



Communication

# Synthesis of Deuterium-Labeled Vitamin D Metabolites as Internal Standards for LC-MS Analysis

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**Abstract:** Blood levels of the vitamin D<sub>3</sub> (D<sub>3</sub>) metabolites 25-hydroxyvitamin D<sub>3</sub> (25(OH)D<sub>3</sub>), 24R,25-dihydroxyvitamin D<sub>3</sub>, and 1 $\alpha$ ,25-dihydroxyvitamin D<sub>3</sub> (1,25(OH)<sub>2</sub>D<sub>3</sub>) are recognized indicators for the diagnosis of bone metabolism-related diseases, D<sub>3</sub> deficiency-related diseases, and hypercalcemia, and are generally measured by liquid-chromatography tandem mass spectrometry (LC-MS/MS) using an isotope dilution method. However, other D<sub>3</sub> metabolites, such as 20-hydroxyvitamin D<sub>3</sub> and lactone D<sub>3</sub>, also show interesting biological activities and stable isotope-labeled derivatives are required for LC-MS/MS analysis of their concentrations in serum. Here, we describe a versatile synthesis of deuterium-labeled D<sub>3</sub> metabolites using A-ring synthons containing three deuterium atoms. Deuterium-labeled 25(OH)D<sub>3</sub> (**2**), 25(OH)D<sub>3</sub>-23,26-lactone (**6**), and 1,25(OH)<sub>2</sub>D<sub>3</sub>-23,26-lactone (**7**) were synthesized, and successfully applied as internal standards for the measurement of these compounds in pooled human serum. This is the first quantification of 1,25(OH)<sub>2</sub>D<sub>3</sub>-23,26-lactone (**7**) in human serum.

**Keywords:** vitamin D; deuterium labeling; liquid-chromatography tandem mass spectrometry; measurement of vitamin D metabolites in blood



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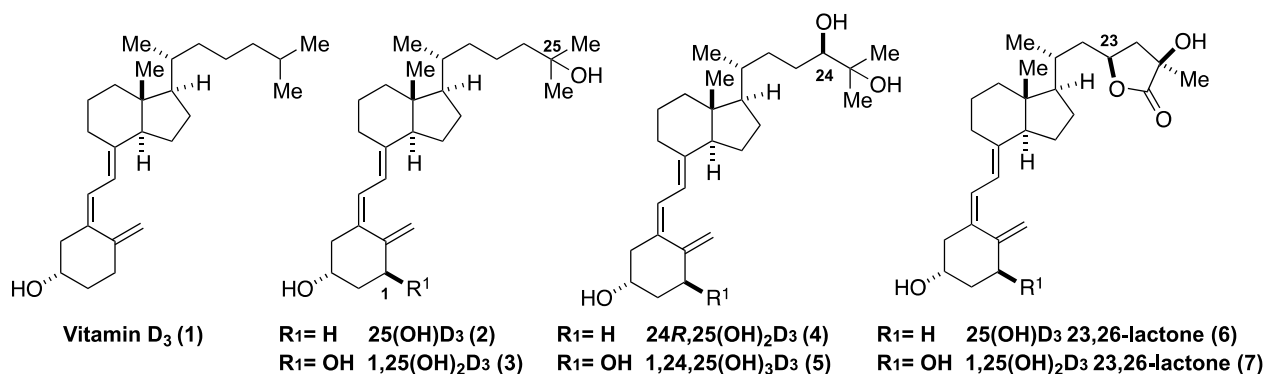


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## 1. Introduction

Vitamin D<sub>3</sub> (D<sub>3</sub>) (**1**) is metabolized by members of the cytochrome P450 (CYP) family to generate more than 50 compounds in vivo. Among them, 25-hydroxyvitamin D<sub>3</sub> (25(OH)D<sub>3</sub>) (**2**) is generated from D<sub>3</sub> (**1**) by CYP2R1 and/or CYP27A1-mediated hydroxylation at C25 in the liver, and the resulting 25(OH)D<sub>3</sub> (**2**) is further metabolized to the active form of D<sub>3</sub>, 1 $\alpha$ ,25-dihydroxyvitamin D<sub>3</sub> (1,25(OH)<sub>2</sub>D<sub>3</sub>) (**3**), by CYP27B1-mediated oxidation at C1 $\alpha$  in the kidneys (Figure 1). 1,25(OH)<sub>2</sub>D<sub>3</sub> (**3**) plays a key role in the regulation of bone metabolism in vivo [1]. In normal conditions, production of **3** from **2** is strictly controlled by the concentrations of calcium and parathyroid hormone (PTH) in the blood, and thus the concentration of **3** in the blood is a useful indicator of functional status [2] and is helpful in the diagnosis of diseases such as hypercalcemia, hyperphosphatemia, rickets, and bone metabolism-related diseases [3,4]. The concentration of **2** in the blood is also useful as an indicator for the diagnosis of various vitamin D deficiency-related diseases [3,4]. Recently, various D<sub>3</sub> metabolites, mostly oxidized at the D-ring side chain, have also been found to show biological activities. For example, 20S,25(OH)<sub>2</sub>D<sub>3</sub>, which is generated by CYP11A1, inhibits the growth of keratinocytes, leukemia cells, and melanoma cells [5–9], while 24R,25(OH)<sub>2</sub>D<sub>3</sub> (**4**), produced by CYP24A1, shows inhibitory activity against various cancer cell lines [10,11]. Further, 25(OH)D<sub>3</sub>-23,26 lactone (**6**) and 1,25(OH)<sub>2</sub>D<sub>3</sub>-23,26 lactone (**7**), which are thought to be final metabolites of D<sub>3</sub>, show antagonistic activity

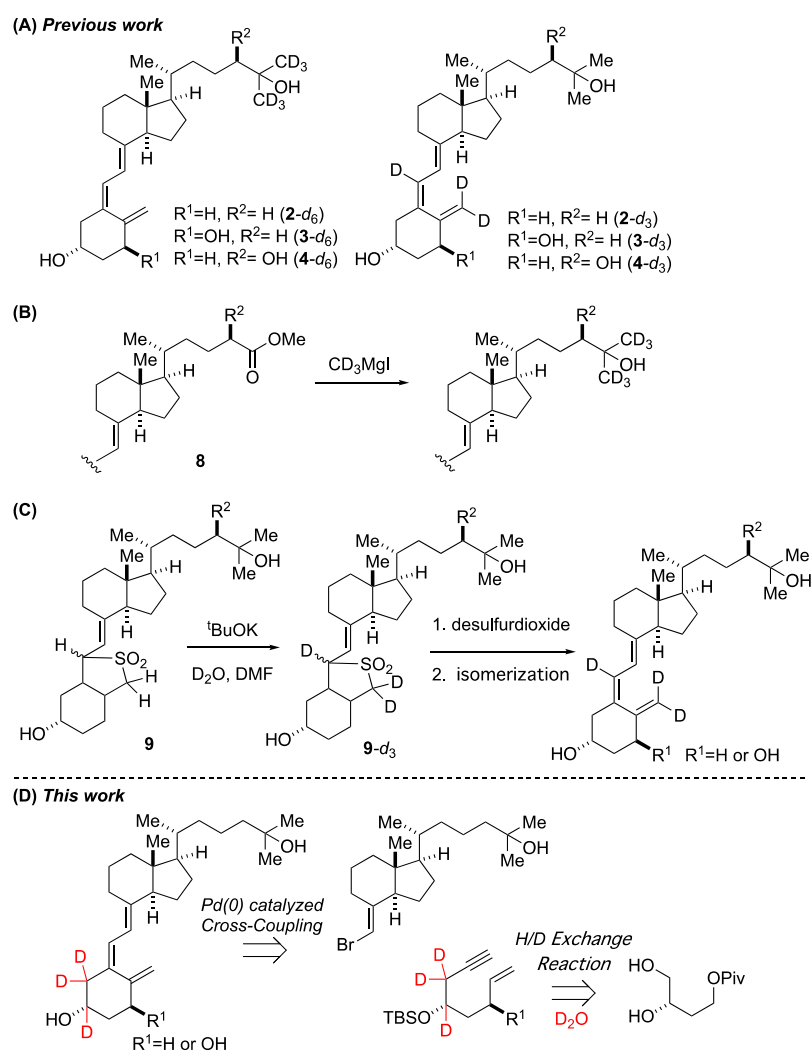
towards  $1,25(\text{OH})_2\text{D}_3$  [12–14], thereby, inhibiting bone formation and resorption. Recently, compound **7** was reported to inhibit fatty acid oxidation [15]. Thus, there is a need to measure the blood levels of these metabolites.



**Figure 1.** Structures of vitamin D<sub>3</sub> (**1**) and its metabolites (**2**)–(**7**).

The concentrations of metabolites **2** and **3** in blood have been measured for clinical purposes by radioimmunoassay (RIA) or chemiluminescent enzyme immunoassay (CLEIA) [16]. They, however, have disadvantages such as the need to handle radioactive materials, and insufficient discrimination of vitamin D metabolites by antibodies [17]. More recently, a liquid-chromatography tandem mass spectrometry (LC-MS/MS) method has been developed to determine the concentration of multiple vitamin D metabolites simultaneously in blood [18]. However, LC-MS/MS-based measurement also has some problems, such as the low ionization efficiency of vitamin D derivatives and interference by contaminants including multiple D<sub>3</sub> metabolites in the blood. To address these issues, several approaches have been investigated. Cookson-type reagents have been developed to improve the ionization efficiency of D<sub>3</sub> metabolites, affording high sensitivity even at low abundance [19,20]. The isotope dilution method has also been applied to avoid interference from contaminants in the blood. This method requires a stable isotope-labeled compound as an internal standard, and so far, deuterium-labeled  $25(\text{OH})_2\text{D}_3$  (**2**),  $1,25(\text{OH})_2\text{D}_3$  (**3**), and  $24R,25(\text{OH})_2\text{D}_3$  (**4**), in which deuterium is introduced at C26, C27, C6, and C19, have been synthesized (Figure 2) [21–25].

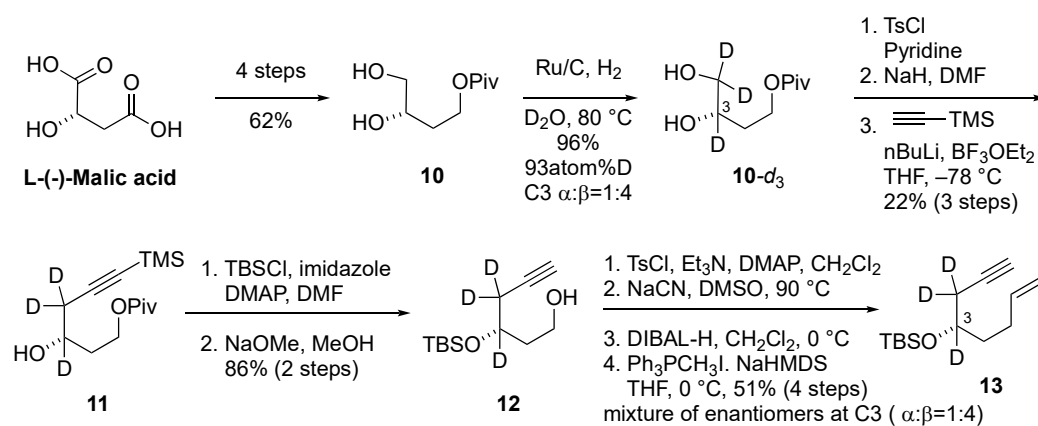
In the synthesis of the deuterium-labeled metabolites **2–4-d<sub>6</sub>**, deuterium was introduced into the side chain at C26 and C27 by reacting esters **8** with deuterated Grignard reagent,  $\text{CD}_3\text{MgBr}$  (Figure 2B) [21,22]. On the other hand, **2–3-d<sub>3</sub>** were synthesized by reacting  $\text{SO}_2$  adducts of cyclic compounds **9** derived from D<sub>3</sub> with deuterium oxide ( $\text{D}_2\text{O}$ ) [23–25]. In both strategies, the range of metabolites that can be synthesized is limited due to the restrictions imposed by the use of steroid precursors. Therefore, a more versatile approach is required. Convergent strategies, with coupling between CD-ring and A-ring moieties, have been widely applied for the synthesis of D<sub>3</sub> derivatives [26,27]. Since the CD-ring structures of the metabolites are diverse, whereas the A-ring structures are relatively constant, we considered that deuterium-labeled A-ring synthons would be suitable for the preparation of a variety of deuterium-labeled D<sub>3</sub> metabolites (Figure 2D). In addition, labeling in the A-ring has an advantage in metabolism studies because the side chains of the D<sub>3</sub> are well known to be enzymatically metabolized easily. In this study, we have developed a synthesis of deuterium-labeled A-ring precursors **13-d<sub>3</sub>** and **16-d<sub>3</sub>** incorporating three deuterium atoms. These precursors were coupled with CD-ring moieties **17** and **18** to afford deuterium-labeled  $25(\text{OH})\text{D}_3$ -*d*<sub>3</sub> (**2-d<sub>3</sub>**) and vitamin D lactones  $25(\text{OH})\text{D}_3$ -23,26-lactone-*d*<sub>3</sub> (**6-d<sub>3</sub>**) and  $1,25(\text{OH})_2\text{D}_3$ -23,26-lactone-*d*<sub>3</sub> (**7-d<sub>3</sub>**). We also confirmed that the concentrations of **2**, **6**, and **7** in human serum could be measured by LC-MS/MS using the corresponding deuterium-labeled compounds as the internal standards (IS) (see Supplementary Materials).



**Figure 2.** Synthetic strategy of deuterium-labeled  $D_3$  metabolites. (A) Structures of reported deuterium-labeled  $D_3$  metabolites. (B) Previous work on the synthesis of 2–4- $d_6$ . (C) Previous work on the synthesis of 2–4- $d_3$ . (D) This work: general synthesis of deuterium-labeled  $D_3$  metabolites.

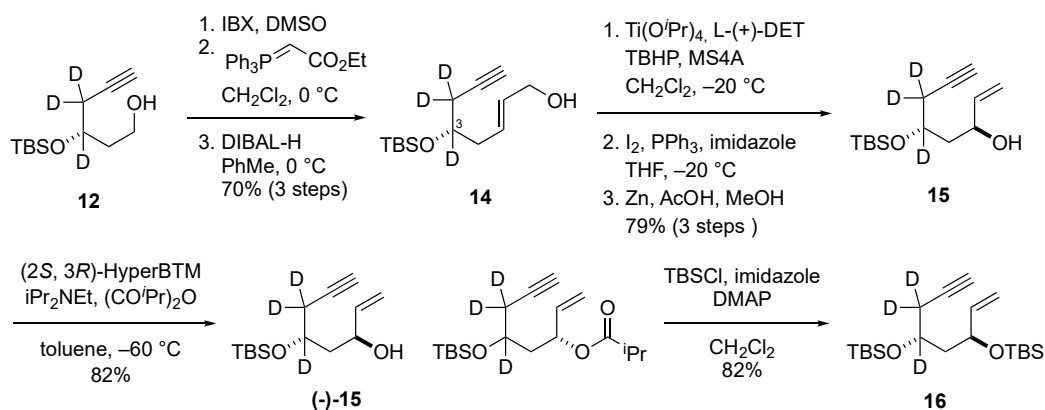
## 2. Results

We employed a convergent strategy using the palladium-catalyzed coupling reaction of enyne-type deuterium-labeled A-ring precursors **13- $d_3$**  and **16- $d_3$**  with bromoolefins **17** and **18** as the CD-ring moieties. The deuterium atoms in **13- $d_3$**  and **16- $d_3$**  were introduced by the H/D exchange at the  $\alpha$ -position of the alcohol, as reported by Sajiki et al. [28]. Our synthesis of deuterium-labeled enyne **13** commenced with the H/D exchange reaction of alcohol **10**, which was obtained from L-(–)-malic acid in 4 steps (Scheme 1) [29]. The alcohol **10** was subjected to the H/D exchange reaction with a catalytic amount of Ru/C in  $D_2O$  at 80 °C under an  $H_2$  atmosphere to afford **10- $d_3$**  deuterium-labeled at C3 and C4 in a 96% yield with over 93% deuteride content [28]. In this reaction, the stereochemistry at C3 was isomerized (4:1 ratio of  $\alpha$ -**10a** and  $\beta$ -**10b**). The deuterium-labeled alcohol **10** (enantiomeric mixture) was converted into alkyne **11** by tosylation of the primary alcohol followed by epoxidation with NaH and reaction with TMS-acetylene (22% yield from **10- $d_3$** ). The hydroxyl group in alkyne **11** was protected with TBS ether, followed by deprotection of the TMS and pivaloyl groups with NaOMe in MeOH to give the alcohol **12** in an 86% yield from **11**. Enyne **13** was obtained in a 51% yield from **12** via 4 steps, (i) tosylation of the primary alcohol; (ii) cyanation with NaCN; (iii) reduction of the nitrile group with DIBAL-H to aldehyde; and (iv) a Wittig reaction with  $Ph_3PCH_3I$  and NaHMDS. It was confirmed by  $^1H$ -NMR that the deuteration rate did not decrease in these reaction steps [30].



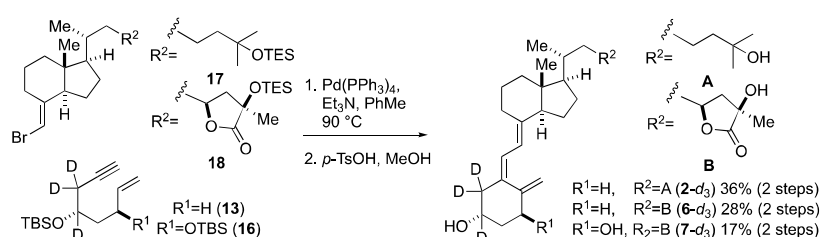
**Scheme 1.** Synthesis of deuterium-labeled enyne **13-d<sub>3</sub>**.

Next, the deuterium-labeled enyne **16** bearing a hydroxyl group at C1 $\alpha$  was synthesized (Scheme 2). The alcohol moiety in **12** was oxidized with an IBX and the resulting aldehyde was reacted with a HWE Wittig reagent to give an unsaturated ester, whose ester group was reduced with a DIBAL-H to give allyl alcohol **14** in a 70% yield from **12** [31,32]. The allyl alcohol **14** was subjected to a Sharpless asymmetric epoxidation with a TBHP in the presence of Ti(O*i*Pr)<sub>4</sub> and L-(+)-DET [33], and the resulting epoxy alcohol was subjected to iodination with iodine and triphenylphosphine followed by treatment with zinc to give a secondary alcohol **15** in a 79% yield (3 steps) [31,32]. The diastereomer ratio at C1 in **15** was 10:1, and the undesired C1 $\beta$  diastereomer was removed by kinetic resolution, using acylation with isopropyl acid anhydride in the presence of (2*S*,3*R*)-HyperBTM [34], to give (-)-**15** in an 82% yield as a single diastereomer. The undesired diastereomer at C3 was also removed via silica gel column purification. The deuterium-labeled enyne **16**, in which the secondary alcohol was protected as the TBS ether, was obtained in an 82% yield.



**Scheme 2.** Synthesis of deuterium-labeled enyne **16-d<sub>3</sub>**.

The palladium-catalyzed coupling reaction of **13-d<sub>3</sub>** and bromoolefin **17** followed by deprotection of the silyl groups provided 25(OH)D<sub>3</sub>-d<sub>3</sub> (**2-d<sub>3</sub>**) in a 36% yield [35]. Next, 25(OH)D<sub>3</sub>-23,26-lactone-d<sub>3</sub> (**6-d<sub>3</sub>**) and 1,25(OH)<sub>2</sub>D<sub>3</sub>-23,26-lactone-d<sub>3</sub> (**7-d<sub>3</sub>**) were similarly synthesized by reacting bromoolefin **18** and enynes **13-d<sub>3</sub>** and **16-d<sub>3</sub>**, respectively [36]. In the synthesis of **2-d<sub>3</sub>** and **6-d<sub>3</sub>**, the undesired diastereomers at C3 $\alpha$  were separated by an HPLC (Scheme 3).



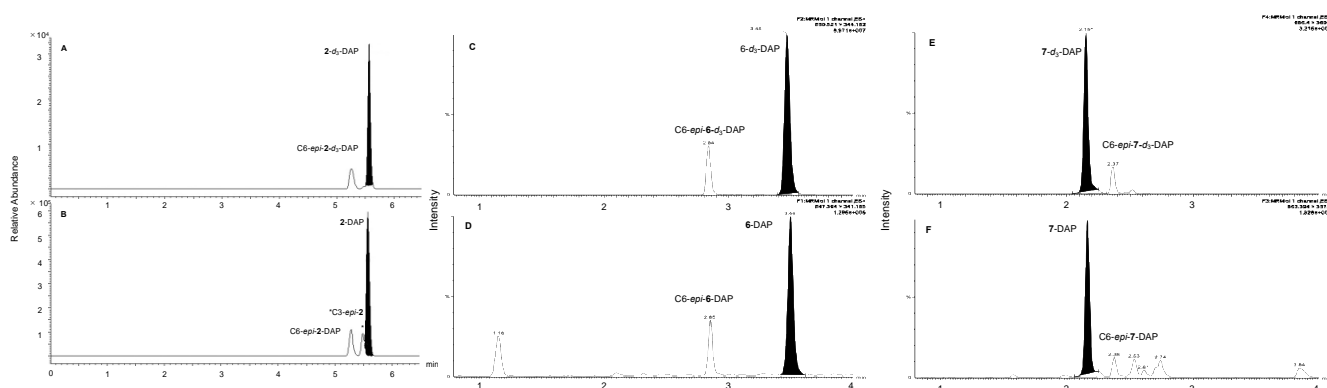
**Scheme 3.** Synthesis of vitamin D<sub>3</sub> metabolites- $d_3$  (2- $d_3$ , 6- $d_3$ , 7- $d_3$ ).

### 2.1. Derivatization of 2, 6, 7 for LC-MS/MS, and Preparation of Calibration Curves

With the deuterium-labeled D<sub>3</sub> metabolites of 2- $d_3$ , 6- $d_3$ , and 7- $d_3$  in hand, we next examined the quantitative analysis of the three D<sub>3</sub> metabolites in pooled human serum by LC-MS/MS. First, we confirmed that our deuterium-labeled D<sub>3</sub> metabolites were suitable as the internal standards for the isotope dilution method in an LC-MS/MS analysis. As described above, D<sub>3</sub> and its metabolites have low ionization efficiency in an LC-MS/MS, and derivatization is necessary to improve the ionization efficiency. Thus, the D<sub>3</sub> metabolites 2, 6, and 7, as well as 2- $d_3$ , 6- $d_3$ , and 7- $d_3$ , were derivatized with a recently developed reagent DAP-PA (4-(4'-dimethylaminophenyl)-1,2,4-triazoline-3,5-dione-phenyl anthracene) [20], and the ion peaks of the DAP adducts were detected by selective reaction monitoring (SRM) under the LC-MS/MS conditions shown in Table 1 (Figure 3).

**Table 1.** Parameters for the LC/MS/MS analysis.

Compound	SRM Transition (m/z)	Cone Voltage (kv)	CE (eV)
25(OH)D <sub>3</sub> -DAP (2-DAP)	619.4 > 341.2	48	28
25(OH)D <sub>3</sub> -23,26-lactone-DAP (6-DAP)	[M + H] <sup>+</sup> [A] <sup>+</sup> 647.4 > 341.2	48	28
1,25(OH) <sub>2</sub> D <sub>3</sub> -23,26-lactone-DAP (7-DAP)	663.4 > 357.2	48	28
25(OH)D <sub>3</sub> - $d_3$ -DAP (2- $d_3$ -DAP)	622.4 > 344.2	48	28
25(OH)D <sub>3</sub> -23,26-lactone- $d_3$ -DAP (6- $d_3$ -DAP)	650.4 > 344.2	48	28
1,25(OH) <sub>2</sub> D <sub>3</sub> -23,26-lactone- $d_3$ -DAP (7- $d_3$ -DAP)	666.4 > 360.2	48	28

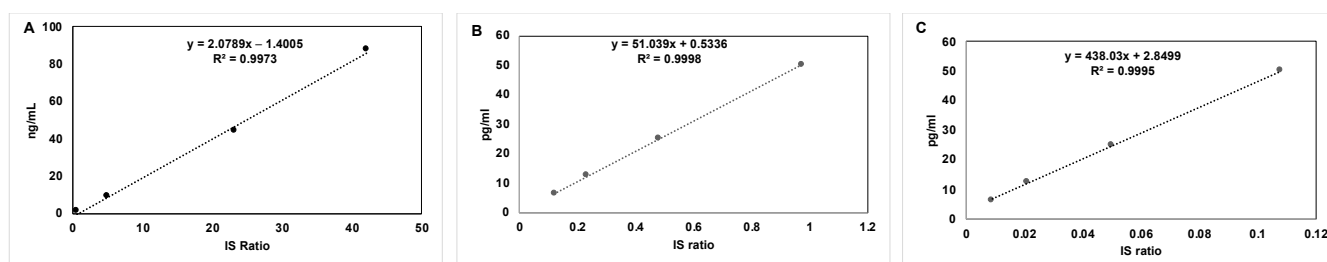


**Figure 3.** SRM chromatograms for the DAP adducts of D<sub>3</sub> metabolites 2, 6, and 7, as well as 2- $d_3$ , 6- $d_3$ , and 7- $d_3$ .

In the case of the DAP-adducts of 2 and 2- $d_3$  (Figure 3A,B), we observed identical ion peaks at the retention time of 5.60 min (abbreviated as  $t_R$ : 5.60 min). Similarly, 6 and 6- $d_3$  showed the same  $t_R$  of 3.50 min, and 7 and 7- $d_3$  showed the same  $t_R$  of 2.15 min, indicating that the deuterium-labeled compounds are suitable as internal standards for the

isotope dilution method. We also observed small peaks at the retention times of 5.20 min (Figure 3A,B), 2.65 min (Figure 3C,D), and 2.37 min (Figure 3E,F) for 2/2- $d_3$ , 6/6- $d_3$ , and 7/7- $d_3$ , respectively. These peaks are due to the epimers at C6 of the DAP adducts, because DAP-PA reacts from both the  $\alpha$ - and  $\beta$ -faces.

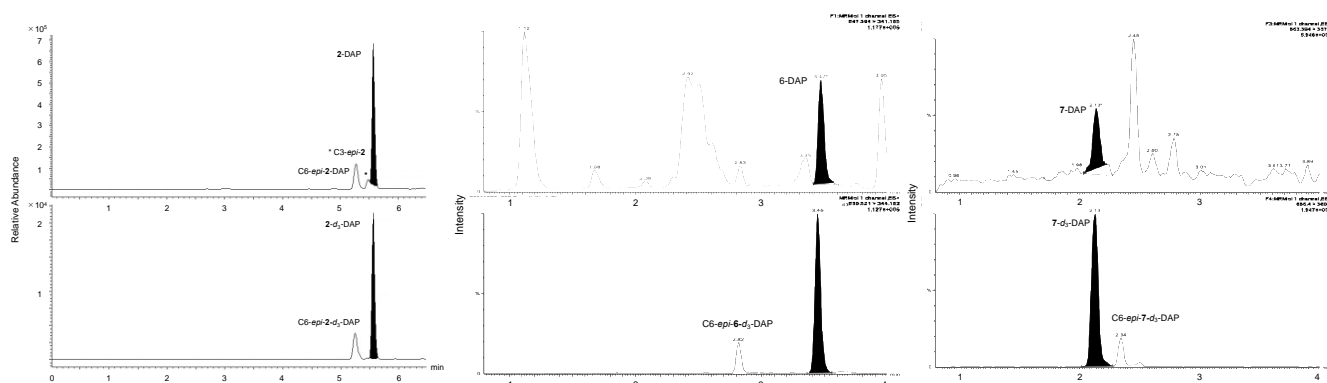
Next, the calibration curves were prepared as follows (Figure 4). A total of 100  $\mu\text{L}$  of each one of the calibrator solutions was mixed with 200  $\mu\text{L}$  of the internal standard solution and evaporated to dryness. After derivatization with DAP-PA, an LC-MS/MS analysis of the unlabeled and labeled DAP-adducts was performed, and the calibration curves were prepared by plotting the concentration of unlabeled DAP-adduct against the ion peak area ratio of unlabeled versus labeled DAP-adduct. All of the calibration curves showed good linearity.



**Figure 4.** Calibration curve of  $D_3$  metabolites; (A) 25(OH) $D_3$  (2); (B) 25(OH) $D_3$ -23,26-lactone (6); and (C) 1,25(OH) $_2D_3$ -23,26-lactone (7).

## 2.2. Quantification of the $D_3$ Derivatives in Human Serum

The levels of 2, 6, and 7 in pooled human serum were quantified by the LC-MS/MS using the isotope dilution method with the constructed calibration curves. The serum was pretreated as follows. An aliquot of serum (100  $\mu\text{L}$ ) was mixed with the internal standards solution (200  $\mu\text{L}$ ). Each sample was loaded onto a supported liquid extraction column (ISOLUTE SLE+ 300  $\mu\text{L}$  sample Volume, Biotage, Uppsala, Sweden) and eluted three times with 600 mL hexane/ethyl acetate (1/1,  $v/v$ ) using a PRESSURE+48 positive pressure manifold (Biotage, Uppsala, Sweden). The combined eluates were evaporated to dryness in a centrifugal evaporator. The ion peaks of the metabolites matched well with those of the corresponding internal standards in the pretreated samples (Figure 5).



**Figure 5.** SRM chromatograms for  $D_3$  derivatives 2, 6, and 7 in pooled human serum.

The concentrations of 2, 6, and 7 in human serum were calculated to be 5.1 ng/mL, 38.3 pg/mL, and 8.9 pg/mL, respectively, based on the area ratios of the detected peaks. The concentrations of 2 and 6 were in agreement with previously reported values [37], while this is the first quantification of 1 $\alpha$ -lactone 7 in human serum.



### 3. Conclusions

We synthesized deuterium-labeled A-ring- $d_3$  synthons **13** and **16** and utilized them for the convergent synthesis of deuterium-labeled  $D_3$  derivatives 25(OH) $D_3$  (**2**), 25(OH) $D_3$ -23, 26-lactone (**6**), and 1,25(OH) $_2D_3$ -23, 26-lactone (**7**). These deuterium-labeled  $D_3$  metabolites were successfully applied as internal standards for the quantification of the metabolites in pooled human serum by LC-MS/MS using the isotope dilution method. This is the first quantification of 1,25(OH) $_2D_3$ -23, 26-lactone (**7**) in human serum.

**Supplementary Materials:** The following supporting information can be downloaded at: <https://www.mdpi.com/article/10.3390/molecules27082427/s1>, Experimental procedures for synthesis and characterization of compounds, Experimental procedure for LC-MS/MS analysis using the isotope dilution method,  $^1H$  and  $^{13}C$  NMR spectra.

**Author Contributions:** Conceptualization, A.N., S.F. and K.N.; investigation, A.N., K.I., R.S., Y.M. and M.I.; analysis, M.T. (Masaki Takiwakiand), Y.K. and S.F.; writing—original draft preparation, A.N. and K.N.; resources, M.O., M.T. (Masayuki Tera) and K.N.; funding acquisition, K.N. All authors have read and agreed to the published version of the manuscript.

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**Conflicts of Interest:** The authors declare no conflict of interest.

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