

Supporting Information

The Skraup and Doebner-von Miller Quinoline Synthesis Revisited — Reversal of the Regiochemistry for γ -Aryl- β , γ -Unsaturated α -Ketoesters

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1. Experimental Procedures

1.1 Synthesis of γ -aryl- β , γ -unsaturated α -ketoesters **2**.¹⁻²

To a solution of pyruvic acid (20 mL, 0.286 mol), substituted or unsubstituted benzaldehyde (0.286 mol) in 15 mL of methanol stirring in an ice bath, a solution of potassium hydrate (24 g, 0.429 mol) in 75 mL of methanol was added. The first 50 mL of the base solution was added slowly and the reaction temperature was kept below 25 °C. The ice bath was then removed and the rest of the base solution was added quickly. Yellow precipitate was formed at once. The reaction temperature was kept at 30 °C for 1 h and then at zero overnight. The yellow crystals were filtered off and washed twice with cold methanol and once with ether. The yellow crystals were air dried to afford the potassium salt.¹

24 mL of acetyl chloride was added to 140 mL of methanol in 250 mL 3-necked round bottom at zero to generate hydrochloric acid. The potassium salt (0.1 mol) was added and the mixture stirred for 30 min before the ice bath was removed. After 2 h the mixture was refluxed overnight. The reaction mixture was evaporated and the yellow solid was extracted with 50 mL of water and two times with 50 mL of dichloromethane. The combined organic phases were washed with 50 mL of saturated sodium carbonate and then 50 mL of water. The organic phase was dried with anhydrous potassium sulfate and evaporated. The yellow crystals of pure γ -aryl- β , γ -unsaturated α -ketoesters **2** were obtained via recrystallization from methanol or ethanol.²

(3E)-2-Oxo-4-phenyl-but-3-enoic acid methyl ester (2a): 37.0% yield in two steps; yellow crystal; mp = 68-70 °C (literature.² 70-71 °C); ¹H NMR (300 MHz, CDCl₃) δ 7.84 (d, J = 16.1 Hz, 1H), 7.60 (dd, J = 9.5, 1.9 Hz, 2H), 7.43-7.31 (m, 4H), 3.91 (s, 3H); ¹³C NMR (75 MHz, CDCl₃) δ 182.4, 162.5, 148.6, 134.0, 131.7, 129.1, 129.1, 120.5, 53.0.

(3E)-4-(4-Methoxyphenyl)-2-oxobut-3-enoic acid methyl ester (2b): 44.0% yield in two steps; yellow crystal; mp = 99-100 °C (literature.³ 109 °C); ¹H NMR (300 MHz, CDCl₃) δ 7.80 (d, J = 16.0 Hz, 1H), 7.56 (d, J = 8.7 Hz, 2H), 7.21 (d, J = 16.0 Hz, 1H), 6.90 (d, J = 8.7 Hz, 2H), 3.89 (s, 3H), 3.82 (s, 3H); ¹³C NMR (75 MHz, CDCl₃) δ 182.2, 162.8, 162.7, 148.5, 131.1, 126.8, 118.0, 114.6, 55.5, 53.0.

(3E)-4-(4-Methylphenyl)-2-oxobut-3-enoic acid methyl ester (2c): 41.0% yield in two steps; yellow crystal; mp = 81-83 °C (literature.⁴ 81 °C); ¹H NMR (300 MHz, CDCl₃) δ 7.76 (d, J = 16.1 Hz, 1H), 7.44 (d, J = 8.1 Hz, 2H), 7.24 (d, J = 16.1 Hz, 1H), 7.14 (d, J = 8.1 Hz, 2H), 3.86 (s, 3H), 2.31 (s, 3H); ¹³C NMR (75 MHz, CDCl₃) δ 182.4, 162.6, 148.7, 142.5, 131.2, 129.8, 129.1, 119.4, 53.0, 21.6.

(3E)-4-(4-Chlorophenyl)-2-oxobut-3-enoic acid ethyl ester (2d): 31.0% yield in two steps; yellow crystal; mp = 76-77 °C (literature.⁵ 77.5-78.5 °C); ¹H NMR (300 MHz,

CDCl_3) δ 7.77 (d, J = 16.1 Hz, 1H), 7.54 (d, J = 8.5 Hz, 2H), 7.39 -7.33 (m, 3H), 4.37 (q, J = 7.1 Hz, 2H), 1.39 (t, J = 7.1 Hz, 3H); ^{13}C NMR (75 MHz, CDCl_3) δ 182.6, 162.0, 146.7, 137.6, 132.5, 130.1, 129.4, 121.0, 62.6, 14.1.

(3E)-4-(4-Nitrophenyl)-2-oxobut-3-enoic acid methyl ester (2e): 23.0% yield in two steps; yellow crystal; mp = 185-187 °C (literature.⁶ 182.5-183.5 °C); ^1H NMR (300 MHz, CDCl_3) δ 8.30 (d, J = 8.8 Hz, 2H), 7.90 (d, J = 16.2 Hz, 1H), 7.81 (d, J = 8.8 Hz, 2H), 7.52 (d, J = 16.2 Hz, 1H), 3.97 (s, 3H); ^{13}C NMR (75 MHz, CDCl_3) δ 181.7, 161.9, 149.2, 145.0, 139.9, 129.6, 124.3, 123.8, 53.4.

1.2 Procedures for the discovery of the reaction

A solution of 2,3-dimethylaniline (**1a**, 0.2 mmol), (3E)-2-oxo-4-phenylbut-3-enoate methyl ester (**2a**, 0.2 mmol) and $\text{Hf}(\text{OTf})_4$ (0.02 mmol) in 2mL of dichloromethane was stirred at room temperature for 48 h. Then, the solution was diluted with 20 mL of CH_2Cl_2 , washed with 5 mL of saturated aqueous NaHCO_3 and then 5 mL of water, dried over anhydrous Na_2SO_4 , filtered, and evaporated under reduced pressure. The products were isolated by flash chromatography on silica gel with ethyl acetate-petroleum ether mixture (1:5, v/v) to give methyl 7,8-dimethyl-2-phenylquinoline-4-carboxylate (**4a**, 44%) and methyl 7,8-dimethyl-4-phenylquinoline-2-carboxylate (**3a**, 18%).

Methyl 7,8-dimethyl-2-phenylquinoline-4-carboxylate (4a): 44.0% yield; colorless crystal; mp = 133-134 °C; ^1H NMR (300 MHz, CDCl_3) δ 8.46 (d, J = 8.7 Hz, 1H), 8.34 (s, 1H), 8.31-8.27 (m, 2H), 7.57-7.42 (m, 4H), 4.06 (s, 3H), 2.88 (s, 3H), 2.53 (s, 3H); ^{13}C NMR (75 MHz, CDCl_3) δ 167.3, 154.6, 148.0, 139.2, 137.7, 135.8, 135.3, 130.7, 129.5, 128.8, 127.4, 122.3, 122.1, 118.5, 52.6, 20.7, 13.7; FTIR (KBr) 2949, 1724, 1432, 1254, 1201, 761, 695 cm^{-1} ; HRMS (FAB) Calcd. For $(\text{M} + \text{H})^+$ $\text{C}_{19}\text{H}_{18}\text{NO}_2$: 292.1332, Found: 292.1335; Anal. Calcd. For $\text{C}_{19}\text{H}_{17}\text{NO}_2$: C, 78.33; H, 5.88; N, 4.81. Found: C, 78.11; H, 5.89; N, 4.79.

Methyl 7,8-dimethyl-4-phenylquinoline-2-carboxylate (3a): 18.0% yield; colorless crystal; mp = 119-120 °C; ^1H NMR (300 MHz, CDCl_3) δ 8.06 (s, 1H), 7.70 (d, J = 8.6 Hz, 1H), 7.51 (s, br, 5H), 7.39 (d, J = 8.6 Hz, 1H), 4.07 (s, 3H), 2.90 (s, 3H), 2.52 (s, 3H); ^{13}C NMR (75 MHz, CDCl_3) δ 166.5, 149.7, 147.2, 146.1, 138.1, 137.9, 136.2, 131.5, 129.6, 128.6, 128.5, 126.2, 122.6, 120.2, 52.9, 20.7, 13.8; FTIR (KBr) 2951, 1718, 1444, 1248, 1131, 765, 706 cm^{-1} ; HRMS (FAB) Calcd. For $(\text{M} + \text{H})^+$ $\text{C}_{19}\text{H}_{18}\text{NO}_2$: 292.1332, Found: 292.1332; Anal. Calcd. For $\text{C}_{19}\text{H}_{17}\text{NO}_2$: C, 78.33; H, 5.88; N, 4.81. Found: C, 78.19; H, 5.92; N, 4.79.

A mixture of **1a** (0.2 mmol) and **2a** (0.2 mmol) in 2 mL of 37% aqueous hydrochloric acid was stirred at room temperature for 24 h. The solution was added saturated aqueous NaHCO_3 until pH = 7, and then was extracted with 20 mL of CH_2Cl_2 . The organic part

was dried over anhydrous Na_2SO_4 , filtered, and evaporated under reduced pressure. The products were isolated by flash chromatography on silica gel with ethyl acetate-petroleum ether mixture (1:5, v/v) to give **4a** (trace, <2%) and **3a** (trace, <2%).

Hydrogen chloride passed through the mixture of **1a** (0.2 mmol) and **2a** (0.2 mmol) in 2 mL of CH_2Cl_2 for 10 min. Then the solution was stirred at room temperature for 24 h. The solution was added saturated aqueous NaHCO_3 until pH = 7, and then was extracted with 20 mL of CH_2Cl_2 . The organic part was dried over anhydrous Na_2SO_4 , filtered, and evaporated under reduced pressure. The products were isolated by flash chromatography on silica gel with ethyl acetate-petroleum ether mixture (1:5, v/v) to give **4a** (trace, <2%) and **3a** (trace, <2%).

Hydrogen chloride passed through the mixture of **1a** (0.2 mmol) and **2a** (0.2 mmol) in 2 mL of CH_2Cl_2 for 10 min. Then the solution was refluxed for 24 h. The solution was added saturated aqueous NaHCO_3 until pH = 7, and then was extracted with 20 mL of CH_2Cl_2 . The organic part was dried over anhydrous Na_2SO_4 , filtered, and evaporated under reduced pressure. The products were isolated by flash chromatography on silica gel with ethyl acetate-petroleum ether mixture (1:5, v/v) to give **4a** (trace, <2%) and **3a** (trace, <2%).

Hydrogen chloride passed through the mixture of **1a** (0.2 mmol) and **2a** (0.2 mmol) in 2 mL of PhMe for 10 min. Then the solution was refluxed for 24 h. The solution was added saturated aqueous NaHCO_3 until pH = 7, and then was extracted with 20 mL of CH_2Cl_2 . The organic part was dried over anhydrous Na_2SO_4 , filtered, and evaporated under reduced pressure. The products were isolated by flash chromatography on silica gel with ethyl acetate-petroleum ether mixture (1:5, v/v) to give **4a** (trace, <2%) and **3a** (trace, <2%).

A mixture of **1a** (0.2 mmol), **2a** (0.2 mmol) and H_2SO_4 (0.2 mmol) in 2 mL of CH_2Cl_2 was refluxed for 24 h. The solution was added saturated aqueous NaHCO_3 until pH = 7, and then was extracted with 20 mL of CH_2Cl_2 . The organic part was dried over anhydrous Na_2SO_4 , filtered, and evaporated under reduced pressure. The products were isolated by flash chromatography on silica gel with ethyl acetate-petroleum ether mixture (1:5, v/v) to give **4a** (trace, <2%) and **3a** (trace, <2%).

A mixture of **1a** (0.2 mmol), **2a** (0.2 mmol) and TFA (0.2 mmol) in 2 mL of boiling CH_2Cl_2 was stirred for 24 h. Then, the solution was diluted with 20 mL of CH_2Cl_2 , washed with 5 mL of saturated aqueous NaHCO_3 and then 5 mL of water, dried over anhydrous Na_2SO_4 , filtered, and evaporated under reduced pressure. The products were isolated by flash chromatography on silica gel with ethyl acetate-petroleum ether mixture (1:5, v/v) to give **4a** (46%) and **3a** (18%).

A mixture of **1a** (0.2 mmol), **2a** (0.2 mmol) and TFA (0.2 mmol) in 2 mL of boiling PhMe was stirred for 24 h. Then, the solution was diluted with 20 mL of CH₂Cl₂, washed with 5 mL of saturated aqueous NaHCO₃ and then 5 mL of water, dried over anhydrous Na₂SO₄, filtered, and evaporated under reduced pressure. The products were isolated by flash chromatography on silica gel with ethyl acetate-petroleum ether mixture (1:5, v/v) to give **4a** (33%) and **3a** (29%).

A mixture of **1a** (0.2 mmol) and **2a** (0.2 mmol) in 2 mL of boiling TFA was stirred for 12 h. TFA was distilled out for reuse, the residue was redissolved in 20 mL of CH₂Cl₂, washed with 5 mL of saturated aqueous NaHCO₃, dried over anhydrous Na₂SO₄, filtered, and evaporated under reduced pressure. The products were isolated by flash chromatography on silica gel with ethyl acetate-petroleum ether mixture (1:5, v/v) to give **4a** (trace, <2%) and **3a** (61%).

A mixture of **1a** (0.2 mmol) and **2a** (0.4 mmol) in 2 mL of boiling TFA was stirred for 12 h. TFA was distilled out for reuse, the residue was redissolved in 20 mL of CH₂Cl₂, washed with 5 mL of saturated aqueous NaHCO₃, dried over anhydrous Na₂SO₄, filtered, and evaporated under reduced pressure. The products were isolated by flash chromatography on silica gel with ethyl acetate-petroleum ether mixture (1:5, v/v) to give **4a** (trace, <2%) and **3a** (80%).

A mixture of **1a** (0.2 mmol) and **2a** (0.4 mmol) in 2 mL of boiling formic acid was stirred for 12 h. Formic acid was distilled out for reuse, the residue was redissolved in 20 mL of CH₂Cl₂, washed with 5 mL of saturated aqueous NaHCO₃, dried over anhydrous Na₂SO₄, filtered, and evaporated under reduced pressure. The products were isolated by flash chromatography on silica gel with ethyl acetate-petroleum ether mixture (1:5, v/v) to give **4a** (trace, <2%) and **3a** (76%).

A mixture of **1a** (0.2 mmol) and **2a** (0.4 mmol) in 2 mL of boiling acetic acid was stirred for 24 h. Acetic acid was distilled out for reuse, the residue was redissolved in 20 mL of CH₂Cl₂, washed with 5 mL of saturated aqueous NaHCO₃, dried over anhydrous Na₂SO₄, filtered, and evaporated under reduced pressure. The products were isolated by flash chromatography on silica gel with ethyl acetate-petroleum ether mixture (1:5, v/v) to give **4a** (trace, <2%) and **3a** (trace, <2%).

1.3 Synthesis of 2-carboxy-4-aryl-quinolines

A mixture of an aniline (**1**, 0.2 mmol) and an γ -aryl- β , γ -unsaturated α -ketoesters **2** (0.4 mmol) in 2 mL of TFA was stirred at reflux for 8-18 h, after which TFA was distilled out for reuse. The residue was redissolved in 20 mL of CH₂Cl₂, and the solution was washed with 5 mL of saturated aqueous NaHCO₃, dried over anhydrous Na₂SO₄, filtered, and

evaporated under reduced pressure. The products **3** were isolated by flash chromatography on silica gel with ethyl acetate-petroleum ether mixture (1:5, v/v).

Methyl 7,8-dimethyl-4-phenylquinoline-2-carboxylate (3a): 80% yield; white solid; mp = 119-120 °C; ¹H NMR (300 MHz, CDCl₃) δ 8.06 (s, 1H), 7.70 (d, *J* = 8.6 Hz, 1H), 7.51 (s, br, 5H), 7.39 (d, *J* = 8.6 Hz, 1H), 4.07 (s, 3H), 2.90 (s, 3H), 2.52 (s, 3H); ¹³C NMR (75 MHz, CDCl₃) δ 166.5, 149.7, 147.2, 146.1, 138.1, 137.9, 136.2, 131.5, 129.6, 128.6, 128.5, 126.2, 122.6, 120.2, 52.9, 20.7, 13.8; FTIR (KBr) 2951, 1718, 1444, 1248, 1131, 765, 706 cm⁻¹; HRMS (FAB) Calcd. For (M + H)⁺ C₁₉H₁₈NO₂: 292.1332, Found: 292.1332; Anal. Calcd. For C₁₉H₁₇NO₂: C, 78.33; H, 5.88; N, 4.81. Found: C, 78.19; H, 5.92; N, 4.79.

Methyl 6-methoxy-4-phenylquinoline-2-carboxylate (3b): 81% yield; white solid; mp = 141-142 °C; ¹H NMR (300 MHz, CDCl₃) δ 8.22 (dd, *J* = 9.2, 2.4 Hz, 1H), 8.08 (d, *J* = 2.4 Hz, 1H), 7.51-7.50 (s, br, 5H), 7.40 (dd, *J* = 9.3, 2.7 Hz, 1H), 7.19 (d, *J* = 2.4 Hz, 1H), 4.04 (s, 3H), 3.77 (s, 3H); ¹³C NMR (75 MHz, CDCl₃) δ 166.1, 159.6, 148.1, 145.0, 144.3, 137.79, 132.6, 129.3, 129.2, 128.8, 128.7, 122.9, 121.8, 103.3, 55.5, 53.1; FTIR (KBr) 3055, 2952, 1740, 1621, 1472, 1226, 1108, 836, 707 cm⁻¹; Anal. Calcd. For C₁₈H₁₅NO₃: C, 73.71; H, 5.15; N, 4.78. Found: C, 73.77; H, 5.14; N, 4.75.

Methyl 6-methyl-4-phenylquinoline-2-carboxylate (3c): 82% yield; white solid; mp = 129-130 °C; ¹H NMR (300 MHz, CDCl₃) δ 8.25 (d, *J* = 8.7 Hz, 1H), 8.11 (s, 1H), 7.71 (s, 1H), 7.58 (d, *J* = 8.8 Hz, 1H), 7.52 (s, br, 5H), 4.07 (s, 3H), 2.48 (s, 3H); ¹³C NMR (75 MHz, CDCl₃) δ 166.2, 149.0, 146.8, 146.5, 139.1, 137.7, 132.5, 130.8, 129.5, 128.7, 128.6, 127.9, 124.4, 121.5, 53.2, 22.1; FTIR (KBr) 3058, 2945, 1720, 1448, 1361, 1253, 1119, 824, 759, 703 cm⁻¹; MS (TOF EI): m/z 277; Anal. Calcd. For C₁₈H₁₅NO₂: C, 77.96; H, 5.45; N, 5.05. Found: C, 77.92; H, 5.47; N, 4.91.

Methyl 6-fluoro-4-phenylquinoline-2-carboxylate (3d): 80% yield; white solid; mp = 134-135 °C; ¹H NMR (300 MHz, CDCl₃) δ 8.40-8.35 (m 1H), 8.17 (s, 1H), 7.61-7.49 (m, 7H), 4.09 (s, 3H); ¹³C NMR (75 MHz, CDCl₃) δ 165.8, 162.0(d, ¹J_{CF} = 249.8 Hz, 1C), 149.4(d, ³J_{CF} = 5.9 Hz, 1C), 146.9(d, ⁴J_{CF} = 2.9 Hz, 1C), 145.2, 137.0, 133.8(d, ³J_{CF} = 9.3 Hz, 1C), 129.4, 129.0, 128.9, 121.8, 120.7(d, ²J_{CF} = 26.0 Hz, 1C), 109.3(d, ²J_{CF} = 23.2 Hz, 1C), 53.3; FTIR (KBr) 3056, 1714, 1465, 1366, 1258, 1231, 1195, 1140, 1108, 836, 785, 710 cm⁻¹; GCT-MS (TOF EI⁺): m/z 281; HRMS (TOF EI⁺) Calcd. For M⁺ C₁₇H₁₂NO₂F: 281.0852, Found: 281.0851; Anal. Calcd. For C₁₇H₁₂NO₂F: C, 72.59; H, 4.30; N, 4.98. Found: C, 72.47; H, 4.22; N, 4.97.

Methyl 6-nitro-4-phenylquinoline-2-carboxylate (3e): 69% yield; yellow solid; mp = 250-251 °C; ¹H NMR (300 MHz, CDCl₃) δ 8.92 (d, *J* = 2.1 Hz, 1H), 8.55 (dd, *J* = 9.3, 2.3 Hz, 1H), 8.50 (d, *J* = 9.2 Hz, 1H), 8.29 (s, 1H), 7.63-7.53 (m, 5H), 4.12 (s, 3H); ¹³C

NMR (75 MHz, CDCl₃) δ 165.3, 152.5, 150.4, 150.1, 147.0, 136.0, 133.0, 129.8, 129.5, 129.3, 127.0, 123.6, 122.8, 122.7, 53.6; FTIR (KBr) 3074, 2955, 1724, 1487, 1367, 1344, 1258, 1140, 797, 704 cm⁻¹; HRMS (FAB) calcd. For (M + H)⁺ C₁₇H₁₃N₂O₄: 309.0870, Found: 309.0869; Anal. Calcd. For C₁₇H₁₂N₂O₄: C, 66.23; H, 3.92; N, 9.09. Found: C, 65.95; H, 3.92; N, 8.93.

Methyl 4-phenylbenzo[*h*]quinoline-2-carboxylate (3f): 81% yield; white solid; mp = 177-178 °C; ¹H NMR (300 MHz, CDCl₃) δ 9.42 (dd, *J* = 8.4, 1.3 Hz, 1H), 8.19 (s, 1H), 7.74-7.64 (m, 5H), 7.47-7.45 (m, 5H), 4.03 (s, 3H); ¹³C NMR (75 MHz, CDCl₃) δ 166.3, 149.4, 146.7, 145.7, 137.9, 133.4, 131.8, 129.9, 129.7, 128.9, 128.7, 128.6, 127.7, 127.6, 126.2, 125.5, 122.7, 122.6, 53.0; FTIR (KBr) 3047, 2947, 1739, 1434, 1244, 1131, 839, 756, 707 cm⁻¹; Anal. Calcd. For C₂₁H₁₅NO₂: C, 80.49; H, 4.82; N, 4.47. Found: C, 80.24; H, 4.97; N, 4.31.

Methyl 1-phenylbenzo[*f*]quinoline-3-carboxylate (3g): 82% yield; white solid; mp = 152-153 °C; ¹H NMR (300 MHz, CDCl₃) δ 8.10 (d, *J* = 9.1 Hz, 1H), 8.07 (s, 1H), 7.92 (d, *J* = 9.1 Hz, 1H), 7.77 (d, *J* = 7.9 Hz, 1H), 7.65 (d, *J* = 8.5 Hz, 1H), 7.44-7.31 (m, 6H), 7.08 (dt, *J* = 7.8, 1.3 Hz, 1H), 4.00 (s, 3H); ¹³C NMR (75 MHz, CDCl₃) δ 165.8, 149.7, 149.5, 145.8, 142.1, 133.7, 132.3, 129.4, 129.2, 128.9, 128.7, 128.6, 128.5, 128.2, 127.7, 126.0, 125.9, 124.8, 53.2; FTIR (KBr) 3027, 2950, 1743, 1240, 1131, 777, 756, 710 cm⁻¹; Anal. Calcd. For C₂₁H₁₅NO₂: C, 80.49; H, 4.82; N, 4.47. Found: C, 80.57; H, 4.89; N, 4.62.

Methyl 4-phenylquinoline-2-carboxylate (3h): 80% yield; white solid; mp = 101-102 °C; ¹H NMR (300 MHz, CDCl₃) δ 8.39 (t, *J* = 8.1 Hz, 1H), 8.17 (s, 1H), 7.98 (t, *J* = 7.7 Hz, 1H), 7.79 (dd, *J* = 7.0, 7.9 Hz, 1H), 7.61-7.53 (m, 6H), 4.11 (s, 3H); ¹³C NMR (75 MHz, CDCl₃) δ 166.0, 150.0, 148.1, 147.4, 137.4, 131.1, 130.1, 129.6, 128.8, 128.7, 127.9, 125.8, 121.3, 53.2; FTIR (KBr) 2953, 1714, 1560, 1457, 1365, 1257, 1131, 1111, 792, 766, 699 cm⁻¹; Anal. Calcd. For C₁₇H₁₃NO₂: C, 77.55; H, 4.98; N, 5.32. Found: C, 77.34; H, 5.06; N, 5.47.

Methyl 4-(4-methoxyphenyl)quinoline-2-carboxylate (3i): 79% yield; white solid; mp = 131-132 °C; ¹H NMR (300 MHz, CDCl₃) δ 8.29 (d, *J* = 8.5 Hz, 1H), 8.05 (s, 1H), 7.94 (d, *J* = 8.5 Hz, 1H), 7.70 (dt, *J* = 7.1, 1.3 Hz, 1H), 7.51 (dt, *J* = 8.2, 1.1 Hz, 1H), 7.40 (dd, *J* = 8.8, 2.1 Hz, 2H), 6.99 (d, *J* = 8.8 Hz, 2H), 4.01 (s, 3H), 3.82 (s, 3H); ¹³C NMR (75 MHz, CDCl₃) δ 166.1, 160.2, 149.8, 148.2, 147.4, 131.1, 130.9, 130.0, 129.7, 128.5, 128.0, 125.8, 121.2, 114.2, 55.4, 53.2; FTIR (KBr) 2955, 1718, 1455, 1368, 1259, 1135, 839, 821, 787, 762 cm⁻¹; Anal. Calcd. For C₁₈H₁₅NO₃: C, 73.71; H, 5.15; N, 4.78. Found: C, 73.85; H, 5.33; N, 4.62.

Methyl 4-*p*-tolylquinoline-2-carboxylate (3j): 82% yield; white solid; mp = 114-115 °C;

¹H NMR (300 MHz, CDCl₃) δ 8.50 (d, *J* = 8.5 Hz, 1H), 8.07 (s, 1H), 7.93 (d, *J* = 8.4 Hz, 1H), 7.71 (dt, *J* = 7.7, 1.2 Hz, 1H), 7.52 (dt, *J* = 8.2, 1.1 Hz, 1H), 7.36 (d, *J* = 8.0 Hz, 2H), 7.28 (d, *J* = 8.0 Hz, 2H), 4.02 (s, 3H), 2.39 (s, 3H); ¹³C NMR (75 MHz, CDCl₃) δ 166.0, 150.2, 148.1, 147.4, 138.8, 134.5, 131.0, 130.1, 129.5, 129.4, 128.6, 128.0, 125.9, 121.3, 53.2, 21.3; FTIR (KBr) 2948, 1723, 1254, 1131, 1111, 825, 780, 728 cm⁻¹; Anal. Calcd. For C₁₈H₁₅NO₂: C, 77.96; H, 5.45; N, 5.05. Found: C, 77.75; H, 5.61; N, 5.00.

Ethyl 4-(4-chlorophenyl)quinoline-2-carboxylate (3k): 83% yield; white solid; mp = 126-127 °C; ¹H NMR (300 MHz, CDCl₃) δ 8.36 (d, *J* = 8.4 Hz, 1H), 8.09 (s, 1H), 7.89 (d, *J* = 8.3 Hz, 1H), 7.78 (t, *J* = 7.6 Hz, 1H), 7.59 (t, *J* = 7.6 Hz, 1H), 7.51 (d, *J* = 8.3 Hz, 2H), 7.45 (q, *J* = 8.3 Hz, 2H), 4.55 (q, *J* = 7.0 Hz, 2H), 1.47 (t, *J* = 7.0 Hz, 3H); ¹³C NMR (75 MHz, CDCl₃) δ 165.4, 148.5, 148.2, 147.8, 135.9, 135.0, 131.3, 130.9, 130.2, 129.0, 128.8, 127.5, 125.3, 121.2, 62.4, 14.4; FTIR (KBr) 3074, 2984, 1718, 1490, 1377, 1253, 1108, 1018, 831, 763 cm⁻¹; Anal. Calcd. For C₁₈H₁₄ClNO₂: C, 69.35; H, 4.53; N, 4.49. Found: C, 69.18; H, 4.57; N, 4.35.

Methyl 4-(4-nitrophenyl)quinoline-2-carboxylate (3l): 72% yield; yellow solid; mp = 205-206 °C; ¹H NMR (300 MHz, CDCl₃) δ 8.33-8.31 (m, 3H), 8.06 (s, 1H), 7.75 (t, *J* = 7.1 Hz, 2H), 7.65-7.56 (m, 3H), 4.01 (s, 3H); ¹³C NMR (75 MHz, CDCl₃) δ 165.7, 148.1, 148.1, 147.5, 147.3, 143.9, 131.4, 130.6, 129.5, 127.0, 124.9, 124.0, 121.1, 53.4; FTIR (KBr) 3066, 2951, 1723, 1514, 1347, 1254, 1133, 1110, 832, 768; Anal. Calcd. For C₁₇H₁₂N₂O₄: C, 66.23; H, 3.92; N, 9.09. Found: C, 66.09; H, 4.13; N, 9.05.

Ethyl 4-(4-chlorophenyl)-6-methoxyquinoline-2-carboxylate (3m): 82% yield; white solid; mp = 171-172 °C; ¹H NMR (300 MHz, CDCl₃) δ 8.25 (d, *J* = 9.3 Hz, 1H), 8.04 (s, 1H), 7.52 (d, *J* = 8.4 Hz, 2H), 7.47 (d, *J* = 8.6 Hz, 2H), 7.42 (dd, *J* = 9.3, 2.7 Hz, 1H), 7.11 (d, *J* = 2.6 Hz, 1H), 4.53 (q, *J* = 7.1 Hz, 2H), 3.80 (s, 3H), 1.47 (t, *J* = 7.1 Hz, 3H); ¹³C NMR (75 MHz, CDCl₃) δ 165.5, 159.7, 146.7, 145.3, 144.3, 136.3, 134.8, 132.9, 130.7, 129.1, 128.9, 123.0, 121.7, 102.9, 62.2, 55.6, 14.4; FTIR (KBr) 3053, 2975, 1729, 1618, 1489, 1471, 1280, 1228, 1110, 1022, 840 cm⁻¹; Anal. Calcd. For C₁₉H₁₆ClNO₃: C, 66.77; H, 4.72; N, 4.10. Found: C, 66.50; H, 4.73; N, 3.82.

Ethyl 4-(4-chlorophenyl)-8-methoxyquinoline-2-carboxylate (3n): 66% yield; white solid; mp = 161-162 °C; ¹H NMR (300 MHz, CDCl₃) δ 8.14 (d, *J* = 4.6 Hz, 1H), 7.57-7.43 (m, 6H), 7.11 (t, *J* = 5.9 Hz, 1H), 4.58-4.50 (m, 2H), 4.11 (d, *J* = 5.3 Hz, 3H), 1.54-1.48 (m, 3H); ¹³C NMR (75 MHz, CDCl₃) δ 165.5, 156.2, 148.4, 146.4, 140.1, 136.2, 134.9, 130.9, 129.2, 128.9, 128.8, 121.9, 116.9, 107.9, 62.4, 56.2, 14.3; FTIR (KBr) 3423, 3046, 2983, 1721, 1466, 1261, 1081, 836, 756 cm⁻¹; Anal. Calcd. For C₁₉H₁₆ClNO₃: C, 66.77; H, 4.72; N, 4.10. Found: C, 66.63; H, 4.80; N, 3.99.

Ethyl 6-chloro-4-(4-chlorophenyl)quinoline-2-carboxylate (3o): 83% yield; white

solid; mp = 220-221 °C; ^1H NMR (300 MHz, CDCl_3) δ 8.31 (d, J = 9.0 Hz, 1H), 8.12 (s, 1H), 7.87 (d, J = 2.3 Hz, 1H), 7.73 (dd, J = 9.0, 2.3 Hz, 1H), 7.56 (d, J = 8.4 Hz, 2H), 7.47 (d, J = 8.4 Hz, 2H), 4.58 (q, J = 7.1 Hz, 2H), 1.50 (t, J = 7.1 Hz, 3H); ^{13}C NMR (75 MHz, CDCl_3) δ 165.0, 148.0, 147.7, 146.5, 135.4, 135.2, 135.1, 132.8, 131.2, 130.8, 129.2, 128.2, 124.2, 122.0, 62.5, 14.4; FTIR (KBr) 3031, 2985, 1716, 1597, 1485, 1450, 1377, 1273, 1253, 1141, 1087, 1020, 847, 829, 789 cm^{-1} ; Anal. Calcd. For $\text{C}_{18}\text{H}_{13}\text{Cl}_2\text{NO}_2$: C, 62.45; H, 3.78; N, 4.05. Found: C, 62.18; H, 3.89, N, 3.93.

Ethyl 8-chloro-4-(4-chlorophenyl)quinoline-2-carboxylate (3p): 71% yield; white solid; mp = 179-180 °C; ^1H NMR (300 MHz, CDCl_3) δ 8.14 (s, 1H), 7.91 (dd, J = 7.4, 0.8 Hz, 1H), 7.83 (dd, J = 8.5, 1.0 Hz, 1H), 7.51 (t, J = 8.2 Hz, 1H), 7.53 (d, J = 8.5 Hz, 2H), 7.46 (d, J = 8.5 Hz, 2H), 4.54 (q, J = 7.1, 2H), 1.50 (t, J = 7.1, 3H); ^{13}C NMR (75 MHz, CDCl_3) δ 165.1, 149.2, 148.2, 144.6, 135.6, 135.5, 135.3, 130.9, 129.4, 129.1, 129.0, 128.5, 124.5, 122.0, 62.5, 14.3; FTIR (KBr) 3052, 2985, 1721, 1447, 1250, 1130, 785, 764 cm^{-1} ; Anal. Calcd. For $\text{C}_{18}\text{H}_{13}\text{Cl}_2\text{NO}_2$: C, 62.45; H, 3.78; N, 4.05. Found: C, 62.18; H, 3.84; N, 3.89.

Ethyl 4-(4-chlorophenyl)-6-nitroquinoline-2-carboxylate (3q): 68% yield; yellow solid; mp = 240-241 °C; ^1H NMR (300 MHz, CDCl_3) δ 8.86 (d, J = 1.9 Hz, 1H), 8.56 (dd, J = 9.3, 2.2 Hz, 1H), 8.52 (d, J = 9.2 Hz, 1H), 8.25 (s, 1H), 7.61 (dd, J = 8.5, 2.1 Hz, 2H), 7.50 (dd, J = 8.4, 2.1 Hz, 1H), 4.60 (q, J = 7.1 Hz, 2H), 1.46 (t, J = 7.1 Hz, 3H); ^{13}C NMR (75 MHz, CDCl_3) δ 164.6, 151.0, 150.9, 150.2, 147.0, 136.2, 134.4, 133.2, 130.8, 129.6, 126.7, 123.6, 122.6, 122.4, 62.9, 14.4; FTIR (KBr) 3070, 2981, 1721, 1487, 1345, 1250, 1112, 837, 745 cm^{-1} ; Anal. Calcd. For $\text{C}_{18}\text{H}_{13}\text{ClN}_2\text{O}_4$: C, 60.60; H, 3.67; N, 7.85. Found: C, 60.60; H, 3.65; N, 7.65.

Ethyl 4-(4-chlorophenyl)-8-nitroquinoline-2-carboxylate (3r): 42% yield; yellow solid; mp = 206-207 °C; ^1H NMR (300 MHz, CDCl_3) δ 8.22 (s, 1H), 8.13 (d, J = 8.5 Hz, 1H), 8.17 (d, J = 7.5 Hz, 1H), 7.68 (t, J = 8.0 Hz, 1H), 7.57 (d, J = 8.5 Hz, 2H), 7.46 (d, J = 8.5 Hz, 2H), 4.52 (q, J = 7.1, 2H), 1.48 (t, J = 7.1, 3H); ^{13}C NMR (75 MHz, CDCl_3) δ 164.7, 149.8, 149.2, 149.0, 139.6, 135.8, 134.8, 130.9, 129.5, 129.3, 128.3, 127.2, 124.4, 122.7, 62.6, 14.2; FTIR (KBr) 2992, 1724, 1531, 1374, 1351, 1255, 1130, 1091, 1016, 842, 767 cm^{-1} ; HRMS (FAB) Calcd. For $\text{M}^+ \text{C}_{18}\text{H}_{14}\text{ClN}_2\text{O}_4$: 357.0637, Found: 357.0641; Anal. Calcd. For $\text{C}_{18}\text{H}_{14}\text{ClN}_2\text{O}_4$: C, 60.43; H, 3.94; N, 7.83. Found: C, 60.40; H, 3.98; N, 7.57.

Ethyl 4-(4-chlorophenyl)-6-hydroxyquinoline-2-carboxylate (3s): 79% yield; white solid; mp = 250-253 °C; ^1H NMR (300 MHz, $\text{DMSO}-d_6$) δ 10.42 (s, 1H), 8.08 (d, J = 8.8 Hz, 1H), 7.84 (s, 1H), 7.65-7.56 (m, 4H), 7.41 (d, J = 8.3 Hz, 1H), 7.07 (s, 1H), 4.38 (q, J = 6.5 Hz, 2H), 1.34 (t, J = 6.5 Hz, 3H); ^{13}C NMR (75 MHz, $\text{DMSO}-d_6$) δ 165.3, 158.6, 158.4, 145.7, 143.2, 136.6, 134.0, 132.8, 131.6, 129.3, 129.1, 121.3, 106.2, 106.1, 61.8,

14.7; FTIR (KBr) 3035, 1731, 1616, 1468, 1279, 1228, 836 cm^{-1} ; HRMS (FAB) calcd. For $(\text{M} + \text{H})^+$ $\text{C}_{18}\text{H}_{15}\text{ClNO}_3$: 328.0735, Found: 328.0743; Anal. Calcd. For $\text{C}_{18}\text{H}_{14}\text{ClNO}_3$: C, 65.96; H, 4.31; N, 4.27. Found: C, 65.86; H, 4.38; N, 4.11.

1.4 Synthesis of Schiff's base **5t**

According to the reported paper,⁷ a mixture of *trans*-cinnamaldehyde (0.10 mol), aniline (0.11 mol) and *p*-toluenesulfonic acid (0.05 mol) in CH_2Cl_2 (40 mL) was stirred at room temperature. When the mixture became turbid due to the formation of water, 25 g of 3 Å molecular sieves was added and the mixture then agitated for 45 min. The mixture was filtered, evaporated under reduced pressure, and recrystallized from ethanol to give pale yellow solid in 83% yield; mp 106-108 °C (literature.⁷ 105-106 °C); ^1H NMR (300 MHz, CDCl_3) δ 8.25 (d, $J = 7.4$ Hz, 1H), 7.54 (d, $J = 7.3$ Hz, 2H), 7.43-7.32 (m, 5H), 7.25 (t, $J = 7.6$ Hz, 3H), 7.16-7.14 (m, 2H); ^{13}C NMR (75 MHz, CDCl_3) δ 161.6, 151.8, 144.0, 135.7, 129.7, 129.3, 129.0, 128.7, 127.6, 126.2, 121.1.

1.5 Procedures for reaction mechanism

A mixture of 2,3-dimethylaniline (**1a**, 10 mmol) and (*3E*)-2-oxo-4-phenylbut-3-enoate methyl ester (**2a**, 20 mmol) in 10 mL of TFA was stirred at reflux for 6 h, after which TFA was distilled out for reuse. The residue was redissolved in 40 mL of CH_2Cl_2 , washed with 10 mL of saturated aqueous NaHCO_3 , dried over anhydrous Na_2SO_4 , filtered, and evaporated under reduced pressure. The products were isolated by flash chromatography on silica gel with ethyl acetate-petroleum ether mixture (1:5, v/v) to give (*3E*)-2-(2,3-dimethylphenylimino)-4-phenylbut-3-enoate methyl ester (**5a**, 2%) and methyl 7,8-dimethyl-4-phenylquinoline-2-carboxylate (**3a**, 51%).

(*3E*)-2-(2,3-Dimethylphenylimino)-4-phenylbut-3-enoate methyl ester (5a**, syn and anti):** 2% yield; yellow foam; ^1H NMR (300 MHz, CDCl_3) δ 7.65-6.61 (m, 10H), 4.03 (3.60) (s, 3H), 2.34 (2.31) (s, 3H), 2.17 (2.11) (s, 3H); ^{13}C NMR (75 MHz, CDCl_3) δ 165.4 (165.0), 158.2 (159.9), 147.5 (149.0), 142.2 (141.0), 137.8 (137.3), 135.4 (135.3), 130.0 (129.8), 128.9 (128.9), 127.9 (127.7), 126.6 (127.6), 126.5 (126.7), 125.7 (125.8), 117.3 (125.5), 116.7 (115.4), 51.88 (51.85), 20.2 (20.1), 14.1 (14.0); FTIR (neat) 3061, 3027, 2951, 1735, 1619, 1579, 1464, 1322, 1193, 1148, 969, 757, 730, 695 cm^{-1} ; GCT-MS (TOF MS EI^+) m/z 293 (M^+), 234, 218; HRMS (TOF MS EI^+) Calcd. For $\text{M}^+ \text{C}_{19}\text{H}_{19}\text{NO}_2$: 293.1416, Found: 293.1415.

A solution of Schiff's base **5a** (0.1 mmol) in 2 mL of TFA was stirred at reflux for 2 h, after that TFA was distilled out for reuse. The residue was redissolved in 20 mL of CH_2Cl_2 , washed with 5 ml of saturated aqueous NaHCO_3 , dried over anhydrous Na_2SO_4 , filtered, and evaporated under reduced pressure. The products were isolated by flash chromatography on silica gel with ethyl acetate-petroleum ether mixture (1:5, v/v) to give methyl 7,8-dimethyl-4-phenylquinoline-2-carboxylate (**3a**, 91%).

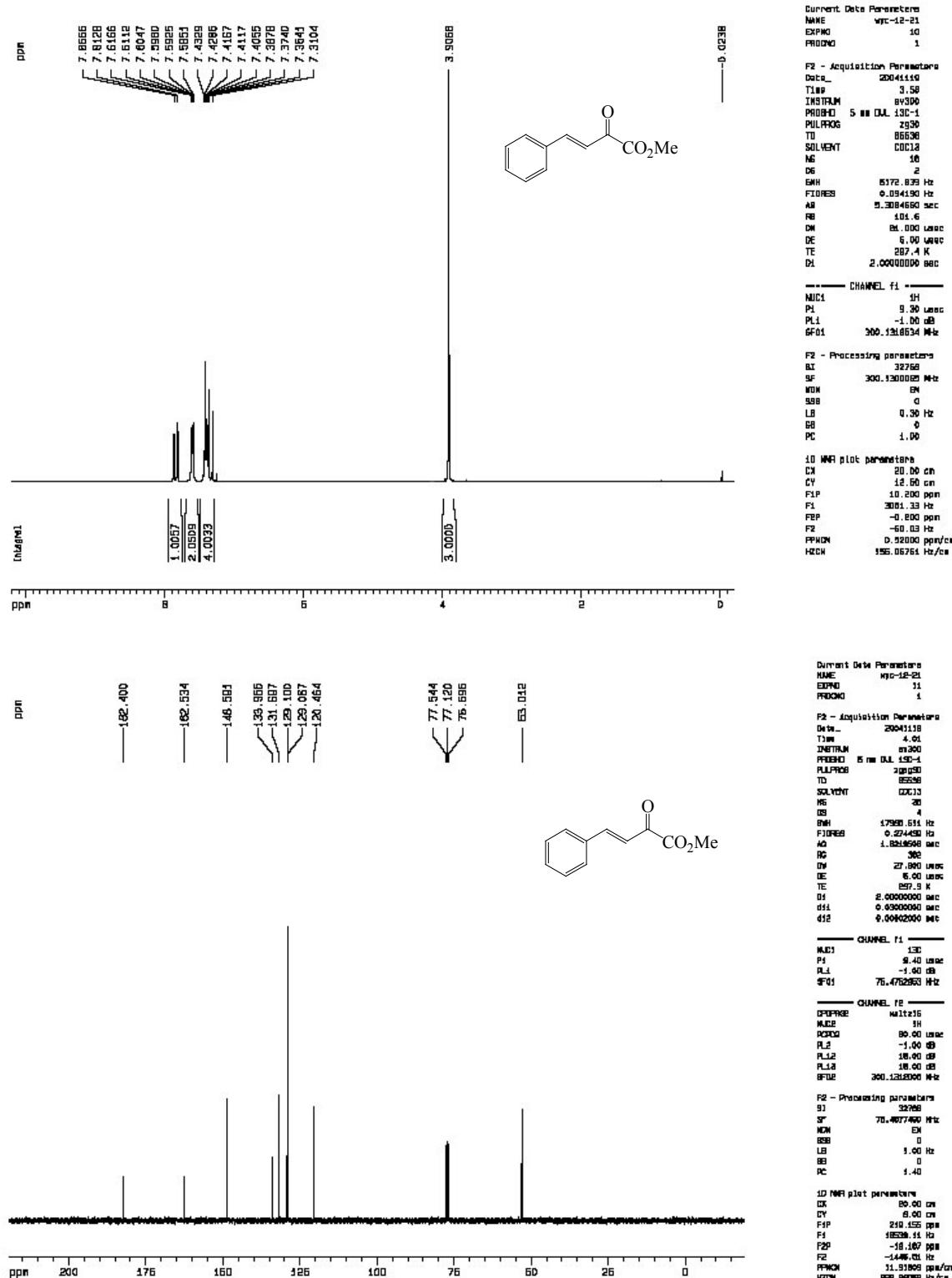
A solution of Schiff's base **5t** (0.1 mmol) in 2 mL of TFA was stirred at reflux for 2 h, after that TFA was distilled out for reuse. The residue was redissolved in 20 mL of CH₂Cl₂, washed with 5 ml of saturated aqueous NaHCO₃, dried over anhydrous Na₂SO₄, but Schiff's base **5t** was not consumed.

References

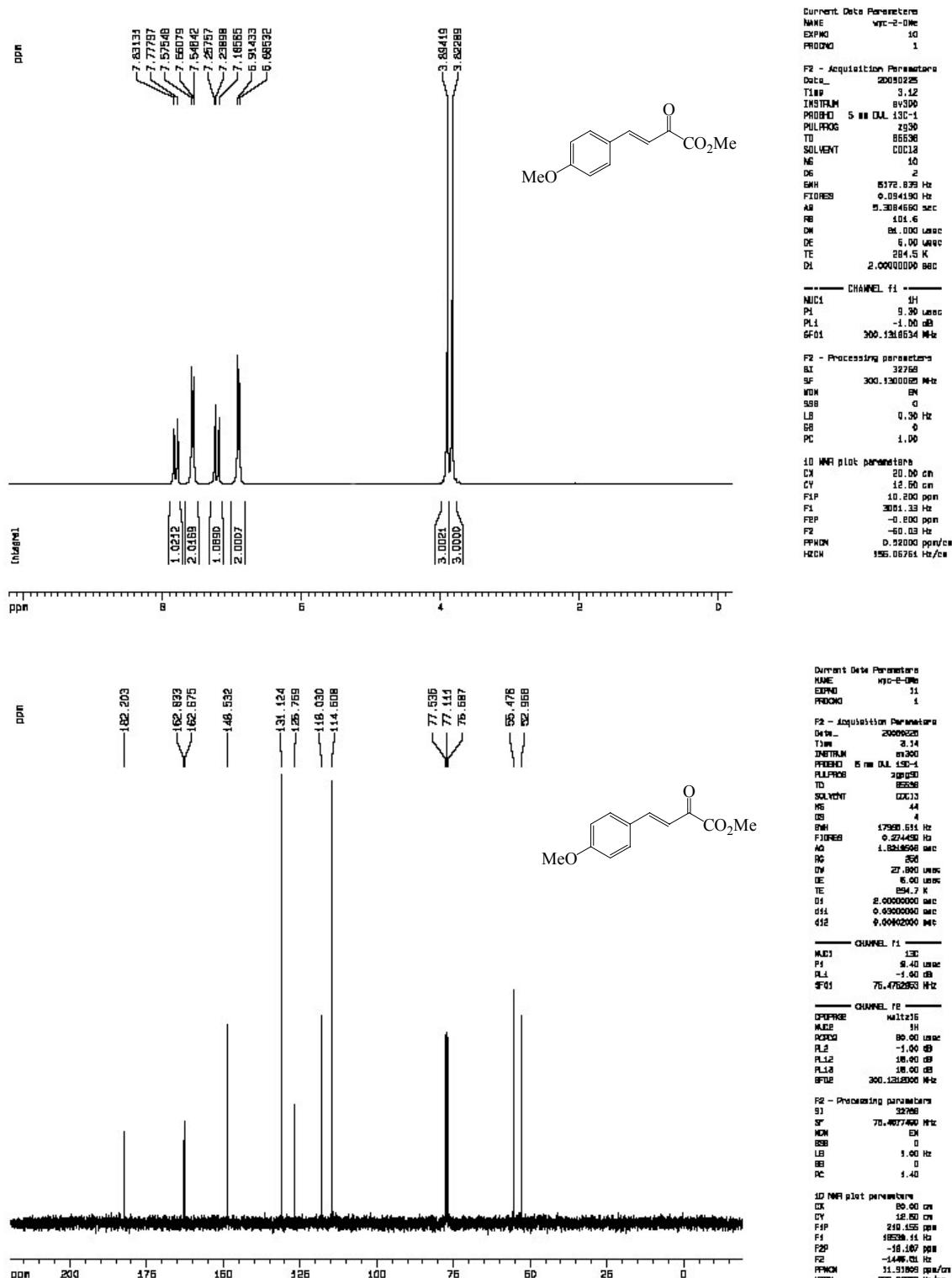
1. Stecher, E. D.; Ryder, H. F. *J. Am. Chem. Soc.*, **1952**, *74*, 4392.
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7. Eisch, J. J.; Dluzniewski, T. *J. Org. Chem.*, **1989**, *54*, 1269.

2. ^1H and ^{13}C NMR spectra of synthesized compounds

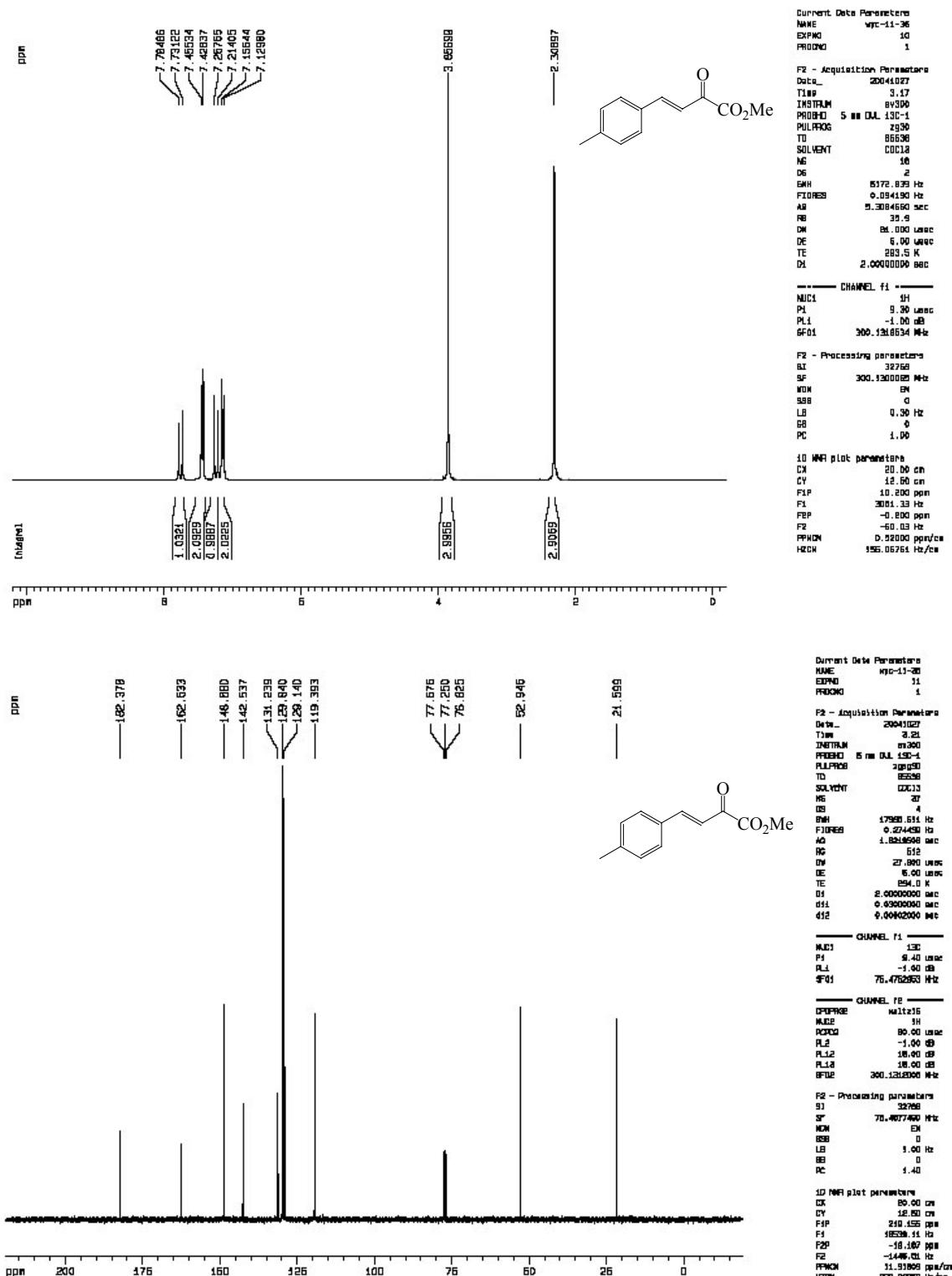
(3E)-2-Oxo-4-phenylbut-3-enoic acid methyl ester (2a)



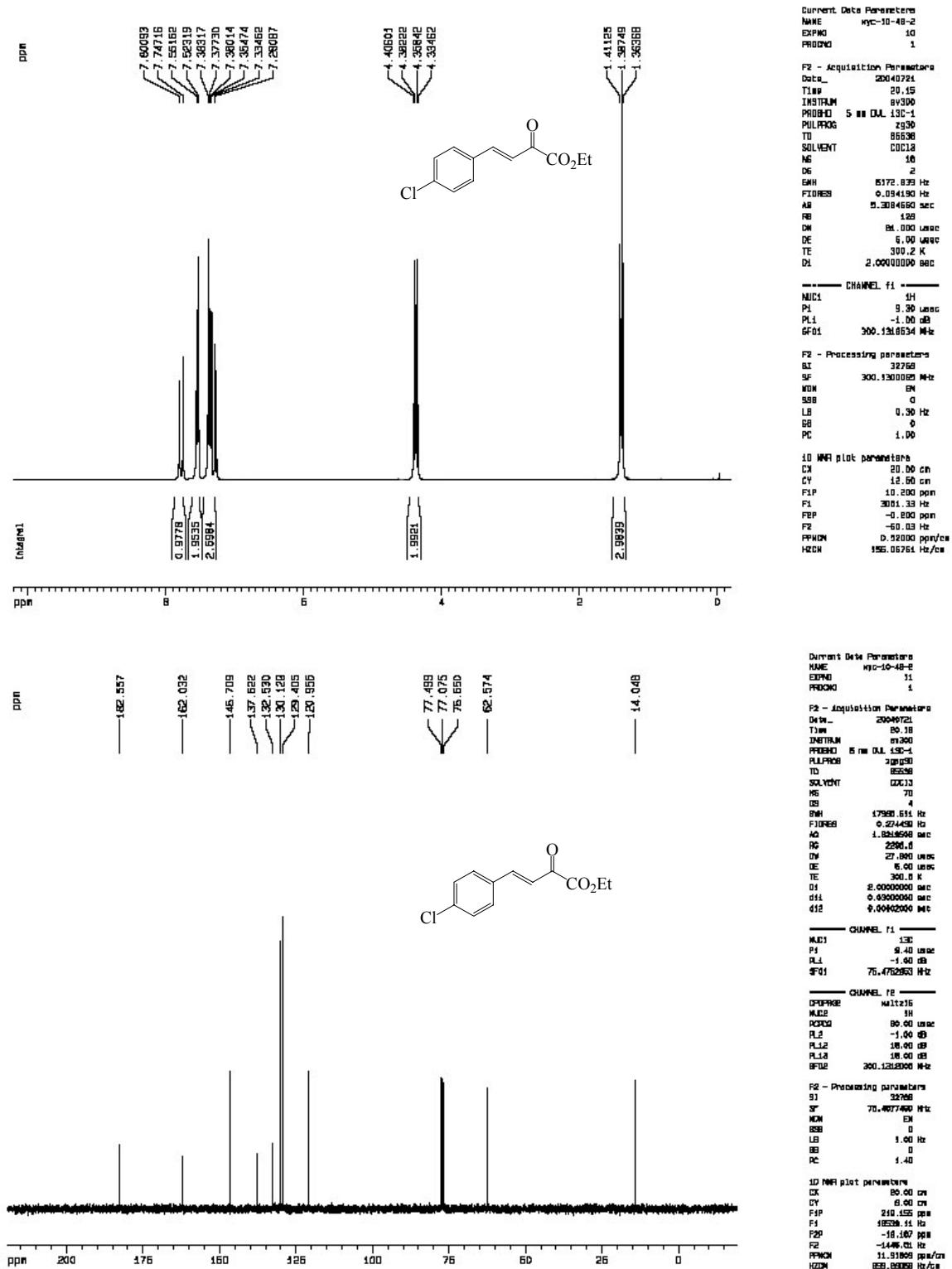
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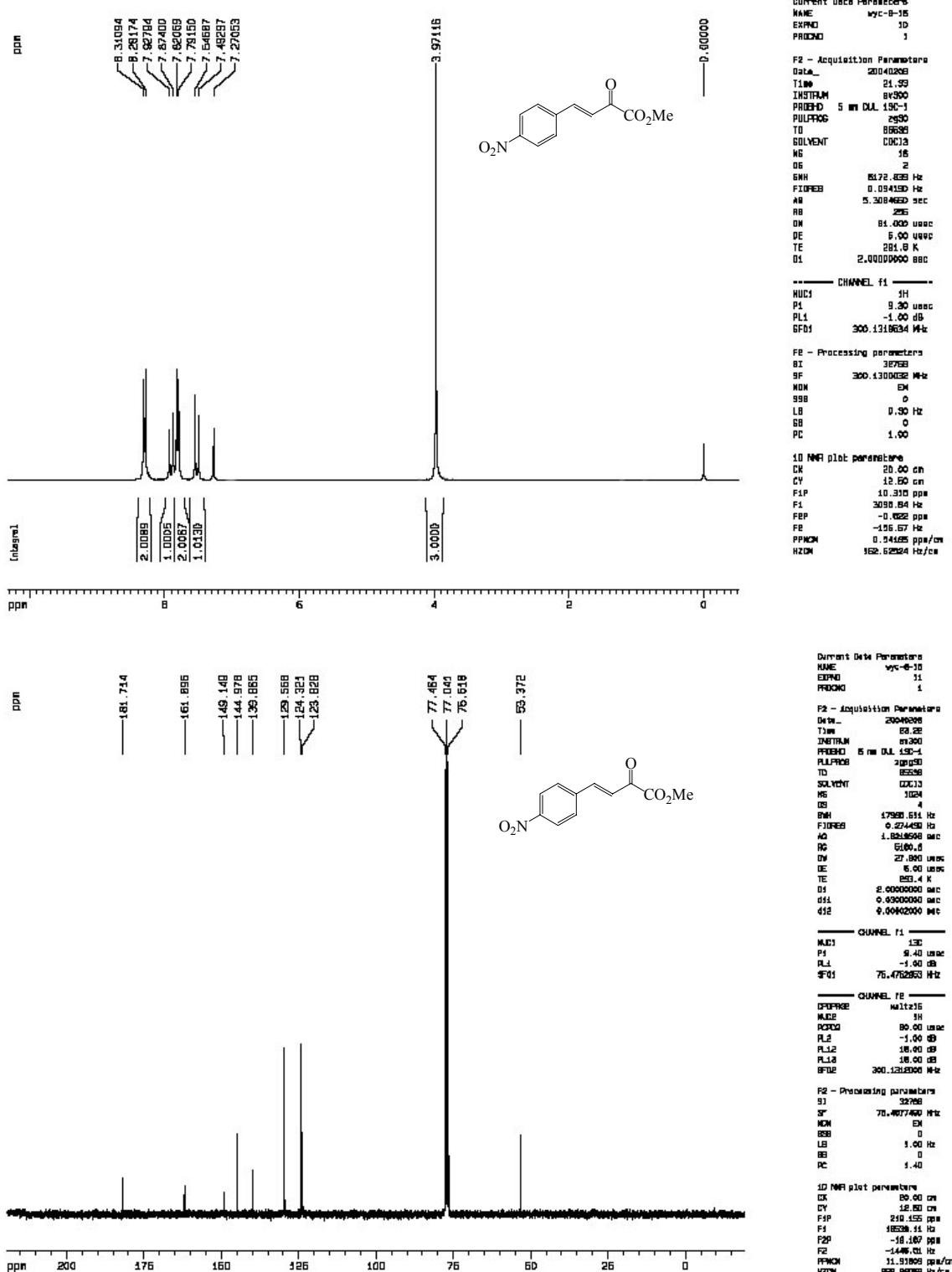
(3E)-4-(4-Methylphenyl)-2-oxobut-3-enoic acid methyl ester (2c)



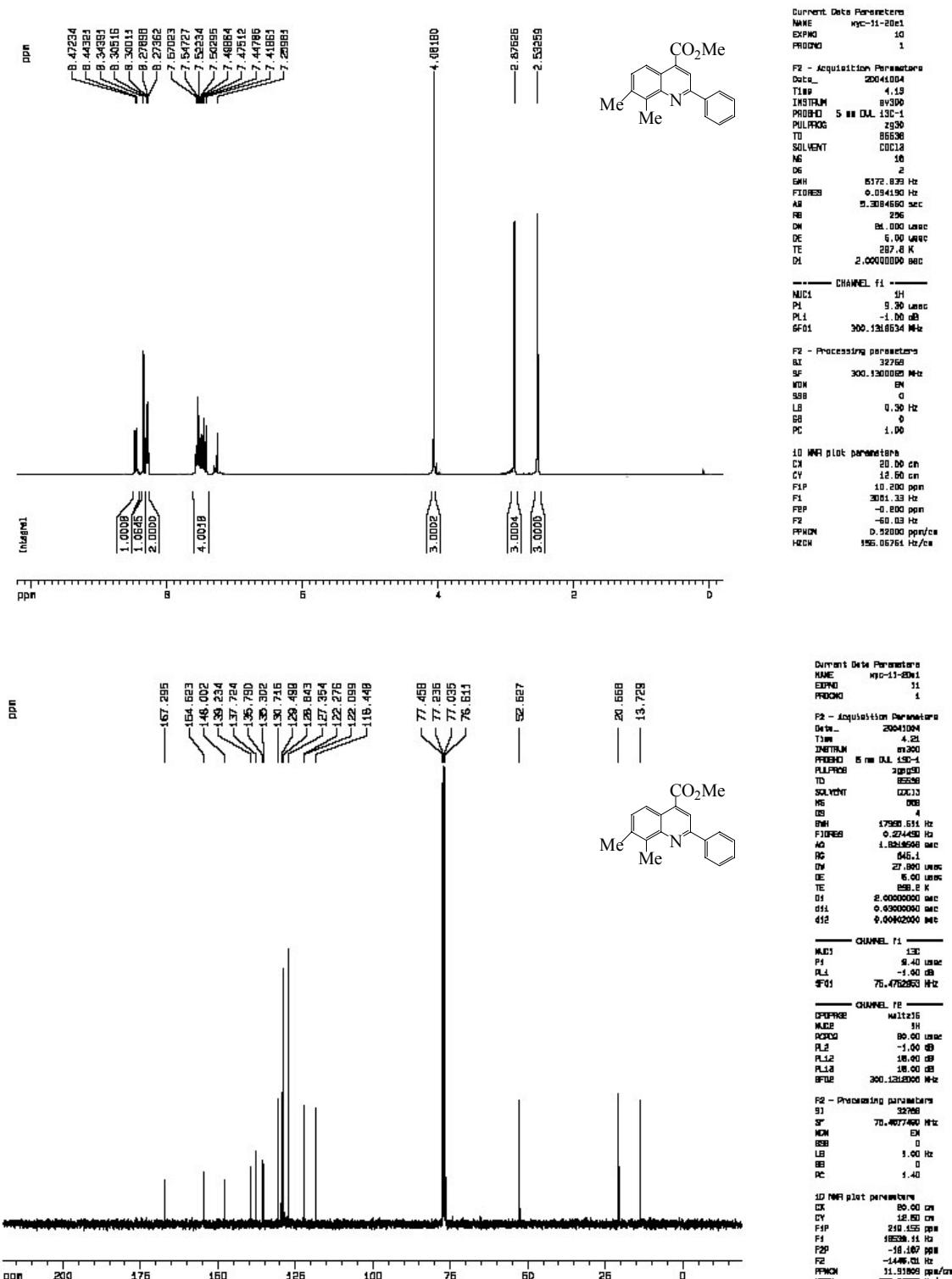
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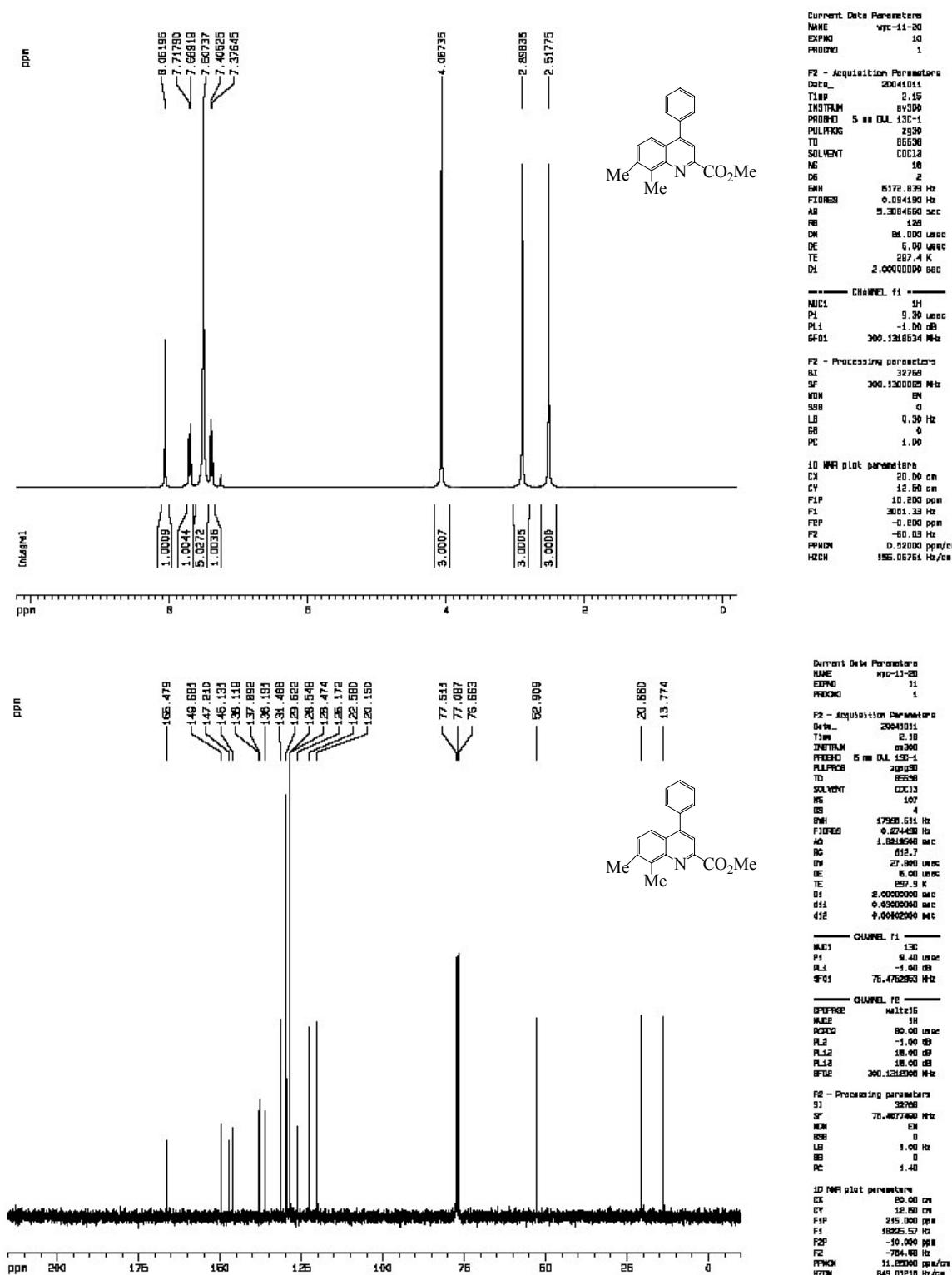
(3E)-4-(4-Nitrophenyl)-2-oxobut-3-enoic acid methyl ester (2e)



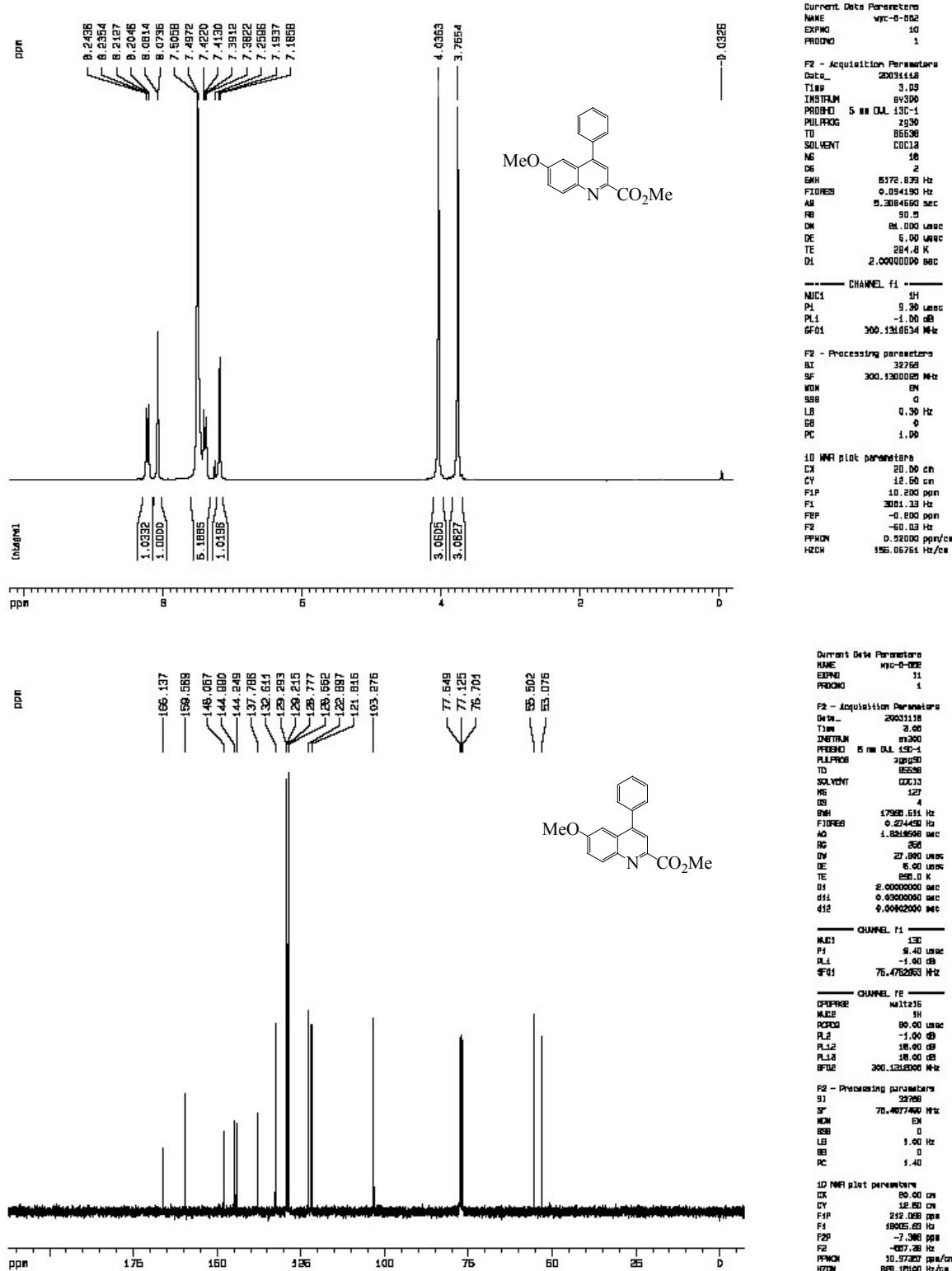
Methyl 7,8-dimethyl-2-phenylquinoline-4-carboxylate (4a)



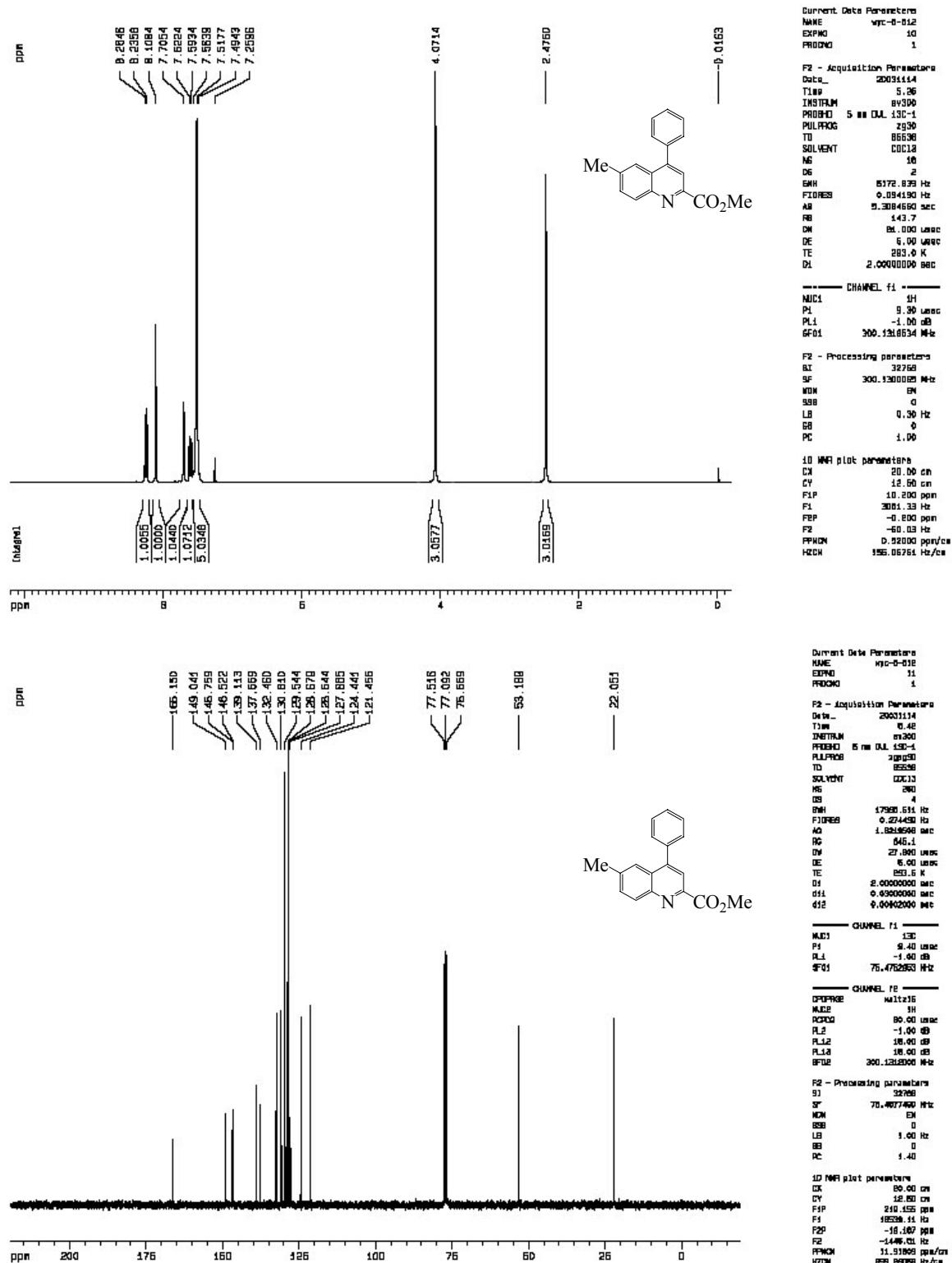
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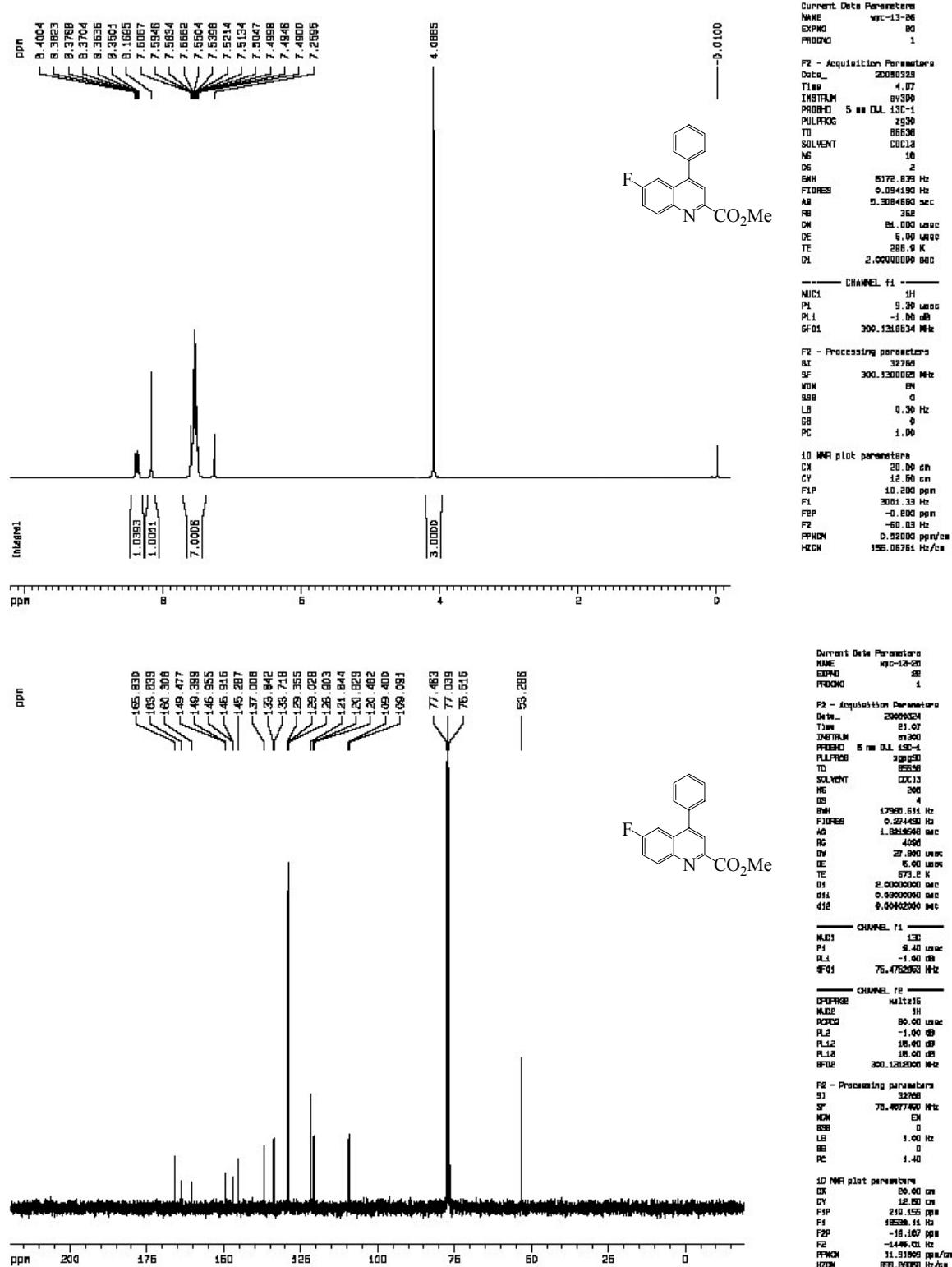
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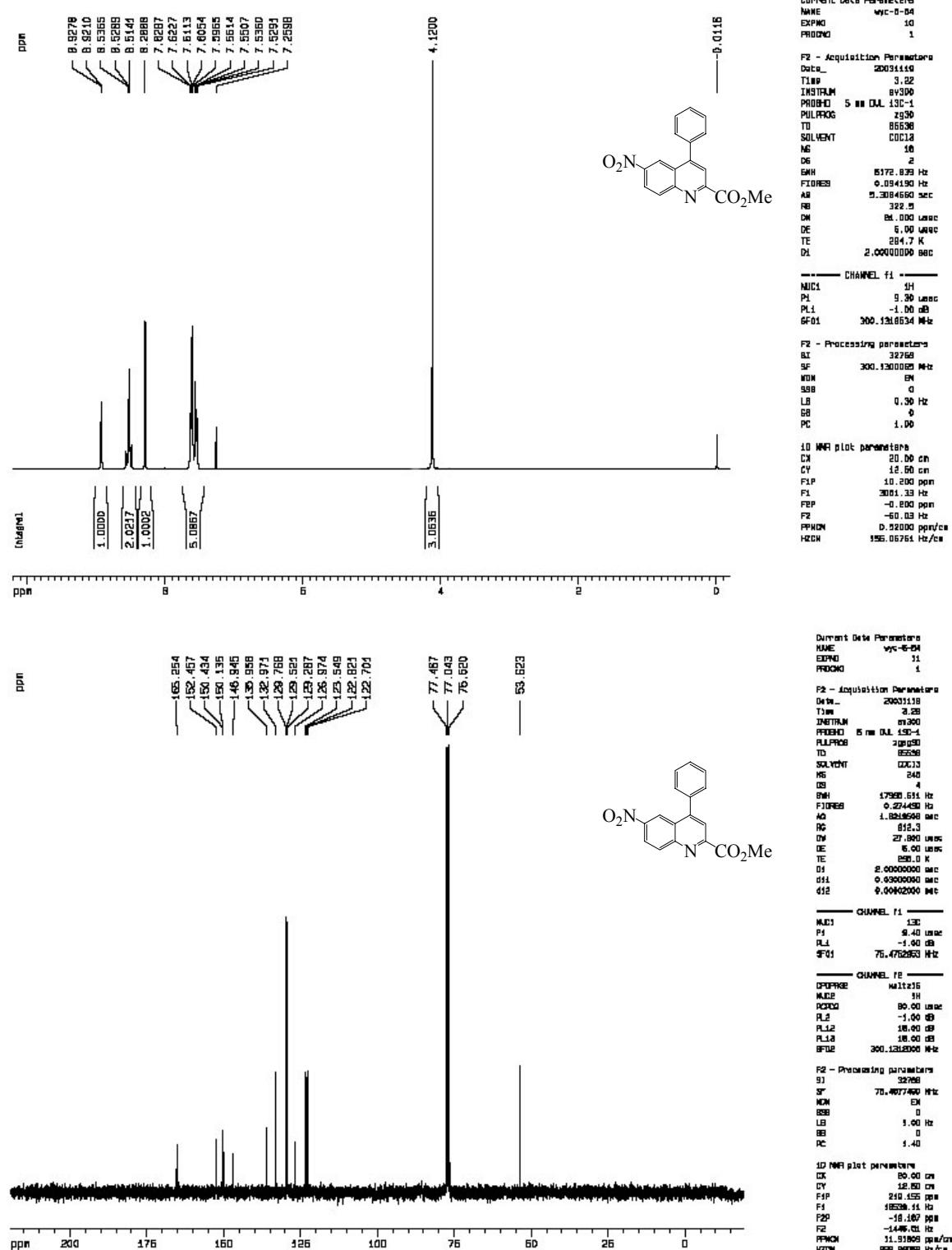
Methyl 6-methyl-4-phenylquinoline-2-carboxylate (3c)



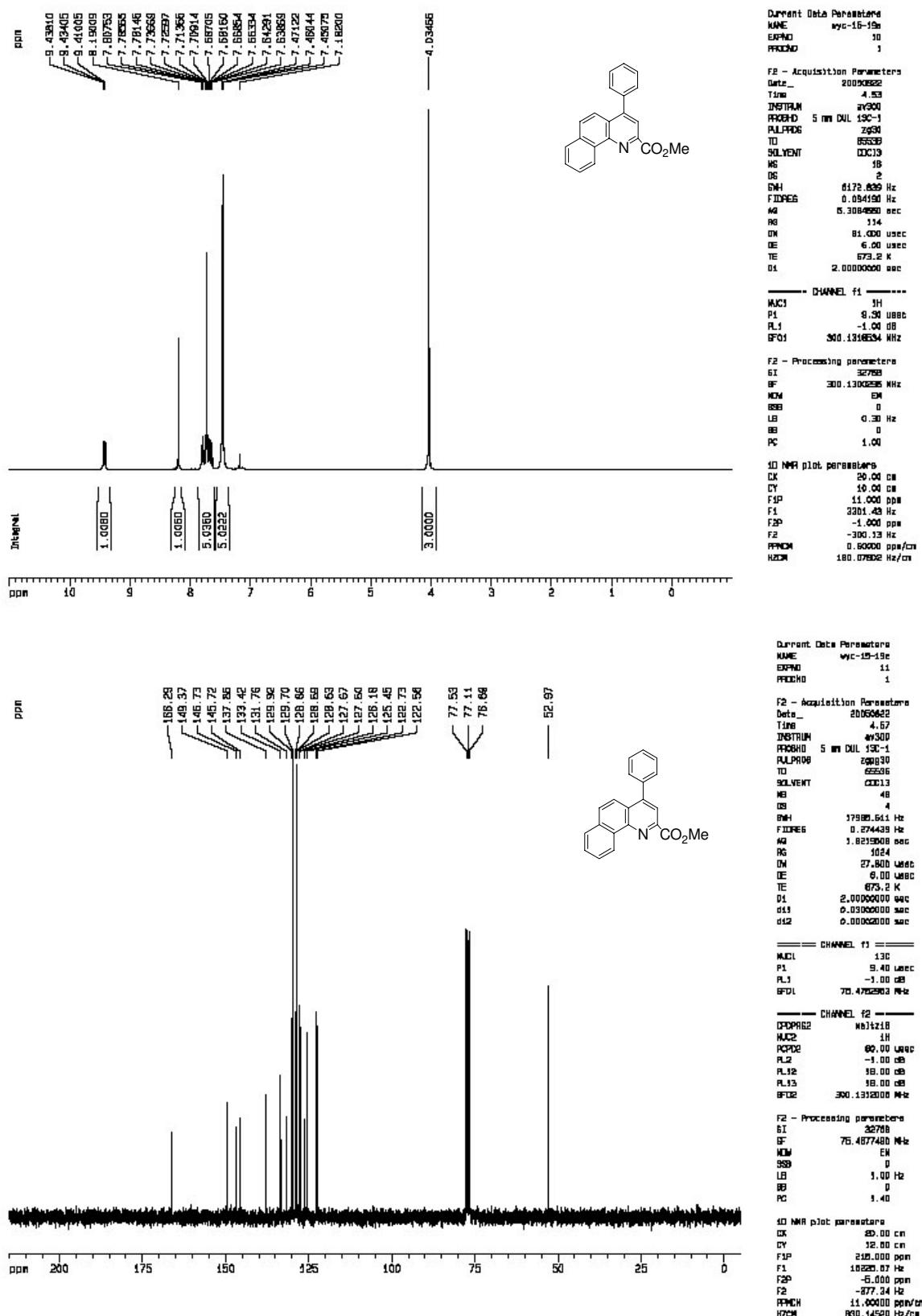
Methyl 6-fluoro-4-phenylquinoline-2-carboxylate (3d)



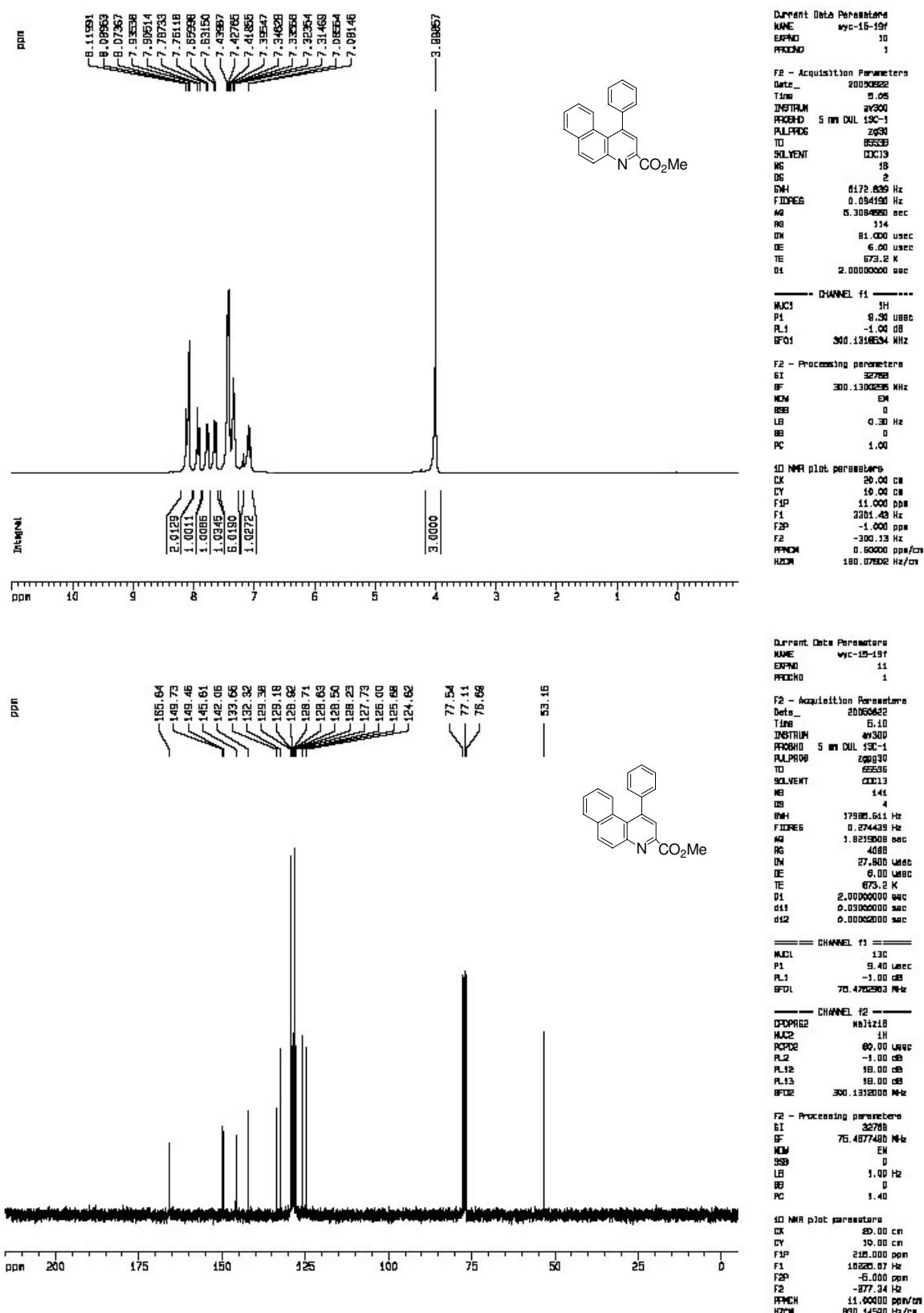
Methyl 6-nitro-4-phenylquinoline-2-carboxylate (3e)



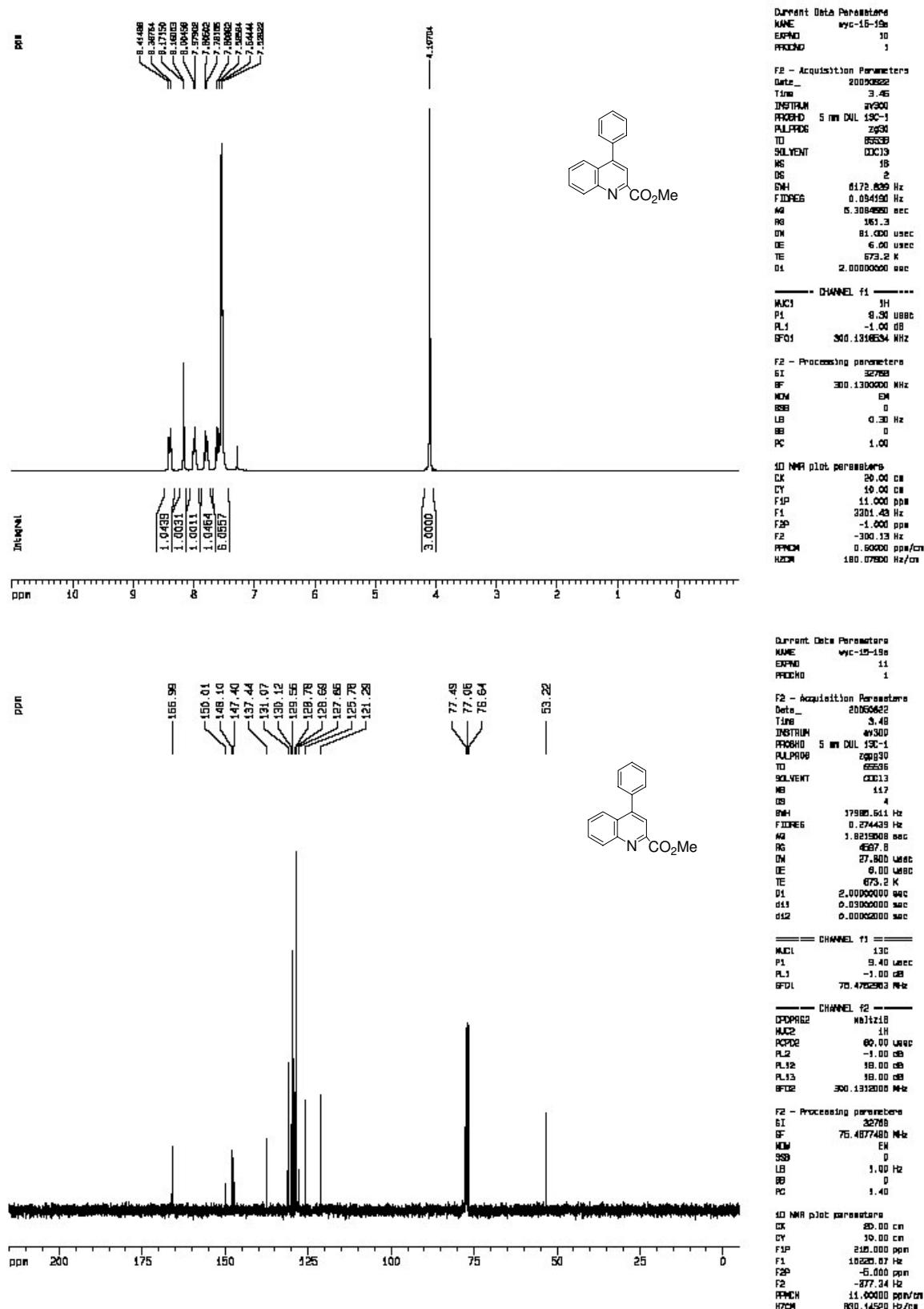
Methyl 4-phenylbenzo[*h*]quinoline-2-carboxylate (3f)



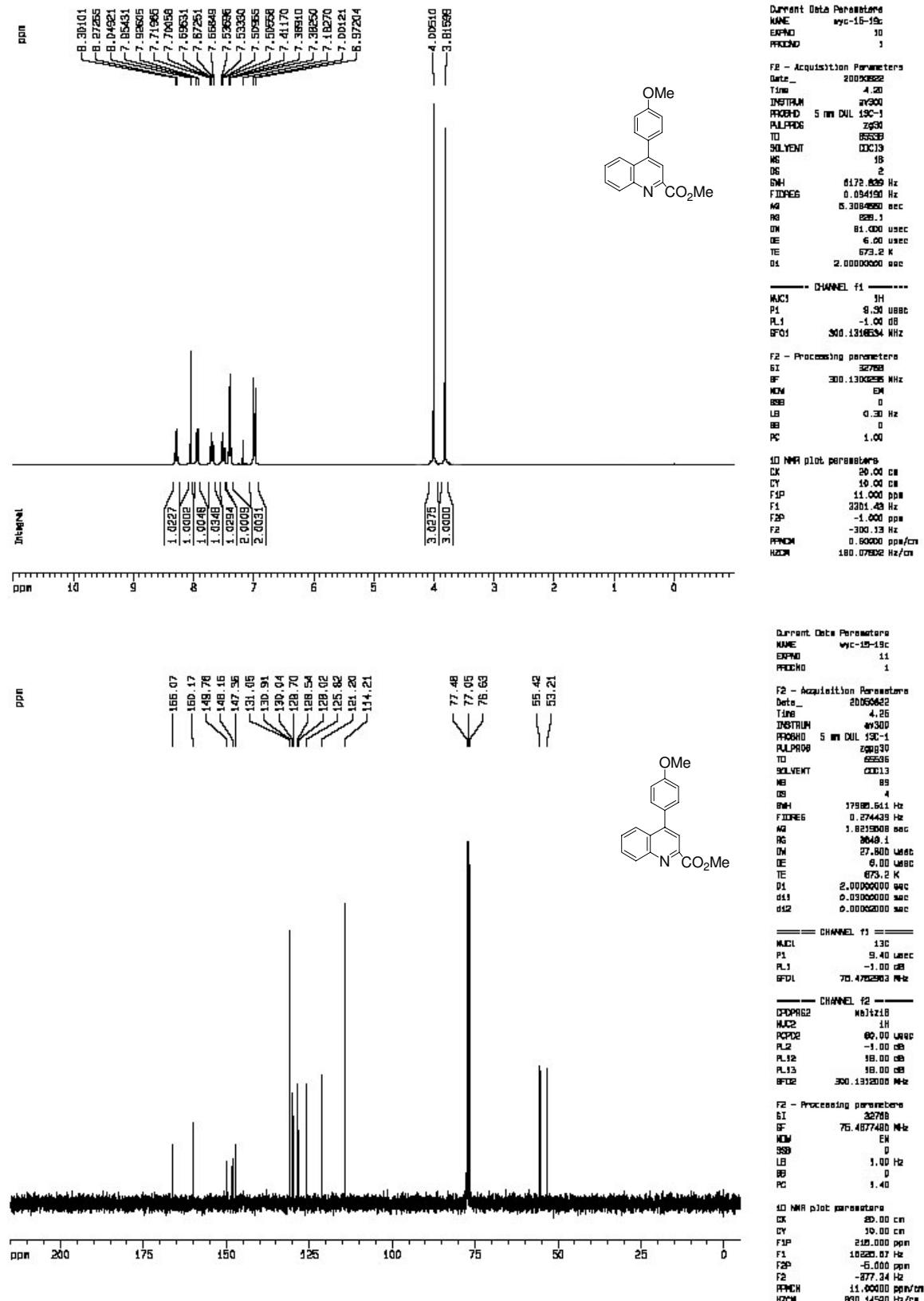
Methyl 1-phenylbenzo[f]quinoline-3-carboxylate (3g)



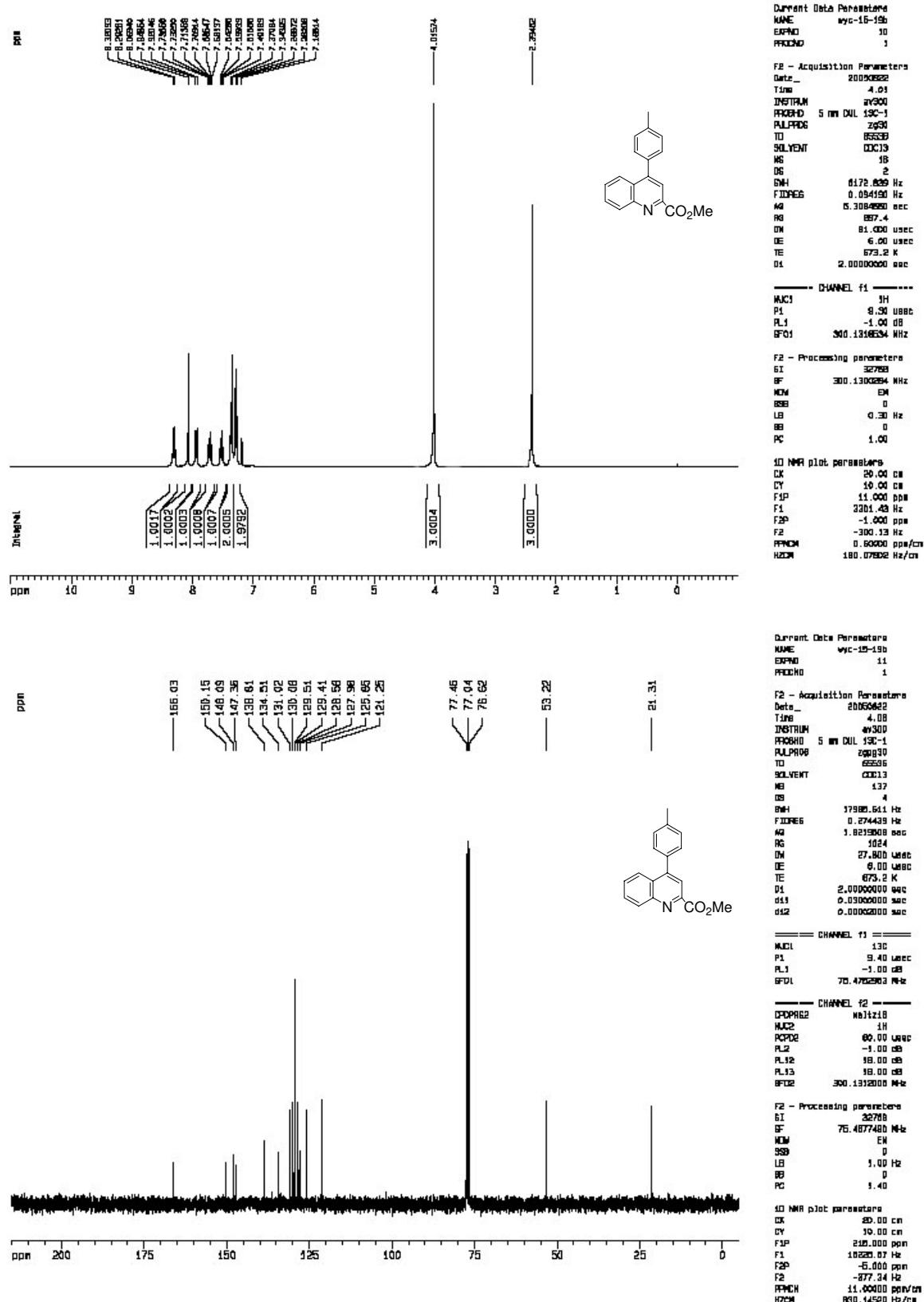
Methyl 4-phenylquinoline-2-carboxylate (3h)



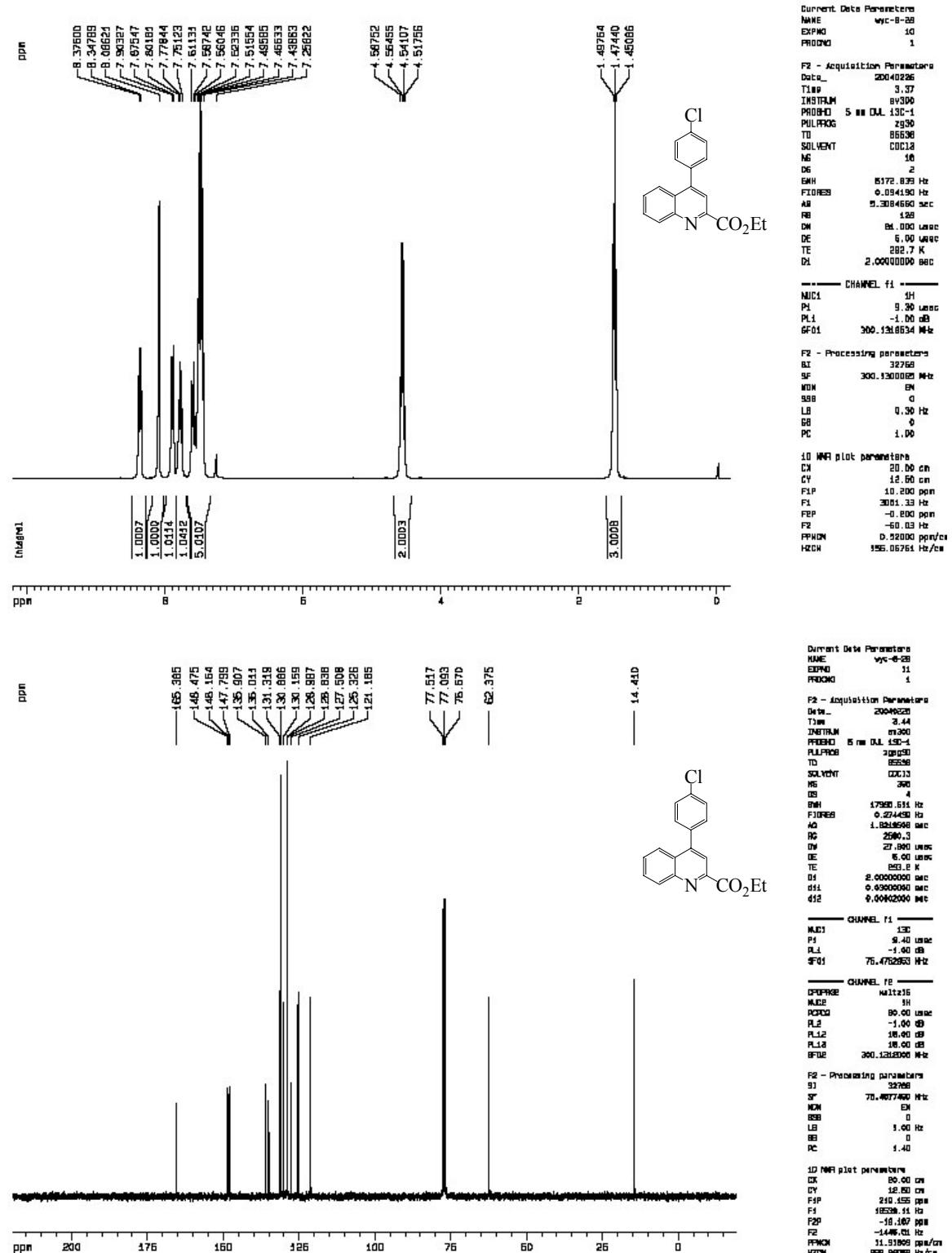
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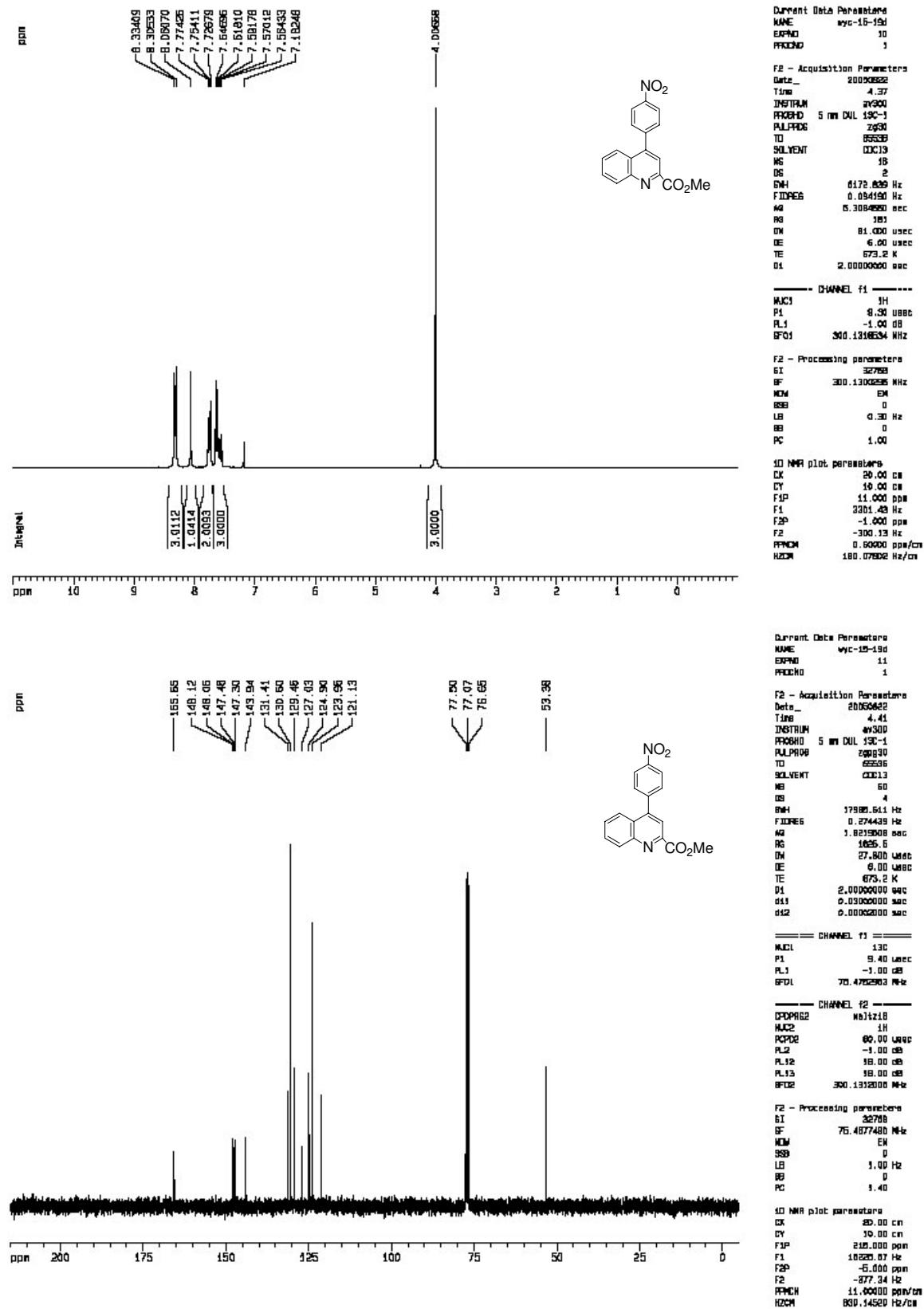
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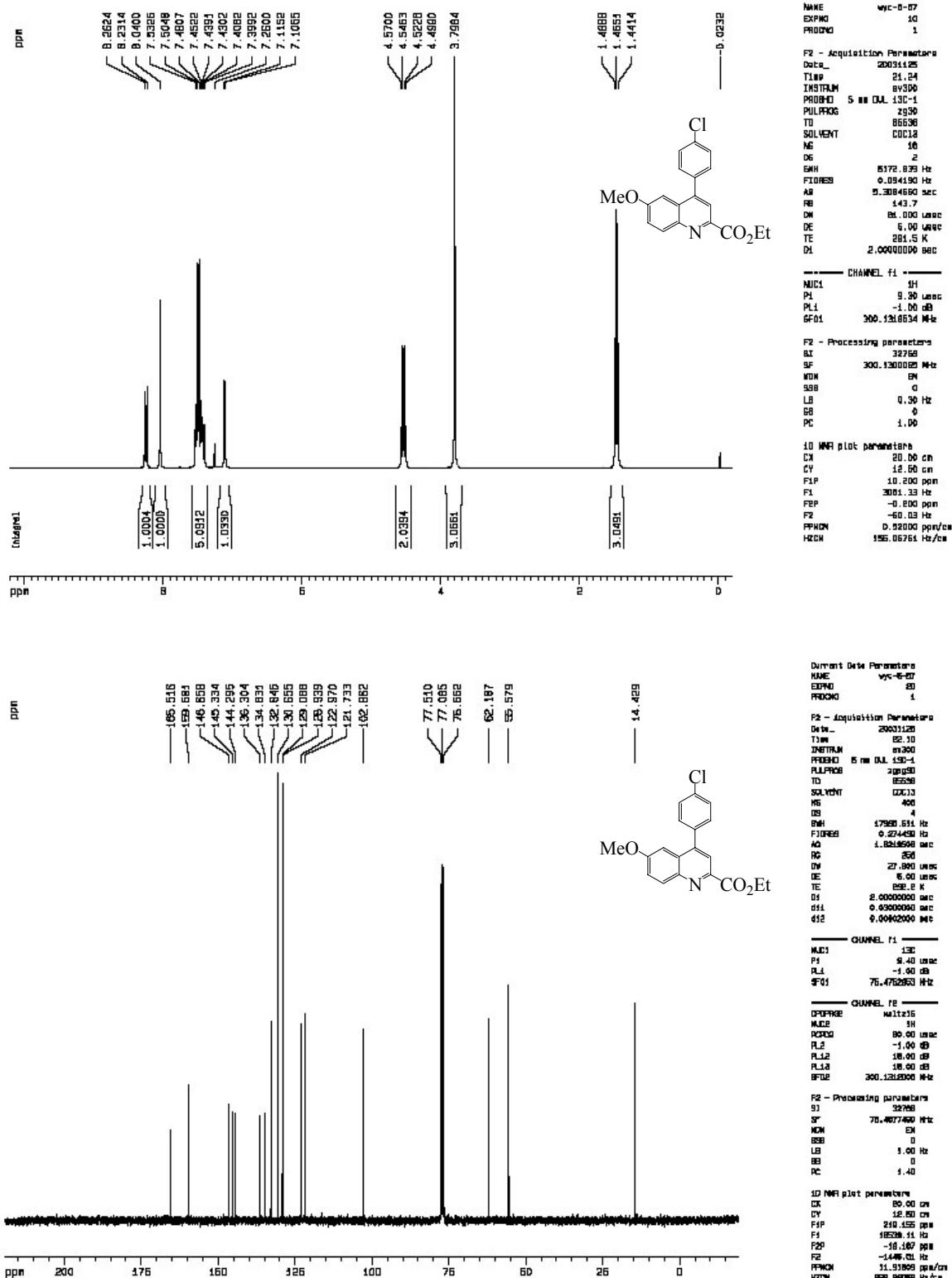
Ethyl 4-(4-chlorophenyl)quinoline-2-carboxylate (3k)



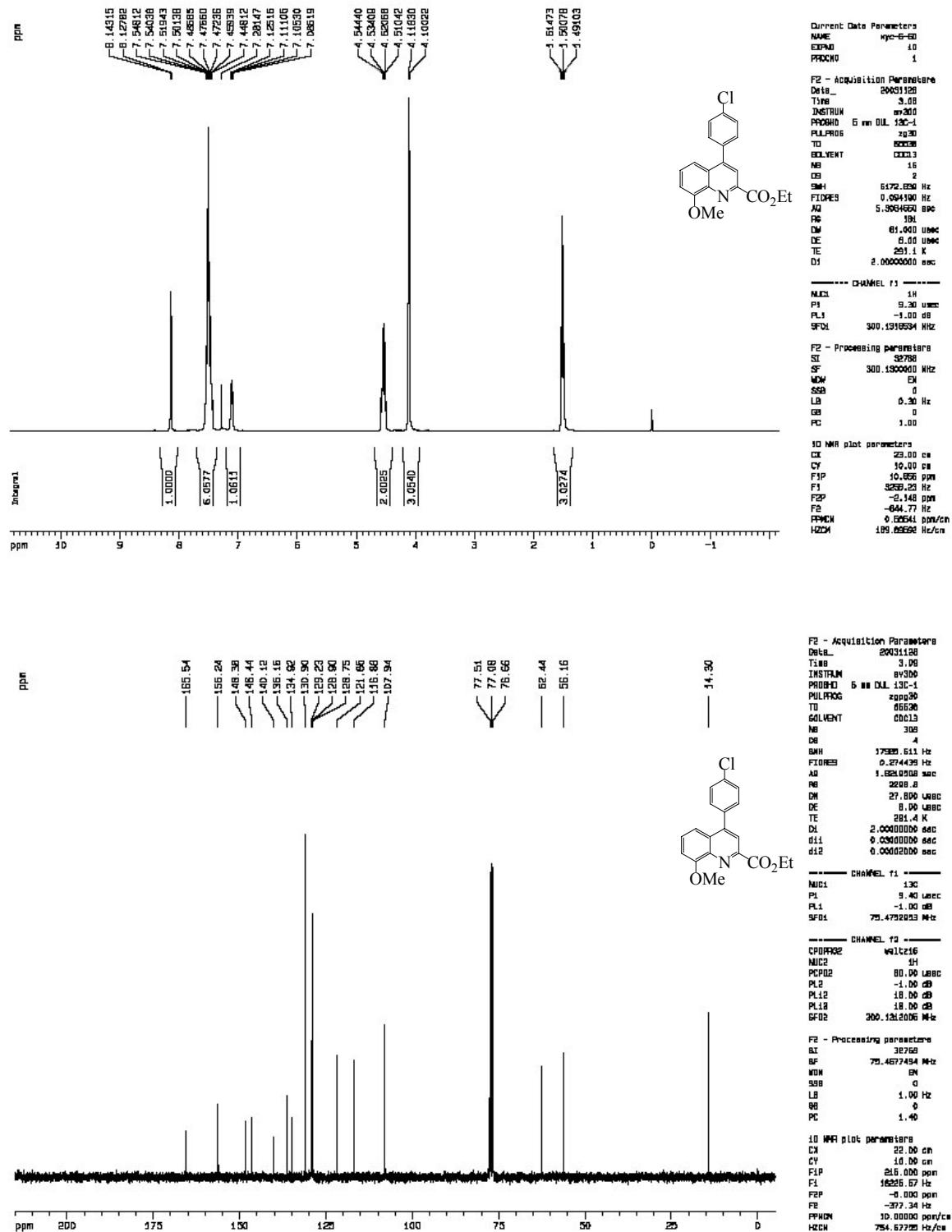
Methyl 4-(4-nitrophenyl)quinoline-2-carboxylate (3l)



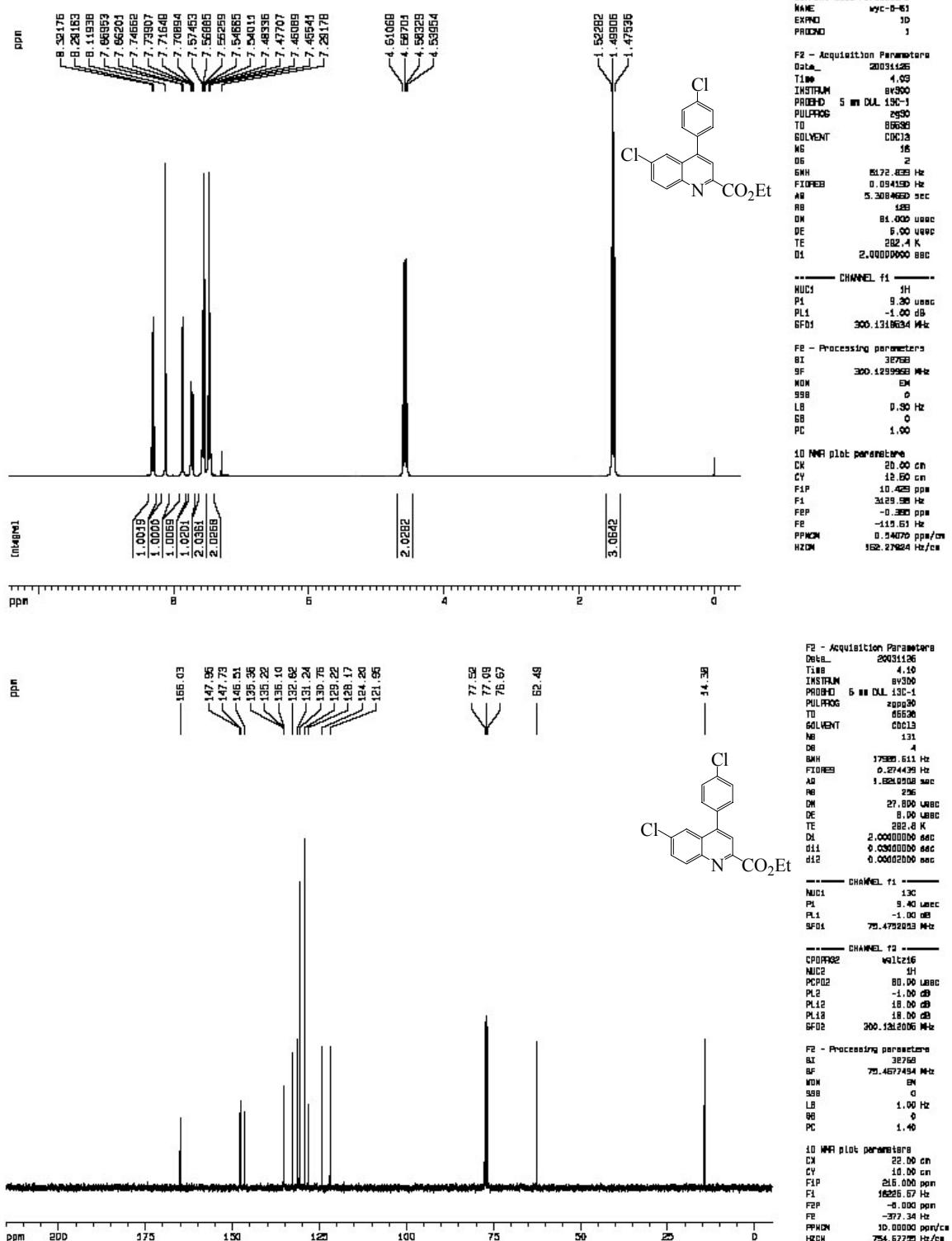
Ethyl 4-(4-chlorophenyl)-6-methoxyquinoline-2-carboxylate (3m)



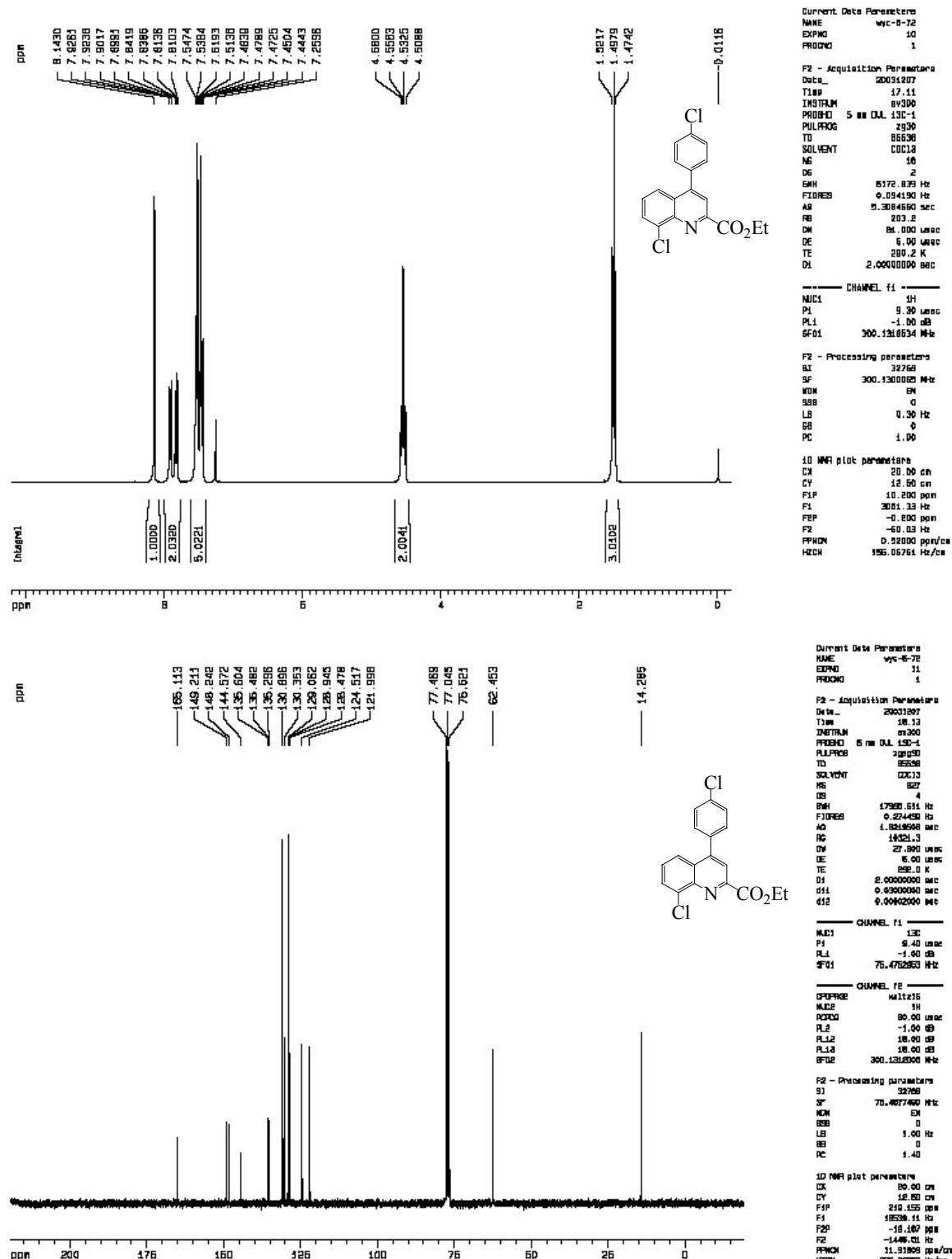
Ethyl 4-(4-chlorophenyl)-8-methoxyquinoline-2-carboxylate (3n)



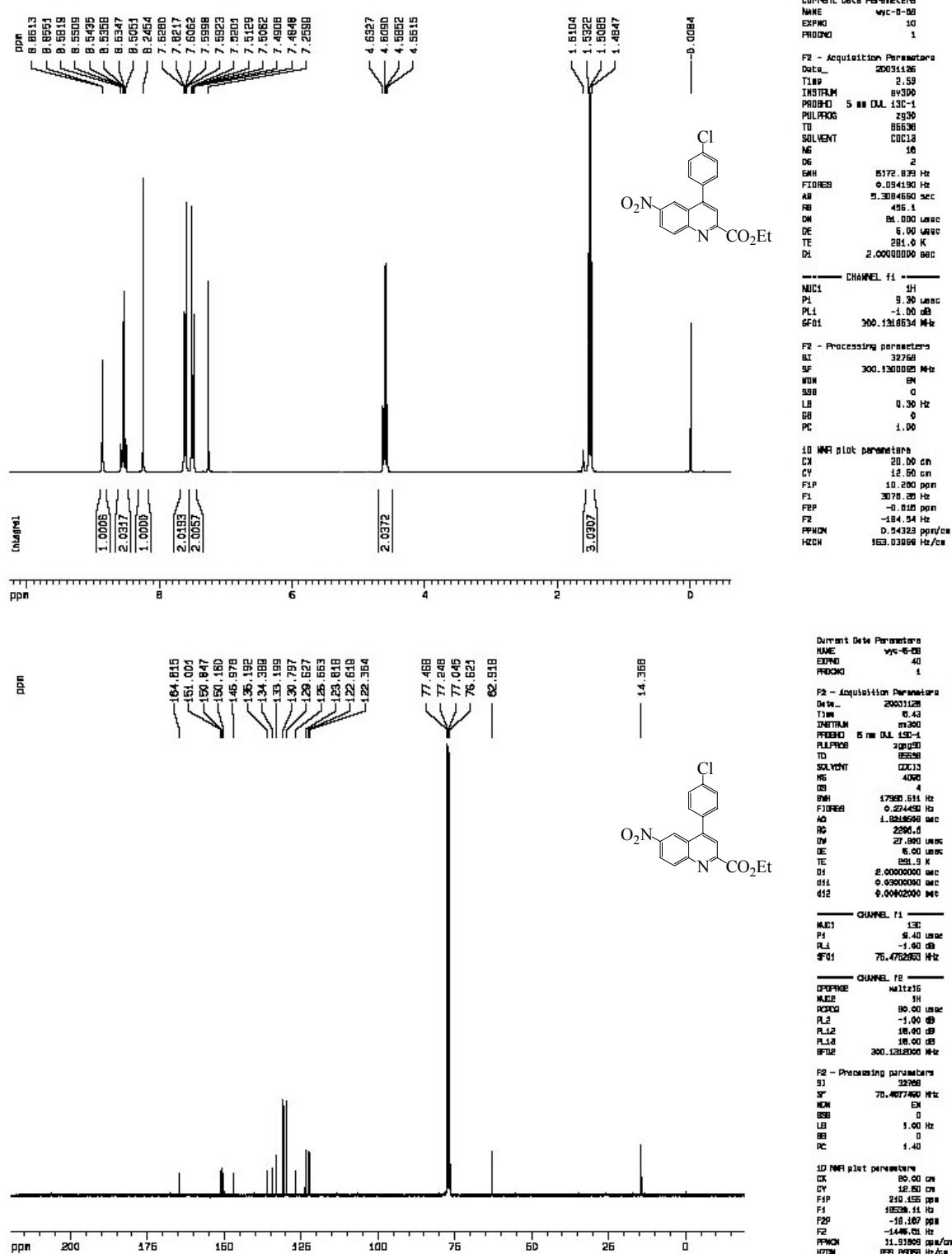
Ethyl 6-chloro-4-(4-chlorophenyl)quinoline-2-carboxylate (3o)



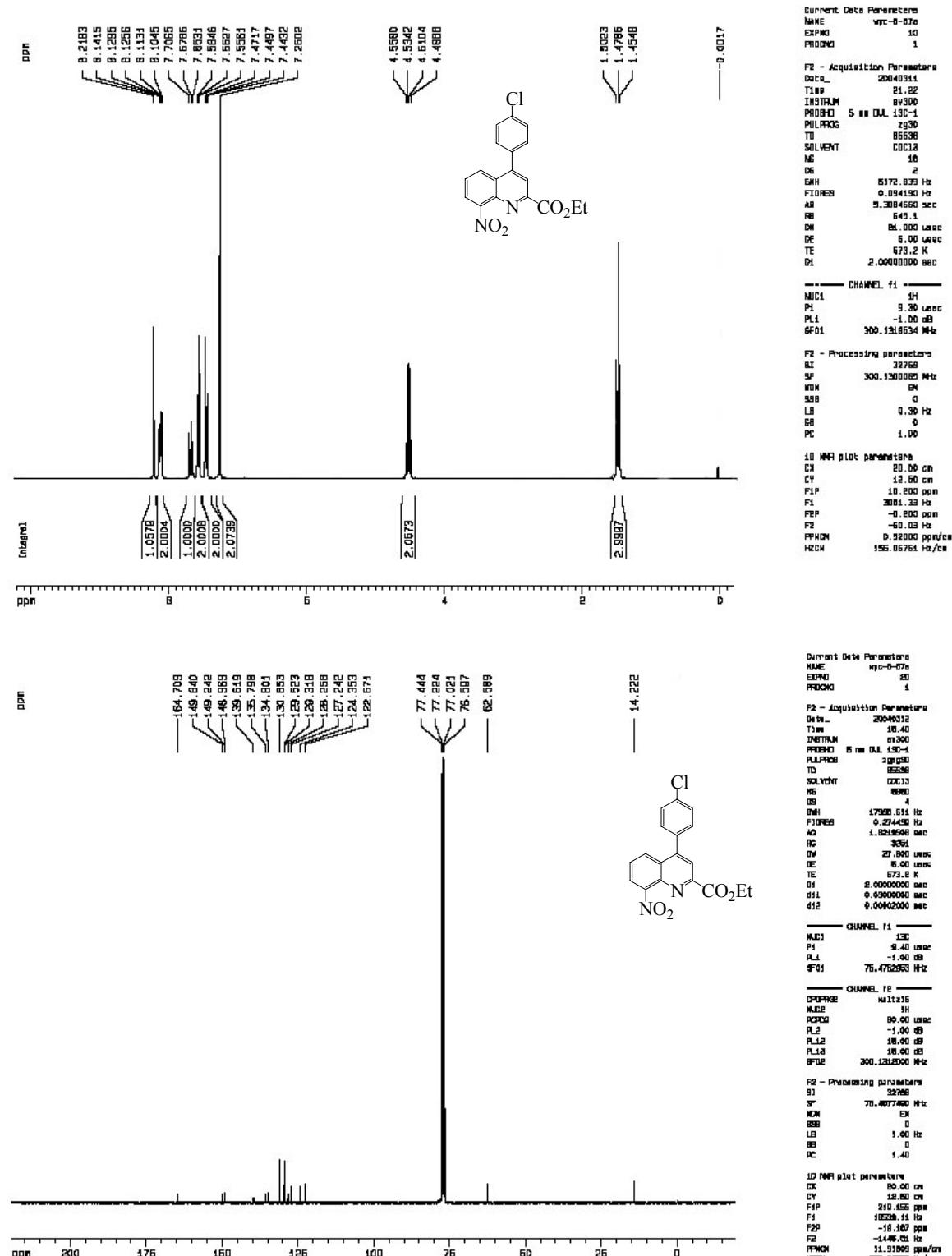
Ethyl 8-chloro-4-(4-chlorophenyl)quinoline-2-carboxylate (3p)



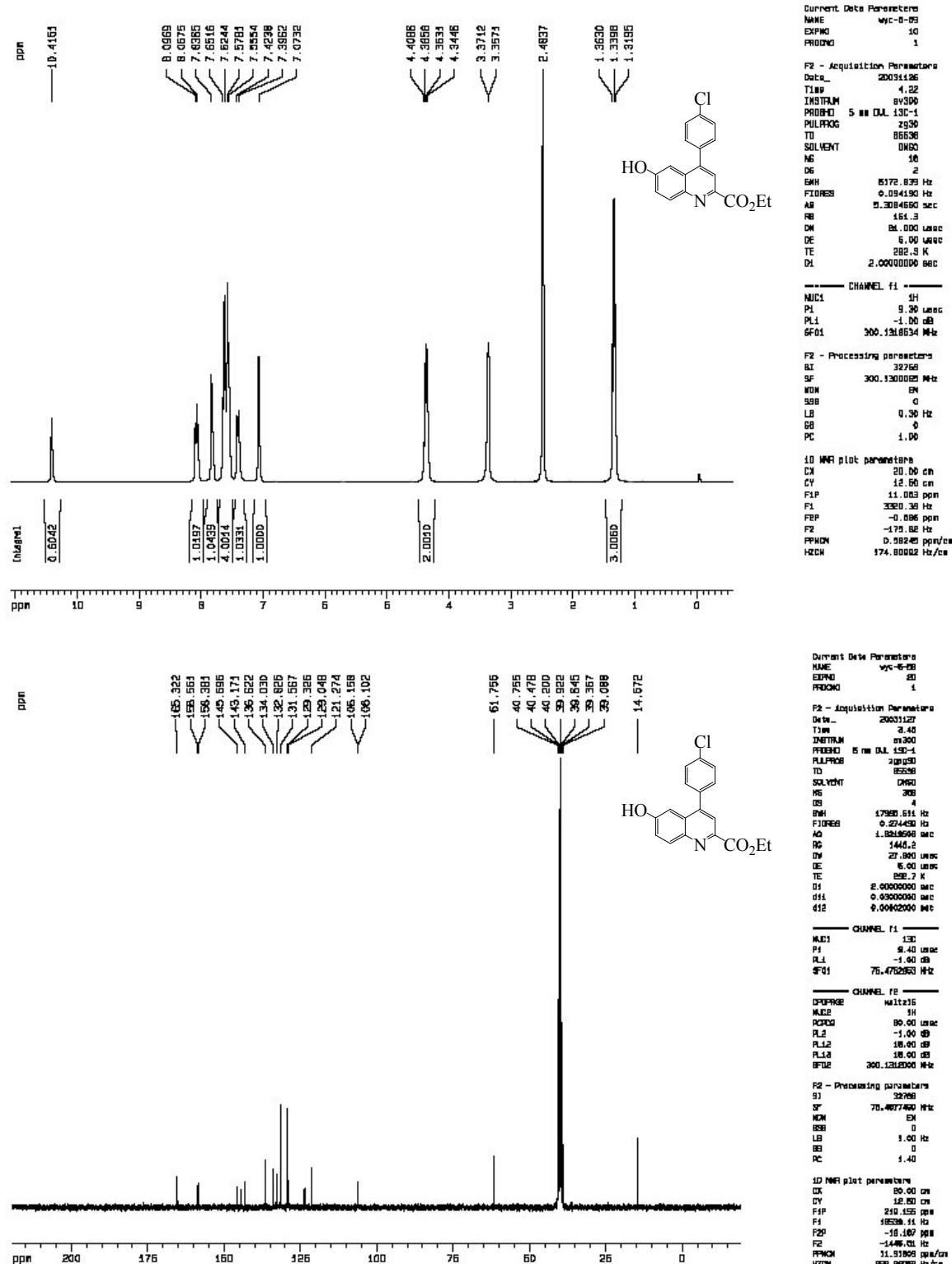
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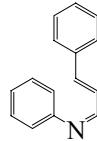
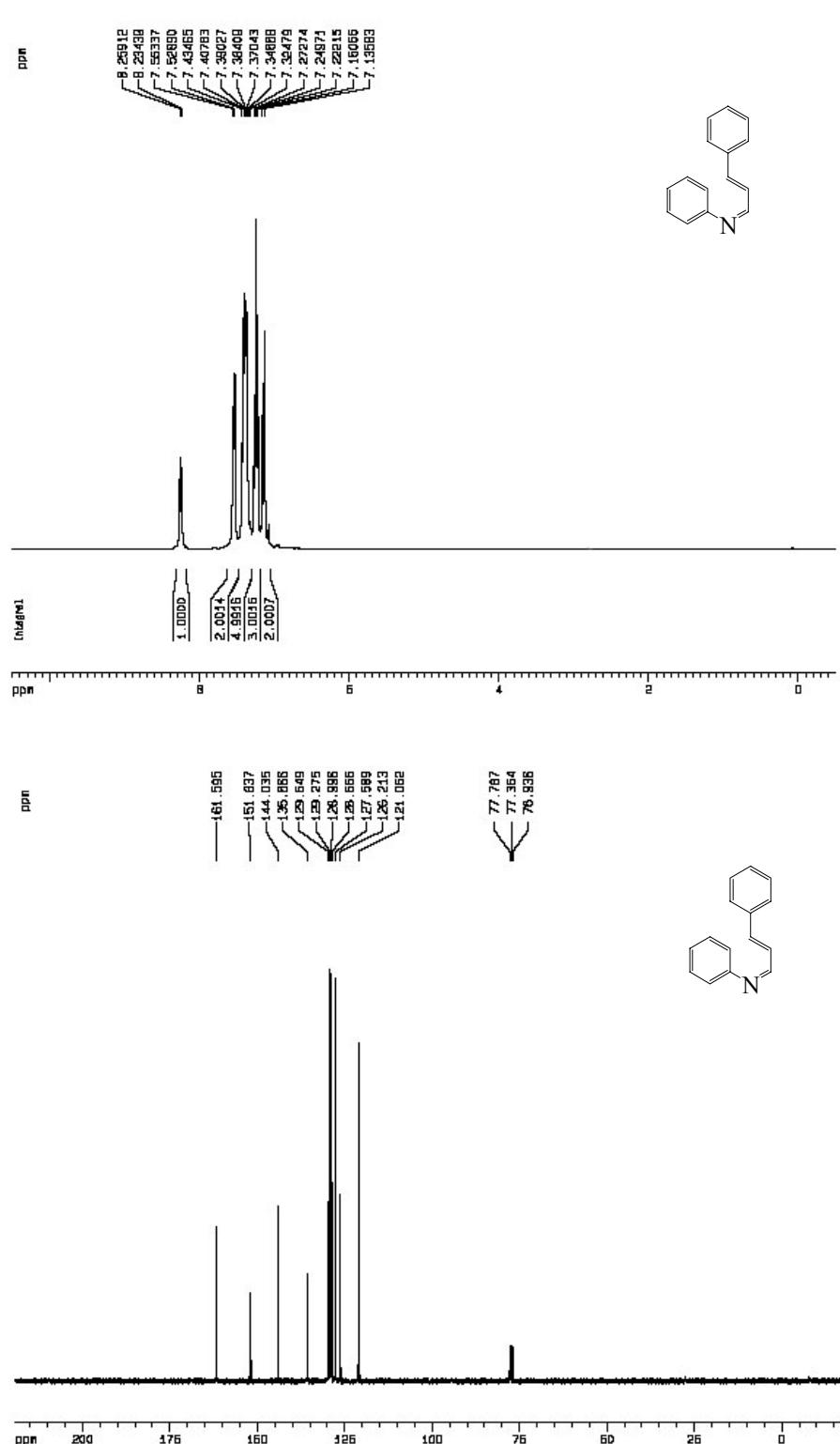
Ethyl 4-(4-chlorophenyl)-8-nitroquinoline-2-carboxylate (3r)



Ethyl 4-(4-chlorophenyl)-6-hydroxyquinoline-2-carboxylate (3s)



Schiff base 5t



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FIDRES  0.084140 Hz
AQ      9.3084660 sec
RG      40.3
DM      0x00000000
DE      6.00 usec
TE      673.2 K
D1      2.00000000000 sec
D1W    1.00000000000 sec
TD0    1.00000000000 sec
R1      1.00000000000 sec

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--- CHANNEL f1 ---
NUC1 1H
P1 9.30 sec
PL1 -1.00 dB
SE01 302 121.0634 MHz

```

F2 - Processing parameters
BI          32768
SF          300.1300000 MHz
WDM         8N
NSB         0
LB          0.30 Hz
GS         0
PC          1.00

```

10 NMR plot parameters

CX	20.00	cm
CY	8.00	cm
F1P	10.000	ppm
F1	3001.36	Hz
F2P	-0.000	ppm
F2	-150.06	Hz
PPWDM	0.55000	ppm/cm ²
HCH	165.07430	Hz/cm ²

Current Date Parameters
NAME mpc-10-8a
EDPNO 31
PRDNO 1

P2 - Acquisition Parameter	
DATA	20000700
TIME	3.00
INSTRUM	ms300
PROBOD	5 mm QUL 150-1
PULPROG	30exp30
TD	32768
SOLVENT	D2O/D3
NS	4
SW	17394.54 Hz
TE/RES	0.024400 sec
ACQ	1.0000000 sec
SC	6798.62
SWF	27.999999 sec
DE	60
TDE	573.2 KHz
D1	2.0000000 sec
D11	0.0320000 sec
D12	0.0400000 sec

CHANNEL F1

MICS	130
P1	9.40 ms
P2	-1.90 dB
P3	75.475253 Hz

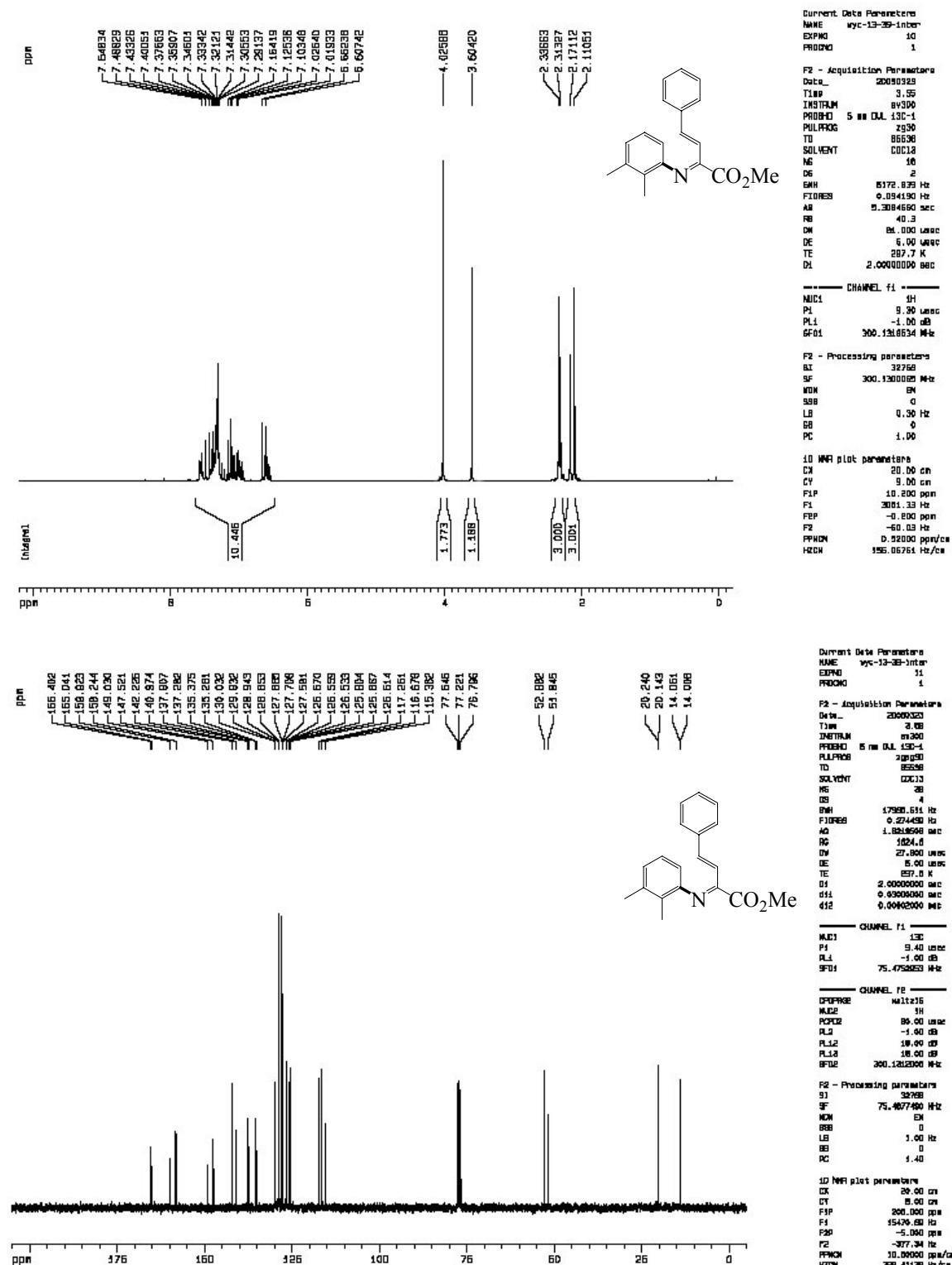
CHANNEL RE

CDP/PAKE	Melitz16
MUDE	3H
PCPDG	80.00 usd
PL2	-1.00 usd
PL12	18.00 usd
PL14	18.00 usd
Total	100.00 usd

R2 - Preprocessing parameters
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 SF 70.4077400 kHz
 NORM EN
 BBS 0
 LS 1.00 Hz
 BB 0

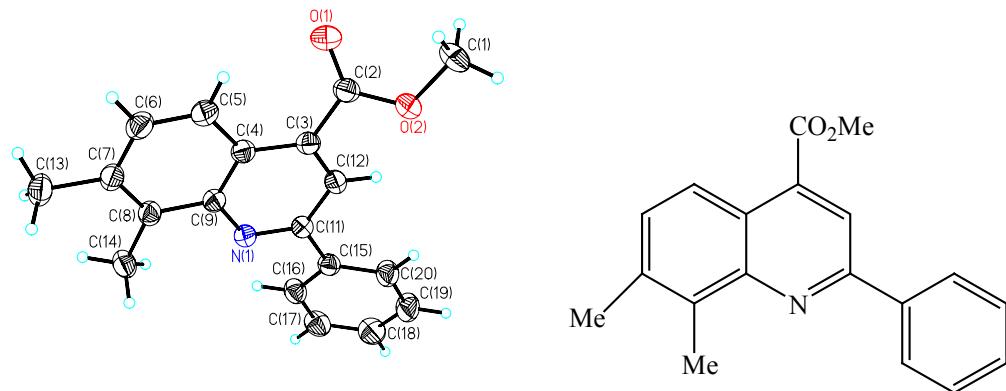
1D NMR plot parameters	
CR	80.00 cm
CY	10.00 cm
F1P	210.155 ppm
F1	1853.01 Hz
F2P	-18.167 ppm
F2	-144.01 Hz
PPMCH	11.91868 ppm/cm
PPMHC	11.91868 ppm/cm

(3E)-2-(2,3-dimethylphenylimino)-4-phenylbut-3-enoate methyl ester (5a, syn and anti)



3. Single-crystal X-ray analyses

3.1 The single-crystal X-ray analysis of **4a**



Methyl 7,8-dimethyl-2-phenylquinoline-4-carboxylate (**4a**)

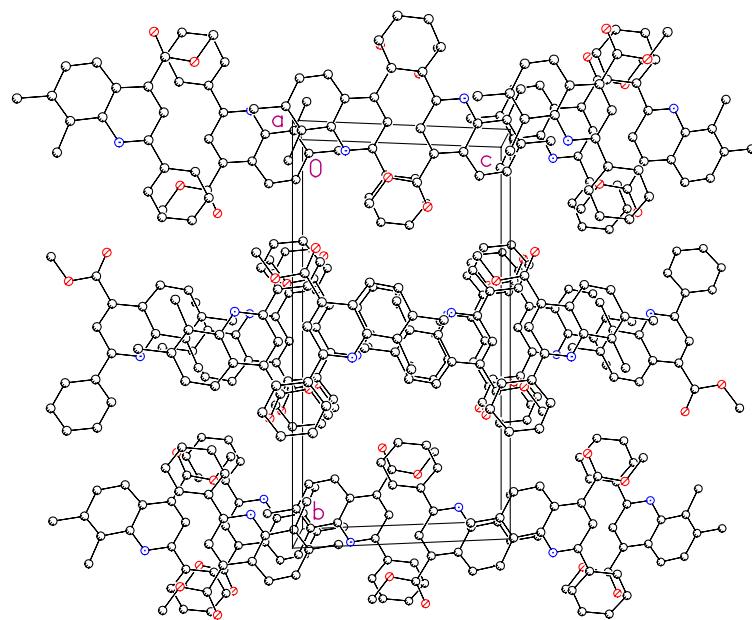


Table 1. Crystal data and structure refinement for **4a**

Identification code	4a
Empirical formula	C19 H17 N O2
Formula weight	291.34
Temperature	293(2) K
Wavelength	0.71073 Å
Crystal system, space group	Orthorhombic, Aba2
Unit cell dimensions	a = 19.877(4) Å alpha = 90 deg. b = 18.927(4) Å beta = 90 deg. c = 8.0012(16) Å gamma = 90 deg.
Volume	3010.2(10) Å^3
Z, Calculated density	8, 1.286 Mg/m^3
Absorption coefficient	0.083 mm^-1
F(000)	1232
Crystal size	0.74 x 0.30 x 0.10 mm
Theta range for data collection	2.05 to 27.38 deg.
Limiting indices	0<=h<=25, -23<=k<=0, 0<=l<=10
Reflections collected / unique	12712 / 1803 [R(int) = 0.0493]
Completeness to theta = 27.38	97.90%
Absorption correction	Empirical
Max. and min. transmission	0.9919 and 0.9407
Refinement method	Full-matrix least-squares on F^2
Data / restraints / parameters	1803 / 1 / 200
Goodness-of-fit on F^2	0.988
Final R indices [I>2sigma(I)]	R1 = 0.0474, wR2 = 0.1278
R indices (all data)	R1 = 0.0614, wR2 = 0.1322
Absolute structure parameter	-2(2)
Extinction coefficient	0.013(2)
Largest diff. peak and hole	0.241 and -0.232 e.Å^-3

Table 2. Atomic coordinates ($x \times 10^4$) and equivalent isotropic displacement parameters ($\text{Å}^2 \times 10^3$) for **4a**

U(eq) is defined as one third of the trace of the orthogonalized U_{ij} tensor.

	x	y	z	U(eq)
O(1)	6230(1)	11704(1)	4927(5)	68(1)

O(2)	5720(1)	10652(1)	4850(4)	55(1)
N(1)	6826(1)	10016(1)	5963(3)	40(1)
C(1)	5094(1)	10987(2)	4385(6)	65(1)
C(2)	6249(1)	11074(2)	5047(5)	46(1)
C(3)	6885(1)	10671(1)	5422(4)	40(1)
C(4)	7497(1)	11015(1)	5170(4)	43(1)
C(5)	8088(1)	10666(1)	5469(4)	39(1)
C(6)	8054(1)	9957(1)	6081(3)	36(1)
C(7)	7401(1)	9655(1)	6309(4)	35(1)
C(8)	7330(1)	8947(1)	6918(4)	37(1)
C(9)	7896(1)	8560(1)	7277(4)	41(1)
C(10)	8537(1)	8874(1)	7078(4)	46(1)
C(11)	8618(1)	9545(1)	6512(4)	44(1)
C(12)	7850(2)	7809(1)	7917(5)	55(1)
C(13)	6634(1)	8651(1)	7150(5)	53(1)
C(14)	8745(1)	11029(1)	5171(4)	42(1)
C(15)	9239(2)	10739(2)	4153(5)	55(1)
C(16)	9839(2)	11094(2)	3884(6)	67(1)
C(17)	9945(2)	11749(2)	4610(6)	66(1)
C(18)	9457(2)	12045(2)	5595(6)	62(1)
C(19)	8857(1)	11690(1)	5884(5)	50(1)

Table 3. Bond lengths [Å] and angles [deg] for **4a**

O(1)-C(2)	1.198(3)	N(1)-C(3)-C(4)	124.4(2)
O(2)-C(2)	1.330(3)	N(1)-C(3)-C(2)	117.8(2)
O(2)-C(1)	1.446(3)	C(4)-C(3)-C(2)	117.8(2)
N(1)-C(3)	1.318(3)	C(5)-C(4)-C(3)	119.9(2)
N(1)-C(7)	1.360(3)	C(4)-C(5)-C(6)	118.0(2)
C(2)-C(3)	1.506(3)	C(4)-C(5)-C(14)	120.1(2)
C(3)-C(4)	1.394(4)	C(6)-C(5)-C(14)	121.9(2)
C(4)-C(5)	1.369(4)	C(11)-C(6)-C(7)	118.1(2)
C(5)-C(6)	1.431(3)	C(11)-C(6)-C(5)	124.4(2)
C(5)-C(14)	1.494(3)	C(7)-C(6)-C(5)	117.6(2)
C(6)-C(11)	1.410(3)	N(1)-C(7)-C(6)	122.46(19)
C(6)-C(7)	1.429(3)	N(1)-C(7)-C(8)	117.13(19)
C(7)-C(8)	1.434(3)	C(6)-C(7)-C(8)	120.4(2)
C(8)-C(9)	1.373(3)	C(9)-C(8)-C(7)	119.3(2)

C(8)-C(13)	1.503(3)	C(9)-C(8)-C(13)	122.0(2)
C(9)-C(10)	1.415(4)	C(7)-C(8)-C(13)	118.7(2)
C(9)-C(12)	1.514(3)	C(8)-C(9)-C(10)	119.3(2)
C(10)-C(11)	1.357(4)	C(8)-C(9)-C(12)	121.5(2)
C(14)-C(15)	1.388(5)	C(10)-C(9)-C(12)	119.2(2)
C(14)-C(19)	1.394(4)	C(11)-C(10)-C(9)	122.5(2)
C(15)-C(16)	1.386(4)	C(10)-C(11)-C(6)	120.3(2)
C(16)-C(17)	1.386(6)	C(15)-C(14)-C(19)	118.9(2)
C(17)-C(18)	1.369(5)	C(15)-C(14)-C(5)	122.0(3)
C(18)-C(19)	1.388(4)	C(19)-C(14)-C(5)	119.1(3)
C(2)-O(2)-C(1)	116.6(2)	C(16)-C(15)-C(14)	120.5(3)
C(3)-N(1)-C(7)	117.61(19)	C(17)-C(16)-C(15)	120.0(3)
O(1)-C(2)-O(2)	124.3(2)	C(18)-C(17)-C(16)	120.0(3)
O(1)-C(2)-C(3)	123.2(2)	C(17)-C(18)-C(19)	120.5(3)
O(2)-C(2)-C(3)	112.5(2)	C(18)-C(19)-C(14)	120.2(3)

Symmetry transformations used to generate equivalent atoms:

Table 4. Anisotropic displacement parameters ($\text{\AA}^2 \times 10^3$) for **4a**

The anisotropic displacement factor exponent takes the form:

$$-2 \pi^2 [h^2 a^*{}^2 U_{11} + \dots + 2 h k a^* b^* U_{12}]$$

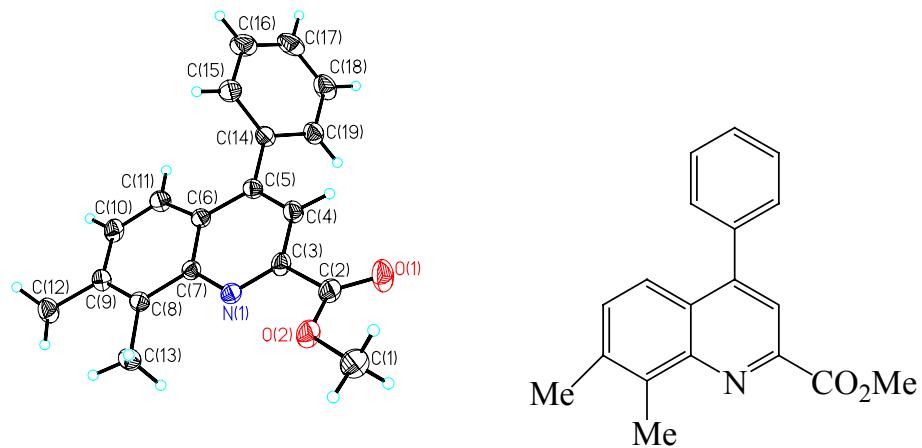
	U11	U22	U33	U23	U13	U12
O(1)	51(1)	43(1)	112(2)	7(1)	-17(2)	9(1)
O(2)	32(1)	49(1)	84(2)	1(1)	-10(1)	8(1)
N(1)	26(1)	39(1)	54(2)	1(1)	-1(1)	3(1)
C(1)	31(1)	68(2)	96(3)	13(2)	-12(2)	8(1)
C(2)	31(1)	45(1)	60(2)	2(1)	-4(1)	5(1)
C(3)	33(1)	37(1)	50(2)	-1(1)	-6(1)	7(1)
C(4)	39(1)	36(1)	53(2)	4(1)	-2(1)	1(1)
C(5)	33(1)	37(1)	46(2)	1(1)	0(1)	-2(1)
C(6)	30(1)	37(1)	41(2)	2(1)	0(1)	0(1)
C(7)	32(1)	35(1)	39(2)	-3(1)	0(1)	1(1)
C(8)	34(1)	36(1)	41(2)	-1(1)	0(1)	-2(1)
C(9)	43(1)	36(1)	46(2)	-1(1)	-4(1)	1(1)
C(10)	36(1)	43(1)	59(2)	3(1)	-9(1)	11(1)
C(11)	29(1)	46(1)	59(2)	1(1)	-5(1)	-1(1)
C(12)	56(2)	41(1)	66(2)	9(2)	-7(2)	4(1)
C(13)	41(2)	44(1)	73(2)	4(2)	4(2)	-4(1)
C(14)	32(1)	39(1)	54(2)	10(1)	-2(1)	0(1)

C(15)	43(2)	52(2)	69(2)	2(2)	6(2)	0(1)
C(16)	39(2)	76(2)	85(3)	16(2)	10(2)	-3(2)
C(17)	40(1)	71(2)	88(3)	28(2)	-3(2)	-16(2)
C(18)	49(2)	48(1)	89(3)	8(2)	-9(2)	-13(1)
C(19)	42(1)	40(1)	67(2)	4(1)	-1(1)	-3(1)

Table 5. Hydrogen coordinates ($x \times 10^4$) and isotropic displacement parameters ($\text{\AA}^2 \times 10^3$) for **4a**

	x	y	z	U(eq)
H(1B)	4749	10634	4278	98
H(1C)	5149	11227	3337	98
H(1D)	4967	11321	5231	98
H(4A)	7503	11480	4799	51
H(10A)	8917	8611	7345	55
H(11A)	9048	9733	6408	53
H(12A)	7386	7670	7985	82
H(12B)	8051	7781	9006	82
H(12C)	8083	7498	7166	82
H(13A)	6665	8176	7562	79
H(13B)	6402	8653	6099	79
H(13C)	6392	8937	7939	79
H(15A)	9166	10303	3648	66
H(16A)	10171	10892	3217	80
H(17A)	10347	11988	4428	79
H(18A)	9528	12487	6073	74
H(19A)	8528	11895	6556	60

3.2 The single-crystal X-ray analysis of **3a**



Methyl 7,8-dimethyl-4-phenylquinoline-2-carboxylate (**3a**)

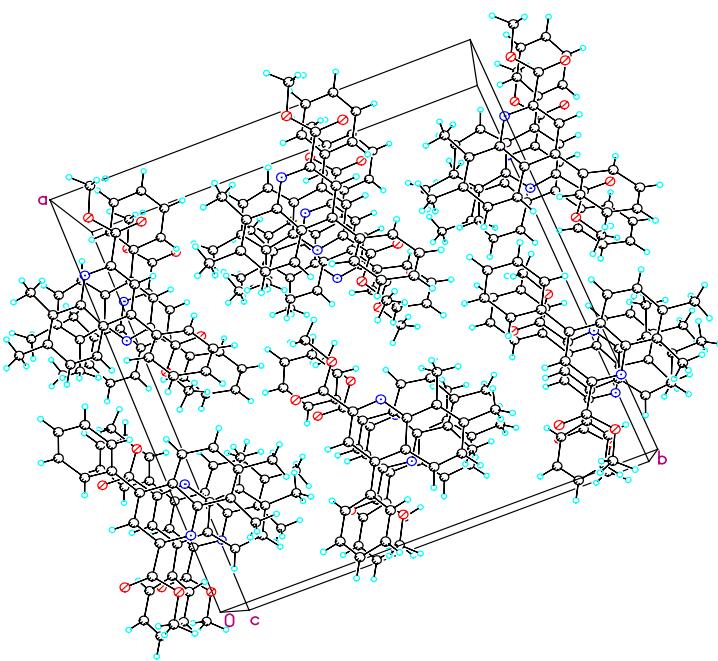


Table 1. Crystal data and structure refinement for **3a**

Identification code	3a
Empirical formula	C19 H17 N O2
Formula weight	291.34
Temperature	293(2) K
Wavelength	0.71073 Å
Crystal system, space group	Monoclinic, P2(1)/c
Unit cell dimensions	a = 7.9843(16) Å alpha = 90 deg.
	b = 19.014(4) Å beta = 109.99(3) deg.
	c = 10.550(2) Å gamma = 90 deg.
Volume	1505.1(5) Å^3
Z, Calculated density	4, 1.286 Mg/m^3
Absorption coefficient	0.083 mm^-1
F(000)	616
Crystal size	0.79 x 0.32 x 0.26 mm
Theta range for data collection	2.32 to 27.47 deg.
Limiting indices	-9<=h<=10, -22<=k<=24, -13<=l<=12
Reflections collected / unique	9208 / 3352 [R(int) = 0.0543]
Completeness to theta = 27.47	97.20%
Absorption correction	Empirical
Max. and min. transmission	0.9784 and 0.9367
Refinement method	Full-matrix least-squares on F^2
Data / restraints / parameters	3352 / 0 / 208
Goodness-of-fit on F^2	1.03
Final R indices [I>2sigma(I)]	R1 = 0.0671, wR2 = 0.1710
R indices (all data)	R1 = 0.0921, wR2 = 0.1904
Largest diff. peak and hole	0.214 and -0.260 e.Å^-3

Table 2. Atomic coordinates ($\times 10^4$) and equivalent isotropic displacement parameters ($\text{\AA}^2 \times 10^3$) for **3a**

U(eq) is defined as one third of the trace of the orthogonalized U_{ij} tensor.

	x	y	z	U(eq)
N(1)	3032(2)	5537(1)	2726(2)	45(1)
O(1)	1641(3)	3008(1)	1090(2)	76(1)
O(2)	749(2)	3708(1)	-710(2)	59(1)

C(1)	136(4)	3104(1)	-1569(3)	74(1)
C(2)	1451(3)	3586(1)	605(2)	52(1)
C(3)	1988(3)	4260(1)	1370(2)	46(1)
C(4)	2030(3)	4325(1)	2726(2)	45(1)
C(5)	1568(3)	3791(1)	3472(2)	55(1)
C(6)	1655(3)	3910(1)	4762(2)	59(1)
C(7)	2161(3)	4566(1)	5402(2)	52(1)
C(8)	2619(3)	5103(1)	4717(2)	48(1)
C(9)	2570(3)	4987(1)	3364(2)	44(1)
C(11)	2971(3)	5461(1)	1465(2)	44(1)
C(12)	2447(3)	4822(1)	759(2)	49(1)
C(13)	2137(4)	4663(2)	6818(2)	67(1)
C(14)	3154(4)	5816(1)	5337(2)	61(1)
C(15)	3489(3)	6083(1)	836(2)	47(1)
C(16)	4491(3)	6615(1)	1645(2)	58(1)
C(17)	4962(4)	7206(1)	1095(3)	69(1)
C(18)	4453(4)	7276(1)	-271(3)	77(1)
C(19)	3452(5)	6757(1)	-1099(3)	78(1)
C(20)	2984(4)	6164(1)	-546(2)	63(1)

Table 3. Bond lengths [Å] and angles [deg] for **3a**

N(1)-C(11)	1.322(2)	C(5)-C(4)-C(9)	117.93(18)
N(1)-C(9)	1.361(2)	C(5)-C(4)-C(3)	125.35(19)
O(1)-C(2)	1.200(2)	C(9)-C(4)-C(3)	116.72(17)
O(2)-C(2)	1.326(3)	C(6)-C(5)-C(4)	120.4(2)
O(2)-C(1)	1.441(3)	C(6)-C(5)-H(5A)	119.8
C(1)-H(1A)	0.96	C(4)-C(5)-H(5A)	119.8
C(1)-H(1B)	0.96	C(5)-C(6)-C(7)	122.5(2)
C(1)-H(1C)	0.96	C(5)-C(6)-H(6A)	118.8
C(2)-C(3)	1.497(3)	C(7)-C(6)-H(6A)	118.8
C(3)-C(12)	1.362(3)	C(8)-C(7)-C(6)	119.38(19)
C(3)-C(4)	1.424(3)	C(8)-C(7)-C(13)	121.7(2)
C(4)-C(5)	1.409(3)	C(6)-C(7)-C(13)	118.9(2)
C(4)-C(9)	1.424(3)	C(7)-C(8)-C(9)	119.32(19)
C(5)-C(6)	1.357(3)	C(7)-C(8)-C(14)	122.00(18)
C(5)-H(5A)	0.93	C(9)-C(8)-C(14)	118.68(18)
C(6)-C(7)	1.409(3)	N(1)-C(9)-C(4)	122.38(17)
C(6)-H(6A)	0.93	N(1)-C(9)-C(8)	117.09(17)

C(7)-C(8)	1.371(3)	C(4)-C(9)-C(8)	120.51(17)
C(7)-C(13)	1.513(3)	N(1)-C(11)-C(12)	121.78(17)
C(8)-C(9)	1.432(3)	N(1)-C(11)-C(15)	116.36(17)
C(8)-C(14)	1.503(3)	C(12)-C(11)-C(15)	121.86(17)
C(11)-C(12)	1.411(3)	C(3)-C(12)-C(11)	120.33(18)
C(11)-C(15)	1.482(3)	C(3)-C(12)-H(12A)	119.8
C(12)-H(12A)	0.93	C(11)-C(12)-H(12A)	119.8
C(13)-H(13A)	0.96	C(7)-C(13)-H(13A)	109.5
C(13)-H(13B)	0.96	C(7)-C(13)-H(13B)	109.5
C(13)-H(13C)	0.96	H(13A)-C(13)-H(13B)	109.5
C(14)-H(14A)	0.96	C(7)-C(13)-H(13C)	109.5
C(14)-H(14B)	0.96	H(13A)-C(13)-H(13C)	109.5
C(14)-H(14C)	0.96	H(13B)-C(13)-H(13C)	109.5
C(15)-C(20)	1.382(3)	C(8)-C(14)-H(14A)	109.5
C(15)-C(16)	1.387(3)	C(8)-C(14)-H(14B)	109.5
C(16)-C(17)	1.376(3)	H(14A)-C(14)-H(14B)	109.5
C(16)-H(16A)	0.93	C(8)-C(14)-H(14C)	109.5
C(17)-C(18)	1.364(4)	H(14A)-C(14)-H(14C)	109.5
C(17)-H(17A)	0.93	H(14B)-C(14)-H(14C)	109.5
C(18)-C(19)	1.377(4)	C(20)-C(15)-C(16)	117.82(19)
C(18)-H(18A)	0.93	C(20)-C(15)-C(11)	122.44(19)
C(19)-C(20)	1.378(3)	C(16)-C(15)-C(11)	119.73(18)
C(19)-H(19A)	0.93	C(17)-C(16)-C(15)	121.3(2)
C(20)-H(20A)	0.93	C(17)-C(16)-H(16A)	119.4
C(11)-N(1)-C(9)	119.42(16)	C(15)-C(16)-H(16A)	119.4
C(2)-O(2)-C(1)	116.78(18)	C(18)-C(17)-C(16)	120.0(2)
O(2)-C(1)-H(1A)	109.5	C(18)-C(17)-H(17A)	120
O(2)-C(1)-H(1B)	109.5	C(16)-C(17)-H(17A)	120
H(1A)-C(1)-H(1B)	109.5	C(17)-C(18)-C(19)	119.9(2)
O(2)-C(1)-H(1C)	109.5	C(17)-C(18)-H(18A)	120
H(1A)-C(1)-H(1C)	109.5	C(19)-C(18)-H(18A)	120
H(1B)-C(1)-H(1C)	109.5	C(20)-C(19)-C(18)	120.0(2)
O(1)-C(2)-O(2)	123.50(19)	C(20)-C(19)-H(19A)	120
O(1)-C(2)-C(3)	125.6(2)	C(18)-C(19)-H(19A)	120
O(2)-C(2)-C(3)	110.86(17)	C(19)-C(20)-C(15)	121.0(2)
C(12)-C(3)-C(4)	119.37(18)	C(19)-C(20)-H(20A)	119.5
C(12)-C(3)-C(2)	119.34(18)	C(15)-C(20)-H(20A)	119.5
C(4)-C(3)-C(2)	121.29(17)		

Symmetry transformations used to generate equivalent atoms:

Table 4. Anisotropic displacement parameters ($\text{\AA}^2 \times 10^3$) for **3a**

The anisotropic displacement factor exponent takes the form:

$$-2 \pi^2 [h^2 a^*^2 U_{11} + \dots + 2 h k a^* b^* U_{12}]$$

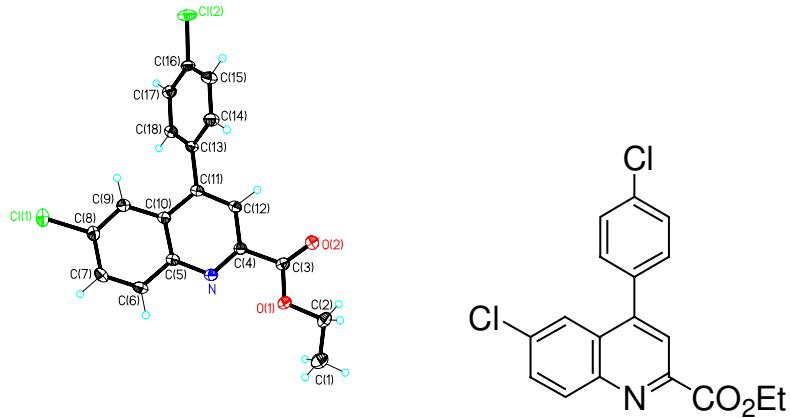
	U11	U22	U33	U23	U13	U12
N(1)	49(1)	43(1)	43(1)	0(1)	15(1)	1(1)
O(1)	115(2)	40(1)	71(1)	2(1)	28(1)	1(1)
O(2)	78(1)	44(1)	52(1)	-9(1)	18(1)	-2(1)
C(1)	91(2)	53(1)	71(2)	-25(1)	21(2)	-5(1)
C(2)	59(1)	43(1)	56(1)	-3(1)	22(1)	2(1)
C(3)	49(1)	39(1)	47(1)	1(1)	15(1)	2(1)
C(4)	46(1)	43(1)	46(1)	3(1)	16(1)	4(1)
C(5)	64(1)	46(1)	56(1)	5(1)	21(1)	-2(1)
C(6)	66(1)	57(1)	58(1)	14(1)	26(1)	4(1)
C(7)	50(1)	63(1)	45(1)	8(1)	18(1)	10(1)
C(8)	45(1)	55(1)	42(1)	1(1)	13(1)	5(1)
C(9)	44(1)	46(1)	43(1)	1(1)	13(1)	3(1)
C(11)	47(1)	43(1)	42(1)	0(1)	16(1)	2(1)
C(12)	61(1)	45(1)	41(1)	-2(1)	20(1)	0(1)
C(13)	69(2)	87(2)	48(1)	10(1)	25(1)	9(1)
C(14)	72(2)	62(1)	51(1)	-11(1)	24(1)	-4(1)
C(15)	53(1)	43(1)	49(1)	0(1)	21(1)	0(1)
C(16)	72(2)	52(1)	53(1)	-6(1)	26(1)	-11(1)
C(17)	87(2)	50(1)	77(2)	-8(1)	38(1)	-16(1)
C(18)	109(2)	53(1)	81(2)	8(1)	50(2)	-11(1)
C(19)	119(2)	66(2)	55(1)	10(1)	37(2)	-8(2)
C(20)	86(2)	53(1)	49(1)	1(1)	24(1)	-9(1)

Table 5. Hydrogen coordinates ($x \times 10^4$) and isotropic displacement parameters ($\text{\AA}^2 \times 10^3$) for **3a**

	x	y	z	U(eq)
H(1A)	-527	2800	-1188	101(11)
H(1B)	1142	2853	-1644	122(13)
H(1C)	-615	3255	-2447	113(11)
H(5A)	1201	3354	3079	66
H(6A)	1371	3546	5241	71
H(12A)	2416	4785	-127	58

H(13A)	2044	4212	7199	113(11)
H(13B)	1134	4948	6792	107(11)
H(13C)	3217	4890	7362	93(10)
H(14A)	2136	6121	5074	190(20)
H(14B)	4061	6009	5032	130(13)
H(14C)	3608	5773	6302	125(12)
H(16A)	4852	6570	2578	69
H(17A)	5626	7559	1654	82
H(18A)	4782	7673	-644	92
H(19A)	3092	6807	-2031	93
H(20A)	2319	5814	-1111	75

3.3 The single-crystal X-ray analysis of **3o**



Ethyl 6-chloro-4-(4-chlorophenyl)quinoline-2-carboxylate (**3o**)

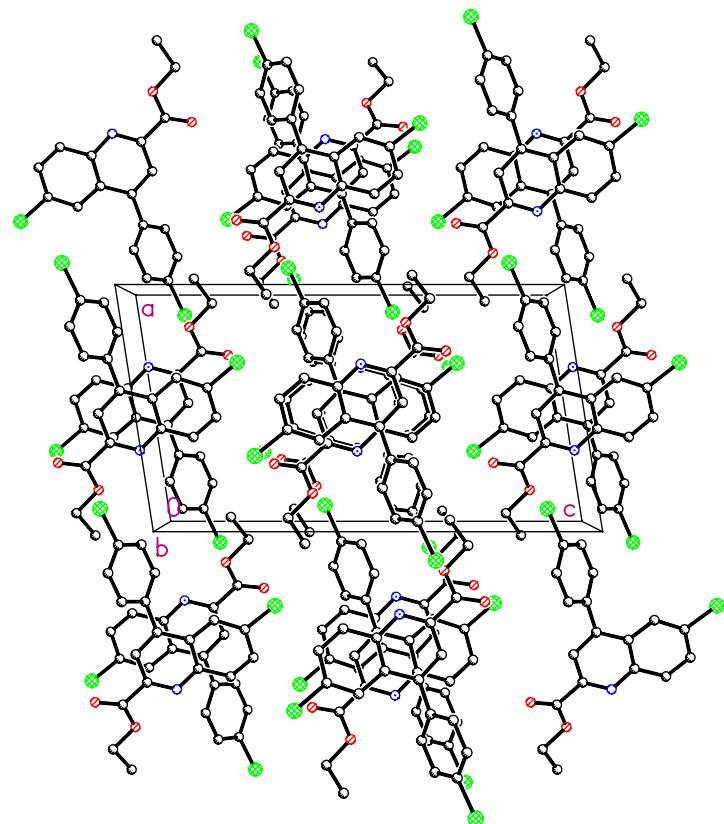


Table 1. Crystal data and structure refinement for **3o**

Identification code	3o
Empirical formula	C18 H13 Cl2 N O2
Formula weight	346.19
Temperature	293(2) K
Wavelength	0.71073 Å
Crystal system, space group	monoclinic, P2(1)/c
Unit cell dimensions	a = 10.3615(4) Å alpha = 90 deg. b = 8.4311(4) Å beta = 98.8060(7) deg. c = 18.5887(6) Å gamma = 90 deg.
Volume	1604.75(11) Å^3
Z, Calculated density	4, 1.433 Mg/m^3
Absorption coefficient	0.413 mm^-1
F(000)	712
Crystal size	0.47 x 0.40 x 0.31 mm
Theta range for data collection	2.22 to 27.44 deg.
Limiting indices	0<=h<=13, 0<=k<=10, -24<=l<=23
Reflections collected / unique	3660 / 3660 [R(int) = 0.0000]
Completeness to theta = 27.44	99.80%
Max. and min. transmission	0.8820 and 0.8297
Refinement method	Full-matrix least-squares on F^2
Data / restraints / parameters	3660 / 0 / 208
Goodness-of-fit on F^2	0.824
Final R indices [I>2sigma(I)]	R1 = 0.0382, wR2 = 0.0987
R indices (all data)	R1 = 0.0633, wR2 = 0.1073
Largest diff. peak and hole	0.277 and -0.330 e.Å^-3

Table 2. Atomic coordinates (x 10^4) and equivalent isotropic displacement parameters (Å^2 x 10^3) for **3o** U(eq) is defined as one third of the trace of the orthogonalized Uij tensor.

	x	y	z	U(eq)
Cl(1)	13123(1)	1777(1)	-2407(1)	67(1)
Cl(2)	9092(1)	507(1)	1134(1)	52(1)
O(1)	18448(1)	4594(2)	1332(1)	42(1)
O(2)	17280(1)	3445(2)	2103(1)	57(1)
N	16764(1)	3660(2)	192(1)	32(1)
C(1)	20424(2)	5990(3)	1657(1)	65(1)

C(2)	19448(2)	4995(3)	1940(1)	55(1)
C(3)	17423(2)	3814(2)	1496(1)	36(1)
C(4)	16441(2)	3413(2)	839(1)	31(1)
C(5)	15856(2)	3260(2)	-392(1)	30(1)
C(6)	16195(2)	3472(2)	-1096(1)	39(1)
C(7)	15367(2)	3049(3)	-1700(1)	42(1)
C(8)	14152(2)	2383(2)	-1625(1)	39(1)
C(9)	13761(2)	2198(2)	-965(1)	36(1)
C(10)	14602(2)	2644(2)	-327(1)	30(1)
C(11)	14279(2)	2468(2)	387(1)	29(1)
C(12)	15226(2)	2828(2)	963(1)	33(1)
C(13)	12971(2)	1937(2)	524(1)	30(1)
C(14)	12877(2)	659(2)	981(1)	37(1)
C(15)	11685(2)	191(2)	1160(1)	39(1)
C(16)	10588(2)	1039(2)	881(1)	35(1)
C(17)	10639(2)	2309(2)	423(1)	39(1)
C(18)	11839(2)	2756(2)	241(1)	37(1)

Table 3. Bond lengths [Å] and angles [deg] for **3o**

Cl(1)-C(8)	1.7418(18)	C(1)-C(2)-H(2B)	110.2
Cl(2)-C(16)	1.7465(16)	H(2A)-C(2)-H(2B)	108.5
O(1)-C(3)	1.325(2)	O(2)-C(3)-O(1)	124.34(16)
O(1)-C(2)	1.452(2)	O(2)-C(3)-C(4)	122.68(16)
O(2)-C(3)	1.201(2)	O(1)-C(3)-C(4)	112.98(14)
N-C(4)	1.315(2)	N-C(4)-C(12)	124.45(15)
N-C(5)	1.365(2)	N-C(4)-C(3)	118.23(14)
C(1)-C(2)	1.472(3)	C(12)-C(4)-C(3)	117.30(14)
C(1)-H(1A)	0.96	N-C(5)-C(10)	123.42(14)
C(1)-H(1B)	0.96	N-C(5)-C(6)	117.69(15)
C(1)-H(1C)	0.96	C(10)-C(5)-C(6)	118.90(15)
C(2)-H(2A)	0.97	C(7)-C(6)-C(5)	121.20(16)
C(2)-H(2B)	0.97	C(7)-C(6)-H(6A)	119.4
C(3)-C(4)	1.503(2)	C(5)-C(6)-H(6A)	119.4
C(4)-C(12)	1.403(2)	C(6)-C(7)-C(8)	119.13(16)
C(5)-C(10)	1.422(2)	C(6)-C(7)-H(7A)	120.4
C(5)-C(6)	1.418(2)	C(8)-C(7)-H(7A)	120.4
C(6)-C(7)	1.353(3)	C(9)-C(8)-C(7)	122.10(17)

C(6)-H(6A)	0.93	C(9)-C(8)-Cl(1)	119.42(15)
C(7)-C(8)	1.406(3)	C(7)-C(8)-Cl(1)	118.48(13)
C(7)-H(7A)	0.93	C(8)-C(9)-C(10)	119.84(16)
C(8)-C(9)	1.358(2)	C(8)-C(9)-H(9A)	120.1
C(9)-C(10)	1.411(2)	C(10)-C(9)-H(9A)	120.1
C(9)-H(9A)	0.93	C(9)-C(10)-C(5)	118.75(14)
C(10)-C(11)	1.424(2)	C(9)-C(10)-C(11)	123.52(15)
C(11)-C(12)	1.371(2)	C(5)-C(10)-C(11)	117.71(14)
C(11)-C(13)	1.486(2)	C(12)-C(11)-C(10)	117.53(14)
C(12)-H(12A)	0.93	C(12)-C(11)-C(13)	119.69(14)
C(13)-C(14)	1.385(2)	C(10)-C(11)-C(13)	122.78(14)
C(13)-C(18)	1.392(2)	C(11)-C(12)-C(4)	120.20(15)
C(14)-C(15)	1.385(2)	C(11)-C(12)-H(12A)	119.9
C(14)-H(14A)	0.93	C(4)-C(12)-H(12A)	119.9
C(15)-C(16)	1.375(3)	C(14)-C(13)-C(18)	118.89(15)
C(15)-H(15A)	0.93	C(14)-C(13)-C(11)	119.41(15)
C(16)-C(17)	1.374(3)	C(18)-C(13)-C(11)	121.57(16)
C(17)-C(18)	1.389(2)	C(13)-C(14)-C(15)	121.17(17)
C(17)-H(17A)	0.93	C(13)-C(14)-H(14A)	119.4
C(18)-H(18A)	0.93	C(15)-C(14)-H(14A)	119.4
C(3)-O(1)-C(2)	115.93(14)	C(16)-C(15)-C(14)	118.52(17)
C(4)-N-C(5)	116.58(14)	C(16)-C(15)-H(15A)	120.7
C(2)-C(1)-H(1A)	109.5	C(14)-C(15)-H(15A)	120.7
C(2)-C(1)-H(1B)	109.5	C(15)-C(16)-C(17)	122.06(16)
H(1A)-C(1)-H(1B)	109.5	C(15)-C(16)-Cl(2)	118.77(14)
C(2)-C(1)-H(1C)	109.5	C(17)-C(16)-Cl(2)	119.16(14)
H(1A)-C(1)-H(1C)	109.5	C(16)-C(17)-C(18)	118.86(17)
H(1B)-C(1)-H(1C)	109.5	C(16)-C(17)-H(17A)	120.6
O(1)-C(2)-C(1)	107.59(17)	C(18)-C(17)-H(17A)	120.6
O(1)-C(2)-H(2A)	110.2	C(17)-C(18)-C(13)	120.49(17)
C(1)-C(2)-H(2A)	110.2	C(17)-C(18)-H(18A)	119.8
O(1)-C(2)-H(2B)	110.2	C(13)-C(18)-H(18A)	119.8

Symmetry transformations used to generate equivalent atoms:

Table 4. Anisotropic displacement parameters ($\text{\AA}^2 \times 10^3$) for **3o**

The anisotropic displacement factor exponent takes the form:

$$-2 \pi^2 [h^2 a^*{}^2 U_{11} + \dots + 2 h k a^* b^* U_{12}]$$

	U11	U22	U33	U23	U13	U12
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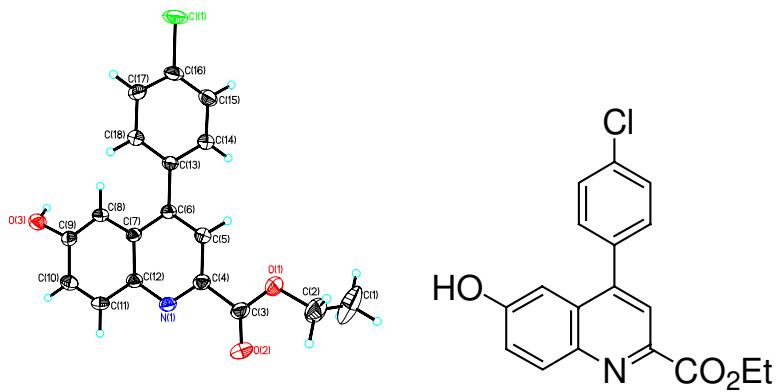
Cl(1)	61(1)	104(1)	34(1)	-16(1)	4(1)	-18(1)
Cl(2)	35(1)	56(1)	70(1)	-1(1)	24(1)	-13(1)
O(1)	32(1)	56(1)	37(1)	3(1)	1(1)	-15(1)
O(2)	44(1)	93(1)	33(1)	10(1)	3(1)	-20(1)
N	28(1)	36(1)	33(1)	3(1)	7(1)	-4(1)
C(1)	51(1)	79(2)	62(1)	-4(1)	1(1)	-29(1)
C(2)	40(1)	76(2)	45(1)	4(1)	-7(1)	-19(1)
C(3)	28(1)	44(1)	37(1)	3(1)	4(1)	-3(1)
C(4)	27(1)	33(1)	32(1)	3(1)	6(1)	-2(1)
C(5)	29(1)	31(1)	32(1)	1(1)	9(1)	-1(1)
C(6)	35(1)	48(1)	38(1)	2(1)	17(1)	-6(1)
C(7)	47(1)	53(1)	29(1)	-1(1)	16(1)	-3(1)
C(8)	41(1)	47(1)	30(1)	-5(1)	5(1)	-2(1)
C(9)	33(1)	42(1)	35(1)	-3(1)	8(1)	-7(1)
C(10)	28(1)	30(1)	32(1)	0(1)	8(1)	-2(1)
C(11)	28(1)	30(1)	32(1)	2(1)	9(1)	-3(1)
C(12)	30(1)	41(1)	30(1)	5(1)	9(1)	-3(1)
C(13)	29(1)	36(1)	28(1)	-2(1)	10(1)	-7(1)
C(14)	31(1)	40(1)	43(1)	7(1)	9(1)	-1(1)
C(15)	39(1)	38(1)	44(1)	7(1)	14(1)	-6(1)
C(16)	29(1)	41(1)	37(1)	-6(1)	13(1)	-11(1)
C(17)	30(1)	44(1)	42(1)	2(1)	7(1)	-2(1)
C(18)	34(1)	42(1)	34(1)	6(1)	7(1)	-5(1)

Table 5. Hydrogen coordinates ($\times 10^4$) and isotropic displacement parameters ($\text{\AA}^2 \times 10^3$) for **3o**

	x	y	z	U(eq)
H(1A)	21097	6280	2049	97
H(1B)	20010	6930	1441	97
H(1C)	20802	5408	1297	97
H(2A)	19854	4038	2158	66
H(2B)	19065	5567	2308	66
H(6A)	17001	3910	-1144	47
H(7A)	15599	3196	-2160	50
H(9A)	12941	1779	-934	43
H(12A)	15063	2684	1436	40
H(14A)	13629	106	1172	45
H(15A)	11627	-678	1461	47

H(17A)	9884	2860	237	46
H(18A)	11887	3608	-71	44

3.4 The single-crystal X-ray analysis of **3s**



Ethyl 4-(4-chlorophenyl)-6-hydroxyquinoline-2-carboxylate (**3s**)

Table 1. Crystal data and structure refinement for **3s**

Identification code	3s
Empirical formula	C18 H14 Cl N O3
Formula weight	327.75
Temperature	293(2) K
Wavelength	0.71073 Å
Crystal system, space group	Monoclinic, P2(1)/c
Unit cell dimensions	a = 10.300(2) Å alpha = 90 deg.
	b = 14.278(3) Å beta = 114.40(3) deg.
	c = 11.790(2) Å gamma = 90 deg.
Volume	1578.9(6) Å ³
Z, Calculated density	4, 1.379 Mg/m ³
Absorption coefficient	0.256 mm ⁻¹
F(000)	680
Crystal size	0.678 x 0.358 x 0.226 mm
Theta range for data collection	2.17 to 25.00 deg.
Limiting indices	0<=h<=12, 0<=k<=16, -14<=l<=12
Reflections collected / unique	2773 / 2773 [R(int) = 0.0000]
Completeness to theta = 25.00	100.00%
Absorption correction	Empirical
Max. and min. transmission	1.1907 and 0.7310
Refinement method	Full-matrix least-squares on F ²
Data / restraints / parameters	2773 / 0 / 209
Goodness-of-fit on F ²	0.98
Final R indices [I>2sigma(I)]	R1 = 0.0493, wR2 = 0.1266
R indices (all data)	R1 = 0.0835, wR2 = 0.1374
Extinction coefficient	0.011(2)
Largest diff. peak and hole	0.313 and -0.314 e.Å ⁻³

Table 2. Atomic coordinates (x 10⁴) and equivalent isotropic displacement parameters (Å² x 10³) for **3s**

U(eq) is defined as one third of the trace of the orthogonalized Uij tensor.

	x	y	z	U(eq)
Cl(1)	112(1)	1245(1)	-2375(1)	84(1)
O(1)	2132(2)	-718(2)	5253(2)	70(1)
O(2)	3679(2)	-122(2)	7064(2)	86(1)

O(3)	6269(2)	3896(1)	3461(2)	59(1)
N(1)	4626(2)	1115(2)	5817(2)	43(1)
C(1)	348(5)	-1221(3)	5868(5)	136(2)
C(2)	1655(4)	-1421(3)	5877(4)	96(1)
C(3)	3149(3)	-123(2)	5953(3)	53(1)
C(4)	3555(3)	546(2)	5178(2)	44(1)
C(5)	2837(2)	560(2)	3871(2)	44(1)
C(6)	3198(2)	1212(2)	3192(2)	39(1)
C(7)	4294(2)	1868(2)	3853(2)	39(1)
C(8)	4699(2)	2607(2)	3277(2)	43(1)
C(9)	5789(3)	3201(2)	3962(2)	45(1)
C(10)	6513(3)	3079(2)	5266(2)	57(1)
C(11)	6131(3)	2387(2)	5845(2)	53(1)
C(12)	4999(2)	1766(2)	5165(2)	41(1)
C(13)	2438(2)	1215(2)	1806(2)	40(1)
C(14)	954(3)	1218(2)	1228(2)	48(1)
C(15)	236(3)	1212(2)	-50(3)	57(1)
C(16)	1013(3)	1205(2)	-757(2)	52(1)
C(17)	2480(3)	1179(2)	-223(2)	54(1)
C(18)	3188(3)	1181(2)	1063(2)	48(1)

Table 3. Bond lengths [Å] and angles [deg] for **3s**

Cl(1)-C(16)	1.744(3)	O(1)-C(3)-C(4)	112.0(2)
O(1)-C(3)	1.336(3)	N(1)-C(4)-C(5)	123.4(2)
O(1)-C(2)	1.445(4)	N(1)-C(4)-C(3)	115.0(2)
O(2)-C(3)	1.193(3)	C(5)-C(4)-C(3)	121.6(2)
O(3)-C(9)	1.348(3)	C(6)-C(5)-C(4)	120.0(2)
N(1)-C(4)	1.325(3)	C(5)-C(6)-C(7)	118.1(2)
N(1)-C(12)	1.359(3)	C(5)-C(6)-C(13)	119.6(2)
C(1)-C(2)	1.372(5)	C(7)-C(6)-C(13)	122.3(2)
C(3)-C(4)	1.496(4)	C(8)-C(7)-C(12)	118.9(2)
C(4)-C(5)	1.408(3)	C(8)-C(7)-C(6)	123.7(2)
C(5)-C(6)	1.375(3)	C(12)-C(7)-C(6)	117.5(2)
C(6)-C(7)	1.425(3)	C(9)-C(8)-C(7)	120.8(2)
C(6)-C(13)	1.492(3)	O(3)-C(9)-C(8)	123.8(2)
C(7)-C(8)	1.408(3)	O(3)-C(9)-C(10)	116.3(2)
C(7)-C(12)	1.419(3)	C(8)-C(9)-C(10)	119.9(2)
C(8)-C(9)	1.373(3)	C(11)-C(10)-C(9)	120.5(2)

C(9)-C(10)	1.415(3)	C(10)-C(11)-C(12)	121.0(2)
C(10)-C(11)	1.349(4)	N(1)-C(12)-C(7)	123.3(2)
C(11)-C(12)	1.420(3)	N(1)-C(12)-C(11)	117.8(2)
C(13)-C(18)	1.387(3)	C(7)-C(12)-C(11)	118.9(2)
C(13)-C(14)	1.393(3)	C(18)-C(13)-C(14)	118.4(2)
C(14)-C(15)	1.378(4)	C(18)-C(13)-C(6)	121.0(2)
C(15)-C(16)	1.374(4)	C(14)-C(13)-C(6)	120.5(2)
C(16)-C(17)	1.376(4)	C(15)-C(14)-C(13)	121.3(3)
C(17)-C(18)	1.385(3)	C(16)-C(15)-C(14)	118.7(2)
C(3)-O(1)-C(2)	118.1(3)	C(15)-C(16)-C(17)	121.8(2)
C(4)-N(1)-C(12)	117.5(2)	C(15)-C(16)-Cl(1)	118.9(2)
C(1)-C(2)-O(1)	113.0(3)	C(17)-C(16)-Cl(1)	119.3(2)
O(2)-C(3)-O(1)	124.4(3)	C(16)-C(17)-C(18)	118.9(3)
O(2)-C(3)-C(4)	123.6(3)	C(17)-C(18)-C(13)	120.9(2)

Table 4. Anisotropic displacement parameters ($\text{Å}^2 \times 10^3$) for **3s**

The anisotropic displacement factor exponent takes the form:

$$-2 \pi^2 [h^2 a^*{}^2 U_{11} + \dots + 2 h k a^* b^* U_{12}]$$

	U11	U22	U33	U23	U13	U12
Cl(1)	100(1)	99(1)	29(1)	-2(1)	3(1)	-20(1)
O(1)	64(1)	80(2)	60(1)	27(1)	20(1)	-15(1)
O(2)	87(2)	120(2)	45(1)	28(1)	20(1)	-13(1)
O(3)	60(1)	62(1)	48(1)	0(1)	14(1)	-17(1)
N(1)	43(1)	56(1)	31(1)	1(1)	15(1)	5(1)
C(1)	149(4)	107(3)	226(6)	78(4)	152(4)	45(3)
C(2)	90(3)	96(3)	98(3)	52(2)	35(2)	-5(2)
C(3)	48(2)	69(2)	44(2)	19(2)	19(1)	13(1)
C(4)	41(1)	53(2)	38(1)	6(1)	17(1)	9(1)
C(5)	40(1)	51(2)	37(1)	1(1)	13(1)	-3(1)
C(6)	37(1)	48(1)	30(1)	0(1)	13(1)	1(1)
C(7)	38(1)	48(2)	31(1)	-2(1)	16(1)	2(1)
C(8)	45(1)	54(2)	28(1)	-4(1)	12(1)	-2(1)
C(9)	47(1)	51(2)	38(1)	-5(1)	19(1)	-7(1)
C(10)	54(2)	74(2)	37(1)	-16(1)	13(1)	-16(2)
C(11)	55(2)	74(2)	26(1)	-8(1)	12(1)	-10(2)
C(12)	40(1)	52(2)	31(1)	-4(1)	15(1)	3(1)
C(13)	43(1)	41(1)	30(1)	1(1)	10(1)	-6(1)
C(14)	45(1)	60(2)	38(1)	-1(1)	15(1)	-4(1)

C(15)	47(2)	65(2)	46(2)	-3(1)	7(1)	-4(1)
C(16)	65(2)	47(2)	31(1)	-3(1)	7(1)	-13(1)
C(17)	65(2)	61(2)	38(1)	-6(1)	24(1)	-15(2)
C(18)	44(1)	61(2)	37(1)	-2(1)	14(1)	-10(1)

Table 5. Hydrogen coordinates ($\times 10^4$) and isotropic displacement parameters ($\text{\AA}^2 \times 10^3$) for **3s**

	x	y	z	U(eq)
H(3A)	5610	4251	3070	89
H(1A)	80	-1713	6284	204
H(1B)	-349	-1173	5023	204
H(1C)	398	-639	6290	204
H(2A)	1605	-2021	5474	115
H(2B)	2351	-1475	6733	115
H(5A)	2119	127	3465	52
H(8A)	4221	2694	2421	52
H(10A)	7258	3479	5727	68
H(11A)	6617	2315	6702	64
H(14A)	437	1224	1715	58
H(15A)	-756	1213	-427	68
H(17A)	2987	1160	-717	65
H(18A)	4179	1159	1434	58

Table 6. Hydrogen bonds for **3s** [Å and deg.]

D-H...A	d(D-H)	d(H...A)	d(D...A)	<(DHA)
O(3)-H(3A)...O(2)#1	0.82	2.23	3.044(3)	174.1
O(3)-H(3A)...N(1)#1	0.82	2.48	2.868(3)	110.4

Symmetry transformations used to generate equivalent atoms:

#1 x, -y+1/2, z-1/2