

Treatment of Wastewater from Petroleum Industry: Current Practices and Perspectives

Sunita Varjani^{1*}, Rutu Joshi², Vijay Kumar Srivastava³, Huu Hao Ngo⁴, Wenshan Guo⁴

¹Gujarat Pollution Control Board, Gandhinagar - 382010, Gujarat, India

²School of Biological Sciences and Biotechnology, Indian Institute of Advanced Research, Gandhinagar - 382007, Gujarat, India

³Sankalchand Patel University, Sankalchand Patel Sahakar, Visnagar-384315, Gujarat, India

⁴Centre for Technology in Water and Wastewater, School of Civil and Environmental Engineering, University of Technology Sydney, Sydney, NSW - 2007, Australia

*Corresponding author: drsvs18@gmail.com

Abstract

Petroleum industry is one of the fastest growing industry and it significantly contributes to economic growth in developing countries like India. The wastewater from a petroleum industry consist a wide variety of pollutants like petroleum hydrocarbons, mercaptans, oil and grease, phenol, ammonia, sulphide and other organic compounds etc. All these compounds are present as very complex form in discharged water of petroleum industry, which are harmful for environment directly or indirectly. Some of the techniques used to treat oily waste/wastewater are membrane technology, photocatalytic degradation, advanced oxidation process, electrochemical catalysis, etc. In this review paper we aim to discuss past and present scenario of using various treatment technologies for treatment of petroleum industry waste/wastewater. The treatment of petroleum industry wastewater involves physical, chemical and biological processes. This review also provides scientific literature on knowledge gaps and future research directions to evaluate the effect(s) of various treatment technologies available.

Key words: Petroleum hydrocarbons; Membrane technology; Photocatalytic degradation; Waste biorefinery; Resource recovery

29 **1) Introduction**

30

31 Water is necessary for life on earth as it is a basic need of all organisms. Rapid industrialization and economic
32 development have led to exponential growth in population and urbanization (Zafra et al. 2015; Chen 2018). The world
33 is witnessing an increase in urbanization and industrialization due to the consumerist approach (Li and Yu 2011;
34 Zhang et al. 2015). The domestic and industrial sectors continuously generate large amount of waste/wastewater at an
35 alarming rate and usually dispose the waste without proper management and treatment (Varjani et al. 2019). Petroleum
36 industries and refineries are important from economic growth point of view (Li and Yu 2011; Li et al. 2014;
37 Abdulredha et al. 2018). Petroleum refineries and petrochemical plants are facing problem of disposing the
38 waste/wastewater generated. Wastewater released by petroleum industries contains different type of organic and
39 inorganic pollutants such as BTEX, sulphides, hydrocarbons, phenol, heavy metals etc (He and Jiang 2008; Usman et
40 al. 2012; Varjani 2017a; Raza et al. 2018). Large quantities of toxic substances are released by activities of petroleum
41 industry such as production process of oil, transportation, oil refinery, petrochemical product, storage and distribution
42 etc. These are harmful for environment and human health (Perera et al. 2012; Viggi et al. 2015; Raza et al. 2018).

43 The treatment of oily wastewater generated from petroleum industry involves different processes such as (1)
44 physical, (b) chemical and (c) biological processes Various processes available for treatment of petroleum
45 hydrocarbons polluted water ranges from applied methods to emerging methods (Usman et al. 2012; Li et al. 2014;
46 Viggi et al. 2015; Varjani 2017b). Characteristics of oily wastewater and effect of petroleum hydrocarbons on
47 environment and human health have been emphasized in this paper. Sections of this paper also discusses about the
48 current treatment process used in petroleum industry and treatment methods used for treating oily wastewater. The
49 current review focuses to generate critical review about the different practices used for wastewater treatment in
50 petroleum industries. This review also focusses on information pertaining to the knowledge gaps and future research
51 directions in this field.

52

53 **2.) Characteristics of oily wastewater**

54

55 Petroleum means “rock oil” and this word is derived from a Latin word “petra” and Greek word “oleum”
56 (Jafarinejad 2016). Petroleum industry generates large amount of oily waste either solid or liquid due to upstream and

57 downstream operations (Varjani and Upasani 2017). Upstream process includes extracting, transporting and storing
58 crude oil and downstream process includes refining of crude oil (Al-Futaisi et al. 2007; Hu et al. 2013; Thakur et al.
59 2018). Depending on a ratio of water and solids waste is categorized into simple wastewater crude oil or sludge. The
60 pH value of oily sludge is usually ranging from 6.5 to 7.5, but it varies depending on sources of crude oil, processing
61 method, reagents used etc. (Hu et al. 2013; Jasmine and Mukherji 2015). Wastewater from petroleum industry contain
62 oil impurities, high level of BOD and COD, high total solids, hydrocarbons and other waste. This waste includes oily
63 sludge, waste catalyst, heavy metals, volatile organic compounds, oil & grease content, high total dissolved salts,
64 ammonia, nitrates, sulfides, etc. (Jasmine and Mukherji 2015). Oily wastewater contains mainly four major types of
65 petroleum hydrocarbons such as aliphatic, aromatic, asphaltenes and compounds containing oxygen, nitrogen and
66 sulfur. It consists organo-metallic complexes with cadmium, lead, nickel and vanadium (Honse et al. 2012; Varjani
67 and Upasani 2017; Thakur et al. 2018). Table 1 summarizes the characteristics of oily wastewater/effluents and
68 standards for their discharge in environment. These pollutants typically disperse, emulsify or dissolve within the oily
69 wastewater. In general aromatic and aliphatic compounds counts up to 75% of petroleum hydrocarbons in oily
70 wastewater (Ward et al. 2003; Perera et al. 2012; Jasmine and Mukherji 2015; Varjani et al. 2018).

71

72 **3) Effect on environment/ human health**

73

74 Petroleum industry discharge huge amount of pollutants in environment. Wastewater released by the petroleum
75 industries contain large quantities of hydrocarbons, heavy metals, phenols and other toxic chemicals (Perera et al.
76 2012; Jasmine and Mukherji 2015; Thakur et al. 2018; Varjani et al. 2018). Different activities of petroleum industry
77 such as transportation, storage, drilling etc. take part in soil contamination by oil. Depending on type of soil lighter oil
78 can move quickly instead of heavier oil through the soil layers (Fakhru'l-Razi et al. 2009; Varjani et al. 2017). Due to
79 ineffectiveness of treatment system, industrial wastewater become harmful to ecosystem and other life forms
80 (Poulopoulos et al. 2005; Veyrand et al. 2013; Varjani 2017b; Al-Hawash et al. 2018). Oily wastewater can affect
81 different components of environment such as human health, drinking water and groundwater resources, air, crop
82 production and aquatic life etc (Zafra et al 2015; Varjani et al. 2017). Accumulation of toxic products in the water
83 bodies leads serious consequences on the ecosystem and living organisms either long term or short term which may
84 be chronic or acute. (Poulopoulos et al. 2005; Usman et al. 2012; Al-Hawash et al. 2018; Varjani et al. 2018).

85

86 4) Current treatment process in petroleum industry

87

88 Different technologies adsorption, coagulation, anaerobic treatment, reverse osmosis, ultrafiltration, chemical
89 destabilization, flocculation, dissolved air flotation (DAF), membrane process etc. have been used to treat wastewater
90 from petroleum industry (Bennett and Peters 1988; Kriipsalu et al. 2003; Sonune and Ghate 2004; Usman et al. 2012;
91 Li et al. 2014; Padaki et al. 2015). Adsorption method has many advantages rather than other technique such as cost
92 benefit, simplicity and adaptability (Sabah et al. 2007; Ahmad et al. 2016). Depending upon concentration and source
93 of contamination different type of treatment technique is required to reduce the toxic level of pollutants (Sonune and
94 Ghate 2004; Hanafy and Nabih 2007; Bennett and Peters 1988; Farajnezhad and Gharbani 2012; Li et al. 2014; Padaki
95 et al. 2015). Mainly the treatment process has been differentiated into three categories (a) primary, (b) secondary and
96 (c) tertiary treatment (fig.1). Generally effluent of petroleum industry is passed through different stages for reducing
97 toxicity level which is show as a schematic diagram in in fig. 2.

98

99 4.1) Primary treatment

100 Primary treatment is usually used for physical operation in petroleum refinery wastewater treatment plant
101 (Ahmad et al. 2016). It is important step as it allows waste for the further secondary treatment unit. Mostly gravity
102 separation applies to classify the floating and settle down material from wastewater. The primary treatment step
103 includes an oil/water separator which can separate oil, water and solids. Gravity separation followed by skimming is
104 carried out for removal of oil from wastewater (Al-Shamrani et al. 2002; Hanafy and Nabih 2007). Oil-water separator
105 such as API oil–water separator is widely used because of low cost, high effectiveness for primary treatment step. API
106 separators work on phenomena of difference in specific gravity to allow heavy material to settle below lighter liquids.
107 Hydrocarbons that float on the surface and the sludge settle down to the bottom (Al-Futaisi et al. 2007; Ahmad et al.
108 2016). API separator is not very much applicable for removal of smaller oil droplets and emulsions (Abdulredha et al.
109 2018). Dissolved air flotation (DAF) is a water treatment process that uses air to increase the capacity of smaller oil
110 droplets and enhance the separation process. DAF unit typically consist chemicals to promote coagulation and increase
111 floc size to make easy separation. In this stage the heterogeneous components of the effluent such as suspended solids
112 colloids or dispersion, immiscible liquids are reduced significantly (Al-Shamrani et al. 2002; Hanafy and Nabih 2007).
113 Colloids and dispersion also delay and damage equipment during proceeding stage (Renault et al. 2009). In Induced
114 air flotation (IAF) system, air is induced by rotor disperse mechanism, the spinning rotor work as a pump and forces

115 to the fluid. The advantages of the IAF process are compact size, low cost and effective removal of free oil and
116 suspended material (Bennett and Peters 1988; Bennett and Shammam 2010).

117 .

118 4.2) Secondary treatment

119 The main purpose of this stage is to reduce contamination level of effluent and make it in permissible limit
120 for discharge into water bodies. Secondary treatment consists coagulation, flocculation and further biological
121 treatment to reduce toxicity of petroleum wastewater (Xu and Zhu 2004; Viggi et al. 2015; Changmai et al. 2017).
122 Petroleum effluent contains large number of refractory components. Poly aluminium chloride is more effective rather
123 than ferric chloride in coagulation process for treatment of petroleum wastewater (Farajnezhad and Gharbani. 2012;
124 Hosny et al. 2016). **Coagulation–flocculation is a process in which chemical product is added to accelerate the**
125 **sedimentation in clarification tank** ~~Coagulation–flocculation consist on the addition of chemical product to accelerate~~
126 ~~the sedimentation in clarification tank~~ (Kriipsalu et al. 2003; Moulai-Mostefa and Tir, 2004, Hosny et al. 2016). The
127 coagulants are organic or inorganic components such as aluminium hydroxide chloride, aluminium sulphate or high
128 molecular weight cationic polymer. The aim of addition of coagulant is to remove 90% of the suspended solids from
129 the wastewater (Lin and Wen 2003; Changmai et al. 2017). Renault et al. (2009) have reported chitosan for efficient
130 coagulation/flocculation process to treat petroleum industry wastewater. (Renault et al. 2009). Biological treatment is
131 the most widely used method for removal of organic compounds in the oil industry wastewater. Biological treatment
132 is mostly classified in to two categories (a) suspended growth process and (b) attached growth process (Chavan and
133 Mukherji 2008; Srikanth et al. 2018). Suspended growth process includes aerated lagoon, membrane bioreactor
134 technology, sequencing batch reactor (SBR) and activated sludge treatment. In the activated sludge process the
135 wastewater enters in to aeration tank where microorganism brought in contact with contaminated wastewater.
136 Microorganisms use the organic material as food and decompose organic matter (Srikanth et al. 2018). N:P ratio has
137 been reported very important parameter for treatment of oily wastewater by using oil degrading bacteria (Chavan and
138 Mukherji 2008).

139

140 4.3) Tertiary treatment

141 Tertiary treatment process includes sand filtration, activated carbon process and chemical oxidation. It is
142 applicable for removal of total suspended solids, dissolved and suspended matter, COD and trace organics such as
143 PAHs (He and Jiang 2008; Li et al. 2014). After secondary treatment process, effluent contain suspended solids

144 depending on operating conditions in the clarifier. Removal of metals and fine solids which could not be settle down
145 in sedimentation process can be removed by sand filtration (Zahrim and Hilal 2013; Ahmad et al. 2016). This process
146 involves passing the wastewater through a filter bed comprised of a filter media. Generally chemical oxidation is used
147 for reduction of residual COD, trace organic compounds and non-biodegradable compounds. This method uses
148 different type of oxidation reagents like hydrogen peroxide, ozone, chlorine dioxide (Usman et al. 2012; Srikanth et
149 al. 2016).

150

151 **5) Treatment methods for oily wastewater**

152

153 It is important to dispose of effluent in a proper manner and it should maintain a minimum concentration
154 level of chemical(s) which is suitable for environment. Thus, innovative research is required to develop new
155 technologies which degrade the complex molecules into simpler forms, which is reliable to maintain water quality
156 (Sonune and Ghate 2004; Ani et al. 2018). Petroleum industries have made few effective technologies for improving
157 treatment capacity through different methods which are mentioned below and also have also been summarized in table
158 2.

159

160 *5.1) Membrane separation technology*

161 Membrane separation processes are dependent on difference in (a) pressure, (b) concentration of pollutants,
162 (c) temperature and (d) electrical potential (Al-Obaidani et al. 2008; Jamaly et al. 2015; Adhama et al. 2018).
163 Depending upon the pore size of utilized membrane process is categorized as a (a) microfiltration (MF), (b)
164 nanofiltration (NF), (c) ultrafiltration (UF) and (d) reverse osmosis (RO), which are mostly applied to treat oily
165 wastewater (Bruggen et al. 2003; Tomaszewska 2007; Jamaly et al. 2015). This separation method play role in physical
166 removal of way of the trapped particle size of contaminants (Hilal et al. 2004). According to membrane pore size
167 ultrafiltration membranes are more effective than microfiltration membranes. Nanofiltration and reverse osmosis can
168 be also used for separating oil from water especially for high salinity water (Zhu et al. 2014; Jamaly et al. 2015). Based
169 on the oil dispersion, oily water can be classified into three types: (a) free-floating oil, (b) unstable oil-water emulsions
170 and (c) stable oil-water emulsions (Srijaroonrat et al. 1999, Hilal et al. 2004; Qiu et al. 2009). Generally free-floating
171 oil can be easily removed by mechanical techniques and also unstable emulsions removed mechanically but sometimes

172 specific chemical additives are required (Salahi et al. 2013; Abdulredha et al. 2018). Membrane separation efficiency
173 is normally identified by oil rejection coefficient (R_0) and it is defined as:

174

$$175 \quad R_0 = \frac{\text{Oil concentration in feed} - \text{Oil concentration in permeate}}{\text{Oil concentration in feed}} \times 100$$

176

177 Usually, an effective membrane has high rejection coefficients for total organic carbon (TOC), total surface charge
178 (TSC) and chemical oxygen demand (COD) (Karhu et al. 2013). Another important parameter in membrane separation
179 is the permeate flux. Permeate flux is defined by:

180

$$181 \quad J = \frac{V_p}{A t}$$

182

183 Where J is the permeate flux, V_p is the permeate volume, A is the membrane effective area, and t is the permeation
184 time (Rezvanpour et al. 2009). The permeate flux also depends on membrane properties such as porosity, pore size
185 and hydrophilicity (Mohammadi et al. 2003; Changmai et al. 2017). Due to small pore size of membrane, oil rejection
186 coefficient is higher and use of large pore size membrane leads high permeate flux (Kocherginsky et al. 2003; Colle
187 et al. 2009; Han et al. 2017).

188

189 5.2) *Advanced oxidation process*

190 The main function of advanced oxidation process is to generate high reactive free radicals. Hydroxyl radicals
191 are effective in destroying organic chemicals because they are reactive electrophiles (Oller et al. 2011; Jamaly et al.
192 2015). Hydroxyl radical is strong oxidant to destroy compound that cannot be oxidized by conventional oxidant.
193 Hydroxyl radical possess faster oxidation rate as compared to H_2O_2 or $KMnO_4$ (Gogate and Pandit 2004; Usman et al
194 2012; Tijani et al. 2014). Generated hydroxyl radicals can attack organic chemicals by radical addition, hydrogen
195 abstraction and/or electron transfer (Gogate and Pandit 2004; Tijani et al. 2014). Generation of hydroxyl radical is
196 commonly accelerated by combining O_3 , H_2O_2 , TiO_2 , UV radiation, electron-beam irradiation and ultrasound (Usman
197 et al 2012; Li et al. 2014; Varjani 2017b).

198

199 5.3) *Electrochemical catalysis*

200 Electrochemical catalysis process is related to oxidation of hydroxyl radical with a highly organic matter
201 between substitution, addition and electron transfer process (Zhang et al. 2015; Mohan et al. 2018). It leads to the
202 degradation of pollutants, mineralization, easy to build airtight circulation with no secondary pollution (Lin et al.
203 2001; Dimoglo et al. 2004; Koper 2005; Bajracharya et al. 2015; Changmai et al. 2017). Electrolysis process is applied
204 to treat oily wastewater which leads to time dependent reduction in chemical oxygen demand (COD). The process
205 consists three steps for effective reduction of pollutants in oily wastewater and do the remediation. The first step is
206 the direct oxidation of oil components at the electrode, by the metal oxide itself or by hydroxyl radicals available at
207 electrode surface. The second step is indirect oxidation of oil components by intermediate oxidizing agents formed
208 and the third step is aggregation of suspended oil droplets by electro-flotation (Santos et al. 2006; Jamaly et al. 2015).
209 Electrochemical catalytic treatment is more effective treatment of oily wastewater (Mohan et al. 2018). Ma and Wang
210 (2006) have performed research with pilot scale plant having double anodes and cathodes. Anode contained active
211 metal and graphite, however cathode contained iron and a noble metal content catalyst with big surface. They have
212 reported approximately 90% reduction of biochemical oxygen demand (BOD) and chemical oxygen demand (COD)
213 and 99% reduction in suspended solids by using the electrochemical process.

214

215 5.4) *Photocatalytic Degradation*

216 Photolytic degradation is very well researched in last decade and conventionally a method for treating
217 petroleum wastewater (Twesme et al. 2006; Do et al. 2010; Yen et al. 2011; Varjani 2017b). Photocatalysis is
218 beneficial process used to oxidize persistent compounds which cannot be oxidized during biological treatment
219 (Vodyanitskii et al. 2016). TiO₂ and ZnO are commonly used photocatalysts in the treatment of petroleum industry
220 wastewater (Santos et al. 2006; Akpan and Hameed 2009; Vodyanitskii et al. 2016). Benefits of TiO₂/UV technique
221 are low cost, faster reaction rates, no sludge production and easy operation at ambient temperature and pressure with
222 complete mineralization of petroleum industry wastewater (Twesme et al. 2006; Akpan and Hameed 2009; Varjani
223 2017b). Park and Choi (2005) have reported photocatalytic hydroxylation of aromatic ring in presence of platinum
224 loaded TiO₂, where they have used water as an oxidant. Photocatalytic degradation of naphthalene present in petroleum
225 industry waste using TiO₂ in presence of inorganic anions had been reported by Lair et al. (2007).

226

227 **6) Knowledge gaps and Future perspectives**

228

229 Petroleum hydrocarbon pollutants are classified as priority pollutants. The pollutants which are present in
230 petroleum industry wastewater can be effectively remediated using different technologies (Yen et al. 2011;
231 Vodyanitskii et al. 2016). Research on study of innovative technologies with minimum environmental and economical
232 influence indicating that it is thrust areas of research (Lin et al. 2001; Santos et al. 2006; Viggi et al. 2015; Vodyanitskii
233 et al. 2016). This review article focuses on improving different technologies used for treatment of wastewater
234 generated by petroleum industry activities. Petroleum industry wastewater contain different toxic substances such as
235 toluene, xylene, benzene, ethylbenzene, phenols and PAHs, etc (Hanafy and Nabih 2007; Renault et al. 2009; Varjani
236 et al. 2017). These pollutants are very difficult to be directly removed by using single method specifically biological
237 treatment which is considered as green approach. Hence advanced chemical or physical treatments in combination
238 with biological treatment are required. Apart from this for biological treatment to identify the successful species which
239 can work *ex-situ* and *in-situ* conditions is a very important task (Varjani 2017b; Chen 2018). It has been reported that
240 integration of various processes may give better results than individual process used for treatment of oily wastewater.
241 But knowledge of integration of technologies is still at its infancy which needs to be explored by researchers. Recent
242 developments are mostly more expensive, require maintenance and it's a time taking process. Hence, all industries do
243 not participate to reduce toxicity of effluent. Future technology needs easy operating systems which are suitable for
244 small, medium and big industries.

245 Huge amount of solid and liquid waste is produced due to petroleum industry activities. Management and
246 handling of waste generated is nowadays is a big issue for local authorities not only in urban areas i.e. municipalities
247 but also other regions in any country (Santos et al. 2006; Li and Yu 2011; Tijani et al. 2014; Jamaly et al. 2015; Varjani
248 and Sudha 2018). Due to increased urbanization and industrialization generation of wastewater, appropriate disposal,
249 treatment and/or recycling is posing more challenges as treatment and disposal costs large amount in terms of money.
250 However, if we consider that “waste” word is placed wrongly and if that be used as a “resource” then resource recovery
251 from wastes is an emerging as thrust area of research and management because it offers environment and social
252 sustainability potentials.

253 Many research groups are focusing their work on recovery of various resources such as energy, bio-products,
254 nutrients (nitrogen and phosphorous), metals from wastewater generated by anthropogenic activities (Li and Yu 2011;
255 Honse et al. 2012; Varjani and Upasani 2017; Mohan et al. 2018; Thakur et al. 2018; Varjani et al. 2019). Valuable
256 pollutants are present in petroleum industrial wastewaters, which can be regarded as resources after recovery (Al-

257 Futaisi et al. 2007; Hu et al. 2013; Jasmine and Mukherji 2015). On one side efficient resource recovery and reuse can
258 create sustainable livelihood whereas on the other side it supports green economy by reducing waste and improve
259 environmental health and cost of recovery. Hence, there is a need to recycle and reuse the waste produced from
260 activities of petroleum industry in an efficient manner. Feasible techniques to produce pollution less products create
261 a new way for environmental and economic sustainability.

262 To optimize the exploitation of petroleum industry waste/wastewater and by-products there is a need to develop
263 sustainable technologies. Focus of the research shall also be thrown on biorefinery concept for development of
264 innovative bio-based industries because waste biorefineries may open up new market opportunities for bio-based
265 products and achieve efficient resource utilization.

266

267 **7) Conclusion**

268

269 Petroleum industry wastewater can be treated by various physical, chemical and biological treatment
270 processes. There are many hazardous components present in waste/wastewater released by the activities of petroleum
271 industry. Successful remediation strategy should be tailored considering environment and human health.
272 Bioremediation is one of the technologies which is gaining a global interest for cleanup of petroleum hydrocarbons.
273 However, integration of various processes gives better results than individual process used for waste/wastewater
274 treatment. The current practice of industrial wastewater treatment is focused to remove pollutants from wastewaters
275 to meet the discharge standards. To resolve hazards associated with petroleum components, a suitable technology that
276 treats waste and generates value addition would be a promising option. Study regarding the recovery of value-added
277 products from waste/wastewater with special reference to different techniques, either separately or by integration,
278 tailoring distinct features of processes are thrust area of research with respect to waste characteristics for production
279 of bio-based non-toxic by-products. Waste biorefinery concept using latest developments in biotechnological and
280 bioengineering options pertaining to recovery of resources from petroleum industry waste/wastewater shall also be
281 explored.

282

283 **References**

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Table Legend

Table 1. Characteristics of oily wastewater/effluents and environmental standards for their discharge

Table 2. Petroleum Industry Wastewater Treatment Technologies

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Table 1. Characteristics of oily wastewater/effluents and environmental standards for their discharge
(Source: G.S.R 186(E), dated 18th March 2008)

Sr. No.	Parameter	Limiting value for concentration*
1	pH	6.0-8.5
2	Chemical oxygen demand	125.0
3	Biological oxygen demand 3 days, 27°C	15.0
4	Oil and Grease	5.0
5	Suspended Solids	20.0
6	Total Kjeldahl nitrogen	40.0
7	Sulphides	0.5
8	Phenols	0.35
9	Cyanide	0.20
10	Benzene	0.1
11	Benzo[a]pyrene	0.2
12	Metal contents	
	Hexavalent Chromium	0.1
	Total Chromium	2.0

Lead	0.1
Mercury	0.01
Nickel	1.0
Zinc	5.0
Copper	1.0
Vanadium	0.2

*Limiting value for concentration is in mg/L, except for pH

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Table 2. Petroleum industry wastewater treatment technologies (Source: Dimoglo et al. 2004; Lin et al. 2001 2000; Vaiopoulou et al. 2005; Knight et al. 1999)

Sr. No.	Name of the techniques used	Treatment details
1	Electrochemical technologies	To remove turbidity, COD, phenol, hydrocarbon and grease by using electro coagulation and electro flotation from petroleum wastewater
2	Ozonation and biological activated carbon (BAC)	To enhance biodegradation process of bio refractory and the growth of a biofilm have been noticed during laboratory scale pre-ozonation and biological activated carbon (BAC)
3	Anaerobic treatment	Anaerobic digestion method has been used for treatment of hydrocarbon pollutants
4	Aerobic sequencing batch reactors	Observer-based time-optimal control: Control strategy regulates and maintains the substrate concentration and feed ratio in the

		reactor. An extended Kalman filter has been used as a nonlinear observer in a petrochemical wastewater treatment
5	Autotrophic denitrification	To remove sulfide from petroleum industry waste water, new alternative treatment for removal of H ₂ S by combination of the biological method and existing stripping CO ₂
7	Wetlands	Some large-scale project and pilot scale studies have been conducted for extraction and pumping stations of oil and gas for treatment using wetland for oil refineries

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532 **Figure Captions**

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534 Figure 1. Petroleum industry wastewater treatment

535 Figure 2. Treatment of wastewater of petroleum industry: Schematic diagram

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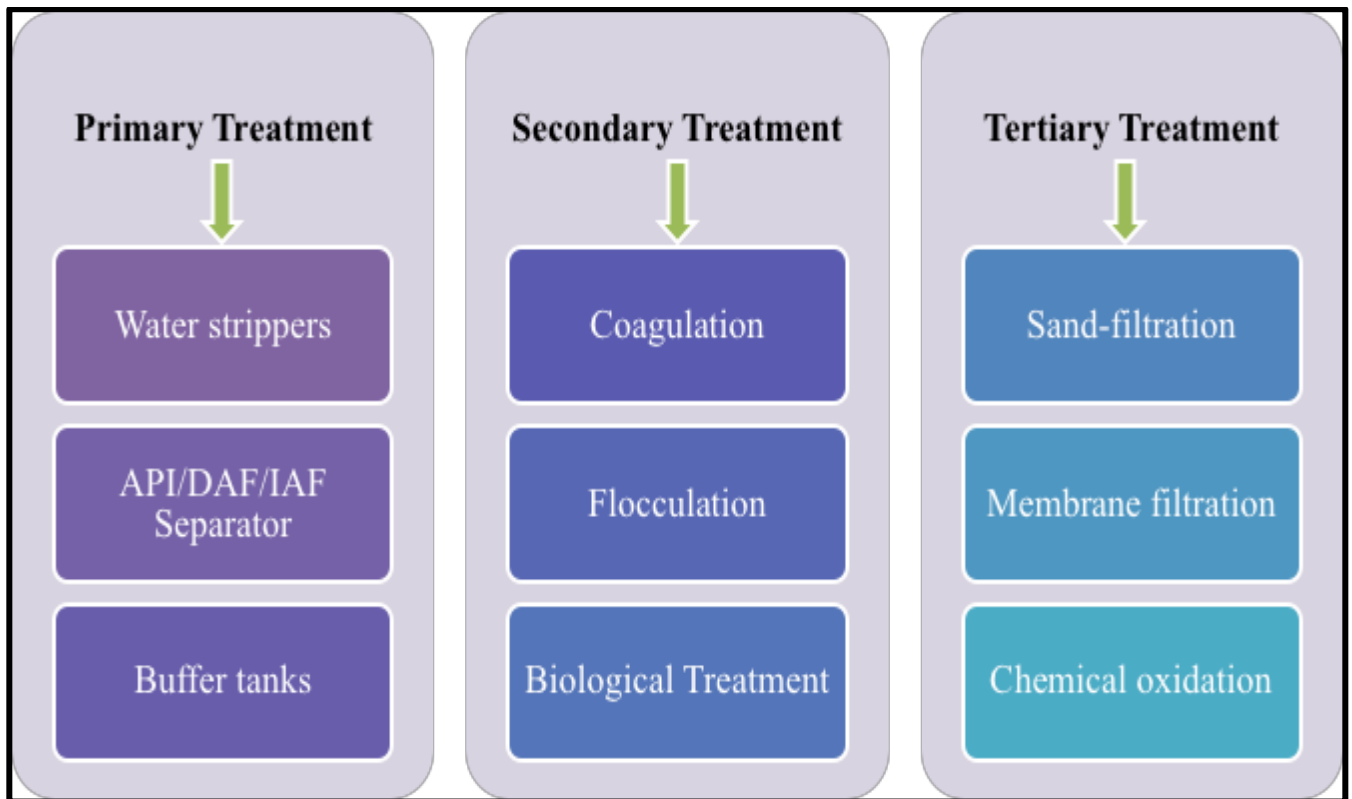
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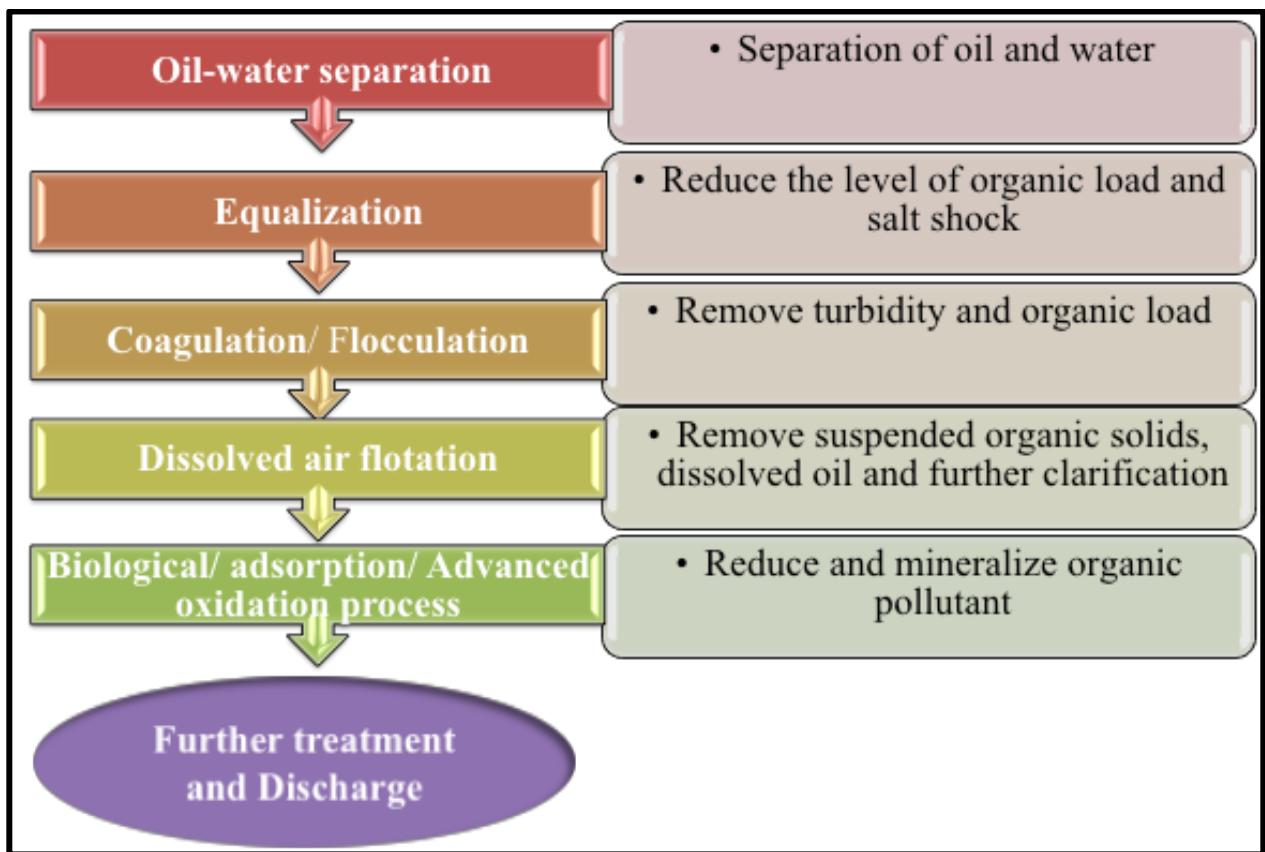
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Figure 1. Petroleum industry wastewater treatment

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570 **Figure 2. Treatment of wastewater of petroleum industry: Schematic diagram**