, P, S, Si, Cl, K, Ti, Sr, Zr, Pb, Sb, Ag, Cs, Hg, Mn, Fe and Zn in PM<sub>2.5</sub> (which had accounted for ~27% of PM<sub>2.5</sub>) 279 over the north-western Himalaya. Sharma et al. (2020a) also reported the elements contribution to  $PM_{10}$  as 16% and 280 13% during post-monsoon and winter seasons, respectively at central Himalaya (Nainital). The similar contribution 281 of elements to the PM<sub>10</sub> over the Himalayan region and other high altitude regions are discussed in our earlier 282 publication (Sharma et al. 2020b).

Figure 5 represents the seasonal enrichment factors (EFs) of the elements (Al, Fe, Ti, Cu, Zn, Mn, Cr, Ni, Mo, Cl, P, S, K, Zr, Pb, Na, Mg, Ca, and B) available in  $PM_{2.5}$  and  $PM_{10}$  samples. Al, Fe, Ti, K, Mg, and Ca in both PM<sub>2.5</sub> and PM<sub>10</sub> have recorded low EFs (< 5) for all the seasons, which indicate that, elements mostly arrived from crustal/soil sources. The element like Cu, Zn, Ni, Pb, Cr, Mo and B have higher EFs (>10) in both PM<sub>2.5</sub> and PM<sub>10</sub> and therefore are likely of anthropogenic origin. The higher the EF of Cr, Ni, Pb and Zn of PM were also attributed to industrial emission (IE) sources. Generally Cu, Mn, Zn, Ni, Cd, Fe, Mo, S and Cr use as a marker for IE in India (Shridhar et al. 2010).

290 The annual and seasonal statistical summary of elements recorded in PM<sub>2.5</sub> and PM<sub>10</sub> samples is tabulated in 291 Table 1. The possible sources of trace metals present in a fine and coarse fraction of particulates can be primarily 292 crustal/mineral dust. In this study, Al, is significantly positive correlated with Fe, Ca, Mg and Ti (as well as with 293  $PM_{10}$ ) and the average Fe/Al ratio is 0.69 (winter: 0.76; summer: 0.74; monsoon: 0.45 and post-monsoon: 0.79), 294 which indicates the dominant source of mineral dust. Similarly, Ca/Al ratio (1.07) indicates that PM over the eastern 295 Himalayan region is rich in Ca mineral as compared to average continental crust. Similar results were also observed 296 in case of elements extracted in  $PM_{25}$  over the eastern Himalaya. Sarin et al. (1979) had reported that the Fe/Al ratio 297 in north Indian plains ranged from 0.55 to 0.63. Kumar and Sarin (2009) reported Fe/Al ratio as 0.59 for PM<sub>2.5-10</sub> at 298 a remote high altitude sampling site (Manora Peak) of western India. McLennan, (2001) recorded the average Ca/Al 299 ratio as 1.07 in PM<sub>10</sub> whereas, the corresponding ratio in the upper continental crust is 0.38. Sharma et al. (2020a) 300 also reported Ca/Al ratio as 1.52 and 1.19 in PM<sub>10</sub> during post-monsoon and winter seasons, respectively at central Himalayas (Nainital), whereas Kumar and Sarin (2009) had recorded the Ca/Al ratio as 0.73 in  $PM_{2.5}$  and 1.74 in PM<sub>10</sub> at Manora Peak, a high altitude site of the western India. The significant positive correlation of Al with Fe, Ca, Mg and Ti of coarse and fine (except few in  $PM_{2.5}$ ) fractions of PM during all the seasons are also indicated the abundance of mineral dust at the sampling site of Darjeeling (Table S1-S8; in supplementary information).

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#### **306 3.2.** Possible sources and source regions

#### 307 3.2.1 Possible sources of PM<sub>2.5</sub> and PM<sub>10</sub>

PCA has been performed with chemical species of  $20PM_{2.5}$  and  $22PM_{10}$  (OC, EC, WSOC, Na, Ca, Mg, Al, Fe, Ti, K, Cl, P, S, Cr, Ni, Cu, Zn, Mn, Mo, Zr, Pb and B) to identify the factor loading to  $PM_{2.5}$  and  $PM_{10}$ . The factor profiles (a yearlong data) of the possible sources of  $PM_{2.5}$  and  $PM_{10}$  extracted by PCA are summarized in Table 2. On the basis of the factor loading PCA resolved the four common sources [biomass burning + fossil fuel combustion (BB+FFC), crustal/soil dust (SD), vehicular emissions (VE) and industrial emissions (IE)] of  $PM_{2.5}$  and  $PM_{10}$  at Darjeeling.

314 Factor 1: The first factor of PM<sub>2.5</sub> represents the BB+FFC, characterized (38.3% of the variance) by highly 315 loaded with OC, EC, WSOC, K, Cl and S. K<sup>+</sup> and Levoglucoson are considered BB (cow dung, crop residue, fuel 316 wood, and wildfires, etc.) marker, whereas presence of Cl in the factor reveals the wood and coal burning (Pant and 317 Harrison, 2012). WSOC/OC and OC/EC mass ratios also evidence the BB+FFC as a one of the sources of PM<sub>2.5</sub> at 318 the observational site of Darjeeling (Sharma et al. 2020b; Chatterjee et al. 2021). The first factor of  $PM_{10}$  represented 319 by high loading (36.7% of the variance) of crustal elements like, Al, Ti, Fe, Ca, Mg, K and Na which inferred the 320 source as crustal/soil/road dust (Begam et al. 2011; Sharma et al. 2014a; Jain et al. 2020b). The abundance of these 321 elements at the study site as crustal origin is also confirmed by EFs (Fig. 5) as well as positive correlations of Al 322 with Ca, Mg and Ti (Table S1-S8).

Factor 2: This factor of  $PM_{2.5}$  was resolved as crustal/soil dust by high loading of Al, Ti, Fe, Ca, Mg, K and Na (17.6% of the variance of factor loading) (Table 2). The EFs of these elements are also suggesting the crustal origin the elements at sampling site (Fig. 5) as well as the positive correlations of Al with Ca, Mg and Ti. Whereas second factor of  $PM_{10}$  represented by BB+FFC with high loadings of OC, EC, WSOC, K, Cl and S (Begam et al. 2011; Pant and Harrison 2012). WSOC/OC and OC/EC ratios are also suggesting the influence of BB+FFC as a source of  $PM_{10}$ 

at Darjeeling (Sharma et al. 2020b; Chatterjee et al. 2021).

329 Factor 3: The third factor of both  $PM_{2.5}$  and  $PM_{10}$  constitutes the vehicular emissions (VE) with the dominant 330 presence of EC, OC, Zn, Mn, Zr and B, indicates the emission derived from road side vehicles (Pant and Harrison, 331 2012; Jain et al. 2020b). Since, EC and OC are majorily emitted from the combustion sources, so these components 332 are considered as important tracers for VE globally (Yin et al. 2010; Begam et al. 2011). Zn and Mn are used as 333 marker of brake and tire wear, two stroke engine emissions (Zn as fuel additive), heavy duty diesel truck emission 334 (Mn as fuel additive) (Kothai et al. 2008; Sharma et al. 2014a). VE is inferred to be one of the major sources of 335 aerosols (Sharma et al. 2020b) at in urban sites of Himalayan region may due to the great influence of tourism 336 activities (Gajananda et al. 2005; Chatterjee et al. 2021).

Factor 4: The fourth source of both  $PM_{2.5}$  and  $PM_{10}$  is characterized as industrial emissions (IE) considering to the higher loading of Cu, Zn, Ni, Cr and Mo in aerosol samples (Table 2). These metals (Cu, Zn, Ni, Cr and Mo) might be originated from the small to medial scale industries, metal processing industries and industrial effluents (Gupta et al. 2007; Jain et al. 2019). Sharma et al. (2020b) also reported the solid waste + IE a source of  $PM_{10}$  in Darjeeling with these marker elements.

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#### 343 3.2.2 Air mass backward trajectory

To investigate the transport pathway of fine (PM<sub>2.5</sub>) and coarse fractions (PM<sub>10</sub>) of aerosols at the observational site of Darjeeling, 5-days air-mass back trajectory was computed using the HYSPLIT model (Draxeler and Ralph. 2003) at an altitude of 100m, 500 m and 100m AGL for all the sampling day (Fig.6). In the present case, air parcels approaching to sampling site of Darjeeling is mainly from Bhutan, Nepal, China, Arunachal Pradesh, IGP (Punjab, Haryana, Uttar Pradesh, Bihar and West Bengal) and surrounding regions during all the seasons. During monsoon season, air mass also approaching to sampling site from Bay of Bengal. Chatterjee et al. (2021) reported the similar air parcels towards the Darjeeling during summer and winter season (Sharma et al. 2020b).

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#### **352 4.** Conclusions

In this study seasonal variation in carbonaceous aerosols and elements in fine and coarse fractions of aerosol ( $PM_{2.5}$  and  $PM_{10}$ ) were estimated to explore the prominent sources of  $PM_{2.5}$  and  $PM_{10}$  in the high altitude of eastern Himalaya during August 2018 - July 2019. The average  $PM_{2.5}$  and  $PM_{10}$  concentrations were recorded as  $37\pm12 \ \mu g$ m<sup>-3</sup> and  $58\pm18 \ \mu g$  m<sup>-3</sup>, respectively during the study. The annual total carbonaceous aerosols in  $PM_{2.5}$  and  $PM_{10}$  were 357 accounted for 20.6% of PM<sub>2.5</sub> and 18.6% of PM<sub>10</sub>, respectively (along with season variation). The concentrations of 358 elements present in PM<sub>2.5</sub> and PM<sub>10</sub> were accounted for 15% and 12%, respectively. During all the seasons, 359 significant positive linear relationship between OC and EC; and OC and WSOC (as well as OC/EC and WSOC/OC) 360 indicate the common sources (BB+FFC) of both PM<sub>2.5</sub> and PM<sub>10</sub>. EFs analysis the elements present in PM indicates 361 the abundance of mineral dust at the eastern Himalaya. PCA resolved the four common sources [(BB+FFC), crustal/SD, (VE) and (IE)] of PM<sub>2.5</sub> and PM<sub>10</sub> at Darjeeling. 5 days HYSPLIT back trajectory air parcels indicate 362 363 that the pollutants approaching to Darjeeling are mainly from Bhutan, Nepal, China, Arunachal Pradesh and IGP 364 region (Punjab, Haryana, Uttar Pradesh, Bihar and West Bengal) during all the seasons as well as the Bay of Bengal 365 (mainly in monsoon season).

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carried out by SKS, AC, TKM and NV. All the authors read and approved the final manuscript.

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383 Data availability: The datasets developed during the current study are available from the corresponding author on
 384 reasonable request.

- **386** Compliance with ethical standards
- 387 **Competing interests**: The authors declare that they have no conflict of interest.
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- 391
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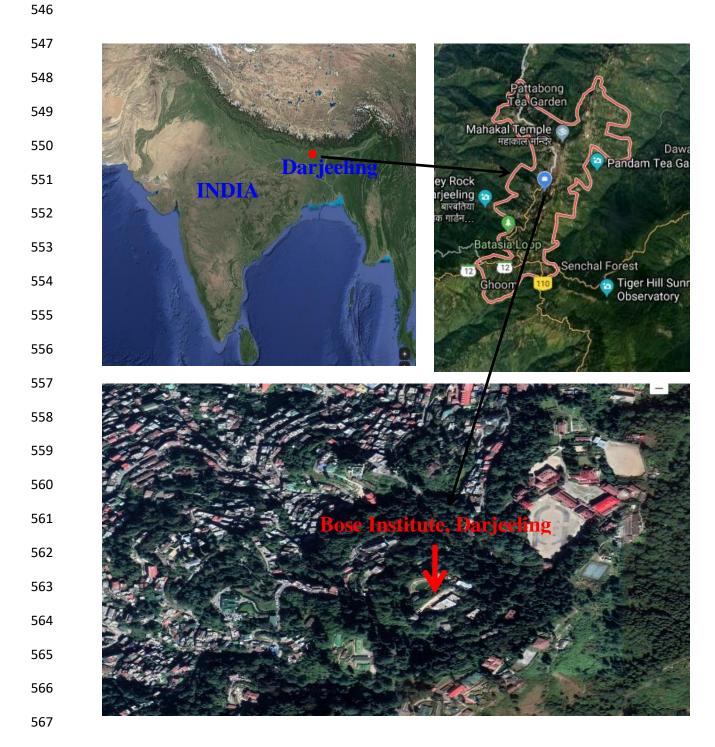
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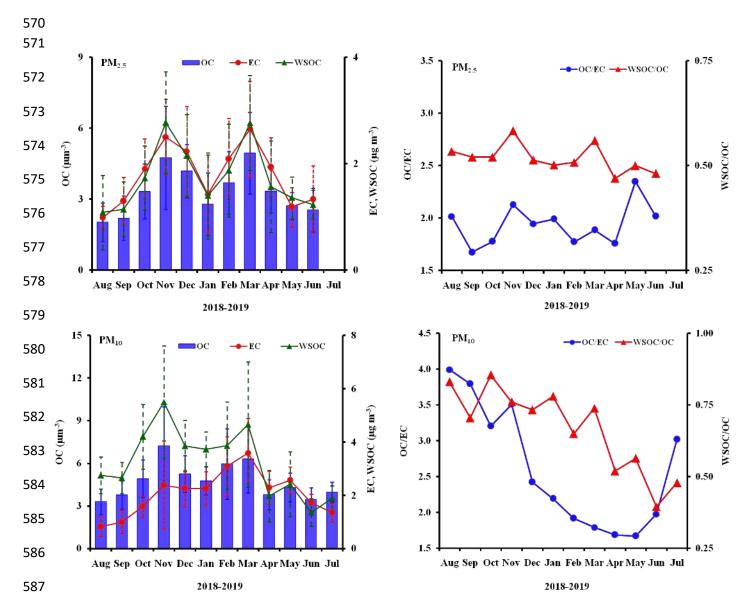
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568 Fig. 1 Map of the observational site (Source: Google maps).



588 Fig. 2: Monthly average concentrations of OC, EC and WSOC and their mass ratios of  $PM_{2.5}$  and  $PM_{10}$  at 589 Darjeeling.

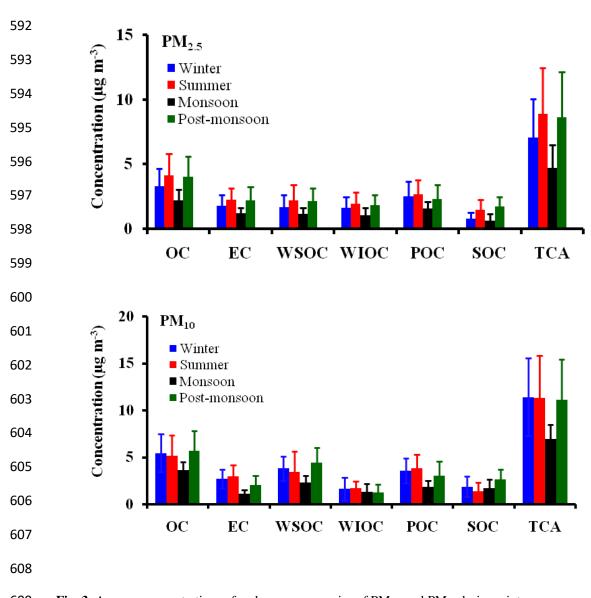
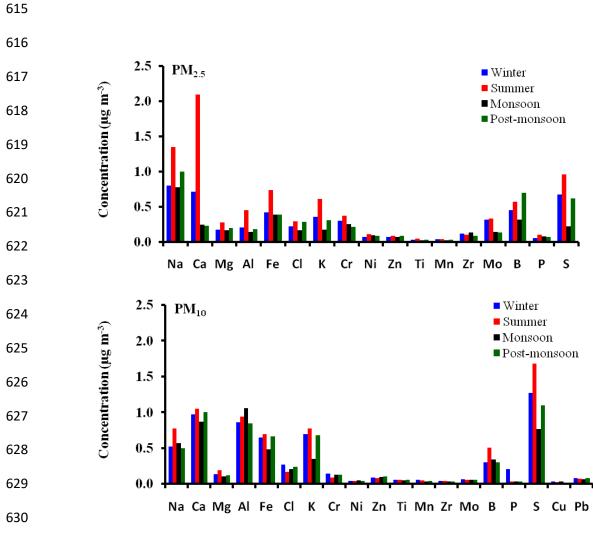
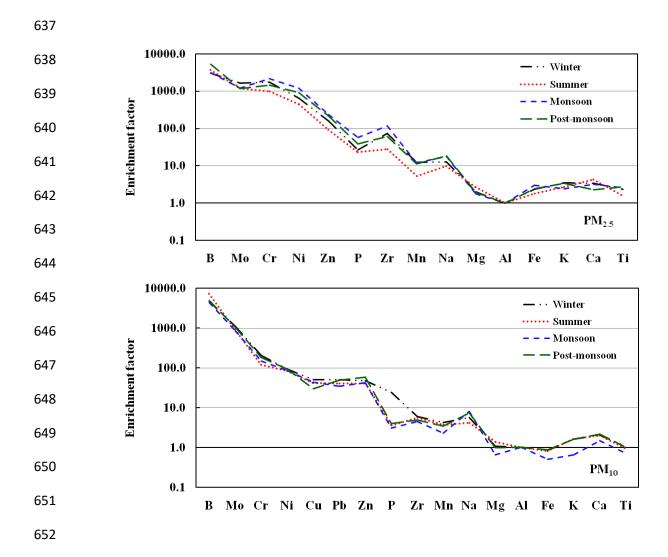


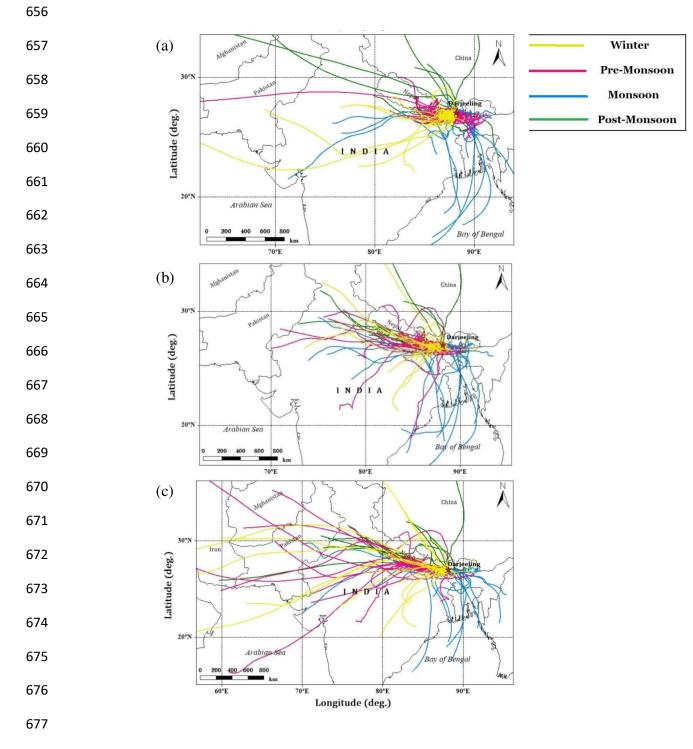
Fig. 3: Average concentrations of carbonaceous species of PM<sub>2.5</sub> and PM<sub>10</sub> during winter, summer, monsoon and
 post-monsoon seasons at Darjeeling.



**Fig. 4**: Average concentrations of major and trace elements in  $PM_{2.5}$  and  $PM_{10}$  during winter, summer, monsoon and post-monsoon seasons at Darjeeling.



**Fig. 5**: Enrichment factor of trace elements of  $PM_{2.5}$  and  $PM_{10}$  during winter, summer, monsoon and post-monsoon seasons at Darjeeling.



678 Fig. 6: Five day air mass backward trajectory at sampling site at (a) 100m, (b) 500m, and (c) 1000 m AGL during679 winter, summer (pre-monsoon), monsoon and post-monsoon seasons.

Species	Annual		Winter		Summer		Monsoon		Post monsoon	
	PM <sub>2.5</sub>	PM10	PM <sub>2.5</sub>	PM <sub>10</sub>	PM <sub>2.5</sub>	PM10	PM <sub>2.5</sub>	PM10	PM <sub>2.5</sub>	PM10
PM	37±12	56±18	31±9	51±17	41±14	64±20	38±8	52±11	40±11.4	55±17
OC	3.56±1.59	5.06±2.03	3.29±1.37	5.43±2.02	4.04±1.67	5.18±2.14	2.21±0.86	3.62±0.87	4.07±1.55	5.69±2.09
EC	1.93±0.93	2.34±1.18	1.80±0.81	2.72±1.01	2.23±0.92	3.01±1.15	1.21±0.43	1.25±0.38	2.19±1.06	2.05±1.02
WSOC	1.88±1.05	3.55±1.75	1.66±0.95	3.80±1.28	2.15±1.19	3.43±2.19	1.14±0.51	2.31±0.77	2.17±0.98	4.44±1.55
WIOC	1.69±0.85	1.51±0.92	1.63±0.85	1.63±1.21	1.94±0.91	1.75±0.68	1.07±0.55	1.31±0.90	1.86±0.78	1.25±0.85
POC	2.35±1.06	3.18±1.31	2.52±1.13	3.56±1.31	2.68±1.10	3.82±1.46	1.57±0.56	1.87±1.31	2.30±1.12	3.07±1.52
SOC	1.19±0.57	2.05±0.98	0.77±0.51	1.87±1.08	1.48±0.76	1.37±0.98	0.64±0.54	1.75±0.91	1.74±0.74	2.62±1.11
TCA	7.63±3.43	10.43±4.25	7.06±2.97	11.41±4.14	8.89±3.54	11.29±4.51	4.74±1.73	6.94±1.55	8.65±3.47	11.15±4.25
Na	1.014±0.392	0.607±0.395	0.801±0.247	0.525±0.343	1.351±0.437	0.776±0.484	0.780±0.272	0.573±0.389	1.002±0.243	0.501±0.239
Mg	$0.209 \pm 0.075$	0.244±0.108	0.172±0.044	0.237±0.066	0.279±0.073	0.191±0.160	$0.168 \pm 0.052$	0.103±0.080	0.195±0.062	0.123±0.041
Ca	0.918±0.813	0.985±0.299	0.716±0.504	0.972±0.178	2.097±0.706	1.051±0.442	0.249±0.213	0.871±0.232	0.229±0.213	1.007±0.182
Cl	$0.250 \pm 0.147$	0.223±0.140	0.221±0.127	0.274±0.126	0.291±0.149	0.171±0.054	0.170±0.050	0.206±0.092	0.283±0.180	0.244±0.118
Р	$0.078 \pm 0.037$	0.077±0.034	0.054±0.029	0.212±0.179	0.105±0.026	0.033±0.024	0.081±0.034	0.032±0.013	0.070±0.039	0.033±0.009
S	$0.667 \pm 0.404$	1.260±0.744	0.675±0.382	1.277±0.768	0.956±0.370	1.680±0.744	0.223±0.181	0.771±0.495	0.620±0.284	1.098±0.629
К	0.388±0.224	$0.650 \pm 0.376$	0.360±0.110	0.699±0.295	0.608±0.226	0.774±0.435	0.174±0.050	0.353±0.242	0.307±0.171	0.681±0.346
Al	0.261±0.164	0.922±0.446	0.203±0.104	0.860±0.291	0.454±0.143	0.943±0.661	0.143±0.065	1.061±0.386	0.179±0.068	0.845±0.215
Fe	0.502±0.195	0.635±0.266	0.423±0.141	0.652±0.189	0.737±0.153	0.694±0.381	0.390±0.087	0.482±0.188	0.391±0.067	0.663±0.142
Ti	0.035±0.015	0.057±0.022	0.031±0.009	0.061±0.017	0.047±0.013	0.058±0.031	0.024±0.008	0.049±0.016	0.032±0.018	0.056±0.013
Cr	0.291±0.143	0.122±0.114	0.299±0.102	0.146±0.125	0.372±0.199	0.093±0.014	$0.254 \pm 0.081$	0.132±0.032	0.215±0.074	0.122±0.114
Ni	0.092±0.025	$0.084 \pm 0.008$	0.072±0.012	0.044±0.008	0.110±0.027	0.045±0.006	0.094±0.014	$0.049 \pm 0.007$	0.090±0.021	0.043±0.011
Cu	-	0.028±0.014	-	0.032±0.008	-	0.029±0.005	-	0.034±0.026	-	$0.018 \pm 0.004$
Zn	$0.079 \pm 0.032$	0.092±0.033	0.069±0.035	0.088±0.032	0.088±0.035	0.081±0.027	0.072±0.016	0.097±0.038	0.084±0.031	0.105±0.034
Mn	0.033±0.012	$0.048 \pm 0.018$	0.037±0.010	0.055±0.013	0.037±0.010	0.053±0.016	0.033±0.012	0.036±0.015	0.031±0.015	0.044±0.021
Мо	0.245±0.169	0.060±0.013	0.318±0.236	0.063±0.010	0.337±0.224	0.058±0.013	0.144±0.034	0.061±0.014	0.133±0.028	0.058±0.015
Zr	0.107±0.094	0.061±0.013	0.117±0.099	0.041±0.009	0.102±0.039	0.042±0.012	0.132±0.112	0.038±0.016	0.086±0.026	0.033±0.011
Pb	-	$0.076 \pm 0.064$	-	0.084±0.010	-	0.072±0.069	-	$0.069 \pm 0.059$	-	0.079±0.067
В	0.533±0.389	0.376±0.220	0.455±0.179	0.302±0.174	0.575±0.324	0.510±0.262	0.321±0.096	0.343±0.203	0.702±0.472	0.307±0.132

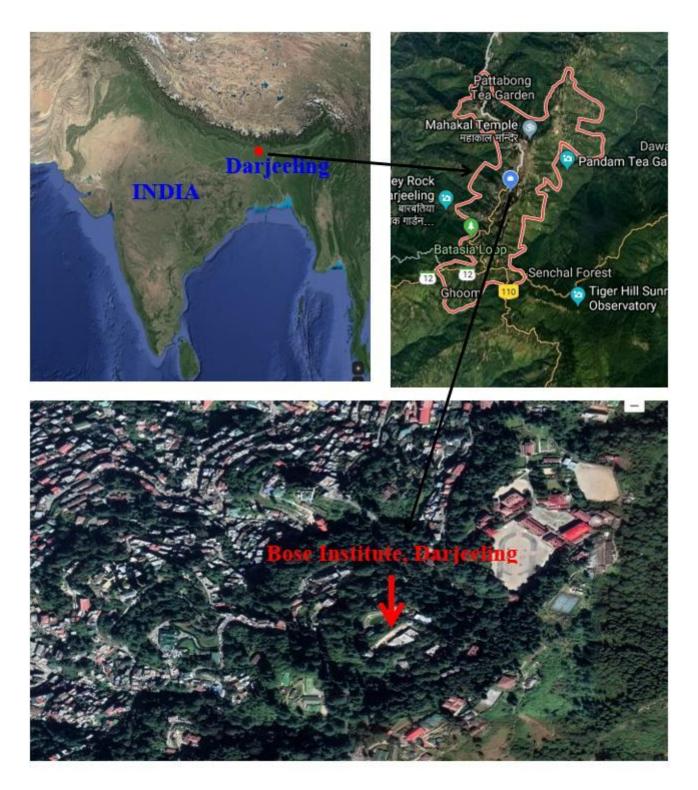
Table 1. Annual average and seasonal concentrations of  $PM_{2.5}$  and  $PM_{10}$  and their chemical species ( $\mu g m^{-3}$ ) in Darjeeling.

 $\pm$  Standard deviation (at  $1\sigma$ )

Species		PN	M <sub>2.5</sub>		PM <sub>10</sub>			
	Factor 1	Factor 2	Factor 3	Factor 4	Factor 1	Factor 2	Factor 3	Factor 4
OC	0.942	0.159	0.482	0.013	0.262	0.912	0.416	0.046
WSOC	0.872	0.095	0.029	-	0.197	0.856	0.174	0.092
EC	0.563	0.138	0.743	0.126	0.182	0.846	0.586	0.186
Na	0.139	0.735	0.031	-	0.851	0.063	0.162	0.142
Mg	0.056	0.832	0.048	0.061	0.909	0.188	0.083	0.098
Ca	0.035	0.877	0.229	0.233	0.935	0.196	0.049	-
Al	0.219	0.898	0.198	0.191	0.860	0.126	0.074	0.368
Fe	0.132	0.852	0.031	0.282	0.882	0.294	-	0.064
Ti	0.383	0.497	0.052	-	0.866	0.257	0.067	0.032
Κ	0.608	0.588	0.316	0.097	0.628	0.696	0.010	0.055
Cl	0.636	0.137	-	0.394	0.072	0.853	0.389	0.095
Р	0.075	0.594	0.160	0.283	0.107	-	0.964	0.068
S	0.574	0.348	0.237	0.361	0.468	0.623	-	0.274
Cr	0.061	0.138	0.235	0.760	0.065	-	0.468	0.674
Ni	-	0.029	0.740	0.396	0.185	-	0.235	0.704
Cu	-	-	-	-	0.109	-	0.425	0.838
Zn	0.281	0.102	0.754	0.467	0.216	0.013	0.530	0.619
Mn	-	0.120	0.436	0.701	0.197	0.056	0.140	0.364
Мо	0.101	0.087	0.891	0.069	-	-	0.037	0.876
Zr	-	-	0.266	0.701	-	-	0.010	0.246
Pb	-	-	-	-	-	0.063	0.273	0.273
В	0.106	-	0.730	0.016	0.523	0.152	0.428	0.428
% Variance	38.5	17.6	11.6	8.7	36.7	16.2	12.4	10.8
Cumulative variance (%)	38.3	55.9	67.5	76.2	36.7	52.9	65.3	76.1
Sources	BB+FFC	Crustal/SD	VE	IE	Crustal/SD	BB+FFC	VE	IE

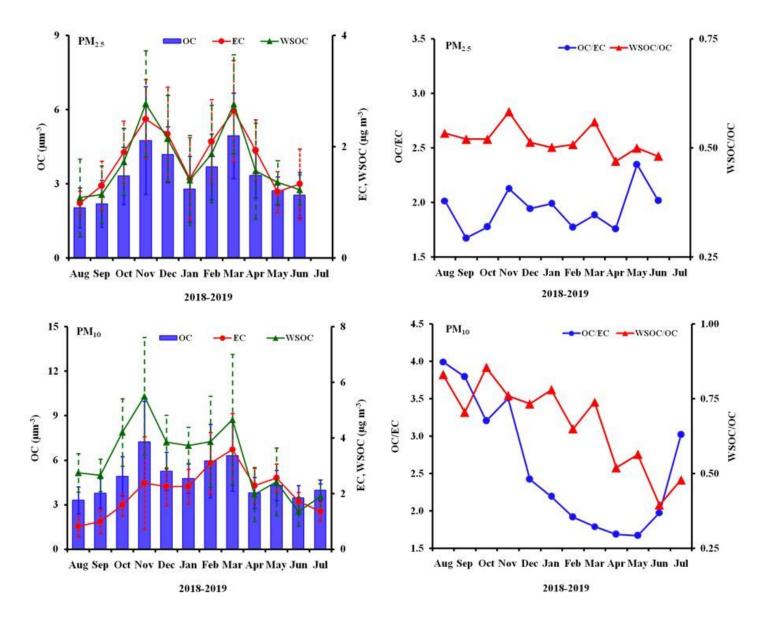
Table 2. PCA factor loading of PM<sub>2.5</sub> and PM<sub>10</sub> samples in Darjeeling, India.

Extraction method: principal component analysis; rotation method: Varimax with Kaiser Normalization; Eigenvalue >1.00; factor loading  $\ge 0.40$ . BB: biomass burning; FFC: fossil fuel combustion; SD: soil dust; VE: vehicular emissions; IE: industrial emission.

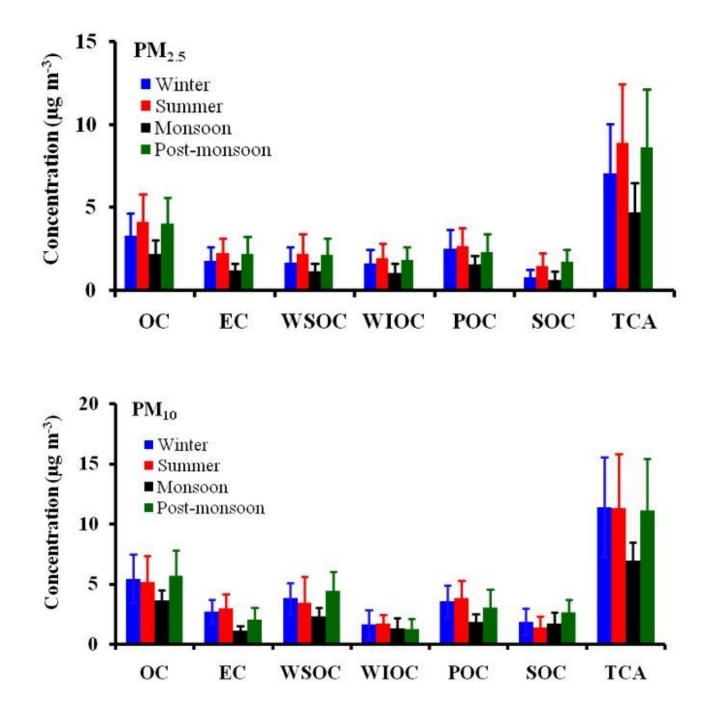


## Figure 1

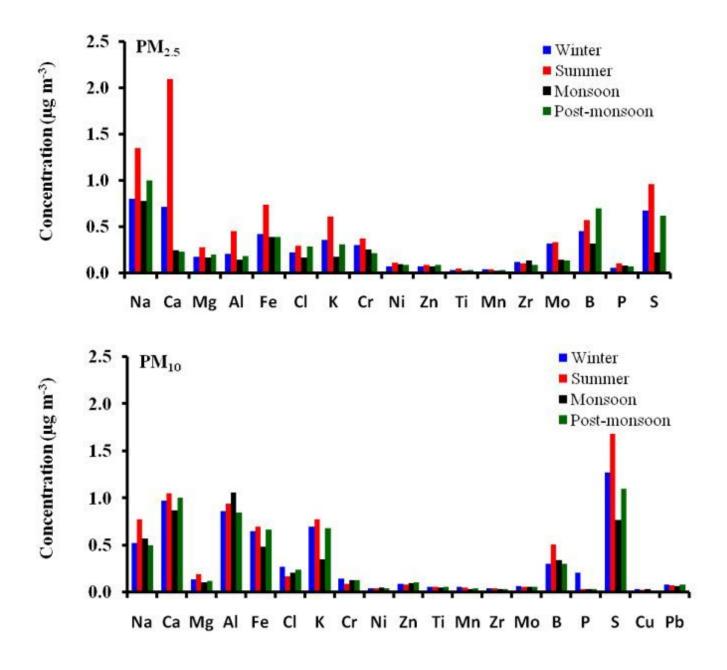
Map of the observational site (Source: Google maps).



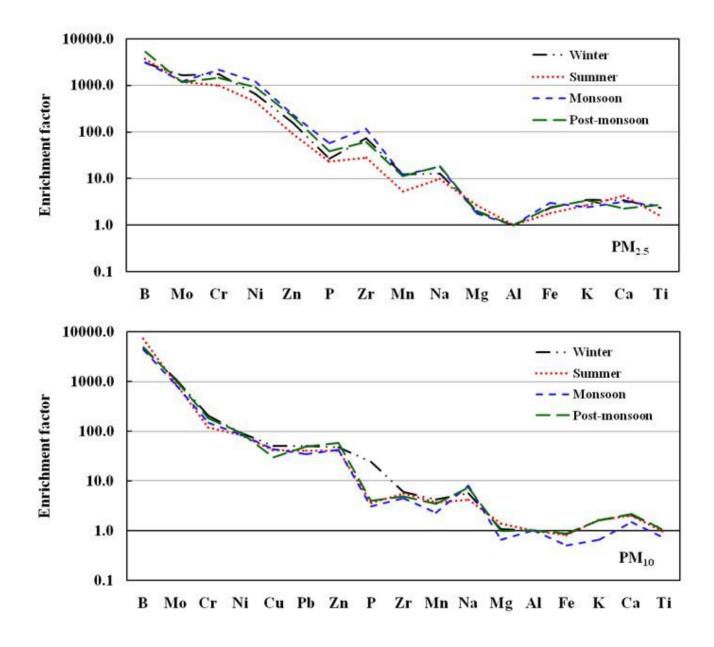
Monthly average concentrations of OC, EC and WSOC and their mass ratios of PM2.5 and PM10 at Darjeeling.



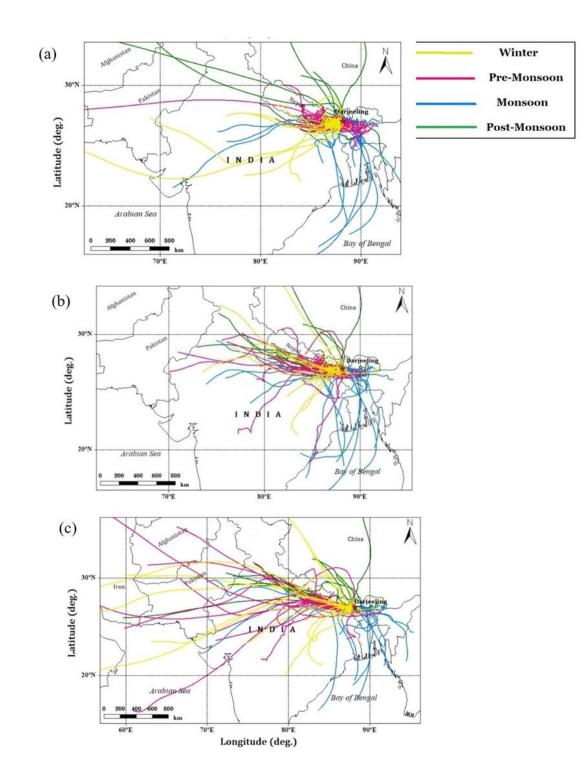
Average concentrations of carbonaceous species of PM2.5 and PM10 during winter, summer, monsoon and post-monsoon seasons at Darjeeling.



Average concentrations of major and trace elements in PM2.5 and PM10 during winter, summer, monsoon and post-monsoon seasons at Darjeeling



Enrichment factor of trace elements of PM2.5 and PM10 during winter, summer, monsoon and postmonsoon seasons at Darjeeling.



Five day air mass backward trajectory at sampling site at (a) 100m, (b) 500m, and (c) 1000 m AGL during winter, summer (pre-monsoon), monsoon and post-monsoon seasons.

## **Supplementary Files**

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