

Thermal Gasification of Rice Husks from Rice Growing Areas in Mwea, Embu County, Kenya

Paul Njogu^{1*}, Robert Kinvua¹, Purity Muthoni¹, Yusuvuki Nemoto²

¹Institute of Energy and Environmental Technology, Jomo Kenyatta University of Agriculture and Technology, Nairobi, Kenva

²Division of Renewable Energy and Environment, Ashikaga Institute of Technology, Ashikaga, Japan Email: ^{*}njogupaul@jkuat.ac.ke, ^{*}njogupl@yahoo.com

Received 5 February 2015; accepted 23 May 2015; published 26 May 2015

Copyright © 2015 by authors and Scientific Research Publishing Inc. This work is licensed under the Creative Commons Attribution International License (CC BY). http://creativecommons.org/licenses/by/4.0/ (i) (ii)

Abstract

Thermal gasification of biomass provides a potential renewable energy resource in rural areas in Kenva. Rice husks are a key byproduct of rice production that are not considered of economic value to millers. Rice husks obtained from Mwea, Embu County, Kenya were converted into syngas using a locally assembled modified updraft gasifier. The syngas production was produced at temperatures between 450°C and 750°C with injection of limited supply of air and purified using series of gas cleaning and cooling devices. Proximate analysis shows that rice husks have a mean content of 21.9% ± 0.3% ash, 9.5% ± 3.3% moisture, 78.8% ± 0.3% volatiles and 91.8% ± 1.3% total solids. Carbonized rice husks have mean contents of $37.8\% \pm 1.2\%$ ash, $3.1\% \pm 0.4\%$ moisture, 62.3% ± 1.2% volatiles and 96.5% ± 0.4% total solids. The study shows that feedstock is consumed at a rate of 25 - 32 Kg/Hr with gas generation rate of 7.76 - 7.78 m³/hr; this translates to a gas yield of 0.31 - 0.35 m³/Kg. Process water was re-circulated at a rate of 2.2 m³/hr within the plant. The total electricity consumption per hour was 1.1 - 1.3 kWh. Carbon monoxide (CO) and temperature were monitored in the working area to assess the safety of the workers and were found to be in the range of 35 - 50 ppm and 24°C - 29.5°C respectively. The two were found to be within safe limits; however, the CO concentrations increased when leakages occurred. Syngas was found to be composed of 16.5% - 17.55% CO, 14.5% - 16.1% CO₂, 4.1% - 4.5% H₂, 6.8% - 7.2% CH₄ and 17.9% -45.7% N₂ among others. The gas was used for direct heating applications and to run modified petrol engines. Carbonized husks were used to make energy briquettes and partly applied to the rice growing pads to improve soil properties. The technology provides energy solutions and aids in the abatement of climate change mitigation and abatement since it provides a permanent carbon sink. The technology provides a value addition chain for rice growers.

^{*}Corresponding author.

How to cite this paper: Njogu, P., Kinyua, R., Muthoni, P. and Nemoto, Y. (2015) Thermal Gasification of Rice Husks from Rice Growing Areas in Mwea, Embu County, Kenya. Smart Grid and Renewable Energy, 6, 113-119. http://dx.doi.org/10.4236/sgre.2015.65010

Keywords

Carbonization, Climate Change, Gasification, Syngas

1. Introduction

The demand for energy in Kenya is growing rapidly due to industrial activities which have led to the high increases in prices [1]. The consumption of fossil fuels has gone with a lot of air pollution and the climate change impacts due to increase in the emission of green house gases such as carbon dioxide and other pollutants.

Rice is grown in four areas in Kenya, namely, Mwea in Kirinyaga county, Ahero and West Kano in Kisumu county and Bunyala in Busia county [2]. Kenya's rice production has ranged between 45,000 tons and 50,000 tons per year. This generates byproducts such as rice husk and rice straw [1]. Rice husk surrounds the paddy grain. During milling of paddy, about 78% of weight is produced as rice, broken rice and bran, and the rest 22% of weight of paddy is produced as husk [1] [2]. The residues can be utilized as animal feed, soil conditioner and a renewable energy resource. The use of rice husks, however, is low due to its low density and poor nutritional value. A small proportion is carbonized in the farms and applied in the rice pads as soil conditioner.

Direct use of rice husks as an energy source is hampered by low density and low heat value. It is thus imperative to convert it into combustible gas. Rice husks contain 75% organic volatile matter and 25% ash [3]. Rice husks can be converted thermally, biologically or chemically to other usable forms of energy like methane gas, liquid fuels (ethanol) and syngas/process gas. This makes it necessary to come up with innovative ways of sustainable utilization of such wastes. The current practice is open burning which leads to transfer of pollutants from land to the atmosphere.

1.1. Thermal-Gasification

Thermal gasification provides one of the most promising approaches of waste to energy (WtE) conversion for rice husks [3] [4]. It makes use of heat in oxygen deficient environments to transform biomass to combustible gas. The combustible gas can be used to power internal combustion engines coupled with electricity generators for electricity generation which can be used in off grid rural electrification in remote areas especially in areas with huge annual biomass yield.

1.2. Process Zones in an Updraft Gasifier

As the fuel moves down in the gasifier, it passes through various reaction zones, namely, drying, pyrolysis, oxidation and reduction as illustrated in Figure 1.

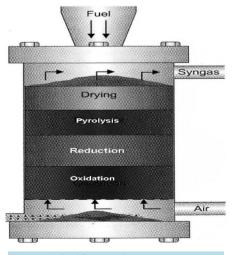


Figure 1. Gasifier reaction zones

1.2.1. Drying Zone

Here, the biomass gets dried due to heat moisture present in the fuel converts to steam:

Main Feedstock + Heat
$$\rightarrow$$
 Dry Feedstock + H₂O_(g) (1)

1.2.2. Pyrolysis Zone

Pyrolysis is a thermo chemical decomposition of biomass into a range of useful products, either in the total absence of oxidizing agents or with a limited supply that does not permit gasification to an appreciable extent. It is one of several reaction steps or zones observed in a gasifier. During pyrolysis, large complex hydrocarbon molecules of biomass break down into relatively smaller and simpler molecules of gas, liquid, and char. It takes place at a temperature greater than 250°C [3]:

Main Feedstock + Heat
$$\rightarrow$$
 simple molecules + $H_2O_{(g)}$ (2)

1.2.3. Oxidation/Combustion Zone

An oxidation or burning zone is formed in the section where air/oxygen is supplied. These combustion reactions are highly exothermic and cause a rapid temperature increase up to 1100° C - 1500° C. The reactions are as follows [3]:

$$C + O_2 \rightarrow CO_2$$
 + 393 MJ/kg·mole (3)

$$2H_2 + O_2 \rightarrow 2H_2O + 242 \text{ MJ/kg} \cdot \text{mole}$$
 (4)

1.2.4. Reaction Zone

This is the main zone of gasifier where we get the syngas. The reactions are as follows:

 $C + O_2 \rightarrow 2CO$ -164.9 MJ/kg·mole (5)

$$C + H_2O \rightarrow CO + 2H$$
 $-122.6 \text{ MJ/kg} \cdot \text{mole}$ (6)

$$\text{CO} + \text{H}_2\text{O} \rightarrow \text{CO}_2 + \text{H}_2 + 42.3 \text{ MJ/kg} \cdot \text{mole}$$
 (7)

$$C + 2H_2 \rightarrow CH_4 + 75 \text{ MJ/kg} \cdot \text{mole}$$
 (8)

$$\text{CO} + 3\text{H}_2 \rightarrow \text{CH}_4 + \text{H}_2\text{O}_{(g)} - 205.9 \text{ MJ/kg} \cdot \text{mole}$$
(9)

The mixture of gases produced in this zone is called syngas or producer gas.

1.2.5. Advantages of Rice Husks Gasification

Syngas production from rice husks presents one of the most promising treatments and disposal methods that incorporate energy recovery. The resultant gas can be used in direct heating applications as well as used as fuel for internal combustion engines. The carbonized husk produced has various applications which include use as activated carbon, and use as odor absorbent in animal pens; the spent husks have also been found to increase the soil conditions of agricultural soils by improving the water retention capacity. The spent husks also provide a climate mitigation opportunity due to production of biochar which presents a permanent sink for carbon [3]-[6]. The high-temperature process refines out corrosive ash elements such as chloride and potassium, allowing clean gas production [7]-[9]. The high temperatures also produce a sterilized product that can be used as a growth medium for sensitive plants such as mushrooms.

2. Methods and Data Sources

2.1. Experimental Set-Up and Design

The study was conducted at the Institute of Energy and Environmental Technology, Jomo Kenyatta University of Agriculture and Technology. Rice husks were sourced from Mwea, Embu County from Nice Rice Millers limited. The project is funded by the Japanese International Cooperation Agency through the JICA-BRIGHT project.

2.2. Gasifier Design and Fabrication

An updraft gasifier was designed and fabricated with cleaning accessories and air injection systems. The gasifier was designed as described by Belonio [10]. The gasifier was made using stainless steel and mild steel sheets for inner and outside cover respectively. Temperature probes and pressure gauges were fitted to monitor operation parameters and the plant run in batch modes. The updraft gasifier is a medium size unit with a 400 mm internal diameter reactor with 75 mm gas outlet, 600 watts air blower, a 370 W water pump for water recirculation, pressure gauges, thermocouples and a temperature read out. The feed were converted to syngas (carbon monoxide (CO), methane (CH₄) and hydrogen (H₂) carbon dioxide (CO₂), nitrogen (N₂) and other traces substances. The reactor has a 150 mm lagging to prevent heat losses and an internal perforated 3 inch pipe for gas uptake and delivery. It is fed from the top and carbonized husk removed from bottom using a screw conveyor.

2.3. Rice Husks Analysis

The feedstock was analyzed to determine quality for use in the gasifier. Proximate analyses were performed to determine the ash content, moisture content, total solids and volatile organic compounds. Standard methods of analysis were used in all cases [11].

2.4. Syngas Production and Upgrading

The feed were dried in the open to achieve moisture contents of <12% then fed to the reactor manually. The raw gas was upgraded by passing through the gas-cleaning and cooling devices; wet scrubbers (concurrent and countercurrent spray water towers), a cyclone separator, a venturi separator, a saw dust adsorbent and a coarse and fine fabric filter.

2.4.1. Syngas Analysis

Syngas was analyzed for compositional analysis using gas chromatographic analyzer with a thermal conductivity detector (GC-TCD). The gas was sampled using gas sampling balloons and injected into the detector without pretreatment. Samples were run in triplicates. The peak areas were determined and the gas composition derived as percentages.

2.4.2. Process Optimization

The gas production was related to the gasification temperatures and air injection rate. The pressure in the system was also monitored in order to come up with the optimal operating conditions. The moisture content was also monitored and related to syngas yields.

2.4.3. Equipment Calibration

The air injection Blower, water pump and the gas flow meter were calibrated before use and corrections made where applicable.

2.4.4. Combustion Emission Analysis

Gasifier emissions were analyzed for carbon monoxide to ensure the safety of the plant workers. Workers were provided with personal protective equipments such as dust coats, helmets, dust masks and eye protection devices. The plant was also designed to allow maximum air flow to and from outdoor environment. A carbon monoxide detector meter was fitted two meters above ground to monitor ambient CO concentrations.

2.4.5. Process Water Analysis

Process water was analyzed before and after use for safe disposal. Parameters measured included temperature, total dissolved solids (TDS), conductivity, pH and the changes monitored.

3. Results and Discussions

3.1. Rice Husk Gasifier

The gasification system is presented in Figure 2 and Figure 3; it includes Feeder hoper, a modified updraft ga-

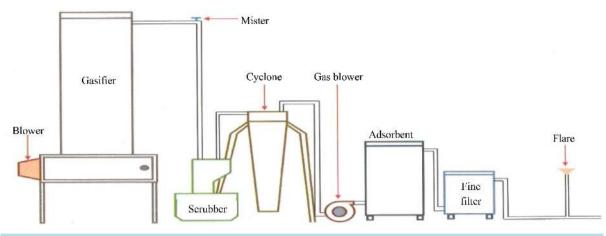


Figure 2. A Schematic for the gasification plant at JKUAT, Kenya.



Figure 3. Gasifier in the Shed.

sifier, Cyclone separator, Air Injection port, Blower, Ash collection device, two water spray towers, A venturi scrubber, process water recirculation unit, A solid adsorbent bed, A fabric Filter, A smoke Vent, Gas transportation pipes and control valves.

3.2. Proximate Analysis Data

Proximate data analysis is presented in **Table 1**. The data shows that the husks contained high concentrations of volatiles followed by fixed carbon, moisture and ash respectively.

3.3. Syngas Composition

The syngas contains an approximate total of 29.2% combustible gas and the rest non combustible gases including nitrogen, carbon dioxide and others (Table 2).

3.4. Gasifier Performance Parameters

Gasifier performance data is presented in Table 3.

3.5. Equipment Specifications

Air injection Blower-Blower from Black and Decker Model KTX 5000, Power rating-600 W. Water Pump-from

P. Njogu et al.

Component		Ash	Moisture		Volatiles	Total solids
Raw rice husk		21.9 ± 0.3	9.5 ± 3.3		78.8 ± 0.7	91.8 ± 0.4
Spent husk		37.8 ± 1.2	3.1 ± 0.4		62.3 ± 1.2	96.5 ± 0.4
11.2.0	·.· 1	1 (0/)				
able 2. Syn	ngas compositional CO	data (%). H ₂	CH ₄	CO ₂	N ₂	Others
	U		CH ₄ 6.8 - 7.2	CO ₂ 14.5 - 16.1	N ₂ 16.9 - 45.7	Others 1.24 - 1.4

Table 3. Performance parameters of gasification.

Parameter	Updraft gasification plant		
Volume syngas produced (m ³ /hr)	7.76 - 7.78		
Gas generation (m ³ /ton)	310 - 350		
Gasification temperature (°C)	450 - 750		
Gas temperature of outlet after cleaning and cooling (°C)	55		
Consumption of biomass (Kg/hr)	25 - 32		
Volume of circulating water (m ³ /hr)	2.2		
CO concentration (ppm)	35 - 50		
Weight reduction per batch (Kg)	7.2 - 7.5		
Temperature in working area (°C)	24 - 29.5		
Ash removing manner	Scooping		
Device of gas cleaning and cooling system	Cyclone separator, 2 spray scrubbers, 1 venturi scrubber, filter (removing dust, tar and cooling), air blower		

Pedrollo Made in Italy Model PKM 60, Supplied by Davis and Shirtliff. Power rating-0.37 W. Gas volume Meter (m^3), maximum pressure-10 kPa, maximum flow rate of 2.5 M^3 /hr.

3.6. Occupational Safety Data

The carbon monoxide concentration in the working area was monitored using a carbon monoxide (CO) detector meter Model CA895 from Chauvin Arnoux. The CO was found to be in the range of 35 - 50 ppm which is within the maximum bearable level for continuous exposure for 8 hours. However higher levels were recorded where leakage occurred. Temperature in working area was in the range of 24° C - 29.5° C which is within safe limits for continuous exposure for 8 hours.

3.7. Process Water Data

Data are presented in **Table 4**. The pH decreased by 33.2% which can be attributed to the dissolution of CO₂, TDS, temperature and conductivity increased by 4.2%, 17.2% and 4.7% respectively.

4. Conclusion and Recommendations

The present study shows great potential for conversion of rice husks into clean gas fuel and carbonized husks. Carbonized husks were used to make energy briquettes and partly applied to the rice growing pads to improve soil properties. The technology provides energy solutions and aids in the abatement of climate change mitigation and abatement since it provides a permanent carbon sink. The technology provides a value addition chain for rice growers.

able 4. Process water data.							
Parameter	Initial	After gasification	% change				
Conductivity (µS)	325 ± 2.7	341.7 ± 4.51	+4.7				
TDS (mg/L)	165.7 ± 2.5	172.3 ± 3.51	+4.2				
Temperature (°C)	23.13 ± 0.5	27 ± 0.53	+17.2				
pH	6.83 ± 0.3	3.8 ± 0.03	-33.2				

Acknowledgements

Authors wish to acknowledge technicians in the engineering workshops. Mwea Rice Millers limited for providing the Rice Husks. The Japan International Cooperation Agency (JICA) and JICA-BRIGHT project for the financial support.

References

- Government of Kenya (2009) National Rice Development Strategic Plan (2008-2018). Ministry of Agriculture, Nairobi.
- [2] TARDA (2009) Tana Delta Irrigation Project.
- [3] Rajvansh, A.K. (2013) Nimbkar Agricultural Research Institute, PHALTAN-415523, 1986. www.narphaltan.org
- [4] Maharashtra, I., Dutta R. and Dutt, G.S. (1981) Producer Gas Engines in Villages of Less-Developed Countries. Science, 213, 731-736. <u>http://dx.doi.org/10.1126/science.213.4509.731</u>
- [5] Kaupp, A. and Goss, J.R. (1983) Technical and Economic Problems in the Gasification of Rice Hulls. Physical and Chemical Properties. *Energy in Agriculture*, **1**, 201-209.
- [6] Kaupp, A. and Goss, J.R. (1981) State of the Art for Small Scale (to 50 kW) Gas Art for Small Scale (to 50 kW) Gas Producer-Engine Producer-Engine Systems. Final Report, Chap 5, U.S.D.A., Forest Service.
- [7] Foley, G. and Barnard, G. (1983) Biomass Gasification in Developing Countries, Technical Report No. 1, Chap 2.4, Earthscan, London.
- [8] Van Der Heijden, S., Szladow, A.J., Barabas, M. and Sirianni, G. (1981) Wood Gasification system for Electricity Production. *Proceedings of 16th Intersociety Energy Conversion Engineering* (16th IECEC), Atlanta, 9-14 August 1981.
- [9] Stickney, R.E., Piamonte, V.N. and Belonio, A.T. (1989) Converting Rice Hull to Gaseous Fuel. Appropriate Technology, 16, 14-16.
- [10] Van Der Heijden, S., Szlakow, A.J., Barabas, M. and Belonio, A.T. (2005) Rice Husk Gas Stove Handbook. Appropriate Technology Center Department of Agricultural Engineering and Environmental Management College of Agriculture Central Philippine University Iloilo.
- [11] Eaton, A.D., Clesceri, L.S., Greenberg, A.E. and Franson, M.A.H. (1995) Standard Methods for the Examination of Water and Wastewater. American Public Health Association, Washington DC.