

Thallium Concentration in Lake Trout from Lake Michigan

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Received: 8 November 2000/Accepted: 7 August 2001

Chronic poisoning by thallium appears to be a phenomenon associated with growing environmental pollution (Malbrain et al. 1997; Chandler et al. 1990; Zhou and Liu 1985). The toxicity of thallium to mammals is similar to that of mercury but higher than that of lead, cadmium, copper and zinc (Manzo et al. 1989; Wallwork-Baber et al. 1985; Zitko 1975). Thallium compounds are remarkably toxic to human beings, the lethal dose ranging from 0.5 to 1 g for a single ingestion (Mulkey et al. 1993; Manzo et al. 1989; Zitko 1975). The concentrations of total dissolved thallium in waters of the Great Lakes exceed those of mercury (Lin et al. 1999; Mason et al. 1997), and the little available information suggests that the levels in aquatic biota are much higher than those of lead, cadmium and mercury (Lin and Nriagu 1999). There are currently few reliable data on thallium levels in Great Lakes fish (Heit et al. 1981; 1985). This report presents the data (whole fish) for samples of lake trout from Lake Michigan. The results are used to estimate the dietary intake and assess the potential risk for people who consume large quantities of fish from the Great Lakes ecosystems.

MATERIALS AND METHODS

This study used a Milli-Q Plus water system (Millipore Corp., Bedford, MA) to generate ultrapure water for the experiments. Nitric and hydrochloric acids (Fisher Scientific) of two different grades, technical and trace metal-free, were employed in the experiments. Technical grade acid was only employed in the labware cleaning processes while trace metal-free grade acid was utilized in both the final rinsing of the labware and in preparing all the standard solutions. In order to minimize possible contamination, all labware was cleaned following a nine-step procedure involving acetone soaking and multiple acid leaching described by Nriagu et al. (1993). A stock solution of Tl(I) (1000 mg/L) was purchased from Perkin-Elmer to prepare the appropriate concentrations of Tl standard solutions. Lake trout samples were collected from three areas (Sheboygan, Sturgeon and Saugatuck) in Lake Michigan during the Spring, Summer and Fall of 1994 and 1995. The age of each fish was determined by counting the year ring on the fish scales. Whole fish samples were ground and stored in glass jars at 4 °C until analyzed. Aliquots of the fish samples were digested with 10 mL of trace-metal free grade concentrated nitric acid in microwave digestion bombs whose operating conditions were controlled as follows: Heating/digestion time: 25 minutes; temperature: < 108 °C; pressure, 100 atmospheres. The leachate was filtered if it was cloudy or had some residual

Table 1. Thallium content in lake trout (wet weight, n=37)

Sample ID	Fish Age (year)	Weight(g)	Growth Factor	Tl (ng/g)
940239003	12	5124	427.0	69.9
940239004	9	3474	386.0	276.0
940239008	6	1031	171.8	231.2
940239010	6	1307	217.8	127.0
940239012	4	455	113.8	496.9
940239014	4	373	93.3	83.5
940239015	5	836	167.2	45.0
940239019	12	4296	358.0	231.7
940239043	4	475	118.8	63.5
940248014	4	1043	260.8	84.1
940248015	4	892	223.0	210.8
940248041	14	5333	380.9	10.7
940248044	10	4939	493.9	95.0
940248050	6	2484	414.0	44.1
940248053	4	997	249.3	110.6
940248057	2	211	105.5	119.3
940248058	13	6218	478.3	229.8
940248061	8	4343	542.9	27.1
940248067	6	3364	560.7	325.3
940248071	3	951	317.0	274.1
940248072	3	928	309.3	239.9
950224001	15	8371	558.1	175.2
950224021	11	5378	488.9	60.0
950224026	6	3390	565.0	309.0
950239052	8	2968	371.0	69.9
950239067	14	6785	484.6	65.0
950239071	11	4762	432.9	150.7
950239073	8	2951	368.9	127.6
950239076	5	1633	326.6	70.3
950239081	4	445	111.3	9.8
950248002	8	3600	450.0	49.1
950248017	2	79.6	39.8	112.8
950248018	9	3923	435.9	88.4
950248028	4	1340	335.0	52.3
950248030	6	2598	433.0	336.2
950248031	7	3322	474.6	58.5
950248032	1	79.6	79.6	183.3
Average				140.8
Standard Deviation				110.5

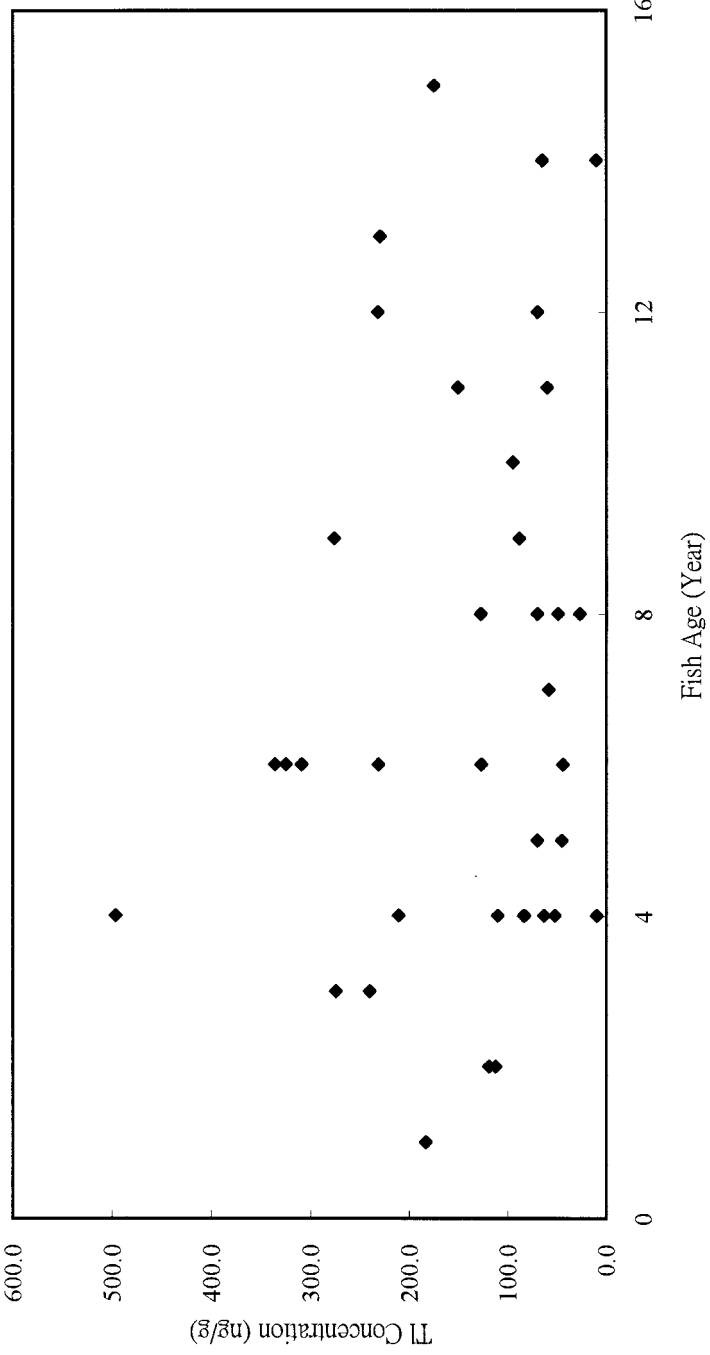


Figure 1. The relationship of thallium concentration and fish age

solids. A Perkin-Elmer 4100ZL graphite furnace atomic absorption spectrometer with a Zeeman background corrector was employed for the final determination of thallium. The analytical conditions of graphite furnace atomic absorption spectroscopy for thallium were: wavelength, 276.8 nm; low slit, 0.7 nm; pretreatment temperature, 400°C; atomization temperature, 1200°C. The method detection limit was 0.2 ng/g. Approximately 30% of the runs were replicates designed to ensure the quality of the analytical procedure.

RESULTS AND DISCUSSION

The average concentration of thallium in 37 lake trout samples was determined to be 140.8 ± 110.5 ng/g wet weight (Table 1) and is similar to reported values for brook trout, white sucker and yellow perch in the Adirondack lakes, USA (Heit 1985). Assuming that the dry weight concentration of thallium in biota is ten times higher than the wet weight concentration (Table 1), the dry weight concentration of thallium in lake trout is estimated to be approximately 1.5 mg/kg. The bioaccumulation factor (BF) of thallium in lake trout calculated to 10000, using an average thallium concentration in Lake Michigan water of 14 ng/L (Lin and Nriagu 1999). The BF is significantly higher than the 27 - 1430 determined in a 300-hour experiment using juvenile Atlantic salmon (Zitko et al. 1975). The relation between thallium concentration in a lake trout and its growth (Figure 1) can be analyzed using a simple regression model. A significant correlation was found between the growth factor (defined as fish weight/fish age) and thallium concentration ($p = 0.055$). Although the highest concentrations of thallium can be expected in the larger, older fish, the value observed seems to vary largely. The variation could not be defined by our limited observation.

Consumption of fish from the Great Lakes is always of some concern because of the high levels of pollutants they contain (Courval et al. 1996; Foran et al. 1989). The consumption advisories for fish from the Great Lakes are generally based on PCB and mercury levels in the fish and rarely consider contaminants such as thallium (US EPA 1997). Simple calculations based on our data suggest that this may be a serious oversight. The acceptable daily intake (ADI) for thallium has been estimated at 0.2 $\mu\text{g}/\text{kg}$ or 14 $\mu\text{g}/70$ kg (Ewers 1988). The Reference Dose (RfD) for thallium suggested by the U.S. EPA (1992) provides a more cautious value (0.07 $\mu\text{g}/\text{kg}/\text{day}$ or 5 $\mu\text{g}/70$ kg/day) for protecting the public from chronic thallium poisoning. The estimated daily dietary intake of thallium by the U.S. population is around 7 $\mu\text{g}/70$ kg (U.S. DHHS 1992), which is one half the ADI.

Assuming a regular consumption of four to eight fish meals per month (one meal = 230 g of flesh), which is considered typical for anglers (Denis et al. 1997), the estimated average daily thallium intake is 12 ~ 16 $\mu\text{g}/70$ kg; this value is comparable to the ADI. However, approximately 15% of the angler population is believed to consume almost twice as much fish as the average angler (Courval et al. 1996). Estimated thallium intake by the high fish consumers is over 20 $\mu\text{g}/70$ kg/day, and may place them at risk for chronic thallium poisoning. This report clearly calls for a detailed assessment of the risk of chronic thallium poisoning that may be associated with consumption of fish from the Great Lakes.

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