

VISCOSITY MEASUREMENTS ON COLLOIDAL DISPERSIONS (NANOFLUIDS) FOR HEAT TRANSFER APPLICATIONS

DAVID C. VENERUS^{1*}, JACOPO BUONGIORNO², REBECCA CHRISTIANSON³, JESSICA TOWNSEND³, IN CHEOL BANG^{4,5}, GANG CHEN², SUNG JAE CHUNG¹⁶, MINKING CHYU¹⁶, HAISHENG CHEN⁶, YULONG DING⁶, FRANK DUBOIS⁷, GRZEGORZ DZIDO⁸, DENIS FUNFSCHILLING⁹, QUENTIN GALAND⁷, JINWEI GAO², HAIPING HONG¹⁰, MARK HORTON¹⁰, LINWEN HU², CARLO S. IORIO⁷, ANDRZEJ B. JARZEBSKI⁸, YIRAN JIANG¹, STEPHAN KABELAC¹¹, MARK A KEDZIERSKI¹², CHONGYOUNG KIM¹³, JI-HYUN KIM⁴, SUKWON KIM¹³, THOMAS MCKRELL², RUI NI⁹, JOHN PHILIP¹⁴, NAVEEN PRABHAT², PENGXIANG SONG¹⁵, STEFAN VAN VAERENBERGH⁷, DONGSHENG WEN¹⁵, SANJEEVA WITHARANA⁶, XIAO-ZHENG ZHAO⁹, SHENG-QI ZHOU⁹

¹ Illinois Institute of Technology, 10 W. 33rd St., Chicago, IL 60616, USA

² Massachusetts Institute of Technology (MIT), 77 Massachusetts Avenue, Cambridge, MA 02139, USA

³ Olin College of Engineering, Olin Way, Needham, MA 02492, USA

⁴ Ulsan National Institute of Science and Technology, School of Energy Engineering, San 194 Banyeon-ri, Eonyang-eup, Ulju-gun, Ulsan Metropolitan City, Republic of Korea

⁵ Tokyo Institute of Technology, 2-12-1 Ookayama, Meguro-ku, Tokyo 152-8550, Japan

⁶ University of Leeds, Clarendon Road, Leeds LS2 9JT, UK

⁷ Université Libre de Bruxelles, Chimie-Physique E.P. CP 165/62 Avenue P.Heger, 1050, Bruxelles, Belgium

⁸ Silesian University of Technology, Department of Chemical and Processing Engineering, ul. M. Strzody 7, 44-100 Gliwice, Poland

⁹ Chinese University of Hong Kong, Department of Physics, G6, North Block, Science Center, Shatin NT, Hong Kong, China

¹⁰ South Dakota School of Mines and Technology, 501 E Saint Joseph Street, Rapid City, SD 57702, USA

¹¹ Helmut-Schmidt University Hamburg, Institute for Thermodynamics, 22039, Hamburg, Germany

¹² National Institute of Standards and Technology (NIST), MS 863, Gaithersburg, MD 20899, USA

¹³ Korea University, Anam-dong, Sungbuk-ku, Seoul 136-713, Republic of Korea

¹⁴ Indira Gandhi Centre for Atomic Research, SMARTS, NDED, Metallurgy and Materials Group, Kalpakkam-603102, India

¹⁵ Queen Mary University of London, School of Engineering and Materials Science, Mile End Road, London, E1 4NS, UK

¹⁶ University of Pittsburgh, Dept. Mechanical Engineering and Materials Science, 648 Benedum Hall, 3700 O'Hara Street, Pittsburgh, PA 15261, USA

* Email: venerus@iit.edu

Fax: x1.312.567.8874

Received: 2.11.2009, Final version: 15.12.2009

ABSTRACT:

This article reports viscosity data on a series of colloidal dispersions collected as part of the International Nanofluid Property Benchmark Exercise (INPBE). Data are reported for seven different fluids that include dispersions of metal-oxide nanoparticles in water, and in synthetic oil. These fluids, which are also referred to as 'nanofluids,' are currently being researched for their potential to function as heat transfer fluids. In a recently published paper from the INPBE study, thermal conductivity data from more than 30 laboratories around the world were reported and analyzed. Here, we examine the influence of particle shape and concentration on the viscosity of these same nanofluids and compare data to predictions from classical theories on suspension rheology.

ZUSAMMENFASSUNG:

In diesem Beitrag werden experimentelle Daten zur Viskosität verschiedener kolloidaler Dispersionen vorgestellt, welche im Rahmen einer internationalen vergleichenden Ringmessung (International Nanofluid Property Benchmark Exercise INPBE) gewonnen wurden. Es werden hierbei sieben unterschiedliche Fluide betrachtet, unter anderem Wasser beziehungsweise synthetisches Öl mit suspendierten metalloxidischen Nanopartikeln. Diese auch Nanofluide genannten Dispersionen werden derzeit intensiv aufgrund ihres möglichen Potenzials als effizientes Wärmeträger-Fluid untersucht. Vor Kurzem wurden in einem weiteren Beitrag der INPBE-Studie Daten zur Wärmeleitfähigkeit von Nanofluiden veröffentlicht, wie sie in den über 30 teilnehmenden Forschungslabors weltweit gemessen wurden. In diesem Beitrag wird der Einfluss der Partikel-Gestalt und der Partikel-Konzentration auf die Viskosität der gleichen Nanofluide untersucht und mit Korrelationen aus der klassischen Rheologie von Suspensionen verglichen.

RÉSUMÉ:

Cet article présente des données de viscosité sur des séries de dispersions colloïdales qui font partie du «International Nanofluid Property Benchmark Exercise (INPBE)». Les données correspondent à sept fluides différents incluant des dispersions de nano particules d'oxyde métalliques dans de l'eau et dans de l'huile synthétique. Ces fluides, également connus comme « nano fluides » font actuellement l'objet d'études pour leur potentiel comme des fluides pour le transfert de chaleur. Dans un article récemment publié dans le cadre d'un étude INPBE, des données de conductivité thermique de plus de 30 laboratoires différents ont été présentées et analysées. Dans ce travail, on étudie l'influence de la forme et de la concentration des particules sur la viscosité des nano fluides qui en sont issus. Ces données sont comparées aux prédictions théoriques des modèles classiques sur la rhéologie de suspensions.

SOMMARIO:

Questo articolo riporta dati sperimentali sulla viscosità di una serie di sistemi di nano-colloidi (nanofluidi) nel contesto del programma internazionale INPBE (International Nanofluid Property Benchmark Exercise). I dati riguardano sette diversi sistemi, comprendenti particelle di ossidi metallici in acqua ed olio sintetico. Questi fluidi sono attualmente soggetti di interesse nella comunità scientifica per la loro possibile applicazione come fluidi termici. I dati di conducibilità termica prodotti da più di 30 laboratori internazionali sono stati riportati e discussi in un articolo su INPBE di recente pubblicazione. Invece qui esaminiamo l'influenza della forma e concentrazione delle nanoparticelle sulla viscosità degli stessi campioni, e confrontiamo i dati con i risultati delle classiche teorie reologiche per sospensioni colloidali.

KEY WORDS: nanofluids, colloidal dispersion, viscosity, thermal conductivity

1 INTRODUCTION

Suspensions of nanometer-sized (1–100 nm), solid particles, also known as colloids or colloidal dispersions, are of great technological importance and can be found in applications ranging from inks to pharmaceuticals. For over a decade, colloidal dispersions containing metallic, or metal-oxide, nanoparticles, sometimes referred to as 'nanofluids,' have been the focus of intensive research because of their potential as heat transfer fluids. This research has been driven by the possibility of having fluids with enhanced thermal conductivity, relative to the suspending liquid (basefluid), while mitigating problems associated with erosion, sedimentation and clogging that plague suspensions of larger particles. Interest in nanofluids as heat transfer fluids has been

further intensified by early studies reporting anomalously large – relative to classical model predictions – enhancement of thermal conductivity. Several review papers on nanofluid heat transfer have been published [1–4].

Transport properties (e.g. electrical or thermal conductivity) of heterogeneous systems are often estimated using effective medium theory. For example, the effective thermal conductivity k of a suspension composed of solid particles with thermal conductivity k_p at volume fraction ϕ (« ϕ ») dispersed in a liquid of thermal conductivity k_f (« k_f ») can be written as

$$\frac{k}{k_f} = 1 + [k]\phi + O(\phi^2) \quad (1)$$

where $[k]$ is defined as follows: $[k] = \lim_{\phi \rightarrow 0} \{(k/k_f - 1)/\phi\}$. Over 100 years ago, Maxwell [5] derived an expression for the thermal (electrical) conductivity of a dilute system of spherical particles that yielded $[k] = 3$. Models for the effective thermal conductivity of heterogeneous systems have been formulated to account for the effects of non-spherical particle shape, particle-particle interactions, and interfacial resistance between the particle and continuous phases [1–4].

There have been numerous experimental studies reporting thermal conductivity data on nanofluids. Much of the reported data show both quantitative and qualitative differences from the effective medium theory prediction in Equation 1. For example, in many early studies, the measured levels of thermal conductivity enhancement for dilute nanofluids ($\phi \leq 0.01$) were significantly larger than the classical prediction given in Equation 1 with $[k] = 3$. Several investigators have reported a dependence of thermal conductivity enhancement on particle size and on temperature, neither of which is consistent with Equation 1. In addition to generating a great deal of optimism about the potential of nanofluids for use as heat transfer fluids, early experimental results also sparked theoretical interest to identify the mechanism(s) responsible for the observed departures from effective medium theory predictions. Further discussion of both experimental and theoretical research on nanofluids can be found elsewhere [1–4]. At present, the study of nanofluids as heat transfer fluids remains both an active and controversial field of research.

It is well known that the addition of solid particles to a liquid can significantly alter its rheological behavior. In shear flows, the apparent viscosity of a fluid η is ratio of the shear stress to the shear rate: $\eta = \sigma/\dot{\gamma}$. The zero-shear rate viscosity is defined as: $\eta_o = \lim_{\dot{\gamma} \rightarrow 0} \{\eta\}$. The zero-shear rate viscosity η_o of a dilute suspension with liquid phase viscosity η_f can be expressed as follows:

$$\frac{\eta_o}{\eta_f} = 1 + [\eta]\phi + O(\phi^2) \quad (2)$$

where the intrinsic viscosity $[\eta]$ of the suspension is defined as: $[\eta] = \lim_{\phi \rightarrow 0} \{(\eta_o/\eta_f - 1)/\phi\}$. The well-known prediction of Einstein [6] for a suspension of spherical particles gives $[\eta] = 2.5$, while for dilute suspensions of particles with large aspect

ratio p , $[\eta] \approx p$ [7]. The linear dependence of η_o on particle volume fraction in Equation 2 typically holds for $\phi \leq 0.03$ [7].

In addition to thermal conductivity, it is also evident that viscosity has a significant impact on the overall performance of a heat transfer fluid. Clearly, pumping a fluid with increased viscosity through a heat exchanger requires an increase in pumping energy, thereby reducing the overall benefit of a higher thermal conductivity fluid. In light of this, there have been a number of studies reporting the viscosity of nanofluids [8–14]. These studies considered nanofluids composed of metal-oxide (Al_2O_3 , CuO , TiO_2 , Fe_2O_3) and metal (Cu), nanometer-sized, spherical particles in water, ethylene glycol, or hydrocarbon oil at concentrations $\phi \leq 0.05$. For all of these nanofluids, Newtonian behavior was observed, and the dependence of relative viscosity η/η_f was approximately linear in particle concentration ϕ . However, the reported intrinsic viscosity $[\eta]$ for these nanofluids was in the range 4–16, which is substantially larger than the theoretical value of 2.5. Prasher et al., [9] and Garg et al., [12] proposed criteria for the overall effectiveness of nanofluids as heat transfer fluids suggesting that $[\eta]$ should be 4–5 times smaller than $[k]$ appearing in Equation 1. The effect of changes in specific heat due to the presence of nanoparticles on the effectiveness of nanofluids as heat transfer fluids has also been investigated [15].

At the first scientific conference centered on nanofluids (Nanofluids: Fundamentals and Applications, September 16–20, 2007, Copper Mountain, Colorado), it was decided to launch an International Nanofluid Property Benchmark Exercise (INPBE), to resolve the inconsistencies in the database and help advance the debate on nanofluid properties. Thermal conductivity data on eight different nanofluids from this study, involving over 30 laboratories around the world, are contained in a recently published paper [16]. A subset of INPBE participants also collected viscosity data on the same set of nanofluids. These data are reported in this paper. The methodology and samples used for the INPBE are described in Section 2. The viscosity data are presented and discussed in Section 3. The findings of this study are summarized in Section 4.

2 INPBE METHODOLOGY

Eight different nanofluids were distributed in four sets to participating INPBE laboratories. To mini-

mize spurious effects due to nanofluid preparation and handling, all organizations were given identical samples from these sets, and were asked to adhere to the same sample handling protocol. The exercise was 'semi-blind,' as only minimal information about the samples was given to the participants at the time of sample shipment. The data were then collected and posted, on the INPBE website (<http://mit.edu/nse/nanofluids/benchmark/index.html>). Additional details of the INPBE methodology and list of participants can be found in the paper containing the thermal conductivity data [16] and at the INPBE website.

Table 1 gives the main characteristics of the nanofluids used in the INPBE study; the sample names in the first column of the table indicate Set#Sample# as designated in the previous paper [16]. Samples S1S1-S1S7 were provided by Sasol. Sample S1S2 is simply de-ionized H₂O and is the base fluid for S1S1; sample S1S7 is a mixture of poly-alpha olefin (PAO) oil (SpectraSyn-10 by Exxon Oil) and 5%wt. dispersant (Solsperse 21000 by Lubrizol Chemical) and is the base fluid for samples S1S3-S1S6. The nanofluids designated as S1S1 and S1S3-S1S6 contain alumina (Al₂O₃) nanoparticles with the shapes and sizes indicated in Table 1. Sample S3S1 was supplied by W. R. Grace & Co. and is a colloidal dispersion of silica (SiO₂) nanoparticles dispersed in de-ionized H₂O stabilized by the addition of small amounts of Na₂SO₄ (pH = 9). Finally, samples in Set 4 were supplied by Dr. Jorge Gustavo Gutierrez of the University of Puerto Rico – Mayaguez (UPRM). Sample S4S2 is a solution of de-ionized H₂O (75% wt) and tetramethylammonium hydroxide [(CH₃)₄NOH] stabilizer (25% wt). Sample S4S1 contains Mn-Zn ferrite (Mn_{1/2}-Zn_{1/2}-Fe₂O₄) nanoparticles dispersed in a based fluid S4S2. To summarize, this study involves seven nanofluids and three base fluids. Note that one of the eight nanofluids that were part of the INPBE study, a dilute colloidal gold dispersion, is excluded from this paper because its rheology was indistinguishable from that of water. Additional details concerning these samples and their characterization can be found elsewhere [16].

Viscosity data on each of the fluids listed in Table 1 were reported by approximately 8-10 different laboratories. All reported data were obtained using well-established methods with commercial viscometers that can be broadly separated into two categories according to the type

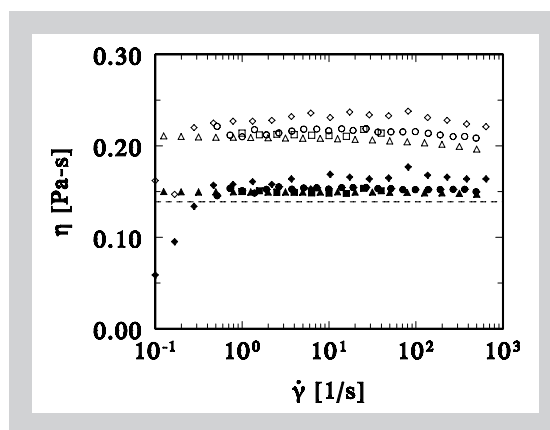


Figure 1: Viscosity versus shear rate for spherical Al₂O₃ nanoparticles in oil: filled symbols $\phi = 0.01$ (S1S3); open symbols $\phi = 0.03$ for (S1S4). Different symbols indicate data from different laboratories. Dashed line is average for oil base fluid (S1S7).

Table 1: Characteristics of nanofluid and base fluid samples.

of deformation applied to the sample. In one category, that includes gravity driven capillary, concentric cylinder and immersion viscometers, sample deformation is non-homogeneous and the (average) rate of deformation is usually not known. In the other category are viscometers that apply a controlled and (approximately) uniform deformation to the sample such as cone-plate viscometers. Only two of the 10 samples considered in this study showed non-Newtonian behavior (i.e., viscosity η was dependent on shear rate $\dot{\gamma}$), so that data from both types of viscometers are included for the eight Newtonian fluids. For the two samples displaying non-Newtonian behavior, only data from cone-plate viscometers will be presented. Finally, all data reported here were collected at room temperature, which was in the range 20-26 °C. We recognize that changes in viscosity of roughly 10% are possible over this range of temperatures.

3 RESULTS AND DISCUSSION

First we examine the viscosity versus shear rate behavior of several fluids listed in Table 1. Data for samples S1S3 and S1S4 from four different laboratories are shown in Figure 1. The dashed line in Figure 1 represents the average viscosity of the base fluid, sample S1S7. From this figure, it is clear that these nanofluids are Newtonian over the range of shear rates considered. This is expected given that these fluids contain spherical particles at relatively modest volume fractions ($\phi = 0.01, 0.03$); this is consistent with previous results on

Sample	liquid	solid (shape)	d or L-d [nm]	ϕ	η_0 [Pa·s]
S1S1	H ₂ O	Al ₂ O ₃ (rod)	80-100	0.01	$12 \pm 2 \cdot 10^{-4}$
S1S2	H ₂ O	--	--	--	$8.9 \cdot 10^{-4}$ *
S1S3	PAO+surf	Al ₂ O ₃ (sph)	100	0.01	$1.7 \pm 0.4 \cdot 10^{-1}$
S1S4	PAO+surf	Al ₂ O ₃ (sph)	100	0.03	$2.4 \pm 0.5 \cdot 10^{-1}$
S1S5	PAO+surf	Al ₂ O ₃ (rod)	80-100	0.01	$2.1 \pm 0.5 \cdot 10^{-1}$
S1S6	PAO+surf	Al ₂ O ₃ (rod)	80-100	0.03	$4.4 \pm 0.8 \cdot 10^{-1}$
S1S7	PAO+surf	--	--	--	$1.4 \pm 0.3 \cdot 10^{-1}$
S3S1	H ₂ O+stab	SiO ₂ (sph)	22	0.32	$4 \pm 0.8 \cdot 10^{-1}$
S4S1	H ₂ O+stab	Mn _{1/2} Zn _{1/2} Fe ₂ O ₃ (sph)	7	0.0017	$3.11 \pm 0.37 \cdot 10^{-3}$
S4S2	H ₂ O+stab	--	--	--	$3.1 \pm 0.41 \cdot 10^{-3}$

Figure 2 (left above): Viscosity versus shear rate for spherical SiO_2 nanoparticles in water: $\phi = 0.32$ (S3S1). Different symbols indicate data from different laboratories. Inset shows relative viscosity versus dimensionless shear rate for one set of data.

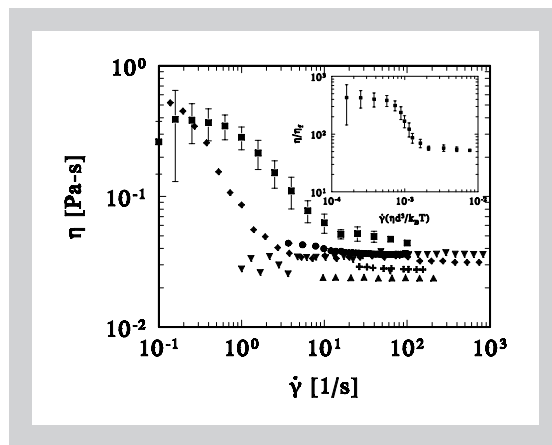


Figure 3 (right above): Viscosity versus shear rate for rod-shaped Al_2O_3 nanoparticles in oil: filled symbols $\phi = 0.01$ (S1S5); open symbols $\phi = 0.03$ for (S1S6). Different symbols indicate data from different laboratories. Dashed line is average for oil base fluid (S1S7).

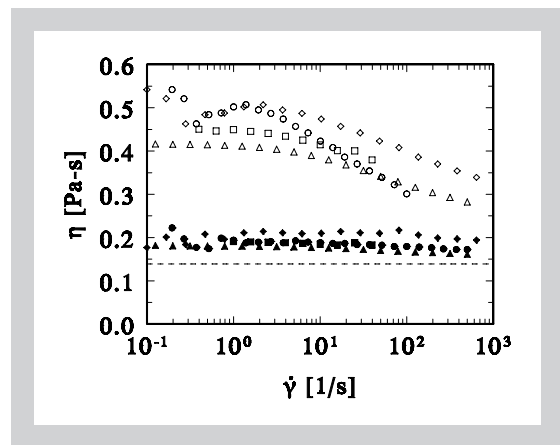
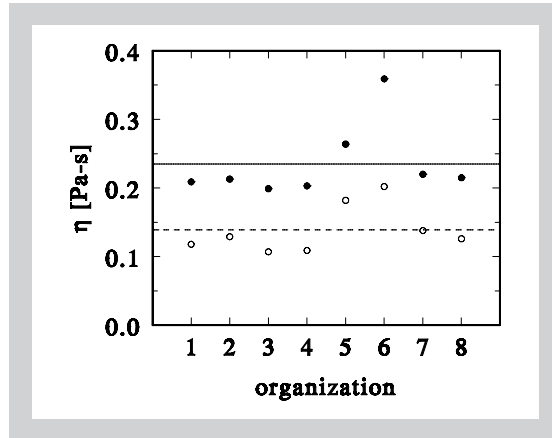


Figure 4 (below): Viscosity of spherical Al_2O_3 nanoparticles in oil as measured by different laboratories: filled symbols are for $\phi = 0.03$ (S1S4); open symbols are for base fluid (S1S7). Solid and dashed lines indicate average values.



similar systems [8 - 14]. It is also evident from Figure 1 that there is good agreement between the data reported by the different laboratories with variations of approximately 10%. Results similar to those shown in Figure 1 were obtained for samples S1S1, S1S7, S4S1 and S4S2; therefore only the average viscosity of these Newtonian fluids will be presented later in this section.

Figure 2 shows the viscosity versus shear rate behavior of the concentrated colloidal silica dispersion sample S3S1. At low shear rates, these data appear to approach a constant viscosity η_0 with a value of approximately 0.4 Pa·s, and a constant viscosity at shear rates greater than 10 s^{-1} . At intermediate shear rates, between $0.1 - 10 \text{ s}^{-1}$, pronounced shear thinning is observed. The scatter among the data from different laboratories is larger in Figure 2 than Figure 1, and is probably the result of the higher sensitivity to shear history of the more concentrated system, as explained next. Shear thinning behavior, which is typical for concentrated suspensions of spherical particles, can be explained by the flow-induced perturbation of particle positions [7]. The rate of particle displacement by flow is given roughly by the shear rate $\dot{\gamma}$, and the rate of particle movement by Brownian motion is roughly $k_B T / \eta d^3$. Hence, when flow perturbs particle positions at a rate faster than they are recovered by thermal fluctuations, the viscosity of the suspension becomes shear-rate dependent. The

inset to Figure 2 shows a commonly used means of rescaling viscosity and shear rate to demonstrate the competition between convection (flow) and diffusion (Brownian motion) in suspensions [7].

Viscosity versus shear rate data from four different laboratories for the nanofluids comprised of rod-shaped Al_2O_3 particles in oil (S1S5 and S1S6) are shown in Figure 3. The dashed line in Figure 3 represents the average viscosity of the base fluid (S1S7). From this figure, we see that the lower concentration ($\phi = 0.01$) fluid is Newtonian, while the higher concentration ($\phi = 0.03$) fluid is non-Newtonian. Again, agreement between the different laboratories is good. Comparison of the data in Figures 1 and 3 for $\phi = 0.03$ shows, as expected, that the viscosity of the fluid with non-spherical particles is more sensitive to shear compared to the fluid with spherical particles. For non-spherical particles, non-Newtonian effects are the result of flow-induced perturbation of particle orientations from the random state that exists at equilibrium [7]. In this case, it is a competition between the tendency for flow to orient particles and thermally driven rotation of particles.

Viscosity data for the nanofluid comprised of spherical Al_2O_3 particles with $\phi = 0.03$ (S1S4) and its oil base fluid (S1S7) as reported by different INPBE participants are shown in Figure 4. For a given laboratory, the reported uncertainty was typically around $\pm 5\%$, which is roughly the size of the symbols in Figure 4. The data in this figure give an idea of consistency of viscosity data reported by different laboratories using different techniques, which is within approximately $\pm 20\%$. Similar results were found for the other nanofluids used in this study.

The relative viscosity versus particle concentration for nanofluids with spherical particles and rod-shaped particles are shown in Figures 5 and 6 respectively. As expected, the data in both figures indicate a linear dependence of η/η_f on ϕ . The measured dependence $[\eta]$, however, is roughly ten times larger than predicted, with

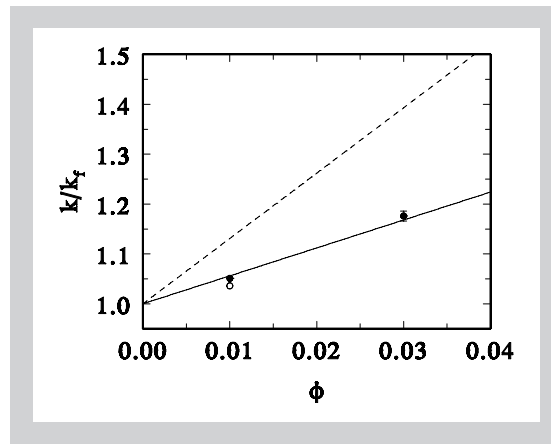
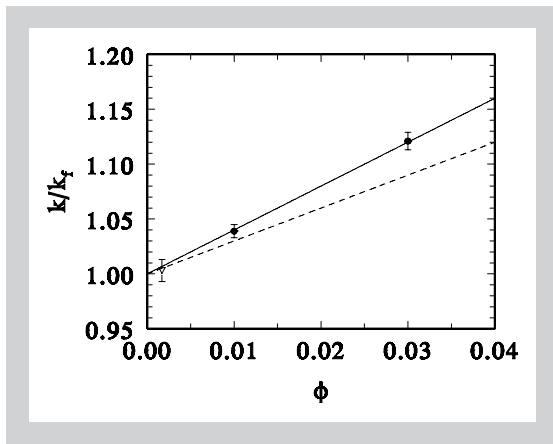
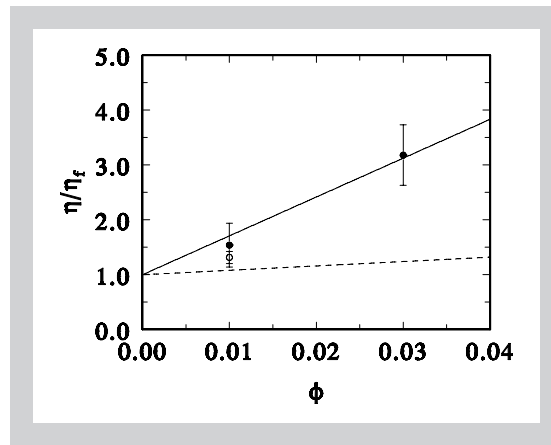
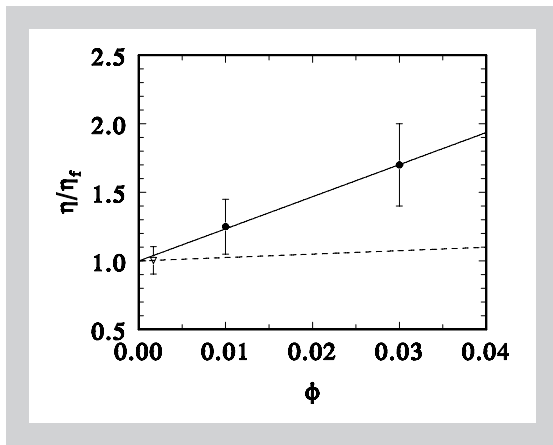


Figure 5 (left above): Relative viscosity η/η_f versus particle concentration ϕ for nanofluids with spherical particles: filled circles for Al_2O_3 particles in oil (S1S3 and S1S4) and open triangle for $\text{Mn}_{1/2}\text{Zn}_{1/2}\text{Fe}_2\text{O}_3$ in water (S4S1). Lines indicate $\eta/\eta_f = 1 + [\eta]\phi$ with prediction $[\eta] = 5/2$ (dashed) and fit $[\eta] = 23.4$ (solid).

Figure 6 (right above): Relative viscosity η/η_f versus particle concentration ϕ for nanofluids with rod-shaped particles: filled circles for Al_2O_3 particles in oil (S1S5 and S1S6) and open circle for Al_2O_3 in water (S1S1). Lines indicate $\eta/\eta_f = 1 + [\eta]\phi$ with prediction $[\eta] = 8$ (dashed) and fit $[\eta] = 70.8$ (solid).

Figure 7 (left below): Relative thermal conductivity k/k_f versus particle concentration ϕ for nanofluids with spherical particles: filled circles for Al_2O_3 particles in oil (S1S3 and S1S4) and open triangle for $\text{Mn}_{1/2}\text{Zn}_{1/2}\text{Fe}_2\text{O}_3$ in water (S4S1). Data taken from INPBE study [16]. Lines indicate $k/k_f = 1 + [k]\phi$ with prediction $[k] = 3$ (dashed) and fit $[k] = 4.0$ (solid).

Figure 8 (right below): Relative thermal conductivity k/k_f versus particle concentration ϕ for nanofluids with rod-shaped particles: filled circles for Al_2O_3 particles in oil (S1S5 and S1S6) and open circle for Al_2O_3 in water (S1S1). Data taken from INPBE study [16]. Lines indicate $k/k_f = 1 + [k]\phi$ with prediction $[k] = 13.1$ (dashed) and fit $[k] = 5.6$ (solid).

$[\eta] = 23.4$ for the spherical particles and $[\eta] = 70.8$ for the rod-shaped particles. The shear thinning behavior observed in alumina nanorod suspensions can also be due to shear induced reorganization of nanorods under confined geometry where the packing can become more efficient. A similar result has been reported in recently published data for nanofluids with spherical particles [8–13], although somewhat lower values of $[\eta] = 4$ –16 were found. A possible explanation for these pronounced differences between theory and experiment is particle agglomeration, which would increase the effective volume fraction of the particles [10–14]. Indeed, light scattering results on these fluids reported elsewhere [16] are consistent with the occurrence of particle agglomeration.

For comparison, we show the relative thermal conductivity versus particle concentration data from the INPBE study [16] in Figures 7 and 8. As with the viscosity data, nanofluids with both spherical and rod-shaped particles show a linear dependence of k/k_f on ϕ . The data for the fluids with spherical particles in Figure 7 show a slightly stronger dependence on particle concentration than predicted: measured $[k] = 4.0$, and predicted $[k] = 3$. This observation is consistent with the presence of particle agglomeration in these systems, and suggests that particle clustering has a larger effect on viscosity than thermal conductivity in nanofluids with spherical particles. The situation

is different for the nanofluids with rod-shaped particles as shown in Figure 8. Here, the observed dependence on particle concentration is weaker than predicted: measured $[k] = 5.6$, and predicted $[k] = 13.1$. Apparently, particle agglomeration can significantly reduce the effective thermal conductivity of nanofluids with rod-shaped particles. A second explanation for the reduction in effective thermal conductivity in these nanofluids is interfacial thermal resistance [16].

4 CONCLUSIONS

Viscosity data have been collected as part of an International Nanofluid Property Benchmark Exercise (INPBE). These data are from approximately 10 different laboratories around the world on a series of 10 different nanofluids and their base fluids. In general, the agreement between different laboratories was good with variations of approximately $\pm 20\%$, which, in part, could be explained by lab-to-lab temperature variations. Two of seven nanofluids showed shear-thinning behavior; the remaining five showed Newtonian behavior. For nanofluids with both spherical and rod-shaped nanoparticles, the dependence of viscosity (relative to the base fluid viscosity) on particle concentration (volume fraction) was significantly stronger than predicted by dilute suspension theory. This discrepancy was attributed to particle agglomera-



Map:
Map showing locations of laboratories participating in the International Nanofluid Properties Benchmark Exercise (INPBE).

tion. In contrast, the observed enhancement in thermal conductivity was slightly larger for the spherical particle fluids, and significantly lower for the rod-shaped particle fluids, than predicted by effective medium theory. As noted in the introduction, criteria for the overall effectiveness of nanofluids as heat transfer fluids have been proposed [9, 12], which suggest $[\eta]$ should be 4 - 5 times smaller than $[k]$. Clearly, the nanofluids considered in this study would fail; this suggests that the overall effect of adding nanoparticles to the base fluid is negative in terms of heat transfer performance.

ACKNOWLEDGEMENTS

This work was made possible by the support of the National Science Foundation under grant CBET-0812804. The authors are also grateful to Sasol and W. R. Grace & Co. for donating some of the samples used in INPBE.

REFERENCES

- [1] Eastman JA, Phillpot SR, Choi SUS: Keblinski P, Nanofluids for thermal transport, *Annu. Rev. Mat. Res.* 34 (2004) 219.
- [2] Keblinski P, Eastman JA, Cahill DG: Nanofluids for thermal transport, *Materials Today*, June (2005) 36.
- [3] Das SK, Choi SUS, Patel HE: Heat transfer in nanofluids – a review, *Heat Transfer Eng.* 27 (2006) 3.
- [4] Yu W, France DM, Routbort JL, Choi SUS: Review and comparison of nanofluid thermal conductivity and heat transfer enhancements, *Heat Transfer Eng.* 29 (2008) 432.
- [5] Maxwell JC: *A Treatise on Electricity and Magnetism*, Clarendon, Oxford (1881).
- [6] Einstein A: Eine neue Bestimmung der Moleküldimensionen, *Ann. der Physik* 19 (1906) 289; Einstein A: Berichtigung zu meiner Arbeit: "Eine neue Bestimmung der Moleküldimensionen," *Ann. der Physik* 34 (1911) 951.

- [7] Russel WB, Saville DA, Schowalter WR: *Colloidal Dispersions*, Cambridge University Press, Cambridge (1989).
- [8] Das SK, Putra N, Thiesen P, Roetzel W: Pool boiling characteristics of nano-fluids, *Int. J. Heat Mass Transfer* 46 (2003) 851.
- [9] Prasher R, Song D, Wang J: Measurements of nanofluid viscosity and its implications for thermal applications, *Appl. Phys. Lett.*, 89 (2006) 133108.
- [10] Chen H, Ding Y, He Y, Tan C: Rheological behaviour of ethylene glycol based titania nanofluids, *Chem. Phys. Lett.* 444 (2007) 333.
- [11] Murshed SMS, Leong KC, Yang C: Investigations of thermal conductivity and viscosity of nanofluids, *Int. J. Therm. Sci.* 47 (2008) 560.
- [12] Garg J, Poudel B, Chiesa M, Gordon JB, Ma JJ, Wang JB, Ren ZF, Kang WT, Ohtani H, Nanda J, McKinley GH, Chen G: Enhanced thermal conductivity and viscosity of copper nanoparticles in ethylene glycol nanofluid, *J. Appl. Phys.* 103 (2008) 074301.
- [13] Anoop KB, Kabelac S, Sundararajan T, Das SK: Rheological and flow characteristics of nanofluids: Influence of electroviscous effects and particle agglomeration, *J. Appl. Phys.* 106 (2009) 034909.
- [14] Shima PD, Philip J, Raj B: Magnetically controllable nanofluid with tunable thermal conductivity and viscosity, *Appl. Phys. Lett.* 95 (2009) 133112.
- [15] Bergman TL: Effect of reduced specific heats of nanofluids on single phase, laminar internal forced convection, *Int. J. Heat Mass Transfer.* 52 (2009) 1240.
- [16] Buongiorno J, Venerus DC, Prabhat N, McKrell T, Townsend J, Christianson R, Tolmachev YV, Keblinski P, Hu L-W, Alvarado JL, Bang IC, Bishnoi SW, Bonetti M, Botz F, Cecere A, Chang Y, Chen G, Chen H, Chung SJ, Chyu MK, Das SK, Di Paola R, Ding Y, Dubois F, Dzido G, Eapen J, Escher W, Funschilling D, Galand Q, Gao J, Gharagozloo PE, Goodson KE, Gutierrez JG, Hong H, Horton M, Iorio CS, Jarzëbski AB, Jiang Y, Jin LW, Kabelac S, Kamath A, Kedzierski MA, Kim C, Kim JH, Kim S, Kieng LG, Leong KC, Manna I, Michel B, Ni R, Patel HE, Philip J, Poulikakos D, Reynaud C, Savino R, Singh PK, Song P, Sundararajan T, Timofeeva E, Triticak T, Turanov AN, Van Vaerenbergh S, Wen D, Witharana S, Yang CC, Yeh W.-H., Zhao X-Z, Zhou S-Q: A Benchmark Study on the Thermal Conductivity of Nanofluids, *J. Appl. Phys.* 106 (2009) 094312.

