Check for updates

OPEN ACCESS

EDITED BY Zhen Liu, Beijing University of Chemical Technology, China

REVIEWED BY Xiaomin Liu, Qingdao University, China Baocun Zhu, University of Jinan, China

CORRESPONDENCE
Gongying Wang
☑ wanggongying1102@126.com
Yanyan Diao
☑ yydiao@ipe.ac.cn

SPECIALTY SECTION

This article was submitted to Antimicrobials, Resistance and Chemotherapy, a section of the journal Frontiers in Microbiology

RECEIVED 06 December 2022 ACCEPTED 09 January 2023 PUBLISHED 06 February 2023

CITATION

Hu Y, Xing Y, Ye P, Yu H, Meng X, Song Y, Wang G and Diao Y (2023) The antibacterial activity and mechanism of imidazole chloride ionic liquids on *Staphylococcus aureus*. *Front. Microbiol.* 14:1109972. doi: 10.3389/fmicb.2023.1109972

COPYRIGHT

© 2023 Hu, Xing, Ye, Yu, Meng, Song, Wang and Diao. This is an open-access article distributed under the terms of the Creative Commons Attribution License (CC BY). The use, distribution or reproduction in other forums is permitted, provided the original author(s) and the copyright owner(s) are credited and that the original publication in this journal is cited, in accordance with accepted academic practice. No use, distribution or reproduction is permitted which does not comply with these terms.

The antibacterial activity and mechanism of imidazole chloride ionic liquids on *Staphylococcus aureus*

Yanhui Hu^{1,2,3,4}, Yuyuan Xing^{2,3,4}, Peng Ye⁵, Haikuan Yu⁶, Xianglei Meng^{2,3}, Yuting Song^{2,3}, Gongying Wang^{1,3*} and Yanvan Diao^{2,3,4*}

¹Chengdu Institute of Organic Chemistry, Chinese Academy of Sciences, Chengdu, China, ²Beijing Key Laboratory of Ionic Liquids Clean Process, CAS Key Laboratory of Green Process and Engineering, State Key Laboratory of Multiphase Complex Systems, Institute of Process Engineering, Chinese Academy of Sciences, Beijing, China, ³University of Chinese Academy of Sciences, Beijing, China, ⁴Innovation Academy of Green Manufacture, Chinese Academy of Sciences, Beijing, China, ⁵Beijing Key Laboratory of Lignocellulosic Chemistry, Beijing Forestry University, Beijing, China, ⁶Senior Department of Orthopedics, Chinese PLA Medical School, Beijing, China

Ionic liquids (ILs) have garnered increasing attention in the biomedical field due to their unique properties. Although significant research has been conducted in recent years, there is still a lack of understanding of the potential applications of ILs in the biomedical field and the underlying principles. To identify the antibacterial activity and mechanism of ILs on bacteria, we evaluated the antimicrobial potency of imidazole chloride ILs (C_nMIMCl) on *Staphylococcus aureus* (*S. aureus*). The toxicity of ILs was positively correlated to the length of the imidazolidinyl side chain. We selected $C_{12}MIMCl$ to study the mechanism of *S. aureus*. Through the simultaneous change in the internal and external parts of *S. aureus*, $C_{12}MIMCl$ caused the death of the bacteria. The production of large amounts of reactive oxygen species (ROS) within the internal parts stimulated oxidative stress, inhibited bacterial metabolism, and led to bacterial death. The external cell membrane could be destroyed, causing the cytoplasm to flow out and the whole cell to be fragmented. The antibacterial effect of $C_{12}MIMCl$ on skin abscesses was further verified *in vivo* in mice.

KEYWORDS

ionic liquids, Staphylococcus aureus, antibacterial activity, mechanism, skin abscess

1. Introduction

In recent years, diseases caused by bacteria have plagued human beings. Although antibiotics have alleviated this problem, excessive use also causes problems such as bacterial resistance. Considering the overuse of antibiotics and the increasing resistance to antibacterial agents, there is an urgent need to develop tunable antibacterial compounds and drug delivery systems to treat high-mortality diseases. Moreover, the COVID-19 pandemic has garnered more attention than ever regarding the need for more effective medical methods to diagnose and cure the disease. New, high-performing antimicrobial and antiviral materials have to be developed. As a combination of salt and organics, ionic liquids (ILs) have been endowed with abundant and diverse properties (Zhang et al., 2014), including low volatility, a wide electrochemical window, a highly tunable structure, and thermal stability, which have been widely used as solvents (Liang et al., 2019; Shamshina and Berton, 2020), extractants (Gao et al., 2020; Wang et al., 2022), catalysts (Li et al., 2021; Sadjadi, 2021; Yuan et al., 2023), electrolytes (Fu et al., 2022), and so on. ILs could be self-assembled into low-dimensional nanoparticles due to their strong interactions between the cations and anions, H-bond direction, and large molecular volume (Dupont, 2011; Chen et al., 2014). Their unique properties have attracted increasing attention in the biomedical field, such as the extraction and conservation of proteins (Schröder, 2017; Veríssimo et al., 2021; Xue et al., 2022) and nuclear

acids (Dinis et al., 2020), drug transport enhancers (Md Moshikur et al., 2020; Zhang et al., 2020), drug carriers (Lu et al., 2020; Moshikur et al., 2021), drug additives (Tang et al., 2016), and disease diagnosis (Zhu et al., 2020) and therapy (Albadawi et al., 2021; Gao et al., 2021). The nanoparticle structure of ILs provides a variety of possibilities for biomedical applications. *Staphylococcus aureus* is a common foodborne pathogenic gram-positive bacterium which can cause a variety of serious infections (e.g., pneumonia, enteritis, pericarditis, and sepsis) in humans and animals through the skin and food (Shangguan et al., 2015; Nithya and Sundrarajan, 2020). In the treatment process, *S. aureus* has strong resistance and low susceptibility to drugs; thus, it is urgent to develop new ingredients to counteract the problem of drug resistance (Hess et al., 2005). In this study, *S. aureus* was selected to study the toxicity and mechanism of ILs on the bacterium.

Although ILs have been extensively studied in various directions for biomedical applications (Egorova et al., 2017), they currently remain in the preliminary stage due to a lack of systematic research and theoretical basis. In addition, there is little relevant research on ILs as an agent in vivo. An exciting avenue for research is to determine whether ILs can be used in developing antibacterial drugs due to their wide variety, adjustable structure, excellent properties, and the mechanism by which they interact with bacteria. Florio et al. (2019) compared and evaluated the antimicrobial efficacy of 15 ILs, including 1-methyl-3-dodecylimidazolium bromide, 1-dodecyl-1methylpyrrolidinium bromide, and 1-dodecyl-1-methylpiperidinium bromide, which had strong inhibitory effects on the biofilm formation of S. aureus or Pseudomonas aeruginosa (P. aeruginosa) (Florio et al., 2019). Brunel et al. (2016) found that triphenylamine phosphonium ILs could self-assemble into nanoparticles, which have a good antibacterial effect on S. aureus. The IL nanoparticles were likely to strongly affect bacterial metabolism (Brunel et al., 2016). Although some researchers have speculated on the mechanism by which ILs interact with S. aureus, it remains unknown how ILs affect bacterial membranes and influence bacterial metabolism. Further studies are needed to better understand the mechanisms involved in antibacterial activity.

Because imidazole chloride ILs are widely available and studied in a variety of fields, a series of C_n MIMCl were chosen to investigate toxicity mechanisms on bacteria and gain additional insight into the effects of IL structure on antibacterial properties. This study explored the toxicity and mechanism of ILs in gram-positive bacteria (*S. aureus*) and designed and developed ILs for infection prevention and control. A skin abscess model was established to demonstrate the antimicrobial efficacy of ILs *in vivo*.

2. Materials and methods

2.1. Strains, culture conditions, and ILs

The *S. aureus* BNCC 186335 and USA300-sfgfp strains were purchased from Forhigh Biotech. The *S. aureus* strain was cultivated in LB medium (18 g/L Nutrient Broth (NB), Solarbio) for 24 h at 37°C and 180 r/min in a shaker incubator. A microplate reader (Tecan, Infinite M2000) was used to measure OD_{600} . ILs (C₂MIMCl, C₃MIMCl, C₄MIMCl, C₆MIMCl, C₈MIMCl, C₁₂MIMCl, C₁₆MIMCl, \geq 99%) were prepared by Shanghai Cheng Jie Chemical Co., Ltd. Then, the ILs were diluted in sterile, distilled water. Cy5 NHS Ester (Cy5, SE) was purchased from Beijing Fanbo Biochemicals Co., Ltd. Then, 4% of the paraformaldehyde fix solution was purchased from Solarbio.

2.2. Antibacterial assay

Iconic liquids were diluted in sterile, distilled water following the corresponding concentration gradient. Then, 50 μ L of bacteria solutions and different ILs were added to 5 ml of LB medium. Afterward, mixed solutions were hatched at 37°C and 180 r/min in a shaker incubator for 24 h. Before measuring OD₆₀₀, the solution was fully shaken to distribute *S. aureus* and ensure the accuracy of the measurement. Finally, 100 μ L of solutions were removed from each sample and placed into 96-well plates to measure OD₆₀₀. To avoid human or machine error, every OD₆₀₀ value was repeated at least three times. EC50s were determined from OD₆₀₀ by GraphPad Prism8. The activity of *S. aureus* cultured with ILs was observed by a confocal microscope, and the fluorescence was quantified.

Then, 200 μ L of IL was added to 10 mL of solid LB medium (18 g/L NB, 1L water, and 15 g agarose), which was melted to liquid in advance and put in 10-cm plates. Then, 100 μ L of bacterial suspension (OD₆₀₀₌ 0.15) with further dilution was spread on solid LB plates and cultured at 37°C for 48 h. The CFUs were counted after being cultured, and the cultures were photographed with a camera.

2.3. Oxidative stress in S. aureus

 C_3 MIMCl or C_{12} MIMCl was added to *S. aureus* (OD₆₀₀ = 0.15) and mixed with the ROS Assay Kit (DCFH-DA), and the mixed solution was placed in a 96-well plate for 30 min at 37°C. Then, the ROS value was detected with the microplate reader at the excitation wavelength of 488 nm and the emission wavelength of 525 nm. Following the above experimental procedure, 200 µL of the mixture was added to a confocal dish, and fluorescence was photographed using confocal laser scanning microscopy (Nikon Corporation, A1, CLSM) at the excitation wavelength of 488 nm with green as the pseudo color.

2.4. Surface morphology characterization and status observation of *S. aureus*

Staphylococcus aureus ($OD_{600} = 0.15$) was cultured in a liquid LB medium with ILs within 3 h at 37°C. Then, the bacteria were washed with PBS three times and fixed in 4% of the paraformaldehyde fix solution at 4°C overnight. The next day, bacteria were washed with PBS three times and treated with ethanol gradient dehydration at different concentrations (30, 40, 50, 60, 70, 80, 90, 95, and 100%). Bacteria were successfully mixed with different ethanol solution concentrations, were left to stand, were dehydrated for 15 min, and were collected by centrifugation. Finally, the bacteria dehydrated with 100% ethanol onto the tin foil was dropped and dried at 37°C for 2 h. The morphological changes of *S. aureus* treated with ILs were characterized by scanning electron microscopy (SEM, HITACHI, SU8020).

S. aureus was cultured with ILs, centrifuged (8000 r/min, 8 min), washed with water (3 times), and fixed with glutaraldehyde solution (PH = 7.4) overnight. They were taken in cross-sections and stained for further transmission electron microscopy (TEM, HITACHI, H-7650B) observation.

2.5. Measurement of *S. aureus* surface zeta potential

S. *aureus* (OD₆₀₀ = 0.25) was treated with C₃MIMCl and C₁₂MIMCl (the final concentrations: 0.01 mM, 0.04 mM, and 0.16 mM) in liquid LB medium at 37°C for 3 h with gentle shaking. Then, *S. aureus* was washed with PBS three times and sterilized with water one time a day. The obtained bacteria were dispersed in 1 mL of water. The uniform suspension was measured for zeta potential using a Zetasizer Nano ZS (Malvern Instruments Ltd., Zetasizer Nano ZS90). As a control group, the bacteria were treated with an LB medium containing corresponding volumes of water at the same conditions. At the same time, we used Cy5, SE labeled C₁₂MIMCl to interact with *S. aureus* and CLSM to observe the interaction between IL and bacteria.

2.6. Mice

BALB/c mice (6–8 weeks old, female) were obtained from Vital River Laboratories (Beijing, China). The Institutional Animal Care and Use Committees approved the animal protocol of the Institute of Process Engineering, Chinese Academy of Sciences (approval ID: IPEAECA 2022103). This study was performed in strict accordance with the Regulations for the Care and Use of Laboratory Animals and the Guideline for the Ethical Review of Animals.

2.7. Skin abscess model and antibacterial activity of ionic liquid *in vivo*

For the skin abscess model, mice were divided into three groups (controls, infected group, and ILs). Mice were anesthetized with pentobarbital sodium (1%) and inoculated with 100 μ l of PBS containing 10¹³ cfu/L *S. aureus* or sterile PBS in the right flank by subcutaneous injection. After abscess formation, 50 μ l of C₁₂MIMCl (2 mM) was injected daily at the abscess site for 3–4 days. We observed the abscess healing continuously. Skin biopsy samples collected on day 11 were fixed and prepared for histopathological evaluation. The tissues were stained with hematoxylin and eosin (H&E) and 4['],6-diamidino-2-phenylindole (DAPI) and visualized by the scanner (3DHISTECH, P250 FLASH).

3. Results and discussion

3.1. Antibacterial activities of C_nMIMCl against *S. aureus*

The antibacterial activities of a series of imidazole chloride ILs against *S. aureus* were determined. To evaluate different ILs' antimicrobial activity against *S. aureus*, different concentrations of

six ILs were used to measure EC50 (Figure 1A). The EC50 values in the LB medium of C2MIMCl, C3MIMCl, C4MIMCl, C6MIMCl, C₈MIMCl, C₁₂MIMCl, C₁₆MIMCl were 76.7, 28.7, 18.2, 3.5, 0.3, 1.9 x 10^{-3} , and 0.5 x 10^{-3} mM for S. *aureus*, respectively. Imidazolium chloride ILs with substituents of 12 and 16 carbon lengths had strong antibacterial effects on S. aureus. The toxicity of S. aureus increased with the length of the cation side chain. It may be due to the increased hydrophobicity. With an increase in C_nMIMCl concentration, the inhibition rate of S. aureus increased. For example, when the concentration of C12 MIMCl was 0.0015, 0.0020, 0.0025, and 0.0030 mM, the inhibition rates of S. aureus were 23.2, 60.5, 83.1, and 87.7%, respectively (Figure 1B). It showed that the toxicity of C_nMIMCl varied for each concentration. S. aureus transfected with a green fluorescent protein (GFP) was incubated with C3MIMCl or C12MIMCl and observed by CLSM. There was no difference in GFP fluorescence intensity between the C₃MIMCl group and the control group, while S. aureus GFP fluorescence intensity in the C12MIMCl group was significantly reduced (Figures 1C, D).

To further verify the toxicity of C_nMIMCl , the less toxic C_3MIMCl and the more toxic $C_{12}MIMCl$ were selected for plate coating experiments (Figure 2). Compared with the control group, there was no bacterial inhibition when the concentration of C_3MIMCl was 0.0040 mM. The inhibition rates were $65.8 \pm 0.8\%$, $89.1 \pm 5.1\%$, and $94.6 \pm 4.7\%$ when the C_3MIMCl concentration was 26 mM, 32 mM, and 40 mM, respectively. $C_{12}MIMCl$ showed a significant inhibitory effect on *S. aureus* compared with C_3MIMCl . When the concentrations of $C_{12}MIMCl$ were 0.0026 mM, 0.0032 mM, and 0.0040 mM, the inhibition rates were $9.7 \pm 3.9\%$, $23.0 \pm 5.1\%$, and $98.4 \pm 0.8\%$, respectively. With ~95% inhibition of *S. aureus*, the concentration used for C_3MIMCl is about 10^4 times higher than that used for $C_{12}MIMCl$.

The toxicity of C_nMIMCl to *S. aureus* was proportional to the length of the imidazolidinyl side chain. At the same time, the longer the alkyl side chain length of the imidazole, the greater the hydrophobicity of C_nMIMCl . Therefore, the toxicity of C_nMIMCl to *S. aureus* increased with the enhancement of hydrophobicity. It is well known that phospholipid is one of the main components of the cell membrane, and the outer surface of the cell membrane is electronegative. The ILs were easily adsorbed to the bacterial surface due to the positive charge of the cationic part, and then, the ILs interacted with the phospholipid bilayer due to hydrophobicity. Therefore, C_3MIMCl with a short carbon chain and $C_{12}MIMCl$ with a long carbon chain were selected for the subsequent research.

3.2. Oxidative stress of *S. aureus* by C₁₂MIMCl

To further investigate the mechanisms leading to the inhibitory and destructive effects of C_{12} MIMCl on *S. aureus*, as well as the significant reduction of the viability of the exposed bacteria induced by ILs, the release of reactive oxygen species (ROS) within *S. aureus* was explored. ROS is a general term for a class of molecules with oxidative activity produced by cells during energy metabolism under aerobic conditions. In bacteria, excess ROS can damage nucleic acids, proteins, and lipids (Brynildsen et al., 2013). Moreover, it can also lead to bacterial oxidative stress and inhibit bacterial metabolism (Ning et al., 2019; Zhao et al., 2022). Bacterial death occurs when



intracellular ROS exceeds the cell's ability to detoxify and repair. Therefore, it is necessary to determine the amount of ROS in bacteria to reveal the mechanism of IL acting on bacteria. Figure 3 shows CLSM images of ROS (green). No significant changes in ROS fluorescence values were observed with 0.16 mM C₃MIMCl compared with the control group. It was found that C12MIMCl could significantly increase ROS in bacteria compared with C3MIMCl at the same concentration, and the ROS gradually increased with an increase in C12MIMCl concentration. Therefore, C12MIMCl could induce ROS release intracellularly.

3.3. Damage to the *S. aureus* membrane by C₁₂MIMCl

3.3.1. Surface morphology observation of S. aureus

SEM images were utilized to observe the morphological changes of S. aureus after incubation with ILs for 3 h. It could be seen that the normal form of S. aureus strains displayed regular and clear edges, smooth surfaces with rounded projections, and complete cell walls. To further investigate the antibacterial activities of short and long cation side chains, the damage of C₃MIMCl and C₁₂MIMCl on the bacterial membrane of S. aureus was evaluated (Figure 4). Furthermore, similar to their control counterparts, the morphologies of the bacteria treated with 0.16 mM C₃MIMCl for 3 h remained intact and smooth. However, the morphology of S. aureus treated with C12MIMCl was collapsed by depression or wrinkling on the surface, indicating that partial or complete membrane lysis might occur. As the concentration of C12MIMCl increased, a greater effect on the surface morphology of S. aureus could be observed. This result indicated that C12MIMCl could damage the membranes of S. aureus.

3.3.2. Changes on the S. aureus surface

Bacterial membrane potential is the proton motive force of bacterial metabolism and respiration. Antibacterial activity strongly correlates with the dissipation of the membrane potential in bacteria (Schilling et al., 2019). When studying the antibacterial mechanisms, the integrity of bacterial membranes is a non-negligible factor. In our study, when the concentration of C12MIMCl was 0.16 mM, C12MIMCl increased the bacterial membrane potential to 16.3%, and the change in membrane potential increased with an increase in ILs concentration. Compared with C3MIMCl, S. aureus showed an increase in zeta potentials after co-culturing with C_{12} MIMCl (0.16 mM) (Figure 5A), indicating that the positively charged C12MIMCl underwent electrostatic interactions with the negatively charged components of the S. aureus membrane. With an increase in C₁₂MIMCl concentration, the change in S. aureus



Bacterial colony photos of Staphylococcus aureus incubated with various concentrations of C₃MIMCl and C₁₂MIMCl for 48 h.



FIGURE 3

Oxidative stress of *Staphylococcus aureus*. (A) CLSM images of ROS release from bacteria in control. (B–E) CLSM images of ROS release from bacteria in the presence of 0.16 mM C_3 MIMCl (B), 0.01 mM C_{12} MIMCl (C), 0.04 mM C_{12} MIMCl (D), 0.16 mM C_{12} MIMCl (E). (F) Quantitative analysis of ROS release by a microplate reader. Data in (F) represent the mean \pm s.d. Statistical significance of (F) was calculated *via* two-tailed unpaired Student's *t*-test. Ns means no significant difference, **P < 0.01.



FIGURE 4

SEM characterization of *Staphylococcus aureus* morphologies and membrane integrities after 3h co-culture with water (control group), 0.16 mM C_3 MIMCl or different concentrations of C_{12} MIMCl (0.003 mM and 0.16 mM).



The changes of *Staphylococcus aureus* surface. (A) Zeta potentials of *Staphylococcus aureus* co-cultured with C_3 MIMCl or C_{12} MIMCl in PBS for 3h. (B) Picture of *Staphylococcus aureus* incubated with Cy5, SE- C_{12} MIMCl. (C–E) CLSM images of C_{12} MIMCl interacting with *Staphylococcus aureus*. Green: *Staphylococcus aureus* transfected with GFP (C), Red: Cy5, SE- C_{12} MIMCl (D), merged (E). Data in (A) represent the mean \pm s.d. Statistical significance of (A) was calculated *via* two-tailed unpaired Student's *t*-test. Ns means no significant difference, ****P < 0.0001.

membrane potential increased. The potential of *S. aureus* is in full correlation with their EC50 values and is concentration dependent. It could be seen that the bacterial color changed after incubation with Cy5, SE-C₁₂MIMCl, indicating that C₁₂MIMCl was adsorbed on the bacterial surface (Figure 5B). Because the *S. aureus* surface was negatively charged, it could attract the IL cation with a positive charge to gather on the bacterial surface by CLSM (Figures 5C–E). It was once again demonstrated that C₁₂MIMCl could interact with bacterial cell membranes.

3.3.3. Observation of the destruction process of S. aureus by ILs

TEM images further confirmed that *S. aureus* was treated with C_{12} MIMCl, which showed membrane detachment, content leakage, and cell disruption. Figure 6 shows the whole process of *S. aureus* being destroyed. The control group and C_3 MIMCl group showed intact and smooth cell membranes. Initially, part of the cell membrane was shed from the bacterial surface in the presence of C_{12} MIMCl (Figure 6A). Subsequently, the cytoplasm flowed out from



FIGURE 6

TEM images of *Staphylococcus aureus* within 3 h treatment with 0.16 mM of C_{12} MIMCL. (A-D): The destruction process of *Staphylococcus aureus* by C_{12} MIMCL.







the broken cell membrane (Figure 6B). Then, the cell membrane of *S. aureus* was completely disrupted (Figure 6C). Finally, *S. aureus* was dead (Figure 6D).

3.4. Antibacterial mechanism of ionic liquids against *S. aureus*

The cell membrane mainly consists of phospholipids with a negative charge on the surface and hydrophobicity. ILs are amphiphilic, and long cationic side chains can be inserted into the phospholipid bilayer, causing cell membrane disorder (Kaur et al., 2020; Liu et al., 2021). According to the above experimental contents, C_{12} MIMCl could be adsorbed on the surface of *S. aureus* by electrostatic action. Afterward, C_{12} MIMCl was inserted into cell membranes driven by hydrophobicity. This led to an oxidative stress reaction in bacteria, that is, the rapid production of a large amount of ROS. At the same time, it changed the permeability of the bacterial cell membrane, which could change the bacterial osmotic pressure, destroying the cell membrane and allowing cytoplasm to flow out of the broken cell membrane. Figure 7 depicts a plausible antibacterial mechanism for interacting C_{12} MIMCl with *S. aureus*.

3.5. The therapeutic effect of C₁₂MIMCl on skin abscess *in vivo*

3.5.1. Observation of a skin abscess

The skin of *S. aureus*-infected mice developed visible abscesses, redness, edema, and ulcers within 1–2 days of infection. The abscess was injected with 2 mM C_{12} MIMCl (50 µl) for 3–4 days. On the second day of C_{12} MIMCl administration, the abscess showed a distinct black scab, and the abscess under the skin gradually decreased, while the abscess in the control group gradually grew larger (Figure 8). In the following days, the scabbed area gradually expanded. The subcutaneous abscess did not disappear completely

until 3–4 days after the C_{12} MIMCl injection. After 8 days, the scab gradually fell off. The skin in the treatment group could be healed for 10–12 days. However, the subcutaneous abscess remained in the control group.

3.5.2. Analysis of skin tissue

Skin biopsy samples collected on day 11 were fixed and prepared for histopathological evaluation (Figures 9A-C). Compared with the control group, the wavy curves of the epidermis and dermis at the subcutaneous abscess sites of S. aureus infection disappeared, indicating that the epidermis and dermis were severely damaged. In addition, there was a large infiltration of monocytes in the dermis. After the C12MIMCl injection, the morphology of the epidermis and dermis gradually recovered, and the number of inflammatory cells decreased significantly. H&E staining of skin sections from the mice revealed that C12MIMCl had a good therapeutic effect on the subcutaneous abscess. After DAPI staining and GFP spontaneous fluorescence scanning, it could be seen that the number of S. aureus in the skin was nearly zero after the C₁₂MIMCl injection. The number of S. aureus cells in the skin remained largely unchanged in the untreated group (Figures 9D-F). In conclusion, C12MIMCl could improve the skin abscess infected with S. aureus in mice.

4. Conclusion

A variety of imidazole chloride ILs with varying length substituents was investigated for their antibacterial mechanisms against S. aureus. Imidazole chloride ILs containing twelve and sixteen carbon lengths had strong antibacterial and anti-biofilm activity against S. aureus. The results showed that the antibacterial efficiency of ILs could be improved by changing the alkyl chain length due to the tunable structure of ILs. The above research demonstrated that long cationic side-chain ILs effectively inhibited S. aureus. This study provided new ideas for creating new antibacterial drugs. ROS release tests confirmed that long cationic side-chain ILs have a strong impact on the metabolism of S. aureus. SEM and membrane potential test experiments revealed that ILs could collapse and damage the surface of S. aureus. TEM images clearly showed how C12MIMCl disrupts the cell membrane of S. aureus, causing the contents to leak. Ultimately, the antibacterial mechanism of C12MIMCl against S. aureus was proven. In vivo tests, C12MIMCl had significant antibacterial effects and accelerated the healing of skin abscesses. After the new coronavirus pandemic in 2019, it is urgent to develop new antibacterial and antiviral drugs. ILs have not been found to be appropriate drugs for clinical use. At present, we know that the toxicity of ILs to S. aureus depends on the length of cationic alkyl chain substituents. In the future, we will expand the types of organism models (e.g., Gramnegative bacteria, viruses, cells, and animals) and design specific properties for different diseases by regulating different types of ILs substituents.

Data availability statement

The original contributions presented in the study are included in the article/supplementary material,

further inquiries can be directed to the corresponding authors.

Ethics statement

The animal study was reviewed and approved by the Institutional Animal Care and Use Committees at the Institute of Process Engineering, Chinese Academy of Sciences.

Author contributions

YH, GW, and YD contributed to the conception and design of the study. YH, PY, and HY performed the experiments. YH, PY, and XM contributed significantly to data analyses. YH and YX organized the pictures of the experiment. YH wrote the manuscript. HY, YX, and YS revised the manuscript. GW and YD oversaw the completion of this study. All authors contributed to the article and approved the submitted version.

Funding

This study was supported by the National Natural Science Foundation of China (21978292 and 21676280), the Innovation Academy for Green Manufacture, the Chinese Academy of Sciences (IAGM2020C20), the Major Program of the National Natural Science Foundation of China (21890762), and the International Partnership Program of the Chinese Academy of Sciences (grant no. 122111KYSB20190060).

Acknowledgments

The authors sincerely appreciate Prof. Suojiang Zhang (IPE, CAS) for his careful academic guidance and great support.

Conflict of interest

The authors declare that the research was conducted in the absence of any commercial or financial relationships that could be construed as a potential conflict of interest.

Publisher's note

All claims expressed in this article are solely those of the authors and do not necessarily represent those of their affiliated organizations, or those of the publisher, the editors and the reviewers. Any product that may be evaluated in this article, or claim that may be made by its manufacturer, is not guaranteed or endorsed by the publisher.

References

Albadawi, H., Zhang, Z., Altun, I., Hu, J., Jamal, L., Ibsen, K. N., et al. (2021). Percutaneous liquid ablation agent for tumor treatment and drug delivery. *Sci. Transl. Med.* 13, eabe3889. doi: 10.1126/scitranslmed.abe3889

Brunel, F., Lautard, C., Garzino, F., Giorgio, S., Raimundo, J. M., Bolla, J. M., et al. (2016). Antibacterial activities of fluorescent nano assembled triphenylamine phosphonium ionic liquids. *Bioorg. Med. Chem. Lett.* 26, 3770–3773. doi: 10.1016/j.bmcl.2016.05.055

Brynildsen, M. P., Winkler, J. A., Spina, C. S., MacDonald, I. C., and Collins, J. J. (2013). Potentiating antibacterial activity by predictably enhancing endogenous microbial ROS production. *Nat. Biotechnol.* 31, 160–165. doi: 10.1038/nbt.2458

Chen, S. M., Zhang, S. J., Liu, X. M., Wang, J. Q., Wang, J. J., Dong, K., et al. (2014). Ionic liquid clusters: structure, formation mechanism, and effect on the behavior of ionic liquids. *Phys. Chem. Chem. Phys.* 16, 5893–5906. doi: 10.1039/C3CP53116C

Dinis, T. B. V., Sousa, F., and Freire, M. G. (2020). Insights on the DNA stability in aqueous solutions of ionic liquids. *Front. Bioeng. Biotechnol.* 8, 547857. doi: 10.3389/fbioe.2020.547857

Dupont, J. (2011). From molten salts to ionic liquids a "nano" journey. Accounts Chem. Res. 44, 1223–1231. doi: 10.1021/ar2000937

Egorova, K. S., Gordeev, E. G., and Ananikov, V. P. (2017). Biological activity of ionic liquids and their application in pharmaceutics and medicine. *Chem. Rev.* 117, 7132–7189. doi: 10.1021/acs.chemrev.6b00562

Florio, W., Becherini, S., D'Andrea, F., Lupetti, A., Chiappe, C., and Guazzelli, L. (2019). Comparative evaluation of antimicrobial activity of different types of ionic liquids. *Mater. Sci. Eng.* C 104, 109907. doi: 10.1016/j.msec.2019.109907

Fu, D., Sun, Y., Zhang, F., Sun, Z., Li, W., Wang, L., et al. (2022). Enabling polymeric ionic liquid electrolytes with high ambient ionic conductivity by polymer chain regulation. *Chem. Eng. J.* 431, 133278. doi: 10.1016/j.cej.2021.133278

Gao, S. R., Fang, S. F., Song, R. Z., Chen, X. C., and Yu, G. R. (2020). Extractive denitrogenation of shale oil using imidazolium ionic liquids. *Green Energ. Environ.* 5, 173–182. doi: 10.1016/j.gee.2020.04.002

Gao, Y. R., Cao, J. F., Shu, Y., and Wang, J. H. (2021). Research progress of ionic liquidsbased gels in energy storage, sensors and antibacterial. *Green Chem. Eng.* 2, 368–383. doi: 10.1016/j.gce.2021.07.012

Hess, D. J., Garni, R. M., Henry-Stanley, M. J., and Wells, C. L. (2005). Escherichia colimodulates extraintestinal spread of *Staphylococcus aureus*. Shock 24, 376–381. doi: 10.1097/01.shk.0000180615.75822.fe

Kaur, N., Fischer, M., Kumar, S., Gahlay, G. K., Scheidt, H. A., and Mithu, V. S. (2020). Role of cationic head-group in cytotoxicity of ionic liquids: probing changes in bilayer architecture using solid-state NMR spectroscopy. *J. Colloid. Interface Sci.* 581, 954–963. doi: 10.1016/j.jcis.2020.08.115

Li, M., Chen, J., Li, L., Ye, C., Lin, X., and Qiu, T. (2021). Novel multi-SO3H functionalized ionic liquids as highly efficient catalyst for synthesis of biodiesel. *Green Energ. Environ.* 6, 271–282. doi: 10.1016/j.gee.2020.05.004

Liang, L., Yan, J. P., He, Q., Luong, T., Pray, T. R., Simmons, B. A., et al. (2019). Scale-up of biomass conversion using 1-ethyl-3-methylimidazolium acetate as the solvent. *Green Energ. Environ.* 4, 432–438. doi: 10.1016/j.gee.2018.07.002

Liu, J., Wang, Y., Wang, C., Gao, J., Cui, W., Zhao, B., et al. (2021). Thermodynamical origin of nonmonotonic inserting behavior of imidazole ionic liquids into the lipid bilayer. *J. Phys. Chem. Lett.* 12, 9926–9932. doi: 10.1021/acs.jpclett.1c02566

Lu, B. B., Zhou, G. X., Xiao, F., He, Q. J., and Zhang, J. H. (2020). Stimuli-responsive poly(ionic liquid) nanoparticles for controlled drug delivery. *J. Mater. Chem. B* 8, 7994–8001. doi: 10.1039/D0TB01352H

Md Moshikur, R., Chowdhury, M. R., Fujisawa, H., Wakabayashi, R., Moniruzzaman, M., and Goto, M. (2020). Design and characterization of fatty acid-based amino acid ester as a new "green" hydrophobic ionic liquid for drug delivery. *ACS Sustain. Chem. Eng.* 8, 13660–13671. doi: 10.1021/acssuschemeng.0c03419

Moshikur, R. M., Ali, M. K., Moniruzzaman, M., and Goto, M. (2021). Recent advances in surface-active ionic liquid-assisted self-assembly systems for drug delivery. *Curr. Opin. Colloid In.* 56, 101515. doi: 10.1016/j.cocis.2021.101515

Ning, H.-Q., Li, Y.-Q., Tian, Q.-W., Wang, Z.-S., and Mo, H.-Z. (2019). The apoptosis of *Staphylococcus aureus* induced by glycinin basic peptide through ROS oxidative stress response. *LWT-Food Sci. Technol.* 99, 62–68. doi: 10.1016/j.lwt.2018.09.028

Nithya, P., and Sundrarajan, M. (2020). Ionic liquid functionalized biogenic synthesis of Ag-Au bimetal doped CeO_2 nanoparticles from *Justicia Adhatoda* for pharmaceutical applications: antibacterial and anti-cancer activities. *J. Photochem. Photobiol. B* 202, 111706. doi: 10.1016/j.jphotobiol.2019.111706

Sadjadi, S. (2021). Magnetic (poly) ionic liquids: a promising platform for green chemistry. J. Mol. Liq. 323, 114994. doi: 10.1016/j.molliq.2020.114994

Schilling, N. A., Berscheid, A., Schumacher, J., Saur, J. S., Konnerth, M. C., Wirtz, S. N., et al. (2019). Synthetic lugdunin analogues reveal essential structural motifs for antimicrobial action and proton translocation capability. *Angew. Chem. Int. Ed. Engl.* 58, 9234–9238. doi: 10.1002/anie.201901589

Schröder, C. (2017). Proteins in ionic liquids: current status of experiments and simulations. *Top. Curr. Chem.* 375, 25. doi: 10.1007/s41061-017-0110-2

Shamshina, J. L., and Berton, P. (2020). Use of ionic liquids in chitin biorefinery: a systematic review. *Front. Bioeng. Biotechnol.* 8, 11. doi: 10.3389/fbioe.2020.00011

Shangguan, J., Li, Y., He, D., He, X., Wang, K., Zou, Z., et al. (2015). A combination of positive dielectrophoresis driven on-line enrichment and aptamer-fluorescent silica nanoparticle label for rapid and sensitive detection of *Staphylococcus aureus*. *Analyst* 140, 4489–4497. doi: 10.1039/C5AN00535C

Tang, W. T., Liu, B., Wang, S. P., Liu, T. L., Fu, C. H., Ren, X. L., et al. (2016). Doxorubicin-loaded ionic liquid-polydopamine nanoparticles for combined chemotherapy and microwave thermal therapy of cancer. *RSC Adv.* 6, 32434–32440. doi: 10.1039/C6RA02434C

Veríssimo, N. V., Saponi, C. F., Ryan, T. M., Greaves, T. L., and Pereira, J. F. B. (2021). Imidazolium-based ionic liquids as additives to preserve the enhanced green fluorescent protein fluorescent activity. *Green Chem. Eng.* 2, 412–422. doi: 10.1016/j.gce.2021.08.001

Wang, Y., Wang, S., and Liu, L. (2022). Recovery of natural active molecules using aqueous two-phase systems comprising of ionic liquids/deep eutectic solvents. *Green Chem. Eng.* 3, 5–14. doi: 10.1016/j.gce.2021.07.007

Xue, Y., Zhang, X. G., Lu, Z. P., Xu, C., Xu, H. J., and Hu, Y. (2022). Enhancing the catalytic performance of candida antarctica lipase b by chemical modification with alkylated betaine ionic liquids. *Front. Bioeng. Biotechnol.* 10, 850890. doi: 10.3389/fbioe.2022.850890

Yuan, X., Liu, J., Wang, Y., Jie, X., Qin, J., and He, H. (2023). Ionic liquids enable highly efficient knoevenagel reaction by dual-responsive emulsion microreactor. *Chem. Eng. J.* 451, 138941. doi: 10.1016/j.cej.2022.138941

Zhang, S. J., Sun, J., Zhang, X. C., Xin, J. Y., Miao, Q. Q., and Wang, J. J. (2014). Ionic liquid-based green processes for energy production. *Chem. Soc. Rev.* 43, 7838–7869. doi: 10.1039/C3CS60409H

Zhang, Y., Cao, Y., Meng, X., Li, C., Wang, H., and Zhang, S. (2020). Enhancement of transdermal delivery of artemisinin using microemulsion vehicle based on ionic liquid and lidocaine ibuprofen. *Colloid. Surfaces B* 189, 110886–110886. doi: 10.1016/j.colsurfb.2020.1 10886

Zhao, Y., Wei, J., Li, C., Ahmed, A. F., Liu, Z., and Ma, C. (2022). A comprehensive review on mechanism of natural products against *Staphylococcus aureus*. *J. Future Foods* 2, 25–33. doi: 10.1016/j.jfutfo.2022. 03.014

Zhu, X. L., Tang, X. X., Lin, H. Y., Shi, S. G., Xiong, H. H., Zhou, Q. J., et al. (2020). A fluorinated ionic liquid-based activatable 19 fMRI platform detects biological targets. *Chem* 6, 1134–1148. doi: 10.1016/j.chempr.2020. 01.023