

Soil, plant, and transport influences on methane in a subalpine forest under high ultraviolet irradiance

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Received: 7 April 2009 – Published in Biogeosciences Discuss.: 5 May 2009

Revised: 7 July 2009 – Accepted: 23 July 2009 – Published: 30 July 2009

Abstract. Recent studies have demonstrated direct methane emission from plant foliage under aerobic conditions, particularly under high ultraviolet (UV) irradiance. We examined the potential importance of this phenomenon in a high-elevation conifer forest using micrometeorological techniques. Vertical profiles of methane and carbon dioxide in forest air were monitored every 2 h for 6 weeks in summer 2007. Day to day variability in above-canopy CH₄ was high, with observed values in the range 1790 to 1910 nmol mol⁻¹. High CH₄ was correlated with high carbon monoxide and related to wind direction, consistent with pollutant transport from an urban area by a well-studied mountain-plain wind system. Soils were moderately dry during the study. Vertical gradients of CH₄ were small but detectable day and night, both near the ground and within the vegetation canopy. Gradients near the ground were consistent with the forest soil being a net CH₄ sink. Using scalar similarity with CO₂, the magnitude of the summer soil CH₄ sink was estimated at ~1.7 mg CH₄ m⁻² h⁻¹, which is similar to other temperate forest upland soils. The high-elevation forest was naturally exposed to high UV irradiance under clear sky conditions, with observed peak UVB irradiance >2 W m⁻². Gradients and means of CO₂ within the canopy under daytime conditions showed net uptake of CO₂ due to photosynthetic draw-down as expected. No evidence was found for a significant foliar CH₄ source in the vegetation canopy, even under high UV conditions. While the possibility of a weak foliar source

cannot be excluded given the observed soil sink, overall this subalpine forest was a net sink for atmospheric methane during the growing season.

1 Introduction

Methane is a greenhouse gas that is primarily produced by metabolism of obligate anaerobic microorganisms (Cicerone and Oremland, 1988). Although there is less CH₄ in the air than CO₂, the radiative forcing of CH₄ is significant and amounts to about one third of the forcing caused by CO₂ (Forster et al., 2007). Atmospheric CH₄ has been increasing during the last several hundred years due to human modification of agricultural systems and other human activities (Dlugokencky et al., 1994; Etheridge et al., 1998; Ferretti et al., 2005). During the last 15 years, the atmospheric increase rate slowed to near zero (Dlugokencky et al., 2003), but CH₄ has recently begun to increase again around the world (Rigby et al., 2008). The atmospheric CH₄ burden represents the net balance of extremely large sources and sinks, and while some of these are fairly well understood, others are not. The causes of the recent changes in methane increase rate remain unknown. There is a need to understand the biogeochemical processes involved in methane production and oxidation, including those in terrestrial ecosystems, if we are to make defensible estimates of the CH₄ cycle in the future.

Until recently, the role of living plants in the methane cycle was thought to be limited to transport of microbially-produced CH₄ through aerenchyma in herbaceous wetland plants and some woody plants (Schimel, 1995; Rusch and



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Rennenberg, 1998; Terazawa et al., 2007). A recent paper by Keppler et al. (2006) suggested the intriguing possibility that vascular plants emit CH₄ directly, under aerobic conditions, and without the involvement of microorganisms. These workers used their initial experiments to estimate the global significance of aerobic plant emission, and reported that as much as 240 Tg CH₄ yr⁻¹ might be produced by living plants (and more from plant litter). This factor is as large as many estimates of global wetland sources, and larger than the combined emissions from rice agriculture and ruminants (Denman et al., 2007), although many studies since then have placed the upper limit closer to 50–100 Tg CH₄ yr⁻¹ (Houweling et al., 2006; Kirschbaum et al., 2006; Ferretti et al., 2007). Not surprisingly, this finding has generated a vigorous scientific debate, focused on the possible global significance of plant methane emission (Bousquet et al., 2006; Houweling et al., 2006; Kirschbaum et al., 2006; Bergamaschi et al., 2007; Miller et al., 2007; Frankenberg et al., 2008; Houweling et al., 2008; Meirink et al., 2008), as well as the underlying mechanism of emission.

Several studies from independent laboratories have now convincingly demonstrated direct CH₄ emission from plants (Keppler et al., 2006; Cao et al., 2008; Keppler et al., 2008; McLeod et al., 2008; Vigano et al., 2008; Wang et al., 2008; Bruggemann et al., 2009; Messenger et al., 2009; Wang et al., 2009), but studies by other groups have failed to find it (Dueck et al., 2007; Beerling et al., 2008; Kirschbaum and Walcroft, 2008). It appears that ultraviolet radiation is required for plant emission (McLeod et al., 2008; Vigano et al., 2008; Messenger et al., 2009), and the lack of UV in some experimental studies may help resolve these conflicting results. UV is not always required, however; very high temperature (Vigano et al., 2008) and physical damage to plant tissues (Wang et al., 2009) can also stimulate emission. Very low rates have been observed under low light without UV in poplar shoot cultures, with strong evidence that microbial organisms are not required for CH₄ production (Bruggemann et al., 2009).

A consensus is emerging that the mechanism of plant emission may be biophysical rather than from direct production of CH₄ via a plant metabolic pathway. Emission occurs upon UV irradiation of synthetic plant pectin and related plant structural compounds (Keppler et al., 2008; McLeod et al., 2008). This is likely via a UV-initiated free-radical chemical process (Sharpatyi, 2007). Addition of chemicals that scavenge reactive oxygen species (ROS) strongly inhibits UV-induced CH₄ emission from pectin, and CH₄ emission in the dark can occur when certain ROS are added to pectin (Messenger et al., 2009). Other environmental stressors can initiate formation of ROS (Apel and Hirt, 2004), so it is possible that factors other than UV might induce CH₄ emission from plants (Messenger et al., 2009).

Most studies investigating CH₄ emission from plants have been conducted in the laboratory, using either artificial lamps or sunshine (e.g., Keppler et al., 2006; Vigano et al., 2008).

The few studies that have directly observed plant CH₄ emission in the field (Sanhueza and Donoso, 2006; Cao et al., 2008; Wang et al., 2008) have used chambers or vials with detached (or rarely intact) plant tissues, which limit or omit UV irradiation relative to a plant's natural environment. At present, we have virtually no basis to compare CH₄ emission rates measured in the lab using artificial lamps to those that might be expected in the field under natural sunlight without chambers.

The soils of most upland forests act as a sink for methane (Dutaur and Verchot, 2007) due to the metabolic activity of methanotrophic microorganisms (King, 1992). In ecosystems dominated by plants that do emit CH₄, the presence of a local soil sink could limit the overall importance of plant production to the atmospheric CH₄ budget. However, to date there have been no studies that were specifically designed to evaluate the ecological significance of plant methane emission from whole ecosystems under natural conditions in the field.

Our primary objective was to examine the general patterns of summer methane exchange in an upland subalpine conifer forest. The forest was located at high elevation and naturally exposed to high UV irradiance, providing an ideal situation to examine plant and soil CH₄ exchange processes under natural conditions. Micrometeorological techniques were used to avoid the microclimatic and radiative artifacts of plant and soil chambers. In this paper we examine the causes for variability of CH₄ in subalpine forest air, including forest biological processes and atmospheric transport. We investigate local soil production and consumption of methane, and the magnitude of soil-atmosphere exchange. Measurements of CO₂ and CO are used to provide some understanding of expected CH₄ patterns within the forest and of transport of urban air masses. Finally, we address the importance of aerobic CH₄ emission from foliage in the context of whole-forest biosphere-atmosphere exchange.

2 Methods

2.1 Study location

The primary study site was a subalpine conifer forest (the Niwot Ridge AmeriFlux forest) in the southern Rocky Mountains of the United States (40.03° N, 105.55° W, 3050 m elevation). The roughly 110-yr old stand is dominated by *Pinus contorta* (lodgepole pine), *Picea engelmannii* (Engelmann spruce), and *Abies lasiocarpa* (subalpine fir), with a sparse understory. The mean annual temperature is 1.5°C, and the mean annual precipitation is 800 mm, roughly 40% of which is snow in winter. Further site details and information about measurement of weather and environmental parameters, energy and CO₂ exchange can be found elsewhere (Monson et al., 2002, 2006).

2.2 In-situ methane and carbon dioxide analysis

Vertical profiles of CH₄ and CO₂ were measured in the forest during 28 June–10 August, 2007 (day of year 179–222). Measurement heights were 21.5 m (above the vegetation canopy), 11, 9, and 7 m (within the canopy, vegetation height was 11–12 m), and 1, 0.5, and 0.1 m (in the sub-canopy air space near the ground). The inlet system (described in detail by Schaeffer et al., 2008b) was time-multiplexed with other instrumentation; CH₄ and CO₂ were measured during one 0.5 h subset of each 2 h period. The full profile for both gases was measured within a half hour, on the same gas samples simultaneously. Air flowed continuously through sampling lines and was sub-sampled for analysis. CH₄ mole fraction was measured using off-axis integrated-cavity output spectroscopy (Model 908-0001, Fast Methane Analyzer, Los Gatos Research, Mountain View, CA). CO₂ and H₂O vapor mole fraction were measured using a non-dispersive infrared gas analyzer (LI-7000, Li-cor, Lincoln, NE). All CH₄ and CO₂ mole fractions are reported relative to World Meteorological Organization (WMO) calibration scales, which are maintained at the National Oceanic and Atmospheric Administration/Earth System Research Laboratory (NOAA/ESRL, <http://www.esrl.noaa.gov/>). During the 1.5 h not devoted to methane measurements, CO₂ and $\delta^{13}\text{C}$ of CO₂ were measured as described by Schaeffer et al. (2008b); these data are not used in the present paper.

A 4-way crossover valve (Valco E4UWE, Houston, TX) was used to route sample air from the selected tower inlet to the selected analyzer. When a given analyzer was not in use, the crossover valve was plumbed to route filtered room air to the analyzer. Immediately downstream of the crossover valve, sample air entered a 2.4 m length of 1.5 mm inner diameter Nafion membrane (Nafion MD-070-96P-2, Toms River, NJ) to remove moisture. The countercurrent flow on the outside of the Nafion membrane was generated by pumping room air through a 80/20 500 g mixture of molecular sieve 4a (Acros Organics, Morris Plains, NJ) and indicating desiccant (Drierite, Xenia, OH). The flow was maintained at 400 sccm (twice that of the sample flow) using a needle valve downstream of the moisture trap. The minimum dew point achieved using this method was -26°C and averaged -18°C . The moisture traps were changed every 7–10 days after which time the dew point averaged -14°C ; in one case the dew point rose as high as 5°C for one 24 h period. After drying, the air passed into the CH₄ analyzer via a 2 micron high flow filter (Nupro SS-44FW4-2, Solon, OH). The flow rate through the analyzer was maintained at 200 sccm using the analyzer's internal pump and a mass flow controller immediately downstream of the instrument. The flow was tuned to optimize both the drying and cell flushing. The exhaust flow of the CH₄ analyzer was routed to the sample cell of the LI-7000 for CO₂ and H₂O measurement, while dry N₂ flowed through the reference cell at 10 sccm.

The CH₄ analyzer was calibrated using whole air from high pressure cylinders that were calibrated for CH₄ at NOAA/ESRL prior to deployment. A two-point calibration was employed (zero, and span, $1713.9 \pm 0.8 \text{ nmol mol}^{-1}$) given the stability of the analyzer. Additionally, a second calibrated CH₄ tank ($1909.2 \pm 0.8 \text{ nmol mol}^{-1}$) was used as a “target tank” to independently gauge the “measurement trueness” (accuracy) and “measurement reproducibility” (precision) (JCGM, 2008). The reproducibility as indicated by replicate measurements of the target tank was $0.35 \text{ nmol mol}^{-1}$. The difference between measured and calibrated value of the target tank was $+2.8 \text{ nmol mol}^{-1}$ (1912.0 – 1909.2), which is within 0.15% of the calibrated target tank (note the reference gas used for calibration was $1713.9 \text{ nmol mol}^{-1}$). When the target tank was measured following calibration with a tank closer to the reference gas ($1842.5 \text{ nmol mol}^{-1}$), the difference was only $1.0 \text{ nmol mol}^{-1}$. This “gain” difference amounts to only 0.01, indicating that, for example, a measured vertical or temporal difference of 20 nmol mol^{-1} would be in error by only about $0.2 \text{ nmol mol}^{-1}$. Like methane, in addition to the seven levels, for CO₂ both a reference and target gas were introduced into the sample cell, in order to calibrate the measurements to the WMO scale and verify the calibration. Although the target gas exhibited a linear CO₂ drift of about $0.03 \mu\text{mol mol}^{-1} \text{ day}^{-1}$, the residual standard deviation from that drift was $0.2 \mu\text{mol mol}^{-1}$; the mean offset from the assigned value was $-2.1 \mu\text{mol mol}^{-1}$ at a distance from the reference gas of $152 \mu\text{mol mol}^{-1}$. For both CH₄ and CO₂, the precision and stability of our analysis system indicates our ability to resolve small atmospheric gradients.

Average CH₄ mole fractions for a given intake height at a given time were determined as the mean of the last 60 s of 200 s dwell time at a given level or gas cylinder; the majority of this time was used to flush the sample cell of the analyzers completely, based on pre-deployment bench tests. For the target tank, the reproducibility was calculated as the one sigma standard deviation of these means over the entire measurement period (day of year 179–222).

Drying air and/or calculating its moisture content was necessary to determine accurate methane mole fractions. In particular, the WMO methane scale is defined as “dry air mole fraction”: $\text{nmol CH}_4 (\text{mol dry air})^{-1}$. Thus a dilution correction must be made to the raw output of a spectrometer that measures the ratio of CH₄ to total pressure, which includes the H₂O vapor partial pressure. For example, a raw measurement of $1800 \text{ nmol CH}_4 \text{ mol}^{-1}$ with a water content of $1000 \text{ mmol H}_2\text{O mol air}^{-1}$ – a dew point of about -26°C at the pressure and typical temperature of the site – is $1801.8 \text{ nmol CH}_4 \text{ mol}^{-1}$ after correction. For a dew point of 4°C ($\sim 12\,500 \text{ mmol H}_2\text{O mol air}^{-1}$ H₂O), typical of daytime ambient conditions, the correction would be about $20 \text{ nmol CH}_4 \text{ mol}^{-1}$. Our primary strategy was to dry the air, and we additionally measured its moisture content with the LI-7000 to be able to correct it regardless of the efficiency

of drying. Moreover, even when undried and uncorrected, small gradients of water vapor do not contribute significant artifacts to observed CH_4 gradients; the largest H_2O vapor gradient we observed was less than $200 \text{ mmol H}_2\text{O mol air}^{-1}$ after the dessicant ran out.

2.3 Automated flask collection for methane and carbon monoxide analysis

Air samples from the primary study site, and a second alpine tundra site, were collected in glass flasks for laboratory analysis of several trace gases, including CH_4 and CO . These data were used to provide an indicator of regional transport of urban air. The tundra site (40.05° N , 105.58° W , 3423 m , NOAA/ESRL site NWR) was located approximately 3 km to the NW and 373 m higher in elevation from the forest site. Samples were collected using an automated flask sampler described by Schaeffer et al. (2008b), with simultaneous collection at both sites, at midday (14:00 h) and at night (02:00 h), every other day. CH_4 and CO mole fractions were measured in the flasks by the NOAA Carbon Cycle Greenhouse Gases Group as described by Schnell (2004). To investigate general patterns of arrival of polluted urban air at the sites, flask data are presented from the period 18 September 2005 to 12 October 2007 (all data available), and for the subset of time during summer 2007 (days 170–230) in this paper.

2.4 Ultraviolet irradiance

Ambient UV irradiance data were obtained from the NOAA-EPA Brewer Spectrophotometer UV Network (<http://www.esrl.noaa.gov/gmd/grad/neubrew/>) for an adjacent site (MRS, CO), located 1.1 km east of the forest site and 127 m lower in elevation. These data include radiant spectral flux density in the 286.5 to 363 nm range, with one full spectrum collected roughly every 15 min during daylight hours. Spectral UV data were integrated to calculate total irradiance in a particular wavelength range, and then averaged on a 30-min basis to match trace gas analysis periods. The measurement range of the instrumentation was narrower than the usual definition of UVA ($315\text{--}400 \text{ nm}$) and UVB ($280\text{--}315 \text{ nm}$) wavelength ranges (McKenzie et al., 2004). This leads to an underestimation in our reported UVA irradiance by roughly a factor of 2 (under clear sky conditions) relative to the irradiance in the unweighted full UVA range (see Fig. 1). There is little UVB at Niwot Ridge in the unmeasured higher-energy region of 280 to 286.5 nm , so our reported irradiance is indicative of the true UVB irradiance.

2.5 Data selection and analysis: “gradients” and “daytime excess”

Daytime and nighttime periods were selected based on analysis of solar radiation versus time of day. CH_4 and CO_2 data are presented for daytime (06:15 to 17:15 h local time) and

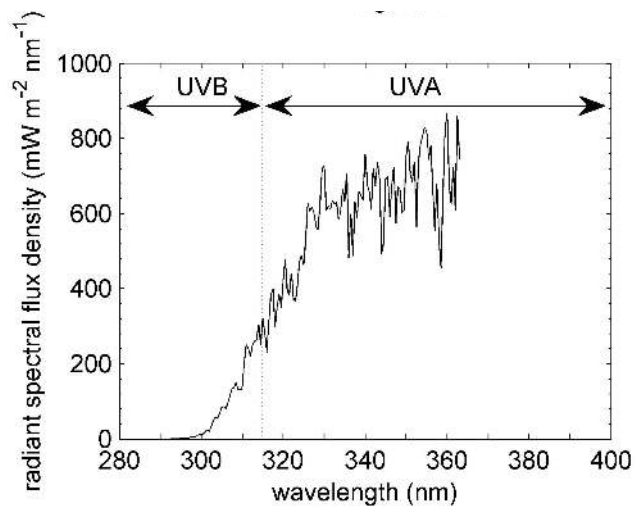


Fig. 1. A typical UV irradiance spectrum at Niwot Ridge (midday on day 203). The vertical line divides the standard UVB (left, $280\text{--}315 \text{ nm}$) and UVA (right, $315\text{--}400 \text{ nm}$) ranges.

nighttime (19:15 to 04:15 h) periods separately, and we ignore the 2-h transition periods between to avoid complexities of storage fluxes. In all, 238 daytime and 164 nighttime profiles were obtained. Data were analyzed separately for the within-canopy inlets ($7, 9, 11 \text{ m}$) and the near-ground inlets ($0.1, 0.5, 1 \text{ m}$), assuming these to be indicative of canopy and soil processes, respectively. There is strong evidence for the validity of this assumption under stable atmospheric conditions at night (Yi et al., 2005; Schaeffer et al., 2008a). During the day, processes from each vertical zone will certainly influence the other. For each group, gradients were calculated as the slope of a least-squares linear regression (mole fraction versus height) through the three grouped inlets. The daytime excess content of each trace gas within the vegetation canopy was calculated as the mean of the mole fractions at the 3 canopy inlets minus the mole fraction at 21.5 m . Daytime excess is used as an index of canopy physiological activity. Note that the daytime excess is in effect a gradient (from within the canopy region to above it) but is expressed as a mole fraction difference. A positive daytime excess should generally indicate a source within the forest, a negative one a sink.

3 Results and discussion

3.1 Environmental conditions and observed CH_4 and CO_2

Environmental conditions during the study are shown in Fig. 2. Most days were sunny in the morning with some afternoon cloud cover, and later in the summer afternoon rainstorms increased soil moisture. Air temperature peaked at 24.5° C , and although there were some cold nights, soils

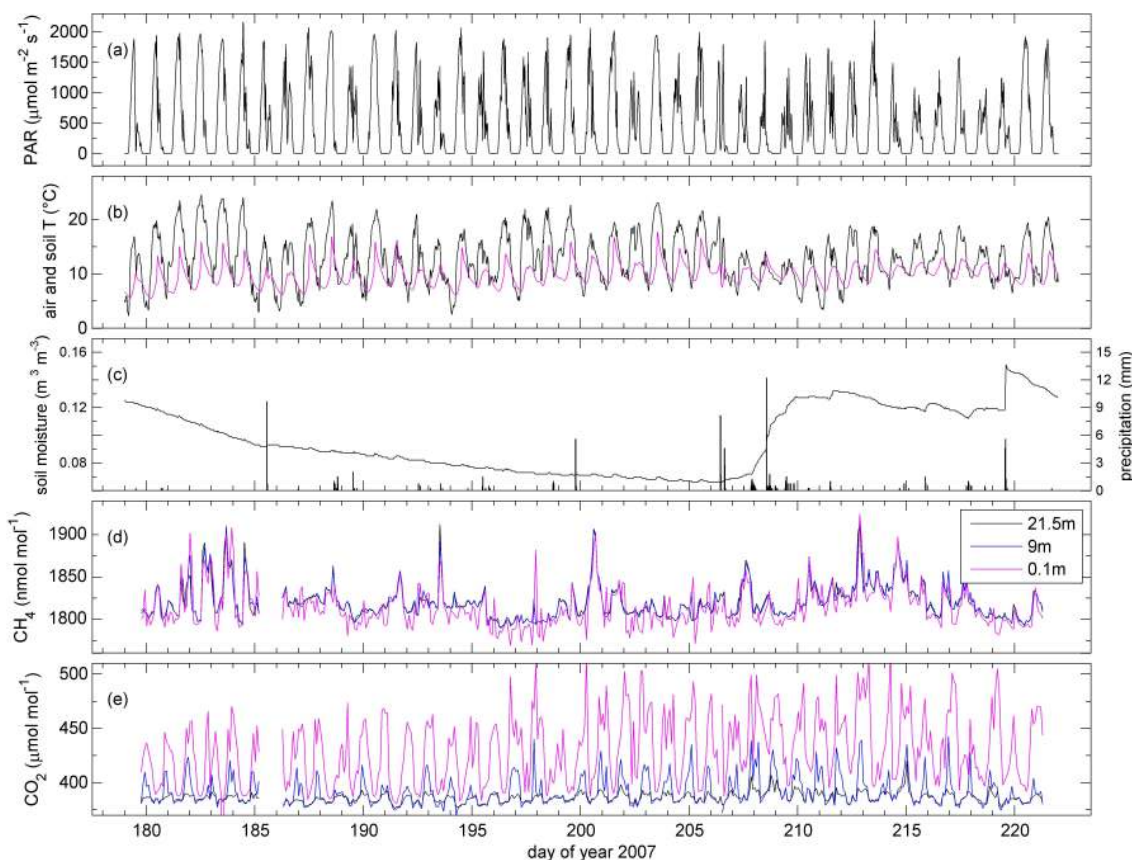


Fig. 2. Environmental conditions, methane, and carbon dioxide mole fractions during the study. (a) above-canopy photosynthetically active radiation (PAR), (b) air temperature at 8 m (black line), and soil temperature at 5 cm depth (pink line), (c) soil moisture at 5 cm depth (line, left axis) and above-canopy precipitation (total in a 30-min period, bars, right axis), (d) CH₄ mole fractions, and (e) CO₂ mole fractions at 3 measurement heights within the forest. The legend in (d) also applies to (e).

never experienced freezing conditions (soil temperature data not shown). Precipitation during June (the month preceding the study) was very low (70% of normal for the month), and July/August precipitation were fairly normal. Near-surface soils during the study were moderately dry, with water content declining for the first 30 days from 12 to 8%, and increasing following the late summer rains to about 15% (field capacity of these soils is $\sim 40\%$).

Several studies have recently highlighted the importance of UV radiation for aerobic methane production by plants (Keppler et al., 2008; McLeod et al., 2008; Vigano et al., 2008; Messenger et al., 2009). The high elevation of the Niwot Ridge forest led to high solar irradiance (Fig. 3). Incident photosynthetically active radiation peaked above $2000 \mu\text{mol m}^{-2} \text{s}^{-1}$, with UVA (in the limited measurement range) exceeding 30 W m^{-2} , and UVB greater than 2 W m^{-2} . On average the daily UV fluxes peaked at 20 and 1.4 W m^{-2} , respectively (Fig. 3). In a laboratory study with artificial lamps, Vigano et al. (2008) reported a fairly high methane emission rate ($55 \text{ ng CH}_4 (\text{g dw})^{-1} \text{ h}^{-1}$) from living conifer foliage (*Pinus ponderosa*) under total UV irradiance of 38 W m^{-2} (note that their UVB irradiance of 24 W m^{-2}

was a much higher portion of the total than ours). As discussed, our UVA values are probably underestimated by a factor of two, so based on the Vigano et al. study one might expect the UV at the Niwot Ridge forest to be sufficient to cause CH₄ emission from *Pinus* species. The dominant tree species are all conifers, including *P. contorta*, which is congeneric to the pine used by Vigano et al. (2008).

Shown in Fig. 2d and e are time series of CH₄ and CO₂ measured above-, within-, and below-canopy. CO₂ exhibited a distinct diel pattern, with highest CO₂ near the ground due to soil respiration. Day to day variability in CH₄ was high. The amplitude of the seasonal cycle of CH₄ at Niwot Ridge is roughly 50 nmol mol^{-1} (Dlugokencky et al., 1994), and change during our summer study period that is driven by the seasonal cycle at this latitude should be maximally around 5 nmol mol^{-1} (Shipham et al., 1998a). Simpson et al. (1999) reported a systematic shift in CH₄ during the growing season in a boreal aspen forest, and attributed it to increased soil production (as soil temperature increased). Neither CH₄ mole fraction nor soil temperature changed markedly over the course of our study (Fig. 2b and d).

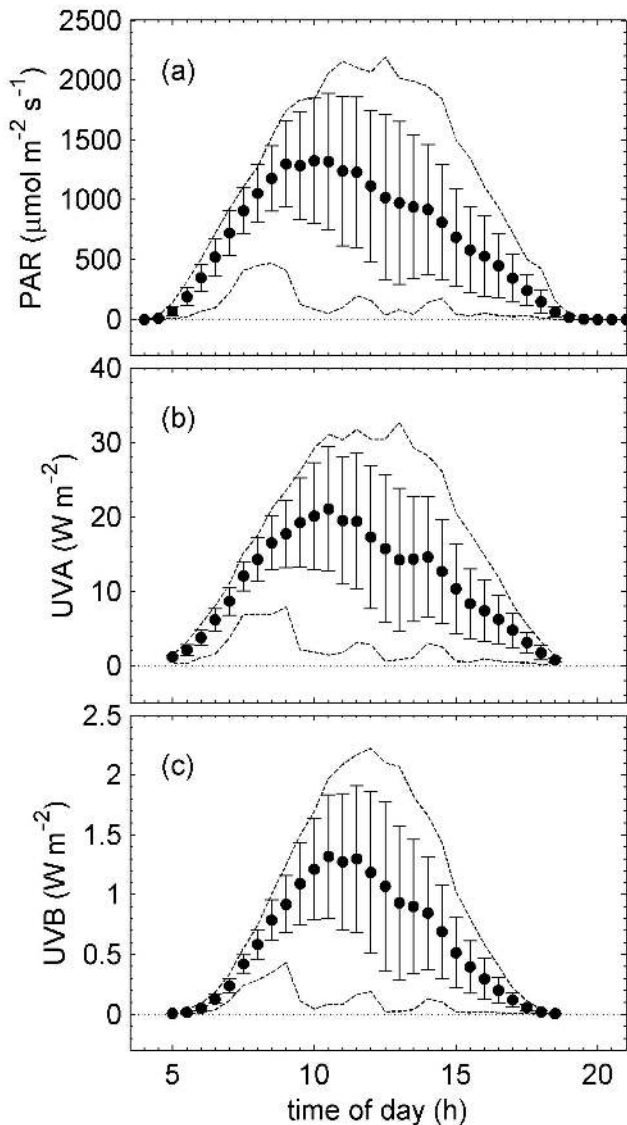


Fig. 3. Diel time course of above-canopy radiant flux densities (a) photosynthetically active radiation, (b) UVA (in the range 315–363 nm), and (c) UVB (286.5–315 nm) irradiance. The data points show the mean over the 43 days of the study, error bars are ± 1 standard deviation, and the dashed lines show the observed ranges for each time period. Note that the UVA sub-range 363–400 nm is not covered by these observations (Fig. 1), so the actual UVA irradiances were likely higher by a factor of 2 than those shown.

It is well-known that CH_4 production occurs in anaerobic soil conditions. There were no abrupt changes in CH_4 associated with precipitation events in the present study, although there is some indication of a baseline methane increase 1–2 days following many precipitation events (Fig. 2c and d). Rainstorms later in the season that increased soil moisture (days 208–210) did not substantially increase the CH_4 baseline. While the soils were never saturated during the study, anaerobic micro-sites (Sey et al., 2008) were likely present

within soil aggregates following the rain events. If CH_4 was produced in soil aggregates it was likely consumed before diffusing to the forest air.

Scientists have observed diel patterns in CH_4 within forests, with highest CH_4 mole fractions at night, and diel amplitudes of 6–100 nmol mol^{-1} (Shipham et al., 1998a; Simpson et al., 1999). We observed some periods of diel patterning in CH_4 (Fig. 2d, days 181–185), but most of the record did not show it. In general, CH_4 mole fractions within the canopy were similar to those above the canopy, but CH_4 near the soil surface tended to be lower regardless of time of day. This was likely caused by soil CH_4 consumption. The common diel pattern in CO_2 (e.g., Jarvis et al., 1997) was caused by a consistent respiratory source (from soils always and from foliage at night) coupled with changes in atmospheric stability. Stability should also influence CH_4 , but the lack of significant diel pattern (Fig. 2d) suggests that additional factors were responsible for the day-to-day variability of CH_4 in the air at the Niwot Ridge forest.

The wind direction in summer at the Niwot Ridge forest was bimodal, with relatively clean air flowing over the Continental Divide from the west at night, and a daytime reversal (Fig. 4). Above-canopy winds were primarily westerly (night) or easterly (midday). Early morning and late afternoon winds were usually westerly. Below-canopy winds were primarily westerly at night, but quite variable in direction during the day (Fig. 4c). The daily wind direction reversal was caused by a well-known regional mountain-plain wind system at this location (Brazel and Brazel, 1983).

The Niwot Ridge forest is situated near the eastern edge of the Rocky Mountains where they meet the Great Plains of the United States. The forest is roughly 60 km northwest of a large urban region on the plains (the Denver, Colorado, metropolitan area). A series of experiments focused on photochemical processes in air established that pollutants from the Denver urban area can be transported to the Niwot Ridge forest by easterly winds in summer (Roberts et al., 1985; Parrish et al., 1990, 1991). Although CH_4 was not measured in those studies, carbon monoxide at Niwot Ridge tends to be higher during summer easterly winds (Parrish et al., 1991). In general, the highest CH_4 observed during the present study was associated with daytime easterly flow (Fig. 4a). High CH_4 periods were strongly correlated with high CO content (Fig. 5), at both the forest and the tundra sites, during summer 2007 and also in the longer record. Similar correlations between CH_4 and anthropogenic hydrocarbon pollutants have been observed in long-term records at Harvard Forest, Massachusetts (Shipham et al., 1998b), suggesting an anthropogenic origin for elevated CH_4 events. Combined with the Niwot Ridge air chemistry studies already mentioned, the data in Fig. 5 are strong evidence that the day to day variability in CH_4 observed during our study (Fig. 2d) was caused by regional transport. High CH_4 events such as days 200, day 207, and day 212 were probably urban in origin and likely not caused by local biological production

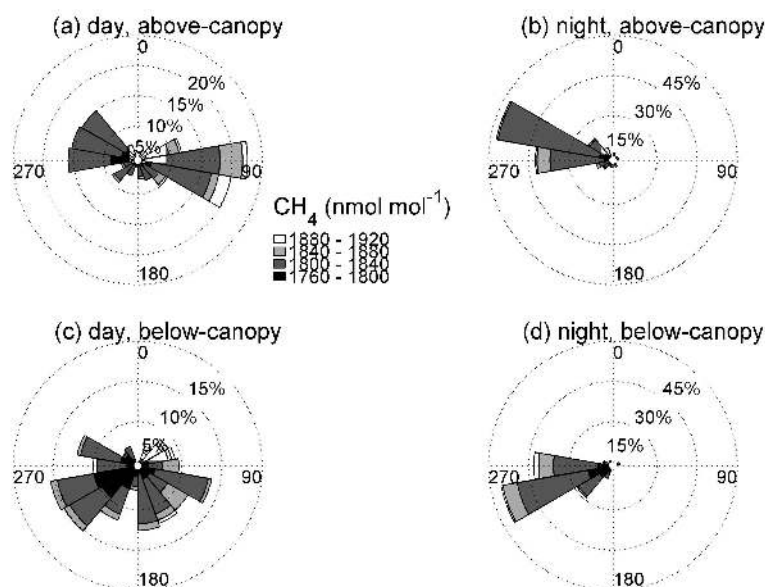


Fig. 4. Wind roses and methane mole fractions during the study. Data are shown for day (a) and (c) and night (b) and (d) separately for above-canopy (a and b, 21.5 m wind and 21.5 m CH₄) and below-canopy (c and d, 2.56 m wind, 1 m CH₄). The radius of each wedge shows the percentage of time that the wind blew from the direction indicated (in 20° bins), and the wedges are divided into 4 gray-scale bins based on the observed CH₄ mole fraction during that period. The area of each gray-scale portion of the wedge is proportional to the relative frequency that the CH₄ mole fraction was observed in the indicated range. The wind percentage scales differ in the panels.

within the forest. Although the easterly flow pattern was common, not all easterly flow was associated with high CH₄ (Fig. 4), presumably due to variability of air mass origin, aging, or route of transport.

3.2 Below-canopy gradients and soil fluxes

Our principal aim was to assess the presence of local forest CH₄ sources and sinks, and there are several possible ways to address this aim. Although the respiratory soil CO₂ source was associated with a consistent diel pattern, the diel pattern was largely absent for CH₄, and variability in mole fraction is not a robust indicator of fluxes. Others have used CH₄ profiles within forests and a canopy-layer budget method to assess forest CH₄ source strength (do Carmo et al., 2006). Inverse Lagrangian analysis is probably the best way to assess the vertical distribution of fluxes within a canopy, and has been used for CH₄ in rice canopies (Leuning et al., 2000). However, given the complexities of radiative transport in clumped conifer canopies (Baldocchi and Collineau, 1994), the central importance of UV radiation for aerobic CH₄ production (McLeod et al., 2008; Vigano et al., 2008; Messenger et al., 2009), and the non-homogeneous nature of the Niwot Ridge canopy, such an analysis is beyond the scope of this paper.

We chose to use gradients and daytime excess (see Methods) to assess whether soils and vegetation at the site were sources or sinks for CH₄. We cannot, however, use observed gradients to calculate meaningful CH₄ fluxes during every 30-min measurement period. Within plant canopies,

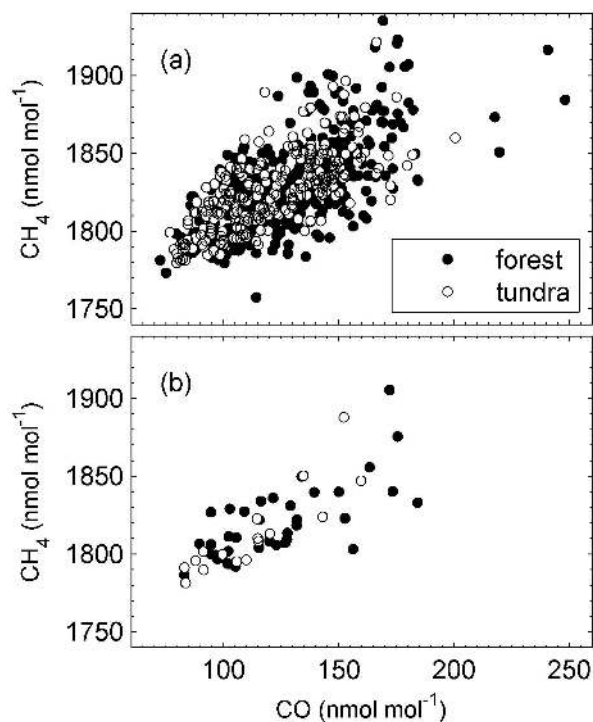


Fig. 5. (a) Methane and carbon monoxide mole fractions at the forest and tundra sites during September 2005 to October 2007 and (b) for the subset of time during summer 2007.

the direction of trace gas flux can sometimes be opposite the direction of the mole fraction gradient, and the assumptions on which flux-gradient theory is based are often violated (Raupach, 1979; Denmead and Bradley, 1985; Cellier and Brunet, 1992). However, over time the direction of the fluxes and gradients should converge in their means, otherwise mass could not be conserved. For example, a persistent soil CH₄ sink should have, in a mean sense, a downward flux direction and a matching gradient, with lower CH₄ nearer the ground. Similarly, a persistent canopy source of CH₄ (such as from aerobic foliar emission) should on average have elevated CH₄ in the canopy relative to the above-canopy air.

In the following discussion, we present the frequency distributions of gradients (and of daytime excess) to assess the direction and magnitude of fluxes. Both CO₂ and CH₄ data are presented because many readers are familiar with spatial and temporal patterns of CO₂ in forest air and can use that knowledge to interpret CH₄ patterns (which are less well-understood). We first describe two example periods in detail, and then show the full distributions of observed gradients. Shown in Fig. 6 are sample night and day profiles for CO₂ and CH₄, each measured within a 30-minute period. The CO₂ profile at night (Fig. 6a) showed high CO₂ near the ground ($\sim 450 \mu\text{mol mol}^{-1}$), decreasing with height to a minimum above the canopy ($\sim 390 \mu\text{mol mol}^{-1}$). The gradients near the ground and within the canopy are shown as solid lines (regressions through each of the 3 groups of inlets). The sign convention used is that a negative slope in the lines shown in Fig. 6 (such as for CO₂ at night, panel a) corresponds to a positive gradient ($\mu\text{mol mol}^{-1} \text{m}^{-1}$) in the figures that follow. By this convention a positive gradient indicates a source, and a negative gradient a sink. The night CO₂ profile shown reflects a soil source, and there was a CH₄ sink in the soil (based on the direction of each gradient in the near-ground inlets). The CO₂ gradient within the canopy indicates a CO₂ source, which includes respiration from soil, living woody tissues, and leaf (needle) dark respiration. Note, however, if we did not know the foliage was a source of CO₂ at night, the profile observed within the canopy (Fig. 6a) might result from the CO₂ source in the soil. In contrast, the apparent canopy CH₄ source (in this example) cannot be attributed to the local soils since they were a sink (see opposite direction of gradients in canopy and near the ground in Fig. 6b).

During the daytime period shown, the soils remained a CO₂ source (Fig. 6c), and a CH₄ sink (Fig. 6d). The negative CH₄ gradient within the canopy could indicate 1) a soil CH₄ sink and no canopy exchange of CH₄, 2) a soil CH₄ sink and a canopy CH₄ sink, or 3) a dominant soil CH₄ sink and a smaller canopy CH₄ source, with relative magnitudes such that the combined net CH₄ flux is a sink. Analysis of gradients during individual measurement periods cannot reliably be used to distinguish between these possibilities. CO₂ within the canopy during the daytime was lowest at the middle canopy inlet (Fig. 6c), which illustrates a potential prob-

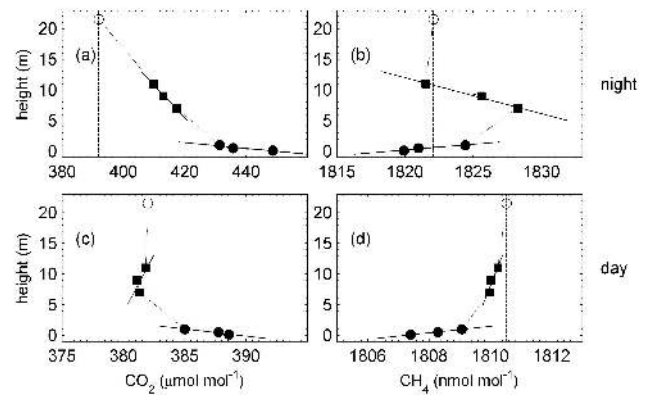


Fig. 6. Example vertical profiles of carbon dioxide (a) and (c) and methane (b) and (d) mole fractions. The top panels show a stable night period (day 192.85) and the bottom panels show a high-UV daytime period (day 192.44). The inlets near the ground are indicated with filled circles, those within the vegetation canopy with filled squares, and above-canopy with an open circle. The vertical dashed line represents the above-canopy observation and the solid lines show linear regressions through the canopy or ground inlets separately. The slopes of the regression lines are used to calculate the gradients of each trace gas for each 30-min measurement period shown in Figs. 7 and 8.

lem with analysis of small within-canopy gradients. While the photosynthetic CO₂ sink in the canopy was clear (since the within-canopy CO₂ was consistently lower than above- and below-canopy at all 3 canopy measurement heights), the regression line at times like these might have either a positive or negative slope based on the relative drawdown at each measurement height, because the gradient is not consistent in direction. In some of the figures that follow, we present both within-canopy daytime gradients and daytime excess (within-canopy mean relative to above-canopy) to clearly indicate the pattern in the canopy.

The frequency distributions of all observed gradients at the 3 near-ground inlets are shown in Fig. 7. Daytime and nighttime periods and subsets of these (high UV or stable nights) are shown separately. In general, gradients of CO₂ near the ground were positive, both during the day (Fig. 7a, mean ± 1 SD of the daytime gradients was $17.7 \pm 14.2 \mu\text{mol mol}^{-1} \text{m}^{-1}$) and at night (Fig. 7c, $20.5 \pm 12.7 \mu\text{mol mol}^{-1} \text{m}^{-1}$), with a large number of profiles measured for each case (number of profiles, n , is indicated above each panel). The CO₂ gradient distributions were positively skewed. CH₄ gradients near the ground were generally negative during the day (Fig. 7e, $-5.4 \pm 5.1 \text{ nmol mol}^{-1} \text{m}^{-1}$) and at night (Fig. 7g, $-6.2 \pm 8.2 \text{ nmol mol}^{-1} \text{m}^{-1}$), and the distributions were strongly negatively skewed, indicating a persistent net soil sink for CH₄. Both CO₂ and CH₄ fluxes were strong enough that the near-ground gradients persisted during the day when greater turbulent mixing had the potential to erase them.

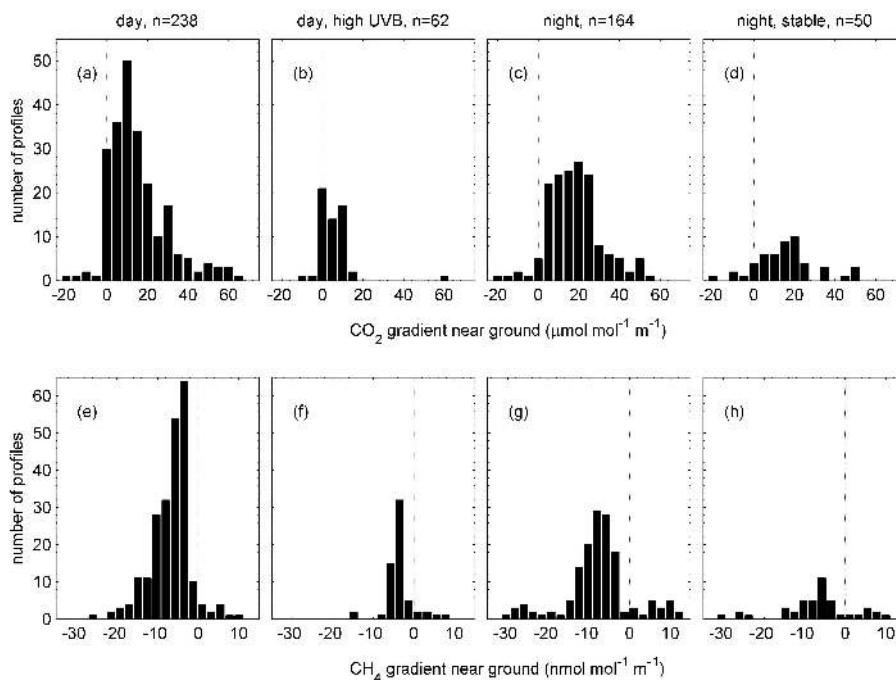


Fig. 7. Frequency distributions of observed CO₂ (top) and CH₄ (bottom) gradients using the near-ground inlets (0.1, 0.5, and 1 m height). The data are separated into all daytime periods (a) and (e), daytime periods with UVB flux density greater than 1 W m⁻² (b) and (f), all nighttime periods (c) and (g), and stable nighttime periods with friction velocity <0.15 m s⁻¹ (d) and (h). Vertical dotted lines show the zero gradient. The sign convention means positive gradients indicate a source, negative gradients a sink. The titles on the 4 top panels also apply to the lower 4 panels.

The soil-source CO₂ gradient pattern has been observed in summer-long mean diel patterns at the Niwot Ridge forest (Bowling et al., 2005), and this is no surprise given the respiratory production of CO₂ by soil organisms, including roots. The soil methane sink, which is common in temperate forest soils (Smith et al., 2000; Le Mer and Roger, 2001), is most likely due to activity of methanotrophs.

If we assume scalar similarity between CO₂ and CH₄, the gradients of each gas near the ground should be related to the respective fluxes by the same eddy diffusivity. This is the basis of the flux-gradient method for estimation of fluxes (Baldochi et al., 1988), and can be formalized as:

$$F_{\text{CO}_2} = -K \frac{d\text{CO}_2}{dz} \quad (1)$$

$$F_{\text{CH}_4} = -K \frac{d\text{CH}_4}{dz} \quad (2)$$

$$F_{\text{CH}_4} = - \left[\frac{F_{\text{CO}_2}}{d\text{CO}_2/dz} \right] \left[\frac{d\text{CH}_4}{dz} \right] \quad (3)$$

where F_{CO_2} and F_{CH_4} represent the flux of each gas, $d\text{CO}_2/dz$ and $d\text{CH}_4/dz$ the summer mean daytime gradients of each gas near the ground, and K the eddy diffusivity. Equation (3) allows us to estimate a summer mean flux of CH₄ from a known summertime CO₂ flux and mean measured gradients of each trace gas. Midsummer soil CO₂

efflux in the Niwot Ridge forest is typically in the range 4–6 μmol m⁻² s⁻¹, and at the soil moisture levels of the present study, around 4 μmol m⁻² s⁻¹ (Scott-Denton et al., 2003; Scott-Denton et al., 2006). Using this value and the mean daytime gradients of 17.7 μmol CO₂ mol⁻¹ m⁻¹ and -5.4 nmol CH₄ mol⁻¹ m⁻¹ from Fig. 7, with Eq. (3) we estimate the magnitude of the soil methane sink as 1.7 mg CH₄ m⁻² d⁻¹, (for comparison with studies below, this equates to 70 μg CH₄ m⁻² h⁻¹ or 17 g CH₄ ha⁻¹ d⁻¹).

We are unaware of any previous reports of CH₄ uptake by high-elevation subalpine forest soils. This is clearly an approximation, but should be of the correct order of magnitude. Our estimate is well within the range of published values for upland forest soil methanotrophy. In a comprehensive review of emission and consumption of methane by soils, Le Mer and Roger (2001) reported the median CH₄ uptake flux for primarily tropical and temperate forest soils was 9.9 g CH₄ ha⁻¹ d⁻¹, with a range of 0.16 to 1659 g CH₄ ha⁻¹ d⁻¹. CH₄ uptake by temperate coniferous and deciduous forest soils in Europe tends to be in the range 0–100 μg CH₄ m⁻² h⁻¹ (Smith et al., 2000). The soils at Niwot Ridge are always saturated during and shortly after the period of snowmelt (Monson et al., 2006; Molotch et al., 2007), usually leading to overland flow (Hood et al., 2003). The snowmelt time period was not covered by our observations, and local soil CH₄ production is quite possible under those conditions.

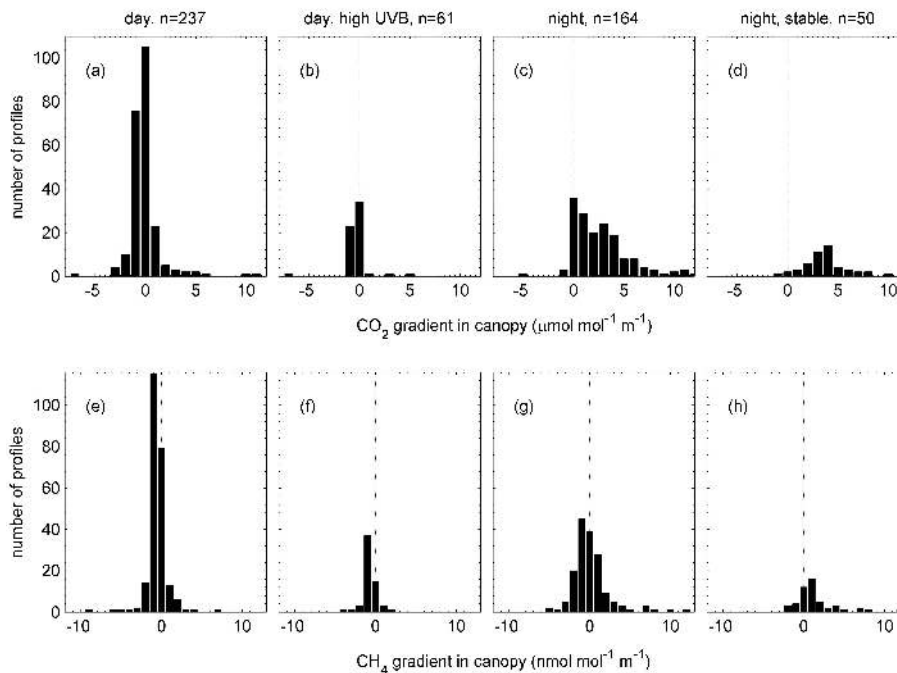


Fig. 8. Same as Fig. 7, but for the within-canopy inlets (7, 9, 11 m height).

3.3 Within-canopy gradients, daytime excess, and canopy fluxes

Gradients of both gases were generally smaller within the vegetation canopy (compared to near the ground) under all conditions (Fig. 8). At night, the within-canopy CO_2 gradient was generally positive as expected for a combined soil and canopy respiratory source. The nocturnal CH_4 gradients observed in the canopy were not principally positive or negative, but more evenly distributed (Fig. 8g). Under stable nighttime conditions usually associated with katabatic flow, the Niwot Ridge forest experiences vertical flow stratification. Tracer experiments and isotopic studies have shown under these stable conditions the air within the vegetation canopy does not mix with air near the soil, and so canopy air should be indicative of canopy physiological processes (Yi et al., 2005; Schaeffer et al., 2008a). The study of Schaeffer and colleagues used several meteorological parameters to indicate these conditions; here we use only the friction velocity u^* . The within-canopy CO_2 gradient on very stable nights was small and positive ($4.4 \pm 2.5 \mu\text{mol mol}^{-1} \text{m}^{-1}$), indicating a respiratory source. The nocturnal CH_4 gradient under very stable conditions was small and positive ($2.0 \pm 2.7 \text{nmol mol}^{-1} \text{m}^{-1}$, Fig. 8h), and also indicated a possible weak canopy source at times. This methane source cannot be attributed to the local soils, since they were clearly a CH_4 sink (Fig. 7).

There are several possible explanations for the apparent nocturnal CH_4 source within the canopy. Aerobic foliar emission at night seems unlikely since UV radiation ap-

pears to be required for plant emission (McLeod et al., 2008; Vigano et al., 2008; Messenger et al., 2009). Plants apparently can emit CH_4 without UV radiation, but at very low rates unless exposed to extremely high temperatures (Vigano et al., 2008) or following foliar injury (Wang et al., 2009), neither of which apply in this case. In a very wet forest with thick moss mats, epiphytes, and termites in the canopy, anaerobic microbial emission is possible (do Carmo et al., 2006), but our forest has a dry canopy except immediately after rain events, and no similar organic microbial habitat within the canopy or termites. The possibility exists for deep soil anaerobic microbial CH_4 production, followed by transport through the stem and out through the stomata, but this mechanism is highly speculative at present except for flood-adapted species that possess aerenchyma (Rusch and Rennenberg, 1998; Terazawa et al., 2007; Megonigal and Guenther, 2008). The evening wind direction reversal from somewhat-polluted to relatively-cleaner flow at Niwot Ridge (Fig. 4) could leave lower CH_4 above the canopy than within at night, and the apparent CH_4 “source” in the canopy could simply reflect this external forcing. We attempted to find evidence for this scenario in our data based on several analyses of wind speed and direction, continuity of wind direction, and storage fluxes, but were unsuccessful. Wetlands in the region are known anaerobic sources of CH_4 (Wickland et al., 2001; Chimner and Cooper, 2003). The most likely reason we observed a positive CH_4 gradient in the canopy under very stable conditions (Fig. 8h) is due to horizontal transport from upwind wetland sources. There are several ponds and boggy areas in the predominant nighttime flow direction

within 1–3 km distance. The flow stratification pattern on stable nights already discussed (Yi et al., 2005; Schaeffer et al., 2008a) probably leads to horizontal transport from these sources.

Within the daytime canopy under high UV radiation, photosynthetic drawdown of CO₂ was apparent (Fig. 8b), and CH₄ gradients were small and usually negative (Fig. 8f). As discussed previously, the sign of the gradient within the canopy in the short term might not correctly reflect the source/sink nature of canopy physiological processes. Shown in Fig. 9 are the frequency distributions of daytime excess CO₂ and CH₄ within the canopy. Generally, the daytime excess indicated sinks for both gases, for all daytime periods, and particularly for high UV periods. On average for the 62 profiles with the highest UV irradiance, the CH₄ mole fraction within the canopy was $-0.93 \pm 4.0 \text{ nmol mol}^{-1}$ (mean \pm SD of the distribution in Fig. 9d) lower than above the canopy (Fig. 9d). The mean CO₂ drawdown by the canopy photosynthetic sink was $-0.8 \pm 1.3 \text{ } \mu\text{mol mol}^{-1}$. While these means were not different from zero, both distributions were strongly negatively skewed, implying a canopy (or possibly soil, in the case of CH₄) sink for both gases.

The fact that negative CO₂ gradients (Fig. 8b) and negative daytime CO₂ excess (Fig. 9b) were detectable in the canopy under high UV irradiance causes us to conclude that turbulence was not sufficient to mask the CO₂ sink. Clear evidence of net photosynthesis in the canopy was obtained, and this provides some confidence that a foliar CH₄ emission source, if present, would be detected (or smaller than our method can resolve). As discussed, it is possible that the canopy could be a source of CH₄ that is masked by a stronger soil CH₄ sink, but the data in Figs. 7–9 strongly indicate that the vegetation canopy at Niwot Ridge is at most a very weak CH₄ source, and the Niwot Ridge forest during our study acted as a net sink for atmospheric methane.

To our knowledge only Vigano et al. (2008) have attempted to find aerobic CH₄ emission from conifers, and only from the single tree species *Pinus ponderosa*, which was found to be an emitter. Our study provides evidence that aerobic foliar CH₄ emission from the dominant conifer species at Niwot Ridge (*Pinus contorta*, *Picea engelmannii*, and *Abies lasiocarpa*) is minimal, and likely does not need to be accounted for in large-scale emission inventories.

We caution, though, that Wang et al. (2009) have recently shown that of ten congeneric *Artemisia* species studied, only one species emitted CH₄ without physical injury to the plant. Hence, demonstrated emission by *P. ponderosa* under UV irradiation (Vigano et al., 2008) does not necessarily imply that all conifers emit CH₄. Similarly, the absence of evidence for strong foliar emission in our forest does not necessarily mean that no conifer forests emit methane from the trees. Given the growing body of literature documenting aerobic methane emission from plants, more work is clearly needed to assess the general phenomenon of methane emission across plant

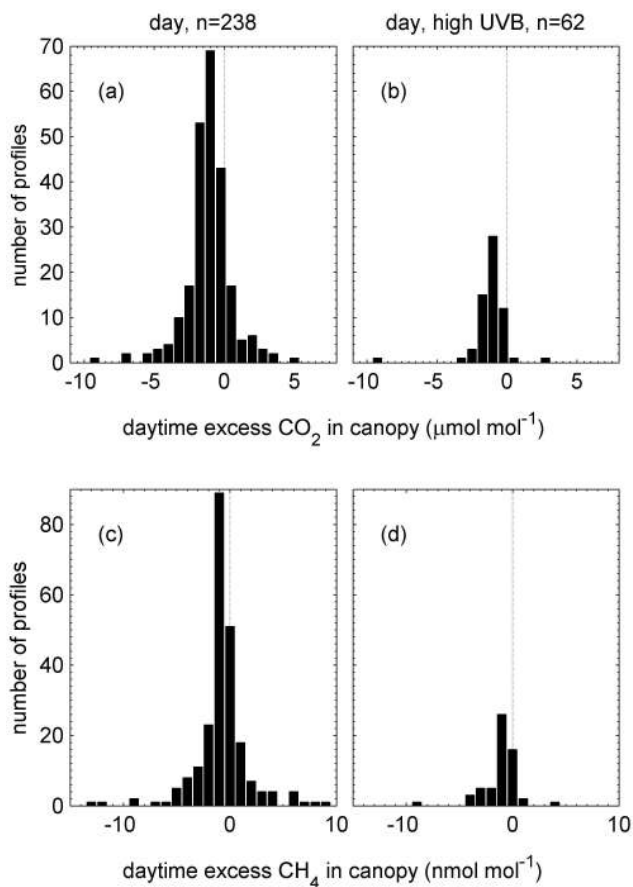


Fig. 9. Frequency distributions of the daytime excess CO₂ (top) and CH₄ (bottom) within the canopy, for daytime periods only. Daytime excess was calculated as the mean of the mole fractions at the 3 canopy inlets (7, 9, and 11 m) minus the mole fraction above-canopy (21.5 m) for a given 30-min period. Negative values indicate lower CO₂ or CH₄ within the canopy compared to above it (a sink, such as during strong photosynthetic drawdown of CO₂). Positive values are suggestive of a canopy source. Data are divided into all daytime periods and high UVB periods as in Figs. 7 and 8.

taxa, and to determine the general ecological significance of plant emission of methane in intact ecosystems.

4 Summary

Patterns of biosphere-atmosphere methane and carbon dioxide exchange were examined in a high-elevation subalpine forest using canopy profile observations over a 6-week summer period. Variability in the CH₄ mole fraction of forest air was related to a mountain-plain wind system and influenced by air mass transport from the Denver, Colorado, urban area. Local soils were a persistent CH₄ sink during the study period. No evidence was found for substantial emission of CH₄ from the foliar vegetation, even under high UV irradiance. We cannot rule out the possibility of a canopy source given

the observed soil sink, but the magnitude of such a vegetation source in this forest must be very weak if it exists at all.

Acknowledgements. Thanks to Ed Dlugokencky and Patricia Lang for calibrating standards and CH₄ flask measurements, Paul Novelli for CO flask measurements, Andrew Crotwell and Doug Guenther for loaning scientific equipment, and Sean Schaeffer for assistance in the field. We are grateful to Patrick Disterhoft and the NEUBrew project members for freely providing UV data. Several colleagues contributed helpful ideas and discussion at AGU including Pat Crill, Gunnar Schade, Torben Christensen, and Todd Rosenstiel. Primary funding for this work was provided by U.S. Dept. of Energy, Office of Science (BER), Grant No. DE-FG02-04ER63904 (to DRB and JBM). We acknowledge additional support from the National Oceanic and Atmospheric Administration Ernest F. Hollings Undergraduate Scholarship Program (to MER) administered by Oak Ridge Institute for Science and Education under a contract between the U.S. Department of Energy and Oak Ridge Associated Universities, from a grant from DOE National Institute for Climatic Change Research (NICCR) Western region (to RKM), and NSF grant DEB 0423662 to the Niwot Ridge LTER program.

Edited by: G. Wohlfahrt

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