Superconductivity in aromatic hydrocarbons

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Abstract

'Aromatic hydrocarbon' implies an organic molecule that satisfies the $(4n+2) \pi$ electron rule and consists of benzene rings. Doping solid aromatic hydrocarbons with metals provides the superconductivity. The first discovery of such superconductivity was made for K-doped picene (K_xpicene, five benzene rings). Its superconducting transition temperatures (T_c 's) were 7 and 18 K. Recently, we found a new superconducting K_xpicene phase with a T_c as high as 14 K, so we now know that K_xpicene possesses multiple superconducting phases. Besides K_xpicene, we discovered new superconductors such as Rb_xpicene and Ca_xpicene. A most serious problem is that the shielding fraction is $\leq 15\%$ for K_xpicene and Rb_xpicene, and it is often ~1% for other superconductors. Such low shielding fractions have made it difficult to determine the crystal structures of superconducting phases. Nevertheless, many research groups have expended a great deal of effort to make high quality hydrocarbon superconductors in the five years since the discovery of hydrocarbon superconductivity. At the present stage, superconductivity is observed in certain metal-doped aromatic hydrocarbons (picene, phenanthrene and dibenzopentacene), but the shielding fraction remains stubbornly low. The highest priority research area is to prepare aromatic superconductors with a high superconducting volume-fraction. Despite these difficulties, aromatic superconductivity is still a core research target and presents interesting and potentially breakthrough challenges, such as the positive pressure dependence of T_c that is clearly observed in some phases of aromatic hydrocarbon superconductors, suggesting behavior not explained by the standard BCS picture of superconductivity. In this article, we describe the present status of this research field, and discuss its future prospects.

'Aromatic hydrocarbon' generally implies a molecule that has $(4n + 2) \pi$ electrons and consists of benzene rings. The crystal and electronic structures of these molecules have been extensively studied by chemists. These molecules are semiconductors with characteristic band gaps, and have recently attracted significant attention from materials scientists and physicists because of their potential application in transistors and solar cells. Also, some hydrocarbons incorporating benzene rings have been investigated for their potential as new conductors. In 1937, London suggested the possibility of 'supraconductivity' in aromatic hydrocarbons. This is the first indication of the relationship of aromatic compound to superconductivity [1]. In the 1950s, Inokuchi and Akamatsu found high conductivity in complexes of perylene with halogens [2]. This pioneer work opened the way towards the realization of an organic conductor. In the 1970s, the main target for an organic conductor changed from aromatic hydrocarbons to charge-transfer complexes such as tetracyanoquinone because of their higher conductivity [3]. Therefore, as suggested above, the development of transistor and solar cell devices has recently focused on using various hydrocarbon molecules, including aromatic hydrocarbons. Thus, the most popular use of organic hydrocarbon molecules is in research on transistor and solar cell devices, and searches on 'hydrocarbon / transistor' find several hundred papers. This implies that hydrocarbons are now some of the most important materials in materials science. In 2010, we reported superconductivity in alkalimetal-doped picene [4], which is classed as an 'aromatic superconductor'; the molecular structures of aromatic hydrocarbons are shown in Fig. 1 together with the lattice structure of picene. This article describes the research on aromatic superconductors reported in the 5 years since this discovery.

1. Superconductivity and features of metal-doped picene solids

When intercalating alkali-metal atoms (K and Rb) into solid picene, superconductivity was observed. The superconducting transition temperatures (T_c 's) found for K-doped picene (K_xpicene) were 7 and 18 K [4]. The 18 K T_c was a new record for organic superconductors, breaking the old record of 14.2 K (in β '-(BEDT-TTF)₂ICl₂ at 8.2 GPa [5]). However, the T_c of 18 K is lower than the highest T_c 's in metal-doped C_{60} 's ($T_c = 33$ K for RbCs₂C₆₀ at ambient pressure (0 GPa) [6] and $T_c = 38$ K for Cs₃C₆₀ at 0.7 GPa [7]), but higher than the highest T_c 's in metal-doped graphites (11.5 K for CaC₆ at 0 GPa [8] and 15.1 K for CaC₆ at 7.5 GPa [9]). Thus, the high T_c found in K_xpicene has attracted much attention because of the expectation of higher T_c 's yet to come. Furthermore, superconductivity in metal-doped aromatic hydrocarbons might lead to a new research field in superconductors because of the diversity of hydrocarbons. Thus, finding superconductivity in K_xpicene was quite significant for basic science, especially in solid state physics and materials science.

However, a most serious problem with metal-doped aromatic hydrocarbon superconductors is that their shielding fraction is still low. The shielding fraction for the 7 K phase of K_xpicene was 15% at 4 K, while it was 1.2% for the 18 K phase of K_xpicene [4]. The low shielding fraction made it impossible to identify the superconducting phases for determination of their crystal structure. During the five years since the discovery of superconductivity in 2010, our group has made an effort to synthesize K_xpicene and other metal-doped picenes that exhibit a high shielding fraction. As reported in Ref. 4, we discovered a superconducting phase of Rb_xpicene, in addition to K_xpicene. Rb_xpicene exhibited a T_c as high as 7 K and a shielding fraction of 10% at 4 K [4]. The first study on Rb_xpicene showed the presence of only a single superconducting phase at 6.9 K. Very recently, we discovered an 11 K phase ($T_c^{\text{onset}} = 11.5 \text{ K}$, $T_c = 11 \text{ K}$) in Rb_xpicene. The M/ H vs. temperature (T) curves are shown in Fig. 2(a), in which M and H refer to the magnetization and applied magnetic field, respectively. The shielding fraction of the 11 K phase was 14% at 3 K. Thus, it was found that Rb_xpicene possesses two distinct superconducting phases, as does K_x picene. The T_c 's and shielding fractions of the superconducting aromatic hydrocarbons are listed in Table 1.

The nominal x values of K_x picene showing $T_c = 7$ and 18 K were 3.3 [4], implying that doping of picene solids with similar amounts of K provides distinct superconducting phases. The nominal x values of Rb_xpicene exhibiting $T_c = 6.9$ and 11 K were 3.1 and 3.8, respectively. It is important to determine the exact x values in each superconducting phase. The most powerful method to determine the exact x value may be Raman scattering. A pronounced peak is observed at 1378 cm⁻¹ in the Raman spectrum of pure solid picene, which can be assigned to superposition of the v_{20} and v_{21} A₁ vibration modes

of the picene molecule [10]. The subscripts of v_{20} and v_{21} refer to the 20th and 21st A₁ vibration modes from the lowest vibration. Kato, Kambe and Kubozono reported that these vibration modes provided strong electron-phonon coupling [11]. When electrons are donated to picene molecules from alkali-metal atoms, the peak clearly shifts to a lower frequency. We measured the Raman scattering of picene, K₁picene, K₂picene, K₃picene, K₄picene and K₅picene solids, and the frequencies of the peak basically took three discrete values, 1378, 1344 and 1313 cm⁻¹, which were consistent with those predicted theoretically for picene, picene²⁻ and picene³⁻ [10]. This implies that only the K_2 picene and K₃picene solids can be prepared when doping picene solids with K atoms at nominal x values of 1 - 5. The same behavior was observed in Rb_x picene [10]. The observed frequencies were 1378, 1345 and 1313 cm⁻¹, which could be assigned to picene, picene²⁻ and picene³⁻, implying the presence of Rb₂picene and Rb₃picene. Systematic M / H - Tplots were prepared for the Kxpicene and Rbxpicene samples, and the correlation between the magnetic behavior and Raman scattering was fully investigated [10], revealing that only samples containing only K₃picene and Rb₃picene showed superconducting behavior. In other words, the superconductivity may be associated with A₃picene phase (A: alkali metal atoms) or the phase relating to A₃picene. Recent reports discuss the exact number of metal atoms required for superconductivity in other aromatic superconductors [12,13].

The crystal structures of superconducting K_x picene solids are a most important research subject. However, conclusive structures have never been found because of the low superconducting shielding fraction of the samples prepared. The crystal structures of 7 and 18 K phases in K_x picene were experimentally suggested by our group [3,9]. The Xray diffraction (XRD) patterns showed that the pure picene solids had a monoclinic lattice (space group: *P*2₁) (see Fig. 1(b)). The lattice constants, *a*, *b*, *c* and β are 8.472(2), 6.170(2), 13.538(7) Å and 90.81(4)°, respectively [4,13], which are consistent with those determined for pure picene single crystals [14]. In the crystals, the herringbone stackinglayer (*ab*-layer) laminates in the direction of the *c*-axis. For the K_xpicene with the 7 K *T_c*, whose nominal x value was 2.9, lattice constant *a* expanded to 8.707(7) Å, while *b* and *c* shrank to 5.912(4) and 12.97(1) Å, respectively; the space group of *P*2₁ still remained in K_xpicene. The lattice constants of pure picene and K_xpicene are listed in Table 2. The lattice constants obtained for K_x picene suggest that K atoms are intercalated into the *ab*plane so as to expand only lattice constant *a*. Theoretical calculation suggests that this is energetically the most stable structure [15].

On the other hand, we were able to selectively synthesize the 18 K phase in K_xpicene with a liquid solvent technique using CH₃NH₂ [10], and measured its XRD patterns. The experimental details are reported in Ref. 10. Raman scattering of the K_x picene (nominal x = 3.1) after the complete removal of CH₃NH₂ showed a red-shift of 67 cm⁻¹ (from the 1378 cm⁻¹ in pure picene) in the peak ascribable to the superposition of the v_{20} and v_{21} A₁ vibration modes, *i.e.* the K₃picene phase. The samples showed superconductivity with $T_c = 18$ K. A (CH₃NH₂)K_xpicene sample (nominal x = 3.0) can be prepared simply by not heating the material to 443 K to completely remove CH_3NH_2 [10], and it shows no superconductivity. The XRD patterns of both samples were analyzed with the space group of $P2_1$. As seen in Table 2, the lattice constants a and c of the $(CH_3NH_2)K_x$ picene sample are larger than those in pure picene, while the b remained almost constant. The c axis expanded markedly from 13.538(7) to 14.476(4) Å, suggesting that the CH₃NH₂ molecules may be intercalated into the space between the herringbone layers (*ab*-layers). The lattice constants a and c in the K_xpicene sample showing $T_c = 18$ K that was prepared using liquid CH₃NH₂ were still larger than those of pure picene, but the c was smaller than that of $(CH_3NH_2)K_x$ picene. For this K_x picene solid (nominal x = 3.1) exhibiting a T_c of 18 K, we concluded that the K atoms might be intercalated into both the *ab*-plane and the space between *ab*-layers. The theoretical calculation suggested the presence of a metastable structure in K_3 picene [16]. The structure is denoted as K₂K₁picene in Refs. 15 and 16, meaning that two K atoms are inserted into the ab-layer while one K atom is intercalated into the space between ablayers, implying different placement of K atoms than in 7 K K_xpicene. If this is the case, the T_c in K₃picene is closely related to the crystal structure, *i.e.*, the crystal structure is a key for determining $T_{\rm c}$. However, at the present stage, although certain crystal structures are suggested for the samples, it remains uncertain whether the suggested structures are associated with the actual superconducting phases, as the sample's shielding fractions are too low.

The strength of electron-phonon coupling in K₃picene was calculated considering intramolecular vibrations of picene³⁻ [11], and the density of states on the Fermi level $(D(\varepsilon_{\rm F})'s)$ was evaluated by comparing the electron-phonon coupling and the experimental $T_{\rm c}$'s of K₃picene with McMillan's formula. The $D(\mathcal{E}_{\rm F})$'s were 1.99 and 2.44 eV⁻¹ molecule⁻¹ spin⁻¹, respectively, for 7 and 18 K phases; where the Coulomb pseudopotential describing electron-electron repulsion, μ^* , was fixed at 0.1 [11]. In other words, if this superconductivity is explainable within the framework of the BCS theorem, the T_c 's observed can be reproduced with the above $D(\varepsilon_F)$ values. The experimental $D(\varepsilon_F)$ was evaluated to be 1.2 states eV⁻¹ molecule⁻¹ spin⁻¹ from the ESR spectrum for the 18 K phase of K₃picene, which is on the same order as the theoretical value (2.44 eV^{-1} molecule⁻¹ spin⁻¹), while the $D(\varepsilon_{\rm F})$ of the 7 K phase could not be evaluated because of the low ESR signal. That is, a larger $D(\mathcal{E}_{\rm F})$ provides a higher $T_{\rm c}$. Thus the approximate estimates of $D(\varepsilon_{\rm F})$ may roughly reproduce the $T_{\rm c}$, although not completely. Casula *et al.* pointed out that the superconductivity could not be explained by considering only the intramolecular phonons, but that intermolecular phonons also play important roles in the emergence of superconductivity [17]. It was suggested that the intermolecular phonons' contribution accounts for $\sim 50\%$ of all the electron-phonon coupling that produces superconductivity. These theoretical calculations were based on the assumption that the superconductivity of K₃picene substantially follows a BCS-type phonon-mediated mechanism [11,17].

Here, we should consider the interesting variation of T_c with pressure. The T_c of the 7 K phase decreased straightforwardly with increasing pressure (*p*), so that $dT_c / dp = -0.3$ K GPa⁻¹ [10]. This can reasonably be interpreted within the framework of the BCS theorem based on the band broadening caused by applied pressure, *i.e.*, the decrease in $D(\varepsilon_F)$. But surprisingly, the T_c of the 18 K phase increased linearly with increasing pressure [10]. The T_c reached 30 K at around 1 GPa. The dT_c / dp was 12.5 K GPa⁻¹, and much larger than in the CaC₆ superconductors that exhibit positive pressure-dependence, typically $dT_c / dp = 0.5$ K GPa⁻¹ [9]. In the case of solid CaC₆, the positive pressure dependence of T_c was explained within the framework of the BCS theorem by the softening of Ca-Ca phonon modes. However, the positive pressure dependence of T_c in

the 18 K phase of K_x picene cannot be simply explained within the BCS model. Aoki and Kariyado pointed out that the topology of the Fermi surface in the K_3 picene and K_2K_1 picene structures described above changes drastically under high pressure [18]. Such a change is called a 'Lifshitz transition'. They suggested that the drastic change in the Fermi surface might explain the behavior in ways that go beyond simple band broadening.

The magnetic behavior of the 7 and 18 K phases of K_x picene, and the 7 and 11 K phases of Rb_xpicene were fully investigated to determine their lower and upper critical fields (H_{c1} and H_{c2}) and other important parameters related to superconductivity such as London's penetration depth ($\lambda_{\rm L}$) and the Ginzburg-Landau coherence length ($\xi_{\rm GL}$). The M vs. H curve of Rb_{3.8}picene (T_c of 11 K) measured at 4.5 K provided an H_{c1} of 80 Oe (see Fig. 2(b)). This diamond curve shows clear superconductivity. The H_{c2} was evaluated as 2000 Oe at 4.5 K. The λ_L and ξ_{GL} were evaluated to be 170 and 41 nm, respectively, with the general equations relating H_{c1} and H_{c2} to λ_L and ξ_{GL} [19]. Table 3 lists the H_{c1} , H_{c2} , λ_L and ξ_{GL} for all metal-doped picene superconductors. Furthermore, it was pointed out that the λ_L that was found was larger than the crystalline size (L) of the superconductor samples (see table 3), which was evaluated with the Debye-Scherrer equation. The fact that $\lambda_L > L$ means that H must penetrate significantly to the superconducting phase and this penetration reduces the shileding fraction. The electrical resistivity (ρ) of superconducting K_xpicene solids was measured in four-terminal measurement mode, and the ρ of the pellet sample of K_{3.5}picene showed zero-resistivity below 8 K, clearly implying superconductivity [20]. The T_c determined from the crossing point of the two lines that map the variation of ρ above and below the T_c was 10.9 K (Fig. 2(a) in Ref. 20). The reproducibility of zero-resistivity was confirmed for the pellet sample of K_{3.1}picene, which showed a T_c of 6.7 K, and zero-resistivity was observed below 5 K [20]. This confirmation of zero-resistivity guarantees that K_xpicene provides superconductivity. The Raman scattering of the samples showed that they contained the K₃picene phase.

For further confirming the superconducting transition in K_xpicene from R - T plot, we measured the temperature dependence of R using the pressurized sample of K_xpicene (nominal x = 3.0); the sample showed the T_c of 6.5 K and the shielding fraction of 0.5% at 2.5 K (Fig. 3(a)), *i.e.*, the 7 K phase of K₃picene. In this experiment, the temperature dependence of *R* was measured at 4.1 GPa. The superconducting transition ($T_c^{\text{mid}} = 5.6$ K) was confirmed from the R - T plot (Figs. 3(b) and (c)) measured using the K_xpicene sample with such a small shielding fraction; the experimental condition such as the current flow at low-temperature was carefully checked in this experiment.

Here we have to discuss why the drop of resistance or zero resistance was observed in the sample with such a low-shielding fraction. One possibility is that the R - Tmeasurement was performed using pelletized sample or pressurized sample. In the pelletized sample or pressurized sample, the connection between superconducting parts must be improved so that the superconducting current flows. Furthermore, tunnel current caused by Josephson effect may flow across the insulating part which may lead to the observation of zero resistance or drop of resistance. Thus, the superconductivity of K_xpicene was also evidenced from the resistance measurement using pelletized or pressurized sample. In addition, the T_c^{mid} shown in Fig. 3(c) follows the pressure dependence of T_c in the 7 K phase of K_xpicene, which was measured previously [10].

The Photoemission spectroscopy (PES) reported by Okazaki *et al.* showed metallic behavior in thin films of K-doped picene [21,22], *i.e.*, the PES peak was observed at the Fermi energy with K-doping of picene thin films and the peak disappeared with excess K-doping. However, the PES reported by other groups showed no metallic behavior [23-25]. Okazaki *et al.* suggested that the observation of metallic behavior originated in the different crystal structures of picene thin-films, which change depending on the substrates used.

We also intercalated the alkali-earth metal atoms Ca and Ba into solid picene. With Ca atoms intercalated, a superconducting transition was confirmed at a nominal 1.5 parts Ca, *i.e.*, Ca_{1.5}picene [13], which indicated a three-electron donation to picene molecules in the superconducting sample, based on the oxidation state of +2 in Ca. The T_c was 7 K and the shielding fraction was 1.25% at 2 K. In the case of Ba, we observed a superconducting transition with $T_c = 6$ K, but the shielding fraction was quite small (1% at 2.0 K). Thus, although the alkali-earth metal-doped picene showed superconductivity,

the shielding fraction was too small for us to determine the structure and physical properties.

We also tried K-doping picene using the liquid NH₃ technique, analogous to the successful synthesis of K_x picene using CH₃NH₂ [10]. The Li_xpicene prepared using liquid NH₃ showed superconductivity, with a T_c of 11 K. Thus, NH₃ was also found to be useful in the synthesis of K_x picene solids. Recently, we observed superconductivity with a T_c as high as 14 K in K_x picene (nominal x = 3.0), and a shielding fraction of 3.9% at 2 K (see Fig. 4(a)). This phase's T_c is different from the 7 and 18 K phases of K_x picene, and probably represents a new superconducting phase. Its structure and physical properties have not yet been determined, and the study of its characteristics is now in progress. Pressing the sample up to 0.2 GPa led to a higher shielding fraction than the above value (Fig. 4(b)), *i.e.*, a shielding fraction of 16% at 5 K was realized after pressing the sample, suggesting an increase in the size of superconducting grains due to a stronger connection. Furthermore, we speculated the stabilization of superconducting phase under pressure through the suppression of competitive electronic phase for superconductivity such as magnetic order which has often been observed in the K_xpicene sample. Anyway, the pressing the sample improves the superconductivity.

We carefully checked the presence of elements (impurities) other than K and C elements in the K_x picene sample using energy dispersive X-ray (EDX) spectroscopy. As seen from Fig. 4(c), only peaks due to C and K were observed together with the small peaks due to O owing to oxidation of sample, Si owing to the glass (when cutting sample tube, Si was introduced to the sample), and Cu originating from Cu tape used for holding the sample. Notably, H cannot be detected by EDX. Thus, the EDX spectrum substantially suggests no elements other than K and C in K_x picene.

2. Metal-doping of solid hydrocarbons other than picene, and their superconductivity

The first successful production of superconductivity by metal-doping of aromatic

hydrocarbons other than picene was achieved in phenanthrene, which has a W-shaped structure consisting of three fused benzene rings [12]. This work was done by Chen's group at the Chinese University of Science and Technology. The K_xphenanthrene sample (nominal x = 3) provided a T_c of 4.95 K with a shielding fraction of 5.3%, while Rb_xphenanthrene (nominal x = 3) showed a T_c of 4.75 K with a shielding fraction of 6.7%. Pressing the K₃phenanthrene sample increased the shielding fraction to 15.5%. Similar behavior was observed in superconducting K_x picene samples [4]. The Raman peak observed at 1441cm⁻¹ for pure phenanthrene downshifted to 1424 and 1419 cm⁻¹, respectively, for K₃phenanthrene and Rb₃phenanthrene, suggesting three-electron donation to the phenanthrene molecule, an electron transfer equal to that in superconducting metal-doped solid picene. Subsequently, Sr_{1.5}phenanthrene and Ba_{1.5}phenanthrene showed superconductivity, with $T_c = 5.6$ and 5.4 K, respectively [26]. The shielding fraction of $Ba_{1.5}$ phenanthrene was reported to be 65.4%. However, it was found that the samples of Ba_{1.5}phenanthrene contained La atoms, and two papers on specific heats were retracted. It should be noted that La metal shows superconductivity with $T_c = 6$ K (face centered cubic (fcc)) and $T_c = 5$ K (double hexagonal close packing (dhcp)) and also positive pressure-dependence of T_c (d T_c / dp = 1.2 K GPa⁻¹ for fcc and 1.1 K GPa⁻¹ for dhcp) [27]. At the present stage, we cannot judge whether the superconductivity of the Ba_{1.5}phenanthrene sample exhibiting the high shielding fraction (~65%) originated from Ba_{1.5}phenanthrene. Further study of Ba_{1.5}phenanthrene may be required. We observed clear superconductivity in the $Sr_{1.5}$ phenanthrene sample provided by the above group, which showed $T_c = 5.3$ K and a shielding fraction of 4% at 2 K (Fig. 5(a)). X-ray photoelectron spectroscopy (XPS) of the Sr₁ sphenanthrene sample showed only C and Sr peaks (Fig. 5(b)), indicating that the superconductivity of the $Sr_{1.5}$ phenanthrene sample can be assigned to Sr-doped phenanthrene. In other words, no elements other than Sr and C were included in the sample. The pressure dependence of $T_{\rm c}$ in K_3 phenanthrene showed a positive-pressure effect, with the T_c increasing linearly with increasing pressure [12]. The 1/ T_c (0 GPa) × dT_c / dp of K₃phenanthrene was 0.26 GPa⁻¹ which means $dT_c / dp \sim 1.2$ K GPa⁻¹. Our group also observed superconductivity in K_xphenanthrene with $T_c = 4.5$ K and a shielding fraction of 1%.

Superconductivity was confirmed in K-doped dibenzopentacene (K_{3.45}DBP, DBP: dibenzopentacene) by Xue *et al.*, which showed a T_c as high as 33 K [28]. Multiple superconducting phases were observed in K_xDBP; $T_c = 7.4$ K for K₃DBP and $T_c = 28.2$ K for K_{3.17}DBP. Thus, the different x's of K_xDBP provide different T_c 's. At present, the T_c of K_{3.45}DBP is the highest reported for metal-doped aromatic hydrocarbons. The shielding fraction of K_{3.45}DBP was 3.2% at 25 K, still fairly low. Our group also observed the superconducting transitions at $T_c = 30$ and 7 K in K_xDBP, with a shielding fraction below 1%.

We prepared K-doped [6]phenacene using the liquid NH₃ technique, and superconductivity was observed with $T_c = 7$ K and a shielding fraction of 1%. Since the shielding fraction was so low, we could not characterize the structure and physical properties. Nevertheless, this is significant because it showed that a superconducting phase can be produced using phenacenes other than phenanthrene and picene. Subsequently, K_x [7]phenacene was prepared, which showed superconductivity with a T_c as high as 11 K. This sample was prepared using the normal annealing method. The shielding fraction was 0.1% at 2 K. Thus, we have prepared new superconductors with various phenacene molecules. We also observed superconductivity in metal-doped coronene [13] and corannulene, but the shielding fraction was still 1%. Such a low shielding fraction suggests the possibility of surface superconductivity, with only the surfaces of crystal grains doped with metal atoms. Furthermore, the degradation of molecules by the reaction of metal atoms with the hydrocarbons is also suggested. The samples exhibiting low shielding fractions should be more precisely characterized to clarify the origin of their superconductivity. The trial is now in progress independently by some groups.

Finally, we plotted the T_c 's of the hydrocarbon superconductors (both aromatic and non-aromatic) as a function of the number of benzene rings. As seen from Fig. 5(c), for superconductors with a higher shielding fraction, the T_c increases with an increasing number of benzene rings. In contrast, when the T_c 's of metal-doped hydrocarbons with low shielding fractions (~1%) are plotted, the T_c does not increase straightforwardly. Metal-doped picene and dibenzopentacene possess multiple superconducting phases, so

various T_c 's are observed [4,28], but the T_c 's of metal-doped phenanthrenes range from 4.5 to 6.0 K [12,26]. The correlation between T_c and the number of benzene rings is confined to the maximum T_c 's in each sample that has a high shielding fraction. To firmly establish the correlation, we have to produce more hydrocarbon superconductors with high shielding fractions.

3. What was clarified for aromatic hydrocarbon superconductors?

This article has shown the problems to be solved concerning aromatic hydrocarbons. In the ~5 years since the discovery of superconductivity in metal-doped picene, many theoretical studies have appeared on aromatic hydrocarbon superconductors, but just a few papers on the preparation of aromatic hydrocarbon superconductors. The reason is probably the difficulty of preparing such superconductors. We fully reviewed superconductivity in carbon based superconductors, including aromatic hydrocarbons, in 2011 [13], and recently a review on aromatic hydrocarbon superconductors was published by Artioli and Malavasi [29]. In that review, Artioli and Malavasi pointed out that 'in order to make the hydrocarbon superconductors a solid future prospect much effort – starting from the experimental side – is still required'.

At the present stage, some groups have succeeded in preparing aromatic hydrocarbon superconductors with a shielding fraction of more than 10%. Very recently, Artioli *et al.* at University of Pavia (Italy) and University of Oslo (Norway) discovered the superconductivity in Sm doped [n]phenacenes (n = 3, 4 and 5); [n]phenacenes (n = 3, 4 and 5) refer to phenanthrene, chrysene and picene [30]. The T_c 's of Sm doped phenanthrene, chrysene and picene were 4.4, 5.4 and 4 K, respectively. Thus, new types of superconducting aromatic materials steadily increase. This allows us to conclude that superconducting materials can be prepared by metal intercalation into solid aromatic hydrocarbons, or if strictly saying, by reaction of metal with solid aromatic hydrocarbons. This is clearly established. However, the study of the structure and composition of these superconducting materials remains incomplete, as suggested in the above review article. Currently, the superconducting shielding fraction of 0.1 - 1.0% can be observed at the probability of 10% out of the K_xpicene and Rb_xpicene samples prepared, while remaining

90% samples mainly exhibited the Curie-type behavior, suggestive of localized magnetic state. The samples with more than 10% shielding fraction are sometimes found, but the clear protocol to synthesize the high-quality superconductor sample has not yet been established. Recently, some other groups, new to this field, have succeeded in preparing superconducting metal-doped picene solids, and some of their papers will be published soon. This may be the trigger for a more extensive pursuit of the origin of aromatic superconductivity.

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Table 1. T_c 's and shielding fractions (SF's) of metal-doped aromatic hydrocarbons. The T_c determination method is described in each reference. The T_c 's of the 14 K phase of K_xpicene and 11 K phase of Rb_xpicene were determined based on the crossing point of two lines (see Figs. 4(a) and 2(a)). The temperatures for SF evaluation are described in text or references. Nominal x values are described in text or references.

	<i>T</i> _c (K)	SF (%)
K _x phenanthrene ^{a)}	4.95	5.3
Rb _x phenanthrene ^{a)}	4.75	6.7
K _x picene (7 K phase) ^{b)}	6.9	15
K _x picene (14 K phase)	14	3.9
		16 (after pressing)
K _x picene (18 K phase) ^{b)}	18	1.2
Rb _x picene (7 K phase) ^{b)}	6.9	10
Rb _x picene (11 K phase)	11	14
K _x dibenzopentacene ^{c)}	7.4	3.6
· · ·	28.2	5.5
	33	3.2

a) Taken from Ref. 12.

b) Taken from Ref. 4.

c) Taken from Ref. 28.

	<i>a</i> (Å)	<i>b</i> (Å)	<i>c</i> (Å)	β (°)	$V(\text{\AA}^3)$
phenanthrene ^{a)}	8.453	6.175	9.477	98.28	490
K ₃ phenanthrene ^{a)}	8.430	6.134	9.417	98.18	482
Rb ₃ phenanthrene ^{a)}	8.450	6.139	9.442	97.99	485
picene ^{b)}	8.472(2)	6.170(2)	13.538(7)	90.81(4)	708
K _x picene (7 K phase) ^{b)} 8.707(7)	5.912(4)	12.97(1)	92.77(5)	667
(CH ₃ NH ₂) _z K _x picene ⁶	²⁾ 8.927(5)	6.151(1)	14.476(4)	94.16(3)	793
K _x picene (18 K phas	e) ^{c)} 8.571(5)	6.270(2)	14.001(3)	91.68(3)	752

Table 2. Lattice constants of metal-doped aromatic hydrocarbons. Nominal x values are described in text or references.

a) Taken from Ref. 12.

b) Taken from Ref. 4.

c) Taken from Ref. 10.

	H_{c1} (Oe)	$H_{c2}(Oe)$	$\lambda_L(nm)$	$\xi_{GL}(nm)$	L (nm)
K _x picene (7 K phase) ⁶	a) 300	>10000	95	<18	14
K _x picene (18 K phase	(^{a)} 400	>10000	77	<18	14
Rb _x picene (7 K phase) ^{a)} 150	800	130	57	29
Rb _x picene (11 K phas	e) 80	2000	170	41	66
K _x dibenzopentacene ^{b)}	200	-	-	-	-

Table 3. Parameters characterizing superconductivity in metal-doped aromatic hydrocarbons.

a) Taken from Ref. 4.

b) Taken from Ref. 28.

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Figure captions

Fig. 1. (a) Molecular structures of some aromatic hydrocarbons and (b) lattice structure of picene.

Fig. 2. (a) M / H vs. T plots (0 GPa) in ZFC and FC modes and (b) M vs. H plot (0 GPa) at 4.5 K for Rb_xpicene (nominal x = 3.8).

Fig. 3. (a) M / H vs. T plots (0 GPa) in ZFC and FC modes and (b) R vs. T plot (4.1 GPa) of 7 K phase of K_xpicene (nominal x = 3.0). (c) Expanded R vs. T plot of (b).

Fig. 4. (a) M / H vs. T plots (0 GPa) in ZFC and FC modes and (b) M / H vs. T plots in ZFC and FC modes measured at 0.2 GPa for K_xpicene (x = 3.0). (c) EDX spectrum for K_xpicene (x = 3.0).

Fig. 5. (a) M / H vs. T plots (0 GPa) in ZFC and FC modes and (b) XPS spectrum for Sr_xphenanthrene (nominal x = 1.5). (c) Plots of T_c against number of benzene rings in hydrocarbon supoerconductors. In (c), the dots (blue and red) refer to the high- T_c and low- T_c phase of picene, coronene and dibenzopentacene which have multiple superconducting phases. The blue dot of phenanthrene corresponds to the superconducting phase. The dots (green) refer to the superconducting phases of [6]phenacene and [7]phenacene.



phenanthrene

picene

dibenzopentacene

(b)





Figure 3





Figure 5

