melting point results in a break-up and spheroidization of the iron rods together with diffusion of iron to the surface and grain boundaries of specimens. The driving force for these processes is a tendency by the system to minimize its interfacial energy. The spheroidizing of iron rods as a means of decreasing the interfacial energy, is aided by the absence of preferred crystallographic orientations. The eutectic is considerably more thermally unstable than both the Al-$CuAl_2$ and Al-Al_3Ni eutectic possibly because of:

a) its higher interfacial energy, and

b) the higher heat treatment temperatures used.

The stability of the eutectic structures was found to be a function of:

a) time and temperature of heat treatment,

b) interrod spacings, and

c) oxygen content of the alloys.

Point a) affects the diffusion characteristics of the system, whereas b) and c) affect the initial interfacial energy of the system.

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Communication: *Some Aspects of the Annealing Behavior of Indium*, by Cuppam Dasarathy, pp. 1784-86.

Page 1784

Column 1, paragraph 2, line 2, should read: ". . . by 63 pct at liquid nitrogen temperature. The deformed . . ."

The Effect of Carbide and Nitride Additions on the Heterogeneous Nucleation Behavior of Liquid Iron, by Bruce L. Bramfitt, pp. 1987-95.

Page 1990

Column 1, paragraph 4, line 4, should read: ". . . with the [110] Fe ‖ [100] TiC . . ."

Page 1991

Column 1, top line, should read: ". . . In this case the (110) of δ iron . . ."

Page 1993

Appendix, Case 1, first line, should read: "As shown in Fig. 9, the (110) of δ iron . . ."