

# Two-dimensional molybdenum carbide (MXene) as an efficient nanoadditive for achieving superlubricity under ultrahigh pressure

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**Abstract:** In this study, a robust macroscale liquid superlubricity with a coefficient of friction of 0.004 was achieved by introducing molybdenum carbide ( $\text{Mo}_2\text{CT}_x$ ) MXene nanoparticles as lubricating additives in a lithium hexafluorophosphate-based ionic liquid at  $\text{Si}_3\text{N}_4$ -sapphire interfaces. The maximal contact pressure in the superlubricity state could reach 1.42 GPa, which far exceeds the limit of the superlubricity regime in previous studies. The results indicate that a composite tribofilm (mainly containing molybdenum oxide and phosphorus oxide) that formed at the interface by a tribochemical reaction contributed to the excellent antiwear performance. Furthermore, the extremely low shear strength of the tribofilm and the interlayers of  $\text{Mo}_2\text{CT}_x$  MXene contributed to the superlubricity. This work demonstrates the promising potential of  $\text{Mo}_2\text{CT}_x$  MXene in improving superlubricity properties, which could accelerate the application of superlubricity in mechanical systems.

**Keywords:** molybdenum carbide ( $\text{Mo}_2\text{CT}_x$ ) MXene; superlubricity; additives; ultrahigh pressure; wear resistance

## 1 Introduction

Friction and wear cause significant energy loss on a mechanical surface, serious material loss, shortened service life, and increased safety accidents involving machinery [1, 2]. Therefore, many researchers have used various methods to reduce friction and wear [3]. In this context, a new concept of superlubricity, with a coefficient of friction (COF) of less than 0.01, was proposed to enhance lubrication performance at the nanoscale or macroscale, which could provide ultralow friction and wear to reduce energy loss. Currently, different superlubricity approaches exist, such as solid, liquid, and solid-liquid coupling superlubricity systems. For solid superlubricity, molybdenum disulfide ( $\text{MoS}_2$ ) [4], graphite sheets [5], the graphene family of materials [6, 7], and diamond-like carbon

[8, 9] could act as typical solid superlubricity materials at the nanoscale or macroscale.

Compared with solid superlubricity, liquid superlubricity is more efficiently realized at the macroscale with lubrication by various liquids, such as phosphoric acid [10], ionic liquid (IL) [11], oleic acid [12], polyalphaolefin oil [13], and biological liquid [14]. To enhance the load capacity and antiwear ability in the liquid superlubricity state, conventional layered materials, including black phosphorus [15], graphene materials [7, 16], hexagonal boron nitride [17], and layered double hydroxides [18], have been used as lubricant additives because of their extremely low shear strength in their interlayers. These layered materials have a similar two-dimensional (2D) layered structure, which can be adsorbed on the worn-out zone after the friction process to reduce wear and

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simultaneously provide extremely low shear strength between friction pairs. However, one limitation of these liquid materials, related to the contact pressure (dividing the normal load by the contact area), must be improved. With lubrication by these liquids, even though the initial contact pressure between the frictional pairs is high (on the order of gigapascals), the final contact pressure during the superlubricity period is still at a much lower order of magnitude. This is caused by the severe wear occurring during the running-in period.

Recently, a novel member of the 2D layered material family, MXene, was formed by a few atoms with thick layers of transition metal carbides, nitrides, or carbon nitride based on MAX-phases [19–21]. This material exhibits excellent electrochemistry, thermoelectricity, lubrication, and other properties, making it suitable for use in energy storage [22], catalysis [23], lubrication [24], etc. Some studies on the MXene lubrication have been reported [25, 26]. For example, in a number of studies,  $Ti_3C_2$  MXene or its composite materials were used as solid coatings to enhance wear resistance and tribological properties [27–32]. Other researchers have investigated MXene nanoparticles as lubricant additives that can reduce friction and wear [33–35]. Although previous research on the MXene lubrication has revealed that MXene has the potential to achieve low friction and wear, industry requirements (ultralow friction and ultrahigh contact pressure) still hinder the application of MXene. In this study,  $Mo_2CT_x$  MXene nanoparticles were synthesized and introduced into lithium hexafluorophosphate-based IL as a nanoadditive. Robust macroscale superlubricity has been achieved under an ultrahigh contact pressure of as much as 1.42 GPa, which is a 136.67% increase compared with previous results in Ref. [36]. The lubricating behavior related to this liquid superlubricity under ultrahigh pressure using  $Mo_2CT_x$  MXene as a lubricating nanoadditive was studied, and the related mechanism was revealed.

## 2 Experimental

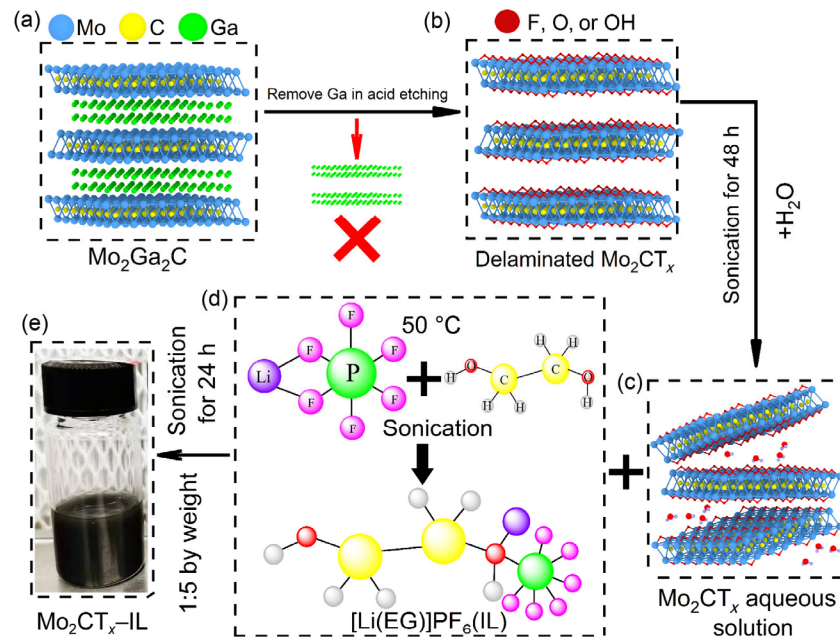
### 2.1 MXene and lubricant synthesis

The precursor  $Mo_2Ga_2C$  powders were synthesized from gallium ingots (Ga, 99.99 wt%, Beijing

Xingrongyuan Co., Ltd., China) and molybdenum carbide powders ( $Mo_2C$ , 99.5 wt%, 325 mesh, Sigma Aldrich Co., Ltd., USA).  $Mo_2C$  and Ga were weighed in a molar ratio of 1:5 [37, 38]. Subsequently, the resulting  $Mo_2Ga_2C$  powders were added to 40 mL of etching solution. The etching solution was prepared by mixing 2.0 g of LiF (99.9 wt%, Aladdin Industrial Co., Ltd., China) and 6-M HCl (11.8 mol/L, Shuangshuang Chemical Reagent Co., Ltd., China). Afterward, the mixture was washed until a pH of approximately 6.0 by using deionized water to obtain  $Mo_2CT_x$  MXene particles. Thereafter, the collected  $Mo_2CT_x$  MXene particles were mixed with deionized water at a ratio of 0.2–5 mg/mL to prepare an aqueous solution of  $Mo_2CT_x$  MXene and sonicated for 48 h. The IL formed *in situ* was synthesized by dissolving 0.1 g of  $LiPF_6$  (purity of 97%, Aladdin Industrial Co., Ltd., China) in 2 g of ethylene glycol (EG) (purity of 99%, Aladdin Industrial Co., Ltd., China) via sonication for 8 h, followed by magnetic stirring at 50 °C for 8 h (Fig. 1(d)). Once the IL was prepared, it was immediately mixed with a  $Mo_2CT_x$  MXene aqueous solution at a ratio of 1:5 by weight, and the mixtures were denoted as  $Mo_2CT_x$ -IL (Fig. 1(e)). The concentration of nanoparticles varied from 0.021 to 0.416 wt%. For comparison, the graphene oxide (GO)-IL and  $MoS_2$ -IL solutions were prepared with the same volume concentration as that of  $Mo_2CT_x$ -IL.

### 2.2 Material characterization

The dried  $Mo_2CT_x$  MXene particles were tested by the atomic force microscope (Icon, Bruker, Germany) to measure the sizes of the  $Mo_2CT_x$  MXene nanoparticles. The crystal structures of the  $Mo_2CT_x$  MXene nanoparticles were detected using the X-ray diffractometer (XRD) (Bruker, Germany) with  $2\theta$  ranging from 5° to 65°. The morphology of the  $Mo_2CT_x$  MXene nanoparticles was analyzed by the high-resolution transmission electron microscope (HRTEM) (2100F, JEM, Japan). The original chemical composition of the  $Mo_2CT_x$  MXene nanoparticles was characterized by the X-ray photoelectron spectroscope (XPS) (PHI Quantera II, Japan). The chemical groups of  $Mo_2CT_x$  MXene and IL were detected by the Fourier transform infrared spectroscope (FTIR) (Vertex 70V, Bruker, Germany) and the Raman spectroscope (HR-800, Horiba, Japan),



**Fig. 1** (a–c) Preparation of  $\text{Mo}_2\text{CT}_x$  MXene aqueous solution and schematic structure of  $\text{Mo}_2\text{CT}_x$  MXene, where the blue, yellow, and green orbs represent molybdenum, carbon, and gallium atoms, respectively, and the red orbs are the functional groups of fluorine, oxygen, or hydroxyl functional-containing groups. (d) Preparation of IL by dissolving lithium hexafluorophosphate (0.1 g) in ethylene glycol (2 g) via 8 h of sonication at 50 °C. (e) Preparation of  $\text{Mo}_2\text{CT}_x$ -IL solution by mixing IL and  $\text{Mo}_2\text{CT}_x$  aqueous solution with a mass ratio of 1:5 via 24 h of sonication.

respectively. The average size of the  $\text{Mo}_2\text{CT}_x$  MXene was measured using a commercial zeta potential instrument (Zetasizer Nano Series, Malvern, UK).

### 2.3 Tribological test

Friction and wear tests were conducted using a universal micro-tribotester in a ball-on-disk mode (UMT-5, Bruker, USA). A fixed  $\text{Si}_3\text{N}_4$  ball was loaded onto a rotatable sapphire disk. Subsequently, 20  $\mu\text{L}$  of the lubricant was dripped onto the contact surfaces in each test. The applied load of the ball varied from 1 to 6 N, corresponding to the initial contact pressures of 1.31–2.40 GPa at the interface. The sliding speed of the disk ranged from 0.004 to 0.375 m/s. All tests were conducted at a room temperature of 25 °C and a relative humidity of 20%–25%.

### 2.4 Surface characterization

The diameter of the wear scar and the roughness of the worn-out zone on the ball and plate were observed using a 3D white-light interferometer (Nexview, ZYGO Lambda, USA). The topographies of the worn zones were acquired by the scanning electron microscope

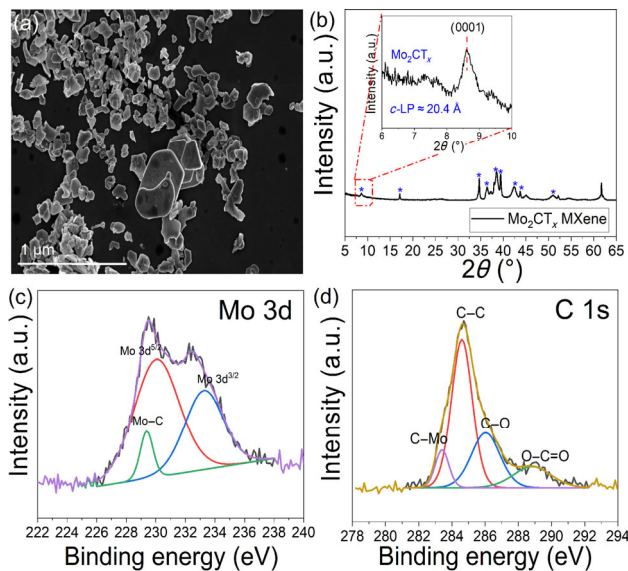
(SEM) (Quanta 200 FEG, FEI, USA). The wear zones on the surfaces were analyzed using the XPS and Raman spectroscopy after the friction test. The cross-sectional structures of the worn zones on the surfaces were observed using HRTEM.

## 3 Results and discussion

### 3.1 $\text{Mo}_2\text{CT}_x$ MXene characterization

$\text{Mo}_2\text{CT}_x$  MXene was synthesized by etching Ga from  $\text{Mo}_2\text{Ga}_2\text{C}$  powders using etchants, and subsequently delaminated by sonication in water (Fig. 1). The  $\text{Mo}_2\text{CT}_x$ -IL solution was prepared by mixing a lithium hexafluorophosphate-based IL and a  $\text{Mo}_2\text{CT}_x$  MXene aqueous solution with a mass ratio of 1:5. After the synthesis of  $\text{Mo}_2\text{CT}_x$  MXene, the morphology of  $\text{Mo}_2\text{CT}_x$  MXene was first observed using SEM (Fig. 2(a)), which showed a highly crystalline structure of  $\text{Mo}_2\text{CT}_x$  MXene nanoparticles.

The lattice structures of  $\text{Mo}_2\text{CT}_x$  MXene were investigated by XRD, as shown in Fig. 2(b), showing the first peak at  $2\theta = 8.6^\circ$ , which was smaller than that of the precursor  $\text{Mo}_2\text{Ga}_2\text{C}$  [39]. This is an indication



**Fig. 2** (a) SEM image of  $\text{Mo}_2\text{CT}_x$  MXene nanoparticles, (b) XRD pattern of  $\text{Mo}_2\text{CT}_x$  MXene, (c) XPS spectra of Mo 3d of the original  $\text{Mo}_2\text{CT}_x$  MXene, and (d) XPS spectra of C 1s of the original  $\text{Mo}_2\text{CT}_x$  MXene.

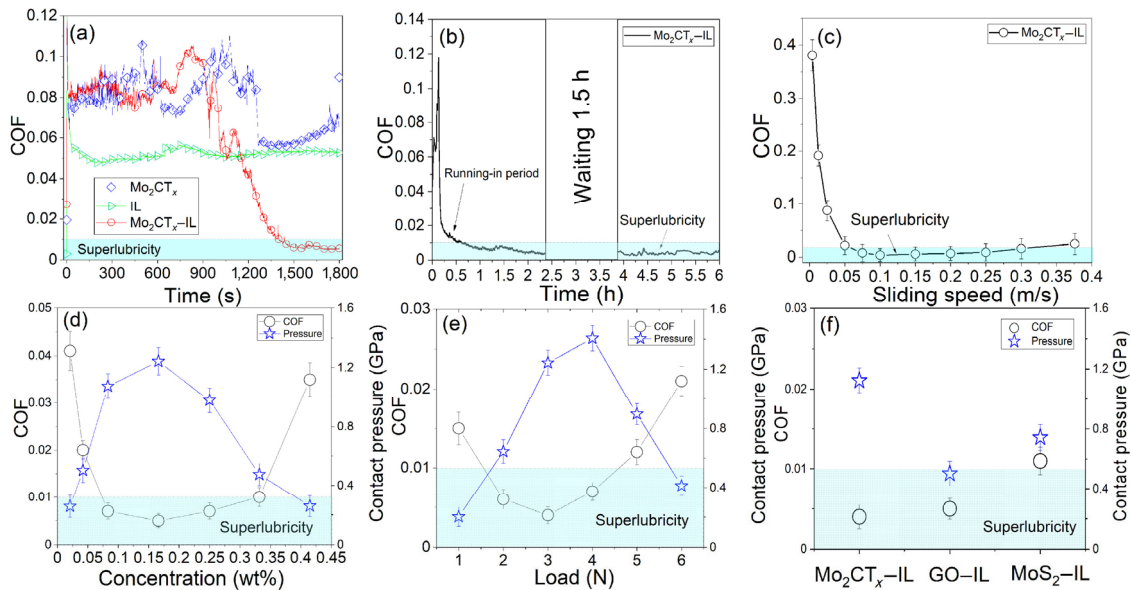
of the successful removal of the gallium layers. The peak at  $8.6^\circ$  was indicative of the typical MAX to MXene transformation, and the (0001) peak broadened and shifted to a lower angle because of an increase in the *c*-lattice parameter (*c*-LP) from 18.1 to 20.4 Å [39–41]. The high-resolution XPS spectra of Mo 3d and C 1s of the original  $\text{Mo}_2\text{CT}_x$  MXene are shown in Figs. 2(c) and 2(d), respectively. For the Mo 3d, there were three main peaks located at 229.8, 230.8, and 233.1 eV. The peaks were assigned to Mo–C, Mo 3d<sup>5/2</sup>, and Mo 3d<sup>3/2</sup> [42], respectively. For C 1s, four main peaks were located at 283.2, 284.6, 286.1, and 289.3 eV. These peaks were assigned to C–Mo, C–C, C–O, and O–C=O bonds, respectively. FTIR was used to detect the reciprocities between IL and  $\text{Mo}_2\text{CT}_x$  (Fig. S1(a) in the Electronic Supplementary Material (ESM)). The peaks at 1,087 and 1,109  $\text{cm}^{-1}$  proved the existence of C–O bonds, and the peak at 1,642  $\text{cm}^{-1}$  proved the presence of Mo. Furthermore, the peaks at 2,875 and 2,938  $\text{cm}^{-1}$  indicated the existence of C–H, and the peak at 3,305  $\text{cm}^{-1}$  indicated the existence of hydroxyl groups [43]. The peaks in the Raman spectra located at 488 and 546  $\text{cm}^{-1}$  are the standard peaks of  $\text{Mo}_2\text{CT}_x$  MXene (Fig. S1(b) in the ESM), confirming the successful preparation of the  $\text{Mo}_2\text{CT}_x$  MXene. The size distribution of  $\text{Mo}_2\text{CT}_x$  MXene in IL was measured using a Zetasizer, which showed that the MXene in IL had

an average size of 531 nm (Fig. S1(c) in the ESM).

### 3.2 Tribological performances

To estimate the macroscale tribological properties of  $\text{Mo}_2\text{CT}_x$ -IL, the friction experiments were performed at the  $\text{Si}_3\text{N}_4$ -sapphire interface with lubrication by IL,  $\text{Mo}_2\text{CT}_x$  aqueous solution, and  $\text{Mo}_2\text{CT}_x$ -IL, respectively, under a load of 3 N and a sliding speed of 0.1 m/s (Fig. 3(a)). The results showed that the COF first displayed an unstable oscillation in the range of 0.08–0.1 from 0 to 1,200 s with lubrication by the  $\text{Mo}_2\text{CT}_x$  MXene aqueous solution. Subsequently, the COF started a sudden drop from 0.1 to 0.058. However, this declining tendency of the COF was not maintained for a long time, and it started to rise after 1,500 s. This performance indicates that the  $\text{Mo}_2\text{CT}_x$  nanoparticles can reduce friction, but they cannot result in an ultralow COF. Thereafter, the IL (without MXene) as a lubricant was tested, showing a robust state with a COF of approximately 0.05 in the entire friction test. This phenomenon indicates that IL has good stability as a lubricant, but it still cannot achieve superlubricity. Based on these results,  $\text{Mo}_2\text{CT}_x$  MXene nanoparticles were selected as lubricating nanoadditives to be introduced into the IL, and the COF of the  $\text{Mo}_2\text{CT}_x$ -IL solution showed a similar tendency to that of  $\text{Mo}_2\text{CT}_x$  during the initial period of 900 s. Subsequently, the COF gradually dropped from 0.09 to 0.005 during the period of 900–1,350 s, which entered the superlubricity regime. Finally, the COF maintained stably at 0.004, and it could be sustained for at least 4.5 h (Fig. 3(b)), corresponding to a sliding distance of 1,620 m.

After the friction test, the wear scar and surface roughness on the  $\text{Si}_3\text{N}_4$  and sapphire surfaces were characterized using 3D white-light interferometry. The results showed that the diameters of the wear scar on the  $\text{Si}_3\text{N}_4$  surface were 55.46 and 91.2  $\mu\text{m}$  after lubrication by  $\text{Mo}_2\text{CT}_x$ -IL and IL, respectively. The corresponding widths of the wear tracks on the sapphire surface were 59.6 and 105.3  $\mu\text{m}$ , respectively (Fig. S2 in the ESM). The average contact pressures (dividing the normal load by the contact area) after lubrication by  $\text{Mo}_2\text{CT}_x$ -IL and IL were 1.24 GPa and 459.48 MPa, respectively. Compared with lubrication by IL, the contact pressure directly increased by 170.46% with lubrication by  $\text{Mo}_2\text{CT}_x$ -IL, indicating that the  $\text{Mo}_2\text{CT}_x$



**Fig. 3** (a) COF vs. time with lubrication by IL,  $\text{Mo}_2\text{CT}_x$  aqueous solution, and  $\text{Mo}_2\text{CT}_x\text{-IL}$  with a load of 3 N and a sliding speed of 0.1 m/s. (b) COF for long-run testing (up to 4.5 h) with lubrication by  $\text{Mo}_2\text{CT}_x\text{-IL}$ . (c) Influence of sliding speed on COF during the superlubricity period with a normal load of 3 N. The concentration of  $\text{Mo}_2\text{CT}_x$  in IL is 0.166 wt%. (d) COFs and corresponding contact pressures during the superlubricity period with lubrication by various concentrations of  $\text{Mo}_2\text{CT}_x\text{-IL}$  with a normal load of 3 N and a sliding speed of 0.1 m/s. (e) COFs and corresponding contact pressures during the superlubricity period under different normal loads (1–6 N) with a concentration of 0.166 wt% and a sliding speed of 0.1 m/s. (f) COFs and corresponding contact pressures during the superlubricity period with different solutions, including  $\text{Mo}_2\text{CT}_x\text{-IL}$ , GO-IL, and  $\text{MoS}_2\text{-IL}$  with the same volume concentration as  $\text{Mo}_2\text{CT}_x\text{-IL}$ . The normal load is 3 N, and the sliding speed is 0.1 m/s.

MXene nanoparticles as additives in IL could reduce friction and suppress wear to enhance the contact pressure. Generally, a high contact pressure is detrimental to the achievement of superlubricity. However, the introduction of  $\text{Mo}_2\text{CT}_x$  MXene additives in ILs could make the contact pressure of superlubricity exceed 1 GPa, which is far higher than that achieved by elastohydrodynamic lubrication (< 300 MPa) [44]. Meanwhile, the surface roughness of the wear scar on the  $\text{Si}_3\text{N}_4$  surface after lubrication by  $\text{Mo}_2\text{CT}_x\text{-IL}$  was 23.2 nm, which was significantly smaller than that with lubrication by IL (56.4 nm). The corresponding surface roughnesses of the wear tracks on the sapphire surface were 18.9 and 33.4 nm with lubrication by  $\text{Mo}_2\text{CT}_x\text{-IL}$  and IL, respectively.

An additional investigation of the influence of the sliding speed on COF was performed with lubrication by  $\text{Mo}_2\text{CT}_x$  MXene nanoparticles as additives in the IL (Fig. 3(c)). The COF decreased from 0.37 to 0.018 when the sliding speed increased from 0.004 to 0.05 m/s, which was noticeably in the non-superlubricity regime. Subsequently, the COF decreased to 0.004 when the sliding speed further increased to 0.075 m/s, which

entered the superlubricity regime. This was followed by maintaining the superlubricity state (the COF varied slightly from 0.004 to 0.008) when the sliding speed increased from 0.075 to 0.25 m/s. When the sliding speed exceeded 0.3 m/s, the COF was greater than 0.01, which was indicative of the failure of the superlubricity state. The evolution of the COF with sliding speed was consistent with the Stribeck curve, indicating that the hydrodynamic effect may also influence the superlubricity state. When the sliding speed was extremely low, the hydrodynamic effect was weak, leading to an increase in the COF. This illustrates that there is an optimal range of sliding speeds in which a robust superlubricity state can be achieved with lubrication by  $\text{Mo}_2\text{CT}_x\text{-IL}$ .

Subsequently, a friction test was carried out with lubrication using different concentrations of  $\text{Mo}_2\text{CT}_x$  MXene as lubricating additives (from 0.021 to 0.416 wt%) in IL (Fig. 3(d)). The extremely low concentration (< 0.042 wt%) clearly cannot lead to the achievement of superlubricity. The COF decreased from 0.042 to 0.021 when the concentration of  $\text{Mo}_2\text{CT}_x$  MXene increased from 0.021 to 0.042 wt%. However, when the

concentration of  $\text{Mo}_2\text{CT}_x$  MXene reached 0.083 wt%, the COF decreased to 0.007, which entered the superlubricity regime. When the concentration of  $\text{Mo}_2\text{CT}_x$ -IL further increased from 0.083 to 0.332 wt%, the results showed that the COF first decreased slightly from 0.008 to 0.004 when the concentration of  $\text{Mo}_2\text{CT}_x$  MXene reached 0.166 wt%, and then it had a slight increase (the COFs were always less than 0.01). However, when the  $\text{Mo}_2\text{CT}_x$  MXene concentration reached 0.416 wt%, the superlubricity state was damaged with a high COF of 0.035. Moreover, the lubrication state became unstable with a violent fluctuation of the COF (Fig. S3 in the ESM). These results indicate that the COF was significantly influenced by the concentration of  $\text{Mo}_2\text{CT}_x$  MXene, and an optimal concentration range (0.083–0.332 wt%) was necessary to achieve the superlubricity state. The diameter and wear volume of the wear scar on the  $\text{Si}_3\text{N}_4$  surface were also measured after the friction test (Figs. S4 and S5 in the ESM), showing that suitable concentrations of 0.083–0.25 wt% of  $\text{Mo}_2\text{CT}_x$  could also suppress the surface wear and meanwhile achieve the superlubricity state, which leads to a high contact pressure ( $> 900$  MPa) for superlubricity. The maximum contact pressure in the superlubricity state could reach 1.24 GPa with a concentration of 0.166 wt% (Fig. 3(d)). However, when the concentration was too low or too high, superlubricity could not be achieved, and there was also severe wear (corresponding to a low contact pressure). This is because there were not enough  $\text{Mo}_2\text{CT}_x$  nanoparticles to enter the contact zone when the concentration was extremely low, and the agglomeration of the  $\text{Mo}_2\text{CT}_x$  nanoparticles occurred when the concentration was very high, both of which were harmful for achieving superlubricity and reducing surface wear.

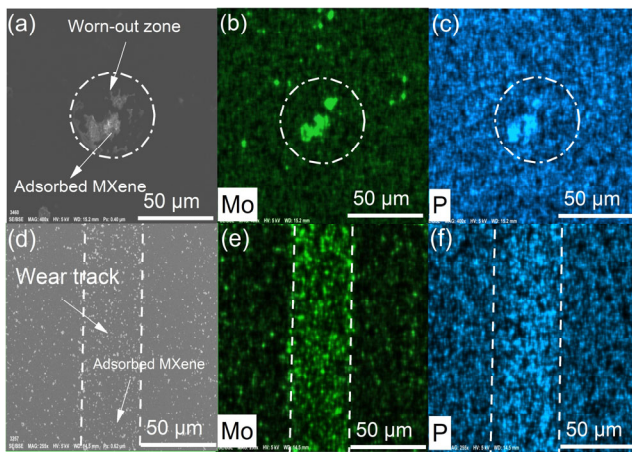
Next, the influence of the normal load on the COF with lubrication by the 0.166 wt% of  $\text{Mo}_2\text{CT}_x$  MXene as additives in IL was investigated (Fig. 3(e)). When a normal load of 1 N was applied, the final COF decreased to 0.015, which belongs to the non-superlubricity regime. After the normal load was increased to 2 N, the COF directly dropped to 0.006, which entered the superlubricity regime. The lowest average value of the COF, 0.004, was achieved by further increasing the normal load to 3 N. However,

when the normal load exceeded 3 N, the COF gradually increased with increasing load and finally reached 0.021 at a normal load of 6 N. These results indicate that the optimal normal load range is 2–4 N, in which superlubricity can be achieved with  $\text{Mo}_2\text{CT}_x$ -IL. When the load was too low or too high, superlubricity could not be achieved, which may be because the tribofilms could not be completely formed with an extremely low load, and the tribofilms were damaged with a very high load. Compared with the corresponding contact pressures after the friction test, high contact pressures ( $> 900$  MPa) could be achieved when the normal load was in the range of 3–5 N, and the maximal contact pressure for superlubricity could reach 1.42 GPa at a normal load of 4 N (Fig. S6 in the ESM).

To verify the excellent lubrication performance of the  $\text{Mo}_2\text{CT}_x$ -IL, two other typical 2D materials (GO and  $\text{MoS}_2$ ) were selected and mixed with IL at the same volume concentration of  $\text{Mo}_2\text{CT}_x$ , referred to as GO-IL and  $\text{MoS}_2$ -IL, respectively. The COF results in Fig. 3(f) were taken as the average values in the stable period, showing that a COF of 0.005 could be achieved by using GO-IL as a lubricant, while the COF would increase to 0.011 by using  $\text{MoS}_2$ -IL as a lubricant. The corresponding wear scars after lubrication by GO-IL and  $\text{MoS}_2$ -IL were 82.74 and 71.8  $\mu\text{m}$ , respectively (Fig. S7 in the ESM), corresponding the contact pressures of 558.23 and 741.31 MPa, respectively. Although the GO-IL could achieve superlubricity, the wear scar was much larger than those of  $\text{Mo}_2\text{CT}_x$ -IL and  $\text{MoS}_2$ -IL, corresponding to a significantly lower contact pressure. This result confirms that molybdenum has much better antiwear ability than graphene.

### 3.3 Analysis of wear interface

After the friction test, the worn-out zone on the  $\text{Si}_3\text{N}_4$  surface was examined using SEM (Fig. 4(a)). Some adsorbed substances can be observed on the  $\text{Si}_3\text{N}_4$  surface in the worn-out zone after lubrication by  $\text{Mo}_2\text{CT}_x$ -IL. To analyze the composition of this area, energy-dispersive X-ray spectroscopy (EDS) mapping was performed in the worn-out zone (Figs. 4(b) and 4(c)). Mo and P were detected on the  $\text{Si}_3\text{N}_4$  surface based on the strong contrast signals in the EDS diagram. The wear track on the sapphire surface was

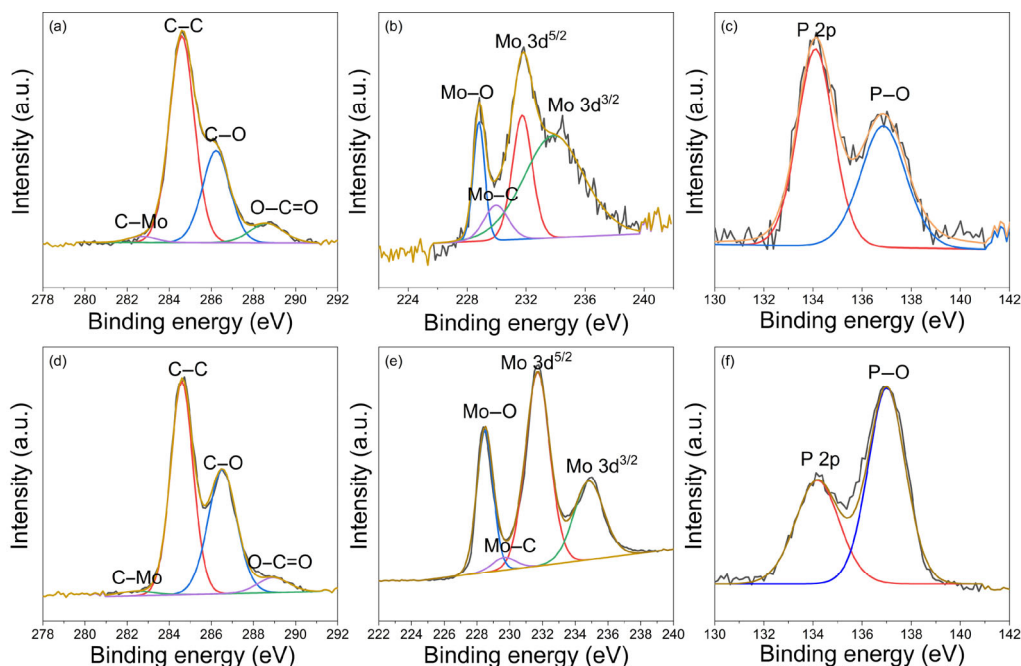


**Fig. 4** Analyses of the wear scar on the  $\text{Si}_3\text{N}_4$  and sapphire surface after lubrication by  $\text{Mo}_2\text{CT}_x$ -IL with a normal load of 3 N and a sliding speed of 0.1 m/s. (a) SEM image of morphology on the  $\text{Si}_3\text{N}_4$  surface. (b, c) EDS elemental mapping of the worn-out zone on the  $\text{Si}_3\text{N}_4$  surface. The elements are Mo and P, respectively. (d) SEM image of morphology on the sapphire surface. (e, f) EDS elemental mapping of the worn-out zone on the sapphire surface. The elements are Mo and P, respectively.

also examined using SEM with EDS (Figs. 4(d)–4(f)), in which adsorbed nanoparticles were observed in the wear track. The EDS results also showed a strong contrast in the Mo and P elements in the wear track. According to these results, it can be inferred that

there may be  $\text{Mo}_2\text{CT}_x$  MXene nanoparticles and IL molecules adsorbed in the worn-out zone through physical adsorption or tribochemical reaction during the friction test.

To determine the change in chemical composition on the friction surfaces and possible tribochemical reaction in the friction process, the worn-out zones on the  $\text{Si}_3\text{N}_4$  and sapphire surfaces after lubrication by  $\text{Mo}_2\text{CT}_x$ -IL were also characterized by XPS (Fig. 5 and Fig. S8 in the ESM). On the worn-out surface of the  $\text{Si}_3\text{N}_4$ , the main element of C 1s contained four main peaks located at 283.0, 284.8, 286.6, and 288.4 eV (Fig. 5(a)), corresponding to C–Mo, carbon in aliphatic chains (C–C), C–O, and O–C=O bonds, respectively. The detected C–O bond may be derived from the IL or oxidation of the  $\text{Mo}_2\text{CT}_x$  MXene nanoparticles. The existence of the C–Mo and O–C=O bonds mainly originated from the  $\text{Mo}_2\text{CT}_x$  MXene (Fig. 2(d)), proving that there were  $\text{Mo}_2\text{CT}_x$  nanoparticles adsorbed on the  $\text{Si}_3\text{N}_4$  surface. For the Mo element (Fig. 5(b)), there were four main peaks located at 228.6, 230.0, 232.1, and 234.4 eV. These peaks were assigned to Mo–O, Mo–C, Mo  $3d^{5/2}$ , and Mo  $3d^{3/2}$ , respectively. In comparison with the original XPS results of  $\text{Mo}_2\text{CT}_x$  MXene in Fig. 2(c), the existence of the Mo–O bond



**Fig. 5** XPS spectra of basic chemical elements on the worn-out zone of  $\text{Si}_3\text{N}_4$  and sapphire surfaces after lubrication by  $\text{Mo}_2\text{CT}_x$ -IL. (a–c) C 1s, Mo 3d, and P 2p on the worn-out zone of the  $\text{Si}_3\text{N}_4$  ball, respectively; (d–f) C 1s, Mo 3d, and P 2p on the worn-out zone of the sapphire surface, respectively.

showed that some  $\text{Mo}_2\text{CT}_x$  MXene participated in the oxidation reaction and produced molybdenum oxides ( $\text{Mo}_x\text{O}_y$ ) in the tribofilm. The production of  $\text{Mo}_x\text{O}_y$  improved the antiwear ability due to the molybdenum oxide, which is a wear-resistant material [45, 46]. For the P 2p, strong peaks were located at 134.5 and 137.3 eV (Fig. 5(c)), which were assigned to P 2p and P–O bonds, respectively. This indicates that phosphorus oxides ( $\text{P}_x\text{O}_y$ ) were produced on the worn-out surface of  $\text{Si}_3\text{N}_4$ . There were also two peaks of F 1s located at 684.6 and 687.8 eV (Fig. S8(a) in the ESM), which corresponded to fluorides and C–F bonds, respectively [47]. This result was consistent with Ref. [11], in which the anion of  $\text{PF}_6^-$  in IL was transformed into fluorides and phosphates by friction. For the Si 2p, Si–N (99.5 eV) and Si–O (101.7 eV) bonds were detected (Fig. S8(b) in the ESM), and the existence of the Si–O bond indicates the presence of a tribochemical reaction between  $\text{Si}_3\text{N}_4$  and water, leading to the formation of a silica layer.

At the same time, a worn-out zone was also detected by XPS on sapphire after lubrication by  $\text{Mo}_2\text{CT}_x$ -IL, and it displayed peaks similar to those of the  $\text{Si}_3\text{N}_4$  surface (Figs. 5(d)–5(f)). The C 1s contained four main peaks located at 282.9 eV (C–Mo), 284.8 eV (C–C), 286.6 eV (C–O), and 288.4 eV (O–C=O), as shown in Fig. 5(d). The detected Mo–O and P–O bonds confirm the production of  $\text{Mo}_x\text{O}_y$  and  $\text{P}_x\text{O}_y$  in the tribofilm on the sapphire surface by a tribochemical reaction. In addition, there were two F 1s peaks located at 684.2 and 685.9 eV on the sapphire surface (Fig. S8(c) in the ESM), which arose from the fluorides and  $\text{AlF}_6^{3-}$ , respectively. The XPS results of Al 2p also prove the generation of Al–F bonds, indicating a tribochemical reaction between  $\text{Al}_2\text{O}_3$  and  $\text{PF}_6^-$  anions (Fig. S8(d) in the ESM). Combining these XPS results and the EDS mapping in Fig. 4 indicates that there were tribofilms formed in the worn-out zone on the  $\text{Si}_3\text{N}_4$  and sapphire surfaces, which were mainly composed of  $\text{Mo}_x\text{O}_y$ ,  $\text{P}_x\text{O}_y$ , and fluorides.

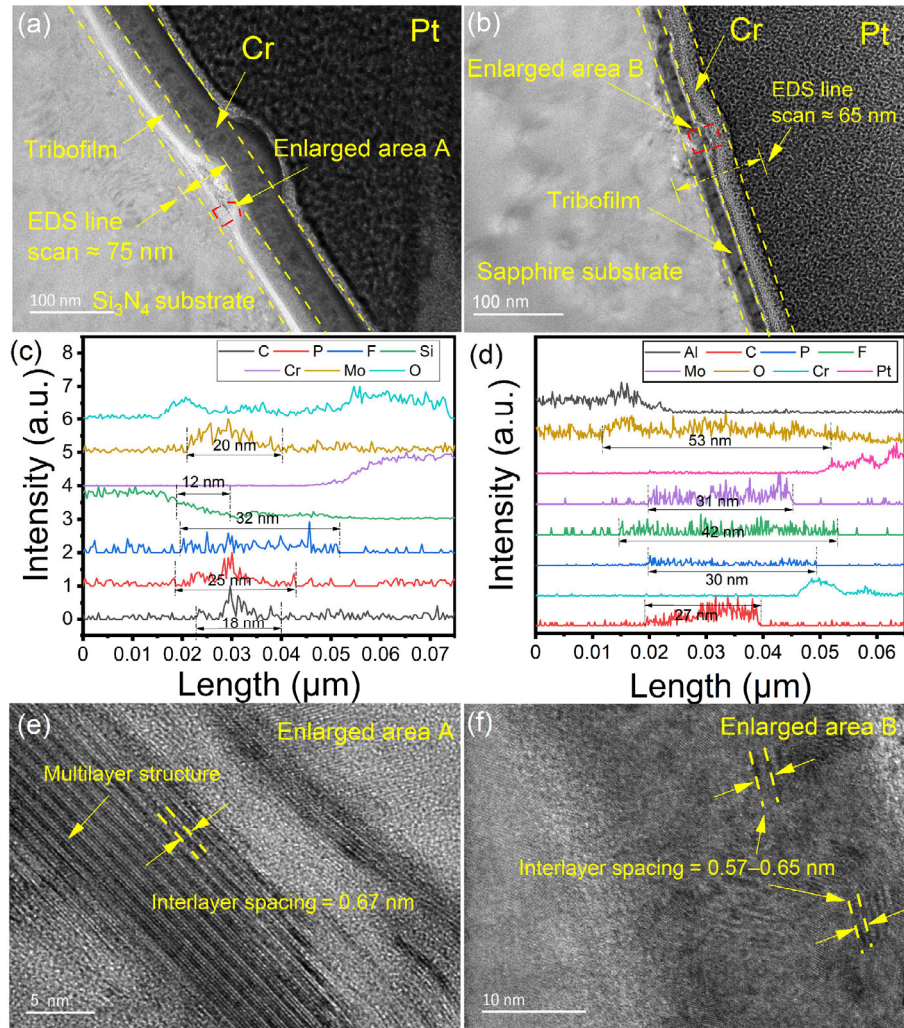
To reveal the structure of the tribofilm further, cross-sectional layers of the worn-out zone on the  $\text{Si}_3\text{N}_4$  and sapphire surfaces were prepared using a focused ion beam after lubrication with  $\text{Mo}_2\text{CT}_x$ -IL (the sampling regions are marked in Fig. S9 in the ESM). HRTEM was then conducted on the cross-sectional samples of the tribofilm in the contact

zone (Fig. 6). The thickness of the tribofilm on the  $\text{Si}_3\text{N}_4$  surface was approximately 45.6 nm (Fig. 6(a)). Furthermore, EDS in line scan mode was performed to analyze the elemental distribution and determine the composition of the tribofilm (Fig. 6(c)). Along with the elemental distributions of P, Mo, and O, the production of  $\text{P}_x\text{O}_y$  and  $\text{Mo}_x\text{O}_y$  occurred in the tribofilm with approximate thicknesses of 25 and 20 nm, respectively. The formation of  $\text{Mo}_x\text{O}_y$  and  $\text{P}_x\text{O}_y$  confirms that the tribochemical reaction occurred for the  $\text{Mo}_2\text{CT}_x$  MXene nanoparticles and IL on the  $\text{Si}_3\text{N}_4$  surface, in accordance with the XPS results in Fig. 5. There was an overlapping region for  $\text{Mo}_x\text{O}_y$  and  $\text{P}_x\text{O}_y$  production, indicating that the tribofilm had a composite structure.

Figure 6(c) shows that the  $\text{Si}_3\text{N}_4$  surface also participated in the tribochemical reaction ( $\text{Si}_3\text{N}_4 + 6\text{H}_2\text{O} \rightarrow 3\text{SiO}_2 + 4\text{NH}_3$ ), leading to the formation of silica on the  $\text{Si}_3\text{N}_4$  surface with an approximate thickness of 12 nm. Additionally, the structure of  $\text{Mo}_2\text{CT}_x$  MXene in the tribofilm was clearly observed in the enlarged HRTEM image (area A in Fig. 6(e)), which showed a multilayer structure with an interlayer spacing of approximately 0.7 nm. The detected layers of  $\text{Mo}_2\text{CT}_x$  MXene were parallel to the sliding direction, which had a beneficial effect on providing low shear strength. Raman detection was performed on the worn-out zone after the friction test (Fig. S10 in the ESM), and the Raman spectra of the worn-out zone of the  $\text{Si}_3\text{N}_4$  surface (Fig. S10(a) in the ESM) exhibited the same bands as the original Raman results in Fig. S1(b) in the ESM, further demonstrating that  $\text{Mo}_2\text{CT}_x$  MXene nanoparticles were adsorbed on the worn-out zone of the  $\text{Si}_3\text{N}_4$  surface.

On the sapphire surface, a tribofilm with a thickness of 57.2 nm was also observed (Fig. 6(b)). The EDS results show that  $\text{Mo}_x\text{O}_y$  and  $\text{P}_x\text{O}_y$  were produced in the tribofilm on the sapphire surface with thicknesses of 31 and 30 nm, respectively (Fig. 6(d)). The higher-magnification HRTEM image (area B) displayed some layer structures with an interlayer spacing of 0.57–0.65 nm in the tribofilm, which corresponded to the  $\text{Mo}_2\text{CT}_x$  nanoparticles. The inherent thickness of  $\text{Mo}_2\text{C}$  is approximately 5 Å [39], and the increase in the interlayer spacing is caused by surface termination or possible intercalated species. The Raman results also prove that the  $\text{Mo}_2\text{CT}_x$  MXene nanoparticles were





**Fig. 6** Cross-sectional structure of the tribofilm on the friction surface after friction test: (a) HRTEM images of the tribofilm formed on the  $\text{Si}_3\text{N}_4$  surface; (b) HRTEM images of the tribofilm formed on the sapphire surface; (c) elemental distributions of Si, P, O, Mo, F, Cr, and C along the cross-sectional profile in the enlarged area of Fig. 6(a) on the  $\text{Si}_3\text{N}_4$  surface; (d) elemental distributions of Al, C, P, F, Mo, O, Cr, and Pt along the cross-sectional profile in the enlarged area of Fig. 6(b) on the sapphire surface; (e) enlarged view of HRTEM image of the tribofilm (area A) formed on the  $\text{Si}_3\text{N}_4$  surface; and (f) enlarged view of HRTEM image of the tribofilm (area B) formed on the sapphire surface.

adsorbed on the sapphire surface (Fig. S10(b) in the ESM).

### 3.4 Mechanism of ultrahigh contact pressure under superlubricity

According to previous studies, it is difficult to achieve superlubricity under a contact pressure of more than 600 MPa (in the absence of coatings). However, in this work, the extended pressure limit of superlubricity was on the order of gigapascals. This implies a specific lubrication mechanism rather than the traditional elastohydrodynamic lubrication when the  $\text{Mo}_2\text{CT}_x$

MXene nanoparticles were used as lubricant additives, which was beneficial for achieving superlubricity under ultrahigh pressures. To determine the lubrication regime in the superlubricity state, the thickness ( $h_c$ ) of the lubricating film at the  $\text{Si}_3\text{N}_4$ –sapphire interface was estimated using the Hamrock–Dowson formula [48]. The calculation in Section S12 in the ESM showed that the liquid film thickness in the superlubricity state was in the range of 41.95–45.67 nm (45.67 nm for 2 N, 43.42 nm for 3 N, and 41.95 nm for 4 N, respectively). The ratio of the liquid film thickness to the equivalent roughness of the two worn-out zones

( $\lambda$ ) was used to classify the lubrication regime, which was calculated using the equation of  $\lambda = \frac{h}{\sqrt{\sigma_1^2 + \sigma_2^2}}$ ,

where  $\sigma$  is the surface roughness (23.2 nm for balls and 18.9 nm for the plate). The calculation yields  $\lambda = 1.41$ – $1.52$ , which implies that the lubrication regime was located in the mixed lubrication regime (containing the boundary and elastohydrodynamic lubrication).

For the running-in period, the COF oscillated unsteadily at 0.09, and slight wear occurred on the  $\text{Si}_3\text{N}_4$  and sapphire surfaces with lubrication by  $\text{Mo}_2\text{CT}_x$ -IL. Although the COF value was larger than that with IL lubrication during the running-in period, the wear was significantly lower than that with IL lubrication. This phenomenon demonstrates that  $\text{Mo}_2\text{CT}_x$  MXene nanoparticles play a dominant role in the inhibition of wear during the running-in period. It is speculated that the  $\text{Mo}_2\text{CT}_x$  MXene nanoparticles are adsorbed in the worn-out zone of the friction pairs from the beginning of the friction test, which can prevent direct contact between the  $\text{Si}_3\text{N}_4$  and sapphire surface. To confirm that the  $\text{Mo}_2\text{CT}_x$  MXene nanoparticles were adsorbed on the friction surfaces within a very short period, XPS was performed on the contact zone of the  $\text{Si}_3\text{N}_4$  surface after lubrication by  $\text{Mo}_2\text{CT}_x$ -IL for running-in periods of 1 and 3 min. The XPS results show a strong Mo 3d and C 1s signal in the contact zone of the  $\text{Si}_3\text{N}_4$  surface (Fig. S11 in the ESM), which are typical peaks of  $\text{Mo}_2\text{CT}_x$  MXene nanoparticles (Figs. 2(c) and 2(d)). These results illustrate the excellent adsorption of  $\text{Mo}_2\text{CT}_x$  MXene nanoparticles from the beginning of the friction test, which prevented the friction pairs from direct contact and thereby suppressed surface wear. Furthermore, the existence of the Mo–O bond during the running-in period indicates that  $\text{Mo}_2\text{CT}_x$ -IL participated in the tribochemical reaction from the beginning of the test. A protective film containing  $\text{Mo}_x\text{O}_y$  formed, which could further decrease abrasion owing to the excellent antiwear property of  $\text{Mo}_x\text{O}_y$  [49]. Compared with other 2D materials, such as GO and  $\text{MoS}_2$ ,  $\text{Mo}_2\text{CT}_x$  MXene nanoparticles exhibited a much better reduction in wear (Fig. S7 in the ESM). Therefore, the antiwear properties of  $\text{Mo}_2\text{CT}_x$  MXene nanoparticles do not only depend on its layer structure, but also are related on the antiwear properties of the  $\text{Mo}_2\text{CT}_x$  MXene nanoparticles themselves, which are not presented in

the traditional 2D materials, such as GO and  $\text{MoS}_2$ . After the running-in period was complete, the COF was directly reduced from 0.09 to 0.004, entering the superlubricity regime.

During the superlubricity period, the lubrication state entered the mixed lubrication regime. A composite structure of tribofilms containing  $\text{Mo}_x\text{O}_y$ ,  $\text{P}_x\text{O}_y$ , silica, and fluorides was formed on the friction surfaces by tribochemical reactions, which can provide extremely low shear strength for the boundary lubrication. Here, the ultralow shear strength occurred for three main reasons. First,  $\text{Mo}_x\text{O}_y$  and  $\text{P}_x\text{O}_y$  containing –OH groups are favorable for combining with water [50, 51]; thus, a water layer can form through double hydrogen bonding between the –OH groups and water molecules, which could reduce the shear strength of the tribofilm. Second, the silica colloid layer could promote the boundary lubrication because it was an excellent boundary lubricating material in a water environment [52], which could further reduce the shear strength of the tribofilm. Third, the existence of  $\text{Mo}_2\text{CT}_x$  MXene nanoparticles in the tribofilm also provided low shear strength, which assisted in reducing friction and wear. Therefore, the COF of the boundary lubrication was very low owing to the combined action of these three principal factors.

For the elastohydrodynamic lubrication, the viscosity of  $\text{Mo}_2\text{CT}_x$ -IL increased from 1.26 to 14.36 mPa·s after the running-in period, which was beneficial for the formation of the elastohydrodynamic lubricating film. Consequently, under the synergistic effect of the boundary lubrication and elastohydrodynamic lubrication, an ultralow COF under an ultrahigh contact pressure was achieved. The results and analysis demonstrate that the introduction of  $\text{Mo}_2\text{CT}_x$  MXene additives into IL can effectively enhance the lubrication performance and suppress surface wear, which further enhances the contact pressure for superlubricity (exceeding 1 GPa). This magnitude of improvement indicates that ultrahigh load capacity and ultralow friction can be achieved simultaneously by MXene-liquid coupling lubrication. The outstanding friction performance of  $\text{Mo}_2\text{CT}_x$  MXene nanoparticles expands the feasibility of MXene materials in the field of lubrication and their potential application in the mechanical industry. Therefore, the synergistic lubrication effect between  $\text{Mo}_2\text{CT}_x$  MXene nanoparticles

and ILs has important implications for the development of the liquid lubrication with superlubricity and super-wear-resistance performance.

## 4 Conclusions

Robust superlubricity can be achieved using  $\text{Mo}_2\text{CT}_x$  MXene nanoparticles in combination with IL at the  $\text{Si}_3\text{N}_4$ -sapphire interface. In this study, the upper limit of the contact pressure for the liquid superlubricity state was greatly improved to 1.42 GPa at the macroscale. The formation of the composite tribofilms induced by the tribochemical reaction, mainly containing  $\text{Mo}_x\text{O}_y$ ,  $\text{P}_x\text{O}_y$ , and adsorbed  $\text{Mo}_2\text{CT}_x$  MXene, suppressed the surface wear and provided an extremely low shear strength, which is the origin of the superlubricity state under ultrahigh contact pressure. This work provides a new method for achieving macroscale superlubricity under ultrahigh contact pressure through the synergistic effect of MXene and liquid molecules. The method has great potential for the application of superlubricity in mechanical systems.

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## Declaration of competing interest

The authors have no competing interests to declare that are relevant to the content of this article.

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