

Subscriber access provided by UNIV OF PITTSBURGH

Article

Synthesis of Dibenzosultams by "Transition-Metal Free" Photoinduced Intramolecular Arylation of N-aryl-2-halobenzenesulfonamides

Walter Damian Guerra, Roberto Arturo Rossi, Adriana Beatriz Pierini, and Silvia Maricel Barolo J. Org. Chem., Just Accepted Manuscript • DOI: 10.1021/acs.joc.6b00330 • Publication Date (Web): 11 May 2016 Downloaded from http://pubs.acs.org on May 11, 2016

Just Accepted

"Just Accepted" manuscripts have been peer-reviewed and accepted for publication. They are posted online prior to technical editing, formatting for publication and author proofing. The American Chemical Society provides "Just Accepted" as a free service to the research community to expedite the dissemination of scientific material as soon as possible after acceptance. "Just Accepted" manuscripts appear in full in PDF format accompanied by an HTML abstract. "Just Accepted" manuscripts have been fully peer reviewed, but should not be considered the official version of record. They are accessible to all readers and citable by the Digital Object Identifier (DOI®). "Just Accepted" is an optional service offered to authors. Therefore, the "Just Accepted" Web site may not include all articles that will be published in the journal. After a manuscript is technically edited and formatted, it will be removed from the "Just Accepted" Web site and published as an ASAP article. Note that technical editing may introduce minor changes to the manuscript text and/or graphics which could affect content, and all legal disclaimers and ethical guidelines that apply to the journal pertain. ACS cannot be held responsible for errors or consequences arising from the use of information contained in these "Just Accepted" manuscripts.



Synthesis of Dibenzosultams by "Transition-Metal Free" Photoinduced Intramolecular Arylation of N-aryl-2-halobenzenesulfonamides

Walter D. Guerra, Roberto A. Rossi, Adriana B. Pierini* and Silvia M. Barolo*

INFIQC, Departamento de Química Orgánica, Facultad de Ciencias Químicas, Universidad Nacional de Córdoba, Ciudad Universitaria, 5000 Córdoba, Argentina.

RECEIVED DATE (to be automatically inserted after your manuscript is accepted if required according to the journal that you are submitting your paper to)

CORRESPONDING AUTHOR FOOTNOTE (Phone number: (+54) 351-5353867, e-mail: adriana@fcq.unc.edu.ar and sbarolo@fcq.unc.edu.ar

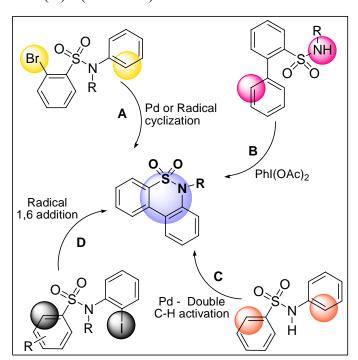
TITLE RUNNING HEAD: Synthesis of Dibenzosultams by Photoinduced Arylation

ABSTRACT: A new and general synthetic route to prepare dibenzosultams is here reported. This approach involves the synthesis of N-aryl-2-halobenzenesulfonamides (3) followed by intramolecular C-C photoinduced arylation under soft conditions without the use of "Transition-Metal". The photostimulated reactions exhibit very good tolerance to different substituent groups with good to excellent isolated yields (42 – 98%) of products. Moreover, it is shown that LED (λ =395 nm) is an efficient light energy source to initiate efficiently the reactions. Theoretical inspection about the mechanism was made to probe the involvement of the radical-anion S_{RN} 1 process.

Introduction

Sulfonamides and their derivatives are known as "sulfa drugs" and are widely used in medicine.¹ This group is an important organic structure within the drug discovery field. Compounds with benzothiazine dioxide or benzosultam core exhibit versatile inhibitory properties against a diverse array of enzymes such as COX-2,² HIV integrase³ or Calpain-1.⁴ Also, benzothiazine dioxide derivatives have been found to play an active role in nuclear factor-κappaB (NFκB) down regulation.⁵

Due to the remarkable importance of benzosultams across medicinal chemistry many synthetic approaches have been developed. ⁶ For dibenzosultams the synthetic approaches are less known and are represented in Scheme 1. These protocols include intramolecular radical cyclization⁷ or palladium-catalyzed intramolecular arylation⁸ of 2-bromo-*N*-alkyl-*N*-arylbenzenesulfonamides ($\bf A$), intramolecular oxidative amination of 2-arylbenzenesulfonamides under "Transition-Metal Free" condition ($\bf B$), ⁹ double C(sp₂)-H palladium-catalyzed intramolecular oxidative coupling of *N*-arylbenzenesulfonamides ($\bf C$)¹⁰ or 1,6 radical addition of *N*-(2-iodophenyl) -*N*-methyl-benzenesulfonamide ($\bf D$)¹¹ (Scheme 1).



Scheme 1. Strategies to synthesize dibenzosultams

Radical nucleophilic substitution involving electron-transfer (ET) steps $(S_{RN}1)^{12}$ is a cyclic process with radicals and radical anions as intermediates. In the $S_{RN}1$ reactions carbanions and anions derived from heteroatoms can be used as nucleophiles to form new C-C or C-heteroatom bonds. This mechanism has proven to be an important synthetic strategy in heterocycle chemistry.

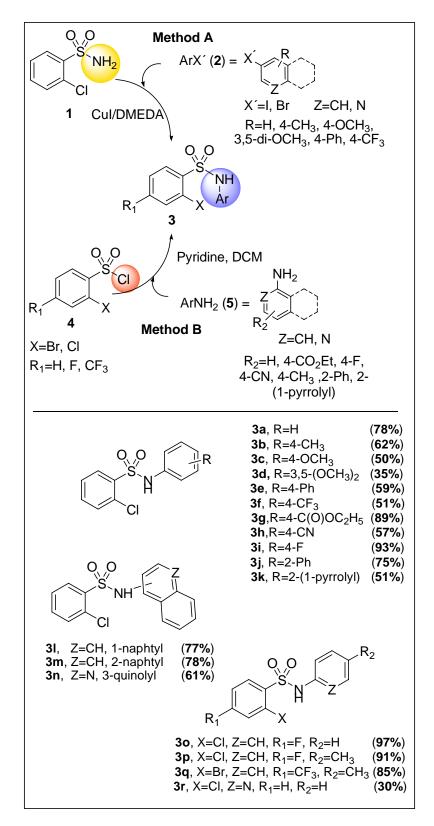
The intramolecular $S_{RN}1$ has been successfully developed to obtain different annulated systems bearing between 5 to 9 members with broad substitution tolerance.¹³ In this context, the reaction of 2-iodobenzensulfonamide with aliphatic ketone enolates to afford 3-subtituted benzothiazines has been reported.¹⁴ However, the reactivity of *N*-aryl-2-halobenzenesulfonamides under $S_{RN}1$ conditions to obtain dibenzosultamas has not been studied yet.

It is important to notice that despite useful synthetic protocols have been investigated to prepare dibenzosultams, still several limitations remain like the use of "Transition-Metal" (Pd), activated substrates like iodide or bromide aromatic precursors, harsh conditions or time-consuming reactions which need to be overcome.

In this context, we developed a novel protocol to synthesize dibenzosultams by intramolecular C-C photoinduced arylation of 2-halo-arylsulfonamides. This protocol involves chloride precursors easily prepared from commercial sources. Besides, the heterocycle could be obtained with a free N-H group that is easily functionalized. Furthermore, the utility of this method is fully demonstrated by exploring the scope toward a broad family of dibenzosultams and greener methodologies. This is a new contribution within our efforts devoted to the synthesis of heterocyclic compounds by "Transition-Metal-free" intramolecular photoinduced arylation reactions. ^{13a}

Results and Discussion

Initially, we investigated different alternative to obtain *N*-aryl-2-halobenzenesulfonamides **3** (Scheme 2). The first strategy involves copper-catalyzed *N*-arylation of 2-chlorobenzenesulfonamide (**1**) with aryl halides **2**¹⁵ (Method A). Under our optimized conditions, ¹⁶ the reaction of **1** with iodobenzene (**2a**) gave the coupling product *N*-phenyl-2-chlorobenzenesulfonamide (**3a**, Ar=Ph, X=Cl) in 78% isolated yield. This protocol was extended to different aryl halides obtaining good to very good isolated yields (50-84%) of the corresponding *N*-aryl-2-chloroarylsulfonamides (**3b-f, m-n** X=Cl) (Scheme 2).



Scheme 2. Synthesis of *N*-aryl-2-haloarylsulfonamides

The other strategy to afford the corresponding sulfonamides 3 involves a known reaction between substituted benzenesulfonyl chlorides 4 and different arylamines 5 with pyridine in DCM at room temperature (Method B).¹⁷ With this methodology we synthesized sulfonamides 3g-l, o-r

with a range of isolated yields from moderate to excellent depending on the corresponding amine (30-97%) (Scheme 2). ¹⁸

N-phenyl-2-chlorobenzenesulfonamide (**3a**) was chosen as model substrate to attempt the intramolecular arylation using *t*-BuOK in DMSO. In this basic medium the initially neutral **3a** (λ_{max} =279 and 281 nm) undergoes an acid-base reaction to give the corresponding anion **3a** (continuum absorption spectrum until 370 nm, see SI).¹⁹

After 3 hours of irradiation (HPI-T metal iodide lamps (400 W)), the reaction of **3a** in DMSO with 2 equiv of t-BuOK afforded the desired product, 6H-dibenzo[c,e][1,2]thiazine 5,5-dioxide (**6a**) in 27% yield, together with 45% of the starting material (entry 1, Table 1). Encouraged by this result, various combinations of irradiation times and equivalents of base were screened (entries 2-5) finding that 3 equivalents of t-BuOK and 3 hours of irradiation gave the best result (97% (86% isolated yield)).²⁰

The other solvents tested were NH_{3(liq)}, CH₃CN and THF. Similar results were obtained in liquid ammonia (-33 °C) (entry 6, Table 1), in CH₃CN the reaction gave a very good yield (72 %) and THF proved to be ineffective for the reaction (entries 7-8, respectively, Table 1).

Table 1. Optimization of the Reaction Conditions ^a

	Base, Solvent, hv, t		S NH
Conditions		6a Yields (%) ^b	
Entry	Solvent/ Base (Equiv)/ hv, t (h)	3a ^c	6a
1	DMSO/ t-BuOK (2)/ hv, 3h	45	27
2	DMSO/ <i>t</i> -BuOK (2)/ hυ, 4h	27	53
3	DMSO/ <i>t</i> -BuOK (2)/ hυ, 6h	23	57
4	DMSO/ <i>t</i> -BuOK (3)/ hv, 2h	13	75
5	DMSO/ t-BuOK (3)/ hυ, 3h	0	97 (86)

6 ^d	NH ₃ / <i>t</i> -BuOK (3)/ hv, 3h	4	95 (84)
7^e	THF/ <i>t</i> -BuOK (3)/ hv, 3 h	75	15
8 ^e	CH ₃ CN/ <i>t</i> -BuOK (3)/ hv, 3 h	18	72
9	DMSO/ NaH (3)/ hv, 3 h	0	75
10	DMSO/ <i>t</i> -BuOK (3)/ hυ, 1.5 h	21	64
11 ^f	DMSO/ <i>t</i> -BuOK (3)/ hυ, 1.5 h	0	92
12	DMSO/ t-BuOK (3)/ dark, 3h	91	0
13 ^g	DMSO/ <i>t</i> -BuOK (3)/ hυ, 1.5 h	97	0
14 ^h	DMSO/ <i>t</i> -BuOK (3)/ hυ, 1.5 h	79	13

^aThe reactions were run in 5 mL of solvent with 0.03 M of **3a** and 0.09M of *t*-BuOK and irradiated for the specific time. Irradiation was conducted in a photochemical reactor equipped with two Philips metal iodide HPI-T 400 W lamps of (air and water refrigerated). ^bYields were determined by CG (internal standard method). Isolated yields are given in parentheses. ^cSubstrate **3a** recovered. ^dThe reaction of liquid ammonia (NH_{3(liq)}) was run in 200 ml (-33°C), with 0.75 mM of **3a** and 2.25 mM of *t*-BuOK. ^eThe reactions in THF or CH₃CN were run in 5 mL. ^f1 equiv of pinacolone was added with respect to the substrate. ^g0.3 equiv of *m*-DNB was added with respect to the substrate. ^h1 equiv of TEMPO was added with respect to the substrate.

When the reaction was carried out without t-BuOK (NaH as base) it proceeded completely which indicates that anion $3a^{-}$ could initiate the reaction (entry 9, Table 1). Also, the reaction time could be shortened under entrainment conditions such as in the presence of pinacolone enolate ions (entry 11 versus entry 10, Table 1).

Mechanistically it is important to notice that there was no reaction under dark conditions (entry 12, Table 1), the photostimulated reaction was completely inhibited by m-dinitrobenzene (m-DNB) (entry 13 versus entry 10, Table 1)^{12c} and equimolecular quantities of TEMPO could partially inhibit the reaction (entry 14 versus entry 10, Table 1). These results exclude a benzyne

and other polar mechanisms and evidence an Electron Transfer (ET) process with formation of radicals. We propose a radical-anion type mechanism ($S_{RN}1$) to be in play.

To extend the scope of the cyclization reaction, sulfonamides previously synthesized were submitted to the t-BuOK/DMSO/hv system, the results being shown in Scheme 3. Modifying the R-substituent in the NH-phenyl moiety of the sulfonamides, with EWG or EDG, **3b-k**, led to full conversion of the substrate after 5h of irradiation providing products **6b-k** in good to excellent isolated yields (54 – 98%). This reveals broad substitution tolerance (CH₃, OCH₃, di-OCH₃, CF₃, F, CN, C(O)OC₂H₅, Ph) and low steric hindrance (tolerate o-substitution, **6j-k**).

The strategy was extended successfully to obtain fused dibensultams after longer irradiation times. 5H-Benzo[e]naphtho[1,2-c][1,2]thiazine 6,6-dioxide (6n) and 6H-benzo[5,6][1,2]thiazino-[3,4-c]quinoline 5,5-dioxide (6n) were obtained after 6n0 with 97% and 69% yields, respectively. 6H-Benzo[e]naphtho[2,1-c][1,2]thiazine 5,5-dioxide (6n0) was obtained in good yield (51%0) after 8n0 h of irradiation without full conversion of the substrate.

The effect of modifying the R-substituent in the sulfonyl-phenyl system was also examined. Substrates with a EWG like F or CF_3 (30-q) were prepared. In these cases the reaction underwent partially without full conversion after 8h to afford the corresponding dibenzosultams 60-q with moderate yields (~45%).

Finally, we attempted the reaction of 2-chloro-*N*-(pyridin-2-yl)benzenesulfonamide (**3r**) as substrate, with a pyridine system, but it was unreactive under our experimental conditions.

A=Conventional "work-up" and purification. B= Direct filtration and recrystallization.

^aReaction time=6h. ^bReaction time= 8h. ^cAcid extraction as "work-up" without further purification.

Scheme 3. Scope of the Intramolecular Photoinduced Arylation to synthesize 6*H*-dibenzothiazines 5,5-dioxide 6.

In order to accomplish the goal of sustainable chemistry we changed the conventional "workup" and purification processes to direct filtration and recrystallization which leaded to a reduction in wastes and in the use of solvents. With this methodology we synthesized dibenzosultams **6c,e,h,l**, with a range of isolated yields from moderate to excellent (Scheme 3).

To extend our study, the irradiation source was changed to LED light. Chemical transformations via visible-LED light is one of the emerging strategies to achieve the increasing demand for more sustainable chemical processes due to their ultra-efficient lighting and low-cost.²¹

The results under visible LED light are summarized in Table 2. Employing a Violet LED light (λmax=395 nm) the reaction undergo full conversion obtaining 82% yields of **6a** (entry 1, Table 2).²² For other studied substrates **3b,i,j,l,m** (entries 2-6, Table 2) the yields obtained were comparable to the classical irradiation source under the same irradiation time. It is interesting to notice that employing this source of irradiation, substrates like **3m,o,p** achieved full conversion with very good yields of dibenzosultams **6m,o,p** (76-94 %), showing the efficiency of LED lights (compare Scheme 3 with entries 6-8, Table 2).

Table 2. Use of Visible-LED lights in the Intramolecular photoinduced arylation^a

$$R_1$$
 R_2
 R_2
 R_2
 R_2
 R_2
 R_2
 R_2
 R_2
 R_3
 R_4
 R_4
 R_5
 R_5
 R_5
 R_5
 R_5
 R_6
 R_7

	y 3 (%) ^b	t (h)	Products	
Entry			6	Yields (%) ^c
1	3a ()	3h	6a	82
2	3b ()	5h	6b	84
3	3i (23)	5h	6i	71
4	3j (15)	5h	6 j	80
5	3l (12)	6h	61	84
6	3m (8)	8h	6m	76

7	30 ()	5h	60	94
8	3p (10)	8h	6p	87

^aThe reactions were run in 5 mL of DMSO with **3** (0.03 M) and 0.09M of *t*-BuOK. Irradiation was performed for the specific time with Violet LED (LEDs ($\lambda = 395 \pm 15$ nm), 3 W, 700 mA). ^bSubstrate **3** recovered. ^cYields were determined by CG (internal standard method).

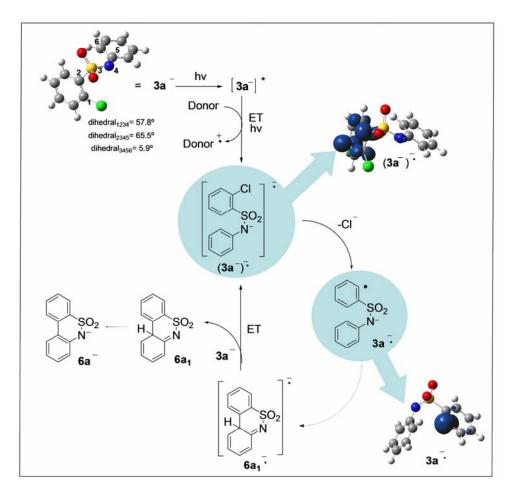
To support our experimental evidences of a radical-anion type mechanism ($S_{RN}1$), we performed a computational study of key mechanistic intermediates and reactive pathways with anion $3a^{-}$ as model. The results are summarized in Scheme 4. The computational data were obtained with M06-2X DFT functional,²³ the 6-311+G* basis set and the PCM continuum solvent model.²⁴

Initially, we propose that anion $3\mathbf{a}$ is formed in presence of excess t-BuO. The most stable conformer of $3\mathbf{a}$, presented in Scheme 4, could form radical dianion $(3\mathbf{a})$ by an ET process. This ET could be achieved from different sources (for example t-BuO). In our case the ET from the base to the excited anion $[3\mathbf{a}]^*$ (S_I state of $3\mathbf{a}$ evaluated with TD-DFT) is exothermic and could be responsible for the initiation pathway.

The (3a') formed bears the unpaired spin density at the π system of the aryl-Cl moiety (Scheme 4). This intermediate could dissociate with a very low activation energy (0.4 kcal/mol) to afford the distonic radical anion 3a* through a transition state characterized by a C-Cl bending transition vector. As known, a σ (C-Cl)- π overlap is required for this π - σ intramolecular dissociative ET to take place.²⁷

After C-Cl fragmentation, the distonic radical anion intermediate $3a^{*}$, represented in Scheme 4, could afford the cyclic radical anion $6a_1^{*}$. In this pathway the C-C coupling is achieved with an activation energy (Ea) of 9.15 kcal/mol. Also, it is important to emphasize that cyclic radical anion $6a_1^{*}$ is 24 kcal/mol more stable than the distonic radical anion $3a^{*}$ and this difference could be the driving force of the reaction.

Finally, $6a_1$ and $(3a^-)^-$ could be formed after an ET from $6a_1^-$ to $3a^-$. The latter ET propagates the reaction cycle. Under the basic media $6a_1$ generate $6a^-$ and upon reaction work up, product 6a was formed.



Scheme 4. Proposed Mechanism. The spin density is shown for (3a) and 3a (isodensity = 0.004).

Conclusions

To conclude, we described an efficient route to synthesize dibenzosultams under "Transition-Metal-Free" conditions starting from *N*-aryl-2-halobenzenesulfonamides easily prepared by two different procedures. The cyclization reactions are promoted by *t*-BuOK and occur in DMSO at room temperature. Many functional groups are tolerated, giving access to a wide range of synthetically relevant heterocycles. Other solvents such as NH₃ and CH₃CN seem promising to perform the reactions. We also explored the use of visible LED light improving some yields of

challenging substrates. Finally, DFT calculations were performed to inspect the energetic and thus confirm our proposal of the radical-anion type $S_{RN}1$ mechanism.

Experimental Section

Computational Procedure. All calculations were performed with the Gaussian09 program. The conformers obtained were refined with complete geometry optimization within the M06-2X DFT functional and the 6-311+ G^* basis set. The geometries thus found were used as starting points for the evaluation of the reaction profiles by using the distinguished reaction coordinate scan. The effect of DMSO as solvent was evaluated through the Tomasi's Polarized Continuum Model (PCM) as implemented in Gaussian09. The inclusion of the solvent in the calculations is a requisite to evaluate valence radical anions. The characterization of stationary points was done by Hessian matrix calculations. The energy informed for TSs, anions and radical anions includes zero-point corrections. The vertical excited singlet stated (S_I) of anion S_I was calculated with TD-DTF the M06-2X functional and the 6-311+ S_I basis set. The energy of S_I was calculated including the PCM contribution under the StateSpecific approach.

General Considerations. Column chromatography was carried out on silica gel. Melting points were determined using a standard melting point instrument and are uncorrected. Gas chromatographic analyses were performed with a flame-ionization detector, on 30 m capillary column of a 0.32 mm x 0.25 μm film thickness, with a 5% phenylpolysiloxane phase. GC-MS analyses were performed employing a 25 m x 0.2 mm x 0.33 μm with a 5% phenylpolysiloxane phase column. 1 H NMR spectra and 13 C NMR spectra were recorded on a 400.16 MHz in CDCl₃, Dimethyl sulfoxide- d_6 (CD₃SOCD₃) or acetone- d_6 (CD₃COCD₃) as solvent with TMS as internal standard. Coupling constants are given in Hz and chemical shifts are reported in δ values in ppm. Data are reported as followed: chemical shift, multiplicity (s = singlet, s br = broad singlet, d = doublet, t = triplet, dd = double doublet, dt = double triplet, ddd = double doublet on the multiplet), coupling constants (Hz), and integration. Copies of 1 H NMR and 13 C NMR spectra are provided.

All new products were further characterized by HRMS. HRMS analyses were carried out using a TOF-MS instrument with an ESI source.

Irradiation was conducted in a reactor equipped with two Philips HPI-T 400 W lamps of metallic iodide (cooled with water) or with LED lights performing at 3W of potency and 700 mV of current.

DMSO as solvent was stored under molecular sieves (4 Å). Anhydrous ethyl ether was stored over Na wire. All solvents were analytical grade. Silica gel (0.063–0.200 mm) was used in column chromatography. Unless otherwise stated, all reagents were purchased from commercial suppliers and used without further purification.

Representative Procedure for synthesis of 2-halo-N-phenylbenzenesulfonamides

Method A: The Cross-coupling copper-catalyzed of 2-chlorobenzenesulfonamide (1) with aryl halides was the procedure to synthesize sulfonamides 3a-f/m-n. A Schlenk tube equipped with a nitrogen inlet and magnetic stirred was charged with 1 (115.0 mg, 0.6 mmol), copper (I) iodide (11.4 mg, 0.06 mmol), aryl halide (iodobenzene, **2a**) (244.4 mg, 1.2 mmol), K₂CO₃ (381.3 mg, 1.8 mmol), acetonitrile (CH₃CN) (4 mL), and N,N-dimethylethane-1,2-diamine (DMEDA) (21.6 mg, 0.3 mmol). The tube was then heated to 90°C for 18 h. The reaction mixture was then cooled to room temperature, 2M HCl (10 mL) was added slowly, followed by EtOAc extraction (15 mL x 3). The organic layers were combined, dried over Na₂SO₄, and filtered. After removal of volatile components from the filtrate, the resulting crude product was purified by column chromatography on silica gel eluting with pentane/EtOAc (100:0 \rightarrow 80:20 %). White solid of 2-Chloro-Nphenylbenzenesulfonamide (3a) was isolated in 78% yield (125.9 mg, 0.468 mmol), m.p. 145-147°C. ¹H NMR (400.16 MHz, CDCl₃) $\delta_{\rm H}$: 8.00 (dd, J = 7.8, 1.4, 1H), 7.49 (dd, J = 8.0, 1.6, 1H), 7.45 (td, J = 7.6, 1.6, 1H), 7.34-7.30 (m, 1H), 7.23-7.19 (m, 2H), 7.13-7.06 (m, 4H). ¹³C NMR $(100.62 \text{ MHz}, \text{CDCl}_3) \delta_C$: 136.1, 135.7, 134.1, 132.0, 131.5, 131.3, 129.3, 127.2, 125.7, 121.6. **GC-MS** (EI) m/z 267 (M⁺, 21), 168 (48), 167 (11), 111 (22), 93 (10), 92 (100), 75 (24), 65 (56), 64 $(8)^{28}$

2-Chloro-N-(p-tolyl)benzenesulfonamide (**3b**). The product was purified by column chromatography on silica gel eluting with pentane/ EtOAc (100:0 \rightarrow 85:15 %). White solid was isolated in 62% yield (104.8 mg, 0.372 mmol), m.p. 151-152 °C. ¹H NMR (400.16 MHz, CDCl₃) $\delta_{\rm H}$: 7.96 (dd, J = 7.8, 1.4, 1H), 7.49 (dd, J = 7.8, 1.4, 1H), 7.44 (td, J = 7.7, 1.6, 1H), 7.30 (td, J = 7.6, 1.2, 1H), 7.02 (br.s, 1H), 7.00 (br.s, 4H), 2.22 (s, 3H); ¹³C NMR (100.62 MHz, CDCl₃) $\delta_{\rm C}$: 136.2, 135.8, 133.9, 132.9, 132.0, 131.5, 131.3, 129.9, 127.2, 122.3, 20.8. **GC-MS** (EI) m/z 281 (M⁺, 9), 111 (15), 107 (8), 106 (100), 79 (30), 78 (12), 77 (35), 75 (18), 51 (9). **HRMS** (TOF, ESI⁺): calcd for C₁₃H₁₂CINNaO₂S (M+Na)⁺: 304.0170; Found: 304.0169.

2-Chloro-N-(4-methoxyphenyl)benzenesulfonamide (3c). The product was purified by column chromatography on silica gel eluting with pentane/ EtOAc (100:0 \rightarrow 85:15 %). Colorless crystal was isolated in 50% yield (89.1 mg, 0.3 mmol), m.p. 134-135 °C. ¹H NMR (400.16 MHz, CDCl₃) $\delta_{\rm H}$: 7.89 (dd, J = 7.8, 1.6, 1H), 7.52 (dd, J = 8.0, 1.2, 1H), 7.45 (td, J = 7.6, 1.6, 1H), 7.29 (td, J =

7.6, 1.2, 1H), 7.06-7.02 (m, 2H), 6.95 (br.s, 1H), 6.74-6.7 (m, 2H), 3.71 (s, 3H). ¹³C NMR (100.62 MHz, CDCl₃) δ_C : 158.1, 136.2, 133.9, 132.0, 131.4, 131.2, 128.1, 127.0, 125.2, 114.4, 55.3. GC-MS (EI) m/z 327 (M⁺, 10), 263 (10), 229 (16), 228 (96), 213 (20), 197 (11), 152 (10), 126 (9), 125 (10), 111 (17). HRMS (TOF, ESI⁺): calcd for C₁₃H₁₂ClNNaO₃S (M+Na)⁺: 320.0119; Found: 320.0106.

2-Chloro-N-(3,5-di-methoxyphenyl)benzenesulfonamide (**3d**). The product was purified by column chromatography on silica gel eluting with pentane/ EtOAc (100:0 \rightarrow 70:30 %). Colorless crystal was isolated in 35% yield (68.7 mg, 0.21 mmol), m.p. 164-165 °C. ¹H NMR (400.16 MHz, CDCl₃) $\delta_{\rm H}$: 8.05 (dd, J = 7.8, 1, 1H), 7.51-7.44 (m, 2H), 7.37-7.33 (m, 1H), 7.01 (s, 1H), 6.29 (d, J = 2.4, 2H), 6.16 (t, J = 2, 1H), 3.69 (s, 6H). ¹³C NMR (100.62 MHz, CDCl₃) $\delta_{\rm C}$: 161.2, 136.1, 134.2, 132.1, 131.6, 131.4, 129.1, 127.2, 99.4, 97.6, 55.4. **GC-MS** (EI) m/z 327 (M⁺, 10), 263 (10), 229 (16), 228 (96), 213 (20), 197 (11), 152 (10), 126 (9), 125 (100), 111 (17). **HRMS** (TOF, ESI⁺): calcd for C₁₄H₁₄CINNaO₄S (M+Na)⁺: 350.0224; Found: 350.0227.

N-([1,1'-Biphenyl]-4-yl)-2-chlorobenzenesulfonamide (**3e**). The product was purified by column chromatography on silica gel eluting with pentane/ EtOAc (100:0 → 80:20 %). White solid was isolated in 59% yield (60.9 mg, 0.177 mmol), m.p. 139-141 °C. ¹**H NMR** (400.16 MHz, CD₃COCD₃) $\delta_{\rm H}$: 9.42 (br.s, 1H), 8.14 (d, *J* = 8.0, 1H), 7.60-7.48 (m, 7H), 7.47-7.28 (m, 5H). ¹³**C NMR** (100.62 MHz, CD₃COCD₃) $\delta_{\rm C}$: 139.8, 137.0, 136.9, 136.4, 134.3, 131.9, 131.7, 131.3, 128.7, 127.4, 127.3, 127.1, 126.4, 120.4. **GC-MS** (EI) *m/z* 343 (M⁺, 16), 169 (14), 168 (100), 166 (4), 141 (24), 139 (5), 115 (17), 111 (4), 75 (4). **HRMS** (TOF, ESI⁺): calcd for C₁₈H₁₄ClNNaO₂S (M+Na)⁺: 366.0326; Found: 366.0329.

2-Chloro-N-(4-(trifluoromethyl)phenyl)benzenesulfonamide (**3f**). The product was purified by column chromatography on silica gel eluting with pentane/ EtOAc (100:0 \rightarrow 80:20 %). White solid was isolated in 51% yield (102.8 mg, 0.306 mmol), m.p. 209-211°C. ¹H NMR (400.16 MHz, CDCl₃) δ_H: 8.10-8.08 (m, 1H), 7.53-7.36 (m, 6H), 7.22 (d, J = 8.4, 2H). ¹³C NMR (100.62 MHz, CDCl₃) δ_C: 139.1, 135.8, 134.5, 132.0, 131.8, 131.4, 127.4, 127.2 (q, J = 33, 1C), 126.7 (q, J = 4.0, 1C), 123.8 (q, J = 273, 1C), 119.9. **GC-MS** (EI) m/z 335 (M⁺, 16), 236 (41), 177 (16), 175 (43), 160 (21), 140 (27), 114 (14), 113 (51), 111 (100), 75 (48), 63 (14), 50 (14). **HRMS** (TOF, ESI⁺): calcd for C₁₃H₉ClF₃NNaO₂S (M+Na)⁺: 357.9887; Found: 357.9896.

2-Chloro-N-(naphthalen-2-yl)benzenesulfonamide (**3m**). The product was purified by column chromatography on silica gel eluting with pentane/ EtOAc (100:0 \rightarrow 70:30 %). White solid was isolated in 78% yield (74.4 mg, 0.234 mmol), m.p. 187-189°C. ¹H NMR (400.16 MHz, CDCl₃) $\delta_{\rm H}$: 8.02 (dd, J = 8.0, 1.6, 1H), 7.73-7.68 (m, 3H), 7.57 (d, J = 2.0, 1H), 7.45-4.36 (m, 3H), 7.30-7.24 (m, 3H). ¹³C NMR (100.62 MHz, CDCl₃) $\delta_{\rm C}$: 136.1, 134.1, 133.5, 133.2, 132.0, 131.5, 131.3,

131.2, 129.4, 127.6, 127.5, 127.2, 126.7, 125.7, 120.9, 118.7. **GC-MS** (EI) m/z 319 (M⁺+2, 5), 317 (M⁺, 13), 218 (25), 143 (7), 142 (64), 140 (5), 116 (10), 115 (100), 89 (6), 75 (6). **HRMS** (TOF, ESI⁺): calcd for $C_{16}H_{12}CINNaO_2S$ (M+Na)⁺: 340.0170; Found: 340.0164.

2-Chloro-N-(quinolin-3-yl)benzenesulfonamide (3n). The product was purified by column chromatography on silica gel eluting with pentane/ EtOAc (100:0 \rightarrow 50:50 %). Light yellow solid was isolated in 61% yield (116.6 mg, 0.366 mmol), m.p. 166-167 °C. ¹H NMR (400.16 MHz, CDCl₃) $\delta_{\rm H}$: 8.68 (d, J = 2.8, 1H), 8.04-8.00 (3H, m), 7.94 (br.s, 1H), 7.73 (d, J = 8.4, 1.2, 1H), 7.64 (ddd, J = 8.3, 7.1, 1.2, 1H), 7.54-7.49 (m, 2H), 7.44 (td, J = 7.8, 1.6, 1H), 7.29 (td, J = 7.6, 1.6, 1H). ¹³C NMR (100.62 MHz, CDCl₃) $\delta_{\rm C}$: 145.9, 145.3, 135.9, 134.5, 132.0, 131.8, 131.3, 129.5, 129.2, 129.1, 127.8, 127.6, 127.5, 127.4, 126.7. **GC-MS** (EI) m/z 320 (M⁺+2, 10), 318 (M⁺, 25), 144 (11), 143 (100), 116 (94), 115 (10), 111 (20), 89 (53), 75 (22), 63 (18). **HRMS** (TOF, ESI⁺): calcd for $C_{15}H_{12}CIN_2O_2S$ (M+H)⁺: 319.0302; Found: 319.0303.

Method B: Sulfonamides **3g-l/o-r** were synthesized by sulfonylation of the corresponding aniline. The aniline (0.72 mmol) was dissolved in dry CH₂Cl₂ (1.2 mL), and the solution was treated with the corresponding 2-halobenzene-1-sulfonyl chloride (0.6 mmol) and pyridine (0.142 g, 1.8 mmol). The mixture was stirred at room temperature for 18 h, diluted with H₂O (15 mL) and extracted with CH₂Cl₂ (3 x 25 mL). The combined organic layers were washed with 1M HCl, brine, dried over Na₂SO₄, and concentrated in vacuum. After removal of volatile components from the filtrate, the resulting crude product was purified by column chromatography on silica gel. Ethyl 4-(2chlorophenylsulfonamido)benzoate (3g) was synthesized from 2-chlorobenzene-1-sulfonyl chloride (4a) and ethyl 4-aminobenzoate and purified eluting with pentane/ EtOAc (100:0 \rightarrow 70:30 %). Light yellow solid was isolated in 89% yield (181.4 mg, 0.534 mmol), m.p. 191-193 °C. ¹H NMR $(400.16 \text{ MHz}, \text{CD}_3\text{SOCD}_3) \delta_H$: 11.22 (br.s, 1H), 8.13 (d, J = 7.6, 1H), 7.81 (d, J = 8.8, 2H), 7.67-7.63 (m, 2H), 7.57-7.53 (m, 1H), 7.20 (d, J = 8.4, 2H), 4.25-4.20 (m, 2H), 1.27-1.23 (m, 3H). ¹³C **NMR** (100.62 MHz, CD₃SOCD₃) δ_C : 165.1, 141.6, 136.0, 135.0, 132.0, 131.7, 130.7, 130.5, 127.8, 124.6, 117.6, 60.5, 14.1. **GC-MS** (EI) m/z 339 (M⁺, 60), 294 (41), 240 (30), 212 (21), 175 (21), 168 (25), 164 (67), 136 (25), 119 (48), 113 (22), 111 (66), 108 (100), 92 (55), 91 (47), 90 (18), 65 (22), 64 (33), 63 (26). **HRMS** (TOF, ESI⁺): calcd for $C_{15}H_{14}CINNaO_4S$ (M+Na)⁺: 362.0224; Found: 362.0228.

2-Chloro-N-(4-cyanophenyl)benzenesulfonamide (**3h**). The product was purified by column chromatography on silica gel eluting with pentane/ EtOAc (100:0 \rightarrow 70:30 %). Light yellow solid was isolated in 57% yield (100.2 mg, 0.342 mmol), m.p. 199-201 °C. ¹H NMR (400.16 MHz, CD₃SOCD₃) δ_H: 11.41 (br.s, 1H), 8.16-8.14 (m, 1H), 7.70-7.55 (m, 5H), 7.21 (d, J = 8.8, 2H). ¹³C NMR (100.62 MHz, CD₃SOCD₃) δ_C: 141.6, 135.9, 135.2, 133.7, 132.1, 131.7, 130.7, 128.0, 118.6,

117.9, 105.3. **GC-MS** (EI) m/z 294 (M⁺ +1, 14), 292 (M⁺, 33), 277 (19), 193 (22), 177 (27), 175 (69), 117 (17), 113 (32), 111 (100), 90 (29), 75 (40), 64 (12), 63 (13). **HRMS** (TOF, ESI⁺): calcd for $C_{13}H_9ClN_2NaO_2S$ (M+Na)⁺: 314.9966; Found: 314.9966.

2-Chloro-N-(4-fluorophenyl)benzenesulfonamide (3i). The product was purified by column chromatography on silica gel eluting with pentane/ EtOAc (100:0 \rightarrow 80:20 %). White solid was isolated in 93% yield (159.4 mg, 0.558 mmol), m.p. 214-215 °C. ¹H NMR (400.16 MHz, CD₃SOCD₃) $\delta_{\rm H}$: 10.59 (br.s, 1H), 8.00-7.98 (m, 1H), 7.64-7.47 (m, 3H), 7.13-7.04 (m, 4H). ¹³C NMR (100.62 MHz, CD₃SOCD₃) $\delta_{\rm C}$: 158.9 (d, J = 240, 1C), 136.3, 134.7, 133.2 (d, J = 3, 1C), 131.8, 131.6, 130.7, 127.7, 121.9 (d, J = 8, 1C), 115.9 (d, J = 23, 1C). **GC-MS** (EI) m/z 287 (M⁺+2, 7), 285 (M⁺, 18), 186 (9), 111 (15), 110 (100), 83 (34), 75 (10), 57 (7). **HRMS** (TOF, ESI⁺): calcd for C₁₂H₉CIFNNaO₂S (M+Na)⁺: 307.9909; Found: 307.9919.

N-([1,1'-Biphenyl]-2-yl)-2-chlorobenzenesulfonamide (**3j**). The product was purified by column chromatography on silica gel eluting with pentane/ EtOAc (100:0 → 80:20 %). White solid was isolated in 75% yield (154.8 mg, 0.45 mmol), m.p. 159-161 °C. ¹**H NMR** (400.16 MHz, CD₃SOCD₃) $\delta_{\rm H}$: 9.72 (br.s, 1H), 7.67 (dd, J = 7.9, 1.5, 1H), 7.59-7.54 (m, 1H), 7.51 (dd, J = 8.0, 1.3, 1H), 7.41-7.23 (m, 9H), 7.01 (dd, J = 7.6, 1.3, 1H). ¹³**C NMR** (100.62 MHz, CD₃SOCD₃) $\delta_{\rm C}$: 139.7, 138.4, 138.2, 133.9, 132.8, 131.8, 130.9, 130.7, 130.2, 129.0, 128.2, 128.0, 128.0, 127.4, 127.4, 127.0. **GC-MS** (EI) m/z 343 (M⁺, 8), 169 (12), 168 (100), 167 (72), 140 (5), 139 (8), 115 (5), 111 (4), 75 (4). **HRMS** (TOF, ESI⁺): calcd for C₁₈H₁₄CINNaO₂S (M+Na)⁺: 366.0326; Found: 366.0322.

N-(2-(1H-Pyrrol-1-yl)phenyl)-2-chlorobenzenesulfonamide (**3k**). The product was purified by column chromatography on silica gel eluting with pentane/ EtOAc (100:0 → 90:10 %). Amber crystal was isolated in 51% yield (100.2 mg, 0.306 mmol), m.p. 122-124 °C. ¹H NMR (400.16 MHz, CDCl₃) δ_H: 8.10 (dd, J = 8.0, 1.6, 1H), 7.56 (dd, J = 8.4, 1.2, 1H), 7.48 (ddd, J = 8.0, 7.2, 1.6, 1H), 7.43 (dd, J = 7.8, 1.4, 1H), 7.24 (1H, ddd, J = 8.3, 7.3, 1.6, 1H), 7.20-7.17 (m, 2H), 7.08 (td, J = 7.6, 1.2, 1H), 6.66 (t, J = 2.2, 2H), 6.37 (t, J = 2.2, 2H). ¹³C NMR (100.62 MHz, CDCl₃) δ_C: 136.2, 134.2, 132.2, 131.1, 131.8, 131.8, 131.1, 128.7, 127.7, 127.0, 124.5, 121.9, 119.0, 110.9. **HRMS** (TOF, ESI⁺): calcd for C₁₆H₁₃ClN₂NaO₂S (M+Na)⁺: 355.0278; Found: 355.0266.

2-Chloro-N-(naphthalen-1-yl)benzenesulfonamide (31). The product was purified by column chromatography on silica gel eluting with pentane/ EtOAc (100:0 \rightarrow 80:20 %). Brown crystal was isolated in 77% yield (146.8 mg, 0.462 mmol), m.p. 166-167 °C. ¹H NMR (400.16 MHz, CDCl₃) $\delta_{\rm H}$: 8.16 (d, J = 8.4, 1H), 7.91 (dd, J = 8.0, 1.6, 1H), 7.81 (d, J = 7.6, 1H), 7.69 (d, J = 8.4, 1H), 7.56-7.44 (m, 4H), 7.30-7.21 (m, 4H). ¹³C NMR (100.62 MHz, CDCl₃) $\delta_{\rm C}$: 136.9, 134.3, 134.0, 131.8, 131.7, 131.5, 131.0, 129.3, 128.3, 127.5, 127.2, 126.8, 126.5, 125.2, 122.0, 121.9. **GC-MS**

(EI) m/z 317 (M⁺, 14), 218 (8), 143 (12), 142 (100), 140 (6), 116 (9), 115 (87), 89 (6), 75 (7). **HRMS** (TOF, ESI⁺): calcd for C₁₆H₁₂ClNNaO₂S (M+Na)⁺: 340.0170; Found: 340.0161.

2-Chloro-4-fluoro-N-phenylbenzenesulfonamide (**30**). The product was purified by column chromatography on silica gel eluting with pentane/ EtOAc (100:0 \rightarrow 80:20 %). White solid was isolated in 97% yield (166.2 mg, 0.582 mmol), m.p. 109-110 °C. ¹H NMR (400.16 MHz, CDCl₃) $\delta_{\rm H}$: 8.01 (dd, J = 8.8, 6.0, 1H), 7.26-7.20 (m, 3H), 7.12-7.09 (m, 3H), 7.06 (br.s, 1H), 7.02 (ddd, J = 8.8, 7.6, 2.4, 1H). ¹³C NMR (100.62 MHz, CDCl₃) $\delta_{\rm C}$: 164.7 (d, J = 257, 1C), 135.5, 134.1 (d, J = 10, 1C), 133.1 (d, J = 11, 1C), 132.4 (d, J = 4, 1C), 129.4, 125.9, 121.6, 119.2 (d, J = 25, 1C), 14.5 (d, J = 21, 1C). **GC-MS** (EI) m/z 287 (M⁺+2, 6), 285 (M⁺, 18), 186 (26), 185 (6), 131 (5), 129 (17), 109 (7), 94 (6), 93 (14), 92 (100), 65 (49), 64 (8), 63 (8). **HRMS** (TOF, ESI⁺): calcd for $C_{12}H_9\text{CIFNNaO}_2\text{S}$ (M+Na)⁺: 307.9919; Found: 307.9919.

2-Chloro-4-fluoro-N-(p-tolyl)benzenesulfonamide (3p). The product was purified by column

chromatography on silica gel eluting with pentane/ EtOAc (100:0 \rightarrow 80:20 %). Light yellow solid was isolated in 91% yield (163.6 mg, 0.546 mmol), m.p. 110-112 °C. ¹H NMR (400.16 MHz, CDCl₃) $\delta_{\rm H}$: 7.97 (dd, J = 9.0, 5.8, 1H), 7.24 (dd, J = 8, 2.4, 1H), 7.03-7.00 (6H, m), 2.24 (s, 3H). ¹³C NMR (100.62 MHz, CDCl₃) $\delta_{\rm C}$: 164.7 (d, J = 257, 1C), 136.1, 134.1 (d, J = 8, 1C), 133.1 (d, J = 11, 1C), 132.8, 132.5, 130.0, 122.3, 119.2 (d, J = 26, 1C), 114.5 (d, J = 22, 1C), 20.8. GC-MS (EI) m/z 301 (M⁺+2, 7), 299 (M⁺, 19), 129 (7), 107 (9), 106 (100), 79 (25), 78 (8), 77 (24), 52 (4), 51 (4). HRMS (TOF, ESI⁺): calcd for C₁₃H₁₁CIFNNaO₂S (M+Na)⁺: 322.0075; Found: 322.0069. 2-Bromo-N-(p-tolyl)-4-(trifluoromethyl)benzenesulfonamide (3q). The product was purified by column chromatography on silica gel eluting with pentane/ EtOAc (100:0 \rightarrow 80:20 %). Colorless solid was isolated in 85% yield (200.1 mg, 0.51 mmol), m.p. 129-130 °C. ¹H NMR (400.16 MHz, CD₃SOCD₃) $\delta_{\rm H}$: 10.70 (br.s, 1H), 8.23 (d, J = 0.8, 1H), 8.19 (d, J = 8.0, 1H), 7.94 (dt, J = 8.2, 0.8, 1H), 7.04 (d, J = 8.4, 2H), 7.00-6.98 (2H, m), 2.16 (s, 3H). ¹³C NMR (100.62 MHz, CD₃SOCD₃) $\delta_{\rm C}$: 142.4, 133.9, 133.7 (q, J = 33, 1C), 133.7, 132.5, 132.2 (q, J = 4, 1C), 129.7, 125.3 (q, J = 4, 1C), 122.4 (q, J = 272, 1C), 120.2, 118.3, 20.2. GC-MS (EI) m/z 395 (M⁺+2, 11), 393 (M⁺, 11), 223 (4), 144 (12), 125 (4), 107 (8), 106 (100), 79 (25), 78 (10), 77 (27), 52 (4). HRMS (TOF,

2-Chloro-N-(pyridin-2-yl)benzenesulfonamide (3r). The product was purified by column chromatography on silica gel eluting with pentane/ EtOAc (100:0 \rightarrow 70:30 %). White solid was isolated in 30% yield (24.2 mg, 0.09 mmol), m.p. 212-213 °C. ¹H NMR (400.16 MHz, CDCl₃) $\delta_{\rm H}$: 14.5 (br.s, 1H), 8.37-8.28 (m, 2H), 7.65 (ddd, J = 9.1, 7.1, 2.0, 1H), 7.45-7.40 (m, 3H), 7.31 (d, J = 8.8, 1H), 6.77 (t, J = 6.4, 1H). ¹³C NMR (100.62 MHz, CDCl₃) $\delta_{\rm C}$: 155.5, 142.9, 139.4, 139.2, 132.9, 132.1, 131.9, 131.1, 126.7, 115.5, 113.2. **GC-MS** (EI) m/z 270 (M⁺+2, 2), 268 (M⁺, 47), 266

 ESI^{+}): calcd for $C_{14}H_{11}BrF_{3}NNaO_{2}S$ (M+Na)⁺: 415.9538; Found: 415.9531.

(100), 264 (74), 172 (22), 168 (16), 133 (22), 124 (30), 121 (24), 118 (15), 109 (32), 98 (17), 79 (16), 78 (16), 62 (14). **HRMS** (TOF, ESI⁺): calcd for $C_{11}H_9ClN_2NaO_2S$ (M+Na)⁺: 290.9966; Found: 290.9965.

Representative Procedure for Photostimulated Reactions. Reactions in DMSO (THF or CH₃CN). The following procedure is representative of all these reactions. The reaction was carried out in a Schlenk tube equipped with a nitrogen inlet and magnetic stirred at r.t. DMSO (5 ml) was dried and deoxygenated, then *t*-BuOK (3.0 equiv, 50.5 mg, 0.45 mmol) was added and after 5 min the corresponding 2-halo-*N*-phenylbenzenesulfonamide (1 equiv, 0.15 mmol) was added and the reaction mixture was irradiated for the corresponding time. In case the 2-halo-*N*-phenylbenzenesulfonamide was an oil, it was added dissolved in anhydrous ethyl ether. The reaction was quenched with ammonium nitrate in excess. The "work-up" of the reaction could have two processes. "Work-up A" the residue was extracted with ethyl acetate (EtOAc) (3 x 30 ml) and the organic extracted was washed with water and dried with anhydrous Na₂SO₄. The solvent was removed under reduced pressure to give the crude products. The products were purified by chromatography on silica gel or quantified by GC using the internal standard method. Or "Work-up B" the residue was filtrate over a bed of silica gel with 300 ml of pentane/ EtOAc 70:30 %. The solvent was removed under reduced pressure and the crude obtained was then recrystallized.

Reaction in Liquid Ammonia. Liquid ammonia (150 ml), previously dried over Na metal, was distilled into a 250 mL three-necked, round-bottomed flask equipped with a cold-finger condenser and a magnetic stirrer under a nitrogen atmosphere. The base *t*-BuOK (3.0 equiv, 50.5 mg, 0.45 mmol) and then the corresponding 2-halo-*N*-phenylbenzenesulfonamides (1 equiv, 0.15 mmol) were added to the liquid ammonia. After 180 min of irradiation the reaction was quenched by addition of NH₄NO₃ in excess, and the ammonia was allowed to evaporate. Water (50 mL) was added to the residue and the mixture was extracted with ethyl acetate (3 x 30 mL). The organic extract was dried over Na₂SO₄, and filtered. The solvent was removed under reduced pressure to leave the crude products. The products were purified by chromatography on silica gel or quantified by GC using the internal standard method.

Isolation and Identification of Products. 6*H*-Dibenzo[c,e][1,2]thiazine 5,5-dioxide (**6a**). The product was purified by column chromatography on silica gel eluting with pentane/ EtOAc (100:0 → 75:25 %). White solid was isolated in 86% yield (29.6 mg, 0.128 mmol), m.p. 195-197°C (lit. 10 194-195°C). The NMR (400.16 MHz, CD₃COCD₃) $\delta_{\rm H}$: 9.89 (br.s, 1H), 8.24-8.19 (m, 2H), 7.89 (dd, J = 7.8, 1.0, 1H), 7.84-7.80 (m, 1H), 7.68 (td, J = 7.6, 0.8, 1H), 7.49 (td, J = 7.6, 1.2, 1H), 7.36-7.31 (m, 2H). The NMR (100.62 MHz, CD₃COCD₃) $\delta_{\rm C}$: 136.8, 135.5, 132.4, 132.3, 130.2, 128.2, 125.4,

125.2, 124.1, 122.4, 121.3, 119.9. **GC-MS** (EI) m/z 232 (M⁺+1, 13), 231 (M⁺, 91), 168 (14), 167 (100), 166 (56), 140 (29), 139 (39), 115 (9), 113 (10), 89 (8), 84 (10), 70 (11), 69 (10), 63 (14). 9-Methyl-6H-dibenzo[c,e][1,2]thiazine 5,5-dioxide (6b). The product was purified by column chromatography on silica gel eluting with pentane/ EtOAc (100:0 \rightarrow 70:30 %). White solid was isolated in 98% yield (36.1 mg, 0.147 mmol). This solid was recrystallized from acetone/pentane as white crystal, m.p. 217-219 °C. ¹H NMR (400.16 MHz, CD₃SOCD₃) δ_{H} : 11.19 (br.s, 1H), 8.23 (d, J = 8.0, 1H), 8.02 (s, 1H), 7.96 (dd, J = 7.8, 1.0, 1H), 7.82-7.78 (m, 1H), 7.65 (t, J = 7.4, 1H), 7.28 (dd, J = 8.2, 1.0, 1H), 7.11 (d, J = 8.0, 1H), 2.369 (s, 3H). ¹³C NMR (100.62 MHz, CD₃SOCD₃) δ_{C} : 134.6, 134.1, 133.2, 132.4, 131.8, 131.1, 128.4, 125.5, 125.4, 121.4, 121.1, 119.8, 20.6. **GC-MS** (EI) m/z 246 (M⁺+1, 9), 245 (M⁺, 69), 181 (21), 180 (100), 178 (9), 153 (6), 152 (17), 151 (7), 127 (5), 90 (9), 89 (5), 77 (11), 76 (7), 75 (5), 63 (6), 51 (5). **HRMS** (TOF, ESI⁺): calcd for C₁₃H₁₁NNaO₂S (M+Na)⁺: 268.0403; Found: 268.0400.

9-Methoxy-6H-dibenzo[c,e][1,2]thiazine 5,5-dioxide (**6c**). The product was purified by column chromatography on silica gel eluting with pentane/ EtOAc (100:0 \rightarrow 60:40 %). White solid was isolated in 77% yield (30.2 mg, 0.115 mmol). This solid was recrystallized from acetone/pentane as light yellow crystal, m.p. 209-210 °C. ¹H NMR (400.16 MHz, CD₃SOCD₃) δ_H: 11.0 (br.s, 1H), 8.29 (d, J = 8.0, 1H), 7.92 (dd, J = 7.8, 1.0, 1H), 7.82-7.78 (m, 1H), 7.70 (d, J = 2.8, 1H), 7.67 (td, J = 7.6, 0.8, 1H), 7.16 (d, J = 8.8, 1H), 7.09 (dd, J = 8.8, 2.8, 1H), 3.86 (s, 3H). ¹³C NMR (100.62 MHz, CD₃SOCD₃) δ_C: 156.1, 134.8, 132.4, 131.7, 129.7, 128.7, 126.0, 123.1, 121.8, 121.2, 117.1, 109.5, 55.6. **GC-MS** (EI) m/z 262 (M⁺+1,11), 261 (M⁺, 69), 183 (14), 182 (100), 155 (8), 154 (70), 153 (14), 128 (23), 127 (30), 126 (12), 77 (10), 75 (8), 51 (8). **HRMS** (TOF, ESI⁺): calcd for C₁₃H₁₁NNaO₃S (M+Na)⁺: 284.0352; Found: 284.0357.

8,10-Di-methoxy-6H-dibenzo[c,e][1,2]thiazine 5,5-dioxide (6d). The product was purified by column chromatography on silica gel eluting with pentane/ EtOAc (100:0 \rightarrow 60:40 %). Light yellow solid m.p. 229-231 °C was isolated in 65% yield (28.4 mg, 0.0975 mmol).. ¹H NMR (400.16 MHz, CD₃COCD₃) $\delta_{\rm H}$: 9.70 (br.s, 1H), 8.59 (d, J=8.4, 1H), 7.90 (dd, J=7.8, 1.0, 1H), 7.66 (td, J=7.8, 1.2, 1H), 7.52 (td, J=7.6, 0.8, 1H), 6.56 (d, J=2.4, 1H), 6.49 (d, J=2.4, 1H), 4.00 (s, 3H), 3.87 (s, 3H). ¹³C NMR (100.62 MHz, CD₃COCD) $\delta_{\rm C}$: 161.5, 159.6, 139.4, 134.7, 131.4, 131.3, 128.7, 126.4, 120.8, 105.5, 96.5, 95.0, 55.5, 55.0. GC-MS (EI) m/z 292 (M⁺+1, 29), 291 (M⁺, 100), 290 (12), 227 (17), 226 (9), 212 (11), 185 (9), 136 (9), 127 (14), 114 (9), 113 (9). HRMS (TOF, ESI⁺): calcd for C₁₄H₁₃NNaO₄S (M+Na)⁺: 314.0457; Found: 314.0469.

9-Phenyl-6H-dibenzo[c,e][1,2]thiazine 5,5-dioxide (**6e**). The product was purified by column chromatography on silica gel eluting with pentane/ EtOAc (70:30 %). White solid was isolated in 94% yield (43.3 mg, 0.141 mmol). This solid was recrystallized from EtOAc /pentane as colorless

flakes, m.p. 235-236 °C. ¹H NMR (400.16 MHz, CD₃SOCD₃) δ_{H} : 11.50 (br.s, 1H), 8.47 (d, J = 8.4, 1H), 8.45 (d, J = 2.0, 1H), 7.96 (d, J = 7.2, 1H), 7.85-7.77 (m, 4H), 7.70 (t, J = 7.6, 1H), 7.50 (t, J = 7.6, 1 7.6, 2H), 7.39 (t, J = 7.6, 1H), 7.30 (d, J = 8.4, 1H). ¹³C NMR (100.62 MHz, CD₃SOCD₃) δ_C : 139.4, 136.0, 135.9, 134.5, 132.6, 131.7, 128.9, 128.8, 128.7, 127.5, 126.8, 126.1, 123.5, 121.8, 121.1, 120.2. **HRMS** (TOF, ESI⁺): calcd for $C_{18}H_{13}NNaO_2S$ (M+Na)⁺: 330.0559; Found: 330.0551. 9-(Trifluoromethyl)-6H-dibenzo[c,e][1,2]thiazine 5,5-dioxide (6f). The product was purified by column chromatography on silica gel eluting with pentane/ EtOAc (100:0 \rightarrow 60:40 %). White solid was isolated in 79% yield (35.5 mg, 0.118 mmol). This solid was recrystallized from ethyl eter/pentane as white crystal, m.p. 234-236 °C. ¹H NMR (400.16 MHz, CD₃COCD₃) δ_H: 10.48 (br.s, 1H), 8.56 (s, 1H), 8.42-8.40 (m, 1H), 8.04 (dd, J = 7.8, 1.0, 1H), 7.89 (ddd, J = 8.0, 7.6, 1.2, 1H), 7.84-7.81 (m, 1H), 7.77 (1H, td, J = 7.6, 1.2, 1H), 7.52 (d, J = 8.4, 1H). ¹³C NMR (100.62) MHz, CD₃COCD) δ_C : 140.0, 135.4, 132.7, 131.1, 129.3, 126.8 (q, J = 4, 1C), 126.1, 125.5 (q, J = 4) 33, 1C), 124.3 (q, J = 270, 1C), 122.7 (q, J = 4, 1C), 122.3, 121.5, 120.2. **GC-MS** (EI) m/z 300 $(M^{+}+1, 26), 299 (M^{+}, 100), 281 (22), 236 (15), 235 (61), 234 (17), 216 (29), 207 (53), 204 (13),$ 185 (28), 166 (23), 140 (14), 139 (17), 93 (15), 69 (16), 58 (74), 57 (18). (26). **HRMS** (TOF, ESI⁺): calcd for $C_{13}H_8F_3NNaO_2S$ (M+Na)⁺: 322.0120; Found: 322.0120.

Ethyl 6H-dibenzo[c,e][1,2]thiazine-9-carboxylate 5,5-dioxide (**6g**). The product was purified by column chromatography on silica gel eluting with pentane/ EtOAc (100:0 \rightarrow 50:50 %). Light yellow solid was isolated in 54% yield (24.6 mg, 0.081 mmol). This solid was recrystallized from acetone/pentane as small white solid, m.p. 263-264 °C. ¹H NMR (400.16 MHz, CD₃SOCD₃) δ_H: 12.0 (br.s, 1H), 8.71 (s, 1H), 8.30 (d, J = 8.1, 1H), 8.03 (d, J = 8.4, 1H), 7.97 (d, J = 7.8, 1H), 7.87-7.83 (m, 1H), 7.72 (t, J = 7.6, 1H), 7.29 (d, J = 8.3, 1H), 4.36 (q, J = 7.1, 2H), 1.36 (t, J = 7.1, 3H). ¹³C NMR (100.62 MHz, CD₃SOCD₃) δ_C: 165.2, 142.2, 134.0, 132.5, 131.0, 130.7, 128.9, 126.2, 125.4, 124.0, 121.4, 120.3, 119.8, 60.7, 14.2. **GC-MS** (EI) m/z 304 (M⁺+1, 13), 303 (M⁺, 70), 275 (35), 259 (18), 258 (100), 211 (9), 194 (17), 167 (10), 166 (36), 165 (9), 164 (13), 140 (21), 139 (46), 138 (9). **HRMS** (TOF, ESI⁺): calcd for C₁₅H₁₃NNaO₄S (M+Na)⁺: 326.0458; Found: 326.0454.

H-*Dibenzo*[*c*,*e*][1,2]thiazine-9-carbonitrile 5,5-dioxide (**6h**). The product was filtrate over a bed of silica gel with 300 ml of pentane/ EtOAc 70:30 %. The solvent was removed under reduced pressure and then was recrystallized from acetone/pentane. Light yellow solid was isolated in 91% yield (35.0 mg, 0.136 mmol), m.p. decomposed over 213°C. ¹**H NMR** (400.16 MHz, CD₃SOCD₃) δ_H: 12.16 (br.s, 1H), 8.80 (d, J = 1.6, 1H), 8.40 (d, J = 8.0, 1H), 7.99 (dd, J = 7.8, 1.0, 1H), 7.91-7.85 (m, 2H), 7.77-7.73 (m, 1H), 7.33 (d, J = 8.4, 1H). ¹³**C NMR** (100.62 MHz, CD₃SOCD₃) δ_C: 140.4, 134.0, 133.6, 132.9, 130.2, 130.1, 129.6, 126.2, 121.4, 121.2, 120.0, 118.6, 106.0. **GC-MS**

(EI) m/z 257 (M⁺+1, 20), 256 (M⁺, 85), 193 (17), 192 (100), 191 (39), 165 (36), 164 (49), 144 (15), 139 (11), 138 (19), 89 (9), 83 (10), 75 (12), 63 (13), 50 (9). **HRMS** (TOF, ESI⁺): calcd for $C_{13}H_8N_2NaO_2S$ (M+Na)⁺: 279.0199; Found: 279.0190.

9-Fluoro-6H-dibenzo[c,e][1,2]thiazine 5,5-dioxide (**6i**). The product was purified by column chromatography on silica gel eluting with pentane/ EtOAc (100:0 \rightarrow 60:40 %). White solid was isolated in 81% yield (30.0 mg, 0.121 mmol). This solid was recrystallized from EtOAc/pentane as white crystal, m.p. 214-215 °C. ¹H NMR (400.16 MHz, CD₃SOCD₃) $\delta_{\rm H}$: 11.41 (br.s, 1H), 8.28 (d, J=7.9, 1H), 8.13 (dd, J=10.2, 2.8, 1H), 7.95 (dd, J=7.8, 1.1, 1H), 7.85-7.81 (m, 1H), 7.71 (td, J=7.6, 1.0, 1H), 7.36 (td, J=8.6, 2.8, 1H), 7.24 (dd, J=8.8, 5.2, 1H). ¹³C NMR (100.62 MHz, CD₃SOCD₃) $\delta_{\rm C}$: 158.7 (d, J=238, 1C), 134.5, 132.9, 132.6, 130.9, 129.3, 126.1, 123.3 (d, J=9, 1C), 121.8 (d, J=8, 1C), 121.2, 117.5 (d, J=23, 1C), 111.8 (d, J=24, 1C). HRMS (TOF, ESI⁺): calcd for C₁₂H₈FNNaO₂S (M+Na)⁺: 272.0152; Found: 272.0153

7-Phenyl-6H-dibenzo[c,e][1,2]thiazine 5,5-dioxide (6j). The product was purified by column chromatography on silica gel eluting with pentane/ EtOAc (100:0 \rightarrow 70:30 %). White solid was isolated in 83% yield (38.3 mg, 0.124 mmol). This solid was recrystallized from EtOAc/pentane as small white needles, m.p. 221-222 °C. ¹H NMR (400.16 MHz, CD₃SOCD₃) $\delta_{\rm H}$: 10.2 (br.s, 1H), 8.27 (d, J = 8.0, 1H), 8.23 (dd, J = 7.8, 1.4, 1H), 7.91 (dd, J = 7.6, 0.8, 1H), 7.85-7.80 (m, 1H), 7.70-7.67 (m, 1H), 7.58-7.41 (m, 7H). ¹³C NMR (100.62 MHz, CD₃SOCD₃) $\delta_{\rm C}$: 138.0, 137.0, 135.7, 133.1, 132.6, 132.4, 131.6, 129.5, 128.9, 128.4, 127.6, 126.5, 126.2, 125.0, 121.6. HRMS (TOF, ESI⁺): calcd for C₁₈H₁₃NNaO₂S (M+Na)⁺: 330.0559; Found: 330.0554.

7-(1H-Pyrrol-1-yl)-6H-dibenzo[c,e][1,2]thiazine 5,5-dioxide (6k). The product was purified by column chromatography on silica gel eluting with pentane/ EtOAc (100:0 \rightarrow 70:30 %). White solid was isolated in 78% yield (34.7 mg, 0.117 mmol). This solid was recrystallized from EtOAc/pentane as light yellow crystal, m.p. 196-197 °C. ¹H NMR (400.16 MHz, CD₃SOCD₃) δ_H:10.4 (br.s, 1H), 8.28 (d, J = 8.0, 1H), 8.21 (dd, J = 7.8, 1.4, 1H), 7.94 (dd, J = 7.8, 1.0, 1H), 7.87-7.82 (m, 1H), 7.73-7.70 (m, 1H), 7.55-7.51 (m, 1H), 7.49 (dd, J = 8.0, 1.6, 1H), 7.14 (t, J = 2.0, 2H), 6.31 (t, J = 2.2, 2H). ¹³C NMR (100.62 MHz, CD₃SOCD₃) δ_C: 135.7, 135.6, 132.7, 132.0, 129.8, 129.3, 127.3, 127.0, 126.7, 124.1, 122.1, 121.8, 109.5. **GC-MS** (EI) m/z 297 (M⁺+1, 12), 296 (M⁺, 75), 233 (14), 232 (94), 231 (71), 229 (12), 205 (29), 204 (100), 164 (14), 151 (11), 139 (16), 115 (16), 102 (18). **HRMS** (TOF, ESI⁺): calcd for C₁₆H₁₂N₂NaO₂S (M+Na)⁺: 319.0512; Found: 319.0513.

5H-Benzo[e]naphtho[1,2-c][1,2]thiazine 6,6-dioxide (61). The product was purified by column chromatography on silica gel eluting with pentane/ EtOAc (70:30 %). Red solid was isolated in 69% yield (29.1 mg, 0.103 mmol). This solid was recrystallized from Acetone/pentane as brown

crystal, m.p. 281-283°C. ¹**H NMR** (400.16 MHz, CD₃SOCD₃) δ_{H} : 11.36 (br.s, 1H), 8.41-8.40 (m, 1H), 8.35 (d, J = 8.0, 1H), 8.30 (d, J = 8.8, 1H), 8.04-8.02 (m, 1H), 8.00 (dd, J = 7.6, 1.2, 1H), 7.93 (d, J = 8.8, 1H), 7.88-7.84 (m, 1H), 7.74-7.66 (m, 3H). ¹³**C NMR** (100.62 MHz, CD₃SOCD₃) δ_{C} : 134.9, 133.6, 132.6, 132.4, 132.2, 128.6, 128.1, 127.6, 127.0, 126.7, 126.5, 125.1, 123.0, 122.3, 121.4, 119.7. **GC-MS** (EI) m/z 282 (M⁺+1, 16), 281 (M⁺, 82), 218 (14), 217 (100), 216 (49), 189 (16), 187 (10), 108 (29), 95 (17), 94 (13)..**HRMS** (TOF, ESI⁺): calcd for C₁₆H₁₁NNaO₂S (M+Na)⁺: 304.0403; Found: 304.0392.

H-Benzo[e]naphtho[2,1-c][1,2]thiazine 5,5-dioxide (**6m**). After evaporation of the solvent the organic phase was purified by column chromatography on silica gel eluting with pentane/ EtOAc (100:0 → 80:20 %) obtaining a mixture of products. After recrystallization of the mixture from acetone/pentane **6m** was afforded as a colorless solid, m.p. decomposed over 223 °C (51 % isolated yield, 21.5 mg, 0.075 mmol). ¹**H NMR** (400.16 MHz, CDCl₃) $\delta_{\rm H}$: 8.54 (d, J = 8.4, 1H), 8.27 (d, J = 8, 1H), 8.09 (d, J = 7.2, 1H), 7.92-7.87 (m, 2H), 7.72 (d, J = 7.2, 1H), 7.63-7.59 (m, 2H), 7.52 (d, J = 7.6, 1H), 7.23 (d, J = 8.4, 1H). **GC-MS** (EI) m/z 282 (M⁺+1, 18), 281 (M⁺, 100), 218 (13), 217 (80), 216 (43), 214 (14), 190 (19), 189 (39), 187 (11), 109 (22), 96 (11), 95 (25), 94 (43), 82 (13). **HRMS** (TOF, ESI⁺): calcd for C₁₆H₁₁NNaO₂S (M+Na)⁺: 304.0403; Found: 304.0392.

H-Benzo[5,6][1,2]thiazino[3,4-c]quinoline 5,5-dioxide (**6n**). An especial procedure was followed to purify this compound. The crude was obtained by extraction from acid media (pH=1, H₂SO₄), with EtOAc (3 x 30 ml), the combined organic layers were washed with H₂O (20 mL) and dried over anhydrous MgSO₄; the solvent was evaporated under vacuum. Without further purification a yellow solid was obtained in 97% yield (41.1 mg, 0.145 mmol), m.p. decomposed over 185 °C. ¹H NMR (400.16 MHz, CD₃SOCD₃) δ_H: 8.94 (s, 1H), 8.63 (d, J = 7.6, 1H), 8.46 (d, J = 7.6, 1H), 8.22-8.14 (m, 2H), 8.00-7.83 (m, 4H). ¹³C NMR (100.62 MHz, CD₃SOCD₃) δ_C: 143.8, 143.7, 136.9, 132.4, 130.6, 130.5, 129.9, 129.2, 128.9, 128.8, 128.6, 125.3, 124.2, 123.4, 121.7. HRMS (TOF, ESI⁺): calcd for C₁₅H₁₁N₂O₂S (M+H)⁺: 283.0536; Found: 283.0530.

2-Fluoro-6H-dibenzo[c,e][1,2]thiazine 5,5-dioxide (**60**). The product was purified by column chromatography on silica gel eluting with pentane/ EtOAc (100:0 \rightarrow 70:30 %). White solid was isolated in 44% yield (16.5 mg, 0.066 mmol). This solid was recrystallized from EtOAc/pentane as colorless crystal, m.p. 214-216 °C. ¹H NMR (400.16 MHz, CD₃SOCD₃) $\delta_{\rm H}$: 11.5 (br.s, 1H), 8.23 (d, J = 7.2, 1H), 8.15 (dd, J = 10.8, 2.4, 1H), 8.00 (dd, J = 8.8, 5.6, 1H), 7.53-7.48 (m, 2H), 7.31-7.27 (m, 1H), 7.21 (d, J = 8.0, 1H). ¹³C NMR (100.62 MHz, CD₃SOCD₃) $\delta_{\rm C}$: 165.4 (d, J = 248, 1C), 137.1, 134.9 (d, J = 9, 1C), 130.9 (d, J = 2, 1C), 124.4 (d, J = 10, 1C), 120.6 (d, J = 4, 1C), 115.9 (d, J = 23, 1C), 112.3 (d, J = 20, 1C). HRMS (TOF, ESI⁺): calcd for C₁₂H₈FNNaO₂S (M+Na)⁺: 272.0152; Found: 272.0164.

2-Fluoro-9-methyl-6H-dibenzo[c,e][1,2]thiazine 5,5-dioxide (**6p**). The product was purified by column chromatography on silica gel eluting with pentane/ EtOAc (100:0 \rightarrow 80:20 %). White solid was isolated in 43% yield (17.0 mg, 0.065 mmol), m.p. 244-245 °C. ¹H NMR (400.16 MHz, CD₃COCD₃) $\delta_{\rm H}$: 8.03-7.98 (m, 3H), 7.44 (td, J = 8.5, 2.4, 1H), 7.34 (d, J = 8.4, 1H), 7.21 (d, J = 8.0, 1H), 2.43 (s, 3H). ¹³C NMR (100.62 MHz, CD₃COCD₃) $\delta_{\rm C}$: 164.7 (d, J = 248, 1C), 135.7 (d, J = 9, 1C), 134.8, 134.1, 132.1 (d, J = 3, 1C), 131.8, 125.9, 124.6 (d, J = 10, 1C), 121.7, 120.2, 115.4 (d, J = 23, 1C), 112.1 (d, J = 25, 1C), 20.0. **GC-MS** (EI) m/z 264 (M⁺+1, 2), 263 (M⁺, 75), 199 (22), 198 (100), 196 (6), 170 (13), 151 (6), 99 (9), 89 (6), 86 (7). **HRMS** (TOF, ESI⁺): calcd for C₁₃H₁₀FNNaO₂S (M+Na)⁺: 286.0308; Found: 286.0304.

9-Methyl-2-(trifluoromethyl)-6H-dibenzo[c,e][1,2]thiazine 5,5-dioxide (6q). The product was purified by column chromatography on silica gel eluting with pentane/ EtOAc (100:0 \rightarrow 80:20 %). White solid was isolated in 42% yield (19.7 mg, 0.063 mmol), m.p. 244-245 °C. ¹H NMR (400.16 MHz, CDCl₃) $\delta_{\rm H}$: 8.21 (s, 1H), 8.11 (d, J = 8, 1H), 7.82-7.79 (m, 2H), 7.30 (d, J = 8, 1H), 7.09 (d, J = 8, 1H), 7.07 (br. s, 1H) 2.48 (s, 3H). ¹³C NMR (100.62 MHz, CDCl₃) $\delta_{\rm C}$: 137.7, 135.7, 134.4 (q, J = 33, 1C), 133.5, 133.0, 132.2, 125.8, 124.8 (q, J = 4, 1C), 123.3 (q, J = 271, 1C), 123.2, 122.6 (q, J = 4, 1C), 122.4, 121.2 21.2. **GC-MS** (EI) m/z 314 (M⁺+1, 13), 313 (M⁺, 98), 249 (23), 248 (100), 229 (8), 228 (14), 180 (26), 179 (15), 178 (15), 152 (10), 151 (7), 114 (9), 69 (9). **HRMS** (TOF, ESI⁺): calcd for C₁₄H₁₀F₃NNaO₂S (M+Na)⁺: 336.0276; Found: 336.0264.

Acknowledgments. We thank Licentiate student Andres G. Teobaldi for conducting some of the reactions. This work was supported in part by the Agencia Córdoba Ciencia, the Consejo Nacional de Investigaciones Científicas y Técnicas (CONICET), Secretaría de Ciencia y Tecnología, Universidad Nacional de Córdoba (SECyT), and the Agencia Nacional de Promoción Científica y Técnica (ANPCyT). W.D.G. gratefully acknowledges the receipt of a fellowship from CONICET.

Supporting Information Available. The screening of optimal conditions for *N*-arylation Coppercatalyzed for the synthesis of *N*-phenyl-2-halobenzenesulfonamides and the Uv-vis spectra for **3a** and anion derivative are presented in Supporting Information. Copies of ¹H NMR and ¹³C NMR spectra for all substrates and products and theoretical section (xyz of stationary points) are available

in Supporting Information. This material is available free of charge via the Internet at http://pubs.acs.org.

References

(1) Winum, J-Y.; Scozzafava, A.; Montero, J-L.; Supuran, C. T. Med. Res. Rev. 2006, 26 (6), 767-792.

- (3) Brzozowski, Z.; Sączewski, F.; Neamati, N. J Bioorg. Med. Chem. Lett. 2006, 16, 5298-5302.
- (4) Wells, G. J.; Tao, M.; Josef, K.A.; Bihovsky, R. J. Med. Chem. 2001, 44, 3488-3503.
- (5) Xie, Y.; Deng, S.; Thomas, C. J.; Liu, Y.; Zhang, Y.-Q.; Rinderspacher, A.; Huang, W.; Gong, G.; Wyler, M.; Cayanis, E.; Aulner, N.; Tobben, U.; Chung, C.; Pampou, S.; Southall, N.; Vidović, D.; Schurer, S.; Branden, L.; Davis, R. E.; Staudt, L. M.; Inglese, J.; Austin, C. P.; Landry, D. W.; Smith, D. H.; Auld, D. S. *Bioorg. Med. Chem. Lett.* **2008**, *18*, 329-335.
- (6) Majumdar, K. C.; Mondal, S. Chem. Rev. 2011, 111, 7749-7773.
- (7) Biswas, D.; Samp, L.; Ganguly, A. K. Tetrahedron Lett. 2010, 51, 2681-2684.
- (8) (a) Rayabarapu, D. K.; Zhou, A.; Jeon, K. O.; Samarakoon, T.; Rolfe, A.; Siddiqui, H.; Hanson, P. R. *Tetrahedron* **2009**, *65*, 3180-3188. (b) Rousseaux, S.; Gorelsky, S. I.; Chung, B. K. W.; Fagnou, K. *J. Am. Chem. Soc.* **2010**, *132*, 10692-10706.
- (9) Li, Y.; Ding, Q.; Qiu, G.; Wu, J. Org. Biomol. Chem. 2014, 12, 149-155.
- (10) Laha, J. K.; Jethava, K. P.; Dayal, N. J. Org. Chem. 2014, 79, 8010-8019.
- (11) Ujjainwalla, F.; da Mata, M. L. E. N.; Pennell, A. M. K.; Escolano, C.; Motherwell, W. B.; Vazquez, S. *Tetrahedron.* **2015**, 71 (38), 6701-6719.
- (12) For reviews see: (a) Budén, M. E., Martín, S. E., Rossi, R. A. *Recent Advances in the Photoinduced Radical Nucleophilic Substitution Reactions*, in CRC Handbook of Organic Photochemistry and Photobiology, 3th ed., Eds. Griesbeck, A. G., Oelgemöller, M., and Ghetti, F. CRC Press Inc. Boca Raton, 2012, Chapter 15, p.p. 347-368. (b) The S_{RN}1 Reaction, Bardagí, J. I., Vaillard, V. A., Rossi, R. A., in Encyclopedia of Radicals in Chemistry, Biology & Materials, Chatgilialoglu, C., Studer, A., Eds., John Wiley & Sons Ltd, Chichester, UK 2012, pp 333-364. (c) Rossi, R. A.; Pierini, A. B.; Peñéñory, A. B. *Chem. Rev.* 2003, 103, 71-167. (d) Rossi, R. A.,

⁽²⁾ Wang, J.; Limburg, D.; Carter, J.; Mbalaviele, G.; Gierse, J.; Vazquez, M. J. *Bioorg. Med. Chem. Lett.* **2010**, *20*, 1604-1609.

- Guastavino, J. F., Budén, M. E., *The S_{RN}I Reaction*, in "*Arene Chemistry: Reaction Mechanisms and Methods for Aromatic Compounds*". Part 2. Nucleophilic Aromatic Substitution, Editor J. Mortier, John Wiley & Sons Ltd, Chichester, UK, 2016, Chapter 10, pp 243-268.
- (13) For some examples of intramolecular S_{RN}1 reactions see: (a) Guerra, W. D.; Rossi, R. A.; Pierini A. B.; Barolo, S. M. *J. Org. Chem.* **2015**, *80*, 928-941. (b) Budén, M. E.; Dorn, V. B.; Gamba, M.; Pierini., A. B.; Rossi, R. A. *J. Org. Chem.* **2010**, *75*, 2206-2218. (c) Barolo, S. M.; Wang, Y.; Rossi, R. A.; Cuny, G. D. *Tetrahedron* **2013**, *69*, 5487-5494.
- (14) Layman, W. J.; Greenwood, T. D.; Downey, A. L.; Wolfe, J. F. J. Org. Chem. 2005, 70, 9147-9155.
- (15) (a) He, H.; Wu, Y-J. *Tetrahedron Lett.* **2003**, *44*, 3385-3386. (b) Wei Deng, W.; Liu, L.; Zhang, C.; Liu, M.; Guo, Q-X. *Tetrahedron Lett.* **2005**, *46*, 7295-7298. (c) Wanga, X.; Guram, A.; Ronk, M; Milne, J. E.; Tedrow, J. S.; Faul, M. M. *Tetrahedron Lett.* **2012**, *53*, 7-10.
- (16) Reactions carried out under N₂ with 0.6 mmol of 2-chlorobenzenesulfonamide (1), 1.2 mmol of aryl halide (2), 0.06 mmol of CuI, 0.3 mmol of DMEDA, 1.8 mmol of K₂CO₃ in 2 mL of CH₃CN. For more details see Supporting Information, Screening of Optimal Conditions for *N*-arylation Copper-catalyzed.
- (17) Ryan, S. J.; Francis C. L.; Savage, J. P. Aust. J. Chem. 2013, 66, 874-881.
- (18) Reaction carried out under N₂ with 0.6 mmol of sulfonyl chloride (**4a-c**), 0.72 mmol of corresponding aniline (**5**), 1.8 mmol of pyridine in 1.2 mL of CH₂Cl₂ at room temperature for 18h.
- (19) Uv-vis spectra for compounds $3a (7.4 \times 10^{-5} \text{ M})$ and anion derivative $3a^{-}$ are presented in supporting information.
- (20) The reaction was tested with 4 equivalents of *t*-BuOK with the same results.
- (21) Ravelli, D.; Dondi, D.; Fagnoni, M.; Albini, A. Chem. Soc. Rev. 2009, 38, 1999-2011.
- (22) Initially, the reaction employing a Blue LED light (LEDs ($\lambda = 455 \pm 15$ nm), 3 W, 700 mA) shown to proceed without full conversion even after 6h.
- (23) (a) Frei, R.; Wodrich, M. D.; Hari, D. P.; Borin, P-A.; Chauvier, C.; Waser, J. J. Am. Chem. Soc. **2014**, 136 (47), 16563–16573. (b) Zhao, Y.; Truhlar, D. G. Theor. Chem. Acc. **2008**, 120, 215-241. (c) Zhao, Y.; Truhlar, D. G. Acc. Chem. Res. **2008**, 41, 157-167.
- (24) (a) Miertus, S.; Scrocco, E.; Tomasi, *J. Chem. Phys.* **1981**, *55*, 117-129. (b) Miertus, S.; Tomasi, J. *Chem. Phys.* **1982**, *65*, 239-245. (c) Cossi, M.; Barone, V.; Cammi, R.; Tomasi, J. *Chem. Phys. Lett.* **1996**, *255*, 327-335.
- (25) Schmidt, L. C.; Argüello, J. E.; Peñéñory, A. B. J. Org. Chem. 2007, 72, 2936-2944.
- (26) The first excited state of 3a can be achieved with an energy of 101 kcal/mol ($\lambda = 282$ nm).

(27) Vera, M. A.; Pierini A. B. J. Org. Chem. 2003, 68, 9191-9199.

(28) Cao, X.; Bai, Y.;: Xie, Y.; Deng, G-J. J. Mol. Catal. A: Chem. 2014, 383-384, 94-100.