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Published by American Chemical Society. Copyright © American Chemical Society. However, no copyright claim is made to original U.S. Government works, or works produced by employees of any Commonwealth realm Crown government in the course of their duties. Vibrational Sum-Frequency Spectroscopic Investigation of the Structure and Azimuthal Anisotropy of Propynyl-Terminated Si(111) Surfaces

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ABSTRACT

Vibrational sum-frequency generation (VSFG) spectroscopy was used to investigate the orientation and azimuthal anisotropy of the C-H stretching modes for propynyl-terminated Si(111) surfaces, Si-C=C-CH₃. VSFG spectra revealed symmetric and asymmetric C-H stretching modes in addition to a Fermi resonance mode resulting from the interaction of the asymmetric C-H bending overtone with the symmetric C-H stretching vibration. The polarization dependence of the C-H stretching modes was consistent with the propynyl groups oriented such that the Si–C=C bond is normal to the Si(111) surface. The azimuthal angle dependence of the resonant C-H stretching amplitude revealed no rotational anisotropy for the symmetric C-H stretching mode and a 3-fold rotational anisotropy for the asymmetric C-H stretching mode in registry with the 3-fold symmetric Si(111) substrate. The results are consistent with expectation that the C–H stretching modes of a –CH₃ group are decoupled from the Si substrate due to a $-C \equiv C -$ spacer. In contrast, the methyl-terminated Si(111) surface, Si-CH₃, was previously reported to have pronounced vibronic coupling of the methyl stretch modes to the electronic bath of bulk Si. Vacuum-annealing of propynyl-terminated Si(111) resulted in increased 3-fold azimuthal anisotropy for the symmetric stretch, suggesting that removal of propynyl groups from the surface upon annealing allowed the remaining propynyl groups to tilt away from the surface normal, into one of three preferred directions towards the vacated neighbor sites.

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I. INTRODUCTION

Chemical functionalization of inorganic semiconductor surfaces has been used to tune the interfacial dipole, relative band-edge positions, manipulate surface-state densities, improve chemical stability, and alter the optoelectronic properties of the resulting materials.¹⁻⁸ The orientation and conformation of chemisorbed or physisorbed molecules can contribute to the electronic coupling between the bound adsorbate and semiconductor material, thereby regulating on a molecular scale interfacial electron transfer, charge injection and extraction, charge transport, and electron tunneling at the interface.⁹⁻¹⁰ The energetics and coupling at the interface can be instrumental in controlling adsorbate-substrate energy transfer and in influencing reaction rates and mechanisms at the interface. A molecular-level understanding of functionalized semiconductor surfaces is thus necessary to elucidate interfacial chemistry and adsorbate-semiconductor interactions, which in turn can be used to inform device engineering.

Covalent attachment of organic moieties to oxide-free crystalline Si surfaces, while preserving the ideal electronic properties of H-terminated Si surfaces (H–Si(111)), would be an attractive route for the fabrication of highly passivated Si surfaces for applications in micro- and nanoelectronics,^{6, 11} photonics, solar-energy conversion,¹² and chemical and biological sensors.¹³⁻ ¹⁴ Methyl termination of Si(111) surfaces (CH₃–Si(111)) by a wet-chemical process yields atomically flat Si with all surface Si sites terminated.¹⁵⁻¹⁸ Although CH₃–Si(111) surfaces exhibit enhanced resistance to air oxidation compared to H–Si(111) surfaces,¹⁹⁻²⁰ –CH₃ groups are limited by a lack of methods to impart secondary functionalization.

Functionalization of the Si(111) surfaces with alkynyl moieties, such as ethynyl and propynyl groups, has the advantage of allowing for elaboration through the unsaturated -C=C- units while still, in principle, maintaining nearly full Si–C surface termination.^{4, 21-24} Propynyl-

terminated Si(111) surfaces (CH₃CC–Si(111)) exhibit nearly complete termination of the Si(111) surface while maintaining a relatively low, air-stable, surface recombination velocity.²⁴⁻³⁰ Previous characterization of CH₃CC–Si(111) surfaces has also indicated that the CH₃CC– groups are oriented normal to the surface by infrared spectroscopy, the surfaces exhibit a (1×1) surface unit cell by low-energy electron diffraction, and the surfaces exhibit broad atomic terraces by atomic-force microscopy.²⁴ However, a full understanding of the chemical and electronic structures of Si surfaces functionalized with alkynyl moieties is still lacking, owing to the paucity of surface-sensitive spectroscopic techniques suitable for the detailed characterization of molecular monolayers.

Vibrational sum-frequency-generation (VSFG) spectroscopy provides surface selectivity and submonolayer sensitivity for in-situ investigations of the molecular structure and dynamics at surfaces.³¹⁻³⁴ VSFG spectroscopy is a second-order nonlinear spectroscopic technique that, under the electric dipole approximation, occurs only in the region of the material in which the inversion symmetry is broken, specifically at the interface. The vibrational signatures of an adsorbate in the mid-IR region are sensitive to the local environment, so VSFG spectroscopy can be used in unraveling the adsorbate-adsorbate and adsorbate-substrate interactions.³⁵⁻⁴⁰ The input and output beam polarizations of VSFG can be independently controlled to produce information about the molecular orientation of the surface species.⁴¹

The molecular structure and rotational dynamics of CH_3 –Si(111) surfaces has been elucidated using polarization-selected VSFG spectroscopy.⁴²⁻⁴³ The results indicated a strong electronic interaction between the –CH₃ vibrational modes and the Si surface, but the extent to which this coupling interaction is influenced by the proximity of the –CH₃ group to the Si(111) surface has not been investigated. While both CH_3CC –Si(111) and CH_3 –Si(111) surfaces have

vibrational signatures arising from –CH₃ groups, the distance between the –CH₃ groups and the Si substrate is larger for CH₃CC–Si(111) samples, allowing one to probe the dependence of adsorbate-substrate interactions on the distance between the terminal –CH₃ group and the Si(111) surface. The work reported herein describes a polarization-dependent VSFG study of the propynyl-terminated Si(111) surface, to elucidate the surface structure and the adsorbate-substrate interaction of this system.

II. EXPERIMENTAL

II.A. Materials and Methods. Water with a resistivity of $\geq 18.2 \text{ M}\Omega$ cm was obtained from a Barnstead E-Pure system. Ammonium fluoride (NH₄F(aq), 40%, semiconductor grade, Transene Co., Inc., Danvers, MA) was purged with Ar(g) (99.999%, Air Liquide) for 1 h prior to use. All other chemicals were used as received. Czochralski-grown n-type Si wafers (Virginia Semiconductor, Fredericksburg, VA) were double-side polished, doped with phosphorus to a resistivity of 1 Ω cm, 381 ± 25 µm thick, and oriented to within 0.1° of the (111) crystal plane.

II.A.1. Preparation of H–Si(111) Surfaces. The n-Si wafers were cut into 1 cm × 4 cm pieces and rinsed sequentially with water, methanol (\geq 99.8%, BDH), acetone (\geq 99.5%, BDH), methanol, and water. The wafers were oxidized and organic contaminants were removed by immersing the wafers for 10 min in a piranha solution (1:3 v/v of 30% H₂O₂(aq) (EMD): 18 M H₂SO₄ (EMD)) at 95 °C. The wafers were removed and then rinsed with copious amounts of water. The oxide was removed by immersing the wafers in buffered hydrofluoric acid (HF(aq), Transene Co. Inc., Danvers, MA) for 18 s, rinsing with water, and immediately placing the wafers for 9.0 min in an Ar(g)-purged solution of NH₄F(aq).^{24, 44-45} The wafers were agitated at

the start of each minute of etching, to remove bubbles that formed on the surface. The samples were then removed from the etching solution, rinsed briefly with water, and dried under Ar(g).

II.A.2. Preparation of Cl–Si(111) Surfaces. The H–Si(111) samples were transferred to a N₂(g)-purged glove box with <10 ppm O₂(g). An initiating amount (<1 mg mL⁻¹) of benzoyl peroxide 98%, Sigma-Aldrich) was added to a saturated solution of PCl₅ (99.998% metal basis, Alfa Aesar) in chlorobenzene (anhydrous, 99.8%, Sigma-Aldrich). The wafers were rinsed with chlorobenzene and immersed in the PCl₅ solution at 90 \pm 2 °C for 45 min.^{24, 45-46} The wafers were removed from the reaction and rinsed with chlorobenzene followed by hexanes (anhydrous, mixture of isomers, 99%, Sigma-Aldrich).

II.A.3. Preparation of Propynyl-Terminated Si(111) Surfaces. The Cl–Si(111) surfaces were immersed in a 1.0 M solution of 1-propynyllithium (CH₃CCLi, BOC Sciences, Shirley, NY) at 45 \pm 2 °C for 15 h inside foil-wrapped test tubes. The wafers were removed from the solution and rinsed with hexanes, followed by methanol (anhydrous, 99.8%, Sigma-Aldrich), submerged in methanol, and removed from the glove box. The samples were sonicated for 10 min in methanol, rinsed with water, and dried under Ar(g). The samples were broken into 1 cm × 1 cm squares, rinsed again with water, dried with Ar(g), and sealed under Ar(g) inside polypropylene centrifuge tubes, to enable transport from Caltech to the University of Southern California. X-ray photoelectron spectroscopy of the samples after preparation was consistent with results reported previously,²⁴ indicating complete termination of the surface by CH₃CC–groups and the absence of detectable levels of silicon oxide.

II.B. VSFG Spectroscopy. A detailed description of the VSFG apparatus can be found elsewhere.⁴² In brief, the femtosecond mid-infrared and picosecond visible pulses used in the broadband SFG spectroscopy setup were generated by splitting the 4 W output of a

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regeneratively amplified Ti:sapphire laser system (Spectra Physics Spitfire seeded by a KM Laboratories Oscillator) operating at 1 kHz repetition rate. The system delivered <100 fs pulses centered at ~800 nm. An optical parametric amplifier assembly (TOPAS-C, Light Conversion)-difference frequency generator (NDFG, Light Conversion) generated tunable femtosecond mid-IR pulses (fwhm 350 cm⁻¹). The narrowband visible pulses were generated from the uncompressed fundamental by passing the beam through an external compressor (Newport Corporation) and an etalon (TecOptics, fwhm 17 cm⁻¹).

To generate the SFG signal, the IR and visible pulses were spatially and temporally overlapped at the sample stage. SFG spectra were recorded using a triple-grating monochromator in conjunction with a CCD cooled by liquid nitrogen. SFG spectra were collected under an atmosphere of dry air obtained from a purge gas generator. To focus on the C–H stretching region of the CH₃CC–Si(111) samples, the IR pulse was centered at 3000 cm⁻¹. The beam was focused on the sample using a 25 cm focal length CaF₂ lens for the IR beam and a 45 cm focal length BK7 lens for the visible beam to yield a 200 μ m diameter spot. The intensities of the visible and the IR beams were 6–7 mW and 6–9 mW, respectively. Prior to VSFG measurements, sample annealing was performed to remove surface impurities for 16–20 h under ultrahigh vacuum at the temperatures ranging from 200–320 °C (see text).

III. RESULTS

III.A. VSFG and Anisotropy Studies of $CH_3CC-Si(111)$ Surfaces. Figure 1 presents SFG spectra of the C–H stretching region for PPP (SFG-visible-IR) and SSP polarization combinations, for as-prepared $CH_3CC-Si(111)$ samples at room temperature. A 270 fs time delay between the IR and the visible pulses was used to collect both the PPP and the SSP spectra, to suppress the nonresonant response of the silicon substrate.^{42-43, 47-48} The PPP spectrum showed three resonant vibrational modes of the terminal –CH₃ group, which are assignable to the symmetric C–H stretch (r⁺) at ~2865 cm⁻¹, the asymmetric C–H stretch (r⁻) at ~2962 cm⁻¹, and the Fermi resonance (d_{FR}^+) of the r⁺ mode with the C–H asymmetric bending overtone, at ~2933 cm⁻¹.^{24, 42, 49-51} The SSP spectrum, which exhibited a significant reduction in signal intensity compared with the PPP spectrum, only showed signatures ascribable to the symmetric C–H stretch (r⁺) at ~2863 cm⁻¹ and the Fermi resonance (d_{FR}^+) at ~2935 cm⁻¹.



Figure 1. PPP (red, left axis) and SSP (blue, right axis) polarized VSFG spectra of the CH₃CC– Si(111) surface for the C–H stretching region. The time delay between the IR and the visible pulses was 270 fs. The corresponding molecular motions are shown alongside the VSFG spectra. The molecular frame of reference (a, b, c) used in this work is indicated in the figure relative to the proposed orientation of the CH₃CC– groups on the surface.

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The azimuthal dependence of the resonant amplitudes of the r^+ and r^- modes of the CH₃CC–Si(111) samples was measured^{42, 52-54} for azimuthal angles ϕ from 0° to 360°, and are shown for PPP polarization conditions in Figure 2. The data points represent the resonant C–H stretching mode amplitudes, which were obtained from the fits of the PPP spectra collected by in-plane rotation of the sample at 30° intervals (Table S2, Supporting Information). The polar plot for the r^+ mode (Figure 2A) showed only small changes in the amplitude of the resonant C–H stretching signal with ϕ , whereas the plot for the r^- mode showed a pronounced 3-fold dependence on the azimuthal angle (Figure 2B).



Figure 2. Polar plots showing the azimuthal dependence of the resonant amplitude of the C–H symmetric stretch r^+ (A) and C–H asymmetric stretch r^- (B), respectively, of the PPP spectra for as-prepared CH₃CC–Si(111) samples. Data points were collected from 0° to 360° in 30° increments, and the blue solid lines are fits described in the text (Eq. (2)).

III.B. Effects of Annealing on the Adsorbate-Substrate Coupling Interaction. The stability of the CH₃CC–Si(111) samples toward annealing in vacuum was investigated by VSFG spectroscopy. Samples were heated to 200–320 °C for 16–20 h under vacuum, and were then

allowed to cool to room temperature for characterization by VSFG spectroscopy. The VSFG spectra for the annealed samples (Figure 3) exhibited the same set of SFG transitions that were seen for the as-prepared samples, with the r^+ and the d_{FR}^+ modes observed for both SSP and PPP polarizations, and the r^- mode only observed for the PPP polarization. After annealing, the peak intensities were reduced and the peak centers shifted by ~5–9 cm⁻¹ to higher energy relative to the spectra obtained before annealing of the sample.



Figure 3. PPP and SSP spectra of the C–H stretching region of $CH_3CC-Si(111)$ samples annealed at 320 °C for 16–20 h under vacuum. The decrease in intensity is consistent with the partial loss of the CH_3CC- units from the Si(111) surface upon annealing. The slight blue-shift of the peaks may be due to the changed local environment of the propynyl group remaining on the surface.

Figure 4 shows the dependence of the resonant amplitudes of the r^+ and r^- modes, respectively, of the annealed CH₃CC–Si(111) samples on the azimuthal angle ϕ for PPP polarization conditions. The fits of the azimuthal rotational anisotropy data, collected for inplane rotation intervals of 10° (Table S3, Supporting Information), are shown superimposed on the data points. After annealing, a 3-fold rotational anisotropy of the symmetric stretch r^+ mode was readily observed, in contrast to the behavior prior to annealing. The 3-fold rotational anisotropy exhibited by the r^- mode was slightly enhanced by the annealing step. The rotational anisotropy of the resonant amplitudes of the r^+ and r^- modes was generally more pronounced after annealing, indicating a change in the interaction between the CH₃CC– adsorbates and the Si substrate.



Figure 4. Polar plots showing the azimuthal dependence of the resonant amplitudes for PPP spectra of (A) the C–H symmetric stretch (A(r^+)) and (B) the C–H asymmetric stretch (A(r^-)) for CH₃CC–Si(111) samples annealed at 200 °C for 16–20 h under vacuum. Data points were collected from 0° to 360° in 10° increments, and the blue solid lines are fits described in the text (Eq. (2)).

IV. DISCUSSION

IV.A. VSFG Orientation Analysis of as-Prepared CH₃CC–Si(111) Samples. The sum frequency spectra of the CH₃CC–Si(111) samples were analyzed by macroscopic (ensemble) averaging of the resonant hyperpolarizability tensor elements ($\beta_{lmn}^{(2)}$, defined in the molecular frame (*a*, *b*, *c*)) that represent the methyl group vibrations (Figure 1).⁴¹ $\beta_{lmn}^{(2)}$ for the *i*-th vibrational mode is expressed as a product of the vibrational transition dipole moment and the Raman polarizability tensor:

$$\beta_{lmn,i}^{(2)} \propto \frac{\partial \alpha_{lm}}{\partial q_i} \frac{\partial \mu_n}{\partial q_i} \tag{1}$$

where q_i is the normal coordinate and the indices l, m, and n represent axes in the molecular frame of reference (a,b,c). Different vibrational modes have different symmetries, and the PPP and SSP polarization combinations sample different β tensor elements,⁵⁵⁻⁵⁷ which are also governed by the molecular symmetries. Hence, the absence of the r⁻ mode in the SSP spectrum provides a qualitative indication of the orientation of the CH₃CC– moieties perpendicular to the Si(111) surface.

The terminating CH₃CC– group can be approximately described as having C_{3v} symmetry, with *c* being the C₃ symmetry axis and the *ac* plane lying on one of the σ_v symmetry planes.^{55, 57} With respect to this molecule fixed coordinate system, the three C–H vibrational modes observed in the SFG spectra can be categorized into two different types of vibrations with different hyperpolarizability tensor elements.⁵⁸⁻⁵⁹ The r⁺ and the d⁺_{FR} modes account for changes in the transition dipole moments along the *c* axis (perpendicular to the Si(111) surface plane), whereas the r⁻ mode accounts for changes along a direction that is perpendicular to the *c* axis (parallel to

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the Si(111) surface plane). The corresponding β tensor elements for these modes can also be categorized into two groups: the totally symmetric r^+ mode corresponds to two isotropic independent β tensor elements, $\beta_{aac}^{(2)} = \beta_{bbc}^{(2)}$ and $\beta_{ccc}^{(2)}$, and the r^- mode corresponds to only one independent β tensor element $\beta_{aca}^{(2)} = \beta_{caa}^{(2)}$.

With the symmetries of the β tensor elements for the r⁺ and r⁻ modes taken into account, along with knowledge of the input/output polarization combinations, the SSP spectra (which sample tensor element $\chi_{_{3yx}}^{(2)}$ defined in the laboratory frame of reference (x, y, z)) would contain no peaks if the CH₃CC– units were all oriented parallel to the Si(111) surface plane. In contrast, if the CH₃CC– units were all oriented perpendicular to the Si(111) surface plane, the SSP spectrum would contain only peaks corresponding to vibrations perpendicular to the Si(111) surface plane. Because the SSP spectra in Figures 1 and 3 exhibit signals ascribed to the r⁺ and d⁺_{FR} modes, the data are thus consistent with the CH₃CC– groups being oriented perpendicular to the Si(111) surface. The PPP spectra in Figure 1 and Figure 3 contain peaks corresponding to both the parallel and perpendicular vibrations, as the spectrum is comprised of four different susceptibility tensor elements ($\chi_{xxx}^{(2)}$, $\chi_{xxx}^{(2)}$, $\chi_{xxx}^{(2)}$, and $\chi_{xxz}^{(2)}$) sampling molecular hyperpolarizabilities in all possible directions. The perpendicular orientation of the CH₃CC– groups with respect to the Si(111) surface plane is consistent with the conclusions from variableangle transmission IR data on CH₃CC–Si(111) samples.²⁴

IV.B. Azimuthal Dependence of the Resonant C–H Stretch Amplitudes for as-Prepared CH₃CC–Si(111) Samples. Both the adsorbate vibrational modes and the bulk Si crystal polarizability contribute to the effective second-order vibrational response from the functionalized Si(111) surface, as has been previously observed in VSFG studies of CH₃–Si(111) surfaces.⁴² The coupling between the adsorbate vibrational modes and electronic bath of bulk Si

crystal, therefore, contributes to the azimuthal behavior of the resonant vibrational r^+ and r^- modes for as-prepared CH₃CC–Si(111) samples. Each of these contributions is comprised of an isotropic component and an anisotropic component dependent on the crystal symmetry. For the Si(111) surface, which has 3-fold rotational symmetry, the generated sum frequency field, E^{SFG} , has a form:

$$E^{\rm SFG} = (A + C\cos(3\phi + \phi_0))E^{\rm vis}E^{\rm IR}$$
⁽²⁾

where A and C are the isotropic and anisotropic contributions to the response, respectively, ϕ is the azimuthal angle within the (111) plane with a phase correction of ϕ_0 , and E^{vis} and E^{IR} are the input visible and mid-IR fields, respectively.

The observed azimuthal dependence of the VSFG signals can be connected to the resonant vibrational modes via the unique β tensor elements that describe the –CH₃ group vibrations. The non-zero β tensor elements for the r⁺ mode are symmetric with respect to the C₃ axis ($\beta_{aac}^{(2)} = \beta_{bbc}^{(2)}$ and $\beta_{ccc}^{(2)}$) because the IR transition moment for the r⁺ mode is along the C₃ axis (which is along *c*), and the Raman polarizability tensor is isotropic about the symmetry axis. Therefore, if the C₃ axis of the terminal –CH₃ group is normal to the surface, as evidenced by the results reported herein, no azimuthal anisotropy for the r⁺ mode is expected regardless of the inplane orientation of the –CH₃ group. The lack of clear rotational anisotropy for the r⁺ mode under PPP polarization conditions (Figure 2A) is consistent with this expectation. In contrast, the β tensor elements for the r⁻ mode is in-plane and also expected to be anisotropic, as long as the molecules do not freely rotate in-plane. As a result, the r⁻ mode shows a 3-fold rotational anisotropy (Figure 2B) that reflects the 3-fold symmetry of the crystalline Si(111) surface. The

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3-fold anisotropy of the r^- mode also indicates that the CH₃CC– groups do not rotate freely at room temperature, but spend most of their time in one of the three energetic minima in registry with the Si(111) surface, analogous to previously reported VSFG results for CH₃–Si(111) samples.⁴² This is additionally consistent with steric considerations on CH₃–Si(111) or CH₃CC– Si(111) surfaces that predict the interlocking of adjacent –CH₃ on the surface.^{18, 60} This observation supports a close packing structure of the CH₃CC– groups on the Si(111) surface, suggesting that the surfaces are fully-terminated.

Previous VSFG studies on the CH₃–Si(111) surface showed a clear 3-fold azimuthal anisotropy for the r^+ mode,⁴² in contrast to the results reported herein, which showed no such anisotropy of the r^+ mode for as-prepared CH₃CC–Si(111) samples. Because symmetry considerations suggest that the r^+ mode should not exhibit rotational anisotropy, the results for CH₃–Si(111) surfaces were attributed to coupling between the adsorbate vibrational modes and the above-band-gap Raman polarizability of the Si bulk.⁴² In the CH₃–Si(111) system, the coupling between the C–H vibrational modes and the electronic structure of the Si bulk is strengthened by the close proximity of the –CH₃ group to the Si surface (the –CH₃ symmetric stretch mode involves displacement of the carbon atom, and thus likely the attached Si atom). Thus, the resultant Raman polarizability derivative for the –CH₃ group vibrations can be expressed as:

$$\frac{\partial \alpha_{lm}}{\partial q_i} = \frac{\partial \alpha_{Si}}{\partial q_i} + \frac{\partial \alpha_{Me}}{\partial q_i}$$
(3)

where α_{Si} and α_{Me} are the electronic polarizability of the Si substrate and the –CH₃ group, respectively. The polarizability scales with volume, so α_{Si} is expected to be much larger than α_{Me} , and the measured VSFG signal is dominated by the above-band-gap electronic response of the Si substrate. Hence the molecular hyperpolarizabilities for the CH₃–Si(111) surface that correspond to the r^+ and r^- modes show the same 3-fold anisotropy of the Si(111) bulk regardless of the orientations of their transition dipole moments with respect to the surface plane. The extent of the adsorbate-substrate coupling decreases as the –CH₃ groups are further removed from the Si(111) surface by the –C=C– spacer. The hyperpolarizability tensor elements for the r^+ mode observed for CH₃CC–Si(111) samples thus only reflect the totally symmetric Raman polarizability tensor elements of the C–H symmetric vibrations of the methyl group uncoupled from the Si substrate. The results reported herein suggest that the distance provided by the –C=C– group is sufficient to substantially suppress the vibronic interaction between the –CH₃ group vibrations and the Si bulk.

IV.C. Effects of Sample Annealing on the Structure of the CH₃CC–Si(111) Surface. The azimuthal anisotropy of the r^+ and r^- modes for CH₃CC–Si(111) was readily observed following sample annealing to 200–320 °C (Figure 4). The appearance of an azimuthal dependence for the r^+ mode and the increase in the observed anisotropy for the r^- mode can potentially be attributed to the loss of CH₃CC– groups from the surface following annealing. Previous thermal stability studies of CH₃CC–Si(111) surfaces in vacuum have indicated loss of CH₃CC– groups upon annealing to temperatures \geq 200 °C, with ~0.76 ML remaining after annealing for 30 min at 200 °C.²⁴ The decrease in SFG signal intensity of the C–H stretching modes upon annealing in vacuum (Figure 3) supports the conclusion that CH₃CC– groups are removed from the surface by this thermal process. The removal of a fraction of CH₃CC– units results in vacancies on the Si(111) surface, which, in an otherwise closely packed structure, could allow the remaining CH₃CC– groups adjacent to the defect sites to tilt away from the surface normal. Thus, a propynyl molecule after annealing could tilt in one of the three

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azimuthal directions towards its neighbors. Figure 5 depicts the proposed removal of a CH₃CC– unit by annealing and the subsequent reorganization of the surface structure.

As the direction of the C₃ symmetry axis of the CH₃CC– group tilts away from the surface normal, the β tensor elements for the r⁺ mode will have resolvable components parallel to the surface, and thus the SFG signal will exhibit 3-fold azimuthal anisotropy, as observed (Figure 4A). The observed decrease in SFG intensity is consistent with the partial loss of the CH₃CC– units from the Si(111) surface upon annealing, and with tilting of the molecular axis away from surface normal. The tilt of the C₃ symmetry axis away from the surface normal also brings the CH₃CC– units closer to the Si surface, which may result in an increased coupling between the CH₃CC– groups and the Si surface. The higher modulation depth of the r⁻ azimuthal response (Figure 4B) is also consistent with this tilting.



Figure 5. Annealing causes loss of CH_3CC- units from the Si(111) surface creating vacancy defects. CH_3CC- units adjacent to vacancies can tilt towards the Si(111) surface to relieve steric strain. As CH_3CC- units tilt toward the Si(111) surface plane, the terminal $-CH_3$ group acquires in-plane component of the 2nd order susceptibility, thus resulting in the 3-fold anisotropy observed in the VSDFG spectra.

V. CONCLUSIONS

The VSFG studies of the CH₃CC-Si(111) surface emphasize the importance of the molecular surface structure in determining the vibronic interactions between the semiconductor bulk and the terminating substituents. The observation of only vibrational modes with transition dipole moments oriented perpendicular to the surface plane for SSP polarization conditions provides evidence that the CH_3CC - groups are oriented perpendicular to the surface. The absence of azimuthal anisotropy observed for the r⁺ mode, in addition to the observed 3-fold azimuthal anisotropy for the r⁻ mode, is consistent with expectations for C-H stretching vibrations for a $-CH_3$ group is decoupled from the Si substrate due to the distance introduced by the $-C \equiv C$ - spacer group. This behavior stands in contrast to results observed for CH₃-Si(111) surfaces, which showed significant electronic coupling between C-H vibrational modes and the above-band-gap Raman polarizability of the Si bulk. The -C=C- group present in CH₃CC-Si(111) surfaces thus appears to effectively isolate the terminal $-CH_3$ group from the Si surface, decoupling the C-H vibrational modes from the bulk electronic structure of the Si. Annealed CH₃CC-Si(111) surfaces exhibited a more pronounced azimuthal anisotropy for the r^+ and $r^$ modes compared with as-prepared samples. Loss of CH₃CC- groups from the surface upon annealing and subsequent tilting of the remaining bound CH₃CC- groups towards the neighbor vacancies and away from the surface normal results in the in-plane component of the hyperpolarizability of the tilted $-CH_3$ group which leads to the observed 3-fold asymmetry.

ASSOCIATED CONTENT

Supporting Information

The supporting information is available free of charge on the ACS publications website at DOI:

Fitting parameters used to analyze the rotational anisotropy of the resonant amplitudes for PPP spectra before and after annealing.

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Notes

The authors declare no competing financial interest.

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REFERENCES

1. Ashkenasy, G.; Cahen, D.; Cohen, R.; Shanzer, A.; Vilan, A., Molecular Engineering of Semiconductor Surfaces and Devices. *Accounts of Chemical Research* **2002**, *35*, 121-128.

2. Filler, M. A.; Bent, S. F., The Surface as Molecular Reagent: Organic Chemistry at the Semiconductor Interface. *Progress in Surface Science* **2003**, *73*, 1-56.

3. Loscutoff, P. W.; Bent, S. F., Reactivity of the Germanium Surface: Chemical Passivation and Functionalization. *Annu Rev Phys Chem* **2006**, *57*, 467-495.

4. Maldonado, S.; Plass, K. E.; Knapp, D.; Lewis, N. S., Electrical Properties of Junctions between Hg and Si(111) Surfaces Functionalized with Short-Chain Alkyls. *The Journal of Physical Chemistry C* **2007**, *111*, 17690-17699.

Seitz, O.; Vilan, A.; Cohen, H.; Hwang, J.; Haeming, M.; Schoell, A.; Umbach, E.; Kahn,
 A.; Cahen, D., Doping Molecular Monolayers: Effects on Electrical Transport through Alkyl
 Chains on Silicon. *Adv Funct Mater* 2008, *18*, 2102-2113.

6. Vilan, A.; Yaffe, O.; Biller, A.; Salomon, A.; Kahn, A.; Cahen, D., Molecules on Si: Electronics with Chemistry. *Adv Mater* **2010**, *22*, 140-159.

7. Li, Y.; O'Leary, L. E.; Lewis, N. S.; Galli, G., Combined Theoretical and Experimental Study of Band-Edge Control of Si through Surface Functionalization. *The Journal of Physical Chemistry C* **2013**, *117*, 5188-5194.

8. Wong, K. T.; Lewis, N. S., What a Difference a Bond Makes: The Structural, Chemical, and Physical Properties of Methyl-Terminated Si(111) Surfaces. *Accounts of Chemical Research* **2014**, *47*, 3037-3044.

9. Patitsas, S. N.; Lopinski, G. P.; Hul'ko, O.; Moffatt, D. J.; Wolkow, R. A., Current-Induced Organic Molecule–Silicon Bond Breaking: Consequences for Molecular Devices. *Surf Sci* 2000, , L425-L431.

10. Tao, F. F.; Bernasek, S., *Functionalization of Semiconductor Surfaces*; John Wiley & Sons, Inc.: Hoboken, NJ, 2012.

11. Chen, Y.; Wang, X.; Erramilli, S.; Mohanty, P.; Kalinowski, A., Silicon-Based Nanoelectronic Field-Effect Ph Sensor with Local Gate Control. *Appl Phys Lett* **2006**, *89*, 223512.

12. Warren, E. L.; Atwater, H. A.; Lewis, N. S., Silicon Microwire Arrays for Solar Energy-Conversion Applications. *The Journal of Physical Chemistry C* **2014**, *118*, 747-759.

Touahir, L., et al., Molecular Monolayers on Silicon as Substrates for Biosensors.
 Bioelectrochemistry 2010, *80*, 17-25.

14. Kirk, J. T.; Fridley, G. E.; Chamberlain, J. W.; Christensen, E. D.; Hochberg, M.; Ratner,
D. M., Multiplexed Inkjet Functionalization of Silicon Photonic Biosensors. *Lab on a Chip* 2011,
11, 1372-1377.

The Journal of Physical Chemistry

15. Becker, J. S.; Brown, R. D.; Johansson, E.; Lewis, N. S.; Sibener, S. J., Helium Atom Diffraction Measurements of the Surface Structure and Vibrational Dynamics of Ch₃–Si(111) and Cd₃–Si(111) Surfaces. *The Journal of Chemical Physics* **2010**, *133*, 104705.

 Bansal, A.; Li, X.; Lauermann, I.; Lewis, N. S.; Yi, S. I.; Weinberg, W. H., Alkylation of Si Surfaces Using a Two-Step Halogenation/Grignard Route. *J Am Chem Soc* 1996, *118*, 7225-7226.

17. Bansal, A.; Lewis, N. S., Stabilization of Si Photoanodes in Aqueous Electrolytes through Surface Alkylation. *The Journal of Physical Chemistry B* **1998**, *102*, 4058-4060.

18. Yu, H.; Webb, L. J.; Ries, R. S.; Solares, S. D.; Goddard, W. A.; Heath, J. R.; Lewis, N.

S., Low-Temperature Stm Images of Methyl-Terminated Si(111) Surfaces. *The Journal of Physical Chemistry B* 2004, *109*, 671-674.

19. Royea, W. J.; Juang, A.; Lewis, N. S., Preparation of Air-Stable, Low Recombination Velocity Si(111) Surfaces through Alkyl Termination. *Appl. Phys. Lett.* **2000**, *77*, 1988-1990.

20. Webb, L. J.; Lewis, N. S., Comparison of the Electrical Properties and Chemical Stability of Crystalline Silicon(111) Surfaces Alkylated Using Grignard Reagents or Olefins with Lewis Acid Catalysts. *J. Phys. Chem. B* **2003**, *107*, 5404-5412.

21. Teyssot, A.; Fidélis, A.; Fellah, S.; Ozanam, F.; Chazalviel, J. N., Anodic Grafting of Organic Groups on the Silicon Surface. *Electrochimica Acta* **2002**, *47*, 2565-2571.

22. Hurley, P. T.; Nemanick, E. J.; Brunschwig, B. S.; Lewis, N. S., Covalent Attachment of Acetylene and Methylacetylene Functionality to Si(111) Surfaces: Scaffolds for Organic Surface Functionalization While Retaining Si–C Passivation of Si(111) Surface Sites. *J Am Chem Soc* **2006**, *128*, 9990-9991.

23. Rohde, R. D.; Agnew, H. D.; Yeo, W.-S.; Bailey, R. C.; Heath, J. R., A Non-Oxidative Approach toward Chemically and Electrochemically Functionalizing Si(111). *J Am Chem Soc* **2006**, *128*, 9518-9525.

24. Plymale, N. T.; Kim, Y.-G.; Soriaga, M. P.; Brunschwig, B. S.; Lewis, N. S., Synthesis, Characterization, and Reactivity of Ethynyl- and Propynyl-Terminated Si(111) Surfaces. *The Journal of Physical Chemistry C* **2015**, *119*, 19847-19862.

25. Puniredd, S. R.; Assad, O.; Haick, H., Highly Stable Organic Modification of Si(111) Surfaces: Towards Reacting Si with Further Functionalities While Preserving the Desirable Chemical Properties of Full Si–C Atop Site Terminations. *J Am Chem Soc* **2008**, *130*, 9184-9185.

26. Bashouti, M. Y.; Sardashti, K.; Schmitt, S. W.; Pietsch, M.; Ristein, J.; Haick, H.; Christiansen, S. H., Oxide-Free Hybrid Silicon Nanowires: From Fundamentals to Applied Nanotechnology. *Progress in Surface Science* **2013**, *88*, 39-60.

27. Soria, F. A.; Paredes-Olivera, P.; Patrito, E. M., Chemical Stability toward O₂ and H₂o of Si(111) Grafted with –Ch₃, –Ch₂ch₂ch₃, –Chchch₃, and –Ccch₃. *The Journal of Physical Chemistry C* **2015**, *119*, 284-295.

28. Puniredd, S. R.; Assad, O.; Haick, H., Highly Stable Organic Monolayers for Reacting Silicon with Further Functionalities: The Effect of the C–C Bond Nearest the Silicon Surface. *J Am Chem Soc* **2008**, *130*, 13727-13734.

29. Assad, O.; Puniredd, S. R.; Stelzner, T.; Christiansen, S.; Haick, H., Stable Scaffolds for Reacting Si Nanowires with Further Organic Functionalities While Preserving Si–C Passivation of Surface Sites. *J Am Chem Soc* **2008**, *130*, 17670-17671.

The Journal of Physical Chemistry

30. Qin, G.; Santos, C.; Zhang, W.; Li, Y.; Kumar, A.; Erasquin, U. J.; Liu, K.; Muradov, P.; Trautner, B. W.; Cai, C., Biofunctionalization on Alkylated Silicon Substrate Surfaces Via "Click" Chemistry. *J Am Chem Soc* **2010**, *132*, 16432-16441.

31. Zhu, X. D.; Suhr, H.; Shen, Y. R., Surface Vibrational Spectroscopy by Infrared-Visible Sum Frequency Generation. *Phys Rev B* **1987**, *35*, 3047-3050.

32. Richmond, G. L., Molecular Bonding and Interactions at Aqueous Surfaces as Probed by Vibrational Sum Frequency Spectroscopy. *Chem. Rev.* **2002**, *102*, 2693-2724.

33. Kim, J.; Somorjai, G. A., Molecular Packing of Lysozyme, Fibrinogen, and Bovine Serum Albumin on Hydrophilic and Hydrophobic Surfaces Studied by Infrared–Visible Sum Frequency Generation and Fluorescence Microscopy. *J Am Chem Soc* **2003**, *125*, 3150-3158.

34. Fu, L.; Liu, J.; Yan, E. C. Y., Chiral Sum Frequency Generation Spectroscopy for Characterizing Protein Secondary Structures at Interfaces. *J Am Chem Soc* 2011, *133*, 8094-8097.

35. Guyot-Sionnest, P.; Dumas, P.; Chabal, Y. J.; Higashi, G. S., Lifetime of an Adsorbate-Substrate Vibration: H on Si(111). *Phys Rev Lett* **1990**, *64*, 2156-2159.

36. Lin, S. H.; Villaeys, A. A., Theoretical Description of Steady-State Sum-Frequency Generation in Molecular Adsorbates. *Phys Rev A* **1994**, *50*, 5134-5144.

37. Caudano, Y.; Silien, C.; Humbert, C.; Dreesen, L.; Mani, A. A.; Peremans, A.; Thiry, P.
A., Electron–Phonon Couplings at C₆₀ Interfaces: A Case Study by Two-Color, Infrared–Visible
Sum–Frequency Generation Spectroscopy. *Journal of Electron Spectroscopy and Related Phenomena* 2003, *129*, 139-147.

38. Rupprechter, G.; Morkel, M.; Freund, H.-J.; Hirschl, R., Sum Frequency Generation and Density Functional Studies of Co–H Interaction and Hydrogen Bulk Dissolution on Pd(111). *Surf Sci* **2004**, *554*, 43-59.

39. Sovago, M.; Campen, R. K.; Wurpel, G. W. H.; Müller, M.; Bakker, H. J.; Bonn, M., Vibrational Response of Hydrogen-Bonded Interfacial Water Is Dominated by Intramolecular Coupling. *Phys Rev Lett* **2008**, *100*, 173901.

40. Stiopkin, I. V.; Weeraman, C.; Pieniazek, P. A.; Shalhout, F. Y.; Skinner, J. L.; Benderskii, A. V., Hydrogen Bonding at the Water Surface Revealed by Isotopic Dilution Spectroscopy. *Nature* **2011**, *474*, 192-195.

41. Wang, H.-F.; Velarde, L.; Gan, W.; Fu, L., Quantitative Sum-Frequency Generation Vibrational Spectroscopy of Molecular Surfaces and Interfaces: Lineshape, Polarization, and Orientation. *Annu Rev Phys Chem* **2015**, *66*, 189-216.

42. Malyk, S.; Shalhout, F. Y.; O'Leary, L. E.; Lewis, N. S.; Benderskii, A. V., Vibrational Sum Frequency Spectroscopic Investigation of the Azimuthal Anisotropy and Rotational Dynamics of Methyl-Terminated Silicon(111) Surfaces. *The Journal of Physical Chemistry C* **2012**, *117*, 935-944.

43. Shalhout, F. Y.; Malyk, S.; Benderskii, A. V., Relative Phase Change of Nearby Resonances in Temporally Delayed Sum Frequency Spectra. *J. Phys. Chem. Lett.* **2012**, *3*, 3493–3497.

44. Webb, L. J.; Rivillon, S.; Michalak, D. J.; Chabal, Y. J.; Lewis, N. S., Transmission Infrared Spectroscopy of Methyl- and Ethyl-Terminated Silicon(111) Surfaces. *J. Phys. Chem. B* **2006**, *110*, 7349-7356.

The Journal of Physical Chemistry

45. O'Leary, L. E.; Johansson, E.; Brunschwig, B. S.; Lewis, N. S., Synthesis and Characterization of Mixed Methyl/Allyl Monolayers on Si(111). *The Journal of Physical Chemistry B* 2010, *114*, 14298-14302.

46. Rivillon, S.; Chabal, Y. J.; Webb, L. J.; Michalak, D. J.; Lewis, N. S.; Halls, M. D.; Raghavachari, K., Chlorination of Hydrogen-Terminated Silicon (111) Surfaces. *Journal of Vacuum Science and Technology A* **2005**, *23*, 1100-1106.

47. Lagutchev, A.; Hambir, S. A.; Dlott, D. D., Nonresonant Background Suppression in Broadband Vibrational Sum-Frequency Generation Spectroscopy. *J. Phys. Chem. C* 2007, *111*, 13645-13647.

48. Stiopkin, I. V.; Jayathilake, H. D.; Weeraman, C.; Benderskii, A. V., Temporal Effects on Spectroscopic Line Shapes, Resolution, and Sensitivity of the Broad-Band Sum Frequency Generation. *J. Chem. Phys.* **2010**, *132*, 234503.

49. Ferguson, G. A.; Raghavachari, K., Collective Vibrations in Cluster Models for Semiconductor Surfaces: Vibrational Spectra of Acetylenyl and Methylacetylenyl Functionalized Si(111). *The Journal of Chemical Physics* **2007**, *127*, 194706.

50. Yates, D. J. C.; Lucchesi, P. J., Infrared Spectra of Acetylene and Acetylene Derivatives Adsorbed on Alumina and Silica. *The Journal of Chemical Physics* **1961**, *35*, 243-255.

51. Yang, F.; Hunger, R.; Roodenko, K.; Hinrichs, K.; Rademann, K.; Rappich, J., Vibrational and Electronic Characterization of Ethynyl Derivatives Grafted onto Hydrogenated Si(111) Surfaces. *Langmuir* **2009**, *25*, 9313-9318.

52. Tom, H. W. K.; Heinz, T. F.; Shen, Y. R., Second-Harmonic Reflection from Silicon Surfaces and Its Relation to Structural Symmetry. *Phys Rev Lett* **1983**, *51*, 1983-1986.

53. Sipe, J. E.; Moss, D. J.; van Driel, H. M., Phenomenological Theory of Optical Secondand Third-Harmonic Generation from Cubic Centrosymmetric Crystals. *Phys Rev B* **1987**, *35*, 1129-1141.

54. Mitchell, S. A.; Boukherroub, R.; Anderson, S., Second Harmonic Generation at Chemically Modified Si(111) Surfaces. *J. Phys. Chem. B* **2000**, *104*, 7668-7676.

55. Hirose, C.; Akamatsu, N.; Domen, K., Formulas for the Analysis of Surface Sum -Frequency Generation Spectrum by Ch Stretching Modes of Methyl and Methylene Groups. *The Journal of Chemical Physics* **1992**, *96*, 997-1004.

56. Hirose, C.; Yamamoto, H.; Akamatsu, N.; Domen, K., Orientation Analysis by Simulation of Vibrational Sum Frequency Generation Spectrum: Ch Stretching Bands of the Methyl Group. *The Journal of Physical Chemistry* **1993**, *97*, 10064-10069.

57. Wang, H.-F.; Gan, W.; Lu, R.; Rao, Y.; Wu, B.-H., Quantitative Spectral and Orientational Analysis in Surface Sum Frequency Generation Vibrational Spectroscopy (Sfg-Vs). *Int Rev Phys Chem* **2005**, *24*, 191-256.

58. Zhuang, X.; Miranda, P. B.; Kim, D.; Shen, Y. R., Mapping Molecular Orientation and Conformation at Interfaces by Surface Nonlinear Optics. *Phys Rev B* **1999**, *59*, 12632-12640.

59. Gautam, K. S.; Schwab, A. D.; Dhinojwala, A.; Zhang, D.; Dougal, S. M.; Yeganeh, M. S., Molecular Structure of Polystyrene at Air/Polymer and Solid/Polymer Interfaces. *Phys Rev Lett* **2000**, *85*, 3854-3857.

60. Nemanick, E. J.; Solares, S. D.; Goddard, W. A.; Lewis, N. S., Quantum Mechanics Calculations of the Thermodynamically Controlled Coverage and Structure of Alkyl Monolayers on Si(111) Surfaces. *The Journal of Physical Chemistry B* **2006**, *110*, 14842-14848.

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