

State-to-state modeling of a recombining nitrogen plasma experiment

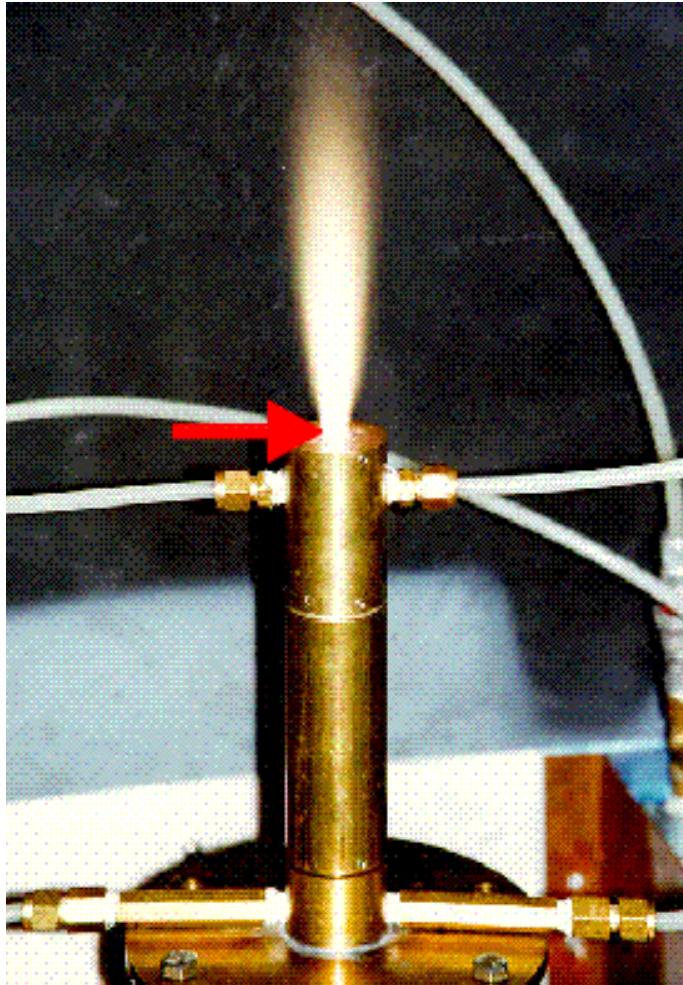
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and
Stanford University, Mechanical Engineering Dept.

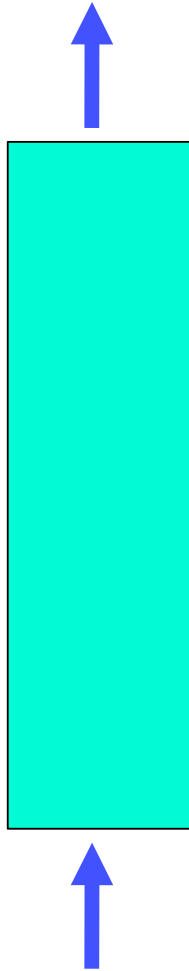


*Symposium in Honour of Prof Mario Capitelli on the Occasion of His 70th Birthday
Università degli Studi di Bari - Aldo Moro
January 31- February 1, 2011*

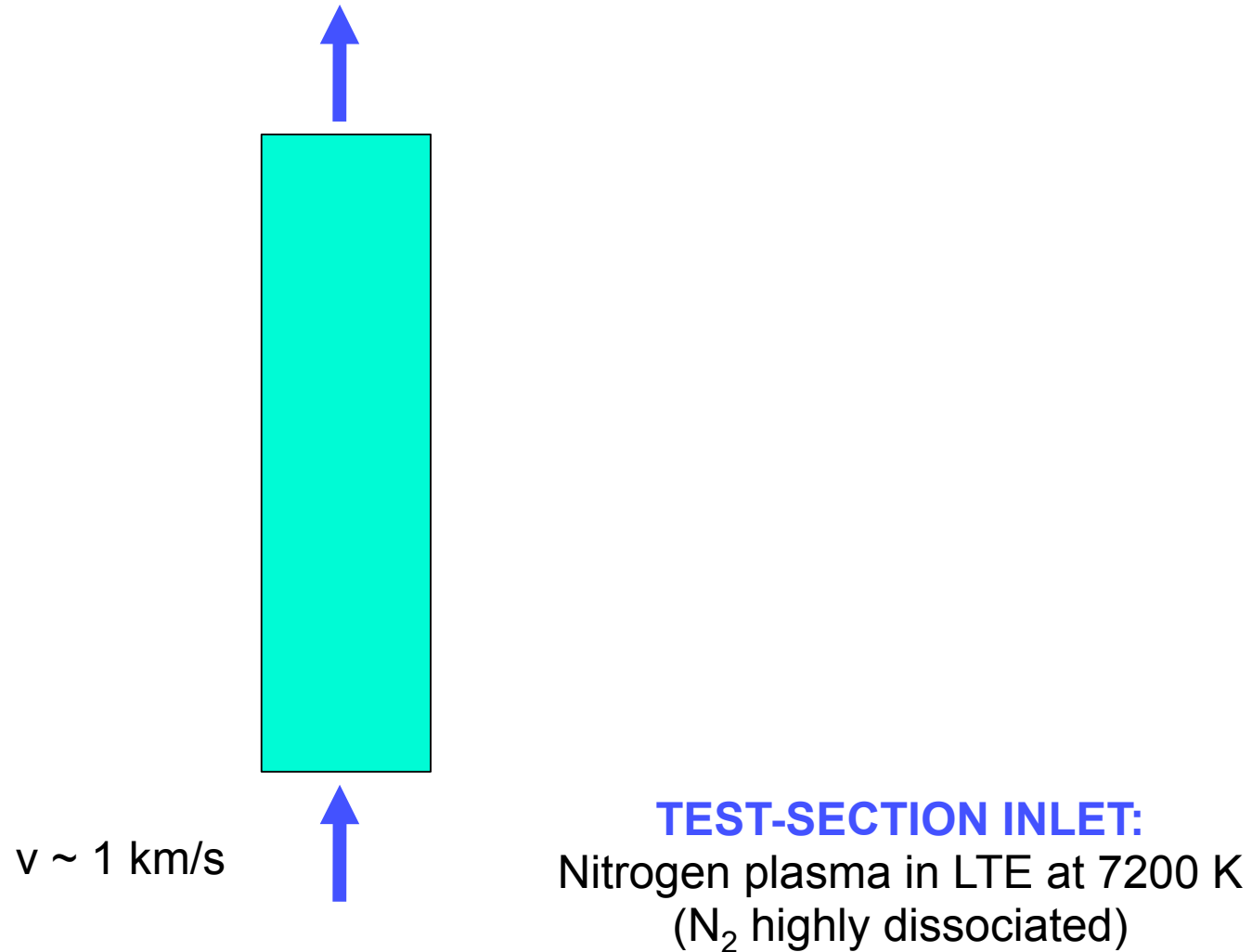
Recombining nitrogen plasma at atmospheric pressure



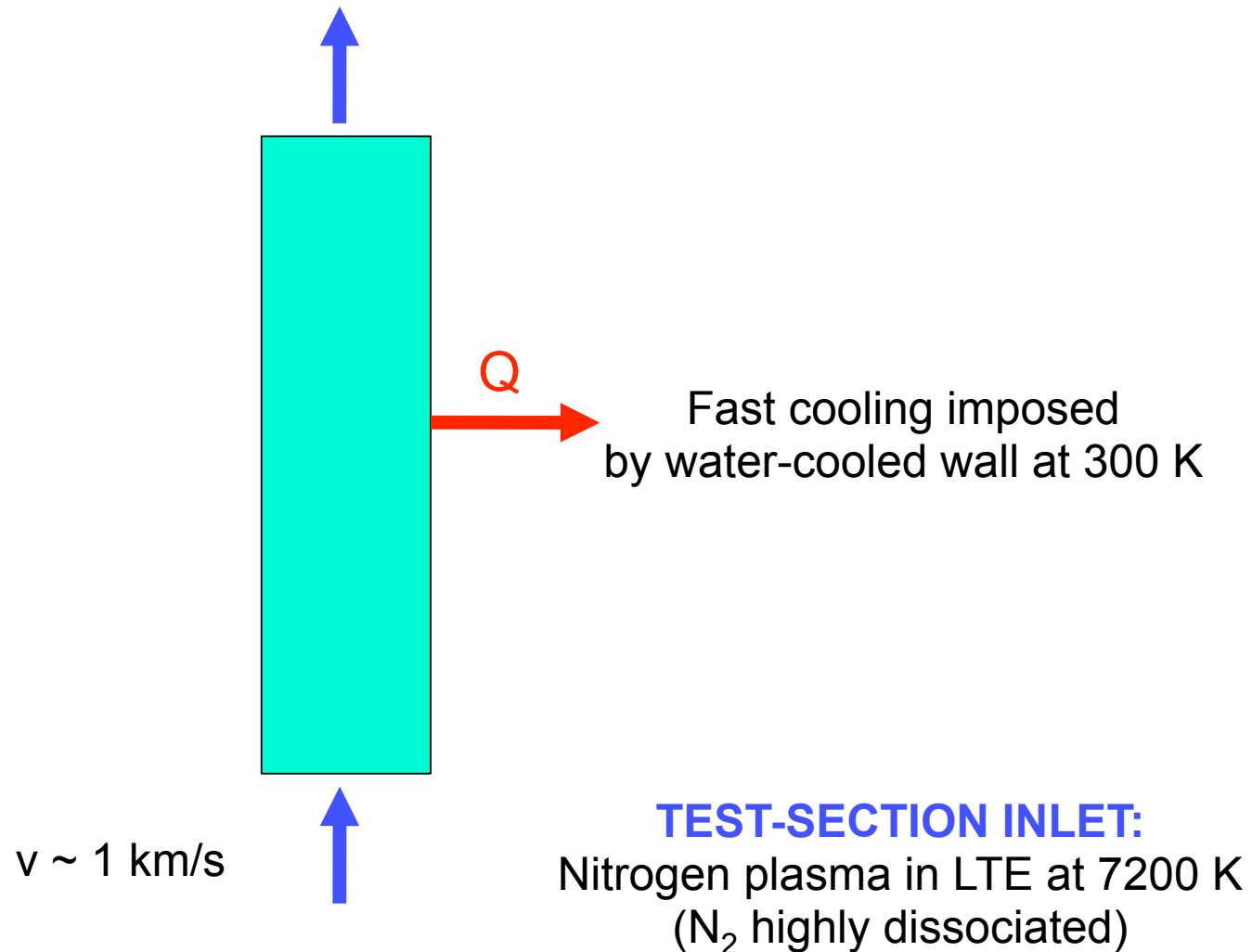
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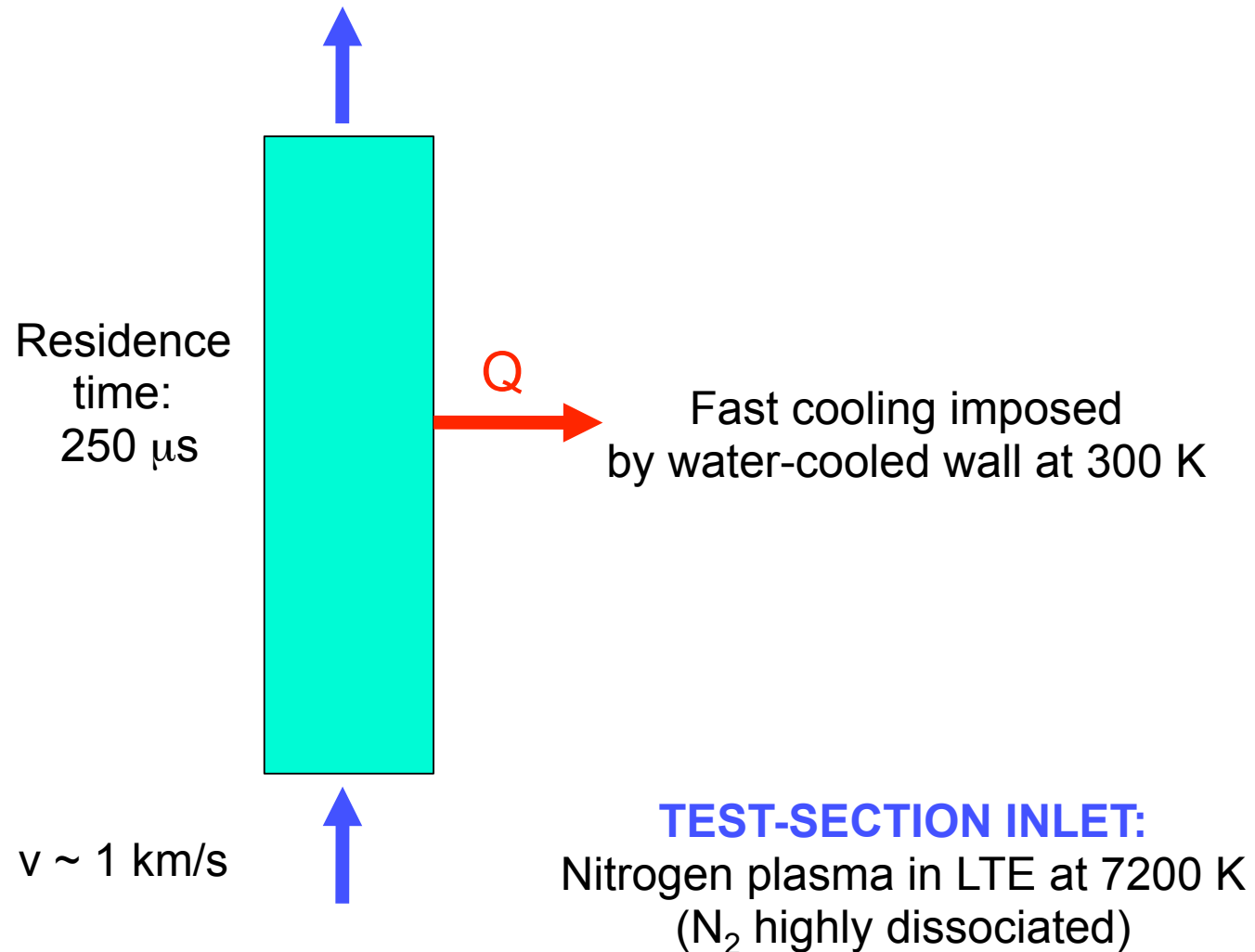
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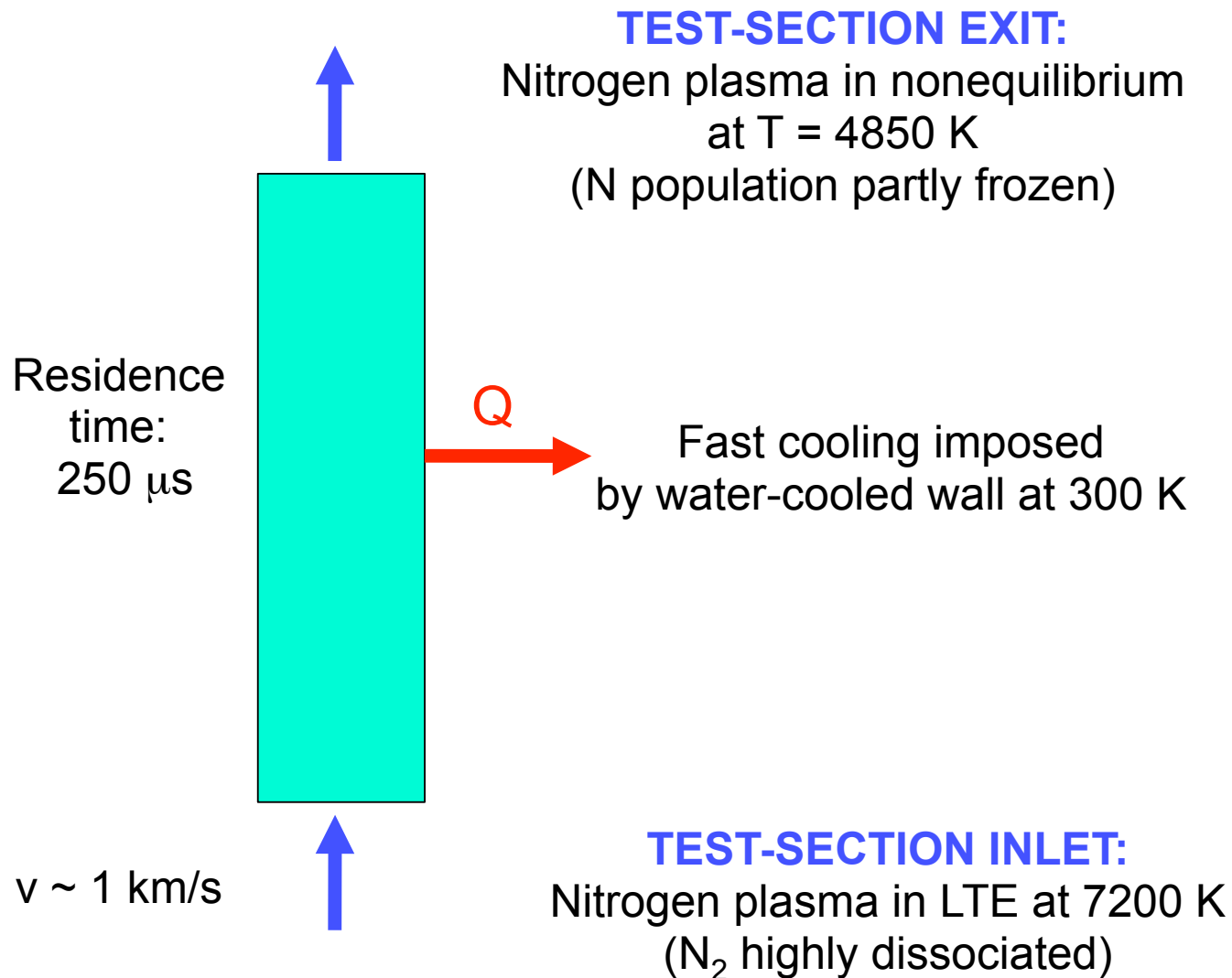
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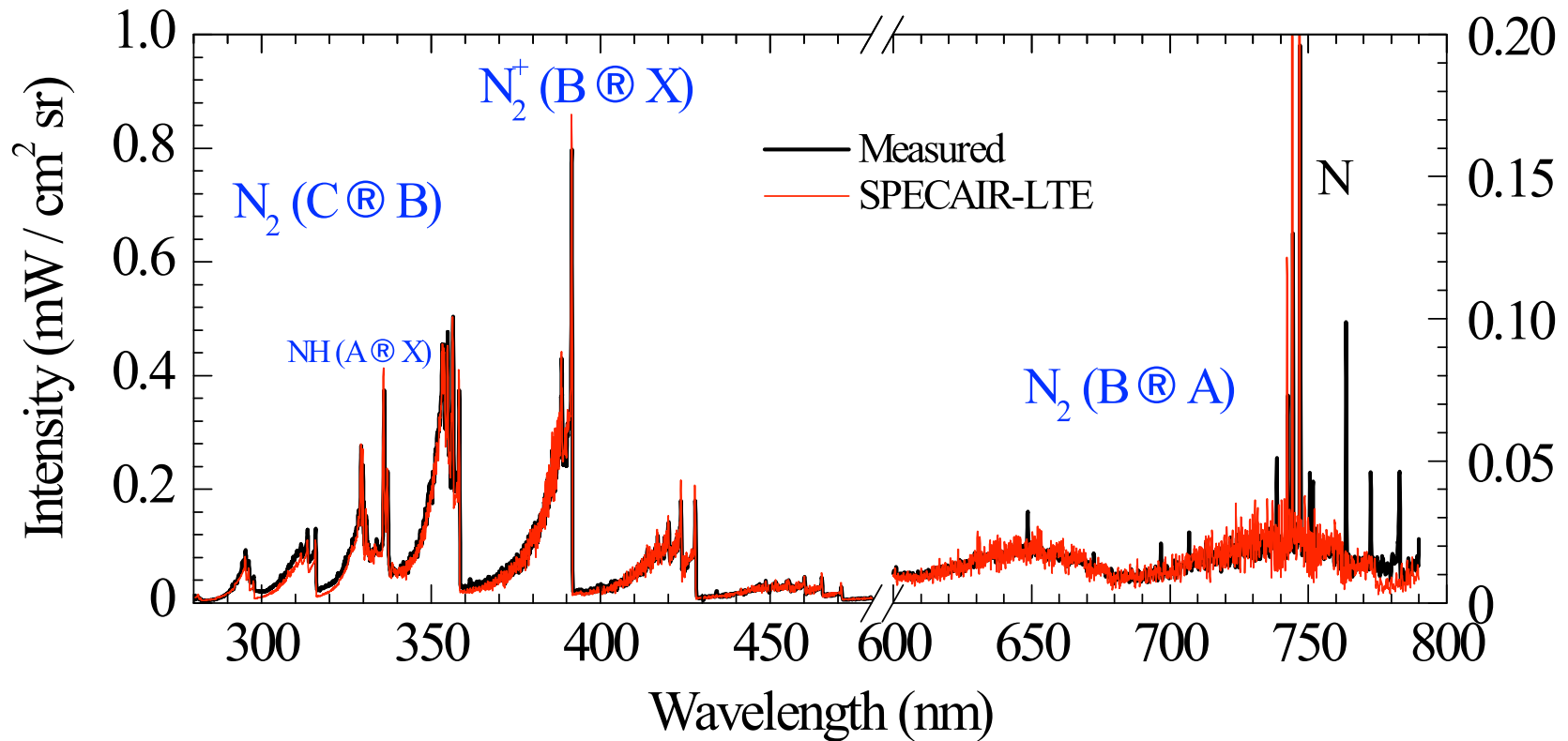
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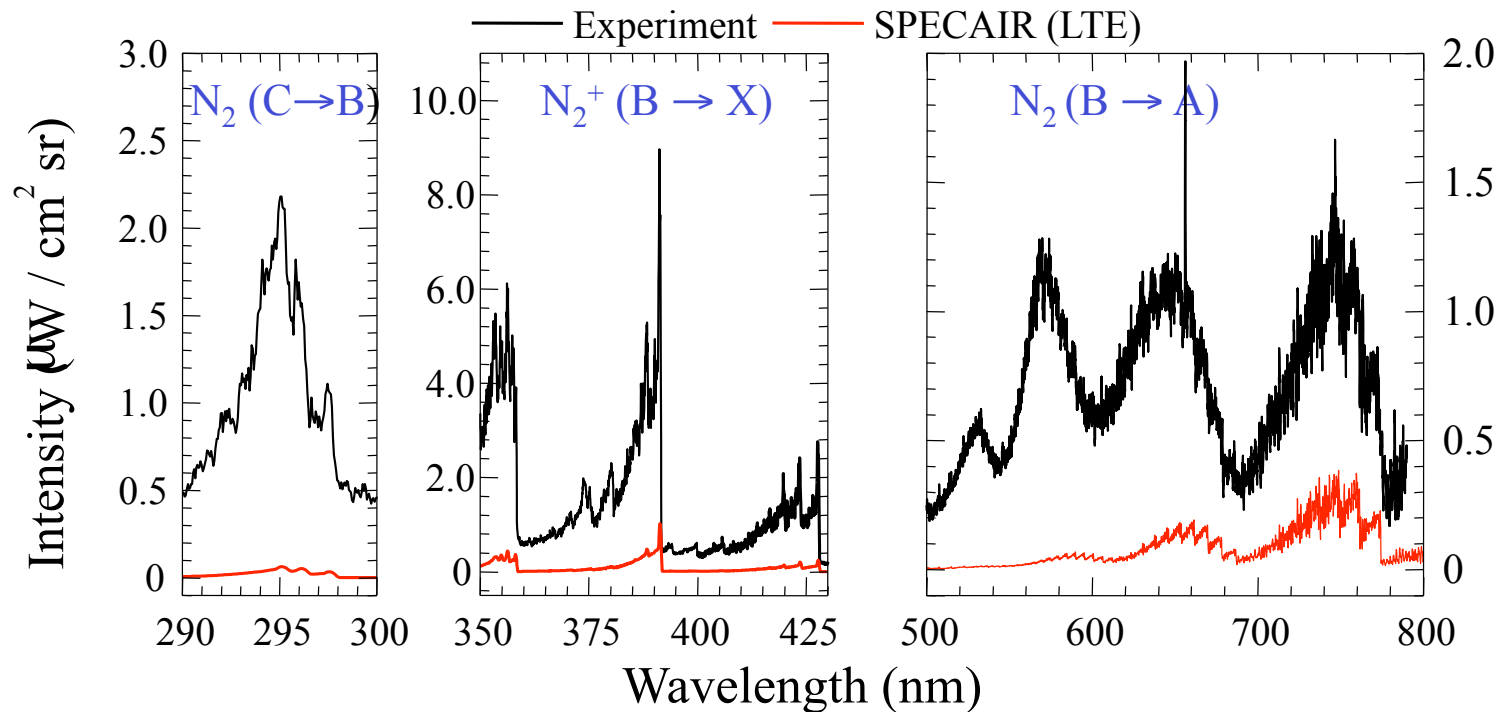


Emission Spectrum at Test-Section Inlet



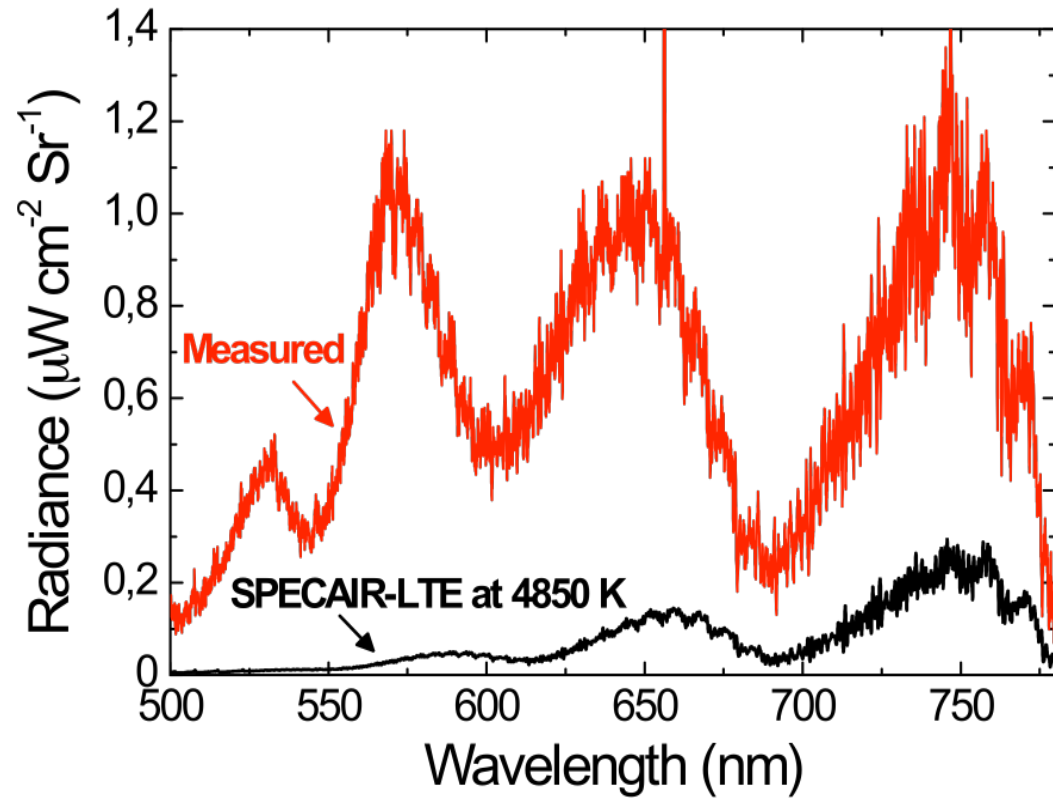
- Test-section inlet: Plasma close to LTE

Emission Spectrum at Test-Section Exit

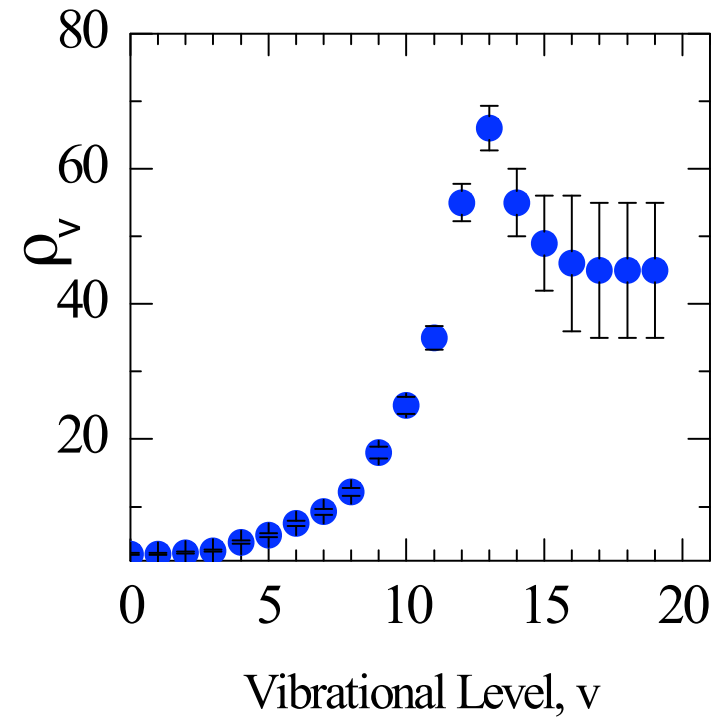
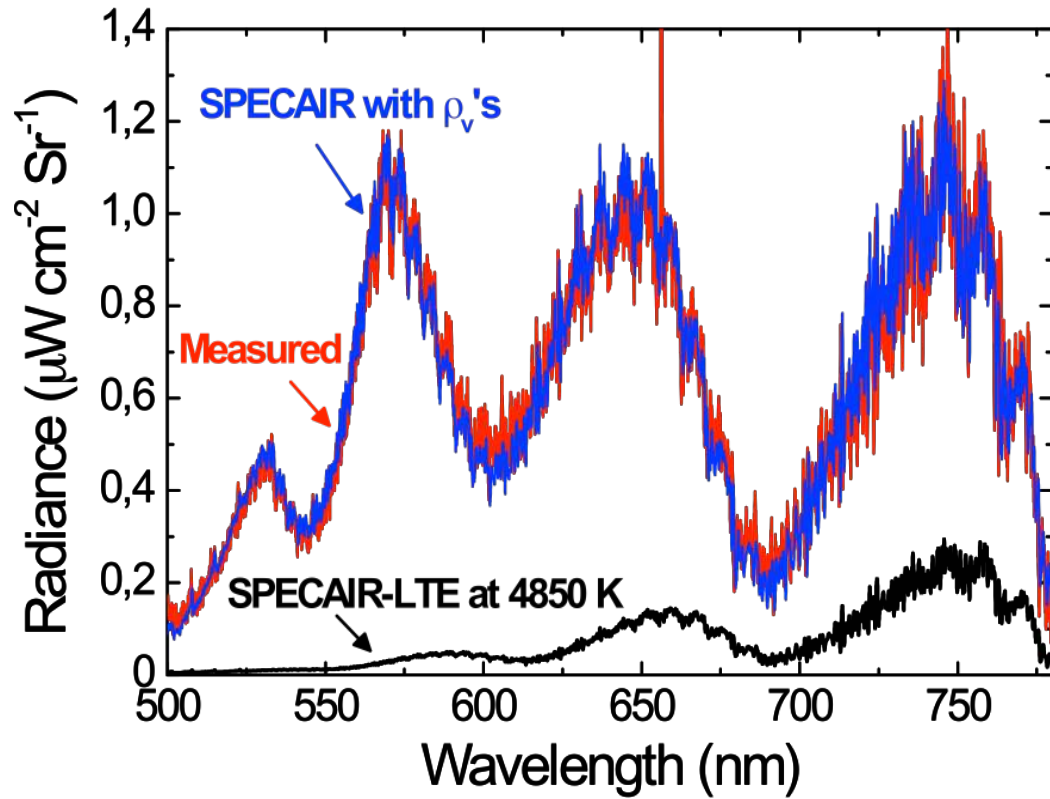


- Emitting states (N_2^+ B, N_2 C, N_2 B) are overpopulated with respect to LTE.

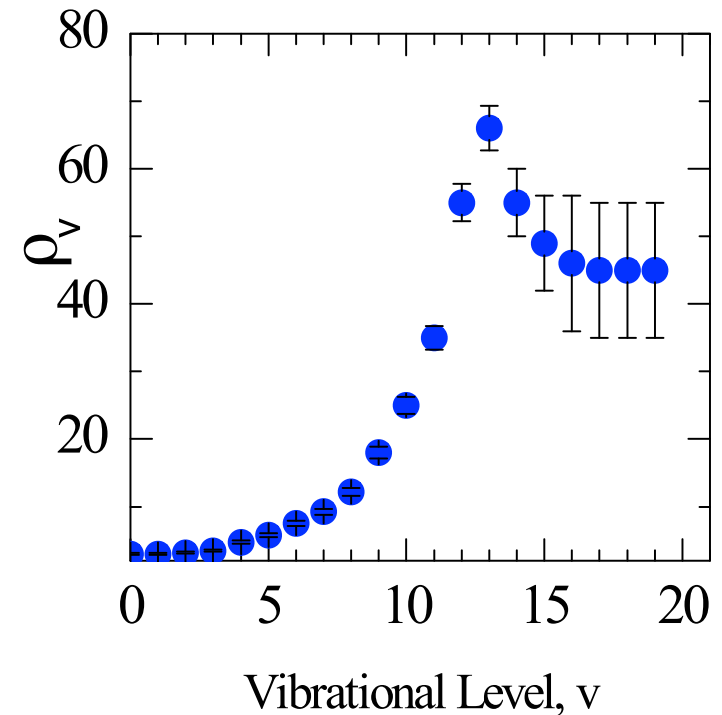
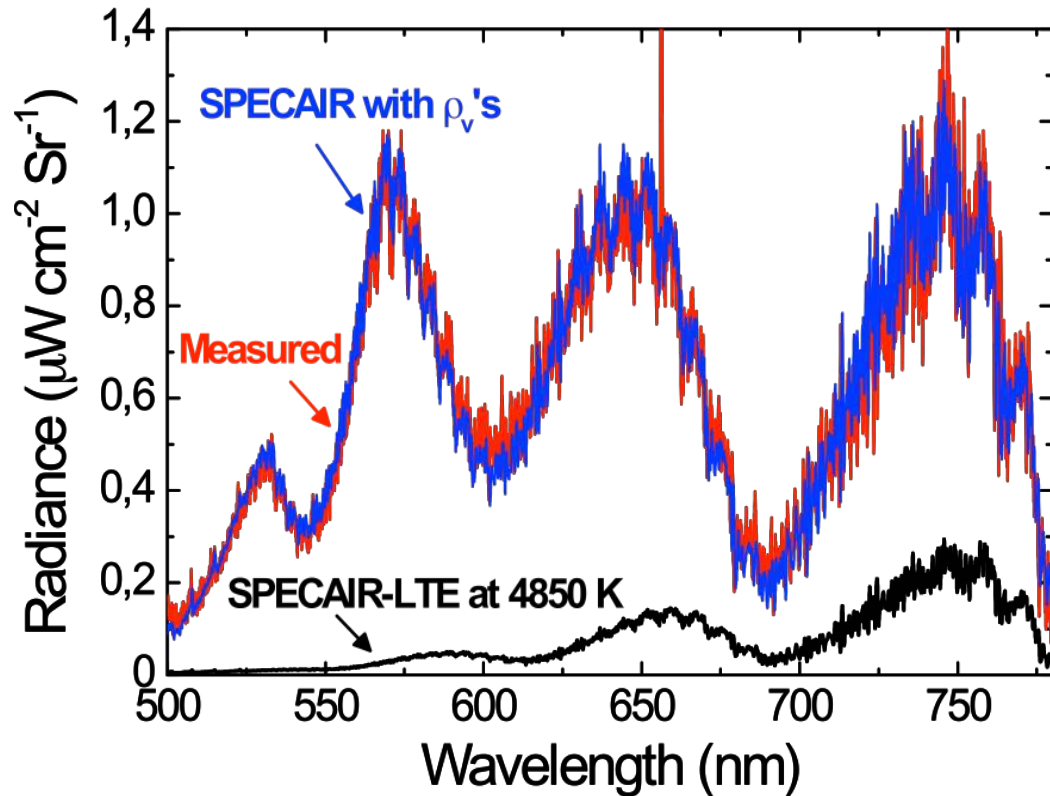
Analysis of the N₂ B-A spectrum at exit of test-section



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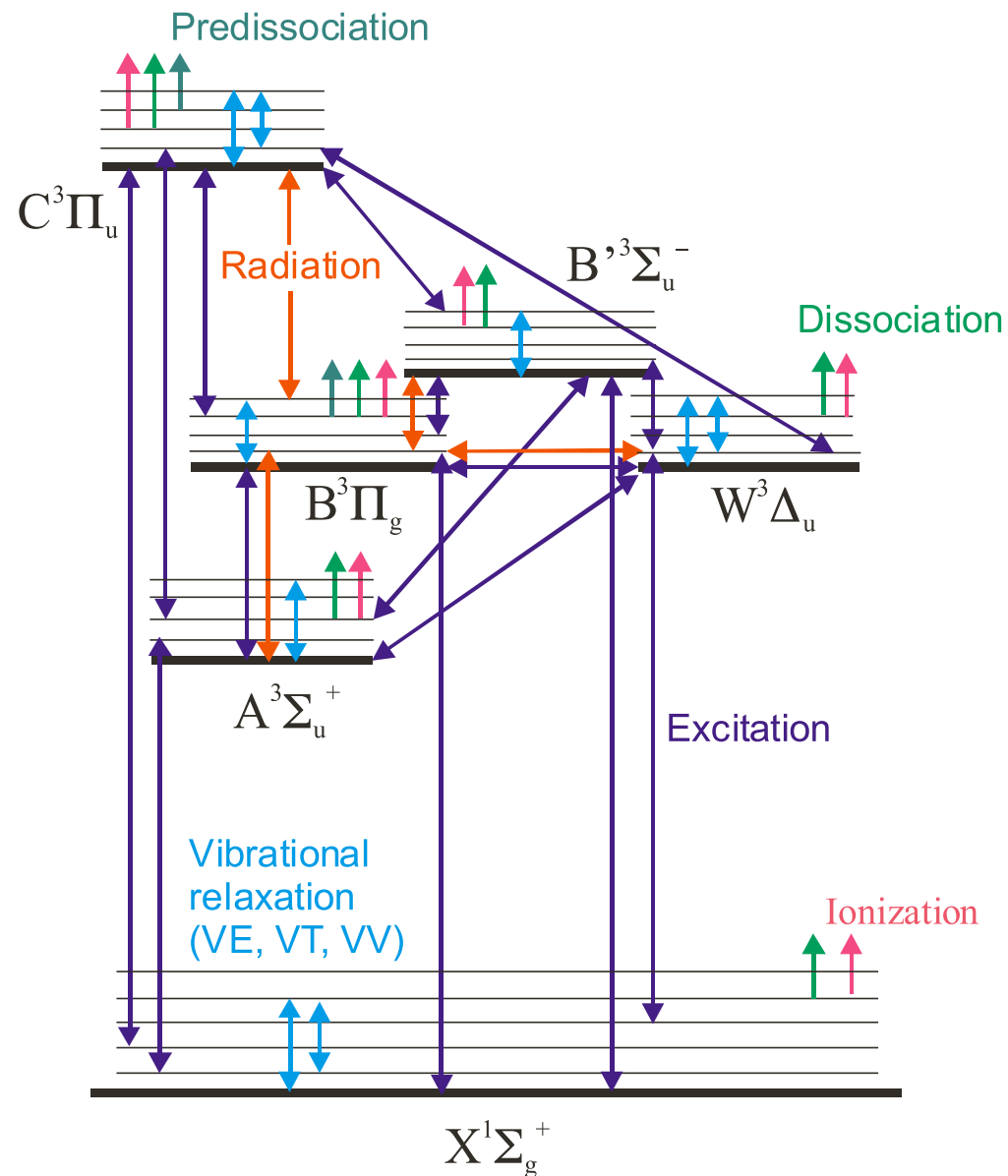


- Non-Boltzmann distribution of the vibrational levels
⇒ CR model needed to analyze experiment

Vibrationally Specific State-to-State CR Model of nitrogen

- Predict nonequilibrium populations of
 - N (22 electronic levels)
 - N⁺ (1 electronic level)
 - N₂ X (v=0-47), A (0-27), B (0-30), W(0-37), B' (0-41), C (0-4)
 - N₂⁺ X (v=0-52), A (0-63), B (0-24)
 - Electrons⇒ 357 states + electrons
- Input parameters
 - Number density of N nuclei and electrons
 - P
 - T_e: kinetic temperature of electrons (Maxwellian)
 - T_g: kinetic temperature of heavy species (Maxwellian)
 - T_{rot} = T_g

Vibrationally Specific State-to-State CR Model of nitrogen



~46,000 state-to-state reactions

Collisional-Radiative Model of N₂

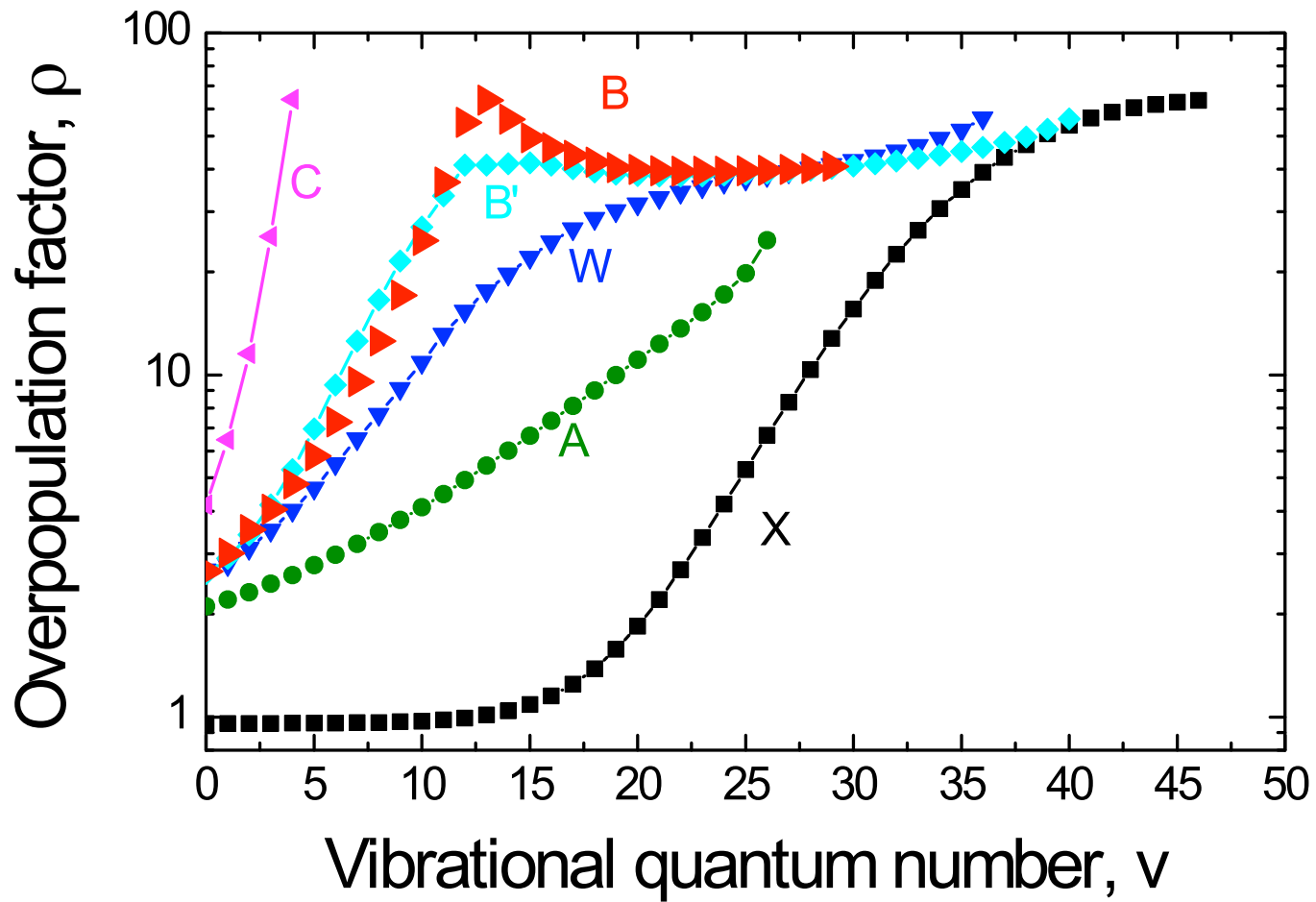
- **Vibrational state-specific reactions considered:**
 - Electron-impact excitation of N, N₂, N₂⁺
 - Electron-impact ionization of N and N₂
 - Electron and heavy-particle impact dissociation of N₂ and N₂⁺
 - Dissociative recombination of N₂⁺
 - Charge exchange: N₂⁺ + N ⇌ N₂ + N⁺
 - VE: electron-impact vibrational excitation of N₂ X
 - VT: vibrational-translational transfer
 - VV: vibrational-vibrational transfer
 - Radiation
 - Predissociation

Assumptions for CR simulations at test-section exit

- $P=1$ atm and $T=4850$ K
- $n_e=7.3 \times 10^{13}$ cm⁻³ (measured value)
- n_N adjusted to provide best match to experiment
- EEDF is Maxwellian
(Capitelli et al., JTHT, 12, 478-481, 1998)

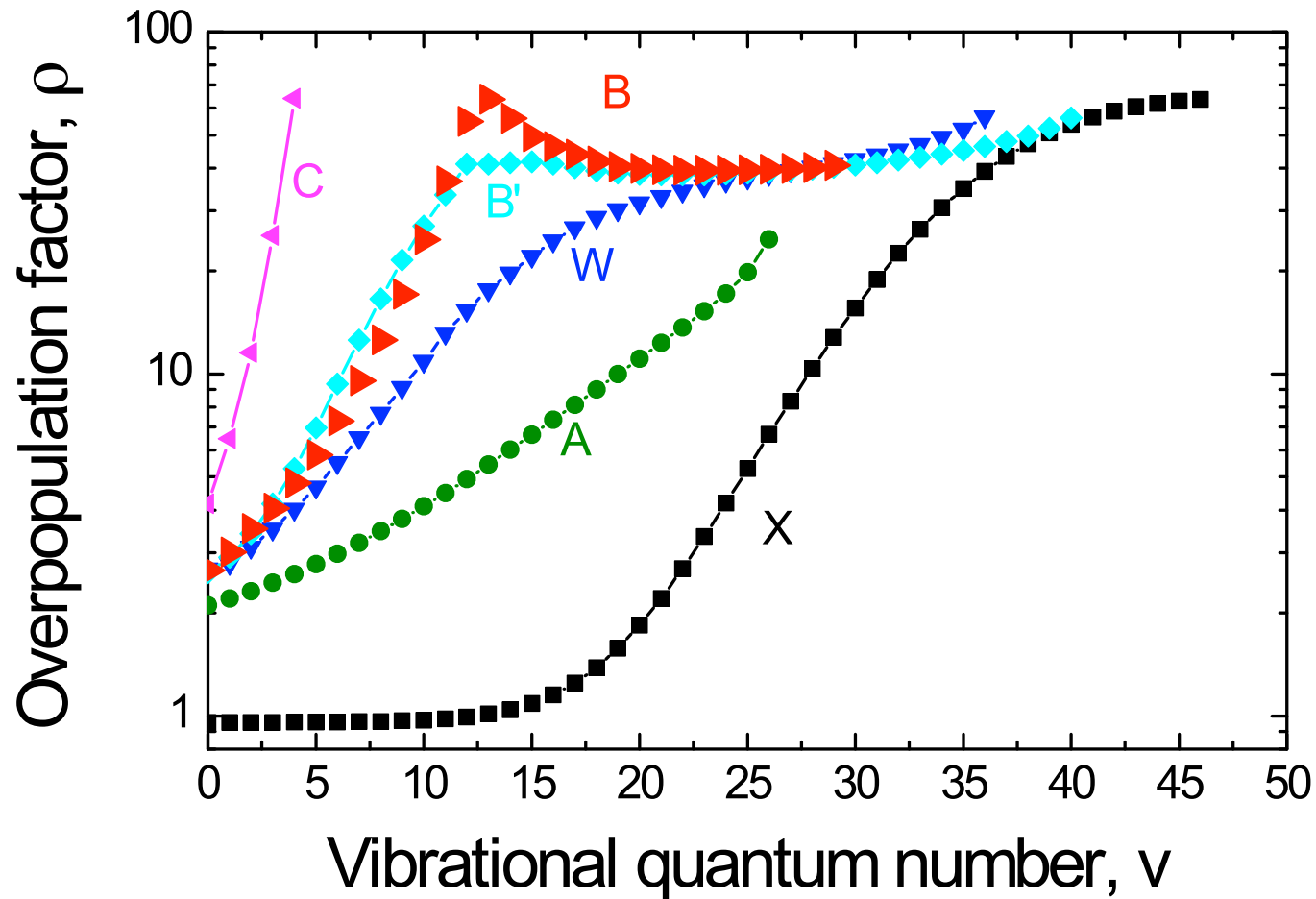
CR model results: vibronic distributions

$T = 4850 \text{ K}$, $P = 1 \text{ atm}$, $\rho_e = 180$, $\rho_N = 8.1$



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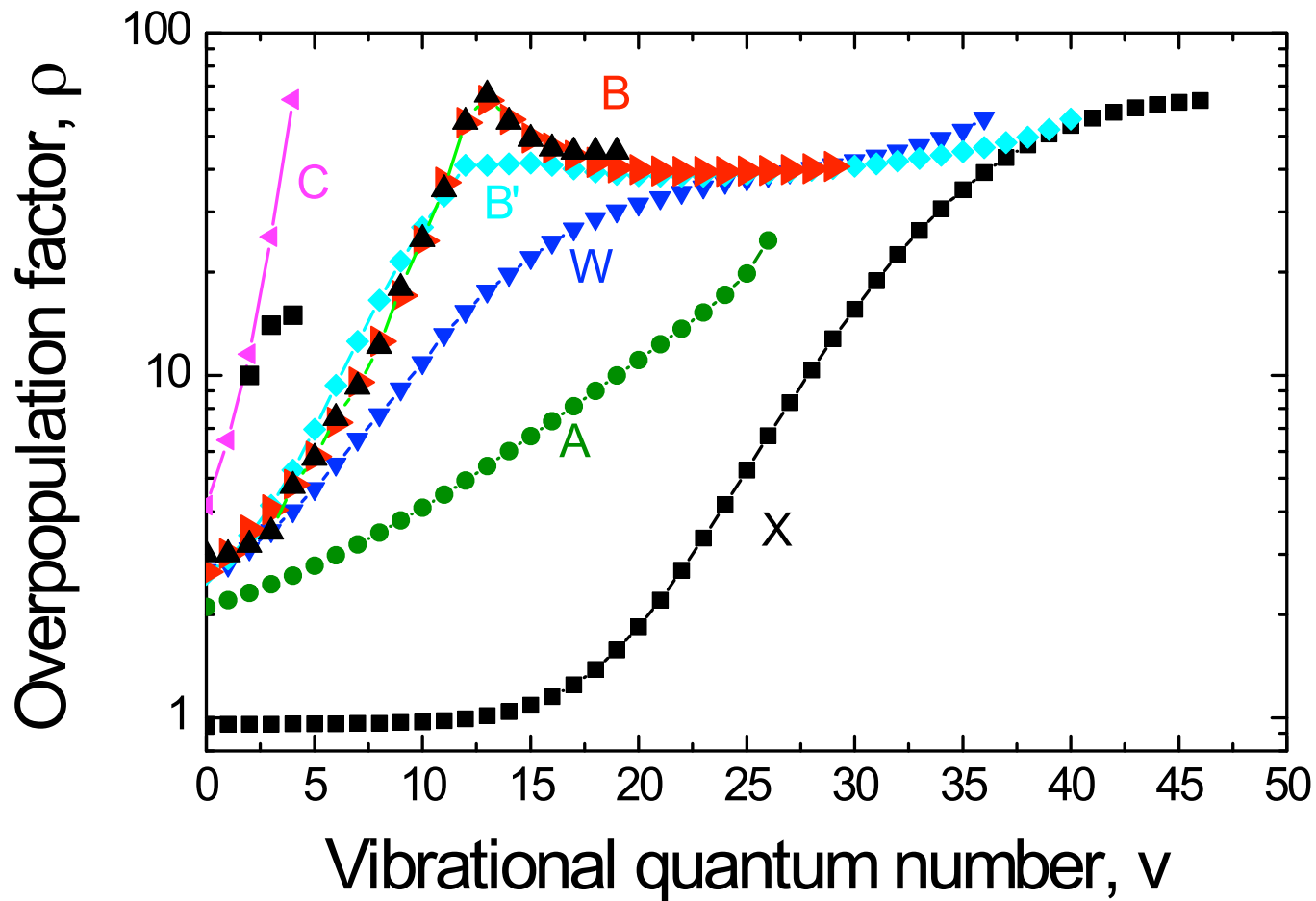
$T = 4850 \text{ K}$, $P = 1 \text{ atm}$, $\rho_e = 180$, $\rho_N = 8.1$



- Peak at $v=13$ is due to inverse predissociation: $\text{N} + \text{N} \rightarrow \text{N}_2 (\text{B}, v=13)$

CR model results: comparison with experiment

$T = 4850 \text{ K}$, $P = 1 \text{ atm}$, $\rho_e = 180$, $\rho_N = 8.1$



- Peak at $v=13$ is due to inverse predissociation: $\text{N} + \text{N} \rightarrow \text{N}_2 (\text{B}, v=13)$

Conclusions

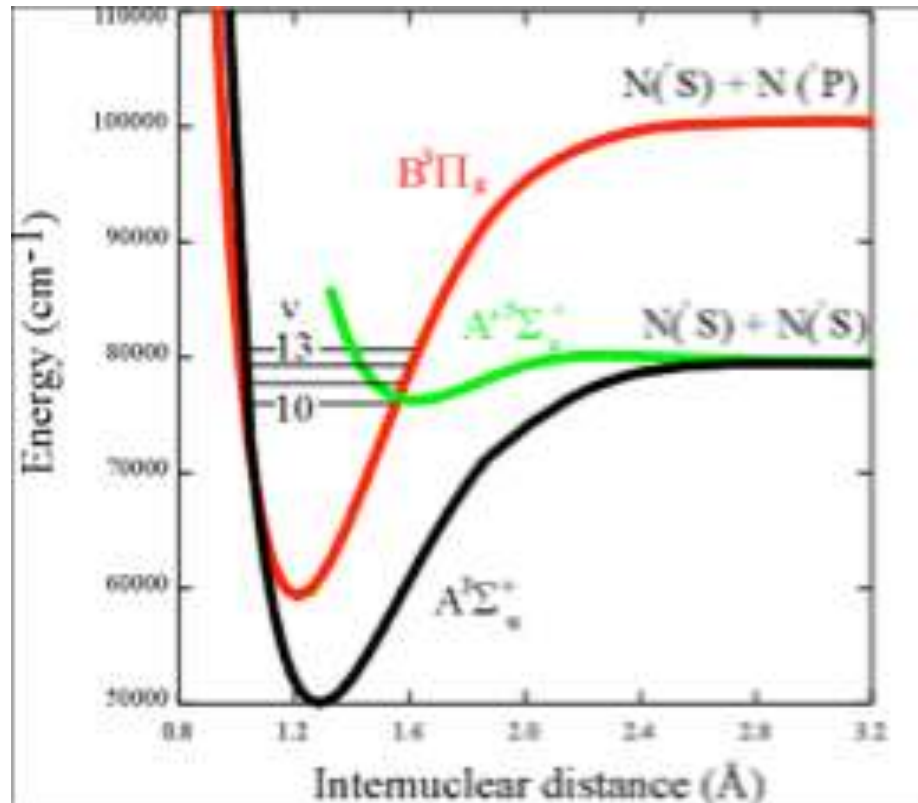
- State-to-state model developed to predict vibrational populations in ground and excited states of nitrogen
- Good agreement with measured vibrational distributions
- Overpopulation of N_2 B, $v=13$ provides a convenient way to measure **absolute densities of N atoms** by emission spectroscopy (application to expanding flows,...)

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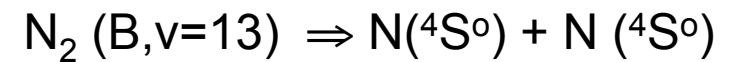
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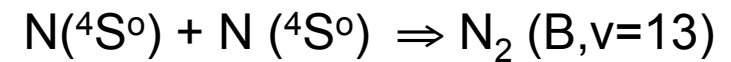
Predissociation/Inverse Predissociation



- **Predissociation:**



- **Inverse Predissociation**



Rate equation for N₂, B, v=13

$$\begin{aligned}
 \frac{dn_{N_2, B, 13}}{dt} &= -n_{N_2, B, 13} \left\{ k_{v=13}^{pred} + \sum_{v''} A_{13v''} + n_M (k_{VT}^{13 \otimes 14} + k_{VT}^{13 \otimes 12}) + n_e k_{ion}^e + \sum_{Y=X, A, B', W, C} n_e k_{B, 13 \otimes Y}^e \right\} \\
 &\quad + n_N^2 k_{v=13}^{inv.pr.} + \sum_{Y', v'} n_{N_2, Y', v'} A_{v'13} + n_{N_2, B, 14} n_M k_{VT}^{14 \otimes 13} + n_{N_2, B, 12} n_M k_{VT}^{12 \otimes 13} + n_e^2 n_{N_2^+} k_{rec}^e \\
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 &= 0 \quad \text{at steady-state}
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Rate equation for N₂, B, v=13

2.6x10⁸ s⁻¹

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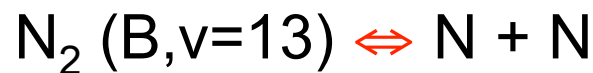
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$$n_{N_2, B, 13} k_{v=13}^{pred} \approx n_N^2 k_{v=13}^{inv.pr.}$$



N atom concentration

Previous analysis: $k_{\text{pred}} n_{\text{N}_2(\text{B},13)} \cong k_{\text{inv.pred.}} (n_{\text{N}})^2$

Always true: $k_{\text{pred}} n_{\text{N}_2(\text{B},13)}^{\text{equil}} = k_{\text{inv.pred.}} (n_{\text{N}}^{\text{equil}})^2$

Thus: $\frac{n_{\text{N}_2(\text{B},13)}}{n_{\text{N}_2(\text{B},13)}^{\text{equil}}} = \left(\frac{n_{\text{N}}}{n_{\text{N}}^{\text{equil}}} \right)^2 \Rightarrow \rho_{\text{N}} = \sqrt{\rho_{\text{N}_2(\text{B},13)}}$

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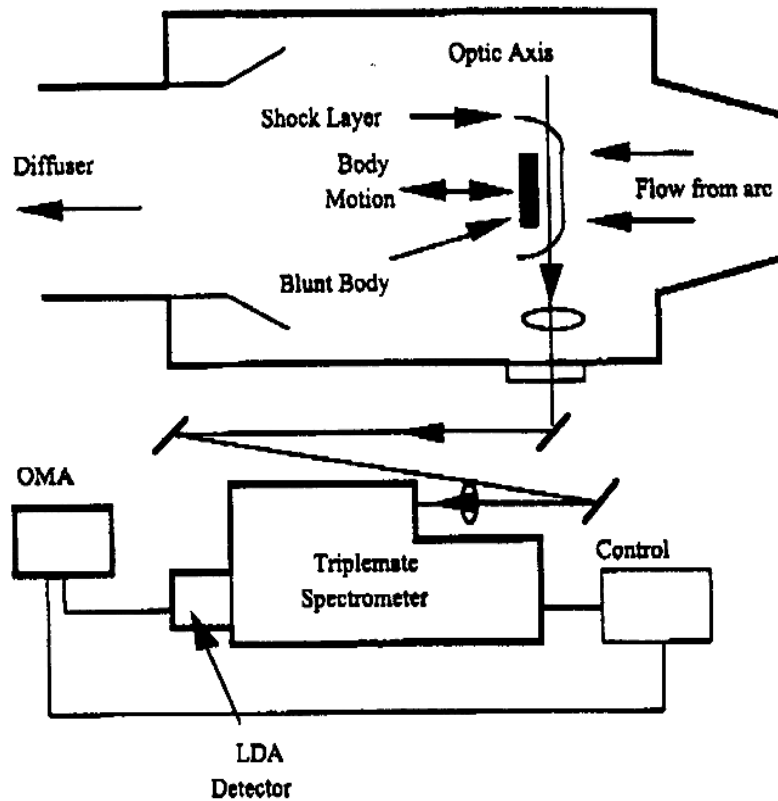
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- Diagnostic tool for ground state N concentration:
 - $\rho_{\text{B},v=13} = 66 \pm 4$ (measured) $\Rightarrow \rho_{\text{N}} = 8.1 \pm 0.3$
 - $[\text{N}]_{\text{eq},4850\text{K}} = 1.8 \times 10^{16} \text{ cm}^{-3}$
- $\Rightarrow [\text{N}] = 1.5 \pm 0.4 \times 10^{17} \text{ cm}^{-3}$

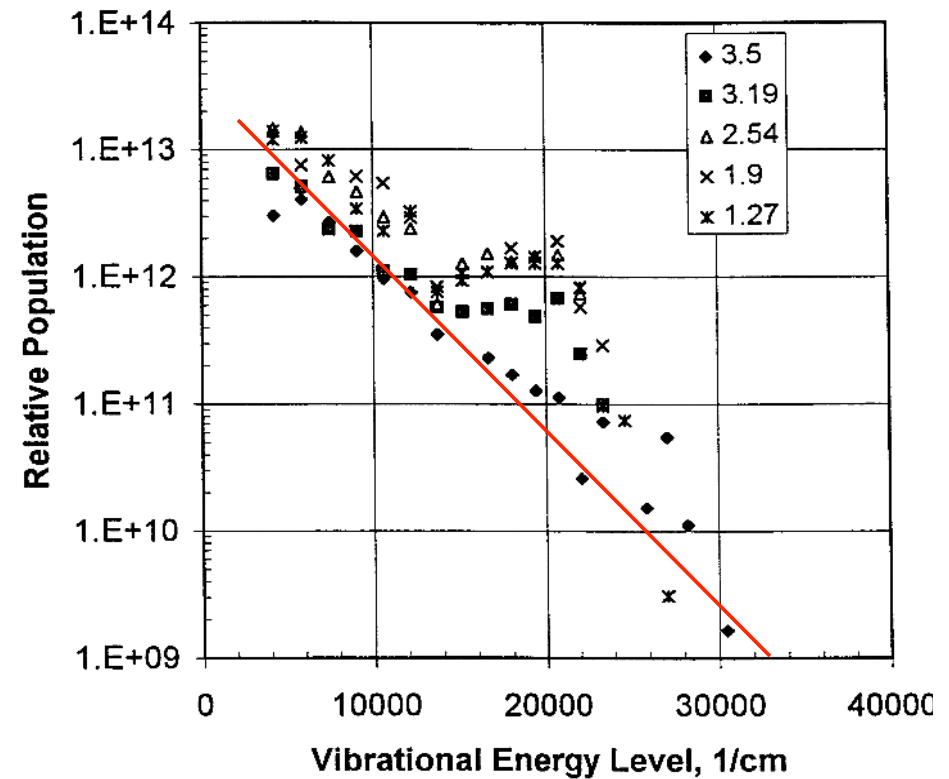
Nonequilibrium distribution of N₂ B state in arcjet measurements at Johnston Space Center

Experimental set-up

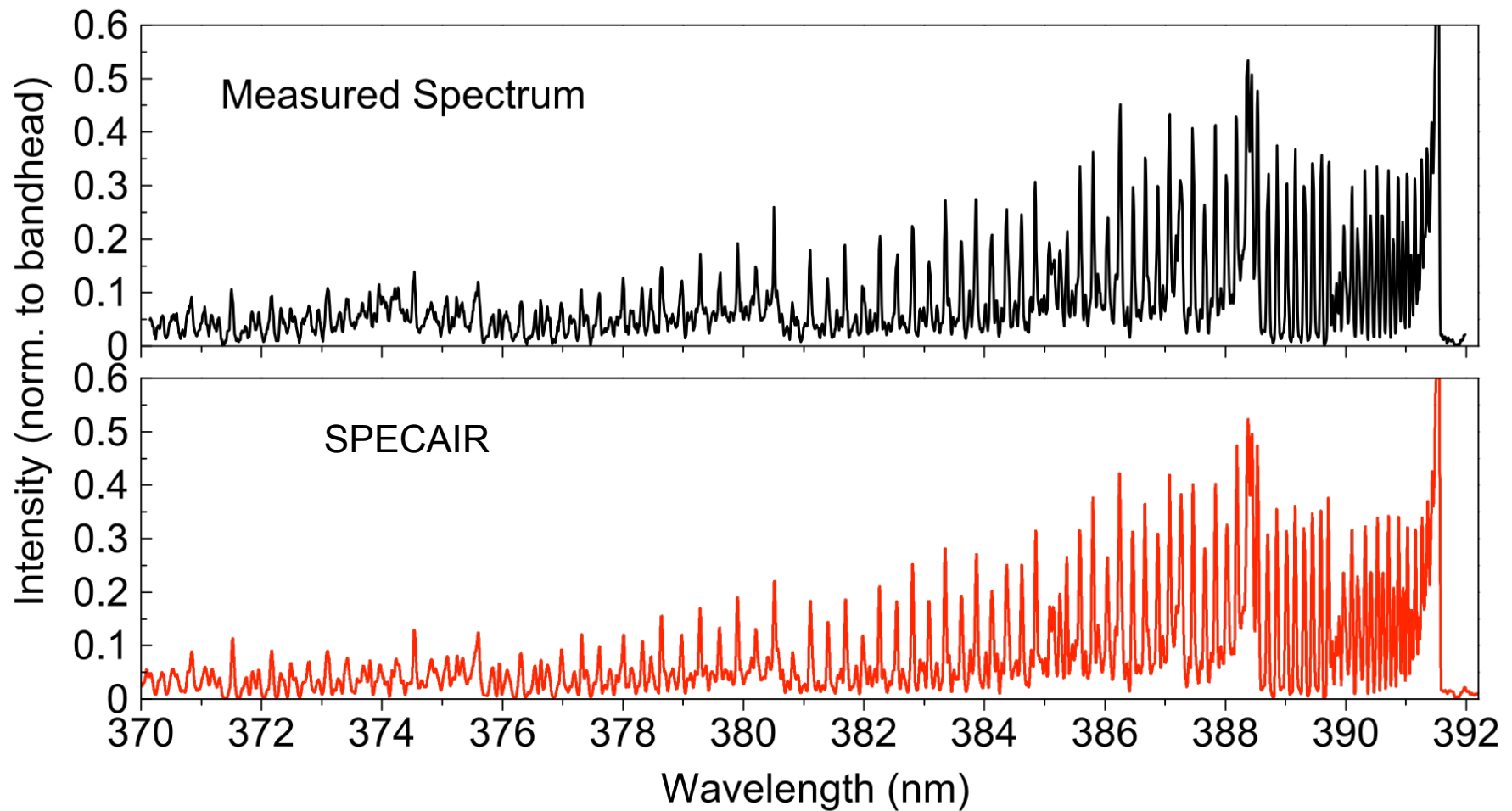


Blackwell, Scott, and Arepalli, JSC, 1997

N₂ B state vibrational populations as a function of distance from body

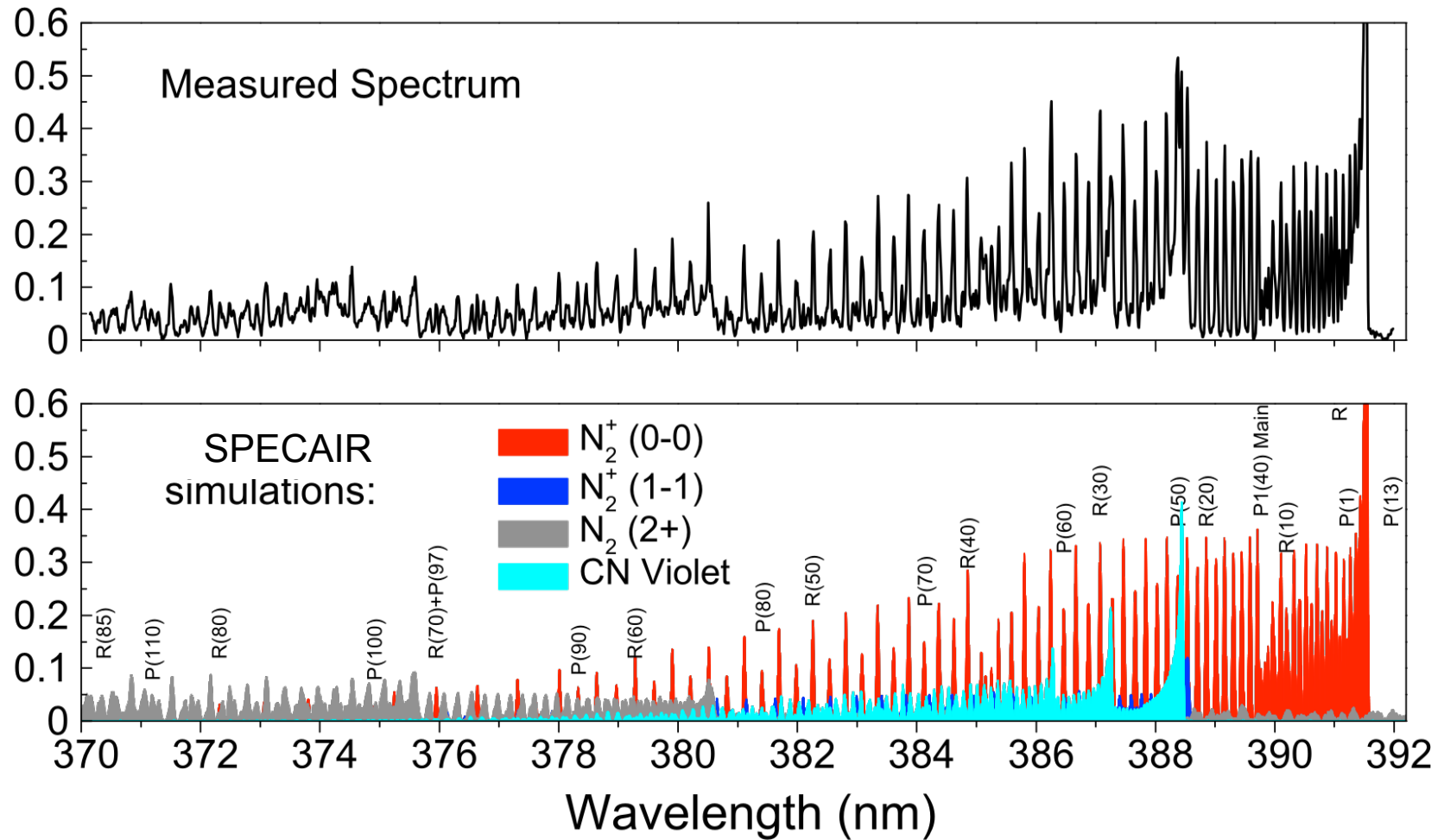


Rotational temperature at 15 cm using N_2^+

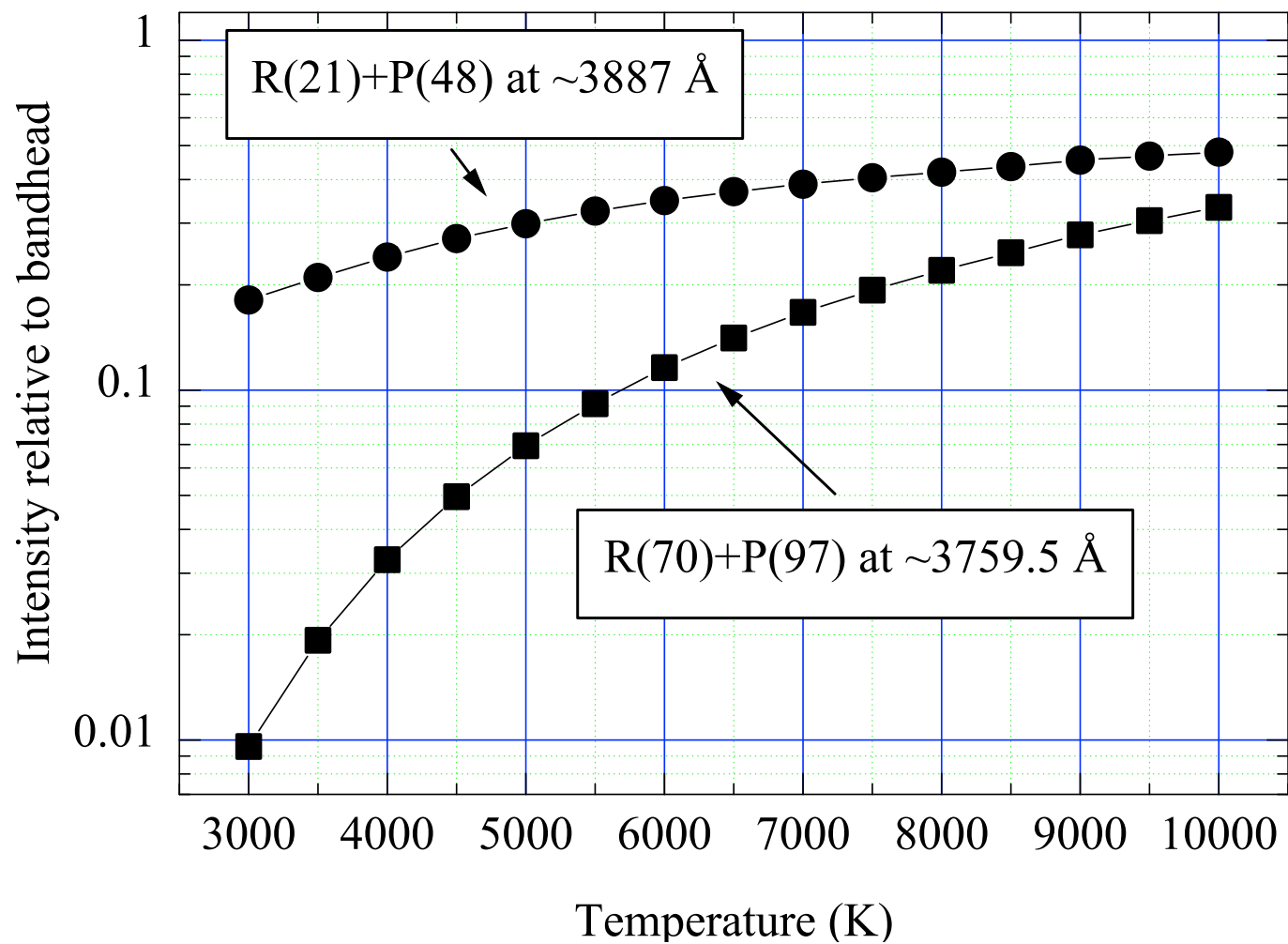


• Best fit: $T_{\text{rot}} = T_{\text{vib}} = 4850 \pm 100 \text{ K}$

Rotational Temperature at 15 cm using N_2^+



Rotational temperature: [R(70)+P(97)] vs. Bandhead



Calculation of vibrational state specific rate coeff.



- Rovibrational state-specific rate coefficient:

$$K_{N_2 X v'' J''}^{N_2^+ A v' J'}(T_e) = \frac{8\pi}{\sqrt{m_e}} \frac{1}{(2\pi k T_e)^{3/2}} \int_0^{+\infty} \varepsilon \sigma_{N_2 X v'' J''}^{N_2^+ A v' J'}(\varepsilon) \exp\left(-\frac{\varepsilon}{k T_e}\right) d\varepsilon$$
$$\downarrow$$
$$\sigma_{N_2 X v'' J''}^{N_2^+ A v' J'} = q_{X v''}^{A v'} \delta_{J''}^{J'} \chi(\varepsilon, \Delta E_{X v'' J''}^{A v' J'})$$

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$$\chi(\varepsilon, \Delta E_{X v''}^{A v' J'}) = 4\pi a_0^2 N \left(\frac{R}{\Delta E_{X v''}^{A v' J'}} \right)^2 \frac{1}{t + u + 1} \left[\frac{\ln t}{2} \left(1 - \frac{1}{t^2} \right) + 1 - \frac{1}{t} - \frac{\ln t}{t + 1} \right]$$

$$t = \varepsilon / \Delta E_{X v''}^{A v' J'}, \quad u = U / \Delta E_{X v''}^{A v' J'}$$

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$$t = \varepsilon / \Delta E_{X v''}^{A v' J'}, \quad u = U / \Delta E_{X v''}^{A v' J'}$$

- Average over rotational levels (in Boltzmann distribution at T_r):

$$K_{N_2 X v''}^{N_2^+ A v' J'}(T_e, T_r) = \frac{1}{Q_{rot}^{N_2 X v''}(T_r)} \times \sum_{J''} (2J'' + 1) \exp\left[-\frac{F_{N_2 X v''}(J'')}{k T_r}\right] \sum_{J'} K_{N_2 X v''}^{N_2^+ A v' J'}(T_e)$$