

Full Paper

Synthesis and Antifungal Bioactivities of 3-Alkylquinazolin-4-one Derivatives

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Abstract: A simple, efficient, and general method has been developed for the synthesis of various 3-alkylquinazolin-4-one derivatives from quinazolin-4-one treated with alkyl bromides under phase transfer catalysis condition. The structures of the compounds were characterized by elemental analysis, IR, ¹H-NMR and ¹³C-NMR spectra. Title compound 6-bromo-3-propylquinazolin-4-one (**3h**) was found to possess good antifungal activity.

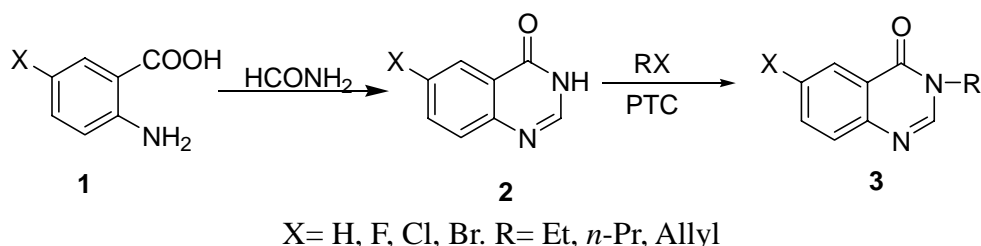
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Introduction

A variety of the reports regarding synthetic studies of quinazolinone derivatives have been presented due to the chemical and biological interests to the quinazoline [1, 2]. Quinazoline compounds are widely used in agrochemicals as plant virucides [3], antifungal agents [4] and herbicides [5]. According to recent data, quinazoline nucleus has attracted the attention of medicinal chemists due to its well known anticancer activity, and many substituted quinazoline derivatives have recently earned great interest in chemotherapy as antitumor drugs [6, 7]. In order to find potential new plant fungicides and anticancer agents, we had designed and synthesized a series of

N-aryl-4-aminoquinazoline compounds, among which some compounds were found to possess moderate bioactivity [8]. In this report we designed and synthesized a series of 3-alkylquinazolin-4-one derivatives and investigated their bioactivities. The synthetic route was shown in Scheme 1. The structures of **3** were firmly established by well defined IR, ¹H-NMR, ¹³C-NMR and elemental analysis. Preliminary bioassay tests showed that some compounds displayed antifungal activity against three fungi at 50 µg/mL *in vitro*, but with a degree of variation. It was found that title compound **3h** displayed strong *in vitro* antifungal activity on hyphal growth of *F. oxysporum*, *Valsa mali* and *Gibberella zeae*.

Scheme 1 Synthetic route to the title compounds.



Results and Discussion

In order to optimize the reaction conditions, the synthesis of **3e** was carried out under various conditions. The effects of KOH concentration, reaction time, reaction temperature, presence or absence of a phase transfer catalyst (PTC) on the reaction were investigated and the results are shown in Table 1. When no tetrabutylammonium bromide was used as PTC, the reaction was relatively slow and the product was obtained in 8.5 % refluxed at 88-90 °C for 1 h (Table 1, entry 5), while the use of the PTC obviously accelerated the reaction (Table 1, entries 1-4). While the yield of **3e** increased to 77.1 % also in 1 h when the reaction was catalyzed by 0.08 equiv Bu₄NBr (Table 2, entries 3 and 5). When the amount of Bu₄NBr increased from 3 % to 5 %, 8 % and 10 %, **3e** could be obtained in 38.8 % and 62.2 %, 77.1 % and 77.4 %, respectively (Table 1, entries 1-4). The effect of KOH concentration was also studied. When KOH concentration increased from 10 %, to 20 %, 30 % and 35 % with Bu₄NBr as catalyst, **3e** could be obtained in 48.5 %, 69.1 %, 77.1 % and 83.0 %, respectively (entries 6-7, 3, 8). Interestingly, lower yield of **3e** was obtained when KOH concentration increased from 35 % to 40 % (entries 8-9), which indicated that 35 % was the best KOH concentration. Using 8 mol% tetrabutylammonium bromide as catalyst, the yield of **3e** increased from 55.9 % to 83.0 % when the reaction time was prolonged from 0.5 to 1 h refluxed at 88-90 °C (entries 8, 14 and 15). When the reaction time was prolonged further to 2 h, tiny improvement (83.8 %, entry 13) was obtained, as compared to that of 1 h (83.0 %, entry 8). As for the reaction temperature, it could be seen that the yield was relatively low when the reaction was carried out at 20-65°C (Table 1, entries 10-12) than that at 88-90 °C (entry 8).

Table 1 Different reaction conditions for synthesis of **3e**.

Entry ^a	Reaction time / h	KOH concentration / % (w/w)	The amount of Bu ₄ NBr / mol %	Reaction temperature / °C	Yield ^b / %
1	1	30	3	88-90	38.8
2	1	30	5	88-90	62.2
3	1	30	8	88-90	77.1
4	1	30	10	88-90	77.4
5	1	30	0	88-90	8.5
6	1	10	8	88-90	48.5
7	1	20	8	88-90	69.1
8	1	35	8	88-90	83.0
9	1	40	8	88-90	74.0
10	1	35	8	20-25	4.3
11	1	35	8	40-45	13.3
12	1	35	8	60-65	65.0
13	2	35	8	88-90	83.8
14	0.75	35	8	88-90	74.5
15	0.5	35	8	88-90	55.9

^a Reaction conditions: A mixture of quinazolin-4-one (1 equiv) (**2e**), 1-bromopropane (1 equiv), Bu₄NBr (0.03-0.1 equiv), 10-40 % potassium hydroxide (10 mL) in toluene (10 mL) was stirred at 20-90°C for 0.5-2 h.

^b Yields of isolated products.

The effect of different organic phase was also investigated. When CHCl₃, benzene, ethyl acetate, and toluene was used, the yields of **3e** were 25.5 %, 45.2 %, 54.3 % and 83.0 %, respectively (Table 2). By using toluene as the organic phase, the **3e** synthesis was found to proceed smoothly in PTC condition at 88-90°C.

Table 2 Effect of different organic solvents on the synthesis of **3e**

Entry ^a	Reaction time / h	organic phase	The amount of solvent / mL	Yield ^b / %
1	1	CHCl ₃	10	25.5
2	1	C ₆ H ₆	10	45.2
3	1	CH ₃ COOEt	10	54.3
4	1	Toluene	10	83.0

^a Reaction conditions: A mixture of quinazolin-4-one (1 mmol) (**2e**), 1-bromopropane (1.0 mmol), Bu₄NBr (0.08 mmol), 35 % potassium hydroxide (84 mmol, 10mL) in toluene (10 mL) was stirred at 88-90°C for 1 h.

^b Yields of isolated products.

Using the optimized condition, the best result was obtained when intermediate **2** was treated with 1 equiv of alkyl halides and 84 equiv KOH (35%, 10 mL) under PTC conditions (Bu₄NBr, 0.08 equiv) with toluene as solvent (10 mL) refluxed at 88-90 °C for 1 h. Under these reaction conditions, the amination reaction proceeded smoothly, and the results are summarized in **Table 3**. Compounds **3e** and **3i** were prepared previously in yields of 41% and 33%, respectively [8], by the reaction of metallic derivatives of substituted or unsubstituted quinazolin-4-one with alkyl halides or dialkyl sulfate in ethanol solvent for 6-24 hrs.

Table 3. Yields of the title compounds **3**.^a

Compd	3a	3b	3c	3d	3e	3f
X-	H	F	Cl	Br	H	F
R-	Et	Et	Et	Et	<i>n</i> -Pr	<i>n</i> -Pr
^b Yields /%	79.5	74.5	74.5	77.0	83.0	92.2
Compd	3g	3h	3i	3j	3k	3l
X-	Cl	Br	H	F	Cl	Br
R-	<i>n</i> -Pr	<i>n</i> -Pr	allyl	allyl	allyl	allyl
^b Yields /%	83.1	87.0	70.0	72.7	82.5	70.9

^a All reactions were carried out in toluene (10 mL) at 88-90°C for 1 h under PTC conditions with KOH (84 equiv) used as base.

^b Isolated yields.

The structures of title compounds **3** were established on the basis of their spectroscopic data. They showed IR absorption bands at 3049-3099 (Ar-H), 1664-1678 (C=O), and 1451-1613 cm⁻¹ (skeleton vibration of aromatic ring). In the ¹H NMR spectra of quinazolinone derivatives, the 2-H signal appeared as a singlet in the 7.99-8.08 ppm range, while the 7-H peaks of 6-fluoroquinazolin-4-one derivatives were observed at about 7.72-7.73 ppm as a quartet. The 8-H of 6-bromoquinazolin-4-one derivatives appear as a doublet at 7.57-7.60 ppm.

Antifungal activity bioassay

The antifungal bioassay results are given in Table 4. It could be seen that these newly synthesized 3-alkylquinazolin-4-one derivatives exhibit weak to good antifungal activities. At 50 µg/mL, compounds **3c**, **3h** and **3k** inhibited growth of *Gibberella zeae* at 55.0 %, 50.3 % and 49.1 % respectively. Compounds **3h**, **3c**, **3k**, **3d** and **3l** exhibited good activities on *Fusarium oxysporum* at 47.2 %, 37.5 %, 36.4 %, 33.3 %, 32.4 %, respectively, which is slightly lower than that of hymexazol (55.4 %). Compounds **3l**, **3k**, **3c** showed activities against *Valsa mali* at 30.4 %, 28.3 %, 24.4 %, respectively, which is little higher than *Hymexazol* (22.5 %). Compounds **3h** showed activities against *Valsa mali* at 40.9 %, which is obviously higher than *Hymexazol* (22.5 %).

Table 4. Inhibition effect of 3-alkylquinazolin-4-one derivatives on phytopathogenic fungi.

Compd (50 µg/mL)	<i>Fusarium oxysporum</i>	<i>Gibberella zeae</i>	<i>Valsa mali</i>
3a	17.9	17.6	8.7
3b	14.2	11.8	6.5
3c	37.5	55.0	24.4
3d	33.3	35.4	21.7
3e	21.2	17.3	16.9
3f	4.9	17.0	14.5
3g	20.4	5.5	15.9
3h	47.2	50.3	40.9
3i	15.3	19.9	0.9
3j	13.9	26.7	13.4
3k	36.4	49.1	28.3
3l	32.4	37.0	30.4
Hymexazol	55.4	50.4	22.5

Conclusions

In summary, the present new method of the formation of 3-alkylquinazolin-4-one derivatives under phase transfer catalyst condition offers several advantages: fast reaction rates, less by-products, and high yields. Where direct comparisons were possible our method was found to be superior to reported methods. It was also found that title compound **3h** displayed good antifungal activity.

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Experimental

General

Unless otherwise indicated, all common reagents and solvents were used as obtained from commercial suppliers without further purification. Melting points were determined on a XT-4 binocular microscope (Beijing Tech Instrument Co., China) and were not corrected. The IR spectra were recorded on a Bruker VECTOR 22 spectrometer in KBr disks. ¹H-NMR and ¹³C-NMR spectra (solvent CDCl₃) were recorded on a JEOL ECX 500 MHz spectrometer at room temperature using TMS as internal standard. Elemental analysis was performed by an Elementar Vario-III CHN analyzer. The following compounds were prepared as described in the literature: *quinazolin-4-one* (**2a**): white solid, yield 95.4%, m.p. 214~215°C (lit. [9], m.p. 215.5~216.5°C); *6-fluoroquinazolin-4-one* (**2b**):

white solid, yield 70.8%, m.p. 258-260 °C (lit. [10], m.p. 255-257 °C); 6-chloroquinazolin-4-one (**2c**): white solid, yield 78.5%, m.p. 257-260 °C (lit. [11], m.p. 259-259.5 °C) and 6-bromoquinazolin-4-one (**2d**): white solid, yield 69.5 %, m.p. 264-267°C (lit. [12], m.p. 267-269 °C).

Preparation of 3-Alkylquinazolin-4-one Derivatives **3a-3l**

A mixture of quinazolin-4-one or 6-haloquinazolin-4-one (1.00 mmol), alkyl bromide (1.00 mmol) and tetrabutylammonium bromide (26 mg, 0.08 mmol) was dissolved in toluene (10 mL) and 35% aqueous potassium hydroxide (10 mL). The solution was stirred and refluxed for 1 h and then the organic layer was separated and washed with water. After evaporation of solvent, the crude product was purified by preparative TLC with petroleum ether/ethyl acetate (V/V=1:2) as developing solvent to give title compounds **3**.

3-Ethylquinazolin-4-one (3a). White crystals, yield 79.5%; m.p. 76.5~78.5 °C; (lit. [8], m.p. 99-101 °C, yield not reported). IR: 3049.2, 2960.5, 1674.2, 1612.5, 1562.3, 1471.7 cm⁻¹; ¹H-NMR: δ 8.32 (d, 1H, *J*=8.0Hz, quinazolinone H-5), 8.08 (s, 1H, quinazolinone H-2), 7.76-7.71 (m, 2H, quinazolinone H-6,8), 7.51 (t, 1H, *J*=6.9Hz, quinazolinone H-7), 4.08 (q, 2H, *J*=7.5Hz, CH₂), 1.43 (t, 3H, *J*=7.5Hz, CH₃); ¹³C-NMR: δ 160.8 (quinazolinone C-4), 148.1 (quinazolinone C-2), 146.3 (quinazolinone C-9), 134.1 (quinazolinone C-7), 127.4 (quinazolinone C-5), 127.2 (quinazolinone C-6), 126.6 (quinazolinone C-10), 122.2 (quinazolinone C-8), 42.1 (CH₂), 14.9 (CH₃); *Anal.* Calc. for C₁₀H₁₀N₂O (174.2): C, 68.95; H, 5.79; N, 16.08. Found: C, 68.80; H, 5.79; N, 16.18.

3-Ethyl-6-fluoroquinazolin-4-one (3b). White crystals, yield 74.5%; m.p. 94.5~96.5 °C; IR: 3058.6, 2974.2, 1674.2, 1626.0, 1606.7, 1487.1 cm⁻¹; ¹H-NMR: δ 8.05 (s, 1H, quinazolinone H-2), 7.94 (dd, 1H, *J*=8.6Hz, 2.9Hz, quinazolinone H-5), 7.73 (q, 1H, *J*_{H-F}=4.5Hz, *J*_{H-H}=4.0Hz, quinazolinone H-7), 7.50-7.48 (m, 1H, quinazolinone H-8), 4.08 (q, 2H, *J*=7.5Hz, CH₂), 1.44 (t, 3H, *J*=7.5Hz, CH₃); ¹³C-NMR: δ 162.1 (quinazolinone C-4), 160.1 (quinazolinone C-6), 145.6 (quinazolinone C-2), 144.7 (quinazolinone C-9), 129.8 (quinazolinone C-10), 123.4 (quinazolinone C-8), 122.8 (quinazolinone C-7), 111.7 (quinazolinone C-5), 42.3 (CH₂), 14.8 (CH₃); *Anal.* Calc. for C₁₀H₉FN₂O (192.2): C, 62.49; H, 4.72; N, 14.58. Found: C, 62.34; H, 4.62; N, 14.45.

6-Chloro-3-ethylquinazolin-4-one (3c). White crystals, yield 74.5%; m.p. 104~106°C; IR: 3099.2, 2980.0, 1666.5, 1626.5, 1604.8, 1469.8 cm⁻¹; ¹H-NMR: δ 8.28 (d, 1H, *J*=2.5Hz, quinazolinone H-5), 8.06 (s, 1H, quinazolinone H-2), 7.70-7.65 (m, 2H, quinazolinone H-7,8), 4.07 (q, 2H, *J*=7.5Hz, CH₂), 1.43 (t, 3H, *J*=7.5Hz, CH₃); ¹³C-NMR: δ 159.9 (quinazolinone C-4), 146.6 (quinazolinone C-2), 146.4 (quinazolinone C-9), 134.6 (quinazolinone C-7), 133.1 (quinazolinone C-6), 129.0 (quinazolinone C-5), 126.0 (quinazolinone C-10), 123.2 (quinazolinone C-8), 42.3 (CH₂), 14.8 (CH₃); *Anal.* Calc. for C₁₀H₉ClN₂O (208.6): C, 57.57; H, 4.35; N, 13.43. Found: C, 57.67; H, 4.25; N, 13.43.

6-Bromo-3-ethylquinazolin-4-one (3d). White crystals, yield 77.0%; m.p. 76.5~78.5°C; (lit. [13], neither yield nor any properties given). IR: 3085.3, 2978.1, 1664.6, 1600.9, 1587.9, 1465.9 cm⁻¹; ¹H-NMR: δ 8.44 (d, 1H, *J*=2.0Hz, quinazolinone H-5), 8.07 (s, 1H, quinazolinone H-2), 7.81 (dd, 1H, *J*=9.2Hz, 1.7Hz, quinazolinone H-7), 7.57 (d, 1H, *J*=8.6Hz, quinazolinone H-8), 4.07 (q, 2H, *J*=7.5Hz,

CH₂), 1.43 (t, 3H, $J=7.5$ Hz, CH₃); ¹³C-NMR: δ 159.7 (quinazolinone C-4), 146.9 (quinazolinone C-2), 146.6 (quinazolinone C-9), 137.3 (quinazolinone C-7), 133.5 (quinazolinone C-5), 129.2 (quinazolinone C-10), 123.5 (quinazolinone C-8), 120.8 (quinazolinone C-6), 42.3 (CH₂), 14.8 (CH₃); *Anal. Calc.* for C₁₀H₉BrN₂O (253.1): C, 47.46; H, 3.58; N, 11.07. Found: C, 47.33; H, 3.48; N, 11.18.

3-Propylquinazolin-4-one (3e). White crystals, yield 83.0%; m.p. 82~83°C; (lit. [8], m.p. 95-96 °C). IR: 3075.3, 2964.6, 1674.2, 1610.6, 1562.3, 1471.7 cm⁻¹; ¹H-NMR: δ 8.33 (d, 1H, $J=8.0$ Hz, quinazolinone H-5), 8.05 (s, 1H, quinazolinone H-2), 7.78-7.71 (m, 2H, quinazolinone H-6,8), 7.51 (t, 1H, $J=8.0$ Hz, quinazolinone H-7), 3.98 (t, 2H, $J=7.5$ Hz, NCH₂), 1.87-1.82 (m, 2H, CH₂), 1.01 (t, 3H, $J=7.5$ Hz, CH₃); ¹³C-NMR: δ 161.1 (quinazolinone C-4), 148.0 (quinazolinone C-2), 146.6 (quinazolinone C-9), 134.1 (quinazolinone C-7), 127.4 (quinazolinone C-5), 127.2 (quinazolinone C-6), 126.7 (quinazolinone C-10), 122.1 (quinazolinone C-8), 48.6 (NCH₂), 22.7 (CH₂), 11.1 (CH₃); *Anal. Calc.* for C₁₁H₁₂N₂O (188.2): C, 70.19; H, 6.43; N, 14.88. Found: C, 70.14; H, 6.28; N, 15.03.

6-Fluoro-3-propylquinazolin-4-one (3f). White crystals, yield 92.2%; m.p. 96~98°C; IR: 3082.6, 2960.7, 1676.1, 1610.6, 1608.1, 1481.3 cm⁻¹; ¹H-NMR: δ 8.01 (s, 1H, quinazolinone H-2), 7.95 (dd, 1H, $J=8.6$ Hz, 2.9Hz, quinazolinone H-5), 7.73 (q, 1H, $J_{H-F}=4.5$ Hz, $J_{H-H}=4.0$ Hz, quinazolinone H-7), 7.50-7.46 (m, 1H, quinazolinone H-8), 3.98 (t, 2H, $J=7.5$ Hz, NCH₂), 1.86-1.81 (m, 2H, CH₂), 1.01 (t, 3H, $J=7.5$ Hz, CH₃); ¹³C-NMR: δ 162.2 (quinazolinone C-4), 160.4 (quinazolinone C-6), 146.0 (quinazolinone C-2), 144.8 (quinazolinone C-9), 129.9 (quinazolinone C-10), 123.6 (quinazolinone C-8), 122.7 (quinazolinone C-7), 111.8 (quinazolinone C-5), 48.8 (NCH₂), 22.7 (CH₂), 11.2 (CH₃); *Anal. Calc.* for C₁₁H₁₁FN₂O (206.2): C, 64.07; H, 5.38; N, 13.58. Found: C, 64.18; H, 5.40; N, 13.44.

6-Chloro-3-propylquinazolin-4-one (3g) [CAS number 501917-08-4, but no references given]. White crystals, yield 83.1%; m.p. 130.0~131.5°C; IR: 3085.2, 2956.9, 1674.2, 1612.5, 1556.6, 1469.7 cm⁻¹; ¹H-NMR: δ 8.26 (d, 1H, $J=2.3$ Hz, quinazolinone H-5), 8.03 (s, 1H, quinazolinone H-2), 7.68-7.65 (m, 2H, quinazolinone H-7,8), 3.97 (t, 2H, $J=7.5$ Hz, NCH₂), 1.86-1.82 (m, 2H, CH₂), 1.01 (t, 3H, $J=7.5$ Hz, CH₃); ¹³C-NMR: δ 160.1 (quinazolinone C-4), 146.9 (quinazolinone C-2), 146.7 (quinazolinone C-9), 134.6 (quinazolinone C-7), 133.1 (quinazolinone C-6), 129.1 (quinazolinone C-5), 126.1 (quinazolinone C-10), 123.3 (quinazolinone C-8), 48.8 (NCH₂), 22.7 (CH₂), 11.2 (CH₃); *Anal. Calc.* for C₁₁H₁₁ClN₂O (222.7): C, 59.33; H, 4.98; N, 12.58. Found: C, 59.43; H, 4.94; N, 12.57.

6-Bromo-3-propylquinazolin-4-one (3h). White crystals, yield 87.0%; m.p. 132.5~134.5°C; IR: 3075.2, 2955.0, 1666.5, 1610.6, 1597.1, 1465.9 cm⁻¹; ¹H-NMR: δ 8.44 (d, 1H, $J=2.0$ Hz, quinazolinone H-5), 8.04 (s, 1H, quinazolinone H-2), 7.82 (dd, 1H, $J=9.2$ Hz, 1.7Hz, quinazolinone H-7), 7.57 (d, 1H, $J=8.6$ Hz, quinazolinone H-8), 3.97 (t, 2H, $J=7.5$ Hz, NCH₂), 1.86-1.81 (m, 2H, CH₂), 1.01 (t, 3H, $J=7.5$ Hz, CH₃); ¹³C-NMR: δ 159.9 (quinazolinone C-4), 146.9 (quinazolinone C-2), 146.7 (quinazolinone C-9), 137.3 (quinazolinone C-7), 129.3 (quinazolinone C-5), 129.2 (quinazolinone C-10), 123.5 (quinazolinone C-8), 120.8 (quinazolinone C-6), 48.7 (NCH₂), 22.6 (CH₂), 11.1 (CH₃); *Anal. Calc.* for C₁₁H₁₁BrN₂O (267.1): C, 49.46; H, 4.15; N, 10.49. Found: C, 49.73; H, 4.24; N, 10.45.

3-Allylquinazolin-4-one (3i). White crystals, yield 70.0%; m.p. 63~65°C; (lit. [8] m.p. 65-68 °C). IR: 3061.0, 1676.1, 1610.6, 1562.3, 1473.6 cm⁻¹; ¹H-NMR: δ 8.33 (d, 1H, $J=8.0$ Hz, quinazolinone H-5),

8.04 (s, 1H, quinazolinone H-2), 7.78-7.71 (m, 2H, quinazolinone H-6,8), 7.52 (t, 1H, $J=8.0\text{Hz}$, quinazolinone H-7), 6.03-5.98 (m, 1H, CH=C), 5.32-5.26 (m, 2H, C=CH₂), 4.66 (d, 2H, $J=6.0\text{Hz}$, CH₂); ¹³C-NMR: δ 160.9 (quinazolinone C-4), 148.1 (quinazolinone C-2), 146.3 (quinazolinone C-9), 134.4 (CH=C), 131.9 (quinazolinone C-7), 127.6 (quinazolinone C-5), 127.4 (quinazolinone C-6), 126.9 (quinazolinone C-10), 122.2 (quinazolinone C-8), 119.0 (C=CH₂), 48.4 (CH₂); *Anal. Calc.* for C₁₁H₁₀N₂O (186.2): C, 70.95; H, 5.41; N, 15.04. Found: C, 70.99; H, 5.15; N, 15.02.

3-Allyl-6-fluoroquinazolin-4-one (3j). White crystals, yield 72.7%; m.p. 79.0~80.5°C; IR: 3090.5, 1666.5, 1610.6, 1574.0, 1458.2 cm⁻¹; ¹H-NMR: δ 7.99 (s, 1H, quinazolinone H-2), 7.94 (dd, 1H, $J=8.6\text{Hz}$, 2.9Hz, quinazolinone H-5), 7.73 (q, 1H, $J_{\text{H-F}}=4.5\text{Hz}$, $J_{\text{H-H}}=4.0\text{Hz}$, quinazolinone H-7), 7.51-7.46 (m, 1H, quinazolinone H-8), 6.03-5.95 (m, 1H, CH=C), 5.34-5.26 (m, 2H, C=CH₂), 4.66 (d, 2H, $J=5.6\text{Hz}$, CH₂); ¹³C-NMR: δ 162.4 (quinazolinone C-4), 159.9 (quinazolinone C-6), 145.4 (quinazolinone C-2), 144.8 (quinazolinone C-9), 131.6(CH=C), 129.9 (quinazolinone C-10), 123.3 (quinazolinone C-8), 122.7 (quinazolinone C-7), 119.1 (C=CH₂), 111.6 (quinazolinone C-5), 48.4 (CH₂); *Anal. Calc.* for C₁₁H₉FN₂O (204.2): C, 64.70; H, 4.44; N, 13.72. Found: C, 64.51; H, 4.32; N, 13.64.

3-Allyl-6-chloroquinazolin-4-one (3k). White crystals, yield 82.5%; m.p. 113~115°C; IR: 3088.7, 1678.1, 1610.6, 1574.8, 1470.0 cm⁻¹; ¹H-NMR: δ 8.28 (d, 1H, $J=2.5\text{Hz}$, quinazolinone H-5), 8.03 (s, 1H, quinazolinone H-2), 7.71-7.66 (m, 2H, quinazolinone H-7,8), 6.02-5.96 (m, 1H, CH=C), 5.34-5.26 (m, 2H, C=CH₂), 4.65 (d, 2H, $J=5.5\text{Hz}$, CH₂); ¹³C-NMR: δ 159.8 (quinazolinone C-4), 146.4 (quinazolinone C-2), 146.3 (quinazolinone C-9), 134.8 (CH=C), 133.2 (quinazolinone C-7), 131.5 (quinazolinone C-6), 129.1 (quinazolinone C-5), 126.2 (quinazolinone C-10), 123.1 (quinazolinone C-8), 119.2 (C=CH₂), 48.5 (CH₂); *Anal. Calc.* for C₁₁H₉ClN₂O (220.7): C, 59.88; H, 4.11; N, 12.70. Found: C, 59.80; H, 4.07; N, 12.64.

3-Allyl-6-bromoquinazolin-4-one (3l). [CAS number 459416-38-7, but no references given]. White crystals, yield 70.9%; m.p. 118~120°C; IR: 3068.8, 1670.4, 1610.6, 1577.9, 1464.0 cm⁻¹; ¹H-NMR: δ 8.45 (d, 1H, $J=2.5\text{Hz}$, quinazolinone H-5), 8.05 (s, 1H, quinazolinone H-2), 7.85 (dd, 1H, $J=9.2\text{Hz}$, 1.7Hz, quinazolinone H-7), 7.60 (d, 1H, $J=8.6\text{Hz}$, quinazolinone H-8), 6.02-5.96 (m, 1H, CH=C), 5.34-5.27 (m, 2H, C=CH₂), 4.65 (d, 2H, $J=5.5\text{Hz}$, CH₂); ¹³C-NMR: δ 159.6 (quinazolinone C-4), 146.7 (quinazolinone C-2), 146.5 (quinazolinone C-9), 137.5 (quinazolinone C-7), 131.5(CH=C), 129.4 (quinazolinone C-5), 129.2 (quinazolinone C-10), 123.4 (quinazolinone C-8), 121.1 (quinazolinone C-6), 119.3 (C=CH₂), 48.5 (CH₂); *Anal. Calc.* for C₁₁H₉BrN₂O (265.1): C, 49.84; H, 3.42; N, 10.57. Found: C, 49.84; H, 3.38; N, 10.65.

Bioassays

The antifungal activity of all synthesized compounds **3a-l** was tested against three pathogenic fungi, namely *Fusarium oxysporum*, *Gibberella zeae*, and *Valsa mali*, by the poison plate technique [14]. Compounds **3a-l** were dissolved in acetone (10 mL) before mixing with Potato Dextrose Agar (PDA, 90 mL). The final concentration of compounds **3a-l** in the medium were fixed at 50 $\mu\text{g/mL}$. Three kinds of fungi were incubated in PDA at 25 \pm 1 °C for 5 days to get new mycelium for antifungal assay,

then a mycelia disk of approximately 0.45 cm diameter cut from the culture medium was picked up with a sterilized inoculation needle and inoculated in the center of PDA plate. The inoculated plates were incubated at $25\pm 1^\circ\text{C}$ for 5 days. Acetone in sterilized distilled water served as control, while hymexazole was used as positive control. For each treatment, three replicates were carried out. The radial growth of the fungal colonies was measured on the sixth day and the data were statistically analyzed. The *in vitro* inhibiting effects of the test compounds on the fungi were calculated by the formula $CV = \frac{A-B}{A}$, where A represents the diameter of fungi growth on untreated PDA, B represents the diameter of fungi on treated PDA, and CV represents the rate of inhibition.

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Sample Availability: Samples of the compounds are available from authors.

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