Winter Particulate Pollution over Raipur, India

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Abstract: Particulate air pollution during the winter season in the urban regions of India is severe due to substantial fuel and mineral combustion in adverse climatic conditions. In this work, chemical characteristics and sources of coarse particulate matter (PM_{10}) and particulates associated chemicals during winter period of years 2006–2013 in the polluted city of Raipur, Chhattisgarh, India, are reported. The ambient air coarse particulate (PM_{10}) concentration during the winter period of 2006–2007 ranged from 221 to 760 μ g m⁻³. The major fraction of the PM_{10} was composed of organic carbon, elemental carbon, iron, calcium, and sulfate. Their concentrations were remarkably reduced in the rainy season due to high wind speeds (around 10 kmh⁻¹) and removal with rain. The concentration variations and sources of PM and associated chemical species (i.e., carbons, ions, and metals) in the ambient air are discussed. **DOI:** 10.1061/(ASCE)HZ.2153-5515.0000444. © 2019 American Society of Civil Engineers.

Author keywords: Particulate matters; Carbons; Metals; Ions; Sources.

Introduction

Particulate matter (PM) is composed of a wide variety of airborne materials (i.e., dust, smoke, and soot, among others), which are directly emitted into the air or result from the transformation of gaseous pollutants due to natural and anthropogenic sources (WHO 2006). PM has been reported as criteria pollutants by the USEPA (USEPA 2010). PM ($\leq 10 \ \mu m$) is of most concern for its effects on human health, i.e., chronic lung disease and asthma, lung cancer, heart attacks, exacerbation of chronic obstructive pulmonary disease (COPD), premature death, preterm birth, and low birth weight, among other (Ferrante et al. 2015). Some metals interact with enzymes, cell components, or DNA, for example, to cause adverse

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Note. This manuscript was submitted on December 11, 2018; approved on March 12, 2019; published online on June 19, 2019. Discussion period open until November 19, 2019; separate discussions must be submitted for individual papers. This paper is part of the *Journal of Hazardous, Toxic, and Radioactive Waste*, © ASCE, ISSN 2153-5493. health effects (Fortoul et al. 2015). The optically active components (e.g., elemental and organic carbon and sulfate) of the particulates affect precipitation and cloud-cover events (Bell and Holloway 2007). High particulate concentrations in ambient air have caused synoptic weather in several parts of the world (Awad and Mashat 2016; Ding et al. 2016; Han et al. 2015; Sati and Mohan 2014; Guttikunda et al. 2013). The distribution, composition, and origins of particulates in ambient air of several regions have been reported (Zeb et al. 2018; Ding et al. 2016; Pan et al. 2015; Xu et al. 2015; Tao et al. 2014; Satsangi et al. 2013; Singh et al. 2013; Yadav and Satsangi 2013; Khillare and Sarkar 2012; Kulshrestha et al. 2009; Lakhani et al. 2008; Gupta et al. 2007; Yttri et al. 2007; Begum et al. 2006; Mouli et al. 2006; Oanh et al. 2006; Gupta and Kumar 2006: Duan et al. 2005: Sillanpaa et al. 2005). The aerosol dynamics (i.e., distribution, composition, sources, and impacts) in developing countries like India are complicated due to scarce aerosol data, high emissions, and severe health hazards. In this work, concentration variations, composition, and sources of the chemical species associated with ambient particulates in the most polluted city, Raipur, during the winter season of years 2006-2013 are described.

Materials and Methods

Study Area

The capital city of the Indian state of Chhattisgarh, Raipur (21°24'N; 81°63' E) is surrounded by coal- and mineral-based industries and was therefore selected for the proposed investigation. The Urla industrial area is spread over around 300 ha in the northeast direction, with installation of wide range of various industries. Another industrial area, Silrara, is spread over around 900 ha in the east direction and is home to industries such as sponge iron units and ferroalloy plants, among others. Similarly, many cement plants are located in the east direction of the city within an approximate 70-km radius. The largest steel plant (Bhilai) is located around 20 km away in the northwest direction from Raipur. The Borai, Durg industrial area (around 192 ha) is located approximately 40 km away from Raipur in the north–south direction. At least 300 rice mills are running in neighboring cities such as Durg, Rajnandgaon, Tilda, Bhatapara, Arang, Rajim Nawapara, and Dhamtari, among others. A large industrial city, Korba, where several coal mines and thermal power plants operate, lies in the northeast direction around 150 km away from the city.

Sampling of Particulate

The PM₁₀ sample was collected by using portable air sampler University of California (UC Davis) over a 47-mm (Whatmann OMA, Darmstadt, Germany) quartz filter at flow rate of 19.4 Lmin⁻¹. Similarly, the suspended particulate matter (SPM) sample was collected by use of a high-volume air sampler (flow rate of $1.062.5 \text{ Lmin}^{-1}$) on the quartz filter (Whatmann OMA. 515.6 cm⁻²). They were heated beforehand (600°C for 6 h) to reduce blank values. The samplers were installed on the second floor of building (above a height of around 10 m from the ground) to minimize dust originating from ground-level activities. The filter was loaded in the sampler and run to collect particulates over 12 h (6:00 a.m.-6:00 p.m.). In similar ways, one collection blank was prepared. After collection, the dismounted filter was kept in petri dish and transferred to the laboratory. Similarly, a total of 18 PM₁₀ and SPM samples were collected from 15 sites of the city in the winter period of (2006-2007, Fig. 1). For temporal studies, 15 PM₁₀ samples from three locations i.e., Kota, Kalibadi, and Sankara, were collected in the month of January during years 2008-2013. All loaded filters were kept in a petri dish and refrigerated at -4° C until the analysis.

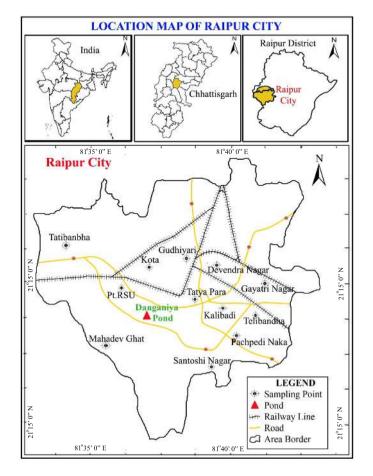


Fig. 1. Map of India with location of the sampling sites in Raipur City and major possible particle sources.

Analysis of Particulate Carbons

The thermal method was used for analysis of the organic (OC) and elemental carbon (EC) as prescribed in the literature (VDI German 1996). All carbons were converted into CO₂ by subsequent measurement with a nondispersive infrared (IR) detector. The sum total concentration of OC and EC was considered as total carbon (TC) in this work. The OC and EC analyses were carried out in triplicate. Their relative standard deviation (RSD) values were observed within $\leq \pm 3.0\%$.

Analysis of Elements

The proton-induced X-ray emission (PIXE) technique was used for measurement of the metal content of the filter. The Istituto Nazionale di Fisica Nucleare Tandetron (INFN, Florence, Italy) accelerator 3-MV facility in Florence, Italy, was used as external beam for the excitation of the elements (Calzolai et al. 2006). The RSD values for the triplicate analysis of the elements fell within $\leq \pm 5\%$.

Analysis of Water-Soluble lons

The ion content of PM was leached out with deionized water (15 mL, 0.054 μ S cm⁻¹) by sonicating at 60°C for 1 day (around 24 h). An ion chromatograph (Dionex DX120, Sunnyvale, California) equipped with anion and cation separation columns and a conductivity detector was used for analysis of the ions. An aliquot of the filtered sample (200 μ L) was injected into the chromatograph for quantifying the ions. The bicarbonate (9 mM) and methane sulfonic (20 mM) eluents were employed as leaching agents for the anions and cations, respectively. The RSD values for triplicate analysis of ions were observed between ±5% and 10%.

Quality Assurance and Quality Control

The precision of the portable air sampler was determined by using a 2.300-sequential speciation air sampler [Partisol model (Thermo Fisher Scientific Air Quality Instruments, Franklin, Massachusetts)]. Similarly, a total of six samples (2×3) by both samplers were collected. The masses of the PM were determined by a Mettler M3 electronic (Columbus, Ohio) balance after blank correction. The RSD values (n = 3) for PM₁₀ collections with the portable and Partisol samplers fell in between $\pm 2.9\%$ and 2.2%. respectively. The carbon analyzer was calibrated with known quantities of propane gases and CO₂ for analysis of the samples. The sensitivity of the PIXE technique for studied elements was standardized by using the reference aerosol filter sample. For ion chromatographic analysis, the calibration curves were prepared by injecting different concentrations of the standard solutions (Emanuel Merck, Darmstadt, Germany) for determination of ions. The precision of the analysis in the proposed work was expressed in terms of standard deviation (\pm SDT).

Results and Discussion

Meteorology

The meteorology remarkably influences the PM concentration in ambient air. The values of temperature (*T*), relative humidity (RH), wind speed (WS), and sunshine (SS) during the period December 2006–February 2007 were recorded in the range of 16.7° C–23.4°C, 52%–70%, 1.5–5.9 km h⁻¹, and 5.3–9.4 h day⁻¹ with an average value of 20°C, 59%, 2.7 km h⁻¹, and 8.2 h day⁻¹, respectively. The lowest wind speed (around 1.0 km h^{-1}) was

			PM			
S. No.	Sampling date	Sampling site	Туре	Mode	Concentration in air (µg m ⁻³)	
1	December 12, 2006	Devendra Nagar (DN)	CA	PM ₁₀	696	
2	January 3, 2007	Gudiari (G)	CA	PM_{10}	650	
3	January 4, 2007	Telebandha (TL)	CA	PM_{10}	321	
4	January 6, 2007	Gayatri Nagar (GN)	RA	PM_{10}	265	
5	January 8, 2007	Santoshi Nagar (SN)	RA	PM_{10}	577	
6	January 9, 2007	Kalibadi (KB)	CA	PM_{10}	462	
7	January 10, 2007	Pachperi Naka (PN)	CA	PM_{10}	308	
8	January 12, 2007	Dangania (D)	RA	PM_{10}	221	
9	January 14, 2007	Mahadevghat Road (MR)	RA	PM_{10}	510	
10	January 15, 2007	RS University (RSU)	RA	PM_{10}	295	
11	January 16, 2007	Tatibandh (TB)	RA	PM_{10}	304	
12	January 19, 2007	Kota (K)	RA	PM_{10}	353	
13	January 21, 2007	Tatyapara (TP)	CA	PM_{10}	420	
14	February 7, 2007	Tara Industrial area (TI)	IA	PM_{10}	384	
15	February 8, 2007	Sankara (SK)	IA	PM_{10}	760	
16	February 5, 2007	Sankera (SK)	IA	SPM	1,266	
17	February 5, 2007	Siltera (SL)	IA	SPM	1,150	
18	February 7, 2007	Tara Industrial area (TI)	IA	SPM	1,577	

Note: CA = commercial area; IA = industrial area; and RA = residential area.

observed during months of December and January. Thereafter, the wind speed steadily increased at rate of around 1.5 km h⁻¹, reaching maxima (round 10 km) in the months of June and July. After July, the wind speed again steadily decreased to its minima in the month of December. Due to the adverse climatic conditions during the months of December and January, air pollutants could not export to the outside environment and therefore increased several times higher than the mean values, causing several climatic and health hazards.

Distribution of Particulate Matter

High PM content in the ambient air is responsible for the origin of complex environmental and health issues in urban areas. The PM concentration in the air during period from December 2006 to February 2007 is summarized in Table 1. Wide PM₁₀ (n = 15) and SPM (n = 3) concentration variations in the air were observed: 221–760 and 1,150–1,577 μ g m⁻³ with an average value of 435 ± 168 and $1,331 \pm 221 \ \mu g m^{-3}$, respectively. The maximum PM₁₀ mass concentration in the air was seen in the industrial (i.e., Sankara) and commercial areas (i.e., Devendra Nagar and Gudhiyari). In contrast, a significant high mass concentration of PM₁₀ in sites such as Santoshi Nagar and Mahadev ghat were also observed, but these were expected due to transport of effluents from the Bhilai Steel Plant. The PM₁₀/SPM average ratio was observed to be 0.33. The wind speed of the air mass varied around $1.0-10 \text{ km h}^{-1}$ with the minimum and maximum value in December and July, respectively. The concentration of PM₁₀ had a good negative correlation (r = -0.86) with the wind speed of the air mass. The minimum PM₁₀ concentration in the air during July was marked due to the high wind speed ($\geq 10 \text{ km h}^{-1}$) and the PM₁₀ removal with rain. The average air-quality standard for urban areas proposed by the USEPA was used as the air-quality criteria. The PM₁₀ concentration in Raipur City during the winter period was found to be around eightfold greater than the permissible limit of 50 μ g m⁻³ (USEPA 2004). Severe ambient particulate pollution in Raipur City was identified, which was much higher than values observed in other parts of the country (Gupta et al. 2007; Begum et al. 2006; Gupta and Kumar 2006; Oanh et al. 2006).

Distribution of Carbon

The major fractions of the aerosols are composed of elemental carbon [EC or black carbon (BC)] and organic carbon. Among them, EC is an optically active component of the PM, being the second culprit for climate change after CO₂. In addition, they also adsorbed heavy metals and organic compounds. The OC were composed of a variety of organic compounds of climatic and environmental interests. The concentration of OC_{10} , EC_{10} , TC_{10} , OC_{SPM}, EC_{SPM}, and TC_{SPM} in the air in the range of 14-109, 11–114, 25–219, 20–35, 41–101, and 64–136 μ g m⁻³, respectively (Table 2). The OC_{10}/OC_{SPM} and EC_{10}/EC_{SPM} average ratios were found to be 1.37 and 0.60, respectively. Tremendously high concentrations of EC were observed in the industrial (i.e., Sankara) and commercial areas (i.e., Devendra Nagar and Gudhiyari), as expected due to emissions from industrial sources. The EC concentration with the PM₁₀ had fair correlation (r = 0.68). However, the maximum OC concentration was observed in the commercial areas, i.e., Devendra Nagar and Gudhiyari, which may be due to vehicle emissions. The OC concentration with the PM₁₀ had fair correlation $(r^2 = 0.84)$ with the PM₁₀. However, the concentration of OC and EC with the wind speed of air mass had a fair negative correlation (r = -0.81 to -0.85 to 0.83) with their lowest concentrations in the month of July, similar to the PM₁₀ scenario. The average ratios of OC_{10}/PM_{10} and EC_{10}/PM_{10} were observed to be 0.08 and 0.09, respectively. The OC_{10}/EC_{10} ratio was between 0.28 and 1.74, with an average value of 0.99 ± 0.36 . The concentrations of the EC and OC seem to be much higher than the values observed in other locations of the country and world (Gupta et al. 2017; Costa et al. 2016; Ding et al. 2016; Xu et al. 2015; Yttri et al. 2007; Duan et al. 2005; Sillanpaa et al. 2005; Latha and Badrinath 2003).

Distribution of Major Elements

The major fraction of the aerosols was composed of crustal elements. The ambient air concentrations of Al, P, Ti, Cr, Mn, Fe, Cu, Zn, and Zr (associated with PM₁₀) varied as 6.5–20.3, 0.2–0.4, 1.0–2.9, 0.1–0.2, 0.5–2.4, 14–78, 0–0.2, 0–1.3, and 0–0.3 μ g m⁻³, respectively (Table 3).

Table 2. Mass concentration of organic carbon (OC), black carbon (BC), and total carbon (TC)

	Concentration in air $(\mu g m^{-3})$			Concentration in PM (%)		
S. No.	OC	EC	TC	OC	EC	TC
1	109	110	219	15.7	15.8	31.5
2	107	96	203	16.5	14.8	31.3
3	28	25	53	8.7	7.8	16.5
4	27	32	59	10.2	12.1	22.3
5	33	19	52	5.7	3.3	9.0
6	38	30	68	8.2	6.5	14.7
7	26	19	45	8.4	6.2	14.6
8	14	11	25	6.3	5.0	11.3
9	26	28	54	5.1	5.5	10.6
10	20	22	42	6.8	7.5	14.3
11	15	23	38	4.9	7.6	12.5
12	15	16	31	4.2	4.5	8.7
13	25	26	51	6.0	6.2	12.2
14	18	39	57	4.7	10.2	14.9
15	32	114	146	4.2	15.0	19.2
Mean PM ₁₀	36	41	76	7.7	8.5	16.2
$\pm SD$	30	35	61	3.8	4.1	7.1
16	23	41	64	1.8	3.2	5
17	20	63	83	1.7	5.5	7.2
18	35	101	136	2.2	6.4	8.6
Mean SPM	26	68	94	1.9	5.0	6.9
±SD	8	30	37	0.3	1.7	1.8

Note: SD = standard deviation.

Table 3. Mass concentration of trace elements in ambient air ($\mu g m^{-3}$)

								40	<i>,</i>
S. No.	Al	Р	Ti	Cr	Mn	Fe	Cu	Zn	Zr
1	13.0	0.2	2.0	0.1	2.4	28	0.20	1.2	0.2
2	10.2	0.4	1.6	0.1	1.3	26	0.10	0.3	0.1
3	7.3	0.2	1.2	0.1	1.0	16	0.01	0.3	0.2
4	6.5	0.3	1.0	0.1	0.5	16	0.01	0.2	0.0
5	15.6	0.3	2.5	0.1	1.2	39	0.01	0.3	0.3
6	11.3	0.4	1.6	0.1	1.0	25	0.05	0.2	0.2
7	6.8	0.2	1.2	0.1	0.5	14	0.01	0.2	0.3
8	6.5	0.3	1.0	0.1	0.8	15	0.02	0.2	0.2
9	12.9	0.3	1.9	0.2	2.2	38	0.07	0.6	0.2
10	8.4	0.2	1.2	0.1	1.1	20	0.04	0.2	0.3
11	9.7	0.4	1.5	0.1	1.1	19	0.01	0.3	0.3
12	10.2	0.3	1.6	0.2	0.8	24	0.04	0.3	0.3
13	13.7	0.3	1.9	0.1	0.9	26	0.04	0.0	0.3
14	10.6	0.4	2.0	0.1	0.6	28	0.04	0.2	0.3
15	20.3	0.3	2.9	0.2	2.2	78	0.04	1.3	0.2
Mean PM ₁₀	10.9	0.3	1.7	0.1	1.2	27.5	0.05	0.4	0.2
$\pm SD$	3.8	0.1	0.5	0.0	0.6	15.9	0.05	0.4	0.1
16	20.7	0.1	2.8	0.2	2.7	82	1.80	0.2	0.2
17	15.5	0.2	2.4	0.2	0.6	54	0.10	0.6	0.1
18	23.6	0.3	3.9	0.3	2.7	87	1.10	0.1	0.2
Mean SPM	19.9	0.2	3.0	0.2	2.0	74.3	1.0	0.3	0.2
\pm SD	4.1	0.1	0.8	0.1	1.2	17.8	0.9	0.3	0.1

The concentration of examined elements in the SPM varied as 15.5–23.6, 0.1–0.3, 2.4–3.9, 0.2–0.3, 0.6–2.7, 54–87, 0.1–1.8, 0.1–0.6, and 0.1–0.2 μ g m⁻³, respectively (Table 3). Among them, the concentration of elements Al, Ti, Mn, Fe, Cu, and Zn with PM₁₀ had a fair correlation (r = 0.63 - 0.83), unlike P, Cr, and Zr. However, their concentrations with wind speed had a negative partial correlation (r = -0.45 to -0.47). The concentration ratio of the corresponding elements (associated with SPM and PM₁₀) was found to be 1.9, 0.7, 1.8, 2.0, 1.7, 2.7, 20, 0.8, and 1.0, respectively.

Table 4. Mass concentration of water aerosol soluble ions in ambient air ($\mu g m^{-3}$)

S. No.	Na ⁺	\mathbf{K}^+	Mg^{2+}	Ca^{2+}	NH_4^+	SO_4^{2-}	NO_3^-	Cl-
1	3.8	15.9	1.5	26.6	1.9	38.4	14.3	12.0
2	4.2	9.3	1.5	29.4	0.8	17.7	13.5	13.9
3	5.3	2.9	1.0	21.1	1.7	14.9	8.6	3.9
4	3.0	2.4	0.7	16.4	0.8	8.5	7.6	3.1
5	10.8	2.3	1.1	31.0	0.2	11.5	5.8	3.3
6	6.2	4.7	1.2	25.7	2.9	23.2	10.5	4.2
7	6.2	3.7	1.0	20.4	1.2	12.9	6.0	3.2
8	3.8	1.7	0.6	11.8	1.2	11.0	3.5	1.0
9	5.9	3.5	1.3	30.2	4.0	28.6	13.9	2.2
10	5.9	4.0	1.3	17.6	4.6	25.7	11.5	5.1
11	6.5	1.8	0.8	12.6	0.8	12.1	6.2	2.9
12	8.1	2.6	1.3	22.9	0.7	15.5	6.7	2.7
13	6.3	2.8	1.1	19.9	0.8	11.5	5.4	4.0
14	8.0	2.4	1.2	11.9	0.9	12.7	4.6	3.4
15	4.4	3.2	1.3	11.5	1.5	17.0	4.3	5.5
Mean PM ₁₀	5.9	4.2	1.1	20.6	1.6	17.4	8.2	4.7
\pm SD	2.0	3.7	0.3	6.9	1.3	8.2	3.7	3.5
16	1.5	1.5	0.6	9.0	1.3	12.7	3.3	0.8
17	1.8	0.8	0.7	10.5	0.2	8.7	1.9	0.6
18	2.2	2.9	1.3	14.0	0.8	18.7	6.1	1.4
Mean SPM	1.8	1.7	0.9	11.2	0.8	13.4	3.8	0.9
$\pm SD$	0.4	1.1	0.4	2.6	0.6	5.0	2.1	0.4

Among them, the crustal elements Al, Ti, Cr, Mn, Fe, and Cu were moderately to strongly enriched in the SPM modes. The maximum concentration of the elements was seen in the Tara and Sankara industrial areas, probably due to emissions from industrial sources. Significantly high metal concentrations in the ambient air of Raipur City were halted, which were observed to be much higher than values detected in other parts of the country and world, probably due to the operation of several coal- and mineral-based industries in the study area (Pan et al. 2015; Yadav and Satsangi 2013; Khillare and Sarkar 2012; Lakhani et al. 2008; Mouli et al. 2006).

Distribution of Water-Soluble lons

Ions in the ambient air were contributed by both local and distant sources. Remarkably high concentrations of ions (Cl^- , NO_3^- , SO_4^{2-} , NH_4^+ , Na^+ , K^+ , Mg^{2+} , and Ca^{2+}) associated with PM_{10} were identified. These ions were water-soluble and affected the quality of the rain and surface water. The concentrations of the aforementioned investigated ions varied in the range of 1.0-3.9, 3.5-14.3, 8.5-38.4, 0.2–4.6, 3.0–10.8, 1.7–15.9, 0.6–1.5, and 11.5–31.0 μ g m⁻³, respectively (Table 4). Similarly, concentrations of examined ions associated with SPM were in the range of 0.6-1.4, 1.9-6.1, $8.7-18.7, 0.2-1.3, 1.5-2.2, 0.8-2.9, 0.6-1.3, and 9.0-14.0 \ \mu g m^{-3},$ respectively. All ions except Na^+ and NH_4^+ had a poor correlation with the PM_{10} concentration (r = 0.36 - 0.73). Their concentration with wind speed had a partial negative correlation ($r \approx$ -0.45), with the minimum values in July, unlike Na⁺ and NH₄⁺ because they were contributed by other sources. PM₁₀/SPM concentration ratios of 5.2, 2.2, 1.3, 2.0, 3.3, 2.5, 1.2, and 1.8 for Cl⁻, NO_3^- , SO_4^{2-} , NH_4^+ , Na^+ , K^+ , Mg^{2+} , and Ca^{2+} were observed. All ions were moderately to significantly enriched in the coarse modes. The $\sum PM_{10}WSI_8$ concentrations varied from 16.9 to 309 μ g m⁻³ with an average value of $119 \pm 108 \ \mu g m^{-3}$. Significant higher loading of the ions in the air of commercial and residential locations (i.e., Devendra Nagar, Gudhiyari, Santoshi Nagar, and Mahadev Ghat) were observed. The concentration of SO_4^{2-} with the NO_3^- in the PM₁₀ had a fair correlation (r = 0.81), which may

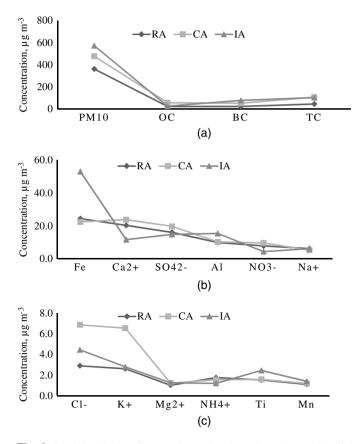


Fig. 2. Spatial variation of PM and associated species. RA = residential area; CA = commercial area; IA = industrial area; OC = organic carbon; BC = black/elemental carbon; and TC = total carbon.

be due to their formation by similar processes. The \sum anion/ \sum cation average equivalent ratio for the PM₁₀ and SPM was 0.40 and 0.45, respectively, indicating the alkaline nature of the air. The relative abundance of 7%, 13%, 27%, 3%, 9%, 7%, 2%, and 32% for Cl⁻, NO₃⁻, SO₄²⁻, NH₄⁺, Na⁺, K⁺, Mg²⁺, and Ca²⁺ in the PM₁₀ was identified. The major fraction (59%) of the PM was contributed by ions SO₄²⁻ and Ca²⁺. Similar concentration ranges for these ions (associated with the coarse particulates) as those recorded for the ambient air of other parts of the country and China were observed (Tao et al. 2014; Satsangi et al. 2013; Singh et al. 2013; Kulshrestha et al. 2009).

Spatial Distribution

Concentrations of the chemical species related to PM in the air are affected by the meteorology and geography of the area. Higher concentrations of carbon and metals were observed in the industrial areas, probably due to coal and mineral combustion emissions (Fig. 2). In contrast, the highest concentrations of species OC, Ca, K, Cu, Cl⁻, NO₃⁻, and SO₄²⁻ were seen in the commercial areas, probably due to contributions from vehicular emissions and atmospheric inputs. However, the higher concentration of two species, Na and NH₄⁺, was marked in the residential area, which may be due to input from municipal wastes.

Temporal Variations

The concentrations of PM and associated species in the air are related to the frequency and strength of emission inventories. The vast industrialization and urbanization of Raipur City is occurring due to the easily available of natural resource materials,

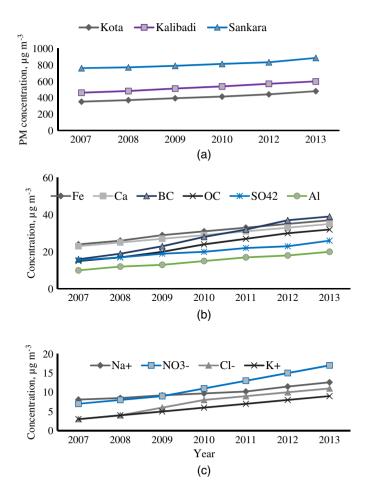


Fig. 3. Temporal variation of PM and associated elements. OC = organic carbon; and BC = black/elemental carbon.

i.e., coal, biomass, and minerals. The concentrations of PM and associated species were found to increase during 2006–2013, as expected due to increases in the quantum of population, industry, vehicles, and construction work, as shown in Fig. 3.

Composition of Aerosols

The urban aerosols are composed of carbons, metals, ions, and organics, among others. The fraction of OC, EC (BC), Fe, Ca^{2+} , SO₄²⁻, Al, NO₃⁻, Na⁺, Cl⁻, K⁺, NH₄⁺, Ti, Mg, Mn, P, Zn, Zr, Cr, and Cu in the PM₁₀ (n = 15) was found to fall in the ranges of 4.2%-16.5%, 3.3%-15.8%, 4.0%-10.3%, 1.5%-6.2%, 2.0%-8.7%, 1.6%-3.3%, 0.6%-3.9%, 0.6%-2.3%, 0.1%-2.1%, 0.4%-2.3%, 0.03%-1.6%, 0.3%-0.5%, 0.2%-0.4%, 0.2%-0.4%, 0.03%-0.14%, 0.01%-0.17%, 0.01%-0.10%, 0.01%-0.06%, and 0.01%-0.03% with average values of 7.7%, 8.5%, 6.2%, 5.1%, 4.2%, 2.6%, 2.0%, 1.5%, 1.0%, 0.9%, 0.4%, 0.4%, 0.3%, 0.3%, 0.08%, 0.08%, 0.06%, 0.03%, and 0.01%, respectively. However, in the SPM, the fraction of all species except Cu was found in the lower orders. The total fraction of the carbons, metals, and water-soluble ions in the PM₁₀ and SPM estimated was found to be 41.4% and 22.1%, respectively. The uncertainty observed in the PM10 and SPM was 58.6% and 77.9%, respectively.

Sources

Local industries, including Bhilai Steel Plant, fully encircle Raipur City. The highest concentrations of PM_{10} , EC, K⁺, Fe, and Al were

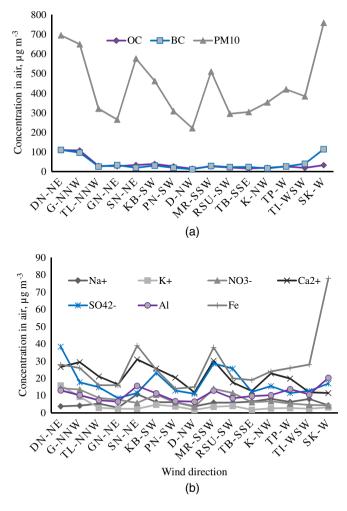


Fig. 4. Influence of wind direction in distribution of PM and associated species. NE = northeast; NNW = north–northeast; NE = north east; SW = southwest; NW = northwest; SSW = south–southwest; SSE = south–southeast; NW = northwest; W = west; and WSW = west–southwest.

found when the air masses come from the northeast direction due to influence of local industries (Fig. 4). The significant concentration of OC and Cl⁻ was seen when the air masses come from the north– northwest direction due to industrial influence of the Bhilai Steel Plant. The remarkable high mass concentration of the secondary aerosol species, i.e., NO_3^- and SO_4^{2-} , was seen when the air masses flowed from the south–southwest direction due to thermal power plants running in Nagpur and Chandrapur areas. The highest concentration of Ca²⁺ was assumed to be due to running of several rice mills in the south–southwest direction of the city.

The widely used principle component analysis-multilinear regression analysis (PCA-MLRA) was used to apportion the emission sources contributing to the ambient particulates in Raipur (USEPA 1997; John et al. 1994). Four factors (Table 5) were identified as per criteria of cumulative percentage variance >80% and eigenvalue of >2 (Stevens 1996). The Varimax rotated factor analysis showed four possible different sources for the OC, BC, metallic elements (ME), and water soluble ions (WSI). Factor 1 comprised of anthropogenic sources with influence of traffic for OC, BC, K⁺, Mg²⁺, SO²⁻₄, NO⁻₃, and Cl⁻ with a variance of 37%. A substantial amount of coal (>10 Mt year⁻¹) was burned in Raipur City and surrounding areas by various industries. This means that the pollutants originated by the fuel combustion as well

Table 5. Factor loadings (varimax normalized)

Species	Factor I	Factor 2	Factor 3	Factor 4
OC	0.95	0.06	-0.02	0.06
BC	0.70	0.41	-0.41	-0.17
Na ⁺	-0.20	0.13	0.86	0.12
K^+	0.89	0.04	0.02	0.28
Mg^{2+}	0.62	0.30	0.30	0.35
Ca ²⁺	0.43	0.02	0.57	0.44
NH_4^+	-0.03	0.03	0.04	0.87
SO_4^{2-}	0.54	0.19	-0.03	0.74
NO ₃	0.61	-0.10	0.24	0.67
Cl-	0.90	0.07	0.21	0.08
Al	0.08	0.94	0.13	-0.03
Р	0.02	0.11	0.47	-0.51
TI	0.08	0.93	0.15	-0.10
Cr	-0.03	0.62	-0.42	0.31
Mn	0.27	0.69	-0.43	0.32
Fe	0.06	0.93	0.09	-0.06
Cu	-0.11	0.09	-0.81	0.06
Zn	0.43	0.68	-0.12	0.07
Variance	5.42	4.33	2.77	2.69
Total	0.28	0.23	0.15	0.14

as blown dust. Factor 2 included mineral sources for Al, Ti, Cr, Mn, Fe, and Zn with a variance of 19%. This pattern of metals indicated the originated from the roasting of the raw materials and blown dust. Factor 3 traced a consistent source for Na⁺ and Ca²⁺ with a variance of 14%. Mainly three sources, i.e., roasting of pyrite and dolomite, fossil fuel combustion, and blown dust, contributed to a continuous presence of Na⁺ and Ca²⁺ in the coarse particulates. Factor 4 predicted secondary sources for specie NH₄⁺, SO₄²⁻, and NO₃⁺ with a variation of 11%.

Conclusion

The severe particulate air pollution (more than eightfold higher than the prescribed value), especially in the period December to January in Raipur City, India, was observed to be due to adverse climatic weather. The particulates were composed of high fractions of BC (9.4%), OC (11.2%), metals (9.2%), and water-soluble ions (26.2%). Anthropogenic activities and atmospheric transformations were expected to be major sources of PM and associated species in the air. Wind speed was observed as one major meteorological factor responsible for the halting of winter particulate pollution.

Acknowledgments

This work was supported through Grant No. ES/48/ICRP/008/2002 funded by the Department of Science and Technology, New Delhi. It was also partially funded by the Spanish Ministry of Education and Science, Madrid.

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