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## Abstract

A simple wave function for the normal state of the hydrogen molecule, taking into account both the atomic and ionic configurations, was set up and treated by a variational method. The dissociation energy was found to be 4.00 v.e. as compared to the experimental value of 4.68 v.e. and Rosen's value of 4.02 v.e. obtained by use of a function involving complicated integrals. It was found that the atomic function occurs with a factor 3.9 times that of the ionic function.

A similar function with different screening constants for the atomic and ionic parts was also tried. It was found that the best results are obtained when these screening constants are equal.

The addition of Rosen's term to the atomic-ionic function resulted in a value of 4.10 v.e. for the dissociation energy.

Attempts to obtain some of the properties of the normal hydrogen molecule by wave-mechanical methods date to the early days of wave mechanics. Heitler and London applied a first-order perturbation method, and Sugiura 2, by evaluating an integral whose value Heitler and London had only estimated, obtained results qualitatively comparable with known experimental data. Eisenschitz and London 3 applied a second-order perturbation treatment and obtained results in poorer agreement with experimental values than the results of previous calculations. For example, Heitler-London-Sugiura's value for the dissociation energy is 3.2 v.e. and the experimental value corrected for the zero-point energy is 4.68 v.e., while Eisenschitz and London obtained 9.5 v.e.. Thus it seems that the perturbation method is not very satisfactory for the treatment of the hydrogen molecule. The variational method, by approaching the value of energy from one side, is safe from the possibility of overshooting the mark. Wang4, using a variational method involving the introduction of a shielding constant as a parameter, obtained 3.7 v.e. for the dissociation energy. Rosen 5, by using the three-parameter function

where  $\mathscr{V}$  is the hydrogenic wave function for the lowest state with a shielding constant Z,  $\mathscr{V}$  is a function symmetrical about the axis but not about a plane through the nucleus perpendicular to it, and  $\mathscr{C}$  is a parameter, has obtained 4.02 v.e. for the dissociation energy. The improvement on the previous value is considerable, but the calculations are rather laborious.

All these calculations were based on the assumption that each of the nuclei always has one electron attached to it, these electrons sometimes interchanging their positions, which leads to the interchange energy. It was suggested by Hund and Mulliken that a truer picture would be given by a wave function \( \frac{\psi\_1 + \psi\_1}{\psi\_2 + \psi\_2} \) which takes account not only of the atomic configuration but also of the ionic configuration, when both electrons are on the same nucleus, the other being completely stripped of electrons. However, a function of the type suggested by Hund and Mulliken would give the hydrogen molecule in the normal state as much ionic character as atomic. There seems to be no reason to assume this, and a logical wave function to take care of the atomic-ionic character of the hydrogen molecule appears to be

C(4, 92+9,42)+(4,42+9,92)

where c is a parameter,  $\psi = Ne^{-Zt_{BI}}$  and  $\psi_i = Ne^{-Zt_{BI}}$ 

It has been shown 7 that the integral

$$E = \frac{\int \varphi^* H \varphi dr}{\int \varphi^* \varphi dr}$$

where H is the Hamiltonian operator and  $\varphi$  is a function which satisfies certain boundary conditions but is otherwise arbitrary, containing, say, some variable parameters, has the property that the lowest value W obtained from varying the numerical parameters is the best approximation to the value of E, and that E - W is always positive or zero. Hence the variational integral presents, as already mentioned, a satisfactory means for evaluating the energy of the normal state of the hydrogen molecule.

The first test for the wave function  $Y = C(Y_1Y_2 + Y_1Y_2) + (Y_1Y_2 + Y_1Y_2)$  would be to consider it a two-parameter function. The results obtained by varying c would then be comparable with the Heitler-London-Sugiura results. It is, however, more convenient to treat  $Y = C(Y_1Y_2 + Y_1Y_2) + (Y_1Y_2 + Y_1Y_2)$  as a three-parameter function, and then, at a certain point in the algebra, to reduce it to a two-parameter function by letting Z = 1.

It is useful to set up the following scheme, due to Slater:

and rewrite our wave function as

$$Y = C(Y_2 - Y_2) + Y_{22} + Y_{22} \tag{1}$$

where  $\frac{1}{2}$  is the atomic Heitler-London term and  $\frac{1}{2}$  is the ionic term. The variational integral

$$W = \frac{\int \psi^* H \psi dr}{\int \psi^* \psi dr} \tag{2}$$

then takes the form

$$W = \frac{W_N}{d} = \frac{\int H[C(4x - 4x) + 4x + 4x]^2 dr}{\int [C(4x - 4x) + 4x + 4x]^2 dr}$$
 (2a)

The wave equation for a hydrogen-like atom, in a system of units where unit of length  $a_o = .52845 \text{ Å}$  and unit of energy  $\frac{\ell^2}{a_o} = 27.06 \text{ volt-electrons}$ , is

$$\nabla^2 \mathcal{Y} + 2(W - V)\mathcal{Y} = 0 \tag{3}$$

or, in a rewritten form,

$$W\psi = -\frac{1}{2}\nabla^2\psi + V\psi \tag{3a}$$

where  $W = -\frac{Z^2}{2}$ . The general wave equation may be written 8,5

$$WY = HY \tag{4}$$
ith  $H = -\frac{1}{2}T^2 + V$  and  $V = \frac{5}{2} = \frac{7}{2}Z_1 = \frac{7}{2}$ 

with 
$$H = -\frac{1}{2}\nabla^2 + V$$
 and  $V = \mathcal{E}\mathcal{E}\frac{Z_iZ_j}{Z_{ij}}$  (5)

the Hamiltonian function for the case of a hydrogen molecule being

$$H = -\frac{1}{2} (P_1^2 + P_2^2) - \frac{1}{z_{A1}} - \frac{1}{z_{A2}} - \frac{1}{z_{B1}} - \frac{1}{z_{B2}} + \frac{1}{z_{I2}} + \frac{1}{z_{AB}}$$
 (6)

In subsequent calculations the letter I with a subscript will be used to indicate the different integrals occurring. The integral SYPAT will be denoted by S.

Leaving out for the present the term ZAR the Hamiltonian we can now set up the following expressions:

$$\int Y_{1}H_{1}Y_{1}d\tau_{1} = \int Y_{1}\left(-\frac{p_{1}^{2}}{2} - \frac{1}{z_{AI}} - \frac{1}{z_{BI}}\right)Y_{1}d\tau_{1} = 
= \int Y_{1}\left(-\frac{Z^{2}}{2} + \frac{Z-I}{z_{AI}} + \frac{1}{z_{BI}}\right)Y_{1}d\tau_{1} = -\left[\frac{Z^{2}}{2} + (I-Z)Z + I_{1}\right] = A$$

$$\int Y_{1}H_{1}Y_{1}d\tau_{1} = -\left[\frac{Z^{2}}{2}S + (Z-Z)I_{2}\right] = B$$
(7a)

Let us denote integrals of the type SyHy dr mutations of the signs of the spin that are necessary to obtain the same spin for functions pertaining to the same electron. Then, in terms of A and B, we obtain

Making use of the relations (8), we find

$$\frac{1}{2}W_{N} = C^{2}H_{II} + H_{IIII} - C^{2}H_{III} + H_{IIII} - 4cH_{IIII} = 
= C^{2}(2A + I_{H}) + 2A + I_{3} + C^{2}(2Bs + I_{c}) + 2Bs + I_{c} + 4c(As + B + I_{s})$$
(9)

and putting I = ZF, where F is a function of  $\rho = ZZ_{AB}$  only, this becomes

$$\frac{1}{2}W_{N} = Z^{2}[[c^{2}+1](1+2sF_{2}-s^{2})+4cF_{2}]+Z[[c^{2}+1](F_{6}-2-2F_{7}-4sF_{2})+$$

$$+F_{3}+c^{2}F_{4}-4c(s+F_{7}s+2F_{2}-F_{3})]=\lambda Z^{2}+\mu Z \tag{9a}$$

where

$$\lambda = (c^2 + 1)(1 + 2SF_2 - S^2) + 4cF_2 \tag{10}$$

$$\mu = (c^2 + 1)/F_6 - 2 - 2F_7 - 45F_2 + F_3 + c^2 F_7 - 4c(s + F_7 + 2F_2 - F_5)$$
 (10a)

The denominator in the expression for W has the value

$$d = \int \psi^* \psi d\tau = 2d_{II} + 2d_{III} - 2d_{III} + 2d_{III} + 8d_{III} =$$

$$= 2c^2 + 2 + 2c^2 s^2 + 2s^2 + 8cs = 2[(c^2 + 1)(1 + s^2) + 4cs] = 2V$$
(11)

As  $\rho = Z_{AB}^c$  , introducing the term  $\frac{1}{Z_{AB}}$  that we left out of the calculations, we have

$$W = \frac{\lambda Z_{+\mu}^{2} Z}{y/z} + \frac{Z}{\rho} = \frac{\lambda}{y} Z_{+}^{2} / \frac{\mu}{y} + \frac{1}{\rho} / Z$$
 (12)

Substituting Z=1 in (12) we obtain W as a function of parameters  $\mathcal C$  and  $\rho$ :

$$W = \frac{\lambda + \mu}{\sqrt{\rho}} + \frac{1}{\rho} = \frac{(c^2 + 1)(-1 - s^2 - 2F_7 - 2F_2 s + F_2) + F_3 + c^2 F_7 - 4c(s + F_7 s + F_2 - F_5)}{(c^2 + 1)(1 + s^2) + 4cs} + \frac{1}{\rho}$$
(13)

This expression for W involves only integrals used by Heitler and London,  $F_6$  being the integral evaluated by Sugiura. Minimizing (13) with respect to C,  $\frac{dV}{dc} = 0$  yields a quadratic in c. Substituting the value of C obtained from this quadratic back in (13) and varying  $\rho$ , the lowest value of W was found to be W = -1.1187 with  $\rho = 1.67$  and C = 6.322. The Sugiura value for W is W = -1.1156, so that the inclusion of the ionic term gives an improvement in the dissociation energy of 0.0031 or 3%.

Going back to (12), we can improve the treatment by minimizing (12) with respect to Z, which is equivalent to Wang's treatment plus an ionic term. We get

$$\frac{dN}{dZ} = \frac{2\lambda}{J} Z + \frac{\mu}{J} + \frac{I}{P} \tag{14}$$

and

$$Z_{min.} = -\frac{1}{2\lambda} \left( \frac{\mu}{5} + \frac{1}{\rho} \right) \tag{15}$$

Substituting (15) into (12) we have

$$W_{Z=min.} = -\frac{1}{4\lambda} \left( \frac{\mu}{\nu} + \frac{1}{\rho} \right)^2 \tag{16}$$

Expression (16) is a function of C and  $\rho$  as was the case for (13). But while in the case of (13)  $\frac{dW}{dc} = 0$  gave a quadratic in C, in the case of (16)  $\frac{dW}{dc} = 0$  gives an equation of the fifth degree in C. However, to evaluate C by direct substitution of different values of C for a given  $\rho$  involves very little labor. The procedure employed was as follows: the best value of C was first obtained for  $\rho = 1.67$ ; then  $\rho$  was varied to get the best value for the obtained C; then C was varied again, and so on, until the variation in either C or  $\rho$  would result in a lower value of W. The lowest value of W obtained from (16) is W = -1.148 with  $\rho = 1.69$ , C = 3.9 and Z = 1.193.

This result compares very favorably with the result of Wang's treatment (improvement of 8% in dissociation energy) and is just very slightly lower than Rosen's result, which requires very elaborate calculations. Thus the results of this calculation show that the ratio

of the coefficients of the atomic and ionic parts of the wave function is about four and not one as was suggested by Hund and Mulliken.

So far it has been assumed that the effective nuclear charge Z is the same for both the atomic and ionic parts. The next step is to introduce a new parameter  $\mathcal{E} = \frac{Z'}{Z}$ , where Z' is the effective nuclear charge in the ionic term, The integrals involving Z only will be denoted as before by I, for the integrals involving  $Z' = \mathcal{E}Z$  an  $\mathcal{E}$  will be added in the subscript, and the integrals involving both Z and Z' will be denoted by the letter J.

Our wave function will be written now as

$$Y = C(Y_1Y_2 + Y_1Y_2) + (\xi_1\xi_2 + \eta_1\eta_2)$$

$$\xi_1 = N'e^{-Z'z_{AI}}$$
and  $\eta_1 = N'e^{-Z'z_{BI}}$ . (17)

are not equal.

Proceeding as before and leaving the term  $\frac{1}{\zeta_{AB}}$  out, we can write the following expressions:

$$\int Y_{1}H_{1}Y_{1}dT_{1} = \int Y_{1}H_{1}Y_{1}dT_{1} = A$$

$$\int Y_{1}H_{1}Y_{1}dT_{1} = \int Y_{1}H_{1}Y_{1}dT_{1} = B$$

$$\int \xi_{1}H_{1}\xi_{1}dT_{1} = \int \gamma_{1}H_{1}\gamma_{1}dT_{1} = A_{E}$$

$$\int \xi_{1}H_{1}\eta_{1}dT_{1} = \int \gamma_{1}H_{1}\xi_{1}dT_{1} = B_{E}$$

$$\int Y_{1}H_{1} \xi_{1} d\tau_{1} = \int Y_{1}H_{1} \eta_{1} d\tau_{1} = -\frac{\varepsilon^{2}Z^{2}}{2}t_{0} + \varepsilon ZJ_{1} - J_{1} - J_{1} = A'$$

$$\int Y_{1}H_{1} \eta_{1} d\tau_{1} = \int Y_{1}H_{1} \xi_{1} d\tau_{1} = -\frac{\varepsilon^{2}Z^{2}}{2}t_{1} + \varepsilon ZJ_{2} - J_{2} - J_{2} - J_{2} = B'$$

$$\int \xi_{1}H_{1}Y_{1} d\tau_{1} = \int \eta_{1}H_{1}Y_{1} d\tau_{1} = -\frac{Z^{2}}{2}t_{0} + ZJ_{1} - J_{1} - J_{1} = A''$$

$$\int \xi_{1}H_{1}Y_{1} d\tau_{1} = \int \eta_{1}H_{1}Y_{1} d\tau_{1} = -\frac{Z^{2}}{2}t_{1} + ZJ_{2} - J_{2} - J_{2} = B''$$
where
$$t_{0} = \int Y_{1}^{2}d\tau \qquad \text{and} \quad t = \int Y_{1}d\tau.$$

In terms of A, B, A, B, A, B, A', B', A", and B", we have

$$H_{II} = H_{II} = 2A + I_{Y}$$

$$H_{III} = H_{III} = -2BS - I_{E}$$

$$H_{IIII} = H_{IIII} = 2A_{E} + I_{3E}$$

$$H_{IIII} = H_{IIIII} = 2B_{E}S_{E} + I_{EE}$$

$$H_{IIII} = H_{IIII} = -H_{IIIII} = -H_{IIIII} = A't + B't_{0} + J_{5}$$

$$H_{IIII} = H_{IIII} = -H_{IIIII} = -H_{IIIII} = A''t + B''t_{0} + J_{5}$$

and 
$$\frac{1}{2}W_{N} = C^{2}(2A+I_{4})+C^{2}(2BS+I_{6})+2A_{2}+I_{3}+2B_{2}+2B_{5}+I_{6}+$$

$$+2C[(A'+A'')t+(B'+B'')t_{6}+2J_{5}] \qquad (20)$$

Introducing integrals F and G, independent of Z, such that I = ZF,  $I_{\epsilon} = \epsilon$  ZF and J = ZG, we get

$$\frac{1}{2}N_{N} = Z^{2}\left\{C^{2}(I+2F_{2}s-s^{2})+2cI(I+\epsilon)G_{1}t+(G_{2}+\epsilon G_{2}')t_{0}+\right. \\
\left.-(I+\epsilon^{2})t_{0}t_{1}^{2}+\epsilon^{2}(I+2S_{\epsilon}F_{2\epsilon}-S_{\epsilon}^{2})\right]+Z\left\{C^{2}(F_{2}+F_{2}-2+\right. \\
\left.-2F_{1}-4F_{2}s\right)+\epsilon(F_{3}+F_{\epsilon}-2-2F_{1\epsilon}-4F_{2\epsilon}S_{\epsilon})+\right. \\
\left.-4cIt(G_{1}+G_{1}')+t_{0}(G_{2}+G_{2}')-G_{5}\right\}=\lambda'Z^{2}+\mu'Z$$
(20a)

The denominator has the value

$$d = 2[c_{+}^{2}c_{5}^{2}+1+s_{\xi}^{2}+4ct_{6}t] = 2V'$$
 (21)

Hence, introducing the term  $\frac{1}{\zeta_{AB}}$  , left out of the calculations, we have

$$W = \frac{\lambda' Z_{+}^{2} \mu' Z}{\nu'} + \frac{Z}{\rho} = \frac{\lambda'}{\nu'} Z^{2} + \left(\frac{\mu'}{\nu'} + \frac{1}{\rho}\right) Z \tag{22}$$

and

$$W_{Z=min} = -\frac{\mathcal{Y}}{4\lambda'} \left( \frac{A'}{4\mathcal{Y}} + \frac{1}{P} \right)^2 \tag{23}$$

The G integrals can be evaluated in elliptic coordinates and they will be found in the appendix.

Expression (23) is a function of  $\rho$ ,  $\varepsilon$  and c. It reduces to (16) when  $\varepsilon=1$ . Starting with the best values of C and  $\rho$  for (16), C=3.9 and  $\rho=1.69$ , values of  $\varepsilon>/$  and  $\varepsilon</$  were tried. It was found that the value  $\varepsilon=1$  gives the lowest energy. This interesting fact shows that the screening constants for the atomic and ionic contributions to the normal state of hydrogen molecule are the same.

The ionic term and the Rosen term giving independently an improvement on Wang's result, the next step is to set up a function which would include both these terms. Making again use of Slater's scheme we can write

$$Y = C(Y + CY')_{\pi} - C(Y + CY')_{\pi} + Y_{\overline{m}} + Y_{\overline{m}}$$
(24)

where 4' is the Rosen term

$$\Psi' = \left(\frac{\alpha^{5}}{\pi}\right)^{1/2} e^{-2Z \epsilon_{AI}} \epsilon_{AI} \cos \theta_{AI} \tag{25}$$

the nuclear charge being taken as 2Z so that the integrals involving  $\psi$  would be integrable.

Following the same procedure as before and remembering that

$$\frac{1}{2}\nabla^2 \psi = \left(-\frac{Z^2}{2} + \frac{Z}{2}\right)\psi \tag{26}$$

and

$$\frac{1}{2}\nabla^{2}\psi' = \left(-\frac{Z^{2}}{Z} + \frac{2Z}{Z}\right)\psi' \tag{26a}$$

we obtain

$$\int Y_{1}H_{1}Y_{1}dT_{1} = A$$

$$\int Y_{1}H_{1}Y_{1}dT_{1} = B$$

$$\int Y_{1}H_{1}Y_{1}dT_{1} = \int Y_{1}H_{1}Y_{1}dT_{1} = -I_{8}$$

$$\int Y_{1}H_{1}Y_{1}'dT_{1} = \int Y_{1}H_{1}Y_{1}'dT_{1} = -\left[\frac{Z^{2}S_{2}}{2} + (I-2Z)I_{10}\right] = D$$

$$= -\left[\frac{Z^{2}S_{2}}{2} + (3-2Z)I_{10}\right] = D$$

$$\int Y_{1}'H_{1}Y_{1}dT_{1} = \int Y_{1}'H_{1}Y_{1}dT_{1} = -\left[\frac{Z^{2}S_{2}}{2} + (I-Z)I_{1}\right] = D$$

$$\int Y_{1}'H_{1}Y_{1}'dT_{1}' = -\left[\frac{Z^{2}}{2} + (I-2Z)I_{11} + I_{12}\right] = E$$

$$\int Y_{1}'H_{2}Y_{1}'dT_{1}' = -\left[\frac{Z^{2}S_{2}}{2} + 2(I-Z)I_{13}\right] = M$$
(27)

where  $S_2 = \int \Psi \varphi' dr$ ,  $S_4 = \int \Psi' \varphi' dr$  and  $I_9 = 2I_{10}$ .

In terms of A, B, D, E and M, we have  $H_{II} = H_{III} = (2 + 2C^2)A + (2C^2 + 2C^4)E + I_4 - (4C + 4C^3)I_3 + 2C^2I_{14} + 4CI_{15} + C^4I_{21} + 4C^3I_{22} + 4C^2I_{23}$   $-H_{III} = -H_{III} = (2S + 4CS_2 + 2C^2S_4)B + (4CS + 8C^2S_2 + 4C^2S_4)D + 4(2C^2S_3 + 4C^3S_2 + 2C^4S_4)M + I_6 + 4CI_{16} + 2C^2(I_7 + I_{20}) + 4C^3I_{18} + 2C^2I_{19} + C^4I_{26}$   $H_{IIII} = H_{III} = H_{III} = H_{III} = -H_{IIII} = -H_{IIII} = -H_{IIII} = -H_{IIII} = -H_{IIII} = -H_{IIII} = H_{IIII} = 2A + I_3$   $H_{IIIII} = H_{IIIII} = 2BS + I_2$ (28)

Substituting I = ZF we get

$$\begin{split} W_{N} &= Z^{2} \left\{ 2 \left[ C^{2} / 1 + C^{2} \right) + 2 C \left[ 3 + C S_{2} \right] + 1 \right] - 2 \left[ 5 + 2 F_{2} \right] \left[ C^{2} / 5 + 2 C S_{2} + C^{2} S_{4} \right] + \\ &+ 2 C + \overline{J} - 4 \left[ S_{2} - 4 F_{0} \right] \left[ C^{2} C \left[ S + 2 C S_{2} + C^{2} S_{4} \right] + C C \right] - 2 \left[ 1 - 4 F_{4} \right] C^{2} C^{2} / 4 + C^{2} \right] + \\ &- 2 \left[ S_{4} + 4 F_{13} \right] C^{2} C^{2} \left[ S + 2 C S_{2} + C^{2} S_{4} \right] \right\} + Z \left\{ \frac{S}{4} - 4 \left[ 1 + F_{1} \right] \left[ C^{2} \left( 1 + C^{2} \right) + 4 C^{2} \right] + 2 C^{2} F_{4} + 3 C F_{5} + 2 C^{2} F_{4} + 3 C F_{5} + 2 C^{2} F_{4} + 3 C^{2} F_{5} + 3 C^{2} C \left[ 1 + C^{2} \right] C^{2} + C C \left[ S + C S_{2} \right] \right] - 2 4 F_{10} \left[ C^{2} C \left[ S + 2 C S_{2} + C^{2} S_{4} \right] + 4 C^{2} C^{2} F_{14} + 4 C^{2} C^{2} \left[ 1 + C^{2} \right] - 8 F_{13} C^{2} C^{2} \left[ S + 2 C S_{2} + C^{2} S_{4} \right] + 4 C^{2} C^{2} F_{14} + 4 C^{2} C^{2} \left[ F_{17} + F_{20} \right] + 3 G^{3} C^{2} F_{10} + 4 C^{2} C^{2} F_{14} + 4 C^{2} C^{2} F_{17} + F_{20} \right] + 3 G^{3} C^{2} F_{10} + 4 C^{2} C^{2} F_{17} + 4 C^{2} C^{2} F_{17} + 5 C^{2} F_{17} + 5 C^{2} F_{17} + 8 C^{3} C^{2} F_{22} + 8 C^{2} C^{2} F_{23} + 8 C C F_{24} + 8 C^{2} C F_{25} + 2 C^{2} F_{25} + 2 C^{2} F_{25} + 2 C^{2} F_{25} + 8 C^{2} C F_{25} + 2 C^{2} F_{25} + 8 C^{2} C F_{2$$

and

$$d = 2C^{2}(1+2C^{2}+5^{4})+2C^{2}[5^{2}+4655_{2}+C^{2}(45_{2}^{2}+255_{4}+46^{2}5_{2}5_{4}+65^{2})$$

$$(30)$$

Finally

$$W = \frac{\lambda_i}{\nu_i} Z^2 + \left(\frac{M_i}{\nu_i} + \frac{1}{\rho}\right) Z \tag{31}$$

and

$$W_{Z=min} = -\frac{\lambda_1}{4\lambda_1} \left( \frac{\mu_1}{\lambda_1} + \frac{L}{\rho} \right)^2 \tag{32}$$

The integrals occurring in (32) will be found in the appendix.

Expression (32) is a function of  $\rho$ , c and c. As both the Wang-ionic and Rosen treatment give the same internuclear distance it is a good approximation to use  $\rho=1.7$  for the Rosen-ionic treatment. Some of the integrals involved are expressed in terms of integral logarithms and are calculated by repeated use of recursion formulae, hence the value  $\rho=1.7$  was chosen in preference to  $\rho=1.69$  as with the former more accurate values could be obtained from the available tables of integral logarithms. With  $\rho=1.7$ , c and c were varied until the combination of c and c giving the lowest value for W was arrived at. The results obtained are c0 = 1.1515, c0 = 5.7, c0 = .07 and c0 = 1.190; c0 has not been varied as such a variation would require very long computations and the expected improvement in the value for W was small.

The comparison of the results of the different calculations can be seen from Fig. 1 and from the following table.

Present paper	W	⊿E(v.e.)	Je(gm-cm2)	$v_e (cm^{-1})$
Wang-ionic	-1.148	4.00	$4.65 \times 10^{41}$	4750
Rosen-ionic	-1.1515	4.10		
Heitler-London-Sugiura	-1.116	3.14	5.2	4800
Wang	-1.138	3.76	4.59	4900
Rosen	-1.1485	4.02	4.65	4260
Observed	-1.173	4.68	4.537 8	4418 8

The fundamental frequency  $\nu_e$  was calculated by means of Morse  $^9$  curve

$$\Delta E = De^{-2a(\rho_0 - \rho_m)} - 2De^{-a(\rho_0 - \rho_m)}$$
(33)

By substituting in (33) three different values of  $\rho_0$  with the corresponding values of  $\Delta E$  we obtain three simultaneous equations which when solved give D=0.1479 and a=1.216. Then  $\gamma_0$  was calculated from the formula

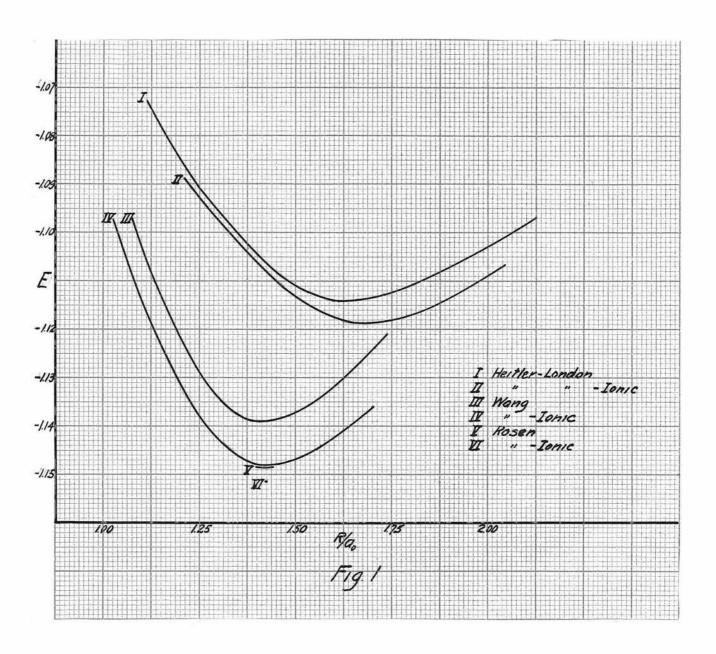
$$V_e = \frac{\alpha}{0.1227} \sqrt{\frac{Q}{M}} \tag{34}$$

where a is in 1/Å, D in wave numbers and  $M = \frac{M_1 M_2}{M_1 + M_2}$  in terms of oxygen = 16.

The table shows that the introduction of the ionic term in the wave function gives an improvement of 8% over Wang's value for the dissociation energy of the hydrogen molecule. Thus from a simple wave function results almost equivalent to Rosen's were obtained. It should be noticed that while the ionic term corresponds

to a definite physical picture, the physical significance of the Rosen term is not quite clear. The addition of the Rosen term to the atomic-ionic wave function gives only about  $2\frac{1}{2}\%$  increase in the value of the dissociation energy.

The writer wishes to thank Professor Pauling for suggesting the problem and for valuable criticism during the progress of the work.



## Appendix

The list of integrals used in this work follows.

The expressions for S and H functions will be found in the discussion of these integrals.

$$S = \int Y_{1}Y_{1}dY_{1} = e^{-P} \left( \frac{\rho^{2}}{3} + \rho + 1 \right)$$

$$S_{1} = \int Y_{1}Y_{1}dY_{1} = 0$$

$$S_{2} = \int Y_{1}Y_{1}dY_{1} = \frac{1}{2}\rho S$$

$$S_{3} = \int Y_{1}Y_{2}dY_{1} = \left( \frac{1}{2} \right) \left( \frac{\rho^{2}}{15} + \frac{2\rho^{3}}{15} - \frac{\rho^{2}}{5} - \rho - 1 \right)$$

$$F_{1} = \int \frac{Y_{1}^{2}}{16\eta} dY_{1} = e^{-P} \left( \frac{\rho^{2}}{15} + \frac{2\rho^{3}}{15} - \frac{\rho^{2}}{5} - \rho - 1 \right)$$

$$F_{2} = \int \frac{Y_{1}Y_{2}}{16\eta} dY_{1} dY_{2} = e^{-P} \left( \frac{1}{2} + \rho \right)$$

$$F_{3} = \int \frac{Y_{1}^{2}Y_{2}^{2}}{16\eta} dY_{1} dY_{2} = \frac{5}{8}$$

$$F_{4} = \int \frac{Y_{1}^{2}Y_{2}^{2}}{16\eta} dY_{1} dY_{2} = e^{-P} \left( \frac{5}{16\rho} + \frac{1}{9} + \rho \right) - e^{-\frac{3\rho}{2}} \left( \frac{5}{16\rho} + \frac{1}{9} \right)$$

$$F_{5} = \int \frac{Y_{1}Y_{2}^{2}}{16\eta} dY_{1} dY_{2} = e^{-P} \left( \frac{5}{16\rho} + \frac{1}{9} + \rho \right) - e^{-\frac{3\rho}{2}} \left( \frac{5}{16\rho} + \frac{1}{9} \right)$$

$$F_{5} = \int \frac{Y_{1}Y_{1}^{2}}{16\eta} dY_{2} dY_{3} = e^{-P} \left( \frac{5}{16\rho} + \frac{1}{9} + \rho \right) - e^{-\frac{3\rho}{2}} \left( \frac{5}{16\rho} + \frac{1}{9} \right)$$

$$F_{7} = \int \frac{Y_{1}Y_{1}^{2}}{16\eta} dY_{1} = 0$$

$$F_{7} = \int \frac{Y_{1}Y_{1}^{2}}{16\eta} dY_{1} = e^{-P} \left( \frac{5}{16\rho} + \frac{1}{2} + \frac{1}{2} + \frac{1}{2} \right)$$

$$F_{7} = \int \frac{Y_{1}Y_{1}^{2}}{16\eta} dY_{1} = e^{-P} \left( \frac{5}{16\rho} + \frac{1}{2} + \frac{1}{2} + \frac{1}{2} \right)$$

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$$F_{7} = \int \frac{Y_{1}Y_{1}^{2}}{16\eta} dY_{1} = e^{-P} \left( \frac{5}{16\rho} + \frac{1}{2} + \frac{1}{2} + \frac{1}{2} \right)$$

$$F_{7} = \int \frac{Y_{1}Y_{1}^{2}}{16\eta} dY_{1} = e^{-P} \left( \frac{5}{16\rho} + \frac{1}{2} + \frac{1}{2} + \frac{1}{2} \right)$$

$$F_{7} = \int \frac{Y_{1}Y_{1}^{2}}{16\eta} dY_{1} = e^{-P} \left( \frac{5}{16\rho} + \frac{1}{2} + \frac{1}{2} + \frac{1}{2} \right)$$

$$F_{7} = \int \frac{Y_{1}Y_{1}^{2}}{16\eta} dY_{1} = e^{-P} \left( \frac{1}{2} + \frac{1}{2} + \frac{1}{2} + \frac{1}{2} + \frac{1}{2} \right)$$

$$F_{8} = \int \frac{Y_{1}Y_{1}^{2}}{16\eta} dY_{1} = e^{-P} \left( \frac{1}{2} + \frac{1}{2} +$$

$$\begin{split} & \{g = \int_{\frac{1}{\sqrt{2}}}^{\frac{1}{\sqrt{2}}} d\eta = \frac{2}{3} \rho e^{\frac{1}{2}}(\rho+1) \} \\ & F_{10} = \int_{\frac{1}{\sqrt{2}}}^{\frac{1}{\sqrt{2}}} d\eta = \frac{1}{2} F_{2} \\ & F_{11} = \int_{\frac{1}{\sqrt{2}}}^{\frac{1}{\sqrt{2}}} d\eta = \frac{1}{\rho} + \frac{3}{\rho^{3}} - e^{-\frac{1}{2}\rho}(\rho^{2} + 3\rho + \frac{11}{2} + \frac{7}{\rho^{3}} + \frac{6}{\rho^{3}} + \frac{3}{\rho^{3}}) \} \\ & F_{13} = \int_{\frac{1}{\sqrt{2}}}^{\frac{1}{\sqrt{2}}} d\eta d\eta = \frac{1}{\rho} + \frac{3}{\rho^{3}} - e^{-\frac{1}{2}\rho}(\frac{1}{2} + \frac{41\rho^{3}}{120} + \frac{301\rho^{2}}{24\rho} + \frac{25\rho}{8} + \frac{3}{\rho^{3}}) \} \\ & F_{14} = \int_{\frac{1}{\sqrt{2}}}^{\frac{1}{\sqrt{2}}} d\eta d\eta = \frac{1}{\rho} + \frac{3}{\rho^{3}} - e^{-\frac{1}{2}\rho}(\frac{1}{2} + \frac{41\rho^{3}}{120} + \frac{301\rho^{2}}{24\rho} + \frac{25\rho}{8} + \frac{25\rho}{8} + \frac{25\rho}{16} +$$

$$F_{22} = \int \frac{4!4!}{2n} dr_1 dr_2 = \int_{P^2} + \frac{9}{P^4} - e^{-2}(\int_{P^2} + \frac{7}{P^2} + \frac{113}{P^2} + \frac{113}{240} + \frac{697p^2}{180} + \frac{3577p}{960} + 8 + \frac{11}{19} + \frac{119}{p^2} + \frac{11}{p^2} + \frac{263}{p^4} + \frac{3577p}{180} + 8 + \frac{11}{19} + \frac{119}{p^2} + \frac{119}{p^2} + \frac{263}{36} + \frac{11}{p^2} + \frac{119}{p^2} + \frac{263}{36} + \frac{11}{p^2} + \frac{11}{p^2} + \frac{263}{36} + \frac{11}{p^2} + \frac{11}{p^2} + \frac{263}{36} + \frac{11}{p^2} + \frac{11}{p^2}$$

$$+ 217S(3,6,p) - 240S(3,4,p) + 51S(3,2,p)$$

$$f_{27} = \int \frac{4,4,4,2}{4,2} d\tau, d\tau_{2} = f_{2} f_{25}$$

$$f_{0} = \int 4,5, d\tau_{3} = \frac{8e^{\frac{3}{2}}e^{\frac{3}{2}}}{(1+e)^{3}}$$

$$f_{0} = \int 4,5, d\tau_{3} = \frac{8e^{\frac{3}{2}}e^{\frac{3}{2}}}{(1+e)^{3}} \left[ e^{-P(1+\frac{4}{P(1-e^{2})}) + e^{-\frac{e_{1}}{P(1-e^{2})}} \right]$$

$$f_{1} = \int \frac{4,5}{4,1} d\tau_{3} = \frac{4e^{\frac{3}{2}}e^{\frac{3}{2}}}{(1+e)^{2}}$$

$$f_{2} = \int \frac{4,5}{4,1} d\tau_{3} = \frac{4e^{\frac{3}{2}}e^{\frac{3}{2}}}{P(1+e^{2})^{2}} \left[ 2ee^{-P+1}P[(1+e)p+2] \right]$$

$$f_{2} = \int \frac{4,5}{4,1} d\tau_{3} = \frac{4e^{\frac{3}{2}}e^{\frac{3}{2}}}{P(1-e^{2})^{2}} \left[ 2ee^{-P+e^{-P}[(1-e^{2})p-2e]} \right]$$

$$f_{3} = \int \frac{4,5}{4,1} d\tau_{3} = \frac{4e^{\frac{3}{2}}e^{\frac{3}{2}}}{P(1-e^{2})^{2}} \left[ 2e^{-P+e^{-P}[(1-e^{2})p+2]} \right]$$

$$f_{3} = \int \frac{4,5}{4,1} d\tau_{3} = \frac{4e^{\frac{3}{2}}e^{\frac{3}{2}}}{P(1-e^{2})^{2}} \left[ 2e^{-P+e^{-P}[(1-e^{2})p-2e]} \right]$$

$$f_{3} = \int \frac{4,5}{4,1} d\tau_{3} = \frac{4e^{\frac{3}{2}}e^{\frac{3}{2}}}{P(1-e^{2})^{2}} \left[ 2e^{-P+e^{-P}[(1-e^{2})p-2e]} \right]$$

$$f_{4} = \int \frac{4e^{\frac{3}{2}}e^{-\frac{3}{2$$

Most of these integrals were calculated by previous workers in this field. However, a number of them had to be calculated for the present paper, namely,  $t_0$ ,  $t_1$ ,  $t_2$ ,  $t_3$ ,  $t_4$ ,  $t_5$ , and  $t_6$ ,  $t_7$ ,  $t_8$ 

solving the interchange integrals of which  $F_{25}$  is a very complicated example will be discussed.

In expressing an interchange integral in elliptic coordinates  $\frac{1}{2\pi}$  is replaced by its Neumann expansion in terms of Legendre polynomials

The resulting integrals can be broken up into a sum of terms of the form  $H(m,n,\rho)$  and  $S(m,n,\rho)$  where

$$H(m,n,\rho) = \int_{\lambda_{1}}^{\infty} n e^{-\rho \lambda_{1}} Q_{0}(\lambda_{1}) d\lambda_{1} \int_{\lambda_{2}}^{\lambda_{1}} n e^{-\rho \lambda_{2}} d\lambda_{2} + \int_{\lambda_{1}}^{\infty} n e^{-\rho \lambda_{1}} Q_{0}(\lambda_{1}) d\lambda_{1} \int_{\lambda_{2}}^{\lambda_{1}} n e^{-\rho \lambda_{2}} d\lambda_{2} + \int_{\lambda_{1}}^{\infty} n e^{-\rho \lambda_{2}} d\lambda_{1} \int_{\lambda_{1}}^{\lambda_{1}} n e^{-\rho \lambda_{2}} d\lambda_{2} = \int_{\lambda_{1}}^{\infty} n e^{-\rho \lambda_{2}} d\lambda_{1} \int_{\lambda_{1}}^{\infty} n e^{-\rho \lambda_{2}} d\lambda_{2}$$

and

$$Q_o(\lambda) = \frac{1}{\lambda} \ln \frac{\lambda+1}{\lambda-1}$$

These quantities can be treated by the use of functions "

$$A_{n}(\rho) = \int_{\lambda}^{\infty} x^{n} e^{-\rho \lambda} d\lambda$$

$$F_{n}(\rho) = \int_{\lambda}^{\infty} x^{n} e^{-\rho \lambda} Q_{o}(\lambda) d\lambda$$

which satisfy the following recursion formulae

$$A_{n}(\rho) = \frac{1}{\rho} \left[ e^{-\rho} + n A_{n-1}(\rho) \right]$$

$$F_{n}(\rho) = F_{n-2}(\rho) + \frac{1}{\rho} \left[ n F_{n-1}(\rho) - (n-2) F_{n-3}(\rho) - A_{n-2}(\rho) \right]$$

where

$$A_{o}(\rho) = \frac{e^{7\rho}}{\rho}$$

$$F_{o}(\rho) = \frac{1}{2} [(\ln 2\rho + C) \frac{e^{7\rho}}{\rho} - Ei(-2\rho) \frac{e^{\rho}}{\rho}]$$

$$F_{i}(\rho) = \frac{1}{2} [(\ln 2\rho + C) e^{7\rho} (\frac{1}{\rho} + \frac{1}{\rho^{2}}) - Ei(-2\rho) e^{\rho} (-\frac{1}{\rho} + \frac{1}{\rho^{2}})]$$

$$C = 0.5772/6.....$$

From  $A_o$ ,  $E_o$  and  $E_i$ , tables of A and F can be compiled with the use of the recursion formulae.

The S functions obey the following recursion formula

$$S(m, n, \rho) = \frac{1}{\rho} [mS(m-1, n, \rho) + A_{m+n}(2\rho)]$$

with

$$S(0,n,p) = \frac{A_n(2p)}{p}$$

H functions, in turn, are expressed by

where

with

With the help of these functions the interchange integrals can now be calculated.

$$\begin{split} &F_{25} = \frac{1}{Z} \int \frac{4!}{3!} \frac{4!}{3!} \frac{4!}{2!} \frac{4!}{2!} \frac{4!}{2!} \frac{4!}{2!} \frac{2!}{2!} e_{ij} \cos \theta_{ij} \frac{4!}{2!} \cdot \\ & \quad \cdot \underbrace{E \frac{2n+1}{2!} P_{in} \binom{2i}{2i}}_{n_{i}} P_{in} \binom{n_{i}}{2i} \binom{n_{i}}{2i} \binom{n_{i}}{2i} \binom{n_{i}}{2i} \frac{2!}{2i} \binom{n_{i}}{2i} \frac{2!}{2i} \frac{2!}{2i} \cdot \\ & \quad \cdot d\lambda_{i} d\lambda_{i} dn_{i} dn_{i} dn_{i} dn_{j} dn_{j} = \\ & = \underbrace{\frac{2^{6}}{9}^{6}}_{8} \int_{1}^{\infty} \int_{1}^{\infty} \int_{1}^{\infty} \int_{1}^{\infty} \frac{2!}{2!} \binom{n_{i}}{2i} \binom{n_{i}$$

+ Po e [ [2, 2/-1+ 6) + (1-3+6-6)]e 2, Sa(2,) d2,. · [ (2-1)e-ph2d2+d2, [Qo(2)(2-1)e-ph2d2e]+  $+\frac{\rho^{5}}{\sqrt{\mu}}e^{-\rho}\int_{-2}^{2}(2+\frac{8}{\rho}+\frac{18}{\rho^{2}}+\frac{18}{\rho^{3}})+(2+\frac{12}{\rho}+\frac{54}{\rho^{2}}+\frac{174}{\rho^{3}}+\frac{360}{\rho^{4}}+$ + 360 ]e - ph, {[3/2, -322-1 Qo/2,]d2, [1/2-1/2]e - ph2 d2+  $+d\lambda_1 \int_{2}^{\infty} \left[ \frac{3}{2} \lambda_2 - \frac{3\lambda_2^2 - 1}{2} G_0(\lambda_2) \right] (\lambda_2^2 - \frac{1}{3}) e^{-\rho \lambda_2} d\lambda_2 + \int_{16}^{5} e^{\rho}.$  $\cdot \int \left[ -\lambda_{1}^{2} \left( 2 - \frac{8}{\rho} + \frac{18}{\rho^{2}} - \frac{18}{\rho^{3}} \right) + \left( 2 - \frac{12}{\rho} + \frac{54}{\rho^{2}} - \frac{174}{\rho^{3}} + \frac{360}{\rho^{9}} - \frac{360}{\rho^{5}} \right) \right] e^{-\rho \lambda_{1}} \lambda_{1} \cdot$  $\cdot \left\{ \left[ \frac{3}{2} \lambda_1 - \frac{3 \lambda_1^2 - 1}{2} \mathcal{Q}_0(\lambda_1) d\lambda_1 \right] \left( \lambda_2^2 - \frac{1}{3} \right) e^{-\rho \lambda_2} d\lambda_2 + d\lambda_1 \right\} \left[ \frac{3}{2} \lambda_2 + \frac{3}{2} \lambda_2^2 + \frac{3}{2} \lambda_2^2$  $-\frac{3\lambda_{2}^{2}-1}{2}G_{0}(\lambda_{2})](\lambda_{2}^{2}-\frac{1}{3})e^{-\rho\lambda_{2}}d\lambda_{2}$  = = f\_F + f = P - (1+ p)[H(3,2,p) - \frac{1}{3}H(3,0,p)] + (1+ \frac{3}{6} + \frac{6}{6} + \frac{6}{6}). · [H(1,2,p)-\$H(1,0,p]]}+f=ep{(-1+6)[H(3,2,p)-\$H(3,0,p]]+ + 9 /3 S(4,2,p) + 3 S(1,5,p) - 2 S(4,0,p) - 2 S(1,3,p) - 3 H(52,p)+ + 1/2 H(5,0,p) + 1/3,2,p) - 1/4(3,0,p)] + (1+6+27+87+180+ + 180 \[ \frac{3}{2} S(1,3,p) + \frac{3}{2} S(2,2,p) - \frac{1}{2} S(1,1,p) - \frac{1}{2} S(2,0,p) - \frac{3}{2} H(3,2,p) + + 1/ H(3,0,p)+ 1/ H(1,2,p)- 6H(1,0,p)] + + = eP - (1-4+2+ -9)[35(4,2,p)+35(1,5,p)-25(4,0,p)-25(1,3p)-3H(5,2,p)+ + 1/2 H(5,0,p) + 1/2 H(3,2,p) - 1/2 H(3,0,p) + (1-6 + 27 - 87) +

 $+ \frac{180}{p^{5}} - \frac{180}{p^{5}} \Big] \Big[ \frac{3}{2} S(2,2,p) + \frac{3}{2} S(1,3,p) - \frac{1}{2} S(2,0,p) - \frac{1}{2} S(1,1,p) + \\ - \frac{3}{2} H(3,2,p) + \frac{1}{2} H(3,0,p) + \frac{1}{2} H(1,2,p) - \frac{1}{6} H(1,0,p) \Big] \Big\} = \\ = F_{24} + f_{46}^{5} \Big\{ e^{-P} - 2(1+\frac{1}{p})A + 2(1+\frac{3}{p} + \frac{6}{p^{2}} + \frac{6}{p^{3}})B - (1+\frac{7}{p} + \frac{9}{p^{2}} + \frac{9}{p^{3}})C + \\ + (1+\frac{6}{p} + \frac{27}{p^{2}} + \frac{87}{p^{3}} + \frac{180}{p^{4}} + \frac{180}{p^{5}})D \Big] + e^{P} \Big[ 2(-1+\frac{1}{p})A + 2(1-\frac{3}{p} + \frac{6}{p^{2}} - \frac{6}{p^{3}})B + \\ - (1-\frac{7}{p} + \frac{9}{p^{2}} - \frac{9}{p^{3}})C + (1-\frac{6}{p} + \frac{27}{p^{2}} - \frac{87}{p^{3}} + \frac{180}{p^{4}} - \frac{180}{p^{5}})D \Big] \Big\}, \text{ where } \\ A = 3H(3,2,p) - H(3,0,p), B = 3H(1,2,p) - H(1,0,p), \\ C = 9S(4,2,p) + 9S(1,5,p) - 3S(4,0,p) - 3S(1,3,p) - 9H(5,2,p) + \\ + 3H(5,0,p) + 3H(3,2,p) - H(3,0,p), \text{ and } D = 9S(2,2,p) + \\ + 9S(1,3,p) - 3S(2,0,p) - 3S(1,1,p) - 9H(3,2,p) + 3H(3,0,p) + \\ + 3H(1,2,p) - H(1,0,p) \Big\}$ 

$$\int \frac{4\pi}{k_{1}} d\tau_{1} = \frac{7^{2}}{R^{2}} \frac{1}{N} \int \frac{4\pi r^{2}}{R^{2}} e^{-\frac{2r}{N}r} e^{-\frac{2r}{N}r} e^{-\frac{2r}{N}r} \int \frac{4\pi r^{2}}{R^{2}} e^{-\frac{2r}{N}r} e^{-\frac{2r}{N}r} e^{-\frac{2r}{N}r} \int \frac{4\pi r^{2}}{R^{2}} e^{-\frac{2r}{N}r} e^{-\frac{2r}{N}r$$

$$-\frac{aD}{2} \left[ e^{-\beta D} - 1 - \frac{1}{\beta D} \right] + e^{\beta D} - 1 + \frac{1}{\beta D} \right] - \frac{aD}{2} \left[ e^{-\beta D} - 1 - \frac{1}{\beta D} - \frac{2}{\beta^2 D^2} \right] + \\
+ e^{\beta D} \left[ 1 - \frac{2}{\beta D} + \frac{2}{\beta^2 D^2} \right] - 2 \left[ e^{-\beta D} - 1 - \frac{1}{\beta D} \right] - e^{+\beta D} \left[ 1 - \frac{1}{\beta D} \right] \right] d\lambda = \\
= \frac{8 \varepsilon^3 ZD}{(1 + 2 \varepsilon^3 D)} \left\{ -e^{-\frac{1}{2} \left( \frac{1}{\beta^2 D} \right)} + e^{-\frac{1}{2} \left( \frac{1}{\beta^2 D} \right)} + e^{-\frac{1}$$

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