

## Supplementary Data

Results of elemental analysis of all the compounds of the synthesized series  $In_{a-e}$  were found to be satisfactory (S1). Since almost identical infrared and  $^1H$  NMR spectra were observed for all the members of the five homologous series ( $In_a - In_e$ ), FT-IR and  $^1H$  NMR spectral data of the *n*-octyloxy derivatives ( $I8_{a-e}$ ), as examples, are given below:

For  $I8_a$ , FT-IR (KBr,  $\nu_{\text{max.}} \text{ cm}^{-1}$ ): 3067 (C–H aromatic) 2923, 2853 (alkyl group), 1506 (–C=C– aromatic), 1732, 1252 (–COO– group), 1601 (–N=N– group), 839 (p-sub. benzene rings).

$^1H$  NMR (300 MHz,  $\text{CDCl}_3$ )  $\delta$  ppm: 0.89 (t, 3H, –CH<sub>3</sub>), 1.28–1.77 (m, 12H, 6  $\times$  –CH<sub>2</sub>–), 4.08–4.10 (t, 2H, –OCH<sub>2</sub>–), 3.79–3.81 (s, 3H, Ar–OCH<sub>3</sub>), 7.02–7.25, 7.95–8.02, 8.29–8.32 (m, 12 H, p-subst. benzene rings).

For  $I8_b$ , FT-IR (KBr,  $\nu_{\text{max.}} \text{ cm}^{-1}$ ): 3046 (C–H aromatic) 2919, 2851 (alkyl group), 1505 (–C=C– aromatic str.,), 1734, 125160 (–COO– group), 1601 (–N=N– group), 840 (p-sub. benzene rings).

$^1H$  NMR (300 MHz,  $\text{CDCl}_3$ )  $\delta$  ppm: 0.88 (t, 3H, –CH<sub>3</sub>), 1.15–1.77 (m, 12H, 6  $\times$  –CH<sub>2</sub>–), 2.35–2.44 (s, 3H, Ar–CH<sub>3</sub>), 4.11 (t, 2H, –OCH<sub>2</sub>–), 7.17–7.27, 7.94–8.01, 8.32 (m, 12 H, p-sub. benzene rings).

For  $I8_c$ , FT-IR (KBr,  $\nu_{\text{max.}} \text{ cm}^{-1}$ ): 3062(C–H aromatic) 2921, 2853 (alkyl group), 1496 (–C=C– aromatic), 1733, 1259 (–COO– group), 1600 (–N=N– group), 839 (p-sub. benzene rings).

$^1H$  NMR (300 MHz,  $\text{CDCl}_3$ )  $\delta$  ppm: 0.87–0.88 (t, 3H, –CH<sub>3</sub>), 1.28–1.77 (m, 12H, 6  $\times$  –CH<sub>2</sub>–), 4.10–4.14 (t, 2H, –OCH<sub>2</sub>–), 7.14–7.51, 7.90–8.02, 8.3–8.4 (m, 12 H, p-sub. benzene rings).

For  $I8_d$ , FT-IR (KBr,  $\nu_{\text{max.}} \text{ cm}^{-1}$ ): 3063 (C–H aromatic) 2920, 2851 (alkyl group), 1487 (–C=C– aromatic), 1733, 1258 (–COO– group), 1599 (–N=N– group), 1061 (C–Br aromatic), 842 (p-sub. benzene rings).

$^1H$  NMR (300 MHz,  $\text{CDCl}_3$ )  $\delta$  ppm: 0.88 (t, 3H, –CH<sub>3</sub>), 1.30–1.77 (m, 12H, 6  $\times$  –CH<sub>2</sub>–), 4.12 (t, 2H, –OCH<sub>2</sub>–), 7.15–7.36, 7.67, 7.97–8.30 (m, 12 H, p-sub. benzene rings).

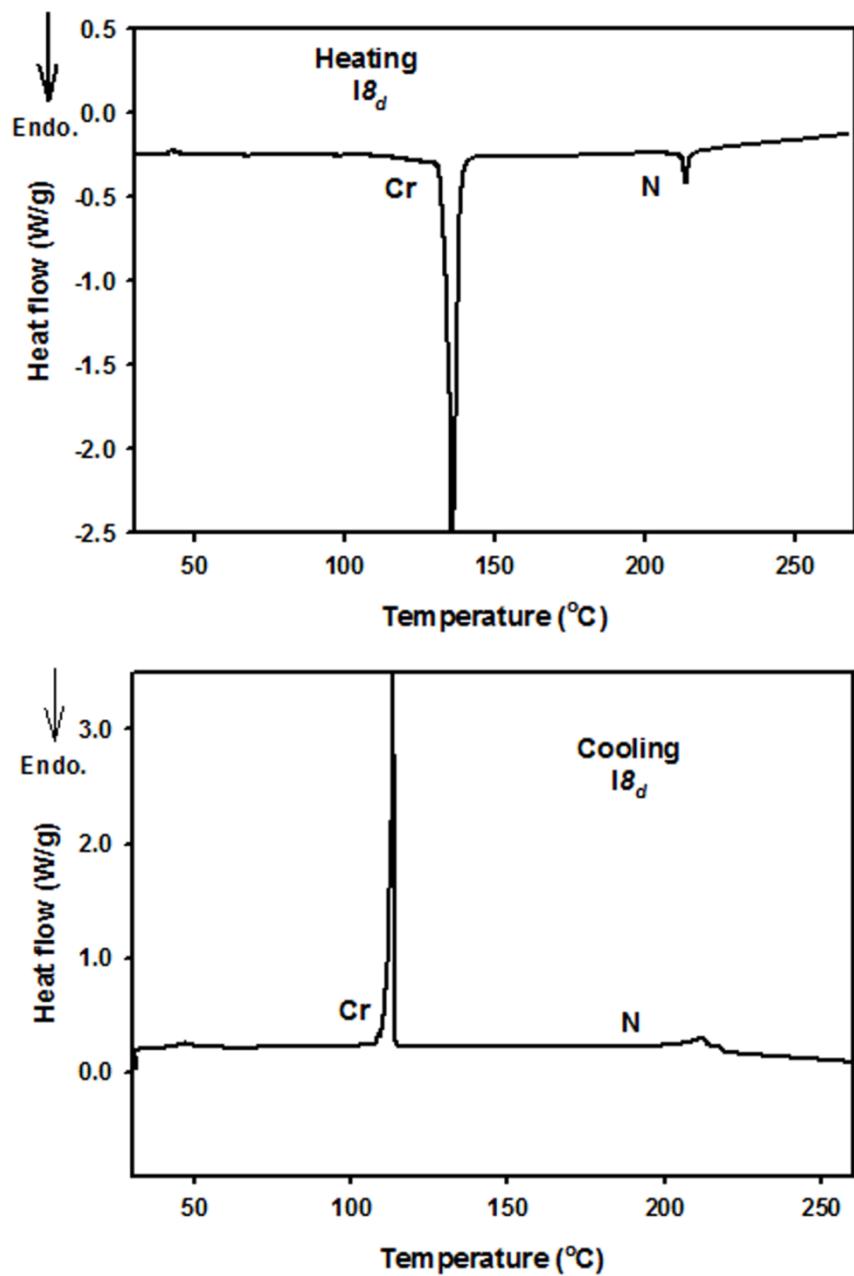
For  $I8_e$ , FT-IR (KBr,  $\nu_{\text{max.}} \text{ cm}^{-1}$ ): 3068 (C–H aromatic) 2923, 2854 (alkyl group), 2238 (–CN), 1501 (–C=C– aromatic), 1732, 1253 (–COO– group), 1601 (–N=N– group), 836 (p-sub. benzene rings).  $^1H$  NMR (300 MHz,  $\text{CDCl}_3$ )  $\delta$  ppm: 0.87 (t, 3H, –CH<sub>3</sub>), 1.28–1.76 (m, 12H, 6  $\times$  –CH<sub>2</sub>–), 4.08–4.10 (t, 2H, –OCH<sub>2</sub>–), 7.14–7.16, 7.58–7.60, 7.97–8.01, 8.31–8.33 (m, 12 H, p-sub. benzene rings).

S1: Elemental analyses of the azo/esters, **In<sub>a-f</sub>**.

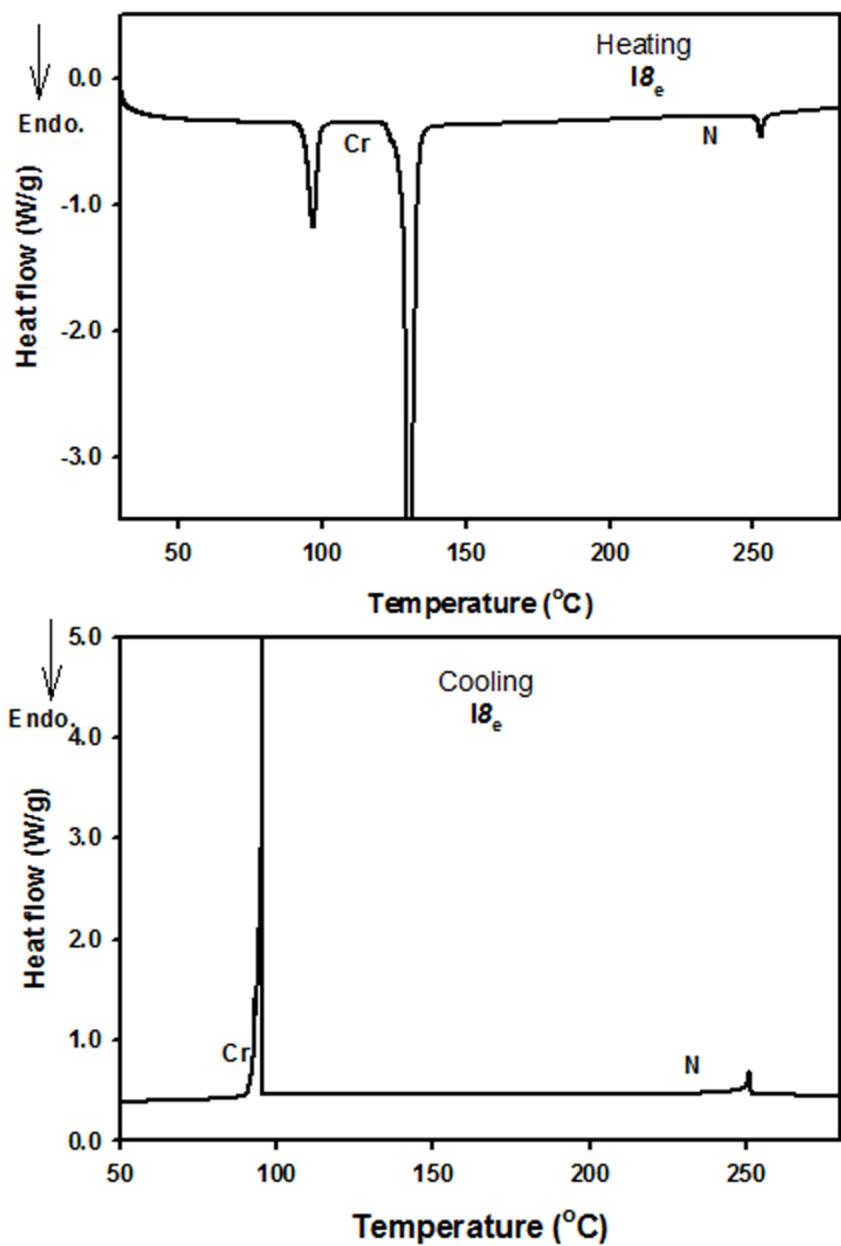


Comp No.	N	X	Mol. Wt.	Elemental Analyses Cal. (Found)			
				% C	% H	% N	% Br
<b>I8<sub>a</sub></b>	8	CH <sub>3</sub> O	460.58	73.02 (73.15)	7.00 (7.21)	6.08 (5.97)	-
<b>I8<sub>b</sub></b>	8	CH <sub>3</sub>	444.58	75.65 (75.74)	7.26 (7.34)	6.30 (6.13)	-
<b>I8<sub>c</sub></b>	8	H	430.55	75.32 (75.41)	6.97(7.21)	6.51 (6,64)	-
<b>I8<sub>d</sub></b>	8	Br	509.45	63.37 (63.12)	5.74 (5.83)	5.50 (5.42)	15.69(15.81)
<b>I8<sub>e</sub></b>	8	CN	455.56	73.82 (74.01)	6.42 (6.28)	9.22 (9.01)	-
<b>II10<sub>a</sub></b>	10	CH <sub>3</sub> O	488.80	73.72 (73.60)	7.42 (7.45)	5.73 (5.91)	-
<b>II10<sub>b</sub></b>	10	CH <sub>3</sub>	472.63	76.24 (76.40)	7.68 (7.45)	5.93 (5.76)	-
<b>II10<sub>c</sub></b>	10	H	458.60	75.95 (75.81)	7.47 (7.35)	6.11 (5.97)	-
<b>II10<sub>d</sub></b>	10	Br	537.50	64.48 (64.61)	6.19 (6.34)	5.21 (5.03)	14.87 (14.70)
<b>II10<sub>e</sub></b>	10	CN	483.61	74.51(74.38)	6.88 (8.62)	5.79 (5.63)	-
<b>II12<sub>a</sub></b>	12	CH <sub>3</sub> O	516.85	74.36 (74.24)	7.80 (8.12)	5.42 (5.24)	-
<b>II12<sub>b</sub></b>	12	CH <sub>3</sub>	500.68	76.80 (76.63)	8.01(7.96)	5.60 (5,45)	-
<b>II12<sub>c</sub></b>	12	H	486.85	76.48 (76.32)	7.87(7.69)	5.75 (5.68)	-
<b>II12<sub>d</sub></b>	12	Br	565.56	65.83 (66.12)	6.59 (6.45)	4.95 (5.21)	14.13 (14.34)
<b>II12<sub>e</sub></b>	12	CN	511.67	75.12 (74.93)	7.29 (7.36)	8.21(8.45)	-
<b>II14<sub>a</sub></b>	14	CH <sub>3</sub> O	544.87	74.94 (75.22)	8.14 (8.25)	5.14 (5.32)	-
<b>II14<sub>b</sub></b>	14	CH <sub>3</sub>	528.74	77.24 (77.51)	8.39 (8.48)	5.30 (5.16)	-
<b>II14<sub>c</sub></b>	14	H	513.91	77.13 (77.39)	8.24 (8.11)	5.45 (5.62)	-
<b>II14<sub>d</sub></b>	14	Br	593.61	66.77 (77.55)	6.96 (7.16)	4.72 (4.91)	13.46 (13.71)
<b>II14<sub>e</sub></b>	14	CN	539.72	75.66 (75.34)	7.66 (7.43)	7.79 (7.92)	-
<b>II16<sub>a</sub></b>	16	CH <sub>3</sub> O	572.92	75.47 (75.52)	8.27 (8.41)	4.89 (4.76)	-
<b>II16<sub>b</sub></b>	16	CH <sub>3</sub>	556.79	77.66 (77.89)	8.51(8.78)	5.03 (4.86)	-
<b>II16<sub>c</sub></b>	16	H	541.96	77.57 (77.34)	8.37 (8.55)	5.17 (5.03)	-
<b>II16<sub>d</sub></b>	16	Br	621.66	67.62 (67.47)	7.13 (7.31)	4.51 (4.73)	12.85(13.11)
<b>II16<sub>e</sub></b>	16	CN	567.77	76.16 (76.41)	7.81(8.03)	7.40 (7.24)	-

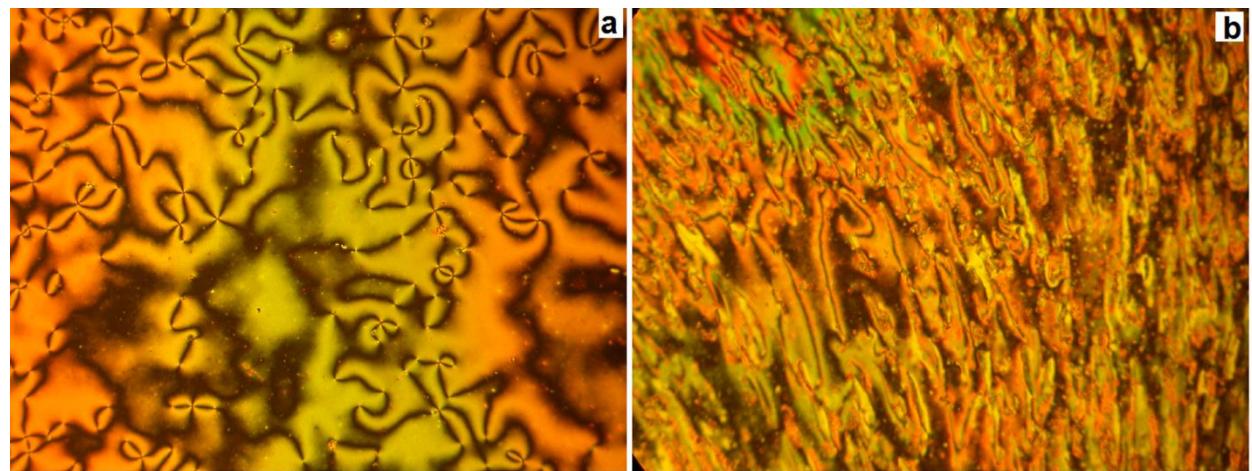
DSC-Curves



Supplementary figure 1. Heating and cooling DSC scans for  $\text{I8}_d$ .



Supplementary figure 2. Heating and cooling DSC scans for  $\text{I8}_\text{e}$ .



Supplementary figure 3. Polarized optical micrographs obtained from cooling of isotropic phases of (a) nematic phase of **I8<sub>d</sub>** at 200°C and (b) of **I8<sub>e</sub>** at 220°C.