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The Catalytic Asymmetric Diels–Alder Reactions and Post-cycloaddition Reductive Transpositions of 1-Hydrizinodienes

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Dienes that enable structural rearrangements in the wake of a Diels–Alder event can afford structurally unique and complex cyclohexenes that can be inaccessible by the direct cycloaddition route.^[1–4] A particular problem in natural product synthesis required a substituted cyclohexene of the type **4**, and we were drawn to the idea that an initial pairing of a hypothetical diene of type **1** with an activated dienophile of type **2** might be followed by a suprafacial, reductive transposition of **3** to the desired cyclohexene **4** (Scheme 1).

In principle, the Diels–Alder chemistry of Fleming’s 1-trimethylsilyl-1,3-butadiene in conjunction with a post-cycloaddition protodesilylation step offers an attractive path to a type **4** structure.^[2,5] While this strategy is feasible, 1-trimethylsilyl-1,3-butadiene displays low levels of regioselectivity in cycloadditions with unsymmetrical dienophiles, and the subsequent protodesilylation step can afford mixtures of epimers when a new stereocenter is produced. Given these circumstances, we designed a 1-hydrizinodiene that allows a stepwise realization of the concept outlined in Scheme 1.^[6] For example, *exo* cycloadduct **6** is produced by a stereospecific union of 1-hydrizinodiene **5** with diethyl maleate and subsequently converted to the isolable hydrazine derivative **7** by a palladium-catalyzed cleavage of the two allyloxy carbonyl groups in **6** (Scheme 2). By the action of a weak base (e.g., sodium acetate), compound **7** is then transformed to the desired cyclohexene **9** via the putative allylic diazene **8**; the spontaneous process that transforms **8** to **9** is formulated as a retroene rearrangement with loss of molecular nitrogen.^[7–9] Interestingly, if the base-induced elimination of methanesulfinic acid from **7** is conducted in CD₃OD, H–D exchange occurs and the ensuing reductive transposition stereospecifically affords the deuterated cyclohexene **10**.

A growing number of examples demonstrate that 1-hydrizinodienes undergo a range of Lewis acid-catalyzed Diels–Alder reactions that are both regio- and diastereoselective as a setup for subsequent, stereospecific reductive transpositions to rearranged cyclohexenes. In this report, we describe our more recent discovery that 1-hydrizinodienes are amenable to

chiral catalyst-controlled, enantioface-selective Diels–Alder cycloadditions, as well as the cycloaddition behavior of new 1-hydrazinodienes for use in chemical synthesis.^[10]

In our effort to merge electron-deficient dienophiles with 1-hydrazinodiene **5** with high margins of stereoselectivity, we discovered that the chiral copper(II) bis(oxazoline) catalysts of Evans and co-workers^[11] mediate efficient, regioselective, and highly stereoselective Diels–Alder reactions of *N*-acryloyl oxazolidinones with diene **5**. Unions of 1-hydrazinodiene **5** with *N*-acryloyl oxazolidinone **11a** were best achieved in methylene chloride at room temperature in the presence of 4 Å molecular sieves and 10 mol% of the freshly prepared copper(II) bis(oxazoline) catalyst. In all cases, *exo* cycloadduct **13a** was produced as the major diastereo-isomer with varying levels of enantioselectivity. The results summarized in Table 1, reveal the impact of the identity of the group R on the chiral bis(oxazoline) ligand and the counterion on Diels–Alder diastereo- and enantioselectivity. The *tert*-butyl bis(oxazoline) ligand afforded excellent levels of diastereo- and enantioselectivities. While the chloride salt of the copper(II) bis(oxazoline) catalyst was unreactive, the hexafluoroantimonate and triflate salts displayed excellent reactivities. The good-to-excellent *exo* diastereoselectivities exhibited in these reactions are consistent with our prior observations on the stereochemical outcomes of 1-hydrazinodiene cycloadditions to C_α-unsubstituted dienophiles.^[6,12] Our hypothesis is that dienophiles lacking α-substitution should undergo *exo* selective Diels–Alder reactions to minimize nonbonded interactions between the Lewis acid-activated carbonyl and the substituents attached to the hydrazine moiety of the diene.^[13]

Having identified the (*S,S*)-(-)-2,2'-isopropylidene-bis(4-*tert*-butyl-2-oxazoline) chiral ligand and the hexafluoroantimonate counter ion as key components of an effective chiral catalyst, we examined a variety of β-substituted *N*-acryloyl oxazolidinones in asymmetric Diels–Alder reactions with 1-hydrazinodiene **5** (Table 2).

Although there was some variation in reaction times, all of the unions leading to *exo* cycloadducts **13a–l** displayed diastereomer ratios of greater than 20:1 and enantiomer ratios ranging from 21–99:1. Evans's copper(II) catalyst **14** is clearly capable of mediating cycloadditions of diverse, β-substituted *N*-acryloyl oxazolidinones to diene **5** with high margins of stereoselectivity.

To further increase the scope of this chemistry, we leveraged our previously described method^[6] to achieve syntheses of an expanded set of hydrazinodienes with diverse substitution patterns. Thus, from simple α,β-unsaturated aldehydes and monoallyloxycarbonyl (Alloc) hydrazine, 1-hydrazinodienes **15–18** (Table 3) were synthesized in three steps^[14] and employed in asymmetric Diels–Alder reactions with α,β-unsaturated imides **11a**, **11b**, **11 f**, and **11l**.

Qualitatively, these new hydrazinodienes were judged to be comparable with respect to reactivity, although dienes **16** and **18** reacted more slowly in relation to the others. All of these chiral catalyst-directed cycloadditions were regioselective and afforded *exo* cycloadducts **19a–k** in good to excellent yields and with diastereomer ratios greater than 20:1. The major, *exo* diastereomers were also produced with high levels of enantioselectivity. X-ray crystallographic analysis confirmed the relative and absolute stereochemical configurations of cycloadduct **19a**; this analysis was fully consistent with the prior observations of Evans and co-workers^[11] on how the architecture of the dienophile-copper(II) BOX complex imparts high levels of stereoface selectivity in Diels–Alder reactions.^[15]

In the wake of the asymmetric Diels–Alder events, it was straightforward to execute the desired reductive transpositions to rearranged cyclohexenes (Table 4). Thus, the Diels–Alder adducts arising from diene **5** and the four dienes shown in Table 3, were smoothly transformed to the isolable hydrazine derivatives **20a–h** by mild, palladium(0)-catalyzed cleavages of the Alloc protecting groups. The reductive transpositions to cyclohexenes **21a–h** were subsequently achieved by warming solutions of compounds **20a–h** in methanol to 50°C. Through a retroene-like rearrangement^[7] of a putative allylic diazene intermediate, molecular nitrogen is expelled, the alkene is shifted to a new position within the six-membered ring, and a new stereochemical relationship is established in this pivotal step.

In the presence of Lewis acids, 1-hydrazinodienes undergo efficient [4+2] cycloadditions with fumarate and maleate esters, as well as α,β -unsaturated aldehydes, ketones, and imides. To gain some insight into the relative reactivity of 1-hydrazinodienes, the HOMO Eigenvalues for 1-dimethylamino-3-*tert*-butyldimethylsilyloxy-1,3-butadiene,^[16] 1-methoxy-3-trimethylsilyloxy-1,3-butadiene,^[17] 1-hydrazinodiene **5**,^[6] and isoprene were calculated as -0.172 , -0.190 , -0.207 , and -0.226 , respectively, by the method of Gaussian 03 B3LYP at the 6-31G(d) level of theory.^[18] By this analysis, the HOMO energy of 1-hydrazinodiene **5** was judged to be less than the HOMO energies of the synergistic dienes of Rawal and Kozmin^[16] and Danishefsky and Kitahara,^[17] but greater than that of isoprene.

As a class, the 1-hydrazinodienes have value in synthesis because they are easily constructed, amenable to efficient and highly stereoselective Diels–Alder reactions with a variety of dienophiles, and enable mild, post-cycloaddition rearrangements to new cyclohexenes that would likely be challenging to produce by alternative methods of synthesis. Our efforts to further extend the utility of 1-hydrazinodienes in organic synthesis are continuing.

Supplementary Material

Refer to Web version on PubMed Central for supplementary material.

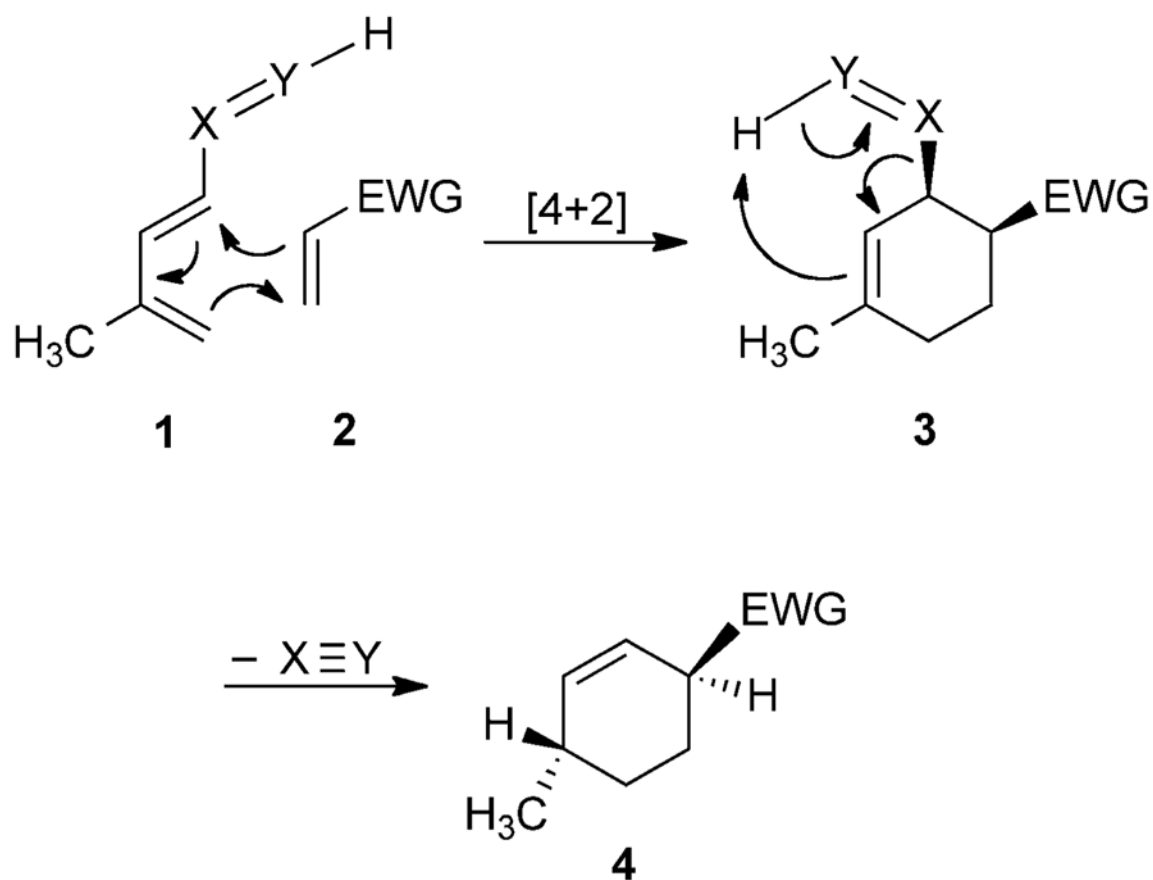
Acknowledgments

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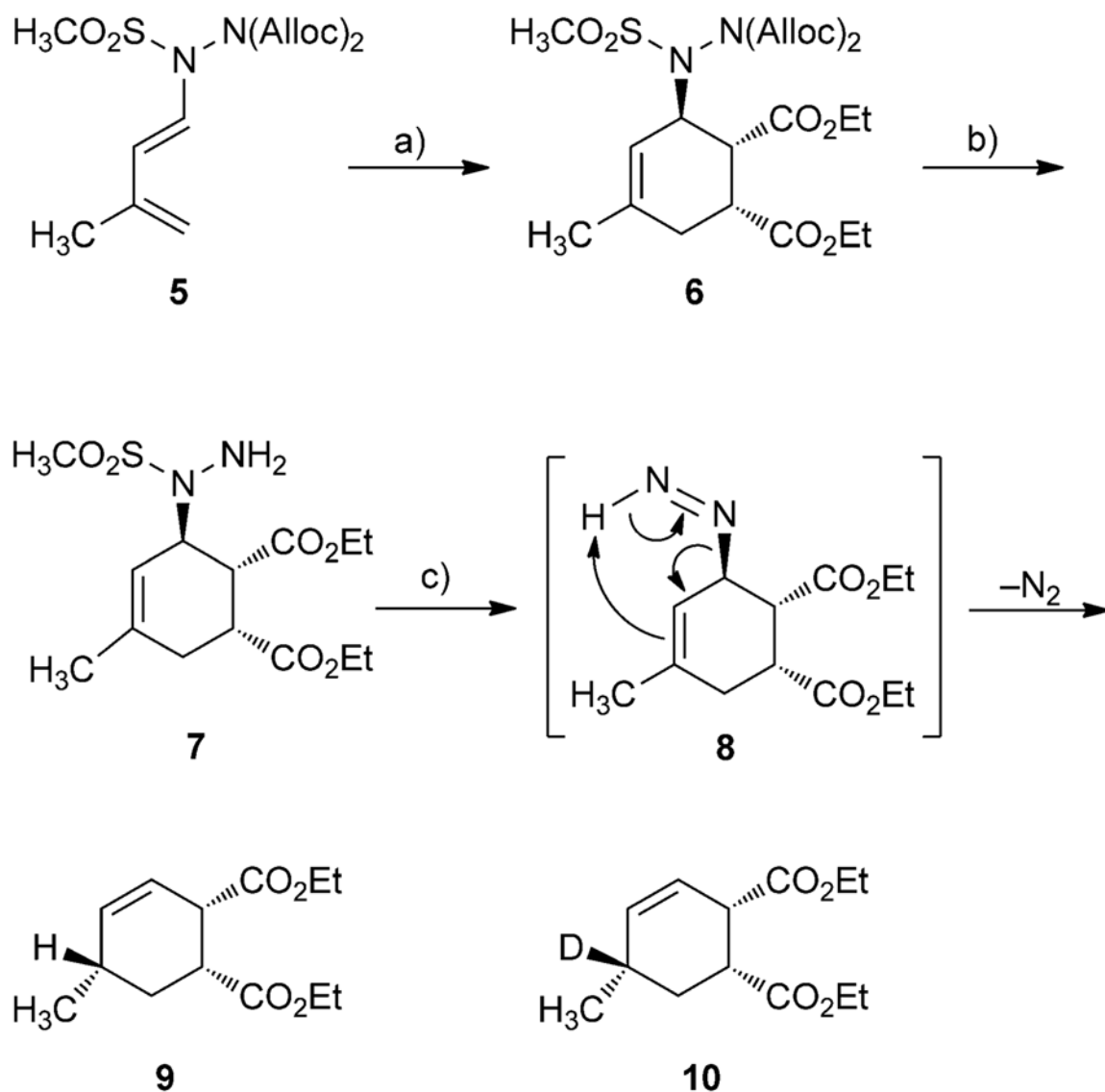
References

1. Evans DA, Bryan CA, Sims CL. *J Am Chem Soc.* 1972; 94:2891–2892.
2. a) Carter MJ, Fleming I. *J Chem Soc Chem Commun.* 1976:679–680. b) Carter MJ, Fleming I, Percival A. *J Chem Soc Perkin Trans 1.* 1981:2415–2434. c) Chan TH, Fleming I. *Synthesis.* 1979:761–786.
3. Huber S, Stamouli P, Neier R. *J Chem Soc Chem Commun.* 1985:533–534. Huber S, Stamouli P, Jenny T, Neier R. *Helv Chim Acta.* 1986; 69:1898–1915. Schoepfer J, Marquis C, Pasquier C, Neier R. *J Chem Soc Chem Commun.* 1994:1001–1002. Arce E, Carreño MC, Belén Cid M, García Ruano JL. *J Org Chem.* 1994; 59:3421–3426. for an excellent review of tandem reactions combining Diels–Alder reactions with sigmatropic rearrangements, see: Neuschütz K, Velker J, Neier R. *Synthesis.* 1998:227–255.
4. For examples of three-component couplings featuring the tandem hetero-Diels–Alder/allylboration chemistry of 1-aza-4-borono-1,3-butadienes, see: Taylor J, Hall DG. *Org Lett.* 2000; 2:3715–3718. [PubMed: 11073683]

5. Fleming and Carter noted in ref. [2a] that “the ability to move the double bond of a Diels–Alder adduct in this way has wide implications in the design of organic syntheses.”
6. Sammis GM, Flamme EM, Xie H, Ho DM, Sorensen EJ. *J Am Chem Soc.* 2005; 127:8612–8613. [PubMed: 15954764]
7. Bumgardner CL, Freeman JP. *J Am Chem Soc.* 1964; 86:2233–2235. Jabbari A, Sorensen EJ, Houk KN. *Org Lett.* 2006; 8:3105–3107. [PubMed: 16805563] For a review of retroene reactions, see: Ripoll JL, Vallée Y. *Synthesis.* 1993:659–677.
8. For selected examples of allylic diazene rearrangements and diazene reactivity in synthesis, see: Sato T, Homma I, Nakamura S. *Tetrahedron Lett.* 1969; 10:871–874. Corey EJ, Cane DE, Libit L. *J Am Chem Soc.* 1971; 93:7016–7021. Hutchins RO, Milewski CA, Maryanoff BE. *J Am Chem Soc.* 1973; 95:3662–3668. Hutchins RO, Kacher M, Rua L. *J Org Chem.* 1975; 40:923–926. Kabalka GW, Yang DTC, Baker JD Jr. *J Org Chem.* 1976; 41:574–575. Danheiser RL, Carini DJ, Fink DM, Basak A. *Tetrahedron.* 1983; 39:935–947. Corey EJ, Wess G, Xiang YB, Singh AK. *J Am Chem Soc.* 1987; 109:4717–4718. Corey EJ, Virgil SC. *J Am Chem Soc.* 1990; 112:6429–6431. Myers AG, Kukkola PJ. *J Am Chem Soc.* 1990; 112:8208–8210. Myers AG, Finney NS. *J Am Chem Soc.* 1990; 112:9641–9643. Steinmeyer A, Neef G. *Tetrahedron Lett.* 1992; 33:4879–4882. Greco MN, Maryanoff BE. *Tetrahedron Lett.* 1992; 33:5009–5012. Wood JL, Porco JA Jr, Taunton J, Lee AY, Clardy J, Schreiber SL. *J Am Chem Soc.* 1992; 114:5898–5900. Myers AG, Zheng B. *J Am Chem Soc.* 1996; 118:4492–4493. Myers AG, Zheng B. *Tetrahedron Lett.* 1996; 37:4841–4844. Myers AG, Movassaghi M, Zheng B. *J Am Chem Soc.* 1997; 119:8572–8573. Myers AG, Movassaghi M. *J Am Chem Soc.* 1998; 120:8891–8892. Ott GR, Heathcock CH. *Org Lett.* 1999; 1:1475–1478. [PubMed: 10825996] Harmata M, Bohnert GJ. *Org Lett.* 2003; 5:59–61. [PubMed: 12509890] Hutchison JM, Lindsay HA, Dormi SS, Jones GD, Vicic DA, McIntosh MC. *Org Lett.* 2006; 8:3663–3665. [PubMed: 16898786] Movassaghi M, Ahmed OK. *J Org Chem.* 2007; 72:1838–1841. [PubMed: 17274659] Qi W, McIntosh MC. *Org Lett.* 2008; 10:357–359. [PubMed: 18092798] Anada M, Tanaka M, Shimada N, Nambu H, Yamawaki M, Hashimoto S. *Tetrahedron.* 2009; 65:3069–3077.
9. Addition of the mild base tetra-*n*-butylammonium acetate to the reaction mixture for the palladium-catalyzed Alloc deprotections enables a one-flask conversion of compound **6** to cyclohexene **9** (see ref. [6]).
10. This study is based on: Xie H. PhD Thesis. Princeton University USA2009
11. a) Evans DA, Miller SJ, Lectka T. *J Am Chem Soc.* 1993; 115:6460–6461. b) Evans DA, Miller SJ, Lectka T, von Matt P. *J Am Chem Soc.* 1999; 121:7559–7573. c) Johnson JS, Evans DA. *Acc Chem Res.* 2000; 33:325–335. [PubMed: 10891050]
12. For discussions of exo selectivities in intermolecular Diels–Alder reactions, see: Lam, Y-h; Cheong, PH-Y.; Blasco Mata, JM.; Stanway, SJ.; Gouverneur, V.; Houk, KN. *J Am Chem Soc.* 2009; 131:1947–1957. [PubMed: 19154113]
13. Methacrolein, a C_α-substituted dienophile, reacts with 1-hydrazinodiene **5** in the presence of diethylaluminum chloride (20 mol%) through a transition state that presumably minimizes steric interactions between the branched methyl group and the groups on the diene, and affords a 92:8 mixture of diastereoisomers favoring the *endo* cycloadduct (see ref. [6]).
14. 1-Hydrazinodienes **15–18** were obtained as stable solids. The syntheses of these compounds, including characterization data, are provided in the Supporting Information.
15. Thus, the configurations of the nitrogen- and carbonyl-bearing stereocenters in **19a** are both assigned as *S*.
16. Kozmin SA, Rawal VH. *J Org Chem.* 1997; 62:5252–5253.
17. a) Danishefsky S, Kitahara T. *J Am Chem Soc.* 1974; 96:7807–7808. b) Danishefsky S. *Acc Chem Res.* 1981; 14:400–406.
18. Frisch, MJ.; Trucks, GW.; Schlegel, HB.; Scuseria, GE.; Robb, MA.; Cheeseman, JR.; Montgomery, JA., Jr; Vreven, T.; Kudin, KN.; Burant, JC.; Millam, JM.; Iyengar, SS.; Tomasi, J.; Barone, V.; Mennucci, B.; Cossi, M.; Scalmani, G.; Rega, N.; Petersson, GA.; Nakatsuji, H.; Hada, M.; Ehara, M.; Toyota, K.; Fukuda, R.; Hasegawa, J.; Ishida, M.; Nakajima, T.; Honda, Y.; Kitao, O.; Nakai, H., et al. Gaussian 03, Revision D.01. Gaussian, Inc; Wallingford CT: 2004.



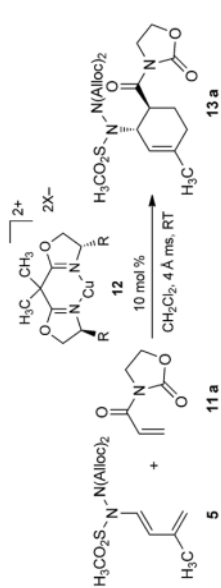
Scheme 1.
A cyclohexene synthesis featuring a post-cycloaddition reductive transposition.

**Scheme 2.**

The Diels–Alder and reductive transposition chemistry of a 1-hydrazone diene. Reaction conditions: a) diethylmaleate, Et_2AlCl , 23 °C, 75%; b) $\text{Pd}_2(\text{dba})_3$, Et_2NH , THF, 23 °C; c) NaOAc , MeOH, 49% over two steps; Alloc: allyloxycarbonyl.

Table 1

Chiral copper(II) bis(oxazoline)-catalyzed Diels–Alder cycloadditions of diene **5** with *N*-acryloyl oxazolidinone **11a**.^[a]



Entry	R	X	Conversion [%] ^[b]	d.r. ^[b]	e.r. ^[c]
1	<i>i</i> Pr	SbF ₆	100	10.2:1	6:1
2	Ph	SbF ₆	100	7.7:1	4:1
3	Bn	SbF ₆	100	8.5:1	6:1
4	<i>t</i> Bu	SbF ₆	100	>20:1	49:1
5	<i>i</i> Bu	OTf	100	>20:1	28:1
6	<i>i</i> Bu	Cl	0	n.d.	n.d.

^[a]Reactions were carried out with diene **5** (0.125 mmol), dienophile **11a** (0.188 mmol), copper(II) bis(oxazoline) catalyst **12** (0.0125 mmol), powdered 4 Å molecular sieves (ms; 19 mg), and CH₂Cl₂ (250 μL) at room temperature for 3 h.

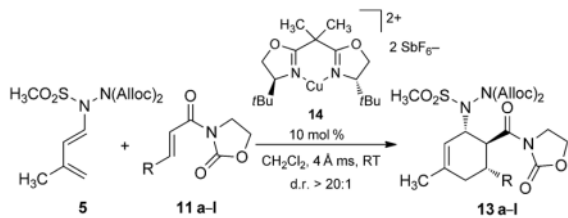
^[b]Conversion and d.r. were determined by analyses of the crude reaction mixtures by ¹H NMR spectroscopy.

^[c]Enantiomer ratios are reported for the major *exo* diastereoisomer and were determined by chiral high-performance liquid chromatography on a Chiralcel OD column.

Alloc: allyloxy carbonyl; d.r.: diastereomer ratio; e.r.: enantiomer ratio; n.d.: not determined.

Table 2

Chiral catalyst-controlled, asymmetric Diels–Alder cycloadditions of diene **5** to β -substituted *N*-acryloyl oxazolidinones **11a–l**.^[a]



Product	R	Yield [%] ^[b]	Reaction time [h]	e.r. ^[c]
13a	H	84	3	49:1
13b	CH ₃	79	4	99:1
13c	CH ₂ CH ₃	74	6	24:1
13d	CH ₂ CH ₂ CH ₃	75	12	99:1
13e	CH(CH ₃) ₂	65	24	99:1
13f	Ph	66	6	49:1
13g	<i>p</i> -CH ₃ Ph	65	12	32:1
13h	<i>p</i> -ClPh	64	12	21:1
13i	<i>p</i> -BrPh	54	12	21:1
13j	<i>p</i> -CF ₃ Ph	60	12	21:1
13k	CH=CHCH ₃	54	18	21:1
13l	CO ₂ CH ₂ CH ₃	83	3	49:1

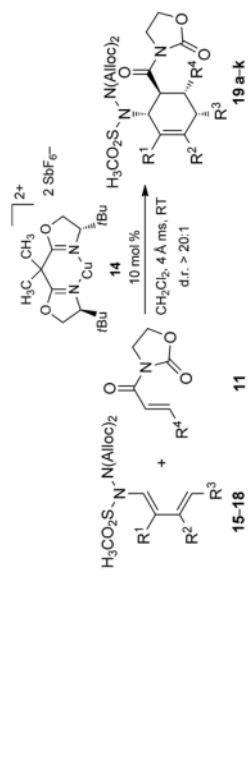
^[a]Reactions were carried out with diene **5** (0.25 mmol), dienophile **11** (0.375 mmol), copper(II) bis(oxazoline) catalyst **14** (0.025 mmol), powdered 4 Å molecular sieves (ms; 37.5 mg), and CH₂Cl₂ (521 μ L) at room temperature.

^[b]Isolated yield after purification by silica gel column chromatography.

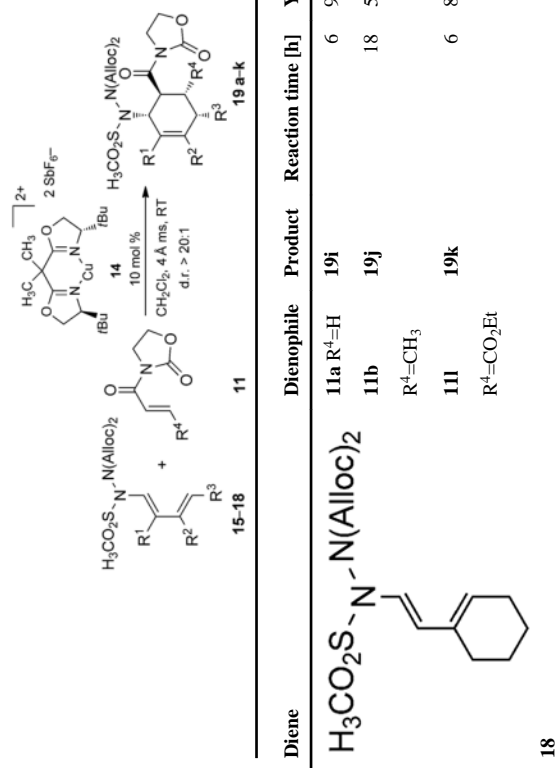
^[c]Enantiomer ratios are reported for the major *exo* diastereoisomer and were determined by chiral high-performance liquid chromatography or supercritical fluid chromatography.

Alloc: allyloxy carbonyl; d.r.: diastereomer ratio; e.r.: enantiomer ratio.

Table 3

Chiral catalyst-controlled, asymmetric Diels–Alder cycloadditions of additional 1-hydrazinodienes.^[a]

Diene	Dienophile	Product	Reaction time [h]	Yield ^[b] [%]	e.r. ^[c]
15 	11a $\text{R}^4=\text{H}$	19a	3	72	49:1
	11b $\text{R}^4=\text{CH}_3$	19b	3	76	32:1
	11f $\text{R}^4=\text{Ph}$	19c	6	78	32:1
	11i $\text{R}^4=\text{CO}_2\text{Et}$	19d	3	66	199:1
16 	11a $\text{R}^4=\text{H}$	19e	12	70	32:1
	11i $\text{R}^4=\text{CO}_2\text{Et}$	19f	12	91	99:1
17 	11a $\text{R}^4=\text{H}$	19g	4	76	66:1
	11i $\text{R}^4=\text{CO}_2\text{Et}$	19h	4	80	199:1

**18**

^[a] Reactions were carried out with diene (0.25 mmol), dienophile (0.375 mmol), copper(II) bis(oxazoline) catalyst **14** (0.025 mmol), powdered 4 Å molecular sieves (ms; 37.5 mg), and CH_2Cl_2 (521 μL) at room temperature.

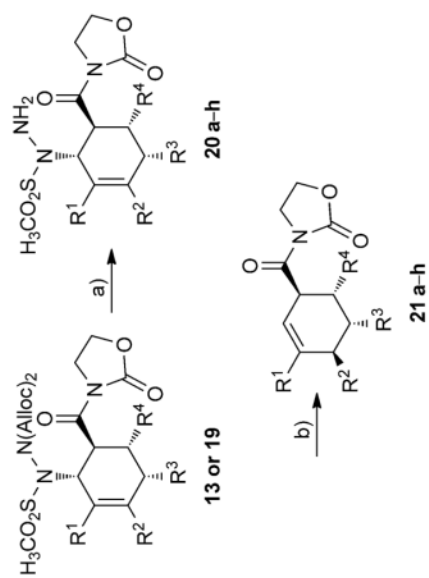
^[b] Isolated yield after purification by silica gel column chromatography.

^[c] Enantiomer ratios are reported for the major *exo* diastereoisomer and were determined by chiral high-performance liquid chromatography or supercritical fluid chromatography.

Alloc: allyloxy carbonyl; d.r.: diastereomer ratio; e.r.: enantiomer ratio.

Table 4

Deprotections and reductive transpositions of selected Diels–Alder products.^[a]



Substrate	Product	R ¹	R ²	R ³	R ⁴	Yield [%] ^[b]
13a	21a	H	CH ₃	H	H	80
13b	21b	H	CH ₃	H	CH ₃	87
13f	21c	H	CH ₃	H	Ph	81
13l	21d	H	CH ₃	H	CO ₂ Et	77
19a	21e	H	Ph	H	H	85
19e	21f	CH ₃	CH ₃	H	H	73
19g	21g	H	Cl	H	H	56
19i	21h	H	CH ₂ CH ₂	CH ₂ CH ₂	H	76

^[a] Reagents and conditions: a) cycloadducts **13** or **19** (0.20 mmol), Pd₂(dba)₃·CHCl₃ (0.01 mmol), 1,2-bis(diphenylphosphino)ethane (0.02 mmol), morpholine (1.6 mmol) in THF (1.0 mL), room temperature, 0.25 h; b) CH₃OH, 50 °C.

^[b] Isolated product yield after purification by silica gel column chromatography for two steps.

Alloc: allyloxy carbonyl; dba: dibenzylideneacetone.