

The combination effects of trivalent gold ions and gold nanoparticles with different antibiotics against resistant *Pseudomonas aeruginosa*

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Abstract Despite much success in drug design and development, *Pseudomonas aeruginosa* is still considered as one of the most problematic bacteria due to its ability to develop mutational resistance against a variety of antibiotics. In search for new strategies to enhance antibacterial activity of antibiotics, in this work, the combination effect of gold materials including trivalent gold ions (Au^{3+}) and gold nanoparticles (Au NPs) with 14 different antibiotics was investigated against the clinical isolates of *P. aeruginosa*, *Staphylococcus aureus* and *Escherichia coli*. Disk diffusion assay was carried out, and test strains were treated with the sub-inhibitory contents of gold nanomaterial. Results showed that Au NPs did not increase the antibacterial effect of antibiotics at tested concentration (40 $\mu\text{g}/\text{disc}$). However, the susceptibility of resistant *P. aeruginosa* increased in the presence of Au^{3+} and methicillin, erythromycin, vancomycin, penicillin G, clindamycin and nalidixic acid, up to 147 %. As an individual experiment, the same group of antibiotics was tested for their

activity against clinical isolates of *S. aureus*, *E. coli* and a different resistant strain of *P. aeruginosa* in the presence of sub-inhibitory contents of Au^{3+} , where Au^{3+} increased the susceptibility of test strains to methicillin, erythromycin, vancomycin, penicillin G, clindamycin and nalidixic acid. Our finding suggested that using the combination of sub-inhibitory concentrations of Au^{3+} and methicillin, erythromycin, nalidixic acid or vancomycin may be a promising new strategy for the treatment of highly resistant *P. aeruginosa* infections.

Keywords Au^{3+} · Au NPs · Antibiotic resistance · Combination effect · *Pseudomonas aeruginosa*

Introduction

The emerging use of antibiotics has led to microbial resistance which is still considered as a major problem in chemotherapy of many infectious diseases [1]. Many mechanisms are involved in the process of antibiotic resistance which mainly include enzymatic degradation and modification of the antibiotic agent [2], modification of the target site of the drug [3], and active reflux and reduced uptake of the drug [4]. *Pseudomonas aeruginosa* is an opportunistic human pathogen Gram negative bacterium which is responsible for infections such as blood stream nosocomial infections, especially in immunocompromised patients and the elderly population of industrial societies [5]. The organism has reputation for having minimal nutritional requirements, the ability to tolerate a wide variety of physical conditions and the ability to resist against new antibiotics such that it has been addressed as “the worst nightmare” of microbiologists and clinical pharmacists for its highly resistant nature [6]. The antibiotic resistance of

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this microorganism is due to a number of mechanisms which mainly include co-operation of multidrug efflux pumps such as MexAB-OprM, a pump system that removes β -lactams, chloramphenicol, fluoroquinolones, macrolides, novobiocin, sulfonamides, tetracycline and trimethoprim, as well as various dyes and detergents [7]; low impermeability of the membrane to drugs and the fact that it readily develops mutational resistance to most antibacterial agents [8]. The problem of multidrug resistance of *P. aeruginosa* has urged many scientists and pharmaceutical companies to search for new potential therapies for this Gram negative bacterium [9–11]. Therefore, studies on new strategies to combat *P. aeruginosa* would be of great value.

An alternative strategy to overcome the problem of resistance is the use of commonly used antibiotics in combination with different natural or chemical agents [12, 13]. To date, many organic and inorganic compounds have been reported to enhance the antibacterial activity of different antibiotics against many bacteria and fungi resistant test strains [14]. Metallic ions and metallic nanoparticles including zinc and silver have particularly shown promise when used in combination with a number of antibiotics such as ciprofloxacin [15], penicillin G, amoxicillin, erythromycin, clindamycin, vancomycin [16], ampicillin, kanamycin, erythromycin and chloramphenicol [17] in different Gram positive or Gram negative test strains. In a recent approach, Rai et al. reported the antibiotic mediated synthesis of Au NPs with potent antimicrobial activity for application in antimicrobial coatings [18].

During our previous attempts to explore new agents that modulate antibiotic resistant, we screened different chemical substances and natural products and reported a number of organic and inorganic compounds which reduced the resistance of various Gram positive and negative bacteria including *Staphylococcus aureus*, *Clostridium difficile* and *Aspergillus* sp. [14, 19]. In particular, our team reported that some monoterpenes (one of the major components of essential oils) enhanced antibacterial activity of nitrofurantoin against resistant strains of *Enterobacteria* [19]. We also showed that the antibacterial activity of fluconazol against different species of *Aspergillus* increased in the presence of different concentrations of *Sarcococca saligna* ethanol extract [14]. In our very recent work, we demonstrated the enhanced activity of ciprofloxacin in the presence of ZnO nanoparticles [15]. We also reported that silver nanoparticles enhance antibacterial activity of different antibiotics against *S. aureus* and *Escherichia coli* [16]. However, our search for finding new compounds that enhance the activity of antibiotics against *P. aeruginosa* was not successful during past years.

To date, gold-based drugs have shown great promise in treatment of various diseases such as auranofin for the treatment of arthritis and triphenylphosphinegold (I) complexes for the treatment of cancer tumors, psoriasis and HIV

infections [20]. Moreover, gold complexes have shown considerable cytotoxic and antimicrobial activity [21–23]. In this study, we report the combination effect of Au^{3+} on antibacterial activity of different antibiotics against clinically resistant strains of *P. aeruginosa*, *S. aureus* and *E. coli*.

Material and methods

Gold materials

Chloroauric acid was purchased from Merck, Darmstadt, Germany. Au NPs used during this investigation were prepared by previously described tannin-free ethanol extract of black tea (*Camellia sinensis*) method [24]. Briefly, an aqueous chloroauric acid solution (10^{-3} M) was added separately to the reaction vessel containing the tannin-free ethanol extract of black tea (10 %v/v), and the resulting mixture was allowed to stand for 15 min at room temperature. The reduction of the Au^{3+} ions by tannin-free ethanol extract of black tea in the solutions was monitored by sampling the aqueous component (2 ml) and measuring the UV–visible spectrum of the solutions. This sample was diluted three times with distilled water, and the UV–visible spectrum of this sample was measured on a Labomed Model UVD-2950 UV–Vis Double Beam PC Scanning spectrophotometer, operated at a resolution of 2 nm. Furthermore, Au NPs were characterized by transmission electron microscopy (model EM 208 Philips). The gold colloid solution was centrifuged ($12000 \times g$) for 60 min. Subsequently the settled Au NPs were washed three times with deionised water. A stock colloid solution (100 mg/ml) was prepared and reserved in 4°C for further experiments.

Antimicrobial assay

The antibacterial activity of Au^{3+} and Au NPs was evaluated at different contents (31.25, 62.5, 125, 250, 500, 1000, 2000 and 4000 $\mu\text{g}/\text{disc}$) on Müller–Hinton agar (MHA) (Difco, Germany) using conventional disk diffusion method against *P. aeruginosa*, *S. aureus* and *E. coli*. Minimum inhibitory content was defined as the lowest content of Au^{3+} or Au NPs creating clear zone of inhibition after 24 h at 35°C . This disk diffusion susceptibility test was also carried out on MHA plates in order to examine the antibacterial activity of candidate antibiotics against resistant test strains. Standard antibiotics disks, listed in Table 1, were purchased from Mast Co., UK. In order to explore a possible combination effect of gold materials and antibiotics, each standard paper disk was impregnated with the sub-inhibitory content of 40 $\mu\text{g}/\text{disk}$ of Au^{3+} and Au NPs. A resistant strain of *P. aeruginosa* was obtained from Imam University Hospital (Tehran, Iran) and identified by conventional microbiological

Table 1 The antibacterial activity of Au^{3+} ions and gold nanoparticles against *Pseudomonas aeruginosa*, *Staphylococcus aureus* and *Escherichia coli*

Compounds ($\mu\text{g}/\text{disk}$)	Zone of inhibition diameter (mm)		
	<i>E. coli</i>	<i>P. aeruginosa</i> (I)	<i>S. aureus</i>
Au^{3+}			
4000	32	30	34
2000	28	24	30
1000	21	20	22
500	14	17	16
250	10	14	13
125	9 (MIC)	12	11
62.5	–	9 (MIC)	9 (MIC)
31.25	–	–	–
Au NPs			
4000	14	14	13
2000	10	11	10 (MIC)
1000	9 (MIC)	9 (MIC)	–
500	–	–	–
250	–	–	–
125	–	–	–
62.5	–	–	–
31.25	–	–	–

Minimum inhibitory content (MIC) was defined as the lowest content of Au^{3+} or Au NPs creating clear zone of inhibition after 24 h at 35°C

and biochemical methods. A single colony of test strains was grown overnight on Mueller–Hinton broth (MHB) medium on a rotator shaker (200 rpm) at 35°C . The inocula were prepared by diluting the cultures with 0.9 % NaCl to a 0.5 McFarland standard and were applied to the plates along with the standard and test disks containing 40 $\mu\text{g}/\text{disk}$ of Au^{3+} and Au NPs. After incubation at 35°C for 24 h, the zones of inhibition were measured. The mean surface area of each inhibition zone (square millimeter) was calculated from the mean diameter of each tested antibiotic. The percent of increase in the inhibition zone areas for different antibiotics against *P. aeruginosa* was calculated as $(b^2 - a^2)/a^2 \times 100$ where a is the inhibition zone in the presence of antibiotic only, and before addition of Au^{3+} and Au NPs, while b represents the inhibition zone in the presence of antibiotic plus Au^{3+} or Au NPs. The same procedure was used for combination of Au^{3+} and Au NPs with antibiotics against additional test strains. All experiments were performed in triplicate.

Additionally, a different clinical isolate of *P. aeruginosa* and two test strains of *S. aureus* and *E. coli* were obtained from Ghods Polyclinic Laboratory (Tehran, Iran), and the same procedure was repeated. To compare the antibacterial activity results of test and control samples and in order to avoid possible errors, a parallel test was run for pure antibiotics with conditions similar to those for antibiotic–gold material

combination. The enhancing effect of Au^{3+} with different antibiotics was further determined against mentioned test strains using the method described above (Table 1).

Results and discussion

In this study, the Au NPs were prepared using a tannin-free ethanol extract of black tea. The inset to Fig. 1 shows the tubes containing this tannin-free extract before (tubes A) and after the reaction with Au^{3+} for 15 min (tubes B). The gold-containing solutions (tubes A and B) that were a transparent yellow at first turned into purple on completion of the reaction by the tannin-free ethanol extract of black tea (tube B). These reaction mixtures were further characterized by UV–visible spectroscopy. As illustrated in Fig. 1, a strong surface plasmon resonance maximum was observed at ca. 527 nm. This peak is assigned to a surface plasmon phenomenon that is well documented for various metal nanoparticles with sizes ranging from 2 to 100 nm [25–27]. Figure 2 shows representative TEM images recorded from the drop-coated film of the as-prepared Au NPs, synthesized by treating the chloroauric acid solution with a tannin-free ethanol extract of black tea (left picture) after 15 min. The particle size histogram of these spherical gold particles, produced with this tannin-free ethanol extract (right illustration) in Fig. 2), shows that the particles range in size from 1.25 to 17.5 nm. It should be mentioned that almost 60 % of prepared Au NPs were in the range 2–6 nm.

The antibacterial effect of Au^{3+} and Au NPs against *P. aeruginosa*, *S. aureus* and *E. coli* has been determined by

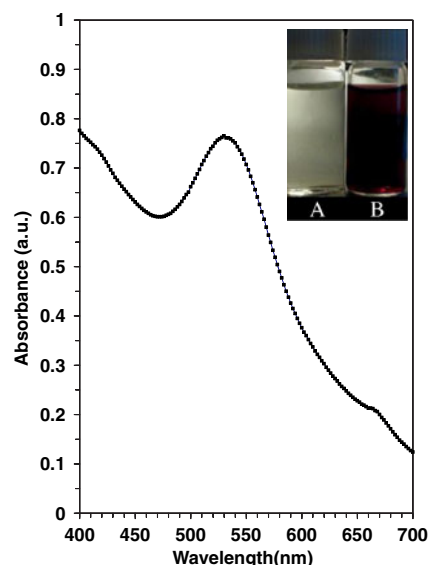
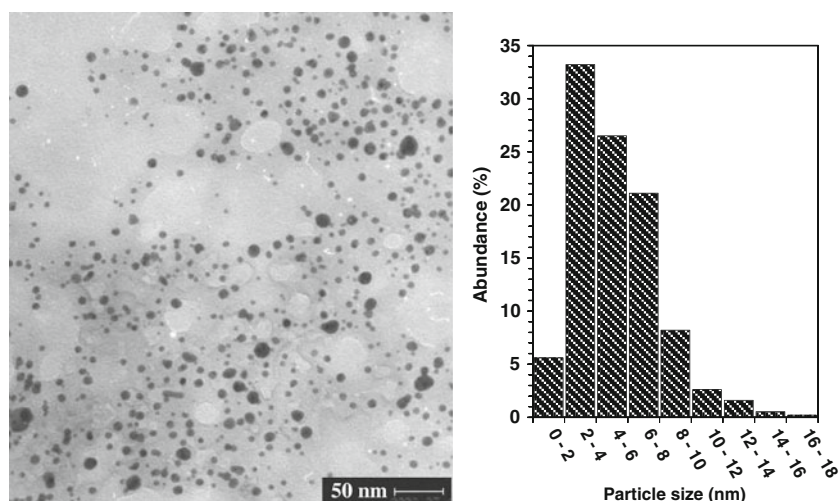


Fig. 1 UV–visible spectrum of gold colloid. Spectrum recorded after adding the tannin-free ethanol extract of black tea (10 ml) to 90 ml of a chloroauric acid solution (1 mM). The curve is recorded after a period of 15 min. The inset shows the solution of chloroauric acid (1 mM) before (a) and after exposure to the tannin-free ethanol extract of black tea (b)

Fig. 2 Transmission electron micrograph recorded from a small region of a drop-coated film of chlorauric acid solution treated with the tannin-free ethanol extract of black tea (left-side picture) for 15 min (scale bars correspond to 50 nm). The related particle size distribution histogram (right-side picture) obtained after measuring the size of 350 individual particles



disk diffusion method and reported in Table 1. Higher concentrations of both Au^{3+} and Au NPs (1000–4000 $\mu\text{g}/\text{disc}$) showed antibacterial activity against above test strains. As antibacterial agent, the Au^{3+} ions were considerably more potent than inert Au NPs against all test strains. Lowest MICs were obtained for Au^{3+} against *P. aeruginosa*, *S. aureus* and *E. coli* (Table 1). In the next step, the antibacterial activity of sub-inhibitory contents of Au^{3+} and Au NPs was investigated in combination with a number of commonly used antibiotics against resistant strains of *P. aeruginosa*, *S. aureus* and *E. coli*. The diameters of inhibition zones (millimeter) in antibiotic disks both in the presence and absence of sub-inhibitory contents of Au^{3+} and Au NPs were calculated. It should be noted that no antibacterial activity was observed for both Au^{3+} and Au NPs at concentrations lower than 62.5 $\mu\text{g}/\text{disc}$. In this investigation the sub-inhibitory content of 40 $\mu\text{g}/\text{disc}$ was chosen which is much lower than the MIC value required to produce antibacterial effect. Therefore, any increase in the antibacterial effect of antibiotics could be attributed not to the cytotoxic effect of Au^{3+} , but to the combination effect of Au^{3+} with antibiotics.

Table 2 shows the inhibition zones (square millimeter) of candidate antibiotics against two different strains of *P. aeruginosa* both in presence and absence of sub-inhibitory content of 40 $\mu\text{g}/\text{disc}$ Au^{3+} . As shown in the Table 2, different antibiotics showed different activities in the presence of Au^{3+} . In both resistant *P. aeruginosa* isolates, the antibacterial activity of penicillin G, methicillin, erythromycin, vancomycin, clindamycin and nalidixic acid increased, while no enhancing effect was observed for the remaining antibiotics. In detail, the surface area of inhibition zones (percent) in resistant *P. aeruginosa* strain 1 plates containing either of methicillin, erythromycin, vancomycin, penicillin G, clindamycin and nalidixic acid increased by 147, 147, 147, 104, 125 and 147 %, respectively.

Furthermore, the effect of Au^{3+} was evaluated using the same set of antibiotics against a different clinical isolate of

P. aeruginosa and the clinical isolates of *S. aureus* and *E. coli*. In *P. aeruginosa* strain 2 group, the sensitivity of the

Table 2 Increase in inhibition zone area (%) of candidate antibiotics against two resistant test strains of *Pseudomonas aeruginosa*, the clinical isolates of *Staphylococcus aureus* and *Escherichia coli* in the presence of Au^{3+} at sub-inhibitory content of 40 $\mu\text{g}/\text{disc}$

Antibiotics ($\mu\text{g}/\text{disc}$)	Increase in inhibition zone area (%)			
	<i>P. aeruginosa</i> (1)	<i>P. aeruginosa</i> (2)	<i>S. aureus</i>	<i>E. coli</i>
Penicillin G 10	104	65	19	0
Amoxicillin 10	0	0	0	0
Methicillin 5	147	146	0	0
Cephalexin 30	0	0	0	30
Cefixime 5	0	0	0	0
Erythromycin 5	147	104	0	0
Gentamicin 10	0	0	7	0
Amikacin 30	0	0	7	9
Tetracycline 30	0	0	10	7
Ciprofloxacin 5	0	0	7	0
Clindamycin 2	125	65	13	65
Nitrofurantoin 300	0	0	8	21
Nalidixic acid 30	147	146	39	0
Vancomycin 30	147	104	0	147

Mean surface area of the inhibition zone (mm^2) was calculated from the mean diameter of each tested antibiotic. The percent of increase in the inhibition zone areas in presence of sub-inhibitory contents of Au^{3+} for different antibiotics against *Pseudomonas aeruginosa* was calculated as $(b^2 - a^2)/a^2 \times 100$ where a is the inhibition zone in the presence of antibiotic only, and before inoculation of Au^{3+} , and b represents the inhibition zone in the presence of antibiotic plus Au^{3+} . No significant inhibition was observed when the combination of antibiotic–Au NPs was used under similar condition

new isolate of *P. aeruginosa* to methicillin, erythromycin, vancomycin, penicillin G, clindamycin and nalidixic acid increased by 146, 104, 104, 65, 65 and 146 %, respectively. However, the inhibition zone was not the same in different clinical isolates of *P. aeruginosa* (104 % and 65 %). In *S. aureus* group, Au^{3+} slightly enhanced the antibacterial activity of penicillin G, gentamicin, amikacin, tetracycline, ciprofloxacin, clindamycin, nitrofurantoin and nalidixic acid. The most enhancing effects in this group were observed for nalidixic acid (39 % increase) and clindamycin (13 % increase). In *E. coli* group, Au^{3+} had enhancing effect on antibacterial activity of cephalexin, amikacin, tetracycline, clindamycin, nitrofurantoin and vancomycin, and the most enhancing effect was observed for cephalexin (30 % increase), clindamycin (65 % increase), nitrofurantoin (21 % increase) and vancomycin (147 % increase).

The test was also repeated with the same set of antibiotics, using the same sub-inhibitory content of 40 $\mu\text{g}/\text{disk}$ Au NPs against resistant *P. aeruginosa* which showed that Au NPs did not have a significant effect on antibacterial activity of antibiotics at a content level of 40 μg (results not shown).

In 2007, Grace and Pandium reported that Au NPs did not have antibacterial effects against a number of microorganisms including *P. aeruginosa*, *S. aureus* and *E. coli*, while coating of Au NPs with antibiotics increased their antibacterial activity [28]. Furthermore, recent studies by Burygin et al. revealed that Au NPs have no enhancing effect on the antibacterial activity of gentamycin [26]. In this study, no significant difference was observed between antibacterial activity of antibiotics alone and their mixture with Au NPs. However, as suggested by Burygin et al., it seems that Au NPs enhance antibacterial activity of antibiotics only when antibiotic is chemically attached on the surface of Au NPs and formed stable conjugates with particles rather than when used in combination with antibiotics as a mixture [29]. Therefore, comparing Au NPs and Au^{3+} , it is plausible that Au^{3+} form more potent mixtures with antibiotics concerning the fact that they are more actively involved in reactions due to their ionic nature. This probably best explains our finding that when used together with antibiotics, Au NPs had no significant effect on antibacterial activity of antibiotics, while Au^{3+} (which form more potent complexes) enhanced the inhibition zone of bacterial growth.

Comparing four groups, it seems that the enhancing effect of Au^{3+} was more considerable in Gram negative *P. aeruginosa* and *E. coli* rather than Gram positive *S. aureus*. It is notable that studies of Marques et al. on MIC (minimum inhibitory concentration) value of Au^{3+} metal compounds in complex with sulphamethoxazole in *P. aeruginosa*, *E. coli* and *S. aureus* suggested no difference between the response of Gram positive and Gram negative bacteria treated with Au compounds in combination with sulphamethoxazole [30], while other studies suggest that the enhancing effect

of Au^{3+} is better in Gram negative bacteria [31]. However, more investigations should be carried out at the molecular level to clarify whether Au^{3+} is more effective on Gram negative rather than Gram positive bacteria.

Fourteen candidate antibiotics were carefully chosen since they represent major classes of antibiotics (penicillins, cephalosporins, macrolides, aminoglycosides, tetracyclines, fluoroquinolones, lincomycin derivatives, nitrofurans and glycopeptides). However, different antibiotics showed different activities in the presence of Au^{3+} . In *P. aeruginosa*, a considerable enhancing effect was observed for methicillin, erythromycin, vancomycin, penicillin G, clindamycin and nalidixic acid, while in *S. aureus* Au^{3+} slightly enhanced the antibacterial activity of penicillin G, gentamicin, amikacin, tetracycline, ciprofloxacin, clindamycin, nitrofurantoin and nalidixic acid. In *E. coli* group, Au^{3+} had an enhancing effect on antibacterial activity of cephalexin, amikacin, tetracycline, clindamycin, nitrofurantoin and vancomycin.

It is already known that Au^{3+} may form a coordination complex with available donor groups such as nitrogen, sulfur and phosphor. Therefore, in the presence of antibiotics such as penicillin, nalidixic acid and clindamycin, Au^{3+} may form coordination complexes with ring nitrogen on these compounds. This interaction may have induced changes in morphology of these compounds, thereby increasing their efficiency. The same mechanism has been proposed for the interaction of gold (III) with zeatin [32]. In beta-lactam and cephalosporins such as penicillin and methicillin, this coordination may have happened between Au^{3+} and free electrons on sulphur and nitrogen donor groups. However, this was not the case for all beta-lactam and cephalosporins tested (amoxicillin and cefixime).

It is also deducible that the enhancing effect of Au^{3+} was more significant in Gram negative *P. aeruginosa* and *E. coli* rather than the Gram positive *S. aureus*. This is potentially interesting since studies by Chudasama et al. on core-shell silver nanostructures [33] and Nomiya et al. on gold (I) complexes [21] also revealed that gold complexes are considerably more effective on Gram negative rather than Gram positive bacteria. However, further investigations are suggested to be carried out on other Gram positive and Gram negative strains to see whether the structure of cell wall in bacteria affects the enhancing effect of gold materials.

So far, a number of elements including copper, lead and zinc have been studied for their interaction with *P. aeruginosa* [28, 34]. As mentioned earlier, the antibiotic resistance of *P. aeruginosa* is primarily due to the co-operation of multidrug efflux pumps and impermeability of bacterial membrane [6]. Therefore, it is possible that Au^{3+} may sensitize *P. aeruginosa* cells by either interfering in the function of these efflux pumps or increasing the permeability of the bacterial membrane. Moreover, *P. aeruginosa* is well known for its resistance to heavy metals through reduction of metal

ions. For instance, it actively resists against Au^{3+} through reducing it to its metallic form [35, 36]. However, when treated with the combination of antibiotic– Au^{3+} , *P. aeruginosa* was more sensitive to the antibiotics. This could be due to the possible new interactions that may form between antibiotics and Au^{3+} which literally hinder *P. aeruginosa* from reducing it to metallic Au and the fact that the organism may not be able to reduce heavy metal ions in the presence of antibiotics. However, more experiments are required to be done in order to determine the underlying mechanism of Au^{3+} in the presence of antibiotics. The mechanism underlying the enhancing effect of Au^{3+} is probably multi-factorial, and a molecular approach is necessary to identify the determinants of this effect.

Conclusion

The result of this work demonstrates that using Au^{3+} , it is possible to enhance the efficacy of a number of commonly used antibiotics against *P. aeruginosa* up to 146 %. This finding is of particular value since resistant strains of *P. aeruginosa* are considered as a major problem in chemotherapy of many infectious diseases. This enhancing effect also occurred in other resistant microorganisms including *S. aureus* and *E. coli*. This suggests that the combination therapy of gold materials such as Au^{3+} could be considered as a new approach and the common antibiotics may have an even broader range of medical applications in the future.

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References

- Levy SB, Marshall B (2004) Antibacterial resistance worldwide: causes, challenges and responses. *Nat Med* 10:122–129
- Wright GD (2005) Bacterial resistance to antibiotics: enzymatic degradation and modification. *Adv Drug Deliver* 57:1451–1470
- Lambert PA (2005) Bacterial resistance to antibiotics: modified target sites. *Adv Drug Deliver* 57:1471–1485
- Kumar A, Schweizer HP (2005) Bacterial resistance to antibiotics: active efflux and reduced uptake. *Adv Drug Deliver* 57:1486–1513
- Wisplinghoff H et al (2003) Current trends in the epidemiology of nosocomial bloodstream infections in patients with hematological malignancies and solid neoplasms in hospitals in the United States. *Clin Infect Dis* 36:1103–1110
- Livermore DM (2002) Multiple mechanisms of antimicrobial resistance in *Pseudomonas aeruginosa*: our worst nightmare? *Antimicrobial Resistance* 34:634–640
- Poole K (2001) Multidrug efflux pumps and antimicrobial resistance in *Pseudomonas aeruginosa* and related organisms. *J Mol Microbiol Biotechnol* 3:255–264
- Li XZ, Zhang L, Poole K (2000) Interplay between the MexA-MexB-OprM multidrug efflux system and the outer membrane barrier in the multiple antibiotic resistance of *Pseudomonas aeruginosa*. *J Antimicrob Chemother* 45:433–436
- Cornelis P (2008) *Pseudomonas*: genomics and molecular biology. Caister Academic, Brussels
- Poole K (2004) Efflux-mediated multiresistance in Gram-negative bacteria. *Clin Microbiol Infect* 10:12–26
- Obritsch MD et al (2005) Nosocomial infections due to multidrug-resistant *Pseudomonas aeruginosa*: epidemiology and treatment options. *Pharmacotherapy* 25:1353–1364
- Tan YT, Tillett DJ, Mackay LA (2000) Molecular strategies for overcoming antibiotic resistance in bacteria. *Mol Med Today* 6:309–314
- Wright GD (2000) Resisting resistance: new chemical strategies for battling superbugs. *Chem Biol* 59:7–16
- Mollazadeh Moghaddam K et al (2010) The Antifungal activity of *Sarcococca saligna* ethanol extract and its combination effect with fluconazole against different resistant *Aspergillus* species. *Appl Biochem Biotech* 162(1):127–133
- Banoee M et al (2010) ZnO nanoparticles enhanced antibacterial activity of ciprofloxacin against *Staphylococcus aureus* and *Escherichia coli*. *J Biomed Mater Res B* 93:557–561
- Shahverdi AR et al (2007) Synthesis and effect of silver nanoparticles on the antibacterial activity of different antibiotics against *Staphylococcus aureus* and *Escherichia coli*. *Nanomedicine* 3:168–171
- Fayaz AM et al (2010) Biogenic synthesis of silver nanoparticles and their synergistic effect with antibiotics: a study against Gram-positive and Gram-negative bacteria. *Nanomedicine* 6:103–109
- Rai A, Prabhune A, Carole C (2010) Antibiotic mediated synthesis of gold nanoparticles with potent antimicrobial activity and their application in antimicrobial coatings. *J Mater Chem* 20:6789–6798
- Raffi F, Shahverdi AR (2007) Comparison of essential oils from three plants for enhancement of antimicrobial activity of nitrofurantoin against Enterobacteria. *Chemotherapy* 53:21–25
- Bowman MC et al (2008) Inhibition of HIV fusion with multivalent gold nanoparticles. *J Am Chem Soc* 130:6896–6897
- Nomiya K, Noguchi R, Oda M (2000) Synthesis and crystal structure of coinage metal (I) complexes with tetrazole (Htetz) and triphenylphosphine ligands, and their antimicrobial activities. A helical polymer of silver (I) complex $[\text{Ag}(\text{tetz})(\text{PPh}_3)_2]_n$ and a monomeric gold (I) complex $[\text{Au}(\text{tetz})(\text{PPh}_3)]$. *Inorg Chim Acta* 298:24–32
- Nomiya K et al (2000) Synthesis, crystal structure and antimicrobial activities of two isomeric gold(I) complexes with nitrogen-containing heterocycle and triphenylphosphine ligands, $[\text{Au}(\text{L})(\text{PPh}_3)]$ (HLspyrazole and imidazole). *J Inorg Biochem* 78:363–370
- Noguchi R, Hara A, Sugie NK (2006) Synthesis of novel gold(I) complexes derived by AgCl-elimination between $[\text{AuCl}(\text{PPh}_3)]$ and silver(I) heterocyclic carboxylates, and their antimicrobial activities. Molecular structure of $[\text{Au}(\text{R}, \text{S-Hpyrrld})(\text{PPh}_3)]$ (H2pyrrld2-pyrrolidone-5-carboxylic acid). *Inorg Chem Commun* 9:355–359
- Banoee M et al (2010) The green synthesis of gold nanoparticles using the ethanol extract of black tea and its tannin free fraction. *Iran J Mater Sci Eng* 7:48–53
- Henglein A (1993) Physicochemical properties of small metal particles in solution: “microelectrode” reactions, chemisorption, composite metal particles, and the atom-to-metal transition. *J Phys Chem B* 97:5457–5471

26. Sastry M et al (1997) pH dependent changes in the optical properties of carboxylic acid derivatized silver colloidal particles. *Colloid Surf A* 27:221–228
27. Sastry M et al (1998) Electrostatically controlled diffusion of carboxylic acid derivatized silver colloidal particles in thermally evaporated fatty amine films. *J Phys Chem B* 102:1404–1410
28. Nirmala Grace A, Pandian k (2007) Antibacterial efficacy of aminoglycosidic antibiotics protected gold nanoparticles: a brief study. *Colloid Surface A* 297: 63–70
29. Burygin GL et al (2009) On the enhanced antibacterial activity of antibiotics mixed with gold nanoparticles. *Nanoscale Res Lett* 4:794–801
30. Marques LL et al (2007) New gold(I) and silver(I) complexes of sulfamethoxazole: synthesis, X-ray structural characterization and microbiological activities of triphenylphosphine(sulfamethoxazolato-N2)gold(I) and (sulfamethoxazolato)silver(I). *Inorg Chem Commun* 10:1083–1087
31. Teitzel GM, Parsek MR (2003) Heavy metal resistance of biofilm and planktonic *Pseudomonas aeruginosa*. *Appl Environ Microb* 69:2313–2320
32. Fowles CC, Smoak EM, Banerjee IA (2010) Interactions of zeatin with gold ions and biomimetic formation of gold complexes and nanoparticles. *Colloid Surf B* 78:250–258
33. Chudasama B et al (2009) Enhanced antibacterial activity of bifunctional Fe₃O₄–Ag core–shell nanostructures. *Nano Res* 2:955–965
34. Aendekerk S, Ghysels B, Cornelis P, Baysse C (2002) Characterization of a new efflux pump, MexGHI–OpmD, from *Pseudomonas aeruginosa* that confers resistance to vanadium. *Microbiology* 148:2371–2381
35. Campbell SC et al (2001) Biogenic production of cyanide and its application to gold recovery. *J Ind Microbiol Biotechnol* 26 (3):134–139
36. Durán N (2007) Antibacterial effect of silver nanoparticles produced by fungal process on textile fabrics and their effluent treatment. *J Biomed Nanotechnol* 3:203–208