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QUARTERLY



Synthesis and antimicrobial activity of new adamantane derivatives $I^{\odot\star}$

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Received: 25 October, 1999

Key words: adamantane derivatives, imides, antimicrobial activity

A series of fourteen derivatives of adamantane was synthesised. The new compound 4-(adamant-1-ylmethoxycarbonyl)phthalanhydride obtained from 1-adamantanemethanol and trimellitic anhydride chloride appeared very useful for preparation of a number of N-substituted phthalimides. Antimicrobial activity of the newly obtained derivatives such as, for example, 4-(adamant-1-ylmethoxycarbonyl)-N-(5-carboxypentamethylene)phthalimide or 4-(adamant-1-ylmethoxycarbonyl)-N-(L-alanyl)phthalimide was tested against *Staphylococcus aureus, Bacillus sp., Micrococcus flavus* and *Enterococcus faecium*. The minimal inhibitory concentration (MIC) for these compounds against *S. aureus* were 0.022 and 0.05 μ g/ml, respectively.

Trimellitic acid anhydride (4-carboxyphthalanhydride) (TMAA) is used in many fields of organic chemistry. Some of ester-imides derived from this anhydride, e.g. compounds with an amino acid as N-substituent and a cholesteryl, phenyl or biphenyl moiety-containing ester group, exhibit liquid-crystal properties [1-3]. Trimellitic anhydride is also frequently used as substrate for poly(ester imide)s synthesis. The combination of the rigid and flat phthalimide core with mesogenic properties of the ester structure endows such polymers with a liquid crystalline character [4–8]. Many compounds possessing imide rings in their structure exhibit biological activity. For example, some imides are valuable

*Presented at the 7th International Symposium on Molecular Aspects of Chemotherapy, September 8-11, 1999, Gdańsk, Poland.

²The study was supported by the Polish Committee for Scientific Research grant 4 PO5F 02712.

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Abbreviations: AM, 1-adamantanemethanol; AU, N-(adamant-1-yl)urea; DMF, dimethylformamide; Me₂SO, dimethylsulfoxide; MIC, minimal inhibitory concentration; TMAA, trimellitic acid anhydride; TMAC, trimellitic anhydride chloride; TNF, tumor necrosis factor.

substrates for the production of anti-seizure drugs [9]. The most known of the clinically useful imides is the antiviral drug amantidine (1-aminoadamantane) [10]. Another field where amantidine and many related derivatives are successfully employed, is the treatment of certain neurological disorders, e.g. Parkinson's disease [11]. There are only few papers concerning the antimicrobial activity of adamantane derivatives [12-14]. In this study we show that certain easily synthesise 1-adamantanemethanol esters of various N-substituted phthalimide 4-carboxylates exhibit a distinct antimicrobial activity. It is worth to note that among structurally similar compounds N-(1-adamantyl)maleimide shows anticancer activity in mice and inhibits herpes simplex virus replication in vitro [14, 15]. N-Adamantylphthalimide induces tumor necrosis factor (TNF- α) enhancing activity induced by 12-O-tetradecanoylphorbol-13-acetate in human leukaemia HL-60 cells [16]. We expect that a combination of both types of compounds, those with a flat phthalimide core and those with a three-dimensional "box-like" adamantane structure, will provide a broad range of active compounds for biological and pharmacological investigations.

MATERIALS AND METHODS

General methods

Melting points were taken in open capillary tubes on a Gallenkamp 5 melting point apparatus and were uncorrected. The structures of products were confirmed by elemental analysis, FTIR and ¹H NMR spectroscopy. The NMR spectra were measured on a Varian Gemini 200 MHz spectrometer in CDCl₃ or d₆-Me₂SO solutions. Column flash chromatography was performed on silica gel 60 (Merck). FTIR spectra were recorded on a Perkin Elmer 2000 apparatus using the KBr pellet method. Adamantane and phthalic acid derivatives were purchased from Aldrich. The specific rotation of compounds was determinated on a Perkin Elmer polarimeter at 20°C in chloroform ($c = 1 \text{ g}/100 \text{ cm}^3$) using the D line of sodium. Preliminary testing of the antimicrobial activity of the newly synthesised compounds was performed by the disc diffusion method using Mueller-Hinton agar medium under standard conditions as described by NCCLS [17]. Sterile filter paper discs were soaked in test compounds solutions prepared in EtOH-Me₂SO mixture (v/v, 1:1). The results were read following 18 h incubation at 37°C (for *M. flavus* at 30°C). Compounds showing distinct antimicrobial activity in the above test were next tested for MIC (minimal inhibitory concentration) in liquid Mueller-Hinton medium according to the appropriate NCCLS protocol, using original stock solutions [18]. Staphylococcus aureus NCTC 4163 and Enterococcus faecium ATCC 6057 were purchased from the National Institute of Hygiene (Warsaw, Poland) Bacillus subtilis H17recand Bacillus subtilis M45rec⁺ were kindly donated by dr. T. Kada from the National Institute of Genetics (Misima, Japan) the other microorganisms used were from own collection of the Department of Pharmaceutical Microbiology, Medical University (Warsaw, Poland).

Syntheses

N-(Adamant-1-yl)trimellitimide (1): 1.92 g (10 mmol) of trimellitic anhydride and 2.28 g (15 mmol) of 1-aminoadamantane were dissolved in dry DMF (15 cm³) and refluxed for 5 h. Then the mixture was cooled and poured into aqueous 1% HCl. Crude product was filtered and crystallised from 90% ethanol to give 1 (1.17 g, 52%); m.p. 261°C; FTIR (cm⁻¹): C=O_{imide} 1706 and 1768. ¹H NMR (CDCl₃) δ (ppm): 1.75–2.52 (3m, H_{adamantane}), 7.85–8.45 (m, H_{arom}.). Elemental analysis: calculated for C₁₉H₁₉NO₄ : C, 70.14; H, 5.89; N, 4.30. Found: C, 70.55; H, 5.95; N, 4.23.

N-(Adamant-2-yl)trimellitimide (2): Compound 2 was synthesised analogously to the procedure described above for 1 starting from

2-aminoadamantane and trimellitic anhydride in a yield 0.90 g (40%); m.p. 228°C; FTIR (cm⁻¹): C=O_{imide} 1714 and 1771. ¹H NMR (CDCl₃) δ (ppm): 1.79–2.92 (3m, H_{adamantane}), 7.85–8.49 (m, H_{arom.}). Elemental analysis: calculated for C₁₉H₁₉NO₄ : C, 70.14; H, 5.89; N, 4.30. Found: C, 70.50; H, 5.89; N, 4.22.

4-(N¹-(Adamant-1-yl)ure-N²-ylcarbonyl))phthalic acid (3): 0.420 g (2 mmol) of trimellitanhydride chloride and 0.39 g (2 mmol) of N-(adamant-1-yl)urea were dissolved in anhydrous pyridine and the reaction mixture was stirred overnight at room temperature. A pale yellow solution was evaporated to an oil and the residue was coevaporated with water to remove the rest of pyridine. Twice crystallisation from methanol water gave 3 (345 mg, 45%) m.p. 205°C; FTIR (cm⁻¹): C=O_{amide} 1680, C=O_{acid} 1695, C=O_{urea} 1728. ¹H NMR (d_6-Me_2SO) $\delta(ppm)$: 1.62–2.62 (3 m, Hadamantane), 7.65-8.22 (m, Harom.). Elemental analysis: calculated for $C_{20}H_{22}N_2H_6$: C, 62.17; H, 5.74; N, 7.25. Found: C, 62.10; H, 5.54; N, 7.41.

4-(Adamant-1-yl-methoxycarbonyl)phthalanhydride (4): The synthesis of 4 was performed from 1-adamantanemethanol and trimellitic anhydride chloride according to the method described previously [19, 20]. Crude product was crystallised with benzene. Yield 42%; m.p. 158° C; FTIR (cm⁻¹): C=O_{ester} 1731, C=O_{anh}. 1785 and 1857.

4-(Adamant-1-ylmethoxycarbonyl)phthalic acid (5): 1.70 g (5 mmol) of 4 was heated with water for 15 min. After cooling the crystals precipitated. Yield 1.43 g (40%); m.p. 197°C. FTIR (cm⁻¹): C=O_{acid} 1702, C=O_{ester} 1732. ¹H NMR (CDCl₃) δ (ppm): 1.65–2.04 (m, H_{adamantane}), 3.99 (s, CH₂O), 7.82–8.54 (m, H_{arom.}). Elemental analysis: calculated for C₂₀H₂₂O₆: C, 67.03; H, 6.19. Found: C, 67.10; H, 6.29.

N-substituted-4-(adamant-1-ylmethoxycarbo-nyl)phthalimides **(6–12)**: 1.36 g (5 mmol) of **4** and appropriate amino acids were refluxed in dry DMF for 3 h. Then mixtures were poured into diluted HCl (1%). Crude products were

purified by column chromatography (silica gel) using chloroform/methanol/acetic acid (v/v, 25:1:traces) as an eluent.

4-(Adamant-1-ylmethoxycarbonyl)-N-carboxymethylenephthalimide **(6)**: (m.p. 242°C, 1.03 g, 65%); FTIR (cm⁻¹): C=O_{imide} 1722 and 1773; ¹H NMR (CDCl₃) δ (ppm): 1.63–2.02 (m, H_{adamantane}), 3.99 (s, CH₂O), 4.52 (s, N– CH₂), 7.90–8.48 (m, H_{arom}.). Elemental analysis: calculated for C₂₂H₂₃NO₆ : C, 66.49; H, 5.83; N, 3.52. Found: C, 66.44; H, 5.89; N, 3.44

4-(Adamant-1-ylmethoxycarbonyl)-N-(2-carboxydimethylene)phthalimide (7): (m.p. 162°C, 0.99 g, 60%); FTIR (cm⁻¹): C=O_{imide} 1723 and 1778; ¹H NMR (CDCl₃) δ (ppm): 1.62–2.02 (m, H_{adamantane}), 2.80 (t, CH₂C=O), 3.97 (s, CH₂O), 4.02 (t, N-CH₂), 7.93–8.44 (m, H_{arom}.). Elemental analysis: calculated for C₂₃H₂₅NO₆ : C, 67.14; H, 6.12; N, 3.40. Found: C, 67.25; H, 6.02; N, 3.52.

4-(Adamant-1-ylmethoxycarbonyl)-N-(3-carboxytrimethylene)phthalimide (8): (m.p. 161°C, 0.95 g, 56%); FTIR (cm⁻¹): C=O_{imide} 1718 and 1779; ¹H NMR (CDCl₃) δ (ppm): 1.63– 1.73 (m, H_{adamantane}), 1.63–1.65 [m, (CH₂)₃], 2.03 (m, -CH₂-), 2.43 (t, CH₂C=O), 3.79 (t, N-CH₂), 3.98 (s, CH₂CO), 7.95–8.45 (m, H_{arom}.). Elemental analysis: calculated for C₂₄H₂₇NO₆ : C, 67.75; H, 6.40; N, 3.29. Found: C, 67.34; H, 6.49; N, 3.33.

4-(Adamant-1-ylmethoxycarbonyl)-N-(5-carboxypentamethylene)phthalimide (9): (m.p. 168°C, 1.23 g, 68%); FTIR (cm⁻¹): C=O_{imide} 1718 and 1779; ¹H NMR (CDCl₃) δ (ppm): 1.63–2.02 (m, H_{adamantane}), 3.99 (s, CH₂O), 4.52 (s, NCH₂), 7.90–8.48 (m, H_{arom}.). Elemental analysis: calculated for C₂₆H₃₁NO₆ : C, 68.86; H, 6.89; N, 3.09. Found: C, 68.76; H, 6.66; N, 2.99.

4-(Adamant-1-ylmethoxycarbonyl)-N-(L-alanyl)phthalimide (10L): (m.p. 117°C, 0.69 g, 42%); FTIR (cm⁻¹): C=O_{imide} 1725 and 1780; ¹H NMR (CDCl₃) δ (ppm): 1.62–2.02 (m, H_{adamantane} and CH₃), 3.97 (s, CH₂O), 4.98 (q, N–CH), 7.90–8.44 (m, H_{arom}.); [α] ²⁰_D = -7.6°. Elemental analysis: calculated for: C₂₃H₂₅- NO₆ : C, 67.14; H, 6.12; N, 3.40. Found: C, 67.28; H, 6.01; N, 3.50.

4-(Adamant-1-ylmethoxycarbonyl)-N-(D-alanyl)phthalimide (10D): (m.p. 118°C, 0.68 g 40%); FTIR (cm⁻¹): C = O_{imide} 1724 and 1780; ¹H NMR (CDCl₃) δ (ppm): 1.62–2.02 (m, H_{adamantane} and CH₃), 3.97 (s, CH₂O), 4.98 (q, N–CH), 7.90–8.44 (m, H_{arom.}); [α] $_{D}^{20}$ = +7.6°. Elemental analysis: calculated for: C₂₃H₂₅-NO₆ : C, 67.14; H, 6.12; N, 3.40. Found: C, 67.27; H, 6.00; N, 3.45.

4-(Adamant-1-ylmethoxycarbonyl)-N-(L-phenylalanyl)phthalimide (11L): (m.p. 115°C, 1.07 g, 55%); FTIR (cm⁻¹): C=O_{imide} 1725 and 1780; ¹H NMR (CDCl₃) δ (ppm): 1.61–2.02 (m, H_{adamantane}), 3.54 (m, CH₂–phenyl), 3.94 (s, CH₂O), 5.08 (m, N–CH), 7.16 (s, H_{phenylala}), 7.37–8.34 (m, H_{arom}); [α] ²⁰_D= -94.5°. Elemental analysis: calculated for: C₂₉H₂₉NO₆ : C, 71.44; H, 6.00; N, 2.87. Found: C, 72.00; H, 6.03; N, 2.99.

4-(Adamant-1-ylmethoxycarbonyl)-N-(D-phenylalanylo)phthalimide (11D): (m.p. 115°C, 1.11 g, 57%); FTIR (cm⁻¹): C=O_{imide} 1725 and 1781; ¹H NMR (CDCl₃) δ (ppm): 1.61–2.02 (m, H_{adamantane}), 3.54 (m, CH₂–phenyl), 3.94 (s, CH₂O), 5.08 (m, N–CH), 7.17 (s, H_{phenylala}), 7.37–8.35 (m, H_{arom}); [α] ²⁰_D = +94.3°. Elemental analysis: calculated for: C₂₉H₂₉NO₆ : C, 71.44; H, 6.00; N, 2.87. Found: C, 71.96; H, 6.03; N, 2.95.

4-(Adamant-1-ylmethoxycarbonyl)-N-(4-carboxyphenyl)phthalimide (12): (m.p. 289°C, 1.38 g, 75%); FTIR (cm⁻¹): C = O_{imide} 1708 and 1775; ¹H NMR (CDCl₃) δ (ppm): 1.64–2.02 (m, H_{adamantane}), 4.02 (s, CH₂O), 7.37–8.60 (m, H_{arom.}). Elemental analysis: calculated for: C₂₇H₂₅NO₆ : C, 70.58; H, 5.48; N, 3.05. Found: C, 71.01; H, 5.50; N, 3.11.

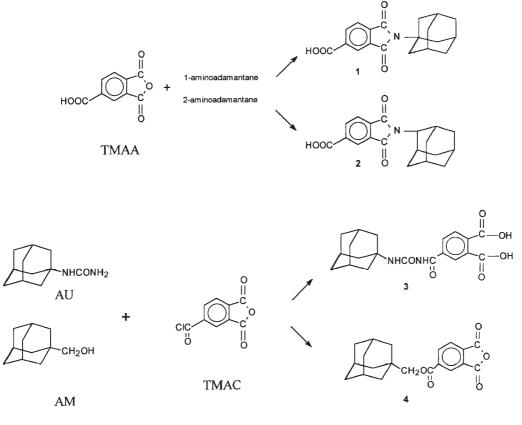
RESULTS AND DISCUSSION

Chemical synthesis

Highly reactive trimellitic acid anhydride (TMAA) (Scheme 1) was utilised for the synthesis of N-substituted phthalimide derivatives in the reaction with 1-aminoadamantane and 2-aminoadamantane. The reaction was performed in refluxing anhydrous DMF to give, respectively: N-(adamant-1-yl)trimellitimide 1 and N-(adamant-2-yl)trimellitimide 2. Stirring of commercially available trimellitic anhydride chloride (TMAC) with N-(adamant-1-yl)urea (AU) in pyridine at ambient temperature followed by hydrolytic opening of the anhydride yielded 4- $(N^1$ -(adamant-1-yl)urea- N^2 -ylcarbonyl))phthalic acid **3**. Esterification of TMAC with 1-adamantanemethanol (AM) analogously to the method described earlier for cholesteryl derivate synthesis [1-3, 19, 20] yielded 4-(adamant-1-ylmethoxycarbonyl)phthalanhydride 4. Thermally facilitated hydrolysis of 4 in water gave 4-(adamant-1-ylmethoxycarbonyl)phthalic acid 5 (Scheme 2). Heating 4 with ω -aminomethylenecarboxylic acids, D- and L-alanine and enantiomers of phenylalanine, as well *p*-aminobenzoic acid analogously to the aforementioned procedure resulted in N-substituted trimellitimides 6–12, respectively.

Microbiological studies

The antimicrobial activity of adamantyl derivatives of phthalimide was first tested by the agar disc-diffusion method against Gram-positive bacteria: Staphylococus aureus, Micrococcus flavus, Enterococcus faecium and certain strains of Bacillus. Gram-negative bacteria: Bordella bronchiseptica, Pseudomonas aeruginosa and strains belong to the family Enterobacteriaceae as well as the fungus Candida albicans were resistant to all tested compounds. Next, the minimal inhibition concentration (MIC) of the most active compounds was determined in liquid Mueller-Hinton medium (Table 1). A particularly strong antimicrobial activity, comparable to that of clinically used antibiotics, was observed for N-amino-acid substituted derivatives 10 and 11. Both chiral isomers, i.e. 10L, 10D, and 11L, 11D were highly active



Scheme 1.

against *Micrococcus flavus* and *Staphylococcus* strains. In contrast, **11L** and **11D** were practically inactive against *Enterococcus faecium*,

whereas **10L** and **10D** showed moderate activity against this microorganism. The most distinct differences in the antimicrobial activ-

Table 1. Sensitivity of Gram-positive *cocci* strains to certain adamantyl derivatives of substituted phthalimides

Bacteria strain	Minimal inhibitory concentration (MIC) in μ g/ml												
Bacteria strain	L	Compound											
	1	2	3**	6	7	8	9	10L	10D	11L	11D	12	
S. aureus ATCC 25923	7.5	7.5	10	7.5	5.0	5.0	0.022	0.22	0.22	0.05	0.05	na	
<i>S. aureus</i> ATCC 6538P	7.5	7.5	10	7.5	5.0	5.0	0.08	0.22	0.22	0.015	0.05	na	
S. aureus NCTC 4163	7.5	7.5	10	7.5	5.0	5.0	0.08	0.22	0.22	0.05	0.05	na	
S. aureus 31	7.5	7.5	na	7.5	5.0	2.5	0.08	0.15	0.15	0.05	0.05	15.0	
<i>M. flavus</i> NCIB 8166	2.5	3.5	na	10.0	2.5	2.5	0.8	0.55	0.55	0.022	0.8	na	
E. faecium ATCC 6057	na*	na	na	na	na	5-7.5	na	7.5	7.5	na	na	na	

*na – no activity at concentration > 50 μ g/ml; **compounds 4 and 5 were inactive against bacteria strains tested.

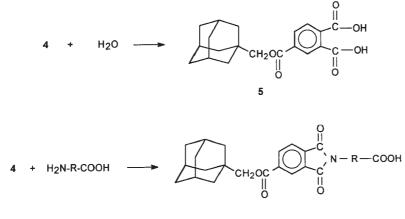
Destania staria	Diameter of growth inhibition area (mm)												
Bacteria strain		C o m p o u n d *											
	1	2	3**	6	7	8	9	10L	10D	11L	11D	12	Control
B. subtilis ATCC 6633	14	13	0	17	18	11	12	29	43	39	35	10	16 ^a
B. cereus ML 98	15	14	10	18	20	17	15	33	25	33	37	10	14^{a}
B. stearother- mophilus ATCC 7953	15	14	10	18	20	18	15	40	28	33	33	10	14^{a}
B. subtilis M45 rec	14	14	10	19	19	17	15	45	33	33	33	10	28^{b}
B. subtilis H17 rec^+	14	14	10	19	19	19	15	39	39	29	33	10	12^{b}

Table 2a. Sensitivity of Bacillus strains to adamantyl derivatives of substituted phthalimide

*800 μ g per 8 mm disc; **compounds **4** and **5** were inactive; ^anitrofurantoine 300 μ g per 8 mm disc; ^b4-nitrochinoline *N*-oxide 1 μ g per 8 mm disc

ity of both chiral isomers 11L and 11D were observed for *S. aureus* ATCC 6538P and *M. flavus* strains.

Noteworthy are large differences in the antimicrobial activity obtained by the disc-diffusion method and MIC method with *Bacillus*





compound	R	compound	R
6	- CH ₂	10Land 10D	— сн — І сн ₃
7	-(CH ₂) ₂ -	11Land 11D	
8	- (CH ₂) ₃ -	12	
9	- (CH ₂) ₅ -		

Scheme 2

	Minimal inhibitory concentration (MIC) in μ g/ml						
Bacteria strain	Compound						
	1-9, 10L and D, 12	11L	11D				
B. subtilis ATCC 6633	>50.0	37.5	30.0				
B. cereus ML 98	>50.0	35.0	30.0				
B. stearothermophilus ATCC 7953	>50.0	30.0	30.0				
B. subtilis M45 rec ⁻	>50.0	18.5	35.0				
B. sublitis H17 rec^+	>50.0	35.0	35.0				

Table 2b. Sensitivity of Bacillus strains to adamantyl derivatives of substituted phthalimides

strains (Tables 2a and 2b). A similar discrepancy was previously observed for these aerobic bacteria in antimicrobial tests of essential oil from *Tanacetum partenium* [21]. The high sensitivity of *Bacillus* strains may be useful in testing the antimicrobial activity of next adamantane derivatives, the synthesis of which we plan in the nearest future.

The genetically modified *Bacillus* rec^+ and rec^- strains employed in the present study were previously shown to be useful in screening the genotoxicity of antibiotics under development [22]. Results obtained in the course of the present study (details not shown) showed no genotoxicity of the newly synthesised adamantane antimicrobials. We plan to expand these studies to include other N-amino-acyl-substituted phthalimidecarboxylic acid esters of adamantane.

REFERENCES

- Białecka-Florjańczyk, E. & Orzeszko, A. (1993) Some novel liquid crystalline derivatives of 4-carboxyphthalimide. *Liq. Cryst.* 15, 255– 258.
- Białecka-Florjańczyk, E., Orzeszko, A. & Śledzińska, I. (1997) Cholesteryl ester imides with oxyethylene and methylene tails. *Mol. Cryst. Liq. Cryst.* 300, 1–8.
- Białecka-Florjańczyk, E., Orzeszko, A., Śledzińska, I. & Górecka, E. (1999) Evidence

of the smectic antiphase in 4-decyloxybiphenyl ester imide derivatives. J. Mater. Chem. 9, 371-374.

- Śledzińska, I., Białecka-Florjańczyk, E. & Orzeszko, A. (1996) Synthesis and liquid crystalline properties of cholesteryl bisester imides with poly(ethylene oxide)s as central spacer. *Eur. Polym. J.* 32, 1345-1350.
- Kricheldorf, H.R. (1999) Liquid-crystalline polyimides. Adv. Polym. Sci. 141, 84-188.
- 6. Loncrini, D.F. (1966) Aromatic poly(ester imide)s. J. Polym. Sci. Part A-1. 4, 1531-1541.
- 7. Kricheldorf, H.R. (1994) Liquid-crystalline polyimides. *Mol. Cryst. Liq. Cryst.* 254, 87–108.
- 8. Kricheldorf, H.R. & Pakull, R. (1988) New polymer synthesis. 20. Liquid-crystalline poly(ester imide)s derived from trimellitic anhydride, α, ω -diaminoalkanes and 4,4'-dihydroxybiphenyl. *Macromolecules* **21**, 551–557.
- Mallawaarachchi, W., Simmonds, R.J. & Parry, D.E. (1989) The geometry of N-hydroxymethyl compounds. Part 3: Geometries of N-(hydroxymethyl)amides related to reactivities. Anti-Cancer Drug Des. 4, 233-240.
- 10. Davies, W.L., Grunert, R.R., Haff, R.F., McGahen, J.W., Neumayer, E.M., Paulshock, M., Watts, J.C., Wood, T.R., Hermann, E.C. & Hoffmann, C.E. (1964) Antiviral activity of 1-adamantanamine (amantidine). Science 144, 862-863.

- 11. Schwab, R.S., England, A.C., Jr., Poskanzer, D.C. & Young, R.R. (1969). Amantadine in the treatment of Parkinson's disease. J. Am. Med. Assoc. 208, 1168-1170.
- 12. Płachta, D. & Starościak, B. (1994) Synthesis of some 1-alkylamino-2,4-adamantylpheno-xy)ethane derivatives. Acta Polon. Pharm. 51, 51-54.
- 13. Papadaki-Valiraki, A., Papakonstantinou-Garoufalias, S., Makaros, P., Chytyroglou-Lada, A., Hosoya, M., Balzarini, J. & De Clercq, E. (1993) Synthesis, antifungal, antibacterial and antiviral effects of some adamantaneketoxime ethers. *Il Farmaco* 48, 1091-1102.
- Wang, J.-J., Wang, S.-S., Lee, Ch.-F., Chung, M.-A. & Chern, Y.-T. (1997) *In vitro* antitumor and antimicrobial activities of N-substituts of maleimide by adamantane and diamantane. *Chemotherapy* 43, 182–189.
- **15.** Takatori, Y. (1992) *N-1-adamantylmaleimide*. Jpn. Kokai Tokyo Koho, Japan Pat. 6028961.
- 16. Sabata, Y., Shichita, M., Sasaki, K., Hashimoto, Y. & Iwasaki, S. (1995) N-Alkylphthalimides: Structural requirement of thalidomidal action on 12-O-tetradecanoylphorbol-13-acetate-induced tumor necrosis factor α production by human leukemia HL-60 cells. Chem. Pharm. Bull. 43, 177–179.

- 17. National Committee for Clinical Laboratory Standards NCCLS Approved Standard Document M7-A. (1985) Villanova, PA, U.S.A.
- 18. National Committee for Clinical Laboratory Standards 5th edn. Approved Standard NCCLS Document M2-A5. (1993) Villanova, PA, U.S.A.
- Białecka-Florjańczyk, E. & Orzeszko, A. (1999) Sposób wytwarzania ciekłokrystalicznego bezwodnika 4-cholesteryloksykarbonyloftalowego. *Pol. Patent* 175659 (in Polish).
- 20. Białecka-Florjańczyk, E. & Orzeszko, A. (1999) Sposób wytwarzania ciekłokrystalicznych estrów cholesterolowych pochodnych bezwodnika 4-cholesteryloksykarbonyloftalowego. *Pol. Patent* 176084 (in Polish).
- 21. Kalodera, Z., Popejnak, S., Blazevic, N., Petrak, T. (1997) Chemical composition and antimicrobial activity of *Tanacetum parthenium* essential oil. *Pharmazie* 52, 885–886.
- 22.Kada, T., Hirano, K. & Shirazu, Y. (1980) Screening of environmental chemical mutagens by the rec-assay system with *Bacillus subtilis*. *Chemical Mutagens* (de Serres, F.J. & Hollaender, A., eds.) vol. 6, pp. 149-173, Plenum Pub. Co., New York.