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THE INFLUENCE OF TEMPERATURE ON FATIGUE-CRACK GROWTH  
IN A MILL-ANNEALED Ti-6Al-4V ALLOY

by

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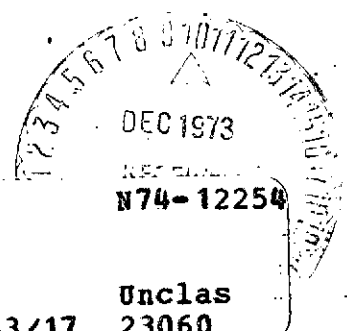
ABSTRACT

To understand the influence of temperature on the rate of fatigue-crack growth in high-strength metal alloys, constant-load-amplitude fatigue-crack growth experiments were carried out using a 1/4-inch-thick (6.35 mm) mill-annealed Ti-6Al-4V alloy plate as a model material. The rates of fatigue-crack growth were determined as a function of temperature, ranging from room temperature to about 290C (or, about 550F/563K) and as a function of the crack-tip stress-intensity factor  $K$ , in a dehumidified high-purity argon environment. The results indicate that the rate of fatigue-crack growth for  $K$  from 10 ksi  $\sqrt{\text{in.}}$  to 30 ksi  $\sqrt{\text{in.}}$  (11 to 33 MN- $\text{m}^{-3/2}$ ), corresponding to growth rates from  $2 \times 10^{-7}$  to  $4 \times 10^{-4}$  inch per cycle ( $5.08 \times 10^{-6}$  to  $1.16 \times 10^{-2}$  mm/cycle) are essentially independent of test temperature in this range. The dependence of the rate of fatigue-crack growth on  $K$  appears to be separable into two regions, with a transition occurring in the range of  $2$  to  $3 \times 10^{-6}$  inch per cycle ( $5.08$  to  $7.62 \times 10^{-5}$  mm/cycle). The transition correlates well with changes in both the microscopic and macroscopic appearances of the fracture surfaces, and suggests a change in the mechanism and the influence of microstructure on fatigue-crack growth.

Limited correlative experiments indicate that dehumidified oxygen and hydrogen have no effect on the rate of fatigue-crack growth in this alloy, while distilled water increased the rate of crack growth slightly in the range tested. Crack growth in vacuum (less than  $5 \times 10^{-6}$  torr) was about one-half that observed in the dehumidified argon environment. Mass spectrometric analysis and other experiments suggest that this difference is produced by residual moisture (well below 30 ppm) in the argon atmosphere. The possible influence of this residual moisture on the observed temperature independence is discussed.

Companion fractographic examinations suggest that the mechanisms for fatigue-crack growth in the various environments are essentially the same. The observation of ductile striations on specimens tested in vacuum ( $4.4 \times 10^{-7}$  to  $2.9 \times 10^{-9}$  torr) is not in agreement with previous investigations on aluminum alloys. Possible reasons for this discrepancy are discussed.

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## I. INTRODUCTION

The importance of fatigue-crack growth resistance in determining the serviceable lives of aircraft structures has been well recognized [1]\*. With the development of aircrafts to operate at flight speeds in excess of the speed of sound, such as the supersonic transport (SST), the fatigue-crack growth resistance of materials over a wide range of temperatures, associated with aerodynamic heating, needs to be considered. Recent investigations have shown that atmospheric moisture and other environments (such as water, salt water, salt and certain organic liquids) can also have a significant effect on the rate of fatigue-crack growth in high-strength alloys [2-16] and that these environmental influences are affected by temperature [10]. Thus, the performance of structures in service will depend on the complex interactions between loads, temperatures and environments encountered during the entire flight profile.

Fatigue-crack growth resistance of several stainless steels and titanium alloys at room temperature and at 550 F ( or, about 290 C/ 563 K) has been evaluated by Hudson [17]. For some of the materials, fatigue-crack growth resistance at -109F (or, about -78C/195K) was also determined [17]. No consistent pattern of behavior with temperature was evident. Since these tests were conducted in air, the results may reflect the combined influences of atmospheric moisture and temperature [10, 11]. In a recent series of experiments, the effect of temperature on the rate of fatigue-crack growth in a 7075-T651 aluminum alloy was examined over a range of temperatures from room temperature to approximately 100C (373K/212F) in distilled water and in dehumidified hydrogen and oxygen environments [10]. Dehumidified high-purity argon (99.9995 percent purity) was used as a reference environment. The results show that fatigue-

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\* See References.

crack growth is controlled by thermally activated processes with apparent activation energies that depend strongly on the crack-tip stress-intensity factor,  $\Delta K$  or  $K$ , given by linear elasticity for all of the test environments [10,11]. The strong effect of  $\Delta K$  on the apparent activation energy for crack growth implies that the influence of temperature on the rate of fatigue-crack growth will be a function of  $\Delta K$ ; the effect of temperature being stronger at the lower  $\Delta K$  levels [10,11]. These findings suggest that the effect of temperature might be incorporated explicitly in design through an absolute rate theory consideration of the fatigue-crack growth process. Verification of the general applicability of this concept and further quantitative development of this approach would be of both technological and fundamental importance.

Since only a limited amount of information of this type is available, an experimental program was initiated to determine the rate of fatigue-crack growth on a single high-strength alloy over a wide range of temperatures and crack growth rates. The experiments were carried out principally in a dehumidified high-purity argon atmosphere to "eliminate" the interaction effects of aggressive environments. Annealed Ti-6Al-4V alloy was selected as model material for this study. Test temperature ranging from room temperature to about 290C (or about 563K/550F) were used. Companion fractographic examinations of selected specimens were made to determine possible changes in crack morphology (or cracking mechanism) with changes in test temperature and environment. The fatigue-crack growth and the fractographic studies will be discussed separately.

## II. THE KINETICS OF CRACK GROWTH

### MATERIAL AND EXPERIMENTAL WORK

#### Material and Specimens

$\frac{1}{4}$ -inch-thick (6.35 mm) plates of mill-annealed Ti-6Al-4V alloy were used in this investigation. \* The nominal chemical composition of this material is given in Table 1. Longitudinal and transverse tensile-test specimens and 3-inch-wide (76.2 mm) by 12-inch-long (305 mm) and single-edge notch (SEN) fracture-test specimens were prepared in triplicate and tested. The tensile properties and crack growth resistance curves (for monotonic loading) are shown in Table 1 and Figure 1 respectively, and are typical of this alloy in the mill-annealed condition. These results suggest that the material was cross rolled, with a cross rolling ratio of nearly 1:1. Optical micrographs show representative  $\alpha$ - $\beta$  structure for this alloy, Figure 2.

Three-inch-wide (76.2 mm) by 14-inch-long (356 mm) center-cracked specimens, oriented in the LT (or RW) orientation,\*\* were used for the fatigue-crack growth studies. The initial center notch, about 0.4 inch (about 10 mm) long, was introduced by broaching. The specimens were precracked in air, or in "ultra-high-purity" grade argon (99.999 percent purity) at a stress ratio R of 0.05 through a sequence of loads that reduce  $\Delta K$  to

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\* Heat No. D-4987. Material conforms to AMS 4911 specification.

\*\* ASTM Committee E-24 on Fracture Testing of Metals has adopted the following convention to denote the plane and direction of cracking [18]. The first letter, in the two letter coding, denotes the direction normal (perpendicular) to the macroscopic crack plane, while the second denotes the direction of crack growth. R, W and T have been used to denote the rolling (longitudinal), width (transverse) and thickness (short transverse) directions respectively. To provide greater consistency and clarity, Committee E-24 will adopt the designations L, T and S, respectively, for these three principal directions in rolled plate product. The new designations are used in this report. For clarity, the old designations are also included parenthetically.

a level that is equal to or less than the selected starting  $\Delta K$  level for the actual experiment. This precracking procedure provided fatigue-cracks of about 0.08 inch (2 mm) in length from the ends of the starter notches, such that subsequent fatigue-crack growth will be through material that has not been altered by the notch preparation procedure and will be unaffected by the starter notch geometry.

### Test Environment

Dehumidified Matheson "Research" grade argon (99.9995 percent purity) was used as the principal test environment. Dehumidified Matheson "Ultra-Pure" grade oxygen (99.95 percent purity) and hydrogen (99.999 percent purity) and distilled water were also used to provide supplemental information. The dehumidified gaseous environments were obtained by passing the respective gases through a gas purifier (Matheson Model 460 purifier with Model 461-R cartridge for moisture), then through a series of cold traps at less than  $-140^{\circ}\text{C}$  ( $\sim 130\text{K}$ ), and finally through a silicone fluid back-diffusion trap and discharged. Distilled water used in the experiments was triple-distilled water purchased commercially. Because of the highly reactive nature of the titanium alloys, limited correlative experiments were conducted to compare test results obtained in dehumidified high-purity argon with those obtained in vacuum at  $2-5 \times 10^{-7}$  torr. \*

### Experimental Procedures

The fatigue-crack growth experiments were carried out under axial loading at  $R = 0.05$  in an 100,000-lb. capacity MTS Systems closed-loop electro-hydraulic

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\* Experiments in vacuum were performed at NASA Langley Research Center.

testing machine operated at either 5 or 10 Hz. \* Load control was estimated to be better than  $\pm 1$  percent. After precracking the specimen, appropriate environmental chambers were clamped in place [7]. For tests in distilled water, cup-type chambers were used. These were simply filled with distilled water. Testing was started after a short period for instrument stabilization. For the dehumidified gaseous environments, a rigorous purging procedure was followed. The gas train was purged with "Ultra-High-Purity" grade argon for at least 30 minutes. During this initial purging operation, the various components of the gas train up-stream from the back-diffusion trap were heated to at least 100C (373K). The cold-traps were then filled with liquid nitrogen, and the environment chamber-specimen assembly was again heated to a minimum of 100C (373K) while maintaining the argon flow. \*\* The specimen was then brought to the desired test temperature by means of electrical resistance heating tapes. (Temperature stability was better than  $\pm 2$ C during a working day.) Appropriate test environment was introduced and was allowed to flow through the system for at least 15 minutes prior to the start of the experiment. Continuous flow at a chamber pressure of about 5 psi (35 kN/m<sup>2</sup>) above ambient was maintained throughout the experiment.

A continuous-recording electrical potential system was used for monitoring crack growth. The detailed experimental procedure and calibration of this method have been described elsewhere [19, 20]. Using a working current of 2.5 amperes, this system provides an average measurement sensitivity of about 0.003 inch (0.076 mm) in half-

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\* Comparison experiments indicated that test frequencies from 1 to 15 Hz. had negligible effect on the rate of fatigue-crack in this material tested in a dehumidified argon environment.

\*\* This step was combined with specimen heating whenever the test temperature was to exceed 100C (373K).

crack length (a) per microvolt ( $\mu\text{v}$ ) change in potential (0.003 in./ $\mu\text{v}$  or 0.076 mm/ $\mu\text{v}$ ), for the  $\frac{1}{8}$ -inch-thick (6.35 mm) Ti-6Al-4V alloy specimens. The variation of measurement sensitivity with crack length for a typical specimen is shown in Figure 3; the actual sensitivity varies somewhat with the thickness of each specimen. Resolution is better than 0.0015 inch (0.038 mm) in half-crack length, a. For comparison, both the electrical potential method and a visual method, using a photogrid technique [17] were used to monitor crack growth on a single specimen. The results show the good agreement between these two methods, Figure 4. (Note that corrections for crack front curvature were needed in making this comparison, since the visual method measures the crack lengths at the specimen surface whereas the electrical potential method provides an average through the specimen thickness).

## RESULTS AND DISCUSSIONS

### Fatigue-Crack Growth Results

Fatigue-crack growth rate data were determined from the slopes of the electrical potential records and correlated with the crack-tip stress-intensity parameter  $\Delta K$ .

$$\Delta K = \Delta\sigma \sqrt{\pi a} \cdot f(a/W) \quad (1)$$

where  $\Delta\sigma$  = range of applied gross-section stress

a = half-crack length

f(a/W) = correction factor for finite-width specimen [21, 22].

W = specimen width

Data for tests conducted in dehumidified argon at temperatures from room temperature to about 290C (or about 70 to 550F/300 to 563K) are shown in Figure 5. These results covered a range of stress-intensity parameters ( $\Delta K$ ) from about 10 to 35 ksi $\sqrt{\text{in}}$ . (11 to 38.5 MN-m $^{-3/2}$ ) with corresponding crack growth rates ( $\Delta a/\Delta N$ ) from about  $2 \times 10^{-7}$



to  $5 \times 10^{-4}$  inch per cycle ( $5.08 \times 10^{-6}$  to  $1.27 \times 10^{-2}$  mm/cycle). Reproducibility was generally within  $\pm 10$  percent. Larger variations at the lower growth rates were principally caused by drift in the crack length monitoring system, associated with small changes in specimen temperature, and possible delay produced by the precracking procedure [24].

The growth rates for selected values of  $\Delta K$ , at the various temperatures, are shown in a standard Arrhenius plot in Figure 6. It is seen that the rates of fatigue-crack growth, for  $\Delta K$  from 10 to 30 ksi  $\sqrt{\text{in}}$ . ( $11$  to  $33 \text{ MN}\cdot\text{m}^{-3/2}$ ) are essentially independent of temperature in the range of 20 to 290C (or, about 70 to 550F/300 to 563K), in contradistinction to that of the 7075-T651 aluminum alloy [10]. The dependence of the rate of fatigue-crack growth on  $\Delta K$  appears to be separable into two regions, with a transition occurring in the range of  $2$  to  $3 \times 10^{-6}$  inch per cycle ( $5.08$  to  $7.63 \times 10^{-5}$  mm/cycle), as shown typically in Figure 8. Using a piece-wise power-law [23] fit to the experimental data,

that is

$$\frac{\Delta a}{\Delta N} = C (\Delta K)^n \quad (2)$$

the exponent  $n$  ranges from 7 to 10 for growth rates ( $\Delta a/\Delta N$ ) between  $2 \times 10^{-7}$  and  $2 \times 10^{-6}$  inch per cycle ( $5.08 \times 10^{-6}$  to  $5.08 \times 10^{-5}$  mm/cycle), and is about 3 in the range  $2 \times 10^{-6}$  to  $5 \times 10^{-4}$  inch per cycle ( $5.08 \times 10^{-5}$  to  $1.27 \times 10^{-2}$  mm/cycle),

Figure 5. This transition suggests a change in the mechanism of fatigue-crack growth and correlates well with a change in the macroscopic appearance of the fracture surface, Figure 7. The fracture is quite coarse (macroscopically) at growth rates below the transition range of  $2$  to  $3 \times 10^{-6}$  inch per cycle ( $5.08$  to  $7.62 \times 10^{-5}$  mm/cycle) and becomes smooth at the higher rates of growth. A more detailed discussion of the changes in fracture mechanism will be given in the fractographic section (Section III) of this paper.

To complement these studies, a limited number of experiments were carried out in dehumidified hydrogen and oxygen and in distilled water to examine the effects of these environments on the rate of fatigue-crack growth, and a correlative study was conducted to determine if any differences existed between data obtained in dehumidified high-purity argon and those obtained in vacuum at  $10^{-6}$  to  $10^{-7}$  torr. The results of these studies are shown in Figures 8 to 11. An additional test was carried out at  $-61^{\circ}\text{C}$  ( $-78^{\circ}\text{F}$ / $212\text{K}$ ) and  $2.9 \times 10^{-9}$  torr; the results of this test are shown in Figure 12. It is seen that data obtained in dehumidified oxygen (Figure 8) and hydrogen (Figure 9) are nearly identical to those for dehumidified argon for  $\Delta K$  ranging from about 15 to 30 ksi/in. ( $16.5$  to  $33 \text{ MN}\cdot\text{m}^{-3/2}$ ). For the same range of  $\Delta K$ , distilled water increased the rate of fatigue-crack growth by about 50 percent at low  $\Delta K$  levels, whereas at the higher  $\Delta K$  levels the rate of growth approached that for dehumidified argon, Figure 10. The results suggest that the rate of fatigue-crack growth in these environments are again independent of temperature.

Data obtained in vacuum ( $2.1 \times 10^{-7}$  and  $4.4 \times 10^{-7}$  torr) at room temperature are some 30 to 50 percent slower than those obtained in dehumidified argon at low  $\Delta K$  levels and tended to converge with the dehumidified argon results at the higher  $\Delta K$  levels, Figure 11. Careful rechecking of testing machine calibration and experimental procedures and additional comparative experiments, using distilled water from a single source as the test environment, showed that these differences are real and significant (the interlaboratory reproducibility being approximately 10 percent). On the basis of these experiments, a mass spectrometric analysis of the dehumidified argon atmosphere was initiated, and further purification of the gases was attempted (see following discussions). Test results obtained in vacuum at  $-61^{\circ}\text{C}$  ( $-78^{\circ}\text{F}$ / $212\text{K}$ ) were essentially the

same as those obtained at room temperature, and suggest that the temperature independence may be extended to this low temperature.

#### Mass Spectrometric Analysis and Companion Experiments

A sample for mass spectrometric analysis was collected by inserting a collection bulb just down-stream of the environment chamber. The standard purging procedure was used with the exception that the system was evacuated with a mechanical pump and back-filled with argon several times before the regular purging sequence to ensure removal of air from the collection bulb. The sample was analyzed in a Hitachi mass spectrometer.\* The results indicate that the amount of nitrogen and oxygen in the dehumidified argon sample was less than 30 and 8 ppm (parts per million) respectively. These contaminants may be introduced by the sample collection procedure (i. e., from air trapped in the stop-cocks), during transfer into the mass spectrometer, or by possible leakage in the environmental system. The determination of moisture level was much more uncertain. The amount of moisture present could not be resolved from the background moisture level in the instrument. Based on the sensitivity limits of the instrument, it was estimated that the moisture content was well below 30 ppm.

Concurrent with this examination, attempts to further purify the test environment were made. It was found that by using a titanium sublimation pump as a getter in the argon stream, reduction in the rate of fatigue-crack growth (comparable to the percentage difference between vacuum and dehumidified argon results), was obtained on another mill-annealed Ti-6Al-4V alloy plate.

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\* Analysis performed by Dr. J. Sturm, Department of Chemistry, Lehigh University.

On the basis of these results, it is clear that residual impurities at levels below 30 ppm can still affect fatigue-crack growth in this titanium alloy. Since the influence of oxygen at ~1 atmosphere (Figure 8) did not significantly increase the rate of fatigue-crack growth, the active impurity is considered to be moisture (water vapor) present in the argon atmosphere.

### Discussion

The absence of temperature dependence for the rate of fatigue-crack growth in the mill-annealed Ti-6Al-4V alloy in the range of  $2 \times 10^{-7}$  to  $5 \times 10^{-4}$  inch per cycles ( $5.08 \times 10^{-6}$  to  $1.27 \times 10^{-2}$  mm/cycle) should be interpreted with care. Although the results are in general agreement with that reported by Hudson [17] for a Ti-8Al-1V-1Mo alloy, the apparent sensitivity of this titanium alloy to residual moisture of less than 30 ppm did not permit a clear resolution of the problem that the influence of temperature associated with deformation may have been compensated by that resulting from environmental embrittlement in a partially saturated environment [10, 25]. If one can assume that the effect of environment has a decreasing branch with temperature as indicated by Johnson and Willner [25] for environments containing a fixed amount of moisture, then the fatigue-crack growth in a truly inert environment may be shown to be dependent on temperature. This particular point still needs to be investigated.

The apparent lack of temperature dependence in the range of growth rates from  $2 \times 10^{-7}$  to  $5 \times 10^{-4}$  inch per cycle ( $5.08 \times 10^{-6}$  to  $1.27 \times 10^{-2}$  mm/cycle) does not imply complete temperature independence over the entire range of growth rates. For growth rates in excess of about  $10^{-4}$  inch per cycle ( $2.5 \times 10^{-3}$  mm/cycle), crack growth approaches the onset of rapid fracture or fracture instability, and is expected to be

related to the fracture toughness of the material,  $K_{Ic}$  or  $K_c$ . Since fracture toughness is known to be dependent on temperature, some influence of temperature on the rate of fatigue-crack growth at these high rates would be expected. It is interesting to note that, by comparing Figures 1 and 11, the value of  $K_{max}$  ( $K_{max} = \Delta K / (1-R)$ ) for transition to rapid fatigue-crack growth (i. e.  $\Delta a / \Delta N > 10^{-4}$  in/cycle or  $2.5 \times 10^{-3}$  mm/cycle) corresponds to the K level for the onset of crack growth under monotonic loading.

Practically, the observed temperature independence permits a degree of simplification in estimating the fatigue performance of structures intended for service at various temperatures. However, the possible interactions between temperature and load under conditions of changing loads and temperatures encountered during service may be quite significant and must be carefully explored. This problem is being investigated currently.

### III. FRACTOGRAPHY

#### MATERIAL AND EXPERIMENTAL WORK

Selected specimens, tested in the various environments and at different temperatures were examined by means of optical microscopy and electron-microfractography to determine possible changes in crack morphology, or cracking mechanism, and the relationship between crack paths and microstructure.

For electron-microfractography, two-stage plastic-carbon replicas shadowed with platinum-carbon were used. The shadowing direction was along the direction of crack prolongation. Replicas were taken from regions of the fracture surface corresponding to observed macroscopic crack growth rates of 1 to  $3 \times 10^{-5}$  inch per cycle (2.54 to  $7.62 \times 10^{-4}$  mm/cycle), and to those above and below the observed macroscopic transition region, 2 to  $3 \times 10^{-6}$  inch per cycle ( $5.08$  to  $7.62 \times 10^{-5}$  mm/cycle), Figures 5 and 7. They were examined in a RCA EMU-3G electron microscope operated at 50 or 100 kV. Specimen tilting was used to enhance contrast [26]; tilt angles up to  $30^\circ$  were used.

In pertinent cases, specimens were plated with nickel and then sectioned to examine the interaction of the propagating crack with the alloy microstructure by optical microscopy.

#### RESULTS AND DISCUSSIONS

##### Effects of Environment and Temperature

Typical electron-micrographs of fatigue fracture surfaces of specimens tested in the various environments and temperatures are shown in Figure 13. The results indicate that there are no significant differences in the failure mode and fracture paths for specimens tested at different temperatures and in the different environments, with ductile

fatigue striations as the predominant feature in all cases. The striation spacings correspond, within expected accuracy, to the observed macroscopic crack growth rate of about  $2 \times 10^{-5}$  inch per cycle ( $5 \times 10^{-4}$  mm/cycle). These observations are consistent with the fact that there were no significant influence of test environment and temperature on the rate of fatigue-crack growth for this alloy.

The observation of ductile striations on specimens of Ti-6Al-4V alloy tested in vacuum (at  $4.4 \times 10^{-7}$  to  $2.9 \times 10^{-9}$  torr) is not in agreement with the results reported by Pelloux [27] and Meyn [28] for aluminum alloys. These workers suggested that the mechanism for fatigue-crack growth is different from that in air, and therefore, should not lead to striation formation. Two probable reasons can be cited to account for the apparent discrepancy, aside from the fact that different alloys are involved. Both reasons are related to the ability to clearly resolve striations by replication electron microscopy. First, the striations in specimens tested in vacuum have a flattened and smeared appearance, Figure 13(c) and (d), and hence, would be more difficult to resolve. Broek showed that by tilting the replica with respect to the electron beam, enhanced contrast may be obtained [26]; regions that appeared relatively "featureless" under one set of viewing conditions were shown to contain fatigue striations and other structural features. This technique was utilized to examine replicas from specimens tested in vacuum. The results indeed demonstrate that, under certain orientations, regions that contain striations can be made to appear nearly featureless, Figure 14. The second possibility is that the rate of fatigue-crack growth, and hence, the striation spacing, could be well below the resolution limits of replication electron-microscopy technique [29]; the limit has been variously estimated at some 200 to 500 Å. This may well be the case for the

work on aluminum alloys reported by Pelloux [27] and Meyn [28]. Since these alloys are known to be quite sensitive to the effect of atmospheric moisture, an order of magnitude reduction in growth rate associated with a change in test environment, from air to vacuum, is possible [2,3,10,12]. The striations, if present, at the reduced rates of growth may not have been resolvable. Regardless, it is clear that the mechanisms for fatigue-crack growth in the Ti-6Al-4V alloy are essentially the same for all the environments investigated.

### Low Growth Rates

For specimens tested at low  $\Delta K$  levels, corresponding to growth rates below about  $2 \times 10^{-6}$  inch per cycle ( $5 \times 10^{-5}$  mm/cycle), a macroscopically observable coarse region of crack growth was observed (see Figure 7). Above approximately  $2 \times 10^{-6}$  inch per cycle ( $5 \times 10^{-5}$  mm/cycle), the macroscopic texture of the fracture surfaces appeared to be fine. This change in fracture appearance corresponds to the transition in the crack-growth-rate versus  $\Delta K$  curves, Figure 5, and suggests a change in the mechanism for fatigue-crack growth. (By alternating between "high" and "low"  $\Delta K$ 's to produce crack-growth-rates above and below the transition range of 2 to  $3 \times 10^{-6}$  inch per cycle ( $5.08$  to  $7.62 \times 10^{-5}$  mm/cycle), alternate fine and coarse regions of crack growth may be produced.)

Typical electron-micrographs of fatigue-crack surfaces in the fine and coarse regions are shown in Figure 15. In the "fine" region, ductile fatigue striations predominate, Figure 15(a); while in the "coarse" region a more irregular fracture surface with regions of "quasi-cleavage" are observed, Figure 15(b). Optical microscopy results (on sections parallel to the plate surface) show extensive crack branching in the



coarse region, and indicate progression of cracks from one second-phase particle and occasional fracturing of the second-phase particles, Figure 16. No detailed mechanism for crack growth can be determined or postulated. Although no striations were observed, a mechanism of crack growth associated with ductile fatigue striation formation could not be ruled out, since at these low rates of growth individual striations (with spacings less than 500 Å) would not be resolvable with the current plastic-carbon replication techniques.

SUMMARY

Constant-load-amplitude fatigue-crack growth experiments were carried out on a  $\frac{1}{4}$ -inch-thick (6.35 mm) mill-annealed Ti-6Al-4V alloy plate in dehumidified argon to study the effects of temperature and crack driving force, characterized by the crack-tip stress intensity factor  $\Delta K$ , on crack growth. A range of temperature from room temperature to about 290C (or, 563K/550F) were investigated. The results indicate that the rate of fatigue-crack growth, for  $\Delta K$  from 10 ksi  $\sqrt{\text{in.}}$  to 30 ksi  $\sqrt{\text{in.}}$  (11 to 33 MN-m<sup>-3/2</sup>), corresponding to growth rates from  $2 \times 10^{-7}$  to  $4 \times 10^{-4}$  inch per cycle ( $5.08 \times 10^{-6}$  to  $1.16 \times 10^{-2}$  mm/cycle) were essentially unaffected by temperature in this range. The dependence of the rate of fatigue-crack growth on  $\Delta K$  appears to be separable into two regions, with a transition occurring in the range of  $2$  to  $3 \times 10^{-6}$  inch per cycle ( $5.08$  to  $7.62 \times 10^{-5}$  mm/cycle). The transition correlates well with changes in both the macroscopic and microscopic appearances of fracture surfaces, and suggests a change in the mechanism and the influence of microstructure on fatigue-crack growth.

Limited correlative experiments indicate that dehumidified oxygen and hydrogen had no effect on the rate of fatigue-crack growth in this alloy, while distilled water increased the rate of crack growth by 30 to 50 percent for  $\Delta K$  from 15 to 30 ksi  $\sqrt{\text{in.}}$  (17.5 to 33 MN-m<sup>-3/2</sup>). Crack growth in vacuum, at less than  $5 \times 10^{-6}$  torr, was about one-half that observed in the dehumidified gaseous environments for the same range of  $\Delta K$ . Mass spectrometric analysis and other experiments suggest that residual moisture, well below 30 ppm, can have a deleterious effect on this alloy. The observed temperature independence may still be caused by the compensating influences of moisture and deformation on fatigue-crack growth, and should be investigated further.

Companion fractographic studies showed that the mechanisms for fatigue-crack growth in the various environments, including vacuum, are essentially the same. The observation of ductile striations on specimens tested in vacuum is not in agreement with previous investigations on aluminum alloys. This discrepancy is believed to be caused principally by problems of contrast and resolution in the replication electron-microfractography technique.

Although the apparent temperature independence provides a degree of simplification in estimating the service performance of structure, the interactions between load and temperature under conditions of changing loads and temperatures encountered during service may be quite significant and should be carefully explored.

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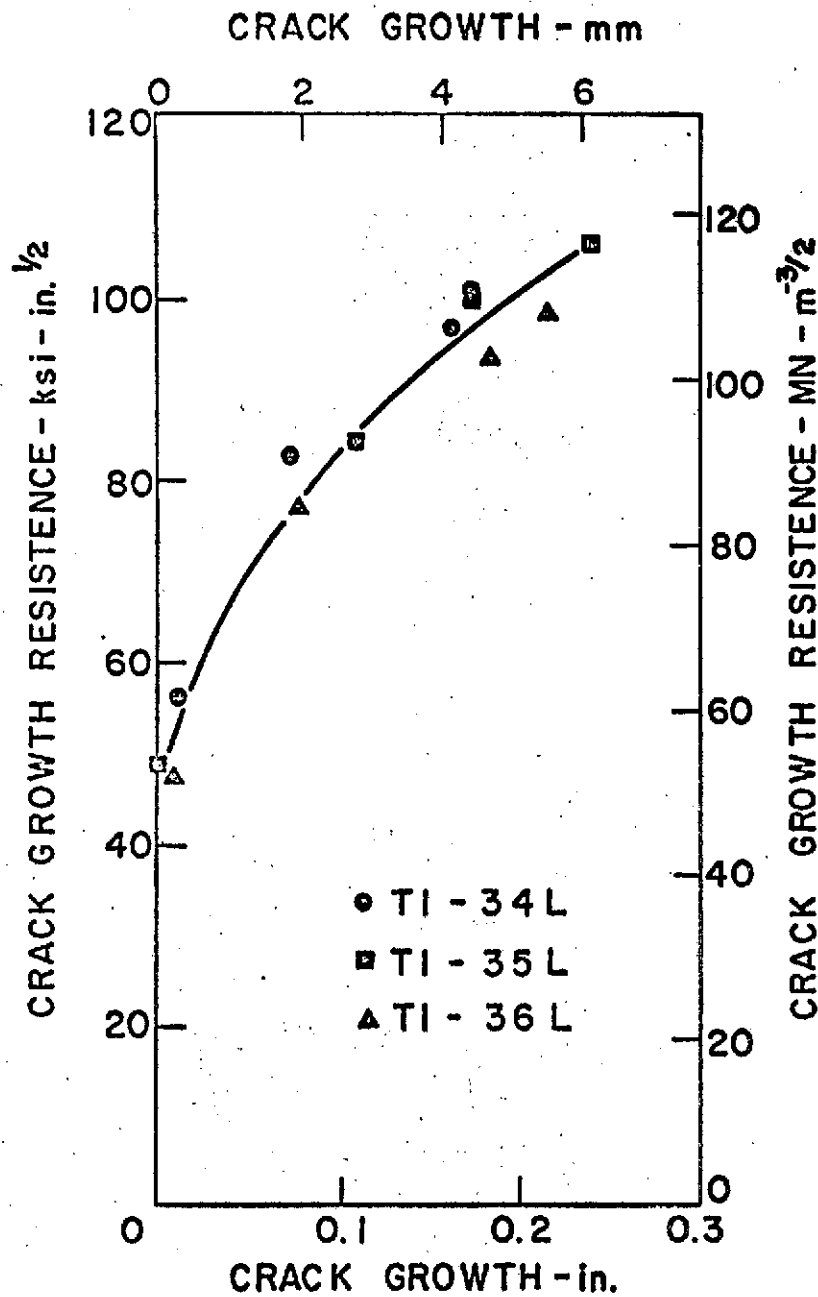
TABLE 1  
**CHEMICAL COMPOSITION AND  
 TENSILE PROPERTIES OF MATERIAL INVESTIGATED**

Nominal Chemical Composition (weight percent)

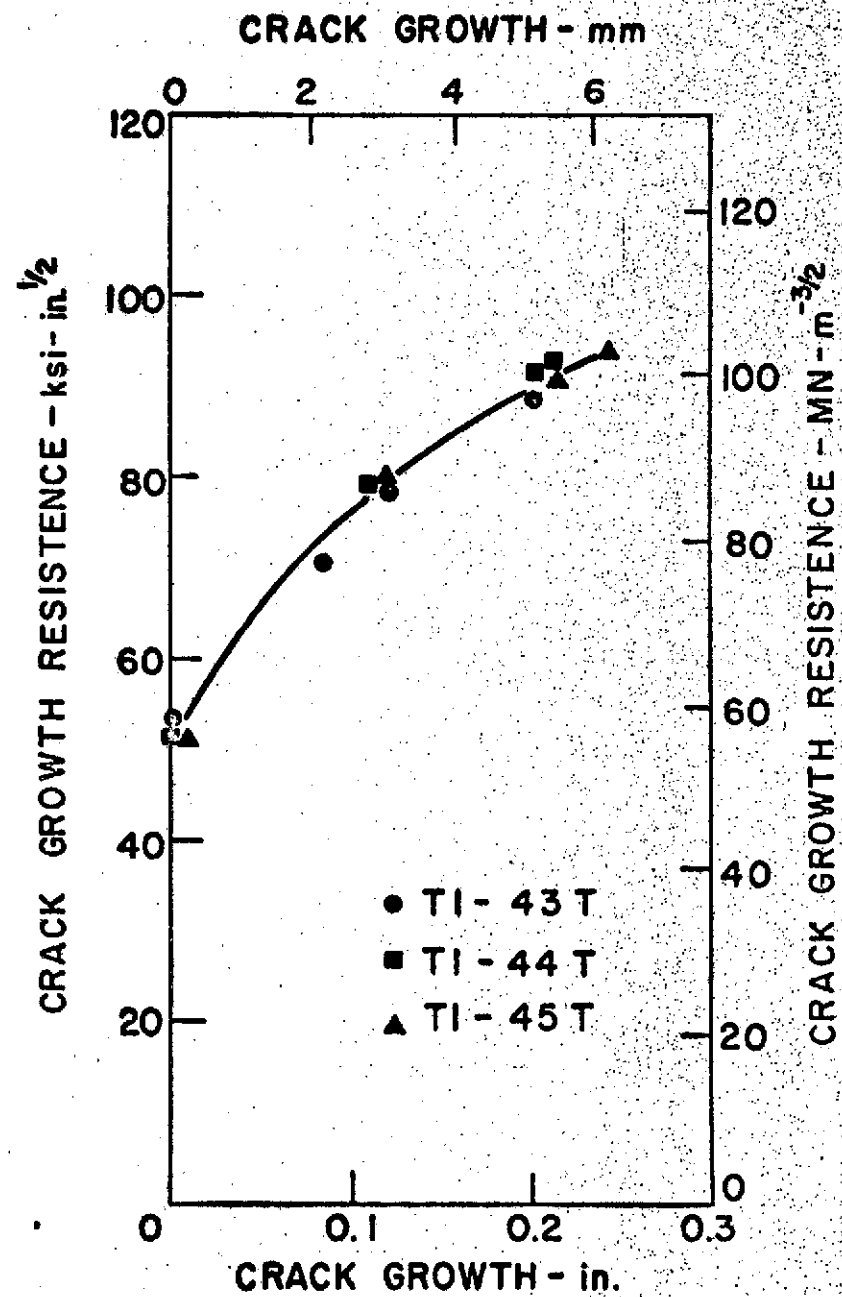
<u>C</u>	<u>Al</u>	<u>V</u>	<u>Fe</u>	<u>N</u>	<u>H</u>	<u>O</u>	<u>Ti</u>
0.10 max.	5.50 to 6.75	3.50 to 4.50	0.30 max.	0.05 max.	0.015 max.	0.20 max.	Balance

Tensile Properties

<u>Specimen No. and Direction</u>	<u>0.2% Offset Yield Strength ksi (MN/m<sup>2</sup>)</u>	<u>Tensile Strength ksi (MN/m<sup>2</sup>)</u>	<u>Elongation in 2 in. percent</u>
<b>Longitudinal</b>			
T1-37L	136.4 (941)	144.3 (995)	14.0
T1-38L	140.8 (971)	147.1 (1015)	14.0
T1-39L	<u>140.2 (967)</u>	<u>146.1 (1008)</u>	<u>14.0</u>
(Average)	139.1 (959)	145.8 (1006)	14.0
<b>Transverse</b>			
T1-40T	139.9 (965)	144.8 (999)	14.0
T1-41T	140.9 (972)	144.8 (999)	14.5
T1-42T	<u>143.0 (986)</u>	<u>146.1 (1008)</u>	<u>14.0</u>
(Average)	141.3 (974)	145.2 (1002)	14.2



(a) Longitudinal (LT or RW)



(b) Transverse (TL or WR)

Figure 1: Crack growth resistance curves for 1/4-inch thick mill-annealed Ti-6Al-4V alloy plate



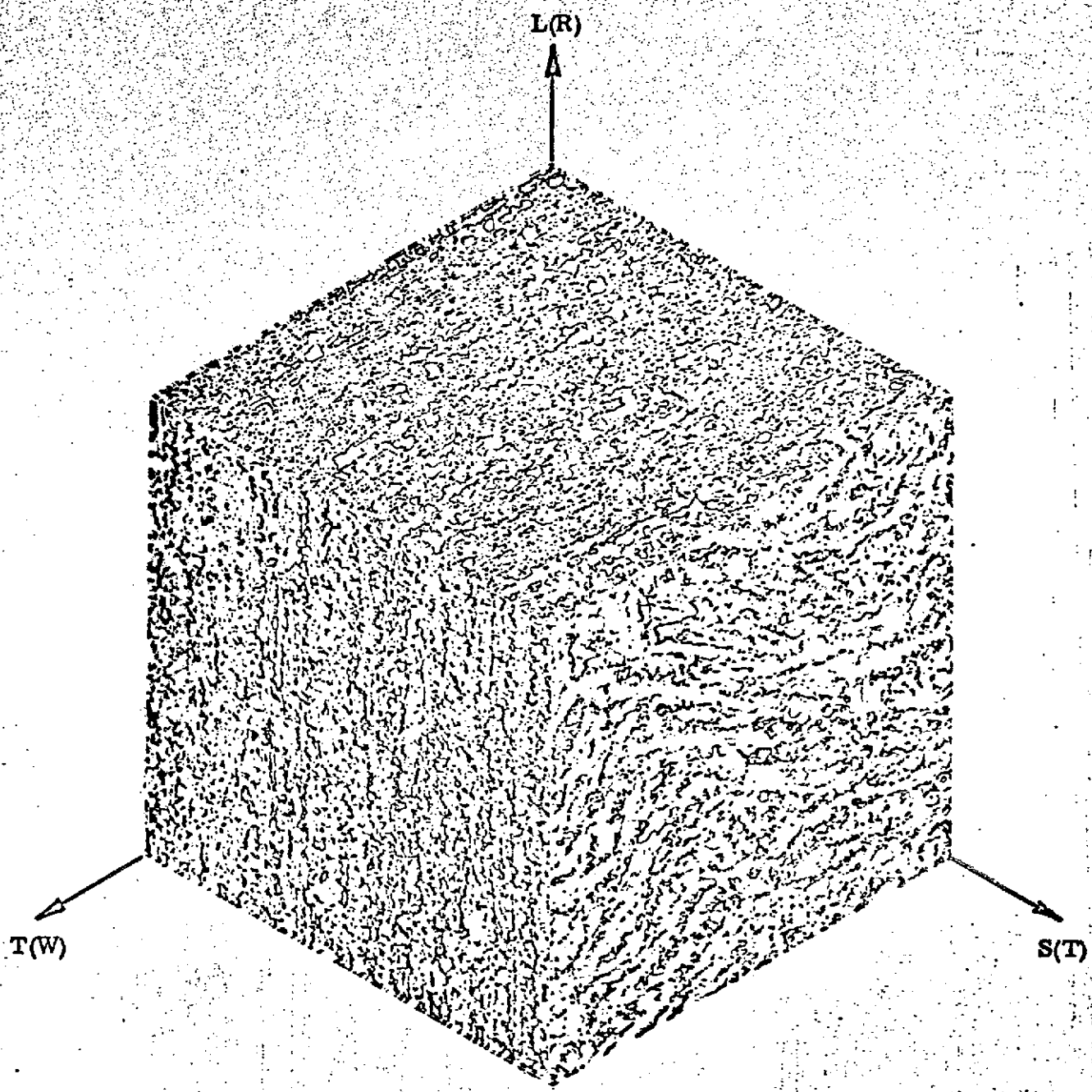


Figure 2: Representative microstructure of alloy investigated (200X)

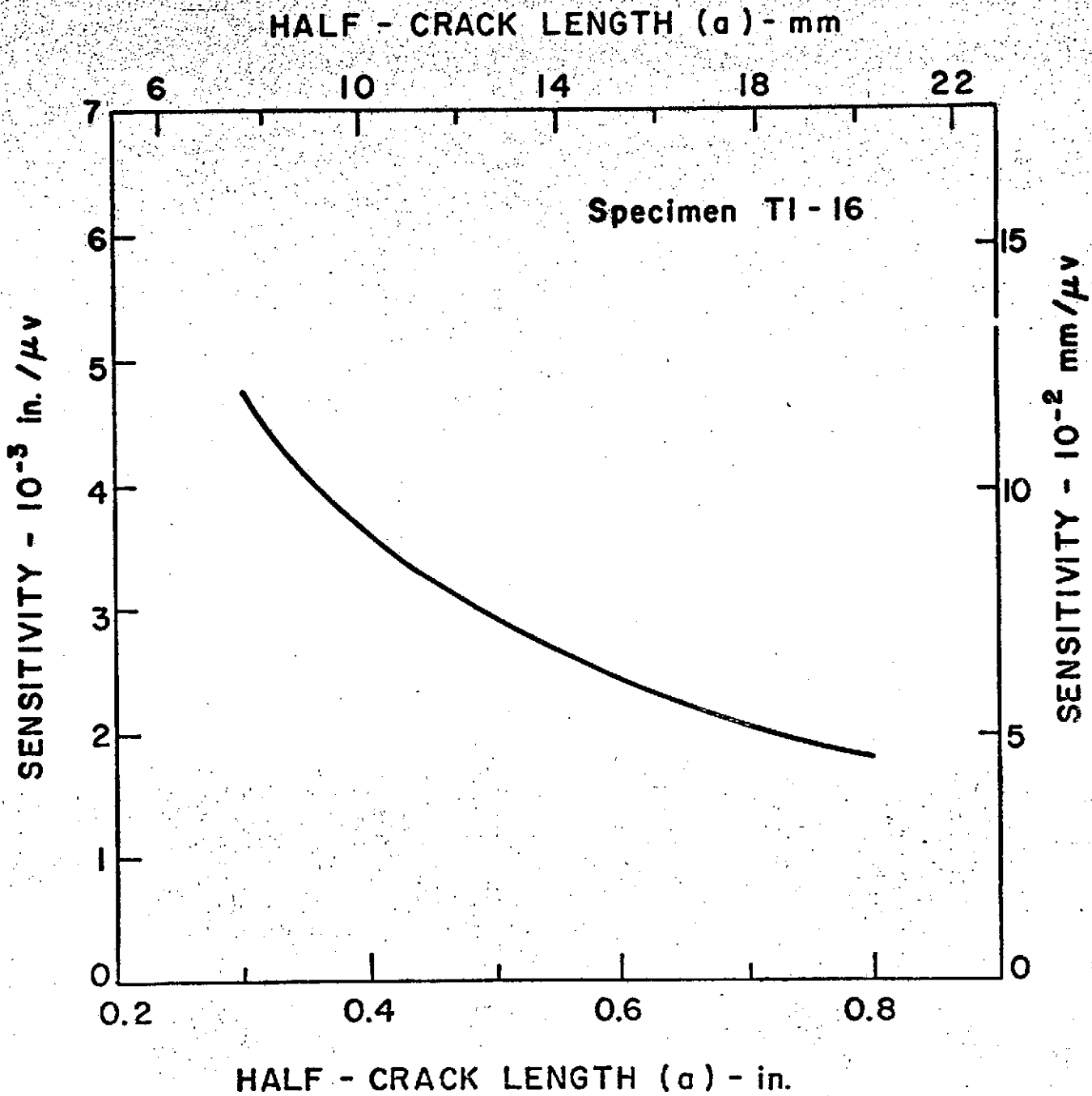
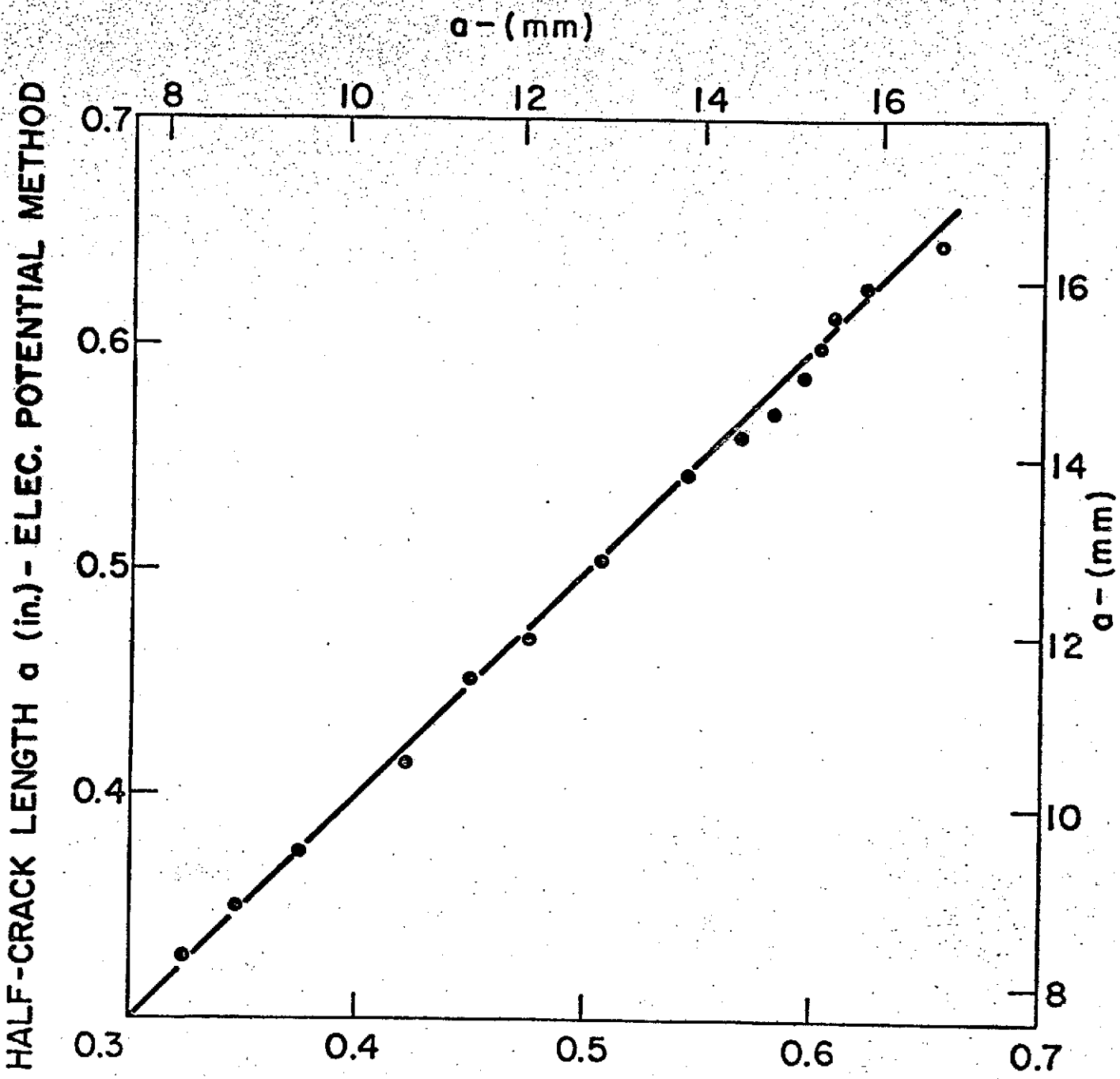


Figure 3: Variation of measurement sensitivity with crack length for a typical test specimen



HALF - CRACK LENGTH  $a$  (in.) - VISUAL METHOD  
(with 0.012 in. added to correct for crack front curvature)

Figure 4: Correlation between crack length measurements from visual and electrical-potential methods

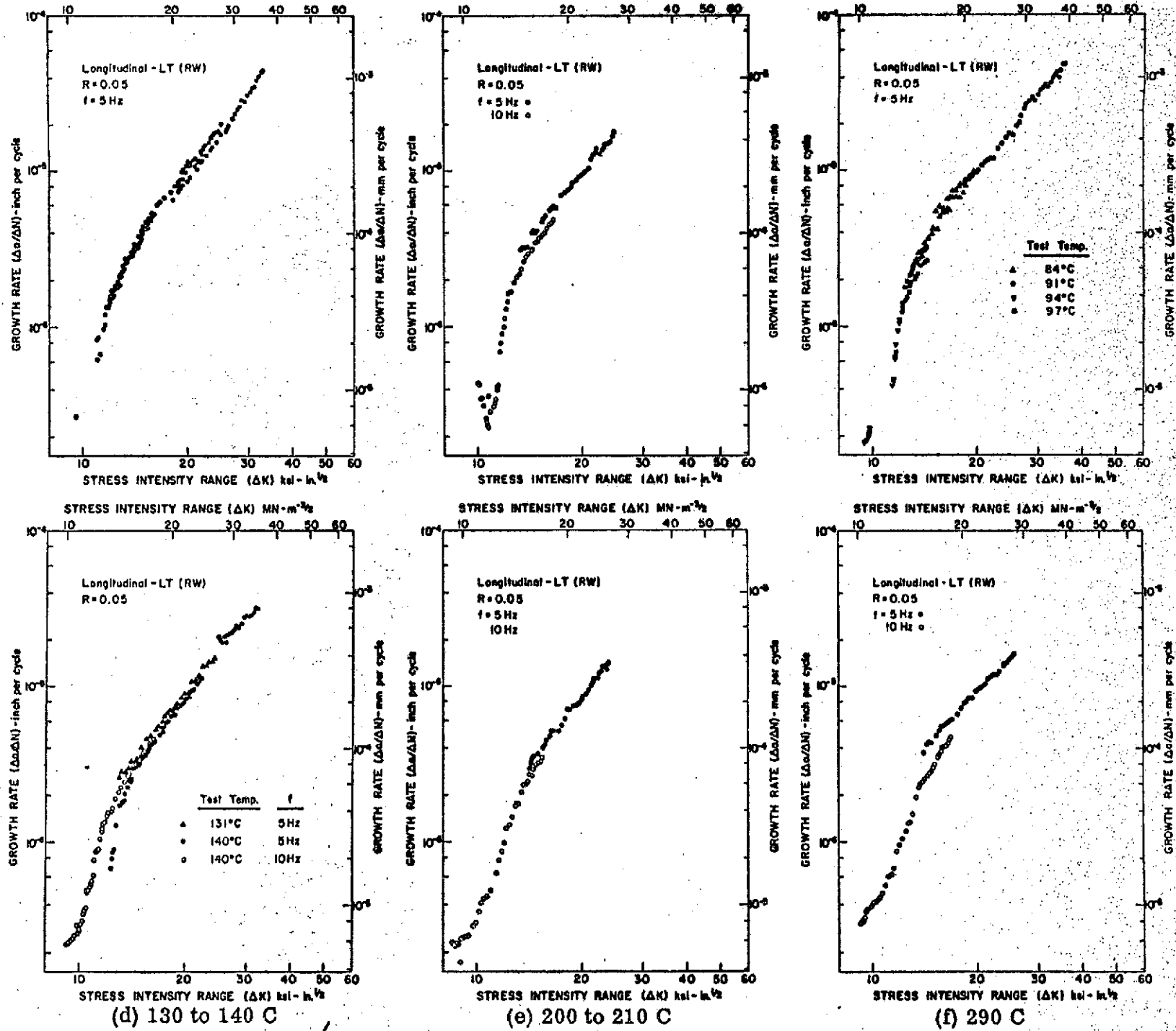


Figure 5: Rate of fatigue-crack growth in dehumidified argon at various temperatures

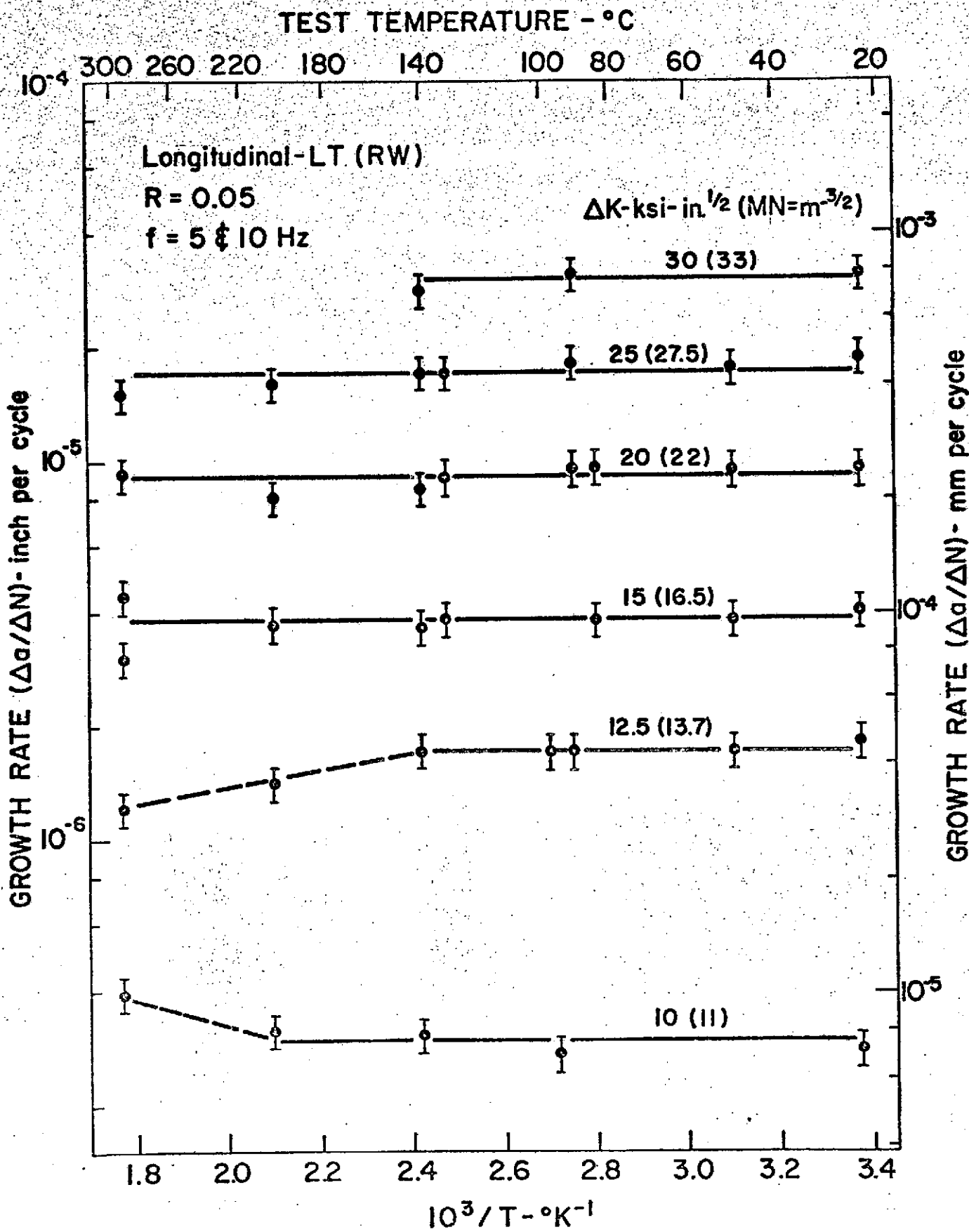


Figure 6: Influence of temperature on fatigue-crack growth in dehumidified argon at various K levels

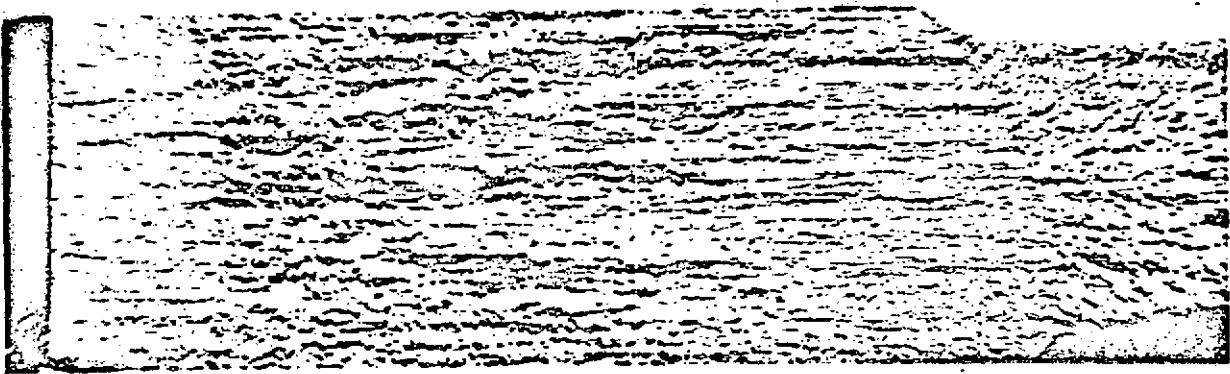


Figure 7: Optical macro-fractograph showing transition from "coarse" to "fine" mode of fracture. (Specimen tested at 300 to 6000 lbs. at 10 Hz. after fatigue precrack. Regions shown are, from left to right, fatigue precrack, coarse region, fine region, and tensile failure.)

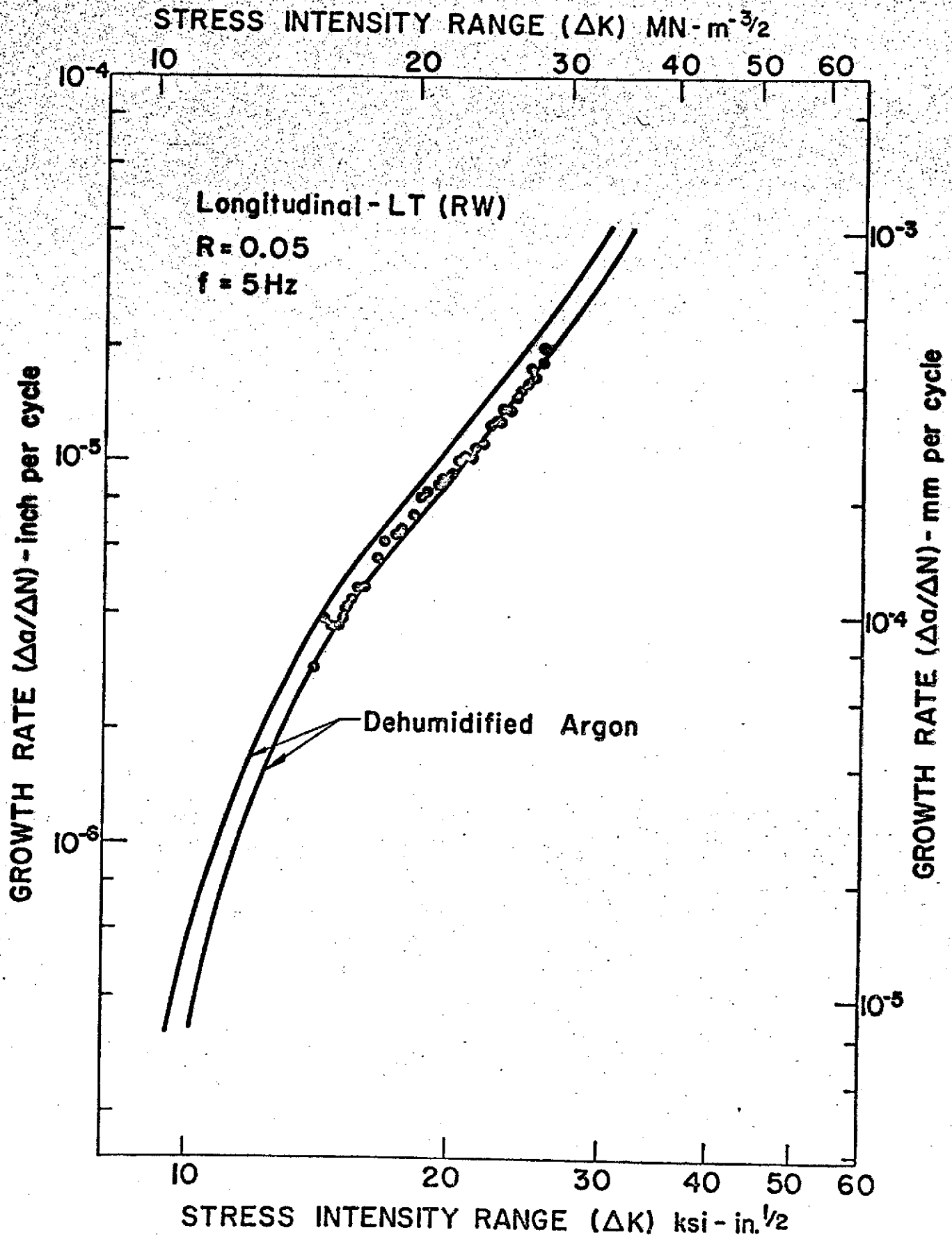


Figure 8a: Rate of fatigue-crack growth in dehumidified oxygen at room temperature

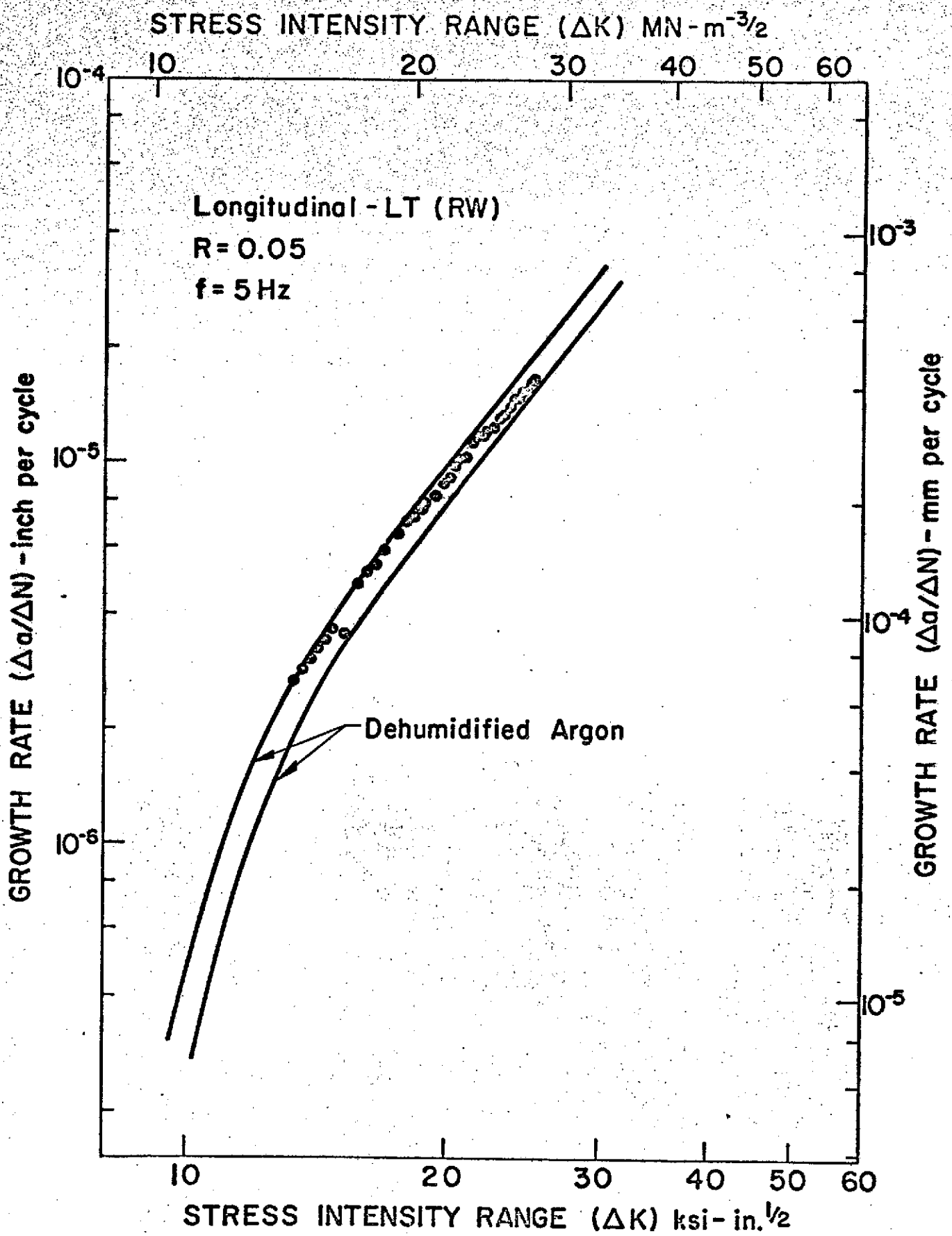


Figure 8b: Rate of fatigue-crack growth in dehumidified oxygen at 130 C



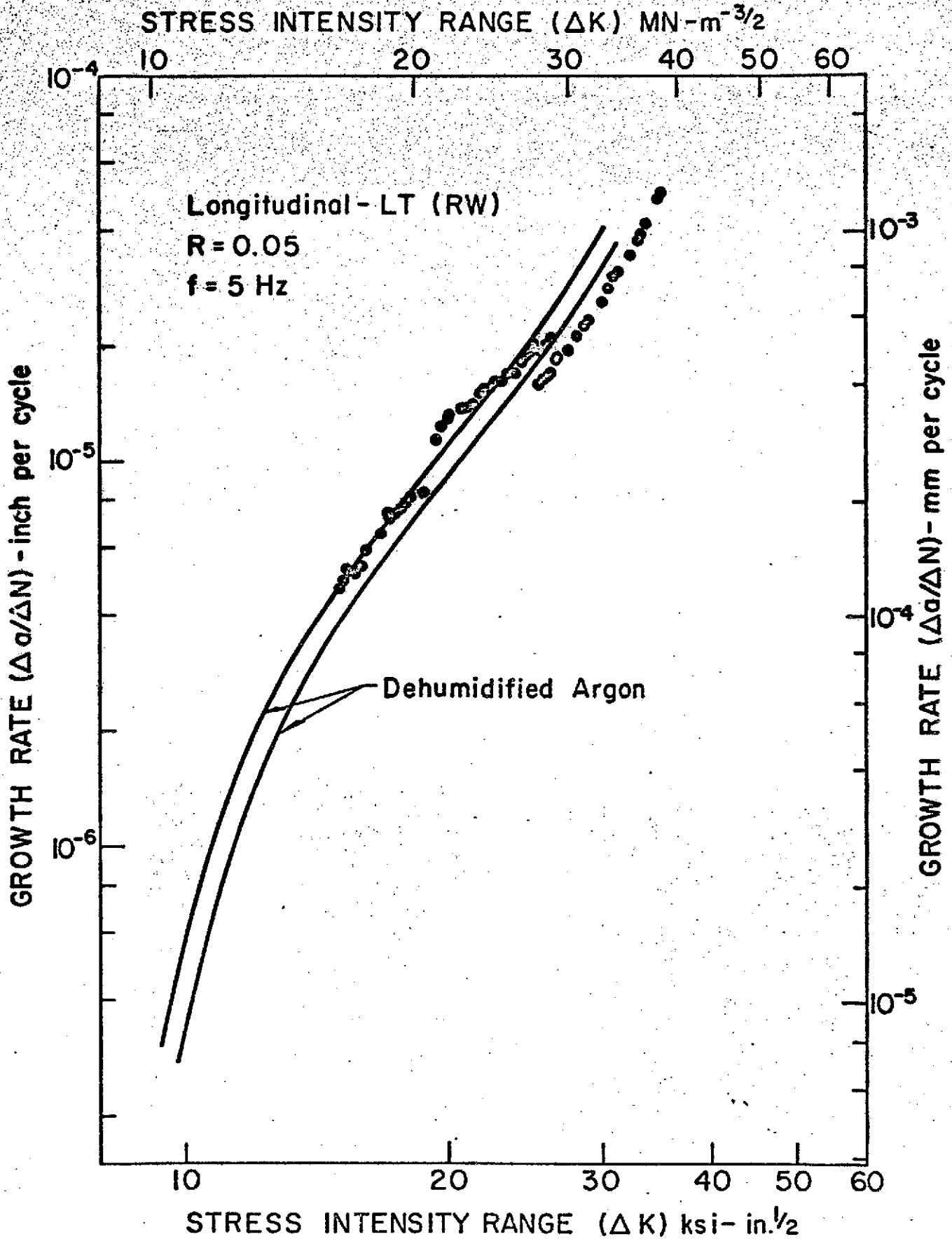


Figure 9: Rate of fatigue-crack growth in dehumidified hydrogen at room temperature

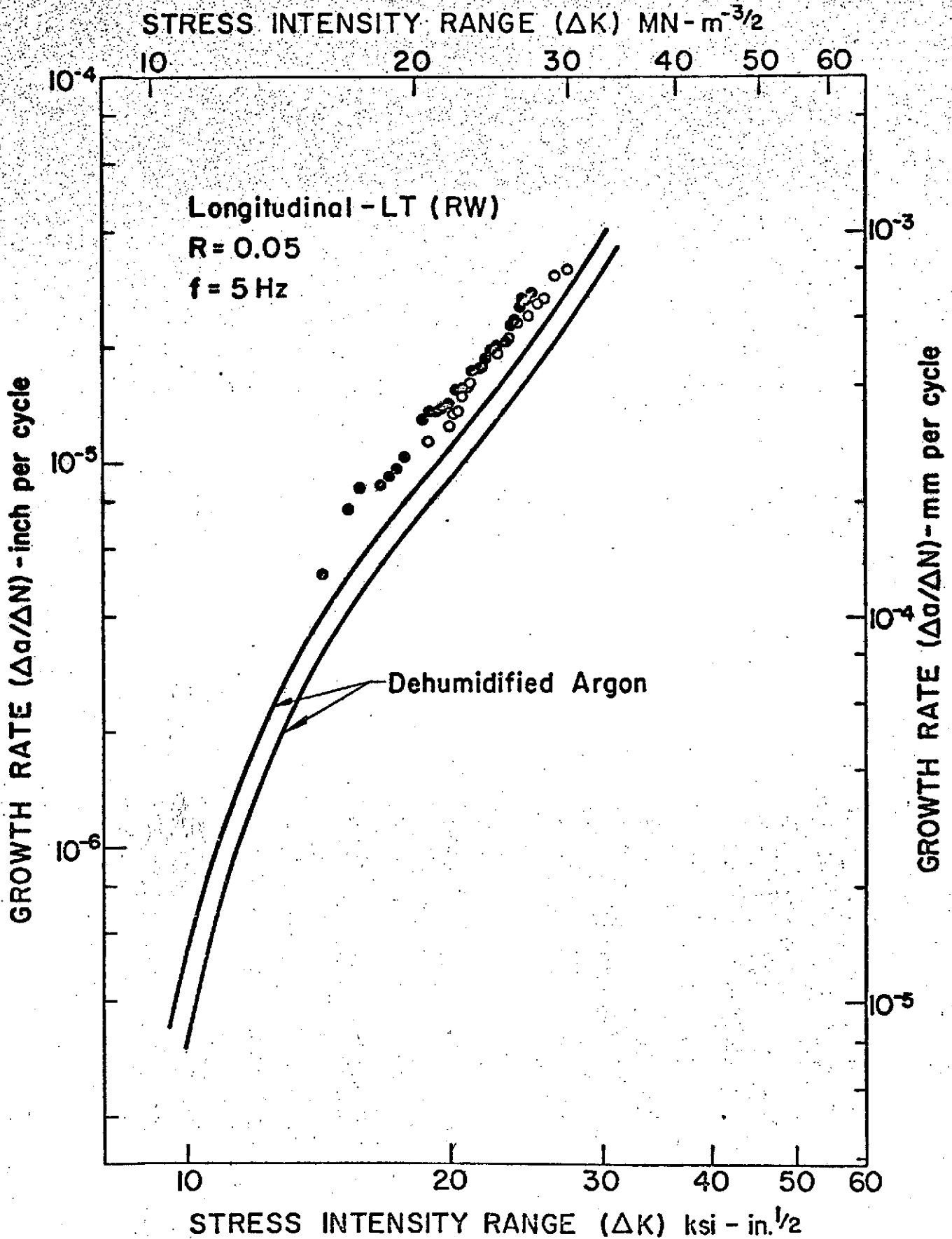


Figure 10a: Rate of fatigue-crack growth in distilled water at room temperature

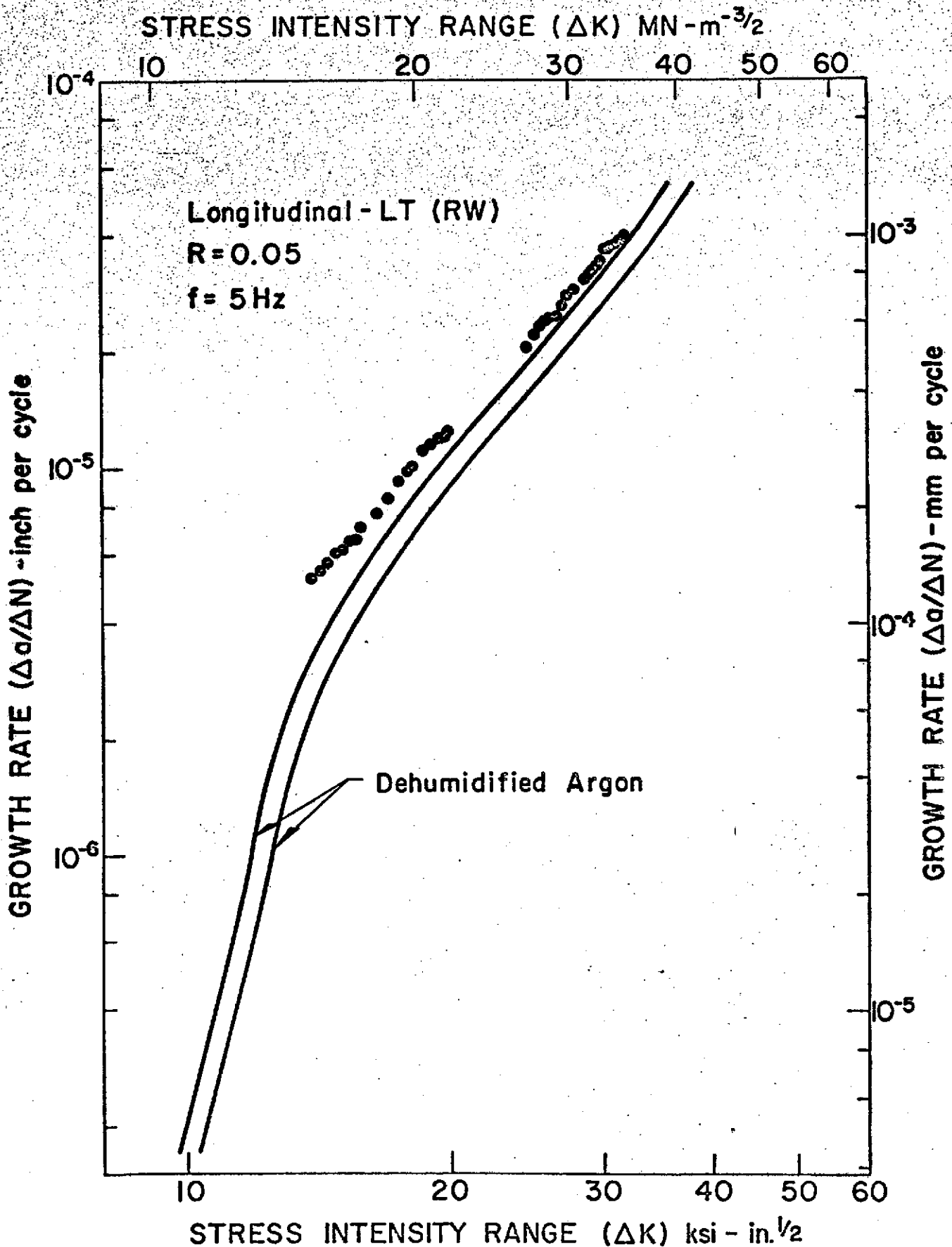


Figure 10b: Rate of fatigue-crack growth in distilled water at 85 C

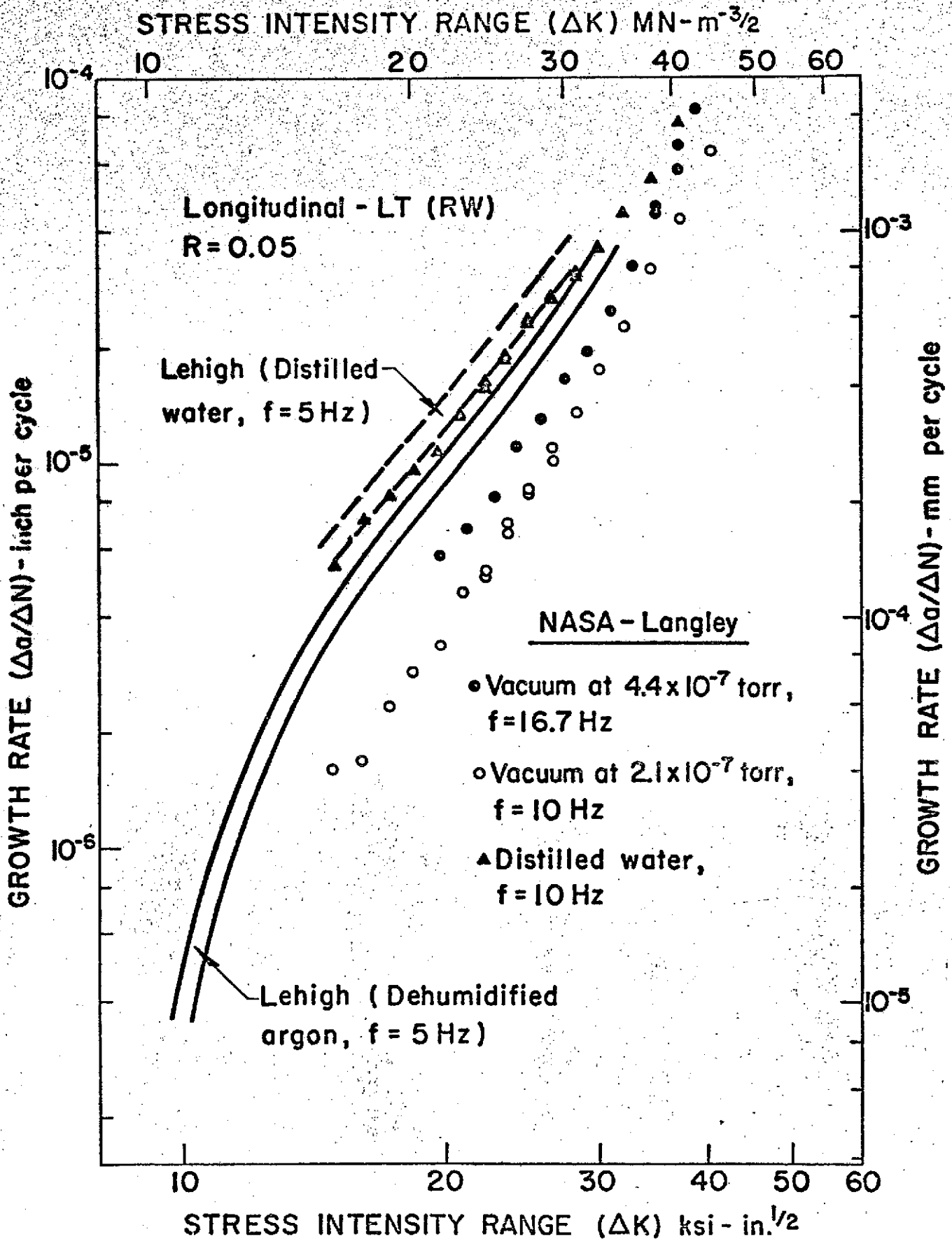


Figure 11: Comparison of NASA and Lehigh test results in vacuum, dehumidified argon and distilled water at room temperature

STRESS INTENSITY RANGE ( $\Delta K$ ) MN-m<sup>-3/2</sup>

10 20 30 40 50 60

Longitudinal - LT (RW)  
R = 0.05

GROWTH RATE ( $\Delta a/\Delta N$ ) - inch per cycle

GROWTH RATE ( $\Delta a/\Delta N$ ) - mm per cycle

NASA - Langley

● Room Temp. at  $4.4 \times 10^{-7}$  torr,  
f = 16.7 Hz

○ Room Temp. at  $2.1 \times 10^{-7}$  torr,  
f = 10 Hz

△ -61°C at  $2.9 \times 10^{-9}$  torr,  
f = 15 Hz

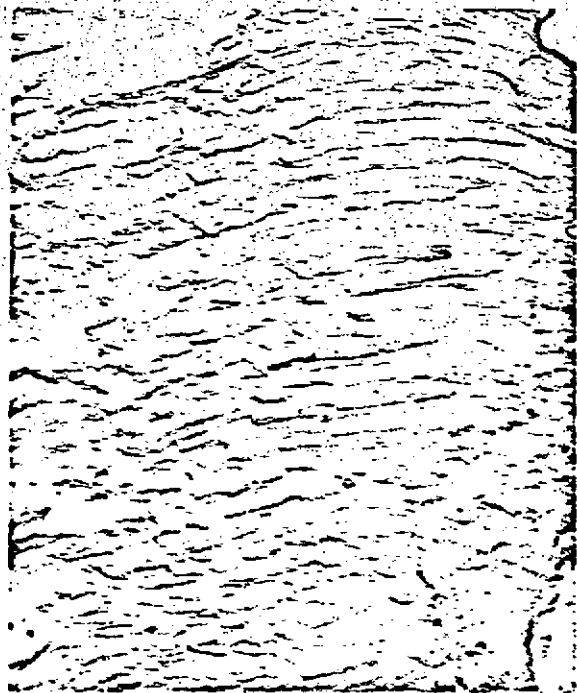
10 20 30 40 50 60

STRESS INTENSITY RANGE ( $\Delta K$ ) ksi - in.<sup>1/2</sup>

Figure 12: Rate of fatigue-crack growth in vacuum at room temperature and -61 C



(a) Argon at RT T1-24



(b) Argon at 130°C T1-18



(c) Vacuum ( $4.4 \times 10^{-7}$  torr) T1-25 at RT

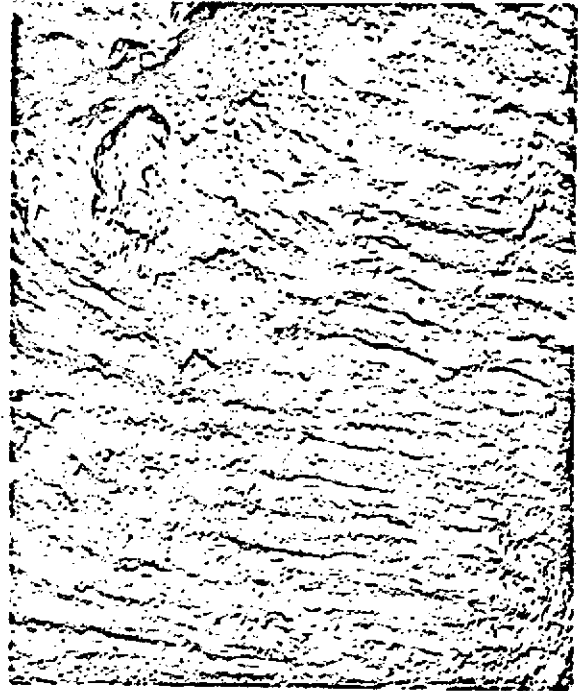


(d) Vacuum ( $2.9 \times 10^{-9}$  torr) T1-26 at -61°C

Figure 13: Typical electron-micrographs of fatigue fracture surfaces of specimens tested in the various environments and temperatures (5700X)

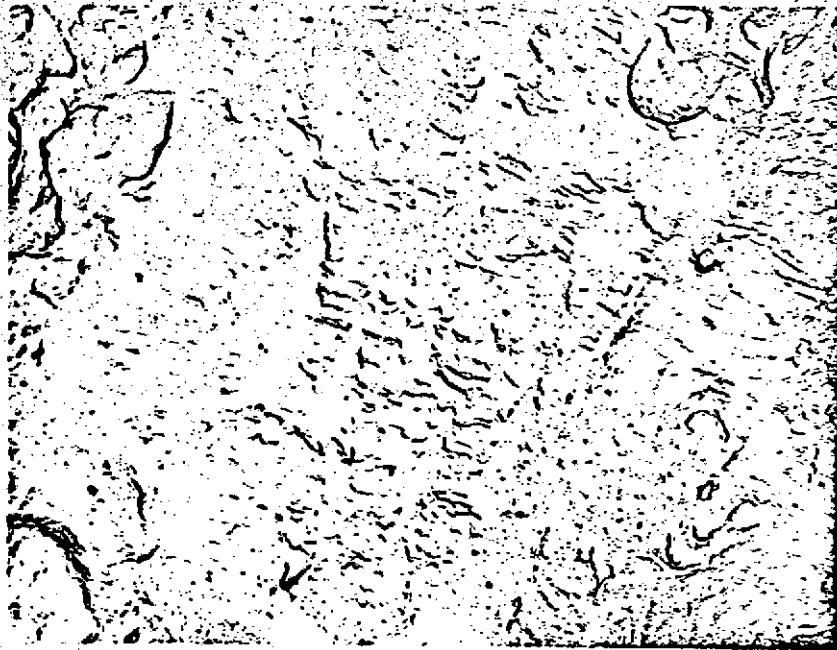


(e) Distilled water at RT T1-23

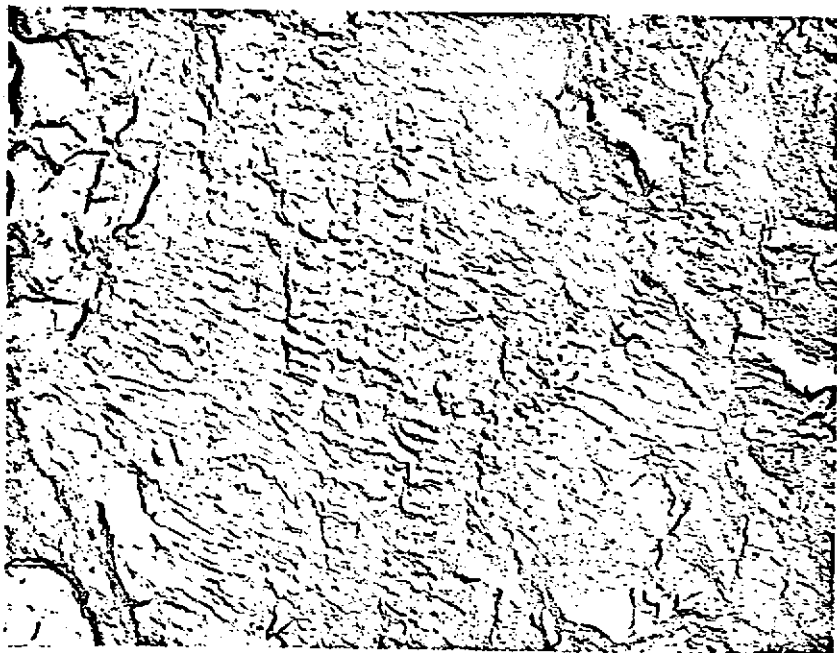


(f) Hydrogen at RT T1-17

Figure 13 (continued): Typical electron-micrographs of fatigue fracture surfaces of specimens tested in the various environments and temperatures (5700X)



(a)  $-15^\circ$  tilt

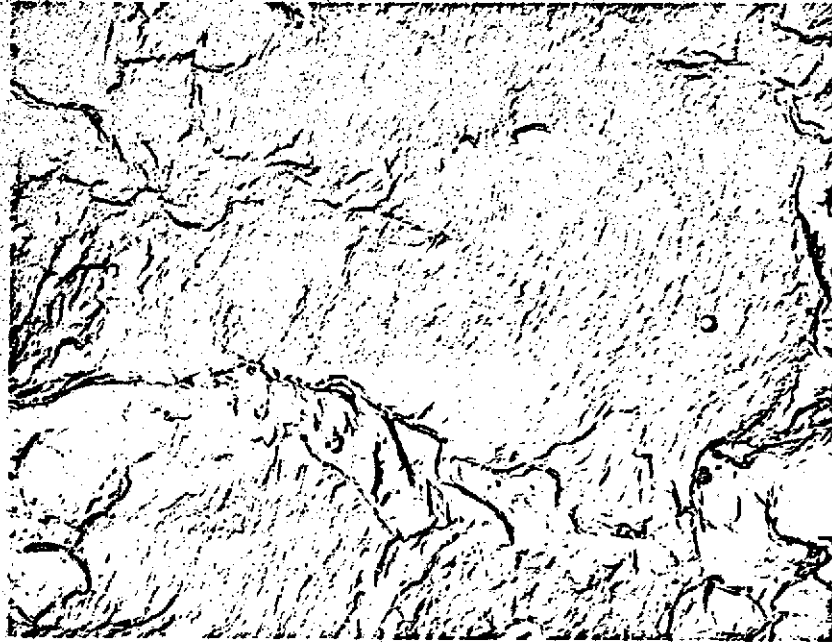


(b)  $+30^\circ$  tilt

Specimen T1-25

Figure 14: Effect of replica tilt angle on fractographic appearance ( $\sim 5300X$ )





(a) "Fine" region

7100X

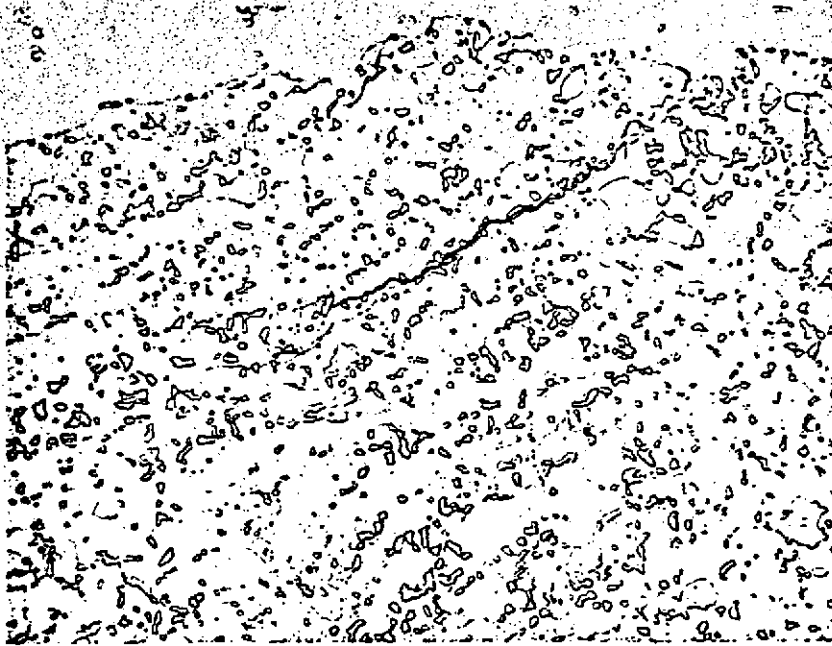


(b) "Coarse" region

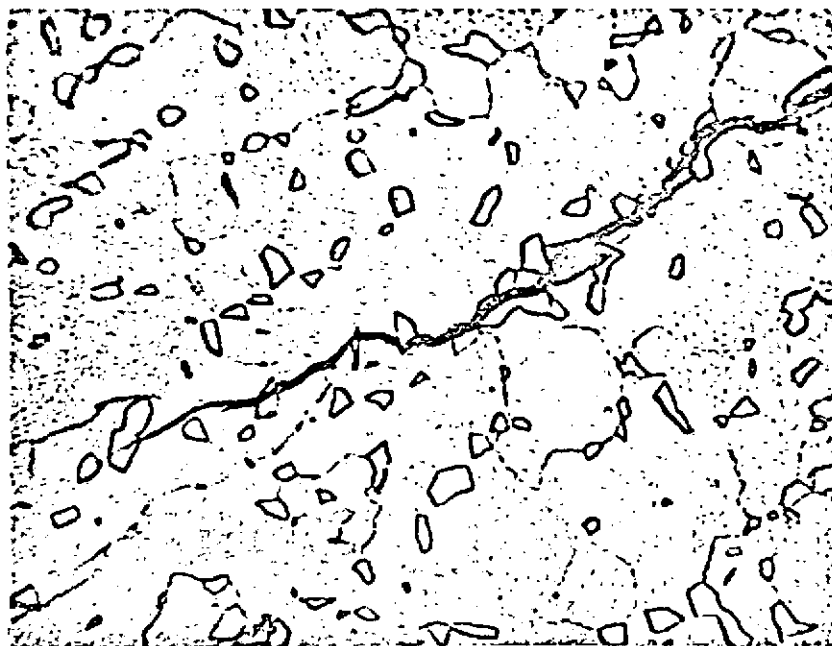
3200X

Specimen T1-30

Figure 15: Typical electron-micrographs of fatigue fracture surfaces in the "coarse" and "fine" regions



(a) 530X



(b) 1200X

Specimen T1-11

Figure 16: Optical micrographs of branching cracks in the "coarse" region