

# Thiazole-Based Thiosemicarbazones: Synthesis, Cytotoxicity Evaluation and Molecular Docking Study

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**Introduction:** Hybrid drug design has developed as a prime method for the development of novel anticancer therapies that can theoretically solve much of the pharmacokinetic disadvantages of traditional anticancer drugs. Thus a number of studies have indicated that thiazole-thiophene hybrids and their bis derivatives have important anticancer activity. Mammalian Rab7b protein is a member of the Rab GTPase protein family that controls the trafficking from endosomes to the TGN. Alteration in the Rab7b expression is implicated in differentiation of malignant cells, causing cancer.

**Methods:** 1-(4-Methyl-2-(2-(1-(thiophen-2-yl) ethylidene) hydrazinyl) thiazol-5-yl) ethanone was used as building block for synthesis of novel series of 5-(1-(2-(thiazol-2-yl) hydrazono) ethyl) thiazole derivatives. The bioactivities of the synthesized compounds were evaluated with respect to their antitumor activities against MCF-7 tumor cells using MTT assay. Computer-aided docking protocol was performed to study the possible molecular interactions between the newly synthetic thiazole compounds and the active binding site of the target protein Rab7b. Moreover, the in silico prediction of adsorption, distribution, metabolism, excretion (ADME) and toxicity (T) properties of synthesized compounds were carried out using admetSAR tool.

**Results:** The results obtained showed that derivatives **9** and **11b** have promising activity ( $IC_{50} = 14.6 \pm 0.8$  and  $28.3 \pm 1.5 \mu M$ , respectively) compared to Cisplatin ( $IC_{50} = 13.6 \pm 0.9 \mu M$ ). The molecular docking analysis reveals that the synthesized compounds are predicted to be fit into the binding site of the target Rab7b. In summary, the synthetic thiazole compounds **1–17** could be used as potent inhibitors as anticancer drugs.

**Conclusion:** Promising anticancer activity of compounds **9** and **11** compared with cisplatin reference drug suggests that these ligands may contribute as lead compounds in search of new anticancer agents to combat chemo-resistance.

**Keywords:** thiazoles, hydrazones, hydrazonoyl halides, docking, Rab7b, MCF-7

## Introduction

Cancer is a broad concept that encompasses a wide variety of diseases, essentially marked by spontaneous growth and cell proliferation, with failures in the division routes known as the cell cycle. It is a major global public health problem as it is the world's second-largest cause of death, with approximate 9.6 million deaths in 2018.<sup>1</sup> Investigation of novel compounds that may be of use in designing new less toxic, selective, and potent anti-cancer agents is still the main challenge for medical chemists. Cisplatin is one of the most effective anticancer agents widely

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used in the treatment of breast cancer. It prevents DNA replication in cancer cells by a ligand replacement reaction with DNA in which a bond is formed between platinum and a nitrogen atom on guanine.<sup>2</sup> Several studies on various diseases have been performed with several sulfur heterocycles, including thiophene and thiazole. The literature reports that thiophene core compounds have drawn significant attention in the areas of drug discovery because of their versatile and wide range of biological activities, which include antimicrobial,<sup>3</sup> anti-inflammatory,<sup>4</sup> antidepressant,<sup>5</sup> analgesic,<sup>6</sup> anticonvulsant.<sup>7</sup> In addition, thiophene, being one of the main scaffolds, is continuously being sought by several researchers to develop potential cancer agents. Thiophene derivatives have been identified as anti-cancer agents for several years and exhibit their influence via different cancer pathways.<sup>8–13</sup> Thiazole-containing drugs, on the other hand, have demonstrated their involvement in a variety of commercially available anti-cancer medications, such as tiazofurin (inhibitor of IMP dehydrogenase),<sup>14</sup> dasatinib (Bcr-Abl tyrosine kinase inhibitor),<sup>15</sup> dabrafenib (inhibitor of enzyme B-RAF),<sup>16</sup> ixabepilone (stabilization of microtubules),<sup>17</sup> and epothilone (inhibition of microtubule function)<sup>18</sup> (Figure 1).

Thiazole-containing compounds depict anticancer activity profile through diverse mechanisms.<sup>19–27</sup>

Molecular hybridization is a beneficial approach to structural alteration involving the integration in a single species of two or more pharmacophores. Over the last several years, hybrid drug design has developed as a prime method for the development of novel anticancer therapies that can theoretically solve much of the pharmacokinetic disadvantages of traditional anticancer drugs.<sup>28,29</sup> Thus, a number of studies have indicated that thiazole-thiophene hybrids and their bis derivatives have important anticancer activity.<sup>30–34</sup> Based on the above-mentioned promising aspects, the strategy of this work includes gathering the two bioactive entities thiophene-thiazole in one compact structure for the purpose of synergism and examined the prepared compounds as anticancer agents. Activity against anticancer cell lines would expect to show remarkable activity. The *in vitro* cytotoxic potential of the newly synthesized compounds was examined against the human breast cancer cell line (MCF-7) using the MTT assay and the results showed compounds **9** and **11b** have promising activity.

Rab proteins are crucial regulators of all aspects of membrane trafficking in all cell types.<sup>35,36</sup> Rab7b belongs

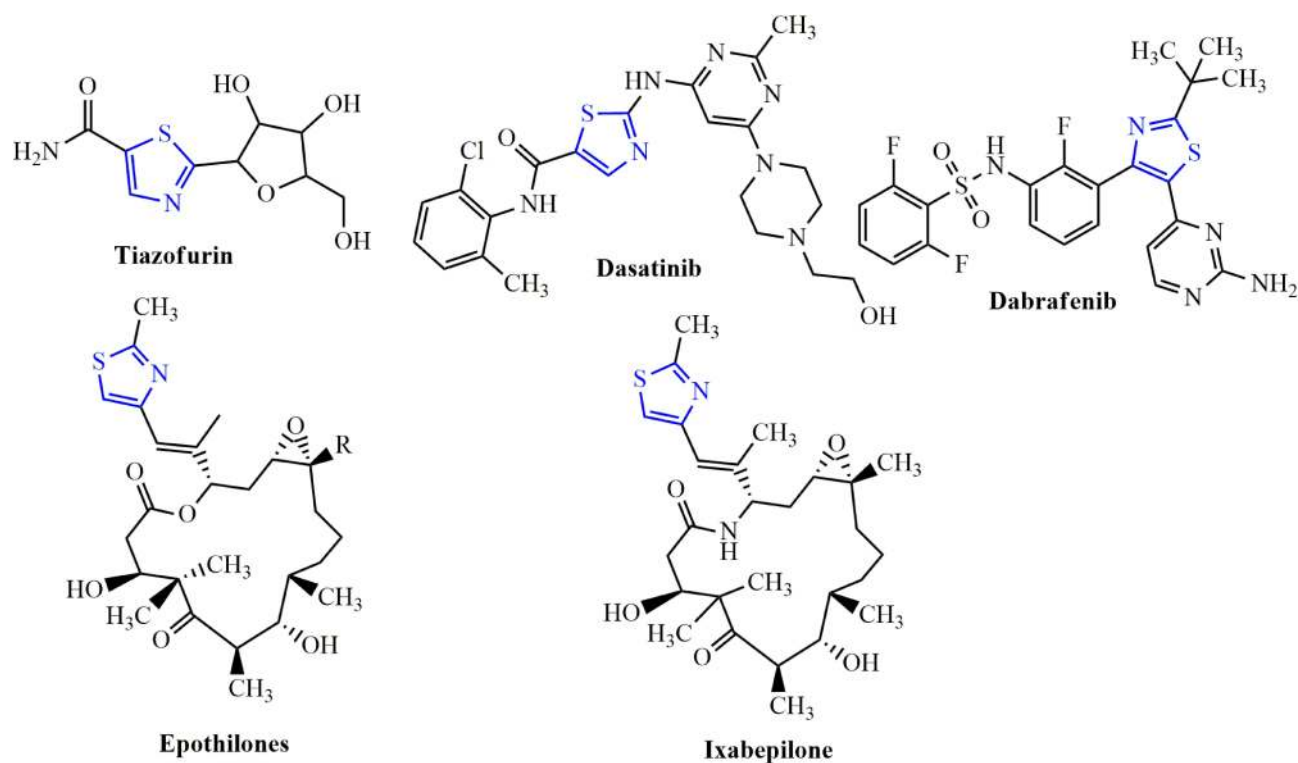


Figure 1 Examples of thiazole bearing anticancer drugs.

to Ras superfamily of small GTPases, which is vital for various cellular processes.<sup>37</sup> Human Rab7b is involved in regulating membrane transport from early to late endosomes; in addition, Rab7b has important roles in lipid metabolism, growth factor signaling, autophagy, and phagolysosome biogenesis.<sup>38</sup> Moreover, the biochemical pathway of Homo sapiens Rab7b is adapted from BioRender web-based tool, as represented in [Figure S1](#), in the [supplementary data](#) section, declared the critical role of the target protein in regulating signal transduction processes leading to cytoskeletal – dependent responses.<sup>39</sup> The cycle of Rab7b protein declares that the target is activated to GTP bound form by GEF (guanine nucleotide exchange factor) and deactivated to GDP bound form by GAP (GTPase-activating protein).<sup>40,41</sup> The active form of Rab7b protein plays a critical role in endocytic trafficking and other degradative pathways like phagocytosis and autophagy.<sup>42</sup> The overexpression of Rab7b is characterized in human cancer progression.<sup>43–46</sup> Finally, human Rab7b protein is targeted for the identification of potent drug candidates against cancer. The molecular docking studies were carried out to understand the molecular interactions between the active site of the target Rab7b and the examined compounds.

## Materials and Methods

### Chemistry

All melting points were determined on an electrothermal apparatus and were left uncorrected. IR spectra were recorded (KBr discs) on a Shimadzu FT-IR 8201 PC spectrophotometer. <sup>1</sup>H-NMR and <sup>13</sup>C-NMR spectra were recorded in DMSO solutions on BRUKER 400 FT-NMR system spectrometer and chemical shifts were expressed in ppm units using TMS as an internal reference. Mass spectra were recorded on a GC-MS QP1000 EX Shimadzu. Elemental analyses were carried out at the Microanalytical Center of Cairo University.<sup>47</sup> The spectral data are shown in [Figure S2](#), in the [supplementary data](#) section.

### Synthesis of Thiazole Derivative 3

A mixture of acetyl thiazole derivative **1** (0.279 g, 1 mmol) and thiosemicarbazide (**2**) (0.091 g, 1 mmol) in EtOH (20 mL) containing drops of HCl was refluxed for 2–4 h (monitored by TLC, using n-Hexane/ethyl acetate (2:1) as elution solvent). The formed solid was filtered and recrystallized from dioxane solvent to give 2-(1-(4-methyl-2-(2-(1-(thiophen-2-yl) ethylidene) hydrazinyl) thiazol-

5-yl) ethylidene) hydrazinecarbothio-amide (**3**). Yellow solid, 80% yield, m.p. 220–222°C (DMF); <sup>1</sup>H-NMR (DMSO-d<sub>6</sub>): δ 2.39 (s, 3H, CH<sub>3</sub>), 2.35 (s, 3H, CH<sub>3</sub>), 2.37 (s, 3H, CH<sub>3</sub>), 7.08–7.56 (m, 4H, Ar-H, NH), 8.73 (br s, 2H, NH<sub>2</sub>), 10.41 (br s, 1H, NH) ppm; <sup>13</sup>C-NMR (DMSO-d<sub>6</sub>): δ 16.1, 17.9, 22.1 (CH<sub>3</sub>), 114.8, 114.9, 115.0, 130.0, 130.1, 134.3, 146.1, 149.4, 160.4 (Ar-C and C=N), 185.1 (C=S) ppm; IR (KBr): ν 3409 (br. 2NH) cm<sup>-1</sup>; MS m/z (%): 354 (M<sup>+</sup> +1, 2), 353 (M<sup>+</sup>, 1), 313 (5), 281 (7), 239 (7), 199 (8), 182 (12), 155 (6), 140 (7), 129 (17), 124 (22), 111 (29), 101 (32), 97 (39), 83 (45), 69 (56), 57 (100). Anal. Calcd for C<sub>13</sub>H<sub>16</sub>N<sub>6</sub>S<sub>3</sub> (352.50): C, 44.29; H, 4.58; N, 23.84. Found: C, 44.33; H, 4.45; N, 23.65%.

### Synthesis of Thiocarbohydrazone Derivative 5

A mixture of acetylthiazole **1** (2.79 g, 10 mmol) and thiocarbohydrazide (**4**) (1.06 g, 10 mmol) in 20 mL EtOH and HCl (2 drops) for 2–4 h (monitored by TLC). The formed solid product was filtered and crystallized from DMF to afford thiocarbohydrazone derivative **5** as yellow solid, 79% yield, m.p. 204–206 °C; <sup>1</sup>H-NMR (DMSO-d<sub>6</sub>): δ 2.11 (s, 3H, CH<sub>3</sub>), 2.33 (s, 3H, CH<sub>3</sub>), 2.43 (s, 3H, CH<sub>3</sub>), 5.40 (br.s, 2H, NH<sub>2</sub>), 7.02–7.62 (m, 4H, Ar-H and NH), 8.68 (br.s, 1H, NH), 9.32 (br.s, 1H, NH) ppm; IR (KBr): ν 3426–3239 (3NH + NH<sub>2</sub>), 1601 (C=N) cm<sup>-1</sup>; MS m/z (%): 368 (M<sup>+</sup> +1, 2), 367 (M<sup>+</sup>, 2), 313 (3), 302 (2), 267 (2), 256 (4), 232 (7), 191 (4), 178 (8), 165 (8), 139 (10), 128 (35), 125 (28), 110 (47), 97 (46), 84 (48), 69 (70), 57 (100). Anal. Calcd for C<sub>13</sub>H<sub>17</sub>N<sub>7</sub>S<sub>3</sub> (367.52): C, 42.48; H, 4.66; N, 26.68. Found: C, 42.36; H, 4.47; N, 26.59%.

### Synthesis of Thiazole Derivatives 7, 9 and 11a, b

General procedure. A mixture of thiosemicarbazone derivative **3** (0.352 g, 1 mmol) and the appropriate α-halocarbonyl compound **6**, **8** and **10a, b** (1 mmol) in EtOH (20 mL) was refluxed for 2–4 h (monitored by TLC). The solvent was evaporated under vacuum pressure and the formed solid was crystallized from the appropriate solvent to give **7, 9** and **11a, b** respectively.

**Compound 7.** Dark green solid, 71% yield, m.p. 210–212 °C; (Dioxane); <sup>1</sup>H-NMR (DMSO-d<sub>6</sub>): δ 2.07 (s, 3H, CH<sub>3</sub>), 2.24 (s, 3H, CH<sub>3</sub>), 2.32 (s, 3H, CH<sub>3</sub>), 2.39 (s, 3H, CH<sub>3</sub>), 2.47 (s, 3H, CH<sub>3</sub>), 7.06–7.10 (t, 1H, Ar-H), 7.23 (d, 1H, Ar-H), 7.31 (br.s, 1H, NH), 7.37 (d, 1H, Ar-H), 7.88 (br.

s, 1H, NH) ppm; IR (KBr):  $\nu$  3414, 3229 (br. 2NH), 1697 (C=O), 1609 (C=N)  $\text{cm}^{-1}$ ; MS m/z (%): 433 ( $\text{M}^+ + 1$ , 2), 432 ( $\text{M}^+$ , 7), 278 (26), 256 (9), 239 (38), 225 (58), 186 (20), 154 (20), 141 (47), 124 (79), 110 (82), 83 (52), 71 (100). Anal. Calcd for  $\text{C}_{18}\text{H}_{20}\text{N}_6\text{O}_3$  (432.59): C, 49.98; H, 4.66; N, 19.43. Found: C, 49.88; H, 4.51; N, 19.35%.

**Compound 9.** Dark green solid, 75% yield, m.p. 140–142 °C (Dioxane);  $^1\text{H-NMR}$  (DMSO- $d_6$ ):  $\delta$  1.24 (t, 3H,  $\text{CH}_3\text{CH}_2$ ), 2.08 (s, 3H,  $\text{CH}_3$ ), 2.28 (s, 3H,  $\text{CH}_3$ ), 2.32 (s, 3H,  $\text{CH}_3$ ), 2.35 (s, 3H,  $\text{CH}_3$ ), 4.18 (q, 2H,  $\text{CH}_2\text{CH}_3$ ), 7.08–7.51 (m, 3H, Ar-H), 8.82 (br.s, 1H, NH), 9.54 (br.s, 1H, NH) ppm;  $^{13}\text{C-NMR}$  (DMSO- $d_6$ ):  $\delta$  13.7, 16.3, 16.8, 18.3, 23.8 ( $\text{CH}_3$ ), 60.6 ( $\text{CH}_2$ ), 114.5, 116.6, 116.7, 127.9, 128.0, 128.2, 128.3, 128.6, 142.7, 142.9, 149.1, 150.2 (Ar-C and C=N), 168.9 (C=O) ppm; IR (KBr):  $\nu$  3427, 3279 (br. 2NH), 1697 (C=O), 1604 (C=N)  $\text{cm}^{-1}$ ; MS m/z (%): 463 ( $\text{M}^+ + 1$ , 2), 462 ( $\text{M}^+$ , 5), 420 (55), 309 (19), 289 (47), 278 (45), 195 (13), 141 (70), 124 (88), 110 (100), 71 (95). Anal. Calcd for  $\text{C}_{19}\text{H}_{22}\text{N}_6\text{O}_2\text{S}_3$  (462.61): C, 49.33; H, 4.79; N, 18.17. Found: C, 49.13; H, 4.99; N, 18.27%.

**Compound 11a.** Red solid, 71% yield, m.p. 260–262 °C (Dioxane);  $^1\text{H-NMR}$  (DMSO- $d_6$ ):  $\delta$  2.27 (s, 3H,  $\text{CH}_3$ ), 2.35 (s, 3H,  $\text{CH}_3$ ), 2.42 (s, 3H,  $\text{CH}_3$ ), 7.07–8.15 (m, 8H, Ar-H and thiazole-H5), 8.41 (br.s, 1H, NH), 9.87 (br.s, 1H, NH) ppm; IR (KBr):  $\nu$  3394, 3238 (br. 2NH), 1604 (C=N)  $\text{cm}^{-1}$ ; MS m/z (%): 488 ( $\text{M}^+ + 1$ , 1), 487 ( $\text{M}^+$ , 11), 431 (31), 414 (53), 386 (16), 333 (58), 289 (36), 235 (57), 210 (36), 168 (29), 139 (66), 110 (100), 71 (41). Anal. Calcd for  $\text{C}_{21}\text{H}_{19}\text{ClN}_6\text{S}_3$  (487.06): C, 51.78; H, 3.93; N, 17.25. Found: C, 51.65; H, 3.82; N, 17.15%.

**Compound 11b.** Dark brown solid, 75% yield, m.p. 180–182 °C (Dioxane);  $^1\text{H-NMR}$  (DMSO- $d_6$ ):  $\delta$  2.11 (s, 3H,  $\text{CH}_3$ ), 2.28 (s, 3H,  $\text{CH}_3$ ), 2.35 (s, 3H,  $\text{CH}_3$ ), 7.06–8.36 (m, 8H, Ar-H and thiazole-H5), 8.81 (br.s, 1H, NH), 9.87 (br.s, 1H, NH) ppm;  $^{13}\text{C-NMR}$  (DMSO- $d_6$ ):  $\delta$  16.2, 19.2, 21.9 ( $\text{CH}_3$ ), 116.5, 118.9, 119.2, 124.6, 125.4, 127.4, 128.3, 128.8, 129.8, 132.3, 144.0, 144.6, 150.6, 152.8, 159.6, 159.8 (Ar-C and C=N) ppm; IR (KBr):  $\nu$  3406, 3236 (br. 2NH), 1598 (C=N)  $\text{cm}^{-1}$ ; MS m/z (%): 498 ( $\text{M}^+ + 1$ , 4), 497 ( $\text{M}^+$ , 6), 383 (53), 344 (61), 329 (42), 289 (34), 247 (43), 234 (46), 221 (18), 150 (13), 141 (11), 124 (52), 97 (30), 71 (31), 58 (100). Anal. Calcd for  $\text{C}_{21}\text{H}_{19}\text{N}_7\text{O}_2\text{S}_3$  (497.62): C, 50.69; H, 3.85; N, 19.70. Found: C, 50.58; H, 3.78; N, 19.66%.

## Synthesis of Thiazole-4-one Derivative 13

A mixture of **3** (0.352 g, 1 mmol) and ethylchloroacetate **12** (1 mmol) in AcOH (20 mL) containing anhydrous

sodium acetate (1 mmol) was refluxed for 2–4 h. (monitored by TLC, using n-Hexane/ethyl acetate (2:1) as elution solvent). The solvent was evaporated under vacuum pressure and the formed solid was crystallized from AcOH to give **13** as Brown solid, 80% yield, m.p. 220–222 °C (Dioxane);  $^1\text{H-NMR}$  (DMSO- $d_6$ ):  $\delta$  2.10 (s, 3H,  $\text{CH}_3$ ), 2.16 (s, 3H,  $\text{CH}_3$ ), 2.25 (s, 3H,  $\text{CH}_3$ ), 4.27 (s, 2H, thiazole- $\text{CH}_2$ ), 6.99–7.08 (t, 1H, Ar-H), 7.51 (d, 2H, Ar-H), 10.53 (br.s, 1H, NH), 11.74 (br.s, 1H, NH) ppm; IR (KBr):  $\nu$  3448, 3261 (br. 2NH), 1697 (C=O), 1612 (C=N)  $\text{cm}^{-1}$ ; MS m/z (%): 393 ( $\text{M}^+ + 1$ , 3), 392 ( $\text{M}^+$ , 13), 380 (40), 309 (33), 293 (22), 248 (57), 183 (16), 166 (16), 156 (24), 124 (100), 110 (98), 97 (54), 71 (73). Anal. Calcd for  $\text{C}_{15}\text{H}_{16}\text{N}_6\text{O}_3$  (392.52): C, 45.90; H, 4.11; N, 21.41. Found: C, 45.88; H, 4.21; N, 21.33%.

## Synthesis of Thiazole Derivatives 14, 15 and 16a, b

General procedure. A mixture of thiocarbohydrazone derivative **5** (0.367 g, 1 mmol) and the appropriate  $\alpha$ -halocarbonyl compound **6**, **8** and **10a, b** (1 mmol) in EtOH (20 mL) was refluxed for 2–4 h. (monitored by TLC, using n-Hexane/ethyl acetate (2:1) as elution solvent). The solvent was evaporated under vacuum pressure and the formed solid was crystallized from appropriate solvent to give **14**, **15** and **16a, b** respectively.

**Compound 14.** Dark brown solid, 85% yield, m.p. 180–182 °C; (Dioxane);  $^1\text{H-NMR}$  (DMSO- $d_6$ ):  $\delta$  2.12 (s, 3H,  $\text{CH}_3$ ), 2.26 (s, 3H,  $\text{CH}_3$ ), 2.31 (s, 3H,  $\text{CH}_3$ ), 2.38 (s, 3H,  $\text{CH}_3$ ), 2.42 (s, 3H,  $\text{CH}_3$ ), 3.36 (br.s, 2H,  $\text{NH}_2$ ), 7.07 (t, 1H, Ar-H), 7.36–7.53 (d, 2H, Ar-H), 11.73 (br.s, 1H, NH) ppm; IR (KBr):  $\nu$  3464, 3387, 3209 ( $\text{NH}_2$  and NH), 1695 (C=O), 1603 (C=N)  $\text{cm}^{-1}$ ; MS m/z (%): 449 ( $\text{M}^+ + 2$ , 8), 447 ( $\text{M}^+$ , 5), 432 (58), 416 (4), 401 (3), 390 (2), 370 (2), 358 (5), 330 (2), 320 (3), 313 (3), 309 (3), 302 (7), 294 (4), 287 (6), 278 (25), 267 (13), 196 (8), 166 (14), 141 (28), 124 (100), 110 (85), 97 (46), 67 (37). Anal. Calcd for  $\text{C}_{18}\text{H}_{21}\text{N}_7\text{O}_3$  (447.60): C, 48.30; H, 4.73; N, 21.90. Found: C, 48.19; H, 4.60; N, 21.74%.

**Compound 15.** Dark brown solid, 70% yield, m.p. 175–177 °C (Dioxane);  $^1\text{H-NMR}$  (DMSO- $d_6$ ):  $\delta$  1.27 (t, 3H,  $\text{CH}_3\text{CH}_2$ ), 2.10 (s, 3H,  $\text{CH}_3$ ), 2.25 (s, 3H,  $\text{CH}_3$ ), 2.32 (s, 3H,  $\text{CH}_3$ ), 2.39 (s, 3H,  $\text{CH}_3$ ), 3.51 (br.s, 2H,  $\text{NH}_2$ ), 4.23 (q, 2H,  $\text{CH}_2\text{CH}_3$ ), 7.06 (t, 1H, Ar-H), 7.38–7.54 (d, 2H, Ar-H), 9.88 (br.s, 1H, NH) ppm; IR (KBr):  $\nu$  3464, 3326, 3176 ( $\text{NH}_2$  and NH), 1697 (C=O), 1607 (C=N)  $\text{cm}^{-1}$ ; MS m/z (%): 477 ( $\text{M}^+$ , 4), 462 (18), 444 (1), 426 (1),

388 (12), 342 (7), 322 (2), 289 (6), 278 (25), 267 (4), 237 (9), 182 (11), 166 (13), 141 (14), 124 (99), 110 (100), 71 (11). Anal. Calcd for  $C_{19}H_{23}N_7O_2S_3$  (477.63): C, 47.78; H, 4.85; N, 20.53. Found: C, 47.55; H, 4.70; N, 20.42%.

**Compound 16a.** Dark brown solid, 77% yield, m.p. 142–144 °C (Dioxane);  $^1H$ -NMR (DMSO- $d_6$ ):  $\delta$  2.09 (s, 3H,  $CH_3$ ), 2.24 (s, 3H,  $CH_3$ ), 2.41 (s, 3H,  $CH_3$ ), 3.47 (br.s, 2H,  $NH_2$ ), 7.01–8.06 (m, 8H, Ar-H and thiazole-H5), 11.50 (br.s, 1H, NH) ppm; IR (KBr):  $\nu$  3387, 3325, 3186 ( $NH_2$  and NH), 1601 (C=N)  $cm^{-1}$ ; MS  $m/z$  (%): 502 ( $M^+$ , 5), 469 (22), 488 (13), 485 (8), 472 (2), 470 (4), 431 (2), 400 (7), 363 (4), 332 (7), 303 (14), 278 (15), 236 (11), 209 (11), 168 (11), 139 (62), 124 (100), 97 (43). Anal. Calcd for  $C_{21}H_{20}ClN_7S_3$  (502.08): C, 50.24; H, 4.02; N, 19.53. Found: C, 50.37; H, 3.85; N, 19.41%.

**Compound 16b.** Dark brown solid, 80% yield, m.p. 180–182 °C (Dioxane);  $^1H$ -NMR (DMSO- $d_6$ ):  $\delta$  2.07 (s, 3H,  $CH_3$ ), 2.27 (s, 3H,  $CH_3$ ), 2.38 (s, 3H,  $CH_3$ ), 3.57 (br.s, 2H,  $NH_2$ ), 7.08–8.43 (m, 8H, Ar-H and thiazole-H5), 11.52 (br.s, 1H, NH) ppm; IR (KBr  $\nu$  3448, 3379, 3217 ( $NH_2$  and NH), 1597 (C=N)  $cm^{-1}$ ; MS  $m/z$  (%): 513 ( $M^+$  + 1, 1), 512 ( $M^+$ , 3), 434 (2), 372 (2), 336 (2), 316 (2), 297 (3), 274 (3), 267 (2), 253 (4), 243 (3), 224 (3), 197 (4), 181 (5), 120 (31), 97 (43), 60 (68), 55 (100). Anal. Calcd for  $C_{21}H_{20}N_8O_2S_3$  (512.63): C, 49.20; H, 3.93; N, 21.86. Found: C, 49.05; H, 3.73; N, 21.71%.

## Synthesis of Thiazole-4-one Derivative 17

A mixture of thiosemicarbazone derivative **5** (0.367 g, 1 mmol) and ethylchloroacetate **12** (1 mmol) in AcOH (20 mL) containing anhydrous sodium acetate (1 mmol) was refluxed for 2–4 h. (monitored by TLC, using n-Hexane/ethyl acetate (2:1) as elution solvent). The solvent was evaporated under vacuum pressure and the formed solid was crystallized from AcOH to give **17** as green solid, 77% yield, m.p. 190–192 °C (Dioxane);  $^1H$ -NMR (DMSO- $d_6$ ):  $\delta$  2.16 (s, 3H,  $CH_3$ ), 2.33 (s, 3H,  $CH_3$ ), 2.69 (s, 3H,  $CH_3$ ), 3.39 (br.s, 2H,  $NH_2$ ), 4.12 (s, 2H,  $CH_2$ ), 7.09–7.54 (m, 3H, Ar-H), 11.89 (br.s, 1H, NH) ppm; IR (KBr):  $\nu$  3436, 3318, 3212 ( $NH_2$  and NH), 1705 (C=O), 1607 (C=N)  $cm^{-1}$ ; MS  $m/z$  (%): 408 ( $M^+$  + 1, 3), 407 ( $M^+$ , 6), 392 (6), 382 (2), 367 (2), 350 (3), 341 (2), 325 (2), 313 (5), 303 (10), 293 (10), 278 (11), 263 (4), 253 (5), 239 (8), 226 (8), 220 (6), 185 (11), 166 (11), 149 (7), 141 (11), 124 (63), 110 (44), 84 (39), 57 (100). Anal. Calcd for  $C_{15}H_{17}N_7OS_3$  (407.54): C, 44.21; H, 4.20; N, 24.06. Found: C, 44.30; H, 4.17; N, 24.01%.

## Anticancer Activity

The synthesized compounds have been cytotoxicity assessed against MCF-7 cells with 24 hours incubation MTT examination at the Regional Center for Mycology and Biotechnology at Al-Azhar University, Cairo, Egypt.

## In vitro Cytotoxic Activity

The two cell cultures for human breast cancer (MCF-7) cell lines were bought from the American Type Culture Collection (Rockville, MD) and preserved in DMEM medium, supplemented by 10%. (Fetal bovine serum), 100U/mL penicillin and 100 U/mL streptomycin. Cells grew at 37°C in a humid atmosphere of 5%  $CO_2$ .

## MTT Cytotoxicity Assay

3-(4,5-Dimethyl-2-thiazolyl)-2,5-diphenyl-2H-tetrazolium bromide (MTT) assay was utilized to evaluate the cytotoxicity of imidazothiazole derivatives versus MCF-7 human cancer cell lines. This method is depending on the dissent the salt of tetrazole by mitochondrial dehydrogenases in the cells. Distributed of the cells in a ( $5 \times 10^4$  cells/well) of 96 well sterile microplate, and at 37°C were incubated in DMSO with series of different concentrations of each tested imidazothiazole derivatives or Doxorubicin (positive control) for 48 h in a serum-free medium prior to the MTT assay. After incubation, the media was carefully removed and (2.5 mg/mL) of 40  $\mu$ L of MTT was added to each well and incubated for an additional 4 hours. Purple formazan-dye crystals were dissolved by adding 200  $\mu$ L of DMSO. At 590 nm, the absorbance was measured utilizing Spectra-Max Paradigm Multi-Mode plate reader. The relative viability of the cell was expressed as a percentage of viable cells compared to untreated control cells. All trials were conducted in three versions and repeated on three different days. All values were represented as  $\pm$  SD. IC50s were determined by probit analysis by SPSP Inc. Analysis (USA, NY, IBM Corp., Armonk).<sup>48</sup>

## In silico Studies

### 3D Structure Generation, Protein Preparation, Active Site Identification

The 3D structure of the target is essential for discovering novel inhibitors against cancer via computer-based docking approach.<sup>49</sup> In fact, to date, the 3D model has not been generated yet, so the homology modeling approach is used to determine the 3D structure of the target Rab7b. The stereochemical quality of the generated model is checked by Ramachandran plot.<sup>50</sup> The active site pockets of the

target were evaluated using Computed Atlas of Surface Topography of proteins CASTp<sup>51</sup> web server.

### Ligand Preparation

The 3D structures of the ligand molecules are generated using ChemDraw Ultra 7.0 and saved as SDF files format by using Open Babel 2.4.1 tool.<sup>52</sup> In-house library of thirteen compounds is generated for study. Energy of the compounds was minimized using Universal Force Field (UFF),<sup>53</sup> to obtain stable confirmations. The energy minimized compounds are then read as input for PyRx virtual screening tool,<sup>54</sup> in order to perform the docking simulation.

### In silico Docking Protocol

To understand ligand–protein interaction, a molecular docking study is performed for of thirteen compounds 1–17 against Rab7b protein using PyRx tool. During the screening process, a maximum of nine conformers is considered for each compound, and then, the conformer with more negative binding energy is elected for further study.<sup>55,56</sup> Two and three-dimensional representations of ligand–protein interactions are visualized using Accelrys discovery studio 3.5 (Accelrys Discovery Studio Visualizer Software 2010).

### ADME Screening

The drug-likeness and physicochemical properties<sup>57,58</sup> of the newly synthesized compounds are predicted using web-based softwares, admetSAR<sup>59</sup> and Mol inspiration.

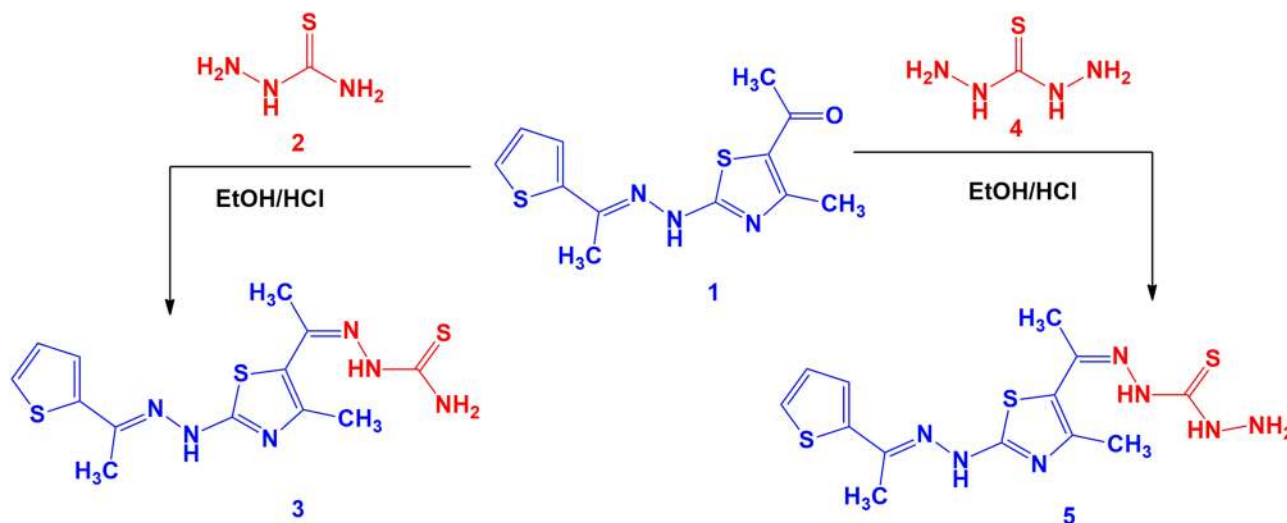
## Results and Discussion

### Chemistry

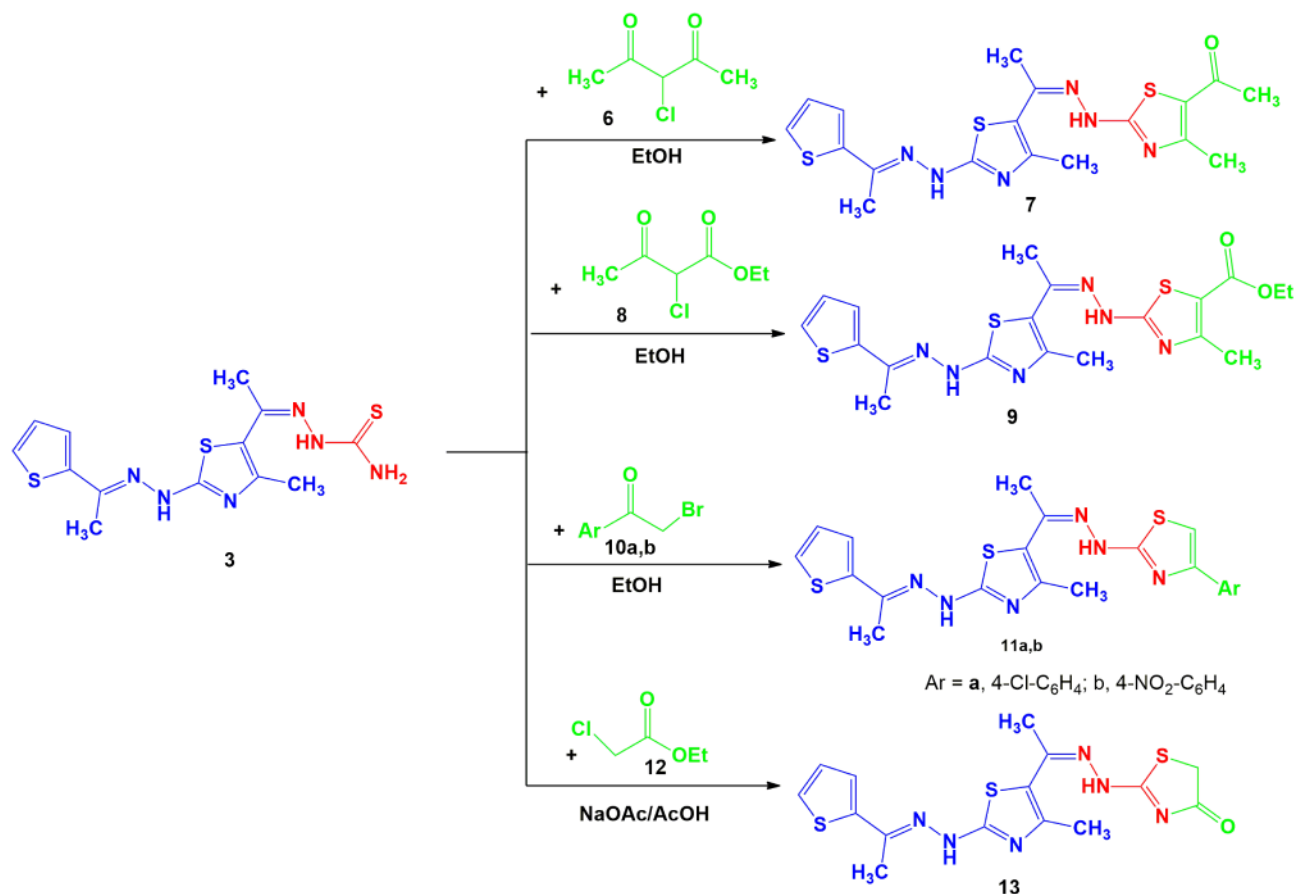
Thiosemicarbazone derivative **3** and thiocarbohydrazone derivative **5** were prepared from reaction of 2-(2-benzylidenehydrazinyl)-4-methylthiazole (**1**)<sup>60</sup> with the respective thiosemicarbazide **2** and thiocarbohydrazone **4** in EtOH/HCl under reflux, respectively (Scheme 1). The chemical structure of the compounds **3** and **5** was elucidated by both spectral data and elemental analysis.

The chemical reactivity of thiosemicarbazone **3** towards  $\alpha$ -halo-compounds was investigated with the aim of synthesizing a series of new thiazole systems. Thus, treatment of thiosemicarbazone derivative **3** with 3-chloropentane-2,4-dione (**6**) in refluxing EtOH yielded the acetylthiazole derivative **7** (Scheme 2). The spectra and elemental analysis of compound **7** were in accordance with suggested structure. The <sup>1</sup>H-NMR spectra of **7** showed the expected signals at  $\delta = 2.07, 2.24, 2.32, 2.39, 2.47$  (5s, 15H, 5CH<sub>3</sub>), 7.06–7.10 (t, 1H, Ar-H), 7.23 (d, 1H, Ar-H), 7.31 (br.s, 1H, NH), 7.37 (d, 1H, Ar-H), 7.88 (br.s, 1H, NH). The mass of compound **7** was determined by mass spectrometry is equal to the calculated value.

In a similar way, thiosemicarbazone **3** reacted with ethyl 2-chloro-3-oxobutanoate **8** in refluxing EtOH to afford the respective thiazole ester **9** (Scheme 2). The <sup>1</sup>H-NMR spectrum exhibited two broad singlet signals of two NH protons at  $\delta 8.82$  and 9.54 ppm, 2signals at  $\delta =$



**Scheme 1** Synthesis of thiosemicarbazone **3** and thiocarbohydrazone **5**.



**Scheme 2** Synthesis of thiazole derivatives **7**, **9**, **11a, b** and **13**.

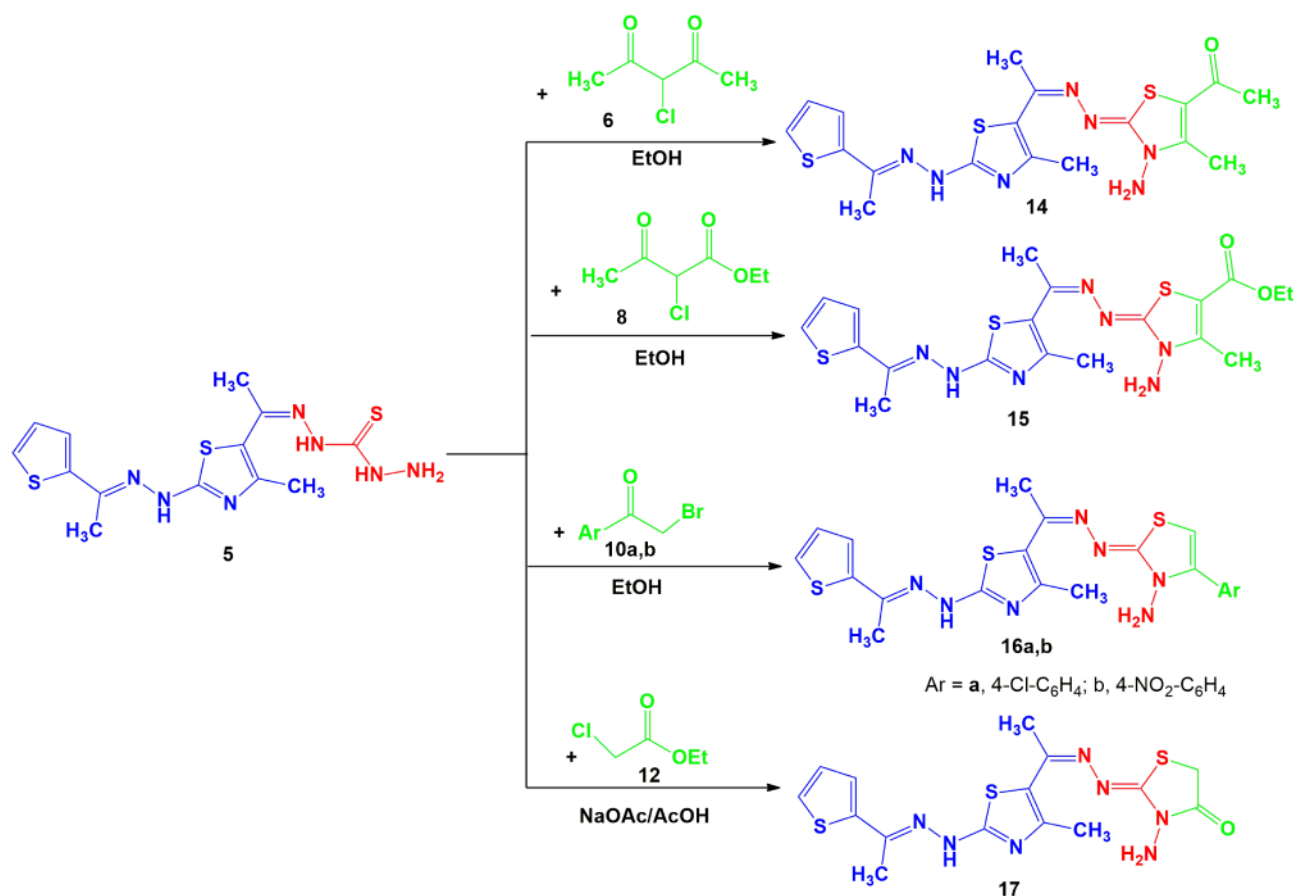
1.24 (CH<sub>3</sub>, t) and 4.18 (CH<sub>2</sub>, q) ppm corresponding to ethyl group, and five singlet signals assignable for 5 CH<sub>3</sub> groups at  $\delta = 2.08, 2.25, 2.28, 2.32$  and  $2.35$  ppm, in addition to the three aromatic protons.

Also by analogy, when compound **3** was reacted with p-substituted phenacyl bromide derivatives **10a, b**, it afforded the products **11a, b** as inferred from their spectral data and elemental analysis (Scheme 2). The <sup>1</sup>H-NMR spectra of compound **11a** showed the expected three singlet signals for the 3CH<sub>3</sub> at  $\delta 2.27, 2.35, 2.42$  ppm, multiplet signal at  $\delta 7.07\text{--}8.15$  ppm (8H), and also two broad singlet signals at  $\delta 8.41$  and  $9.87$  ppm due to 2NH groups. Its IR spectra revealed two NH absorption bands at  $\nu = 3394$  and  $3238$  cm<sup>-1</sup>. Moreover, the mass of compounds **11a, b** determined by mass spectrometry is equal to the calculated values.

In addition, the reaction of compound **3** with ethyl 2-chloroacetate **12** was also studied aiming to prepare new bioactive thiazolone derivative. Thus, refluxing compound **3** with **12** in AcOH\AcONa yielded thiazolone **13** based on spectral data (<sup>1</sup>H-NMR, IR and mass)

(Scheme 2). The mass spectrum of compound **13** exhibited peak at  $m/z$  392. Its <sup>1</sup>H-NMR spectrum showed the existence of a singlet signal at  $\delta = 4.27$  ppm referred to the CH<sub>2</sub> protons of thiazolone ring, in addition to the signals of 3 methyl groups, 3 thiophene protons and the two NH protons (see Experimental).

Next, our study was extended to prepare bioactive N-aminothiazole derivatives from reaction of thiocarbohydrazone **5** with  $\alpha$ -halo-compounds, thus when thiocarbohydrazone derivative **5** was allowed to react with 3-chloropentane-2,4-dione **6** and ethyl 2-chloro-3-oxobutanoate **8** in refluxing EtOH afford the respective N-aminothiazole derivatives **14** and **15** as depicted in Scheme 3. The chemical structures of derivatives **14** and **15** were elucidated by IR, <sup>1</sup>H-NMR and MS. For example, the IR spectrum of derivative **15** showed the stretching bands at  $\nu = 3464, 3326, 3176$  cm<sup>-1</sup> which are attributed to the NH and NH<sub>2</sub> groups, in addition to another band for the conjugated ester carbonyl group at  $\nu = 1697$  cm<sup>-1</sup>. The mass spectrum of derivatives **14** and **15** showed a molecular ion peaks at  $m/z = 447, 477$  which are



**Scheme 3** Synthesis of thiazoles **14**, **15**, **16a, b** and **17**.

compatible with C<sub>19</sub>H<sub>23</sub>N<sub>7</sub>O<sub>2</sub>S<sub>3</sub> and C<sub>18</sub>H<sub>21</sub>N<sub>7</sub>O<sub>3</sub>S<sub>3</sub>, respectively. <sup>1</sup>H-NMR spectrum of compound **15** showed two signals at δ = 3.51 and 9.88 ppm attributed to the NH<sub>2</sub> and NH protons, in addition to the expected signals of the aromatic protons, ester group and 4 methyl groups (see Experimental).

Analogously, the thiosemicarbazone **5** was allowed to react with phenacyl bromides **10a, b** to yield N-aminothiazole derivatives **16a, b** as final product (Scheme 3). The chemical structures of the derivatives **16a, b** were confirmed from the spectral data and elemental analyses. The <sup>1</sup>H-NMR spectra of compound **16a** showed the expected three singlet signals for the 3CH<sub>3</sub> at δ 2.09, 2.24, 2.41 ppm, multiplet signal at δ 7.01–8.06 ppm (8H), and also two broad singlet signals at δ 3.47 and 11.50 ppm due to NH<sub>2</sub> and NH groups, respectively. Moreover, the molecular weight determination of the products **16a, b** was observed in the expected region.

Finally, compound **5** was reacted with ethyl 2-chloroacetate **12** in AcOH/AcONa under reflux to afford N-aminothiazolone derivative **17** as presented in

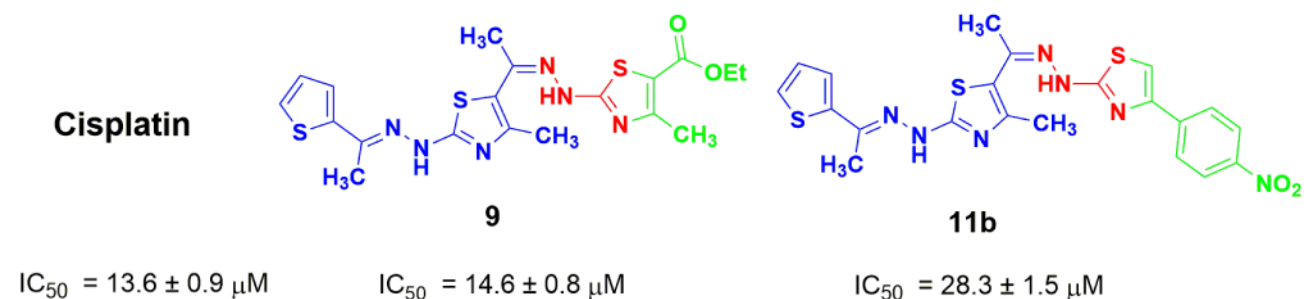
Scheme 3. The <sup>1</sup>H-NMR spectra of compound **17** showed seven signals at δ 2.16, 2.33, 2.69 (3s, 3CH<sub>3</sub>), 3.39 (br.s, NH<sub>2</sub>), 4.12 (s, CH<sub>2</sub>), 7.09–7.54 (m, 3Ar-H), 11.89 (br.s, NH) ppm.

## Anticancer Activity

As a continuation of our studies on 1,3-thiazole and thiophene derivatives as anticancer agents<sup>61–71</sup> and in a search for new anticancer drugs, we herein report the synthesis of a series of new thiazolyl-hydrazono-ethylthiazoles incorporating thiophene moiety and evaluation for their anticancer activities. Thiosemicarbazones (TSCs) have a wide clinical antitumor spectrum with efficacy in various tumor types such as leukemia, breast cancer, pancreatic cancer, non-small cell lung cancer, cervical cancer, prostate cancer and bladder cancer. Several possible mechanisms have been implemented for the anticancer activity of thiosemicarbazones.<sup>72</sup>

Recently, some studies have reported that Rab7b over-expression accelerated the proliferation and growth of human breast cancer (MCF-7) cells.<sup>46</sup> Rab7b protein is





**Figure 2** The most active compounds towards the MCF-7 cell line.

selected as promising target to identify potentially inhibitors as breast cancer therapeutics. Thus, the pharmacological activities of the synthesized thiosemicarbazone **3**, thiocarbohydrazone **5**, and thiazoles **7**, **9**, **11a**, **b**, **13**, **14**, **15**, **16a**, **b**, and **17** were investigated for their MCF-7 using colorimetric MTT assay and reference drug as Cisplatin. The data have been used to plot an exposure response curve in which the concentrations needed to kill half of the cell population ( $IC_{50}$ ) of the tested samples were calculated. Cytotoxic activities were expressed as the mean  $IC_{50}$  of three independent experiments (see [Figure S3a](#) and [S3b](#) in the [supplementary data](#) section). The results are represented in [Figure 2](#) and [Table 1](#).

The results showed that all derivatives displayed concentration-dependent inhibitory effect against tumor cells. Compounds **9** and **11b** have promising cytotoxic activity ( $IC_{50} < 30 \mu M$  which is close to Cisplatin reference drug), and compounds **3**, **7**, **11a** and **16b** exhibited medium activity ( $IC_{50} \sim 30\text{--}80 \mu M$ ) while the rest compounds **5**, **13**, **14**, **15**, **16a** and **17** are inactive ( $IC_{50} > 80 \mu M$ ).

Among the thiazole derivatives, thiazole **9** and **11b** ( $IC_{50} = 14.6 \pm 0.8$  and  $28.3 \pm 1.5 \mu M$ , respectively) showed promising cytotoxic activity close to Cisplatin

( $IC_{50} = 13.6 \pm 0.9 \mu M$ ) towards the MCF-7 cell line. The cytotoxic activity of thiosemicarbazone derivative **3** ( $IC_{50} = 76.4 \pm 0.52 \mu M$ )  $\gg$  thiocarbohydrazone derivative **5** ( $IC_{50} = 292.1 \pm 6.3 \mu M$ ) due to the presence of amino group (electron-donating group). Generally, 1,3-thiazole derivatives have more cytotoxic activity than the respective N-amino-1,3-thiazole derivatives (**7**, **9**, **11a**, **11b**  $>$  **14**, **15**, **16a**, **16b**) due to the presence of amino group. Finally, compound **11b** has more cytotoxic activity than **11a** and also N-amino-1,3-thiazole derivative **16b** has more cytotoxic activity than **16a** may due to the presence of nitro group (electron-withdrawing group) at position 4 of the phenyl ring.

## Molecular Docking Studies

In silico modeling, study<sup>73</sup> is essential to recognize the mechanism of actions of the synthetic compound against Ra7b protein. The FASTA sequence of the target (accession no: Q96AH8) was obtained from NCBI. The reference structure used for protein structure prediction is the crystal structure of the REP-1 protein in complex with monophenylated Rab7b protein (PDB: 1VG0), which was obtained by using BLASTp (Basic Local Alignment Search Tool) webserver. The homology model of the target protein Rab7b was generated using Modeller 9.11 software. Twenty-five protein models were generated, and the best model was selected for further validation. The 3D protein structure was prepared for docking process via removing of water molecules, addition and elimination of polar hydrogen atoms. The homologue model of the target (see [Figure S4](#), in the [supplementary data](#) section), consists of 8  $\alpha$ -helices and 6  $\beta$ -strands obtaining from PDBsum server<sup>74</sup> as represented in [Figure S5](#), in the [supplementary data](#) section.

The physicochemical properties of the target protein were calculated using ProtParam tool.<sup>75</sup> The protein sequence has 199 amino acid residues with molecular

**Table 1** In vitro Cytotoxic Activity of the Newly Synthesized Compounds **3**, **5**, **7**, **9**, **11a**, **b**, **13**, **14**, **15**, **16a**, **b**, and **17** Against MCF-7

Tested Compounds	$IC_{50}$ ( $\mu M$ )	Tested Compounds	$IC_{50}$ ( $\mu M$ )
<b>3</b>	$76.4 \pm 3.4$	<b>14</b>	$99.2 \pm 3.6$
<b>5</b>	$292.1 \pm 6.3$	<b>15</b>	$123.4 \pm 3.9$
<b>7</b>	$65.3 \pm 2.9$	<b>16a</b>	$147.8 \pm 3.6$
<b>9</b>	$14.6 \pm 0.8$	<b>16b</b>	$66.9 \pm 2.5$
<b>11a</b>	$60.2 \pm 2.4$	<b>17</b>	$109.5 \pm 3.7$
<b>11b</b>	$28.3 \pm 1.5$	<b>Cisplatin</b>	$13.6 \pm 0.9$
<b>13</b>	$178.5 \pm 4.3$		

weight 22.510 kDa. The most abundant amino acid residues are LEU, ILE, SER, VAL, LYS, ASP, GLU, and GLN, respectively, in high percentages in Rab7b, as declared in Figure. Leucine has the highest abundance (10.6%), and Tryptophan has the lowest abundance (1.5%), as represented in [Figure S6](#), in the [supplementary data](#) section. The physicochemical parameters predicted a negatively charged protein as the result of the high number of negatively charged residues (aspartic acid 6.5% and glutamic acid 6%) in contrast with positively charged ones (Arginine 5% and lysine 7%). Further, the molecular formula of Rab7b is  $C_{1007}H_{1607}N_{267}O_{298}S_9$ . The atomic composition of the target protein is 3188, with 1007 carbon (C), 1607 hydrogen (H), 267 nitrogen (N), and 9 sulfur (S). In addition, the protein is acidic, with an isoelectric point (pI) of 6.31. The estimated half-life time of Rab7b represented that it can remain intact without being degraded for 30 h in humans, less than 20 h in yeast, and less than 10 h in *E.coli*, and its extinction coefficient is  $26,930 \text{ M}^{-1} \text{ cm}^{-1}$ . Finally, the generated aliphatic index was 98.44, with grand average of hydrophobicity (GRAVY) of  $-0.129$  and instability index was computed to be 36.48, which indicates that the protein is stable.

Once generated, the model was validated using Ramachandran plot, to check its stereochemical quality. RC plot (see [Figure S7](#), in the [supplementary data](#) section) represents 89.4% (160 aa) of the total residues in most favored regions and 10.6% (19 aa) in additional allowed regions, indicating reasonable quality model. The active site prediction tools declared that the amino acid residues Gly18, Gly20, Lys21, Thr22, Ser23, Ala43, Ser44, Asp63, Glu68, and Lys125 are crucial for Rab7b protein. The grid box<sup>76</sup> was then allocated over the predicted binding site region for specific docking, with dimension of  $25 \text{ \AA} \times 25 \text{ \AA} \times 25 \text{ \AA}$ , as shown in [Figure S8](#), in the [supplementary data](#) section. Docking screening was performed for the compounds into active site of the target Rab7b, using PyRx virtual screening tool. Ligand–protein interactions are depicted in [Figure 3](#). [Table 2](#) shows the estimated binding energies which are in the range of  $-6.7$  to  $-5.0$  kcal/mol. The molecule with the lowest binding energy (ie more negative) indicates highest binding affinity to the target protein.<sup>77,78</sup> From the data gotten, the compounds exhibited respectable fitting to the binding pocket of the protein, through a network of non-covalent interactions like hydrogen bonds and  $\pi$ -cation. Compound **1** revealed binding energy of  $-5.4$  kcal/mol and formed two hydrogen

bonds with Thr22 and Ser23 at the distances of 2.78 and 2.98  $\text{\AA}$ , respectively. Compound **3** displayed binding energy of  $-5.2$  kcal/mol besides forming hydrogen bonding interactions Thr22 and Asp63 in distances 2.86 and 2.22  $\text{\AA}$ , respectively. In addition, compound **5** interacted with the target through Asp63, Thr22 and Glu68 at distances of 1.97, 2.20 and 2.37  $\text{\AA}$ , respectively.

Compound **7** was docked to the target through one H-bond with the residue Ser44. Moreover, compound **9** exhibited four H-bond interactions with the amino acid residues Gly18, Gly20, Lys21, and Thr22. The derivatives **11a** ( $-5.8$  kcal/mol) and **11b** ( $-5.1$  kcal/mol) exhibited H-bonds with Thr22 and Ser23, respectively. Introducing of electron-withdrawing groups such as  $-\text{NO}_2$  (strong) and  $-\text{Cl}$  (weak) on phenyl ring causing more cytotoxic activity of the compounds.<sup>79</sup> The compound **13** represented two H-bond interactions with Thr22 at distances of 2.95 and 2.96  $\text{\AA}$ , respectively. Compound **14** was successfully docked to the target protein with maximum binding energy  $-6.7$  kcal/mol, and exhibited two H-bond interactions with Gly20 and Glu68 at 2.66 and 2.31  $\text{\AA}$ , respectively. Compound **15** interacted with the target at the residues Ser23, Ala43 and Lys125, forming three hydrogen bonds and one  $\pi$ -cation interactions at distance of 2.81, 2.98, 3.09 and 3.89  $\text{\AA}$ , respectively. The derivative **16a** ( $-6.3$  kcal/mol) exhibited one H-bond with Thr22 at 2.57  $\text{\AA}$ , while derivative **16b** ( $-6.2$  kcal/mol) showed two H-bonds with the amino acid residues Lys125 and Gly18. Finally, compound **17** showed hydrogen bonding and  $\pi$ -cation interactions with Ser23, Ala43, Asp63, and Lys125 at the distances of 2.95, 2.31, 2.96, 2.39 and 3.64  $\text{\AA}$ , respectively. To further understand the nature of  $\pi$ -cation interactions; lysine Lys125 contains a positively charged amino on its side-chain ( $\text{H}_3\text{N}^+$ ) that is involved in forming  $\pi$ -cation interactions with compounds **15** and **17**. The synthetic thiazole derivatives **1–17** with heterocyclic moieties like thiophene and thiazole are noted to be common pharmacophore groups, which interact with the binding site pocket of the target Rab7b, through non-covalent interactions. The in silico molecular docking study results revealed that all the synthesized compounds having minimum binding energy and good affinity towards the active site pocket, thus, they may be considered as good drug-like small molecules for cancer treatment. On the other hand, the rule of 5 “RO5” methodology is an important way in defining drugability. The results tabulated in [Tables 3](#) and [4](#) indicated that the synthesized compounds fit well with Lipinski rule of five (RO5).<sup>80</sup> The results show that (a)

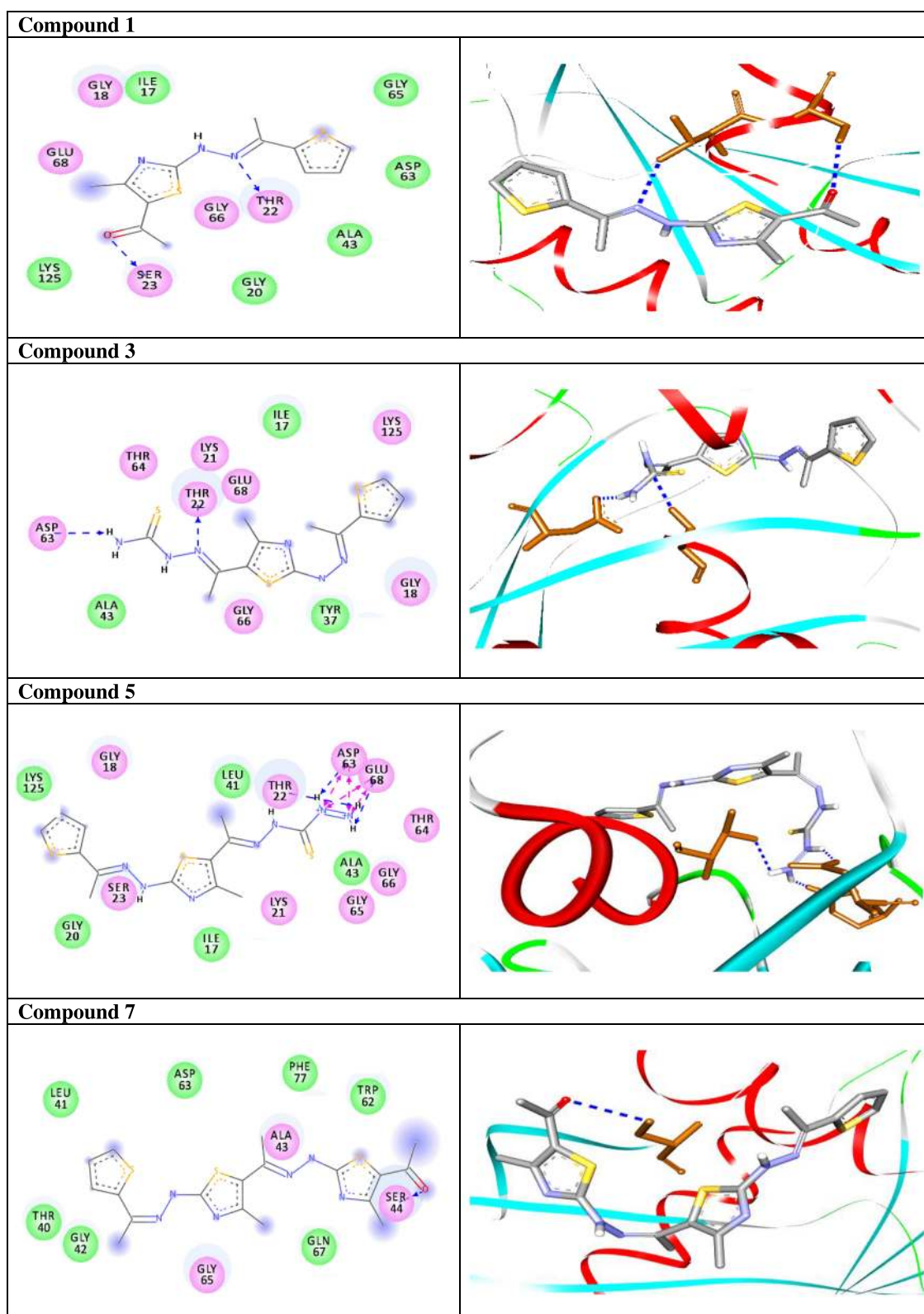
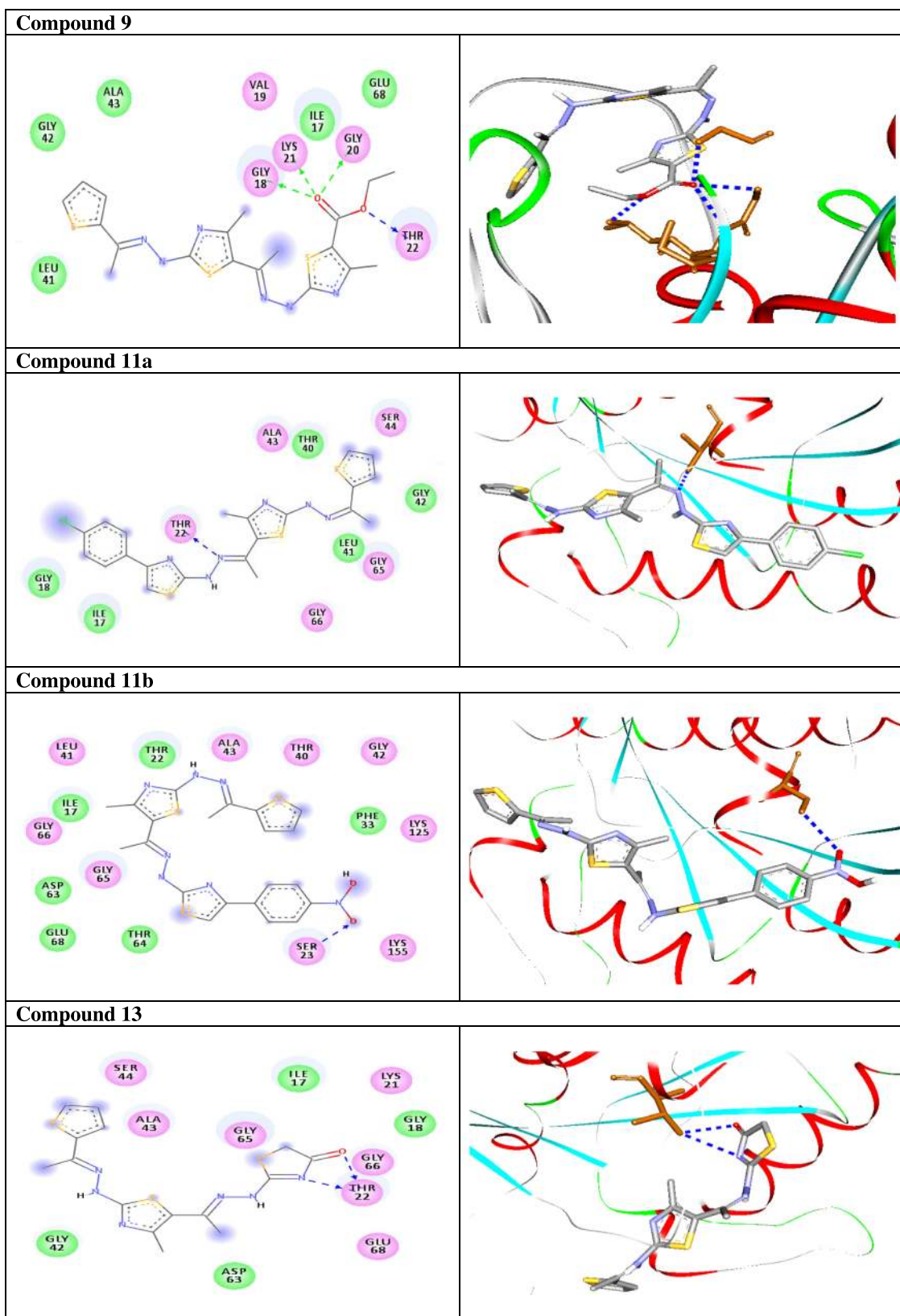
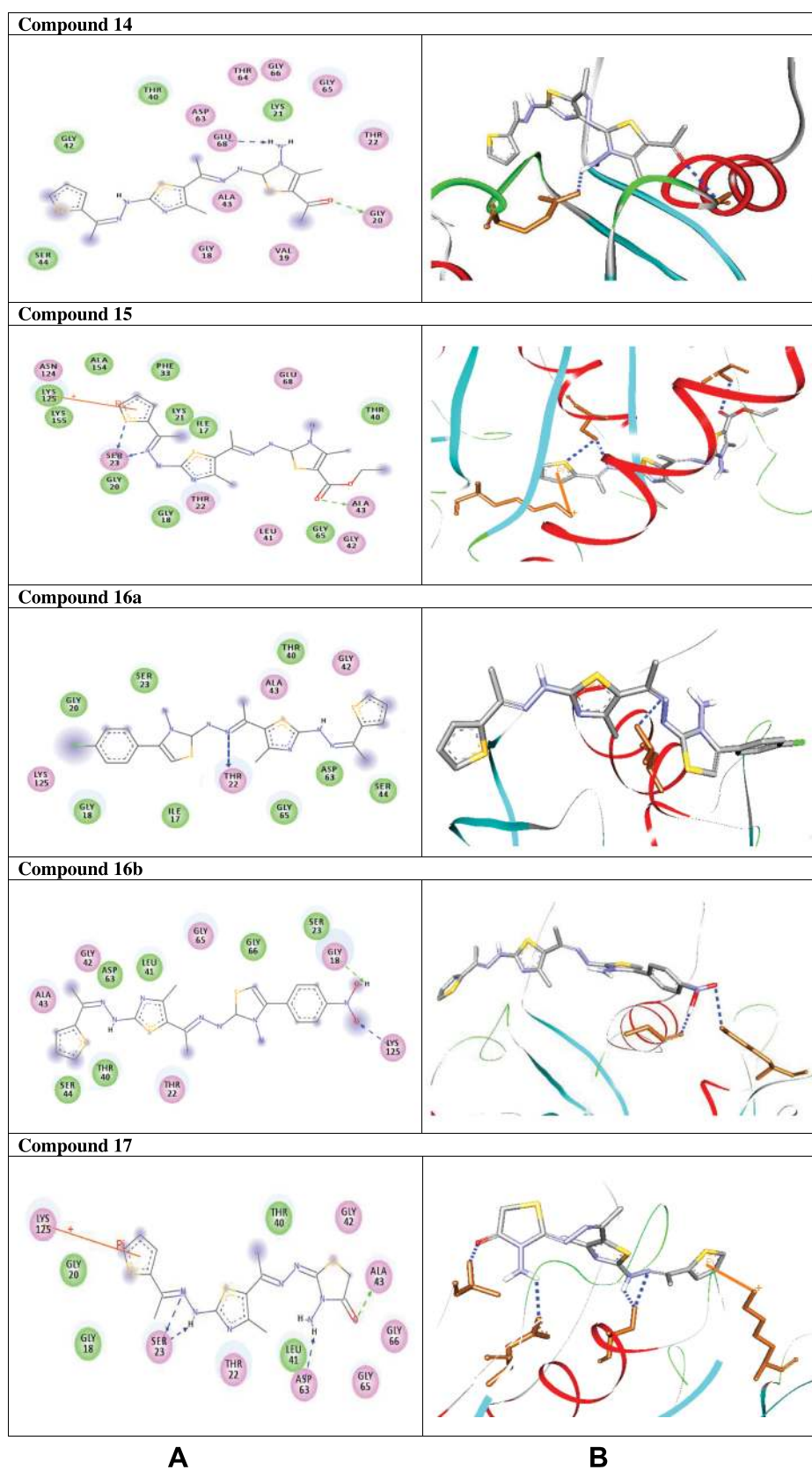


Figure 3a (Continued).



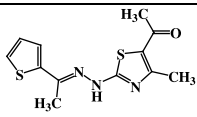
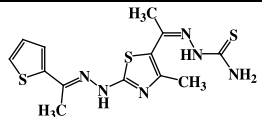
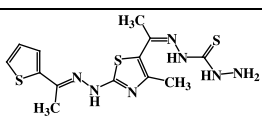
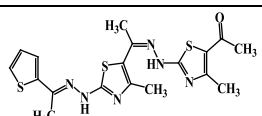
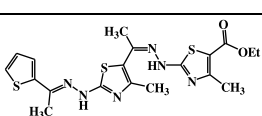
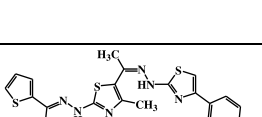
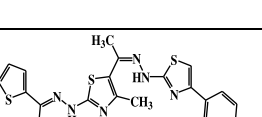
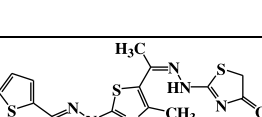
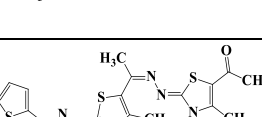
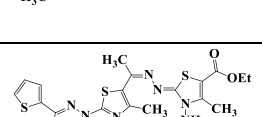


A

B

**Figure 3c** (A) 2D and (B) 3D simulations that show the molecular interactions between compounds I–17 and the Rab7b protein active site region. Hydrogen bonding interactions are represented in green and blue dotted lines, while  $\pi$ -cation interactions are shown in orange lines.

**Table 2** Molecular Docking Results for the Screened Compounds and Rab7b

	<b>2D Structure</b>	<b>Binding Energy kcal/mol</b>	<b>Docked Complex (Amino Acid-Ligand) Interactions</b>	<b>Bond Length (Å°)</b>
<b>1</b>		-5.4	<b>H-bond interactions</b> Thr22: OG1—compound 1  Ser23: OG —compound 1	2.78  2.92
<b>3</b>		-5.2	<b>H-bond interactions</b> Thr22: OG1—compound 3  Asp63: OD1—compound 3	2.86  2.22
<b>5</b>		-6.1	<b>H-bond interactions</b> Asp63: OD1—compound 5 Thr22: OG1—compound 5 Glu68: OE2—compound 5	1.97 2.37 2.20
<b>7</b>		-5.0	<b>H-bond interactions</b>  Ser44: OG—compound 7	  2.86
<b>9</b>		-5.3	<b>H-bond interactions</b> Gly18: N—compound 9 Gly20: N—compound 9 Lys21: N—compound 9 Thr22: OG1—compound 9	2.29 2.80 2.98 2.79
<b>11a</b>		-5.8	<b>H-bond interactions</b>  Thr22: OG1—compound 11a	  2.95
<b>11b</b>		-5.1	<b>H-bond interactions</b>  Ser23: OG—compound 11b	  2.98
<b>13</b>		-6.1	<b>H-bond interactions</b> Thr22: OG1—compound 13  Thr22: OG1—compound 13	2.95  2.96
<b>14</b>		-6.7	<b>H-bond interactions</b> Gly20: N—compound 14  Glu68: OE2—compound 14	2.66  2.31
<b>15</b>		-5.8	<b>H-bond interactions</b> Ser23: OG—compound 15 Ser23: OG—compound 15 Ala43: N—compound 15 <b>π- cation interactions</b> Lys125: NZ—compound 15	2.81 2.98 3.09 3.89

(Continued)

Table 2 (Continued).

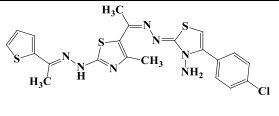
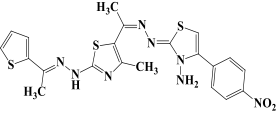
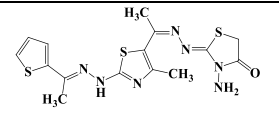
	2D Structure	Binding Energy kcal/mol	Docked Complex (Amino Acid–Ligand) Interactions	Bond Length (Å)
<b>16a</b>		-6.3	<b>H-bond interactions</b> Thr22: OGI–compound <b>16a</b>	2.57
<b>16b</b>		-6.2	<b>H-bond interactions</b> Lys125: NZ–compound <b>16b</b> Gly18: O–compound <b>16b</b>	3.00 2.10
<b>17</b>		-5.8	<b>H-bond interactions</b> Ser23: OG–compound <b>17</b> Ser23: OG–compound <b>17</b> Ala43: N–compound <b>17</b> Asp63: ODI–compound <b>17</b> <b>π-cation interactions</b> Lys125: NZ–compound <b>17</b>	2.95 2.31 2.96 2.39 3.64

Table 3 List of ADME Properties of Synthesized Molecules I–17. The Pharmacokinetic Properties of Compounds are Predicted by Admet SAR Tool

	Molecular Weight (g/mol)	Blood–Brain Barrier (BBB <sup>+</sup> )	Caco-2 Permeability (Caco <sup>2+</sup> )	%Human Intestinal Absorption (HIA <sup>+</sup> )	AMES Toxicity	Carcinogenicity
<b>Reference Range</b>	<b>180–500</b>	<b>–3 to 1.2</b>	<b>&lt; 25 Poor</b> <b>&gt; 500 Great</b>	<b>&lt; 25 Poor</b> <b>&gt; 80 High</b>	<b>Nontoxic</b>	<b>Non-Carcinogenic</b>
<b>1</b>	279.39	0.974	70.98	98.89	Nontoxic	Non carcinogenic
<b>3</b>	352.51	0.966	50.57	98.35	Nontoxic	Non carcinogenic
<b>5</b>	367.53	0.967	55.15	98.35	Nontoxic	Non carcinogenic
<b>7</b>	432.60	0.967	65.15	98.89	Nontoxic	Non carcinogenic
<b>9</b>	462.63	0.975	69.79	98.80	Nontoxic	Non carcinogenic
<b>11a</b>	487.08	0.976	71.38	99.02	Nontoxic	Non carcinogenic
<b>11b</b>	497.63	0.976	80.20	97.44	Nontoxic	Non carcinogenic
<b>13</b>	392.54	0.974	65.13	99.06	Nontoxic	Non carcinogenic
<b>14</b>	447.62	0.975	65.10	97.99	Nontoxic	Non carcinogenic
<b>15</b>	477.64	0.977	73.56	97.66	Nontoxic	Non carcinogenic
<b>16a</b>	502.09	0.977	70.24	98.91	Nontoxic	Non carcinogenic
<b>16b</b>	512.65	0.976	77.70	97.26	Nontoxic	Non carcinogenic
<b>17</b>	407.55	0.977	68.06	99.06	Nontoxic	Non carcinogenic

all newly synthetic thiazole compounds have molecular weights within the limits of 180–500 g/mol, except **16a** and **16b** (b) the compounds have H-bond donating ability <6 (c) the compounds have H-bond accepting ability in the acceptable range (d) the topological polar surface area (TPSA) was found in the acceptable range  $\leq 140$  (e) the logp of the compounds indicates that they are not very lipophilic <5, except **11a**, **11b** and **16a**. Also,

ADMET properties declare that the newly synthetic analogues have better Human Intestinal Absorption (HIA) score, and good Blood-Brain Barrier (BBB) values, which means that they could be better absorbed by the human intestine.<sup>81</sup> In addition, they showed negative toxicity and negative carcinogenicity test. As the newer compounds exhibited the fewest violations of Lipinski rule of five (RO5), our findings suggest that our

**Table 4** Physicochemical Properties of the Title Compounds **1–17**. Logp, Logarithm of Partition Coefficient Between n-Octanol and Water; HBA, Number of HB Acceptors; n Rotatable, Number of Rotatable Bonds; HBD, Number of HB Donors; TPSA, Topological Polar Surface Area

	logp	TPSA A <sup>2</sup>	HBA	HBD	N Rotatable	N Violations	Volume A <sup>3</sup>
Reference Range	< 5	≤ 140	2.0–20.0	0.0–6.0	≤ 10		
<b>1</b>	3.55	54.35	4	1	4	0	237.05
<b>3</b>	2.91	87.70	6	4	6	0	292.88
<b>5</b>	2.41	99.72	7	5	7	0	305.28
<b>7</b>	4.24	91.64	7	2	7	0	363.80
<b>9</b>	4.55	100.87	8	2	9	0	389.58
<b>11a</b>	6.24	74.57	6	2	7	1	396.64
<b>11b</b>	5.52	120.39	9	2	8	1	406.43
<b>13</b>	2.75	91.11	7	2	6	0	320.06
<b>14</b>	4.62	110.04	8	3	6	0	375.47
<b>15</b>	4.93	119.27	9	3	8	0	401.25
<b>16a</b>	4.99	92.97	7	3	6	1	408.31
<b>16b</b>	4.83	138.79	10	3	7	0	418.10
<b>17</b>	2.38	108.34	8	3	5	0	331.73

synthesized compounds **1–17** could be pharmacologically efficient for preclinical use.

## Conclusion

In this context, we herein present an efficient synthesis of a novel series of thiazole linked thiophene conjugates in good yields. The products were screened for their cytotoxic activity against MCF-7 cells and the results obtained showed that derivatives **9** and **11b** have promising activity ( $IC_{50} = 14.6 \pm 0.8$  and  $28.3 \pm 1.5 \mu M$ , respectively) compared to Cisplatin ( $IC_{50} = 13.6 \pm 0.9 \mu M$ ). The molecular docking analysis reveals that the synthesized compounds are predicted to be fit into the binding site of the target Rab7b. In summary, the synthetic thiazole compounds **1–17** could be used as potent inhibitors as anticancer drugs.

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## Author Contributions

All the authors made a significant contribution to the work reported, whether that is in the conception, study design, execution, acquisition of data, analysis and interpretation, or all in these areas; took part in drafting, revising or critically reviewing the article; gave final approval of the version to be published; have agreed on the journal to which has been submitted; and agree to be accountable for all aspects of the work.

## Disclosure

The authors report no conflicts of interest for this work and declare that there is no conflict of interests regarding the publication of this paper.

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