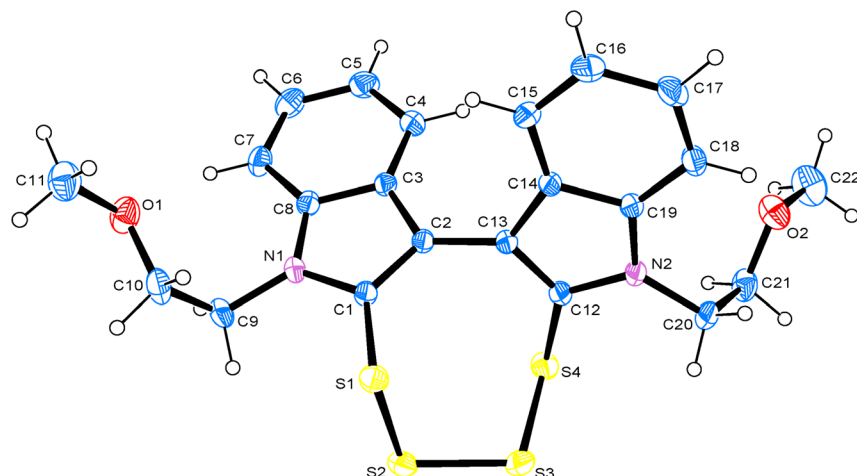


Ming Ni and Penghui Ni\*

# The crystal structure of 5,10-bis(2-methoxyethyl)-5,10-dihydro-[1,2,3,4]tetrathiocino[5,6-*b*:8, 7-*b'*]diindole, C<sub>22</sub>H<sub>22</sub>N<sub>2</sub>O<sub>2</sub>S<sub>4</sub>



<https://doi.org/10.1515/ncrs-2023-0083>

Received February 21, 2023; accepted March 9, 2023;

published online March 23, 2023

## Abstract

C<sub>22</sub>H<sub>22</sub>N<sub>2</sub>O<sub>2</sub>S<sub>4</sub>, triclinic,  $P\bar{1}$  (no. 2),  $a = 8.4865(14)$  Å,  $b = 11.756(2)$  Å,  $c = 12.863(2)$  Å,  $\alpha = 69.693(2)^\circ$ ,  $\beta = 70.812(2)^\circ$ ,  $\gamma = 72.170(2)^\circ$ ,  $V = 1109.6(3)$  Å<sup>3</sup>,  $Z = 2$ ,  $R_{gt}(F) = 0.0280$ ,  $wR_{ref}(F^2) = 0.0753$ ,  $T = 296(2)$  K.

CCDC no.: 2243750

The molecular structure is shown in the figure. Table 1 contains crystallographic data and Table 2 contains the list of the atoms including atomic coordinates and displacement parameters.

## Source of material

All chemicals were purchased from commercial sources and used as received without further purification. A mixture

\*Corresponding author: Penghui Ni, School of Chemistry and Materials Science, Key Laboratory of Functional Metal–Organic Compounds of Hunan Province, Hengyang Normal University, Hengyang, Hunan 421008, China, E-mail: penghuini2020@hynu.edu.cn. <https://orcid.org/0000-0003-4178-3892>

Ming Ni, School of Chemistry and Materials Science, Key Laboratory of Functional Metal–Organic Compounds of Hunan Province, Hengyang Normal University, Hengyang, Hunan 421008, China

**Table 1:** Data collection and handling.

Crystal:	Yellow block
Size:	0.24 × 0.20 × 0.18 mm
Wavelength:	Mo K $\alpha$ radiation (0.71073 Å)
$\mu$ :	0.45 mm <sup>-1</sup>
Diffractometer, scan mode:	Bruker APEX-II, $\varphi$ and $\omega$
$\theta_{\max}$ , completeness:	25.1°, 99%
$N(hkl)_{\text{measured}}$ , $N(hkl)_{\text{unique}}$ , $R_{\text{int}}$ :	5747, 3909, 0.012
Criterion for $I_{\text{obs}}$ , $N(hkl)_{\text{gt}}$ :	$I_{\text{obs}} > 2 \sigma(I_{\text{obs}})$ , 3598
$N(\text{param})_{\text{refined}}$ :	274
Programs:	Bruker [1], SHELX [2, 3], WinGX/ORTEP [4]

of 17.5 g of 1-(2-methoxyethyl)-1*H*-indole, 9.6 g of sulfur and 60 ml of *N,N*-dimethylformamide was stirred under nitrogen in an oil-bath maintained at 150 °C for 6 h, and then allowed to cool slowly overnight in the oil-bath. The yellow crystals which appeared were removed and thoroughly washed with carbon disulfide. The yield of pure 5,10-bis(2-methoxyethyl)-5,10-dihydro-[1,2,3,4]tetrathiocino[5,6-*b*:8, 7-*b'*]diindole, was 11.1 g (47%). Subsequently, dissolve 1 g of the target compound in 30 ml of dichloromethane, heat reflux until the solid is completely dissolved, filtered. Finally, the title crystal was precipitated by controlling solvent volatilization.

## Experimental details

All H-atoms bonded to C atoms were placed geometrically and refined using a riding model with common isotropic

**Table 2:** Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (Å<sup>2</sup>).

Atom	x	y	z	<i>U</i> <sub>iso</sub> */ <i>U</i> <sub>eq</sub>
S1	0.30505 (5)	0.43595 (4)	0.05875 (3)	0.03909 (13)
S2	0.08586 (6)	0.36363 (4)	0.13404 (4)	0.04479 (14)
S3	-0.09841 (6)	0.49793 (4)	0.19691 (4)	0.04544 (14)
S4	-0.04806 (5)	0.49521 (4)	0.34560 (4)	0.04032 (13)
N1	0.52786 (16)	0.28226 (12)	0.19098 (11)	0.0328 (3)
N2	0.02421 (16)	0.72373 (12)	0.29725 (11)	0.0320 (3)
O1	0.88268 (16)	0.14416 (13)	0.08488 (10)	0.0517 (3)
O2	-0.14690 (17)	0.86798 (12)	0.46899 (11)	0.0512 (3)
C1	0.40182 (19)	0.39113 (14)	0.17194 (13)	0.0299 (3)
C2	0.37975 (18)	0.45613 (13)	0.24871 (13)	0.0282 (3)
C3	0.49745 (18)	0.38488 (13)	0.31816 (13)	0.0289 (3)
C4	0.5275 (2)	0.40060 (16)	0.41285 (14)	0.0381 (4)
H4	0.468197	0.470160	0.439187	0.046*
C5	0.6463 (2)	0.31108 (18)	0.46597 (16)	0.0465 (4)
H5	0.667655	0.320343	0.528705	0.056*
C6	0.7357 (2)	0.20591 (18)	0.42634 (16)	0.0494 (5)
H6	0.816184	0.147208	0.463225	0.059*
C7	0.7079 (2)	0.18704 (16)	0.33464 (16)	0.0434 (4)
H7	0.767099	0.116699	0.309526	0.052*
C8	0.58685 (19)	0.27800 (14)	0.28061 (13)	0.0317 (3)
C9	0.5777 (2)	0.18468 (15)	0.13328 (16)	0.0423 (4)
H9A	0.484888	0.190437	0.102136	0.051*
H9B	0.592905	0.104685	0.189665	0.051*
C10	0.7385 (2)	0.18906 (17)	0.03817 (15)	0.0479 (5)
H10A	0.745251	0.138493	-0.009454	0.057*
H10B	0.736857	0.273842	-0.009285	0.057*
C11	1.0371 (3)	0.1533 (2)	-0.00127 (18)	0.0589 (5)
H11A	1.050788	0.104364	-0.051121	0.088*
H11B	1.131502	0.123162	0.033640	0.088*
H11C	1.033589	0.238548	-0.044717	0.088*
C12	0.08661 (19)	0.59964 (14)	0.29602 (13)	0.0302 (3)
C13	0.26297 (19)	0.57474 (13)	0.25999 (12)	0.0283 (3)
C14	0.31247 (19)	0.68843 (13)	0.23922 (12)	0.0280 (3)
C15	0.4707(2)	0.72321 (16)	0.19720 (14)	0.0368 (4)
H15	0.571531	0.665444	0.180230	0.044*
C16	0.4739 (2)	0.84446 (17)	0.18159 (16)	0.0465 (4)
H16	0.577949	0.868785	0.153388	0.056*
C17	0.3219 (2)	0.93216 (17)	0.20771 (17)	0.0467 (4)
H17	0.328003	1.013224	0.197145	0.056*
C18	0.1652 (2)	0.90148 (15)	0.24827 (15)	0.0393 (4)
H18	0.065411	0.960070	0.265174	0.047*
C19	0.16130 (19)	0.77873 (14)	0.26325 (13)	0.0291 (3)
C20	-0.1553 (2)	0.78774 (16)	0.32461(15)	0.0398 (4)
H20A	-0.223086	0.746628	0.307457	0.048*
H20B	-0.167379	0.872246	0.275507	0.048*
C21	-0.2260 (2)	0.79165 (17)	0.44748 (16)	0.0469 (4)
H21A	-0.348532	0.824304	0.462347	0.056*
H21B	-0.204899	0.708319	0.497967	0.056*
C22	-0.2125 (3)	0.8773 (2)	0.58304 (18)	0.0694 (6)
H22A	-0.330950	0.918527	0.594549	0.104*
H22B	-0.149746	0.924073	0.596502	0.104*
H22C	-0.201041	0.795560	0.635280	0.104*

displacement factors  $U_{\text{iso}}(\text{H}) = 1.2$  or  $1.5 U_{\text{eq}}$  (parent C-atom).

## Comment

As a special class of sulfur-rich compounds, polysulfide heterocyclic compounds show important biologically active functions and have great potential in the treatment of major diseases such as cancer and HIV [5, 6]. However, it is difficult to characterize the structure of compounds. So, the synthesis and crystal structure of the title compound are of great significance to studying the application.

Single-crystal structure analysis revealed that the title compound crystallized in the triclinic space group  $P\bar{1}$ . The ORTEP diagram is presented in Figure 1. The bond lengths of S–S in the title molecule is 2.0335(7)–2.0817(7) Å, it is similar to reported in the literature [7–10]. The bond lengths of C1–S1 and C12–S4 in the title molecule are 1.7494(16) and 1.7483(15) Å, respectively. They are similar to reported in the literature [11–14].

**Author contributions:** All the authors have accepted responsibility for the entire content of this submitted manuscript and approved submission.

**Research funding:** This work was supported by the Scientific & Technological Projects of Hengyang City (No. 202150063426), Scientific Research Fund of Hunan Provincial Education Department of China (No. 21B0634), Science Foundation of Hengyang Normal University of China (No. 2020QD07) and Key Laboratory of Functional Metal–Organic Compounds of Hunan Province (2022HSKFJJ024).

**Conflict of interest statement:** The authors declare no conflicts of interest regarding this article.

## References

1. Bruker. *SAINT, APEX2 and SADABS*; Bruker AXS Inc.: Madison, WI, USA, 2012.
2. Sheldrick G. M. *SHELXTL* – integrated space-group and crystal-structure determination. *Acta Crystallogr.* 2015, *A71*, 3–8.
3. Sheldrick G. M. Crystal structure refinement with SHELXL. *Acta Crystallogr.* 2015, *C71*, 3–8.
4. Farrugia L. J. WinGX and ORTEP for Windows: an update. *J. Appl. Crystallogr.* 2012, *45*, 849–854.
5. Wolff L., Bandaru S. S. M., Eger E., Lam H.-N., Napierkowski M., Baecker D., Schulzke C., Bednarski P. J. Comprehensive evaluation of biological effects of pentathiepins on various human cancer cell lines and insights into their mode of action. *Int. J. Mol. Sci.* 2021, *22*, 7631.

6. Desat M. E., Kretschmer R. Facile oxidative addition of O<sub>2</sub> and S<sub>8</sub> by an indium bis(carbene) analogue. *Dalton Trans.* 2019, 48, 17718–17722.
7. Janosik T., Stensland B., Bergman J. Sulfur-rich heterocycles from 2-metalated benzo[b]thiophene and benzo[b]furan: synthesis and structure. *J. Org. Chem.* 2002, 67, 6220–6223.
8. Nicolaou K. C., Hwang C. K., DeFrees S., Stylianides N. A. Novel chemistry of dithiatopazine. *J. Am. Chem. Soc.* 1988, 110, 4868–4869.
9. Mancini A., Aragoni M. C., Bingham A. L., Castellano C., Coles S. L., Demartin F., Hursthouse M. B., Isaia F., Lippolis V., Maninchedda G., Pintus A., Arca M. Reactivity of fluoro-substituted bis(thiocarbonyl) donors with diiodine: an XRD, FT—Raman, and DFT investigation. *Chem. Asian J.* 2013, 8, 3071–3078.
10. Aragoni M. C., Arca M., Demartin F., Devillanova F. A., Garau A., Isaia F., Lelj F., Lippolis V., Verani G. New  $[M(R,R'timdt)_2]$  metal-dithiolenes and related compounds (M = Ni, Pd, Pt; R,R'timdt = monoanion of disubstituted imidazolidine-2,4,5-trithiones): an experimental and theoretical investigation. *J. Am. Chem. Soc.* 1999, 121, 7098–7107.
11. Mancini A., Aragoni M. C., Bricklebank N., Castellano C., Demartin F., Isaia F., Lippolis V., Pintus A., Arca M. Formation of T-shaped versus charge-transfer molecular adducts in the reactions between bis(thiocarbonyl) donors and Br<sub>2</sub> and I<sub>2</sub>. *Chem. Asian J.* 2013, 8, 639–647.
12. Bigoli F., Pellinghelli M. A., Atzei D., Deplano P., Trogu E. F. Synthesis of some 4,5,6,7-tetrathiocino [1,2-b:3,4-b'] diimidazolyl-1,3,8,10-tetraethyl-2,9-dithiones and crystal structure of the tetraethyl derivative. *Phosphorus Sulfur Relat. Elem.* 1988, 37, 189–194.
13. Bigoli F., Pellinghelli M. A., Deplano P., Trogu E. F., Sabatini A., Vacca A. Synthesis and characterization of the complexes  $[Cu(II)(Et_4todit)X_2]_n \cdot nTHF$  (Et<sub>4</sub>todit = 4,5,6,7-tetrathiocino[1,2-b:3,4-b']-diimidazolyl-1,3,8,10-tetraethyl-2,9-dithione; X = Cl, Br). Crystal and molecular structure of  $[Cu(II)(Et_4todit)Cl_2]_n \cdot nTHF$ . *Inorg. Chim. Acta* 1991, 180, 201–207.
14. Bigoli F., Pellinghelli M. A., Deplano P., Trogu E. F. Complexes of 4,5,6,7-tetrathiocino[1,2-b:3,4-b']diimidazolyl-1,3,8,10-tetraethyl-2,9-dithione (Et<sub>4</sub>todit) with group IIb metal halides. Crystal and molecular structure of  $(Cd(II)Et_4toditCl_2)_n$ . *Inorg. Chim. Acta* 1990, 170, 245–249.