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Systematic optimization of quantum junction colloidal quantum dot solar cells

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The recently reported quantum junction architecture represents a promising approach to building a rectifying photovoltaic device that employs colloidal quantum dot layers on each side of the p-n junction. Here, we report an optimized quantum junction solar cell that leverages an improved aluminum zinc oxide electrode for a stable contact to the n-side of the quantum junction and silver doping of the p-layer that greatly enhances the photocurrent by expanding the depletion region in the n-side of the device. These improvements result in greater stability and a power conversion efficiency of 6.1% under AM1.5 simulated solar illumination. © 2012 American Institute of Physics. [<http://dx.doi.org/10.1063/1.4757866>]

Colloidal quantum dots (CQDs) are attractive materials for inexpensive, room temperature and solution processed optoelectronics. These materials have been employed in light emitting devices,¹ field effect transistors (FETs),^{2,3} and photovoltaics.⁴⁻⁷

In photovoltaics, CQD films are typically combined with a wide-bandgap bulk semiconductor which allows for the formation of a p-n junction. This limits the benefits of bandgap tuning in CQD films, since redesign of the bulk semiconductor is required to maintain performance due to band alignment.⁸ The quantum junction (QJ) architecture⁹ has obviated the need for bulk semiconductor materials, and has enabled an efficient implementation of a p-n junction where both sides of the junction were capable of being quantum-tuned. This has opened a path towards all-room temperature processing from solution of the entire junction, obviating the need for the high-temperature sintering steps required when working with many relevant metal-oxide electrodes.

The QJ presents an attractive approach to building junctions of varying quantum dot sizes and thus bandgaps. To fulfil its potential in single-junction devices, the materials and the architecture need to be optimized for a bandgap range centered at 1.3 eV.¹⁰ We took the view that, through careful engineering of electrode materials, and by tuning the CQD properties, we could achieve greater power conversion efficiencies. Prior reports of QJ devices have relied on an indium tin oxide (ITO) transparent contact; an overlying, highly doped PbS CQD p-layer; a lightly doped, PbS CQD n-layer; the latter contacted using a final Al ohmic metal contact.⁹

CQD films typically have low diffusion lengths, in the range from 10–100 nm.^{11,12} As a result, extending the deple-

tion region as far as possible into a lightly doped n-layer is expected to enhance charge collection and thereby increase the photocurrent. This can be achieved if the doping in the p-layer can be considerably increased.

With this goal in mind, we investigated related material systems in which dopants have been introduced during synthesis. Ag ions have been used in doping InAs CQDs.¹³ The valency of Ag is suited to p-doping of PbS CQDs as well, providing one extra hole for each Pb atom that is replaced by Ag in the PbS lattice.

We therefore explored the addition of Ag during PbS CQDs synthesis. PbS CQDs of various sizes are fabricated following a modified published recipe.^{14,15} For Ag incorporation, the Ag precursor solution was prepared by dissolving 144 mg of AgCl in 10 ml of oleylamine (OLA). After the injection of bis(trimethylsilyl) sul-fide (TMS) for the synthesis of CQDs, the heating mantle was turned off to let the flask temperature decrease to ~75 °C. 1.5 ml of the Ag precursor and 3 ml of OLA were then injected and the flask temperature was maintained at 75 °C for 30 min, after which it was allowed to cool to 36 °C. The procedure results in CQDs having Ag impurities incorporated, as verified using x-ray photoelectron spectroscopy (XPS) (S1).²¹ Doping densities were obtained using FET measurements, employing a SiO₂/Si architecture with Au drain and source contacts, where a 40 nm film of CQDs treated with tetramethyl ammonium hydroxide in methanol formed the channel. The extracted doping densities (Figure 1(a) inset) confirmed that the doping in the p-layer had increased substantially as a result of the Ag incorporation. The Ag doping approach resulted in doping densities approaching 10¹⁹ cm⁻³ found to be optimal for device performance.

We then fabricated p-n devices using a layer-by-layer approach,⁹ with a 300 nm n-layer processed in an inert environment with tetrabutylammonium iodide ligands in methanol (10 mg/ml) on a 50 nm air ambient processed p-layer, with tetramethyl ammonium hydroxide in methanol (10 mg/ml) as

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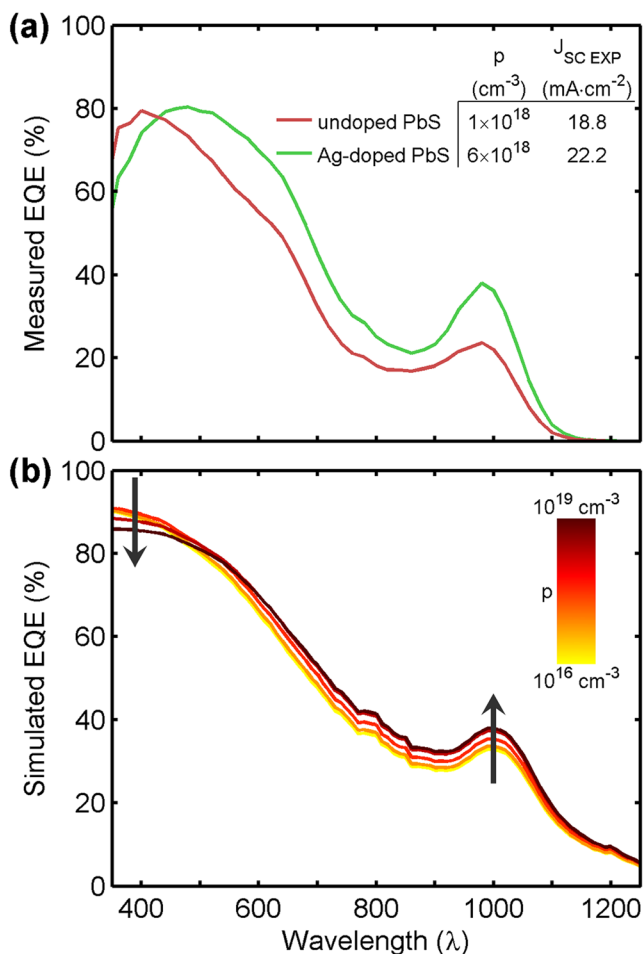


FIG. 1. (a) EQE of devices employing undoped and Ag-doped CQDs showing that Ag doping leads to a greater depletion in the n-layer, inset: hole density (p) as measured by FETs for films made with undoped and Ag-doped PbS CQDs; expected J_{SC} shown as calculated from (a). (b) Simulated EQE curves showing an enhancement in the near infrared and across part of the visible, while a slight reduction around 400 nm due to decreased depletion in the p-layer.

the ligand. A thin, highly doped p-layer, and a thicker, lightly doped n-layer ensure high absorption and depletion in the n-layer, excellent for charge generation and collection. External quantum efficiency (EQE) measurements revealed the benefit from increased doping in the p-layer as a result of Ag incorpo-

ration. The incident spectrum for EQE measurements was generated by passing the output of a 400 W Xe lamp through a monochromator and using appropriate order-sorting filters. The resultant monochromatic beam also passed through an optical chopper operating at 220 Hz coupled to the input of a lock-in amplifier. The EQE (Figure 1(a)) was greatly enhanced across most of the visible spectrum, and markedly so at the infrared excitonic peak where absorption occurs deeper into the device compared to the case of visible-light absorption. The EQE showed a slight decrease at shorter wavelengths, consistent with a decrease in the depletion depth inside the front-side p-layer due to its increased doping.

To explore further this picture of doping and its impact on depletion in the quantum junction, we constructed an optoelectronic model using previously established approaches.¹⁶⁻¹⁸ The resulting EQE curves (Figure 1(b)) demonstrated that, as the doping in the p-layer is increased, the EQE improves across most of the visible and infrared spectrum. Meanwhile, the EQE is reduced at high-energy wavelengths, a fact attributable to the decrease in the depletion region on that side of the junction. The EQE curves further suggest that the overall improvement in EQE across the entire spectrum outweighs any reduction in EQE due to less depletion in the p-layer, also consistent with experimental results.

For a device to be attractive for photovoltaics, it would also need to have excellent long term stability and temperature resilience. The presently employed Al contact is known to react with halides¹⁹ that are used during the fabrication of the n-type film. A highly doped n-type aluminum zinc oxide (AZO) layer could therefore potentially serve to enhance stability while improving current collection by blocking hole recombination (Figure 2(a)). AZO is readily sputtered at room temperature,^{20,21} and a 30 nm contact was deposited at a rate of 0.4 Å/s by sputtering, followed by thermal evaporation of silver (200 nm thick) at a rate of 1.5 Å/s with base pressure of 1×10^{-7} mbar. Figure 2(b) shows the temporal stability of our AZO devices. These devices were encapsulated in a glovebox and tested in an air ambient environment. Encapsulation was done by sealing the device in an epoxy (NOA88, Norland Products Inc), and then curing with an 8 W UV lamp (UVLMS38, UVP Inc.) for 30 min. The devices retained performance for over 60 h under continuous

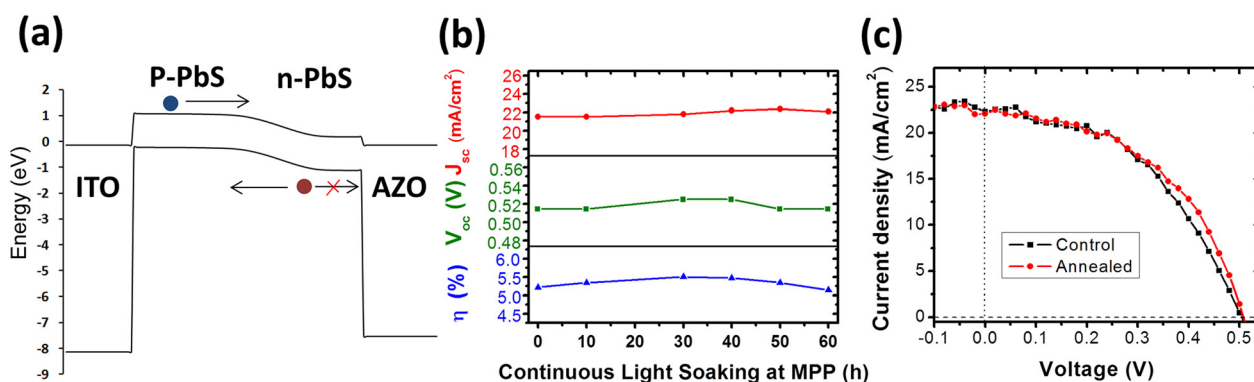


FIG. 2. (a) Energy band diagrams of device employing an AZO electrode in place of an Al contact. X-axis is not to scale. The AZO also serves as a hole blocking layer. (b) Stability of a >5% devices across all performance metrics for 60 h (c) Temperature stability of a QJ device after annealing at 90 °C for 3 h, showing little change in JV characteristics.

illumination. The same devices also exhibited temperature stability to 90 °C (Figure 2(c)), highlighting the robustness of the AZO approach.

Current-voltage characteristics were measured using a Keithley 2400 source-meter in N₂ ambient. The solar spectrum at AM1.5 was simulated with a Xe lamp and filters (Solar Light Company Inc.) with measured intensity at 100 mW cm⁻². Combining the silver doping and AZO contact optimizations, we were able to improve power conversion efficiency, principally by collecting more photocurrent. These advances resulted in devices exhibiting a PCE of 6.1%. Figure 3(a) shows the J-V curve of a champion device, with an open circuit voltage of 0.52 V, short circuit current (J_{SC}) of 23.3 mA/cm², and fill factor (FF) of 50%; where most of the improvement is reflected in the J_{SC} and FF compared with previous reports.⁹ This is consistent with an enhanced depletion region and thus more efficient carrier collection, furthermore aided by hole blocking afforded by using an AZO contact. We constructed sixty QJ devices (Fig. 3(b)) over the course of one month using different batches of PbS CQDs to test the reproducibility of fabrication approach. The average efficiency was 5.6% and the standard deviation was 0.2%. Such a high reproducibility compares favourably with performance variation in previously reported depleted heterojunction devices.⁶

We have shown that by engineering the CQD material through doping and by appropriately choosing ohmic electrodes to our CQD films, we are able to collect more current

and increase efficiency. Furthermore, the choice of the ohmic contact can have a drastic impact on device stability by limiting unwanted chemical reactions and forming a favourable energy alignment for further enhancement in charge collection. We achieve highly reproducible, high performance devices with excellent temperature and temporal stability. This work opens a path towards further optimization of related QJ devices by appropriately tuning the doping and choice of electrical contacts to guarantee efficient charge collection and high efficiency.

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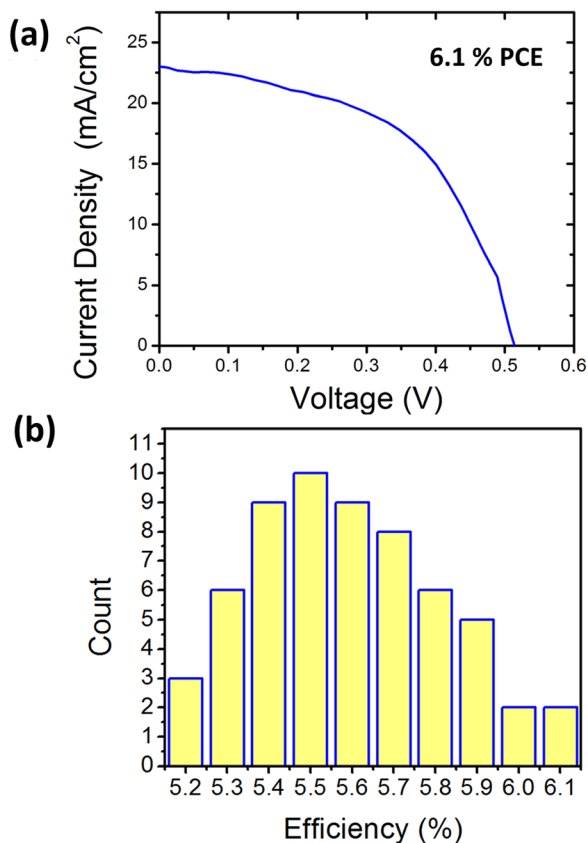


FIG. 3. (a) J-V curve of best performing device showing a PCE of 6.1%. (b) Performance of 60 devices made over 1 month, showing high reproducibility and little deviation in PCE (PCE = 5.6 ± 0.2%).

¹L. Sun, J. J. Choi, D. Stachnik, A. C. Bartnik, B.-R. Hyun, G. G. Malliaras, T. Hanrath, and F. W. Wise, *Nat. Nanotechnol.* **7**, 369 (2012).

²T. P. Osedach, N. Zhao, T. L. Andrew, P. R. Brown, D. D. Wanger, D. B. Strasfeld, L.-Y. Chang, M. G. Bawendi, and V. Bulović, *ACS Nano* **6**, 3121 (2012).

³D. V. Talapin and C. B. Murray, *Science* **310**, 86 (2005).

⁴A. G. Pattantyus-Abraham, I. J. Kramer, A. R. Barkhouse, X. Wang, G. Konstantatos, R. Debnath, L. Levina, I. Raabe, M. K. Nazeeruddin, M. Grätzel, and E. H. Sargent, *ACS Nano* **4**, 3374 (2010).

⁵L. Etgar, W. Zhang, S. Gabriel, S. G. Hickey, M. K. Nazeeruddin, A. Eychmüller, B. Liu, and M. Grätzel, *Adv. Mater.* **24**, 2202 (2012).

⁶J. Tang, K. W. Kemp, S. Hoogland, K. S. Jeong, H. Liu, L. Levina, M. Furukawa, X. Wang, R. Debnath, D. Cha, K. W. Chou, A. Fischer, A. Amassian, J. B. Asbury, and E. H. Sargent, *Nature Mater.* **10**, 765 (2011).

⁷B.-R. Hyun, Y.-W. Zhong, A. C. Bartnik, L. Sun, H. D. Abruña, F. W. Wise, J. D. Goodreau, J. R. Matthews, T. M. Leslie, and N. F. Borrelli, *ACS Nano* **2**, 2206 (2008).

⁸H. Liu, J. Tang, I. J. Kramer, R. Debnath, G. I. Koleilat, X. Wang, A. Fisher, R. Li, L. Brzozowski, L. Levina, and E. H. Sargent, *Adv. Mater.* **23**, 3832 (2011).

⁹J. Tang, H. Liu, D. Zhitomirsky, S. Hoogland, X. Wang, M. Furukawa, L. Levina, and E. H. Sargent, *Nano Lett.* **12**, 4889 (2012).

¹⁰C. H. Henry, *J. Appl. Phys.* **51**, 4494 (1980).

¹¹J. M. Luther, M. Law, M. C. Beard, Q. Song, M. O. Reese, R. J. Ellingson, and A. J. Nozik, *Nano Lett.* **8**, 3488 (2008).

¹²K. W. Johnston, A. G. Pattantyus-Abraham, J. P. Clifford, S. H. Myrskog, S. Hoogland, H. Shukla, E. J. D. Klem, L. Levina, and E. H. Sargent, *Appl. Phys. Lett.* **92**, 122111 (2008).

¹³D. Mocatta, G. Cohen, J. Schattner, O. Mollo, E. Rabani, and U. Banin, *Science* **332**, 77 (2011).

¹⁴M. A. Hines and G. D. Scholes, *Adv. Mater.* **15**, 1844 (2003).

¹⁵J. Tang, L. Brzozowski, D. A. R. Barkhouse, X. Wang, R. Debnath, R. Wolowiec, E. Palmiano, L. Levina, A. G. Pattantyus-Abraham, D. Jankosmanovic, and E. H. Sargent, *ACS Nano* **4**, 869 (2010).

¹⁶D. Zhitomirsky, I. J. Kramer, A. J. Labelle, A. Fischer, R. Debnath, J. Pan, O. M. Bakr, and E. H. Sargent, *Nano Lett.* **12**, 1007 (2012).

¹⁷I. J. Kramer, L. Levina, R. Debnath, D. Zhitomirsky, and E. H. Sargent, *Nano Lett.* **11**, 3701 (2011).

¹⁸I. J. Kramer and E. H. Sargent, *ACS Nano* **5**, 8506 (2011).

¹⁹J. G. Chen, T. P. Beebe, J. E. Crowell, and J. T. Yates, *J. Am. Chem. Soc.* **109**, 1726 (1987).

²⁰G. I. Koleilat, X. Wang, A. J. Labelle, A. H. Ip, G. H. Carey, A. Fischer, L. Levina, L. Brzozowski, and E. H. Sargent, *Nano Lett.* **11**, 5173 (2011).

²¹See supplementary material at <http://dx.doi.org/10.1063/1.4757866> for XPS spectra and composition analysis of Ag-doped PbS CQDs.