

## Synthesis of New Pyrazole and Pyrimidine Steroidal Derivatives

Daniel G. Rivera <sup>1</sup>, Klaus Peseke <sup>2,\*</sup>, Isabel Jomarrón <sup>1</sup>, Alina Montero <sup>3</sup>, Reinaldo Molina <sup>3</sup> and Francisco Coll <sup>1</sup>

<sup>1</sup> Centro de Investigaciones de Producto Naturales, Facultad de Química, Universidad de La Habana, Zapata y G, Vedado, 10400 Habana, Cuba

<sup>2</sup> Fachbereich Chemie der Universität Rostock, D-18051 Rostock, Germany

<sup>3</sup> Centro Bioactivos Químicos, Universidad Central de Las Villas, Santa Clara, Cuba

\* Author to whom the correspondence should be addressed; Fax +49-381-498-6410, E-mail: [klaus.peseke@chemie.uni-rostock.de](mailto:klaus.peseke@chemie.uni-rostock.de); <http://www.fb-chemie.uni-rostock.de/organik/peseke/index.html>

Received: 16 April 2003; in revised form 8 May 2003 / Accepted: 9 May 2003 / Published: 31 May 2003

---

**Abstract:** The synthesis of steroidal heterocycles containing the pyrazole and pyrimidine ring fused to the 16,17-position of the steroid nucleus is reported. Androstenolone acetate (**1**) reacted with carbon disulfide, iodomethane and sodium hydride to furnish 3 $\beta$ -acetoxy-16-[bis(methylthio)methylene]-5-androst-5-en-17-one (**2**). The reactions of **2** with hydrazine hydrate and methylhydrazine afforded the 5'-methylthio-pyrazolo[4',3':16,17]androst-5-en-3 $\beta$ -ols **3a** and **3b**, respectively. Treatment of **2** with amidinium, guanidinium, and isothiuronium salts in the presence of sodium methoxide yielded the 6'-methoxy- pyrimido[5',4':16,17]androst-5-en-3 $\beta$ -ols (**4a-4e**).

**Keywords:** Androstenolone, steroidal heterocycles, push-pull alkenes, pyrazoles, pyrimidines

---

## Introduction

Steroidal heterocycles containing anellated rings in the steroidal moiety have prompted a great interest in the biological study of many heterocyclic compounds. Specially, in the last two decades new compounds were synthesized in which the androstane ring A was condensed with a great variety of heterocyclic rings [1-13]. As should be expected, the fusion of heterocycles to steroids often led to a change in their physiological activities and the appearance of new interesting biological behavior. Thus, several steroidal heterocycles have been obtained exhibiting activity as potential inhibitors of cytochrome P450 enzyme aromatase [14,15], with their subsequent clinical application in the treatment of estrogen-dependent breast cancers, while others which have shown moderate binding affinity for the benzodiazepine receptor [16].

There is no doubt that the main procedure for the synthesis of steroidal heterocycles has been the reaction of  $\alpha$ -hydroxymethylene-oxo-steroids with N,N'-dinucleophiles and N,O-dinucleophiles. We wish to describe in this paper another procedure for the preparation of anellated steroids. The reaction of cyclic and acyclic ketones with carbon disulfide in the presence of a base and an alkylating agent to give  $\alpha$ -oxoketene dithioacetals is known [17,18]. These push-pull activated alkenes were reacted with dinucleophiles to yield heterocyclic compounds [19-21]. Furthermore, the reactions with carbon disulfide have been widely applied to deoxyuloses in order to obtain the corresponding  $\alpha$ -oxoketene dithioacetals with a sugar moiety [22-30]. These sugars proved to be very useful consequently in anellation reactions with dinucleophiles.

On this basis, we used 3 $\beta$ -acetoxy-16-[bis(methylthio)methylene]androst-5-en-17-one (**2**) as precursor in a new approach of steroidal heterocycle synthesis consisting in the fusion between nitrogen heterocycles and the steroid ring D of androstenolone acetate (**1**) at C-16 and C-17. Moreover, in all obtained compounds the  $\Delta^5$ -3 $\beta$ -hydroxy function remains unaffected. Therefore, these molecules are interesting analogues of related hormone 4-en-3-one steroids.

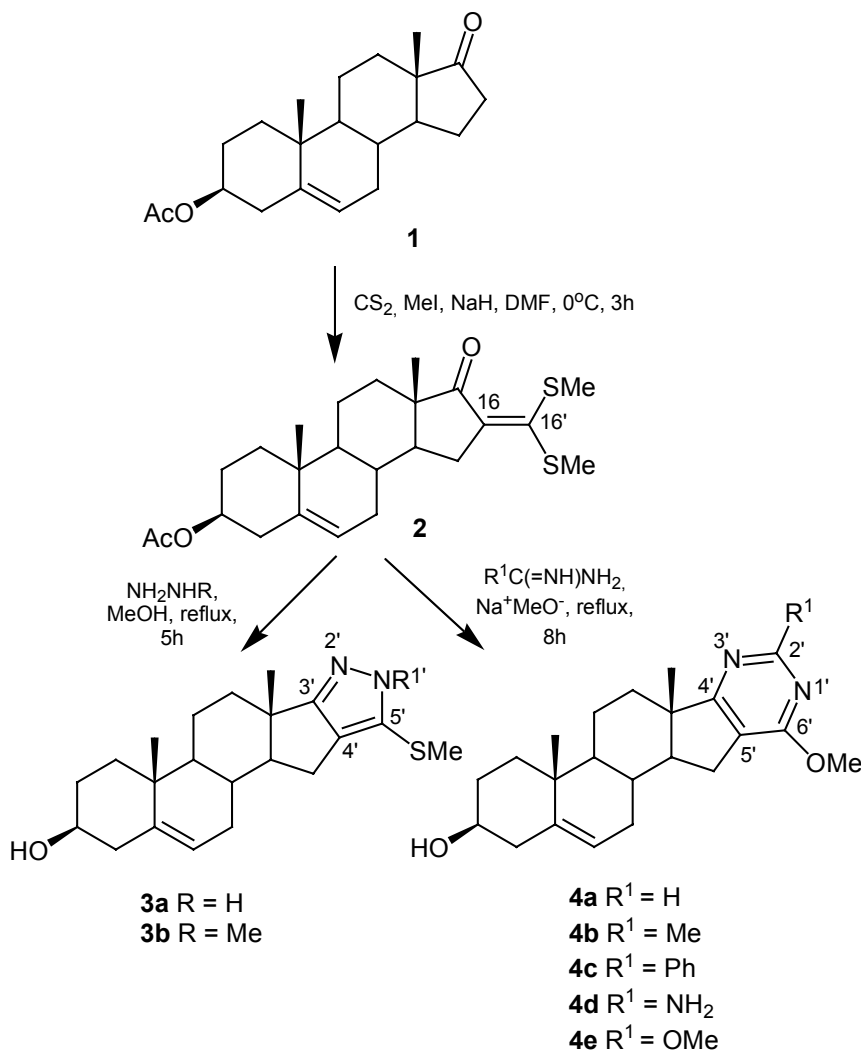
## Results and Discussion

The syntheses started with androstenolone acetate (**1**). This inexpensive compound was chosen because of its synthetic utility and easier accessibility to ring D. Compound **1** was reacted with carbon disulfide, iodomethane and sodium hydride in dimethylformamide as solvent at 0 °C to furnish the corresponding  $\alpha$ -oxoketene dithioacetal **2** in 62 % yield (Scheme 1).

The structure of compound **2** was established by NMR spectra, mass spectral data and elemental analysis. The conclusion from a comparison between the  $^{13}\text{C}$ -NMR data of compounds **1** and **2** was that the appearance of the signal at 149.9 ppm for C-16' and the two corresponding to methylthio groups at  $\delta = 18.5$  and  $\delta = 17.6$  are clear evidence of formation of the desired ketene dithioacetal group. Moreover, the  $^1\text{H}$ -NMR spectrum showed two singlets at  $\delta = 2.45$  and  $\delta = 2.43$  assigned to the methylthio groups, and the singlet belonged to the acetyl group confirming that deprotection of the 3 $\beta$ -hydroxyl did not take place under the reaction conditions. On the other hand, taking in account the

“push-pull” features of **2**, the analysis of its  $^{13}\text{C}$ -NMR spectrum allowed the unequivocal assignment of C-16 at  $\delta = 135.4$  and the signal corresponding to C-16' at  $\delta = 149.9$  [31].

Scheme 1.



With the aim to obtain the pyrazolo anellated androstenediones, compound **2** was allowed to react with 80% hydrazine hydrate and methylhydrazine, both in boiling methanol, affording compounds **3a** and **3b** with yields of 70% and 66%, respectively. The pyrimido-anellated androstenedione derivatives **4** were obtained by reaction of  $\alpha$ -oxoketene dithioacetal **2** with a variety of amidinium, guanidinium and thiuronium salts in dry methanol in the presence of sodium methoxide at reflux. The C-2' substituents in fused pyrimidine rings were selected for their different natures, which allow one to provide the heterocycle moiety with either hydrophobic or hydrophilic features. It can be noted that reaction times for obtaining pyrimidine derivatives are larger than those for pyrazole derivatives; this fact is justified by the known low reactivity of  $\alpha$ -oxoketene dithioacetals toward amidines. Cleavage of acetyl group occurred due to the strong reaction conditions used in the synthesis of both pyrazole and pyrimidine

derivatives. In the second case an additional nucleophilic substitution of a methylthio group occurred because of the high concentration of sodium methoxide required in this synthetic procedure.

Two important spectroscopic features in the  $^1\text{H}$ - and  $^{13}\text{C}$ -NMR spectra of the pyrazole derivatives **3a** and **3b** were the appearance of only one signal, assigned to the methylthio group, and the absence of the classical acetyl signal confirming the expected substitution of one methylthio and the deprotection of the  $3\beta$ -hydroxy group, respectively. In the case of compound **3a** it was not possible to determine which of the two pyrazole tautomers had been formed, although typically an equilibrium exists between them in solution.

Furthermore, as is evident from analysis of  $^{13}\text{C}$ -NMR data of compounds **3** and **4**, chemical shifts of carbon atoms in rings A, B and C have similar values because of their structural differences are located in ring D. The most relevant chemical shifts of steroidal and heterocycle moieties are listed in Table 1.

**Table 1.** Relevant  $^{13}\text{C}$  NMR chemical shifts of compounds **3** and **4**

	<b>3a</b>	<b>3b</b>	<b>4a</b>	<b>4b</b>	<b>4c</b>	<b>4d</b>	<b>4e</b>
C-3	71.5	71.5	71.5	71.5	71.5	71.6	71.5
C-5	141.1	141.2	141.2	141.2	141.2	141.2	141.2
C-6	121.0	121.0	121.0	121.0	121.0	120.9	121.0
C-2'	-	-	169.8	166.5	165.5	165.7	169.8
C-3'	168.1	167.3	-	-	-	-	-
C-4'	123.9	124.9	181.8	181.8	182.4	181.6	181.8
C-5'	131.3	129.3	113.4	115.1	116.2	118.8	113.4
C-6'	-	-	162.8	165.3	163.2	159.9	163.1
C-18	17.9*	18.1*	16.8	16.8	16.9	16.8	16.8
SCH <sub>3</sub>	17.8*	17.9*	-	-	-	-	-
OCH <sub>3</sub>	-	-	53.2	53.3	53.5	53.5	53.6

\* Assignments that could be interchanged.

The loss of the “push-pull” behavior could be certainly corroborated by the  $\delta$  changes at C-16 and C-16' during the formation of these heterocycle rings. In the  $^{13}\text{C}$ -NMR data the shift changes noted at C-2' in the pyrimidine ring of compounds **4** are in agreement with the expected differences. In the  $^1\text{H}$ -NMR spectrum of compound **4e** the lack of the singlet signal for methylthio groups and the appearance of a new singlet at  $\delta = 3.95\text{--}4.00$  which is typical of methoxy groups confirmed the postulated structure.

## Acknowledgments

We are grateful to the Deutscher Akademischer Austauschdienst for scholarships

## Experimental

### General

Melting points were determined on a Boëtius apparatus and are uncorrected. IR spectra were obtained on a Nicolet 205 FT-IR spectrometer.  $^1\text{H-NMR}$  and  $^{13}\text{C-NMR}$  were recorded on a Bruker ACF-250 spectrometer at 250.13 MHz using TMS as internal standard. Chemical shifts are given in ppm and coupling constant ( $J$ ) values are in Hz. The multiplicity of the signals on the  $^{13}\text{C-NMR}$  spectra was determined using the Distorsionless Enhancement Polarization Transfer (DEPT) sequence. Elemental analyses were performed in a Leco CHNS-932 instrument. The mass spectra were recorded in a AMD 402/3 spectrometer AMD Intectra GmbH. Reactions were monitored by thin-layer chromatography (TLC) on precoated plates with silica gel 60 GF<sub>254</sub> (Merck) and spots were visualized with UV light or a spray of vanillin in perchloric acid 1 % w/v and subsequent heating. Flash column chromatography was performed on silica gel 60 (Merck, 70-230 mesh). “Usual work-up” refers to dilution with water, extraction with an organic solvent, washing the extract with HCl (5%) and/or  $\text{KHCO}_3$  (5%) and water, drying over anhydrous  $\text{MgSO}_4$  and removal of the solvent under reduced pressure. The solvents were purified and dried according to recommended procedures.

### *3 $\beta$ -acetoxy-16-[bis(methylthio)methylene]androst-5-en-17-one (2)*

Sodium hydride (60%, 303 mg, 7.6 mmol), carbon disulfide (0.46 mL, 7.6 mmol) and methyl iodide (0.76 mL, 12.1 mmol) were added to a stirred solution of androstenolone acetate **1** (1.000 g, 3.03 mmol) in anhydrous DMF (60 mL). The mixture was stirred for 3 hours at 0 °C, poured into ice-water (200 mL) and extracted with chloroform (3 x 100 mL). The combined extracts were washed with water (3 x 20 mL), dried over anhydrous  $\text{MgSO}_4$  and the solvent was evaporated. The crude product was purified by column chromatography with 10:1 toluene/ethyl acetate elution and crystallized from acetone to furnish **2** (0.8165 g, 62 %) as yellow needles, mp: 165.7–168.0 °C; IR (KBr,  $\text{cm}^{-1}$ ): 1739 (C=O, Ac), 1682 (C=O), 1479 (C=C);  $^1\text{H-NMR}$  ( $\text{CDCl}_3$ ):  $\delta$  = 0.91 (s, 3H, Me-19), 1.04 (s, 3H, Me-18), 2.03 (s, 3H, Ac), 2.44 (s, 3H, SMe), 2.45 (s, 3H, SMe), 4.59 (m, 1H, H-3 $\alpha$ ), 5.40 (d, 1H,  $J$  = 4.9 Hz, H-6);  $^{13}\text{C-NMR}$  ( $\text{CDCl}_3$ ):  $\delta$  = 14.3 (Me-18), 17.6 (SMe), 18.5 (SMe), 19.3 (Me-19), 20.3 (C-11), 21.4 ( $\text{CH}_3\text{CO}$ ), 27.7 (C-2), 30.8 (C-12), 30.9 (C-8), 31.8 (C-7), 32.9 (C-15), 36.7 (C-10), 36.8 (C-1), 38.0 (C-4), 49.7 (C-13), 50.1 (C-14), 50.1 (C-9), 73.7 (C-3), 121.8 (C-6), 135.4 (C-16), 140.0 (C-5), 149.9 (C-16'), 170.5 ( $\text{CH}_3\text{CO}$ ), 205.0 (C-17); MS (EI):  $m/z$  (%): 434 (65.8,  $\text{M}^+$ ), 374 (100); Calc for  $\text{C}_{24}\text{H}_{34}\text{O}_3\text{S}_2$ : C 66.32, H 7.88, S 14.75; Found: C 66.32, H 7.99, S 14.67.

### *5'-Methylthio-pyrazolo[4',3':16,17]androst-5-en-3 $\beta$ -ol (3a)*

Compound **2** (200 mg, 0.46 mmol) was dissolved in anhydrous methanol (50 mL), hydrazine hydrate (80%, 0.8 mL) was added and the mixture was stirred at reflux for 4 hours until all the starting

material had disappeared as indicated by TLC (disappearance of yellow color). Then the reaction mixture was concentrated and usual work-up ( $\text{CHCl}_3$ ) yielded a crude product, which was purified by column chromatography with 3:1 chloroform/ethyl acetate elution to give 117.8 mg (66 %) of **3a**, mp (ethanol): 176.2–178.0 °C; IR (KBr,  $\text{cm}^{-1}$ ): 3354 (OH), 3168 (NH), 1454, 1436, 1408, 1373 (C=C, C=N);  $^1\text{H-NMR}$  ( $\text{CDCl}_3$ ):  $\delta$  = 1.02 (s, 3H, Me-19), 1.07 (s, 3H, Me-18), 2.42 (s, 3H, SMe), 3.53 (m, 1H, H-3 $\alpha$ ), 4.27 (broad signal, NH), 5.38 (d, 1H,  $J$  = 5.20 Hz, H-6);  $^{13}\text{C-NMR}$  ( $\text{CDCl}_3$ ):  $\delta$  = 17.8 (SMe), 17.9 (Me-18), 19.4 (Me-19), 20.4 (C-11), 30.3 (C-12), 30.8 (C-8), 31.5 (C-7), 32.1 (C-2), 33.7 (C-15), 36.7 (C-10), 37.1 (C-1), 40.9 (C-13), 42.2 (C-4), 50.5 (C-9), 62.1 (C-14), 71.5 (C-3), 121.0 (C-6), 123.9 (C-4'), 131.3 (C-5'), 141.1 (C-5), 168.1 (C-3'); MS (EI):  $m/z$  (%): 358 (21.7,  $\text{M}^+$ ), 287 (100); Calc. for  $\text{C}_{21}\text{H}_{30}\text{N}_2\text{OS}$ : C 70.35, H 8.43, N 7.81, S 8.94; Found: C 70.08, H 8.69, N 8.02, S 8.78.

#### *1'-Methyl-5'-methylthio- pyrazolo[4',3':16,17]androst-5-en-3 $\beta$ -ol (3b)*

Compound **2** (200 mg, 0.46 mmol) was dissolved in anhydrous methanol (50 mL), methyl hydrazine (1 mL) was added and the mixture was stirred at reflux for 5 hours until all the starting material had disappeared as indicated by TLC. Then the reaction mixture was concentrated and usual work-up ( $\text{CHCl}_3$ ) yielded a crude product, which was purified by column chromatography with 4:1 chloroform/ethyl acetate elution to afford 129.4 mg (70 %) of **3b**, mp (ethanol): 181–183 °C; IR (KBr,  $\text{cm}^{-1}$ ): 3349 (OH), 1455, 1371, 1356, 1349 (C=C, C=N);  $^1\text{H-NMR}$  ( $\text{CDCl}_3$ ):  $\delta$  = 1.00 (s, Me-19), 1.06 (s, Me-18), 2.36 (s, 3H, SMe), 3.53 (m, 1H, H-3 $\alpha$ ), 3.80 (s, 3H, NMe), 5.38 (d, 1H,  $J$  = 4.9 Hz, H-6);  $^{13}\text{C-NMR}$  ( $\text{CDCl}_3$ ):  $\delta$  = 17.9 (SMe), 18.1 (Me-18), 19.3 (Me-19), 20.5 (C-11), 30.3 (C-12), 30.8 (C-8), 31.5 (C-7), 32.1 (C-2), 33.7 (C-15), 36.3 (NMe), 36.7 (C-10), 37.1 (C-1), 40.8 (C-13), 42.2 (C-4), 50.5 (C-9), 61.8 (C-14), 71.5 (C-3), 121.0 (C-6), 124.9 (C-4'), 129.3 (C-5'), 141.2 (C-5), 167.3 (C-3'); MS (EI):  $m/z$  (%): 372 (38.6,  $\text{M}^+$ ), 327 (100); Calc. for  $\text{C}_{22}\text{H}_{32}\text{N}_2\text{OS}$ : C 70.92, H 8.66, N 7.52, S 8.61; Found: C 70.58, H 8.74, N 7.53, S 8.49.

#### *6'-Methoxy- pyrimido[5',4':16,17]androst-5-en-3 $\beta$ -ol (4a)*

Compound **2** (100 mg, 0.23 mmol) was dissolved in a solution of sodium (26.5 mg, 1.15 mmol) in anhydrous methanol (50 mL), formamidinium acetate (120 mg, 1.15 mmol) was added and the mixture was stirred at reflux for 8 hours. Usual work-up ( $\text{CHCl}_3$ ) yielded a crude product, which was purified by column chromatography to furnish 50.74 mg (68 %) of **4a**, mp (acetone): 178–180 °C; IR (KBr,  $\text{cm}^{-1}$ ): 3450 (OH), 1593, 1558 (C=C, C=N);  $^1\text{H-NMR}$  ( $\text{CDCl}_3$ ):  $\delta$  = 0.98 (s, 3H, Me-19), 1.08 (s, 3H, Me-18), 3.53 (m, 1H, H-3 $\alpha$ ), 3.98 (s, 3H, OMe), 5.38 (d, 1H,  $J$  = 5.20 Hz, H-6), 8.60 (s, 1H, H-2');  $^{13}\text{C-NMR}$  ( $\text{CDCl}_3$ ):  $\delta$  = 16.8 (Me-18), 19.4 (Me-19), 20.5 (C-11), 26.5 (C-15), 30.6 (C-8), 31.6 (C-7), 31.3 (C-12), 32.9 (C-2), 36.8 (C-10), 37.1 (C-1), 42.2 (C-4), 46.0 (C-13), 50.6 (C-9), 53.5 (MeO), 55.6 (C-14), 71.6 (C-3), 118.8 (C-5'), 121.0 (C-6), 141.2 (C-5), 159.9 (C-6'), 165.7 (C-2'), 181.6 (C-4'); MS (EI):  $m/z$  (%): 354 (74.22,  $\text{M}^+$ ), 243 (100); Calc. for  $\text{C}_{22}\text{H}_{30}\text{N}_2\text{O}_2$ : C 74.54, H 8.53, N 7.90; Found: C 74.28, H 8.31, N 7.68.

*6'-Methoxy-2'-methyl- pyrimido[5',4':16,17]androst-5-en-3β-ol (4b)*

Compound **2** (100 mg, 0.23 mmol), sodium (26.5 mg, 1.15 mmol) in anhydrous methanol (50 mL) and acetamidinium hydrogen chloride (110 mg, 1.15 mmol) was reacted as described for preparation of **4a** to furnish 58.4 mg (75 %) of **4b**, mp (acetone): 210–212 °C; IR (KBr, cm<sup>-1</sup>): 3363 (OH), 1591, 1565 (C=C, C=N); <sup>1</sup>H-NMR (CDCl<sub>3</sub>): δ = 0.96 (s, Me-19), 1.07 (s, Me-18), 2.60 (s, Me-2'), 3.53 (m, 1H, H-3α), 3.95 (s, OMe), 5.38 (d, 1H, *J* = 5.20 Hz, H-6,); <sup>13</sup>C-NMR (CDCl<sub>3</sub>): δ = 16.8 (Me-18), 19.4 (Me-19), 20.5 (C-11), 25.8 (Me-2'), 26.1 (C-15), 30.6 (C-8), 31.2 (C-12), 31.6 (C-7), 32.9 (C-2), 36.7 (C-10), 37.1 (C-1), 42.2 (C-4), 46.0 (C-13), 50.6 (C-9), 53.2 (MeO), 55.7 (C-14), 71.5 (C-3), 115.1 (C-5'), 121.0 (C-6), 141.2 (C-5), 165.3 (C-6'), 166.5 (C-2'), 181.8 (C-4'); MS (EI): *m/z* (%): 368 (76.95, M<sup>+</sup>), 353 (100); Calc. for C<sub>23</sub>H<sub>32</sub>N<sub>2</sub>O<sub>2</sub>: C 74.96, H 8.75, N 7.60; Found: C 74.45, H 8.62, N 7.33.

*6'-Methoxy-2'-phenyl- pyrimido[5',4':16,17-c]androst-5-en-3β-ol (4c)*

Compound **2** (100 mg, 0.23 mmol), sodium (26.5 mg, 1.15 mmol) in anhydrous methanol (50 mL) and benzamidinium hydrogen chloride (180 mg, 1.15 mmol) was reacted as described for preparation of **4a** to furnish 57.12 mg (62 %) of **4c**, mp (acetone): 241–243 °C; IR (KBr, cm<sup>-1</sup>): 3396 (OH), 1593, 1552 (C=C, C=N); <sup>1</sup>H-NMR (CDCl<sub>3</sub>): δ = 1.02 (s, Me-19), 1.10 (s, Me-18), 3.54 (m, 1H, H-3α), 4.09 (s, OMe), 5.38 (d, 1H, *J* = 5.20 Hz, H-6), 7.43–7.46 (m, 3H, Ph), 8.43–8.47 (m, 2H, Ph); <sup>13</sup>C-NMR (CDCl<sub>3</sub>): δ = 16.9 (C-18), 19.4 (C-19), 20.5 (C-11), 26.4 (C-15), 30.7 (C-8), 31.3 (C-12), 31.6 (C-7), 33.1 (C-2), 36.8 (C-10), 37.1 (C-1), 42.3 (C-4), 46.1 (C-13), 50.8 (C-9), 53.3 (MeO), 55.8 (C-14), 71.6 (C-3), 116.2 (C-5'), 121.1 (C-6), 128.2, 128.3, 130.0, 138.3 (Ph), 141.2 (C-5), 163.2 (C-6'), 165.5 (C-2'), 182.4 (C-4'); MS (EI): *m/z* (%): 430 (100, M<sup>+</sup>); Calc. for C<sub>28</sub>H<sub>34</sub>N<sub>2</sub>O<sub>2</sub>: C 78.10, H 7.96, N 6.51; Found: C 77.93, H 7.82, N 6.34.

*2'-Amino-6'-methoxy- pyrimido[5',4':16,17]androst-5-en-3β-ol (4d)*

Compound **2** (100 mg, 0.23 mmol), sodium (26.5 mg, 1.15 mmol) in anhydrous methanol (50 mL) and guanidinium nitrate (140 mg, 1.15 mmol) was reacted as described for preparation of **4a** to furnish 49.98 mg (64 %) of **4d**, mp (acetone): 218–221 °C; IR (KBr, cm<sup>-1</sup>): 3472 (OH), 3318, 3209 (NH<sub>2</sub>), 1602, 1568 (C=C, C=N); <sup>1</sup>H-NMR (CDCl<sub>3</sub>): δ = 1.00 (s, Me-19), 1.06 (s, Me-18), 3.06 (s, 2H, NH<sub>2</sub>), 3.53 (m, 1H, H-3α), 5.38 (d, 1H, H-6, *J* = 4.9 Hz); <sup>13</sup>C-NMR (CDCl<sub>3</sub>): δ = 16.8 (C-18), 19.4 (C-19), 20.5 (C-11), 26.1 (C-15), 30.6 (C-8), 31.2 (C-12), 31.6 (C-7), 32.9 (C-2), 36.7 (C-10), 37.1 (C-1), 42.2 (C-4), 46.0 (C-13), 50.6 (C-9), 53.2 (MeO), 55.7 (C-14), 71.5 (C-3), 113.4 (C-5'), 121.0 (C-6), 141.2 (C-5), 162.8 (C-6'), 169.8 (C-2'), 181.8 (C-4'); MS (EI): *m/z* (%): 369 (100, M<sup>+</sup>); Calc. for C<sub>22</sub>H<sub>31</sub>N<sub>3</sub>O<sub>2</sub>: C 71.51, H 8.46, N 11.37; Found: C 71.38, H 8.64, N 11.13.

*2',6'-dimethoxy-pyrimido[5',4':16,17]androst-5-en-3 $\beta$ -ol (4e)*

Compound **2** (100 mg, 0.23 mmol), sodium (26.5 mg, 1.15 mmol) in anhydrous methanol (50 mL) and *S*-methylisothiuronium sulfate (320 mg, 1.15 mmol) was reacted as described for preparation of **4a**, to furnish 52.11 mg (59 %) of **4e**, mp (acetone): 214–216° C; IR (KBr, cm<sup>-1</sup>): 3463 (OH), 1591, 1559 (C=C, C=N); <sup>1</sup>H-NMR (CDCl<sub>3</sub>):  $\delta$  = 0.96 (s, Me-19), 1.07 (s, 3 Me-18), 3.53 (m, 1H, H-3 $\alpha$ ), 3.95 (s, OMe), 4.08 (s, OMe), 5.38 (d, 1H, *J* = 5.20 Hz, H-6); <sup>13</sup>C-NMR (CDCl<sub>3</sub>):  $\delta$  = 16.8 (C-18), 19.4 (C-19), 20.5 (C-11), 26.4 (C-15), 30.7 (C-8), 31.3 (C-12), 31.6 (C-7), 33.1 (C-2), 36.8 (C-10), 37.1 (C-1), 42.3 (C-4), 46.1 (C-13), 50.8 (C-9), 53.2 (MeO), 53.6 (MeO), 55.8 (C-14), 71.6 (C-3), 113.4 (C-5'), 121.1 (C-6), 141.2 (C-5), 169.8 (C-2'), 181.8 (C-4'), 163.1 (C-6'); MS (EI): *m/z* (%): 384 (100 M<sup>+</sup>); Calc. for C<sub>23</sub>H<sub>32</sub>N<sub>2</sub>O<sub>3</sub>: C 71.84, H 8.39, N 7.29; Found: C 71.67, H 8.44, N 7.18.

**References**

1. Newman, H. C.; Potts, G. O.; Ryan, W. T.; Stonner, F. W. *J. Med. Chem.* **1970**, *13*, 948–951.
2. Newman, H. C. *J. Med. Chem.* **1971**, *14*, 1246–1246.
3. Newman, H. C.; Stonner, F. W. *Helv. Chim. Acta* **1972**, *55*, 2014–2017.
4. Campbell, M. M.; Craig, R. C. *J. Chem. Soc. Perkin Trans. I* **1980**, 766–774.
5. Bajwa, J. S.; Sykes, P. J. *J. Chem. Soc. Perkin Trans. I* **1980**, 1019–1024.
6. Ramadas, S. R.; Krishna, M. V. *Indian. J. Chem. Sect. B* **1983**, 1093–1095.
7. Bajwa, J. S.; Sykes, P. J. *J. Chem. Soc. Perkin Trans. I* **1980**, 481–486.
8. Marini-Bettolo, R.; Tsai, C. S.; Tsai, T. Y.; Wiesner, K. *Heterocycles* **1981**, *15*, 305–308.
9. Ramadas, S. R.; Krishna, M. V. *Heterocycles* **1981**, *16*, 2169–2171.
10. Singh, H.; Yadav, M. R.; Garg, S. P.; Sharma, R.; Paul, D. *Heterocycles* **1985**, *23*, 2931–2938.
11. Pradhan, S. K.; Akamanchi, K. G. *Heterocycles* **1989**, *28*, 813–839.
12. Charalambos, C. *J. Heterocycl. Chem.* **1996**, *33*, 539–558.
13. Shamsuzzaman, A. S.; Akram, M. K. *Heterocycles* **1998**, *48*, 329–334.
14. Li, S.; Parish, E. J.; Rodriguez-Valenzuela, C.; Brodie, A. M. H. *Bioorg. Med. Chem.* **1998**, *6*, 1525–1529.
15. Fink, B. E.; Mortensen, D. S.; Stauffer, S. R.; Aron, Z. D.; Katzenellenbogen, J. A. *Chem. Biol.* **1999**, *6*, 205–219.
16. Grandolini, G.; Ambrogi, V.; Perioli, L. *Farmaco* **1996**, *51*, 203–207.
17. Dieter, R. K. *Tetrahedron* **1986**, *42*, 3029–3096.
18. Tominaga, Y. *J. Heterocycl. Chem.* **1989**, *26*, 1167–1204.
19. Borrmann, D. In *Houben-Weyl, Methoden der Organischen Chemie*; Müller, E., Ed.; Thieme: Stuttgart, 1969; Band VII/4, p. 404.
20. Schaumann, E. In *Houben-Weyl, Methoden der Organischen Chemie*; Klamann, D., Ed.; Thieme: Stuttgart, New York, 1985; Band E 11, p. 232.
21. Peseke, K.; Feist, H.; Quincoces, J. *Targets in Heterocyclic Systems – Chemistry and Properties*, Vol. 5; Attanasi, O.A., ed; Societa Chimica Italiana: Rome, 2001; pp. 299–326.



22. Peseke, K.; Feist, H.; Cuny, E. *Carbohydr. Res.* **1992**, *230*, 319–325.
23. Peseke, K.; Feist, H.; Köll, P. *Carbohydr. Res.* **1993**, *247*, 315–322.
24. Gómez, M.; Quincoces, J.; Peseke, K.; Reinke, H. *J. Carbohydr. Chem.* **1999**, *18*, 57–68.
25. Peseke, K.; Thiele, G.; Michalik, M. *Liebigs Ann.* **1995**, 1633–1636.
26. Gómez, M.; Quincoces, J.; Kuhla, B.; Peseke, K.; Reinke, H. *J. Carbohydr. Chem.* **1998**, *17*, 57–68.
27. Methling, K.; Aldinger, S.; Peseke, K.; Michalik, M. *J. Carbohydr. Chem.* **1999**, *18*, 429–439.
28. Kuhla, B.; Peseke, K.; Thiele, G.; Michalik, M. *J. Prakt. Chem.* **2000**, *342*, 240–244.
29. Gómez, M.; Peseke, K.; Reinke, H.; Quincoces, J.; Michalik, M. *J. Prakt. Chem.* **2000**, *342*, 389–395.
30. Herrera, L.; Feist, H.; Michalik, M.; Quincoces, J.; Peseke, K. *Carbohydr. Res.* **2003**, *338*, 293–298
31. Michalik, M.; Peseke, K.; Radeglia, R. *J. Prakt. Chem.* **1985**, *327*, 103–108.

*Sample availability:* Available from MDPI

© 2003 by MDPI (<http://www.mdpi.org>). Reproduction is permitted for non commercial purposes.